# Easy Oxidatively Induced Silicon-Carbon Bond Activation in Organoplatinum Chemistry

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#### **Electronic Supporting Information**

#### Synthetic procedures

The complex  $[Pt_2Me_4(\mu-SMe_2)_2]$ ,  $[Pt_2(CD_3)_4(\mu-SMe_2)_2]$  and ligand  $Me_2Si(2-C_5H_4N)_2$ , bps, were prepared according to the literature.<sup>1,2</sup>

**[PtMe<sub>2</sub>(bps)], 1**. To a stirring solution of bps (0.75 g, 3.48 mmol) in ether (10 mL) was added [Pt<sub>2</sub>Me<sub>4</sub>( $\mu$ -SMe<sub>2</sub>)<sub>2</sub>] (1.0 g, 1.74 mmol). After 1h, the product precipitated from solution as a pale brown solid, which was separated, washed with ether (3 x 2 mL) and pentane (3 x 2 mL), and dried under vacuum. Yield: 88%. Anal. Calcd. for C<sub>14</sub>H<sub>20</sub>N<sub>2</sub>PtSi: C, 38.26; H, 4.59; N, 6.37. Found: C, 38.53; H, 4.85; N, 6.33 %. NMR in CDCl<sub>3</sub>:  $\delta(^{1}H) = 0.71$  [s, 6H,  $^{2}J_{PtH} = 80$  Hz, PtMe]; 0.77 [s, 3H, SiMe]; 1.11 [s, 3H, SiMe]; 7.21 [dd, 2H,  $^{3}J_{HH} = 7$ , 5 Hz, H<sup>5</sup>]; 7.55 [d, 2H,  $^{3}J_{HH} = 7$  Hz, H<sup>3</sup>]; 7.67 [t, 2H,  $^{3}J_{HH} = 7$  Hz, H<sup>4</sup>]; 8.97 [d, 2H,  $^{3}J_{HH} = 5$  Hz,  $^{3}J_{PtH} = 25$  Hz, H<sup>6</sup>];  $\delta(^{13}C) = -18.5$  [PtMe]; -3.9 [SiMe]; -1.7 [SiMe]; 125.4 [C5]; 130.4 [C3]; 133.7 [C4]; 152.4 [C6]; 163.1 [C2]. Complex 1-d<sub>6</sub> was prepared similarly from [Pt<sub>2</sub>(CD<sub>3</sub>)<sub>4</sub>( $\mu$ -SMe<sub>2</sub>)<sub>2</sub>]. The <sup>1</sup>H NMR spectrum was the same except that the methylplatinum resonance at  $\delta(^{1}H) = 0.71$  was absent.

#### [Me(bps)Pt-OSiMe(2-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>PtMe<sub>3</sub>][B(OCH<sub>2</sub>CF<sub>3</sub>)(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>],

