

**Electronic Supporting Information for the Article:**

**A Graphene-based Fluorescent Nanoprobe for Silver(I) Ions  
Detection by using Graphene Oxide and a Silver-Specific  
Oligonucleotide**

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**Experimental Section**

**Materials**

The DNA oligonucleotide (SSO, 5'-FAM-**CTCTCTCTCTTCATTTTCAACACAACACAC-3'**) was synthesized by Takara Biotechnology Co. (Dalian, China) and purified by HPLC. The colored bases in this sequence are the binding site for Ag<sup>+</sup>. 3-morpholinopropanesulfonic acid (MOPs) was purchased from sigma. AgNO<sub>3</sub>, LiNO<sub>3</sub>, Ca(NO<sub>3</sub>)<sub>2</sub>, Mg(CH<sub>3</sub>COO)<sub>2</sub>, Cu(NO<sub>3</sub>)<sub>2</sub>, Zn(NO<sub>3</sub>)<sub>2</sub>, Cd(NO<sub>3</sub>)<sub>2</sub>, Co(NO<sub>3</sub>)<sub>2</sub>, Mn(CH<sub>3</sub>COO)<sub>2</sub>, Ni(NO<sub>3</sub>)<sub>2</sub>, Pb(NO<sub>3</sub>)<sub>2</sub>, Hg(NO<sub>3</sub>)<sub>2</sub> and FeCl<sub>3</sub> were of analytical grade and used as received. All solutions were prepared with Milli-Q water (18 MΩ cm<sup>-1</sup>) from a Millipore system. Environmental water sample was taken from a nearby river and centrifuged to remove the insoluble impurities.

**Preparation of GO**

GO was synthesized from graphite powder based on the Hummer's method. Briefly, graphite power (4 g) was oxidized in a hot solution (80°C) of concentrated H<sub>2</sub>SO<sub>4</sub>(24 mL) containing K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> (8 g), and P<sub>2</sub>O<sub>5</sub> (8 g). The resulting dark blue mixture was thermally isolated and slowly cooled to room temperature over a period of 6 h. The

mixture was diluted to 300 mL, and then filtrated with a filter membrane of 0.22  $\mu$ m (Generay Biotech Co., Ltd., Shanghai, China) and dried overnight at 60°C. These preoxidized graphite powder (2 g) was added to 92 mL of cold H<sub>2</sub>SO<sub>4</sub> (0°C), to which KMnO<sub>4</sub> (12 g) was gradually added under continuous stirring in ice-bath. After 15 min, NaNO<sub>3</sub> (2 g) was added to the mixture. The solution was further stirred for 2 h at 35 °C and distilled water (200 mL) was added. The reaction was stopped with the addition of a mixture of 560 mL of distilled water and 10 mL of H<sub>2</sub>O<sub>2</sub> (30 %). The product was washed with HCl (1:10) and then with water, and then suspended in distilled water. The brown dispersion was extensively dialyzed to remove residual metal ions and acids, and then exfoliated via sonication for 1.5 h (300 W). Unexfoliated graphite oxide was removed by centrifugation (3000 rpm, 5 min) using Centrifuge himac-CF 16RX (Hitachi, Japan).

### Fluorescence assay for Ag(I) Ions

Ag(I) ions of different concentrations were incubated in MOPS buffer (1 mL, pH 7.0) containing 50 mM of NaNO<sub>3</sub> and 10 nM of SSO for 5 min at 23°C. Then 20 $\mu$ L of GO (0.5  $\mu$ g/ $\mu$ L ) was added to this mixture and the fluorescence measurement was carried out 2 min after the GO addition at 23°C.

### Instruments

The fluorescence spectra were measured using a Hitachi F-4500 spectrophotometer equipped with a Xenon lamp excitation source. The excitation wavelength was  $\lambda$ =494 nm, and the fluorescence measurements were carried out at 23°C.

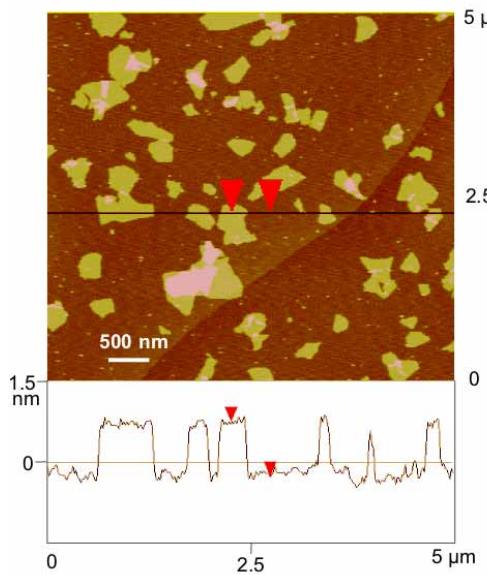


Figure S1 AFM tapping-mode image of the as-prepared GO sheets and the height profile along the dashed line in panel.

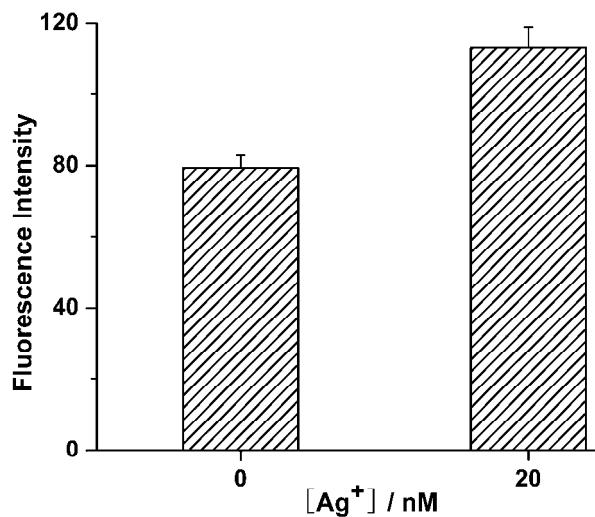


Figure S2 Comparison of the response of LOQ (20 nM of Ag<sup>+</sup>) with the background signal (0 nM of Ag<sup>+</sup>).

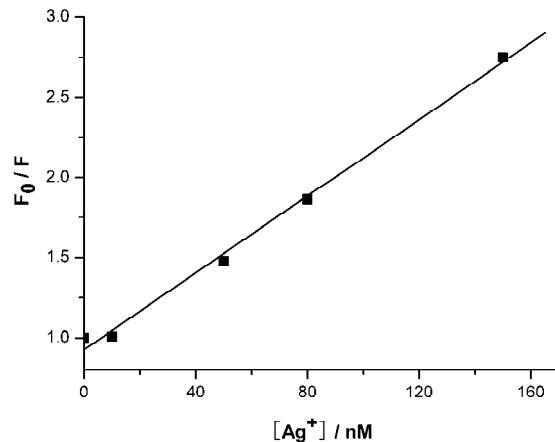


Figure S3 Stern-Volmer plot of the FAM-labeled SSO probe quenched by  $\text{Ag}^+$ .