

Supporting Information

Ligand Exchange Reactions in Cu(III) complexes: Mechanistic Insights by Combined NMR and DFT Studies

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Experimental section

All cuprate samples were prepared by a method described by John et al.¹ The synthesis of the cuprate was directly done in Et₂O-*d*₁₀ to exclude protonated Et₂O. The protonated Et₂O from the ¹³C-labelled MeLi solution was removed as much as possible before the addition to the Cu-salt suspension. Having the cuprate in hand it was cooled down to 170K and a solution MeI in DEE-*d*₁₀ was added depending on the synthesis. The amount of MeI was adjusted to the synthesis of the cuprate solution.

- (1) John, M.; Auel, C.; Behrens, C.; Marsch, M.; Harms, K.; Bosold, F.; Gschwind, R. M.; Rajamohanan, P. R.; Boche, G. *Chem. Eur. J.* **2000**, *6*, 3060-3068.

NMR Data Collection and Processing

The NMR spectra were recorded on a Bruker Avance 600 spectrometer equipped with a 5mm broadband triple resonance Z-gradient probe.

¹H, ¹³C HMBC measurements were carried out with a standard Bruker pulse program using 32 number of scans, 16 dummy scans, TD(F2) = 16k and TD(F1) = 400 with a relaxation delay of 2s. The processing parameters were TD(F1) = 1k and TD(F2) = 1k. The temperatures for all measurements were controlled by a Bruker BVTE 3900 temperature unit.