**2[B(OCH<sub>2</sub>CF<sub>3</sub>)(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>]**. To a solution of **1** (0.09 g, 0.45 mmol) in CF<sub>3</sub>CH<sub>2</sub>OH (5 mL) was added a solution of B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub> (0.11 g, 0.45 mmol) in CF<sub>3</sub>CH<sub>2</sub>OH (5 mL). The mixture was stirred for 2 d, then the volume of the solution was reduced to 2 mL, and the mixture was stored at 0 °C for 2 d., to give the product as colourless crystals, which were separated and dried under vacuum. Yield 37 %. Anal. Calcd. for C<sub>47</sub>H<sub>39</sub>BF<sub>18</sub>N<sub>4</sub>O<sub>2</sub>Pt<sub>2</sub>Si<sub>2</sub>.2H<sub>2</sub>O: C, 36.97; H, 2.84; N, 3.67. Found: C, 36.77; H, 2.32; N, 3.62 %. NMR in acetone-*d*<sub>6</sub>:  $\delta(^{1}\text{H}) = 0.15$  [s, 3H, <sup>2</sup>*J*<sub>PtH</sub> = 69 Hz, MePt<sup>IV</sup> *trans* N]; 0.64 [s, 3H, <sup>2</sup>*J*<sub>PtH</sub> = 75 Hz, MePt<sup>II</sup> *trans* N]; 0.95 [s, 3H, bps SiMe]; 1.17 [s, 3H, <sup>2</sup>*J*<sub>PtH</sub> = 68 Hz, MePt<sup>IV</sup> *trans* N]; 1.22 [s, 3H, OSiMe]; 1.23 [s, 3H, <sup>2</sup>*J*<sub>PtH</sub> = 78 Hz, MePt<sup>IV</sup> *trans* O]; 1.63 [s, 3H, bps SiMe]; 3.57 [q, 2H, <sup>3</sup>*J*<sub>HF</sub> = 12 Hz, CH<sub>2</sub>]; 7.01 [m, 1H, H<sup>5d</sup>]; 7.38 [m, 1H, H<sup>5a</sup>]; 7.61 [m, 1H, H<sup>5c</sup>]; 7.64 [m, 1H, H<sup>5b</sup>]; 7.74 [m, 1H, H<sup>3a</sup>]; 7.83 [m, 1H, H<sup>4d</sup>]; 7.88 [m, 1H, H<sup>3d</sup>]; 7.96 [m, 1H, H<sup>4a</sup>]; 8.07 [m, 2H, H<sup>3b</sup>, H<sup>4b</sup>]; 8.11 [m, 1H, H<sup>4c</sup>]; 8.72 [m, 1H, <sup>3</sup>*J*<sub>PtH</sub> = 21 Hz, H<sup>6b</sup>]; 8.96 [m, 1H, <sup>3</sup>*J*<sub>PtH</sub> = 65 Hz, H<sup>6a</sup>].  $\delta(^{19}\text{F}) = -75$  [br, 3F, CF<sub>3</sub>]; -133 [br m, 6F, F<sup>2</sup>, F<sup>6</sup>]; -164 [m, 3F, F<sup>4</sup>]; -168 [br, 6F, F<sup>3</sup>, F<sup>5</sup>].

[Me<sub>3</sub>Pt(HOSiMe(2-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>][PhCO<sub>2</sub>], 3c. To a solution of complex 1 (0.05 g, 0.12 mmol) in acetone (10 mL) was added dibenzoyl peroxide (0.03 g, 0.12 mmol). The mixture was stirred for 2 h., the volume was reduced to 1 mL, and pentane (5 mL) was added to precipitate the product as a white solid, which was separated, washed with ether (3 x 2 mL) and pentane (3 x 2 mL), and dried under vacuum. NMR in CD<sub>2</sub>Cl<sub>2</sub>:  $\delta(^{1}H) =$ 

1.04 [s, 3H, Si-Me]; 1.05 [s, 6H,  ${}^{2}J_{PtH}$  = 70 Hz, MePt *trans* N]; 1.14 [s, 3H,  ${}^{2}J_{PtH}$  = 75 Hz, MePt *trans* O]; 2.12 [s, 1H, OH]; 7.37 [m, 4H, H<sup>5</sup>,H<sup>m</sup>]; 7.46 [m, 1H, H<sup>p</sup>]; 7.75 [m, 2H, H<sup>3</sup>]; 7.80 [m, 2H, H<sup>4</sup>]; 7.95 [m, 2H, H<sup>o</sup>]; 8.58 [m, 2H,  ${}^{3}J_{PtH}$  = 20 Hz, H<sup>6</sup>].

## Monitoring of reactions by <sup>1</sup>H NMR and ESI-MS.

In a typical reaction, complex **1** (1 mg) was dissolved in CD<sub>3</sub>OD (1 mL), and then excess hydrogen peroxide (0.01 mL) was added to the solution to give [Me<sub>3</sub>Pt(HOSiMe(2-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>][OH], **3b**. The <sup>1</sup>H NMR spectrum was recorded immediately and then at intervals for several days. NMR in CD<sub>3</sub>OD after 4 min.:  $\delta(^{1}H) = 0.94$  [s, 3H, SiMe]; 0.98 [s, 6H, <sup>2</sup>*J*<sub>PtH</sub> = 69 Hz, MePt *trans* N]; 1.15 [s, 3H, <sup>2</sup>*J*<sub>PtH</sub> = 75 Hz, MePt *trans* O]; 7.43 [m, 2H, H<sup>5</sup>]; 7.80 [m, 2H, H<sup>3</sup>]; 7.86 [m, 2H, H<sup>4</sup>]; 8.62 [m, 2H, <sup>3</sup>*J*<sub>PtH</sub> = 20 Hz, H<sup>6</sup>];  $\delta(^{13}C) = -$ 10.7 [PtMe]; -10.4 [PtMe]; -4.6 [SiMe]; 123.4 [C5]; 129.1 [C3]; 136.5 [C4]; 146.5 [C6]; 165.8 [C2]. The similar reaction with **1**-*d*<sub>6</sub> to give **3b\*** gave identical <sup>1</sup>H NMR data except that the resonance at  $\delta(^{1}H) = 0.98$  was absent. The complexes decomposed slowly over several days.

Similar reactions in MeOH were monitored by ESI-MS: Complexes **3b** and **3b**\* gave an envelope of peaks at m/z = 456 and 462 respectively, as expected for  $[Me_3Pt(HOSiMe(2-C_5H_4N)_2]^+$  and  $[Me_3Pt(HOSiMe(2-C_5H_4N)_2]^+$ - $d_6$ , and also at 911 and 923 respectively, as expected for  $[H\{Me_3Pt(OSiMe(2-C_5H_4N)_2\}_2]^+$  and  $[H\{Me_3Pt(OSiMe(2-C_5H_4N)_2\}_2]^+$ - $d_{12}$ . **[PtI\_2Me\_2(bps)], 4**. To a stirred solution of **1** (0.02 g, 0.455 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) was added excess iodine (0.012 g). The mixture was stirred for 2 h., the solvent was removed under vacuum, and the red product was washed with pentane (3 x 3 mL) and ether (3 x 3 mL) and dried under vacuum. Yield 84%. NMR in CD<sub>2</sub>Cl<sub>2</sub>:  $\delta(^{1}H) = 0.81$  [s, 6H, Si-Me]; 2.39 [s, 6H,  $^{2}J_{PtH} = 71$  Hz, PtMe]; 7.42 [m, 2H, H<sup>5</sup>]; 7.78 [m, 2H, H<sup>3</sup>]; 7.86 [m, 2H, H<sup>4</sup>]; 9.36 [m, 2H,  $^{3}J_{PtH} = 25$  Hz, H<sup>6</sup>].

**Structure Determinations:** Data were collected at low temperature (150 K) using a Nonius Kappa-CCD area detector diffractometer. Details are in the cif files. Figures are shown with 25% thermal ellipsoids.

### Crystal data:

1:  $C_{14}H_{20}N_2PtSi$ , fw = 439.50, monoclinic, C2/c, a = 15.2967(7), b = 12.8601(6), c = 15.8369(8) Å,  $\beta$  = 102.191(3)°, V = 3045.1(3) Å<sup>3</sup>, Z = 8, d(calc) = 1.917 Mg m<sup>-3</sup>, R1 = 0.0469, wR2 = 0.1124 [I > 2s(I)].

**2**:  $C_{47}H_{39}BF_{18}N_4O_2Pt_2Si_2$ , fw = 1490.99, triclinic, P-1, a = 11.812(2), b = 13.535(3), c = 16.700(3) Å,  $\alpha = 79.94(3)$ ,  $\beta = 80.78(3)$ ,  $\gamma = 79.76(3)$ °, V = 2564.0(9) Å<sup>3</sup>, Z = 2, d(calc.) = 1.931 Mg m<sup>-3</sup>, R1 = 0.0437, wR2 = 0.0972 [I > 2s(I)].

**3c**:  $C_{21}H_{26}N_2O_3PtSi$ , fw = 577.62, monoclinic,  $P2_1/n$ , a = 14.305(3), b = 10.130(2), c = 14.985(3) Å,  $\beta$  = 99.04(3) °, V = 2144.6(7) Å<sup>3</sup>, Z = 4, d(calc) = 1.789 Mg m<sup>-3</sup>, R1 = 0.0298, wR2 = 0.0732 [I > 2s(I)].

4:  $C_{14}H_{20}I_2N_2PtSi$ , fw = 693.30, monoclinic,  $P2_1/n$ , a = 9.3692(3), b = 13.7960(5), c = 14.4724(5) Å,  $\beta = 100.964(2)^\circ$ , V = 1836.52(11) Å<sup>3</sup>, Z = 4, d(calc) = 2.507 Mg m<sup>-3</sup>, R1 = 0.0507, wR2 = 0.1369 [I > 2s(I)].

## References

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- 2 J.D. Scott and R.J. Puddephatt, Organometallics, 1983, 2, 1643.