Increased formation of Me_4Cu^- applying an excess of MeLi

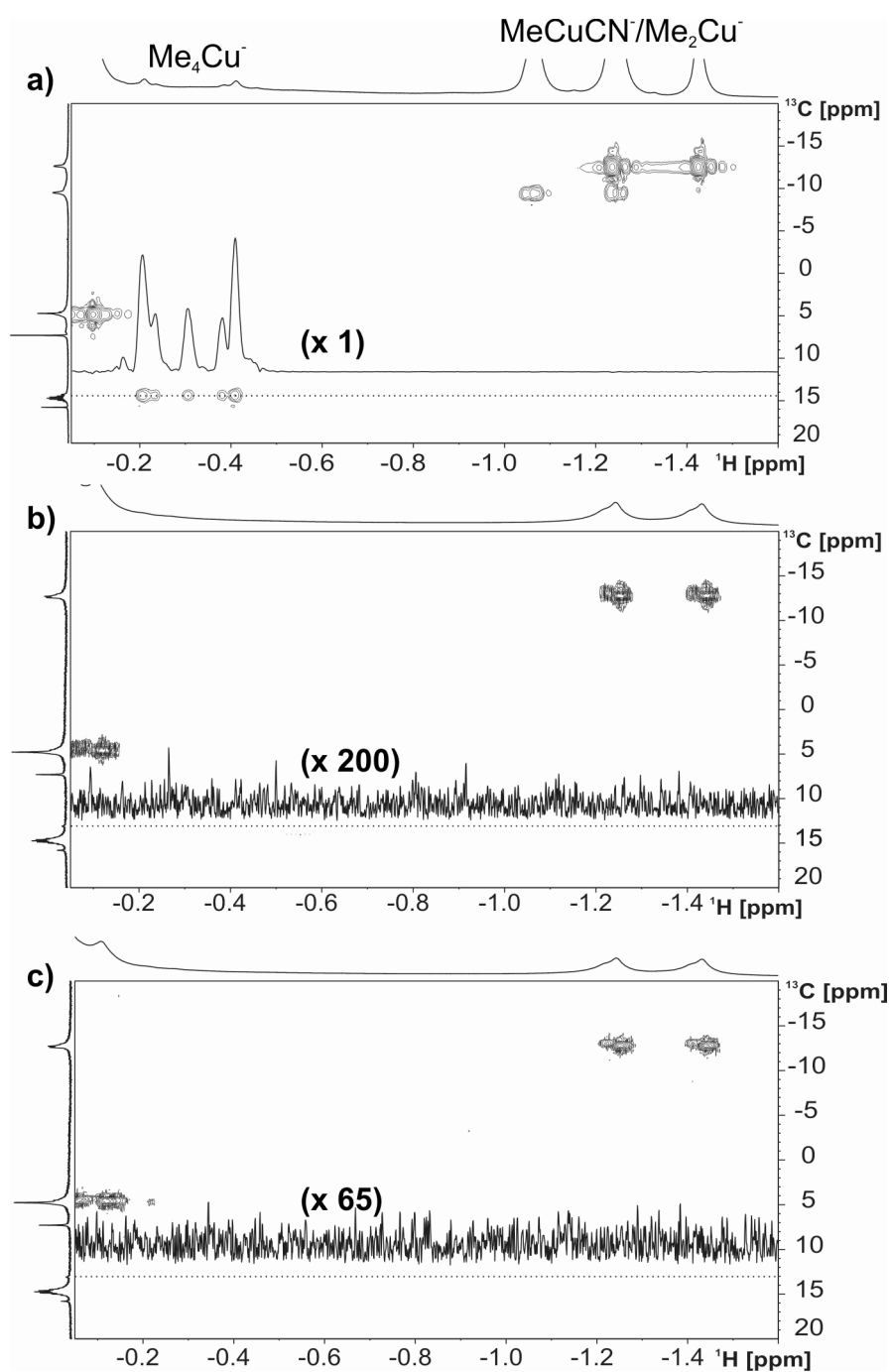


Fig. S1 Sections of ^1H , ^{13}C HMBC spectra at -63°C in diethyl ether showing the signals of Me_4Cu^- , MeCuCN^- and Me_2Cu^- . To visualize the increased concentration of Me_4Cu^- in the sample with excess of MeLi, in addition, the row of the HMBC spectrum showing Me_4Cu^- is presented as insert with their scaling factors given in brackets.

a) $\text{Cu}^{13}\text{CN} + 3 \text{ } ^{13}\text{CH}_3\text{Li} + \text{CH}_3\text{I}$

b) $\text{Cu}^{13}\text{CN} + 2 \text{ } ^{13}\text{CH}_3\text{Li} + \text{CH}_3\text{I}$ (number of scans identical to a)

c) $\text{Cu}^{13}\text{CN} + 2 \text{ } ^{13}\text{CH}_3\text{Li} + \text{CH}_3\text{I}$ (number of scans three times higher than in a)

Analysis of the isotopic pattern of ethane in the reaction of ^{13}C -labelled cyanocuprate converted with MeI and an excess of ^{13}C -labelled MeLi

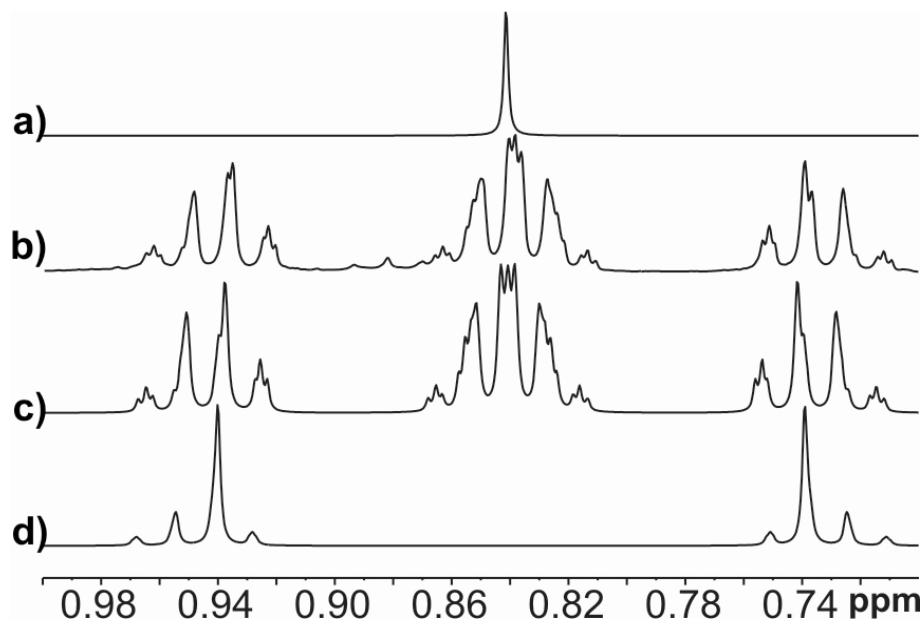


Fig. S2 Simulated ^1H spectra of a) $\text{H}_3\text{C}-\text{CH}_3$, c) $\text{H}_3\text{C}-^{13}\text{CH}_3$ and d) $\text{H}_3^{13}\text{C}-^{13}\text{CH}_3$ in comparison with b) the experimental spectrum of the resulting products of ^{13}C -labelled cyanocuprate converted with MeI and an excess of ^{13}C -labelled MeLi.

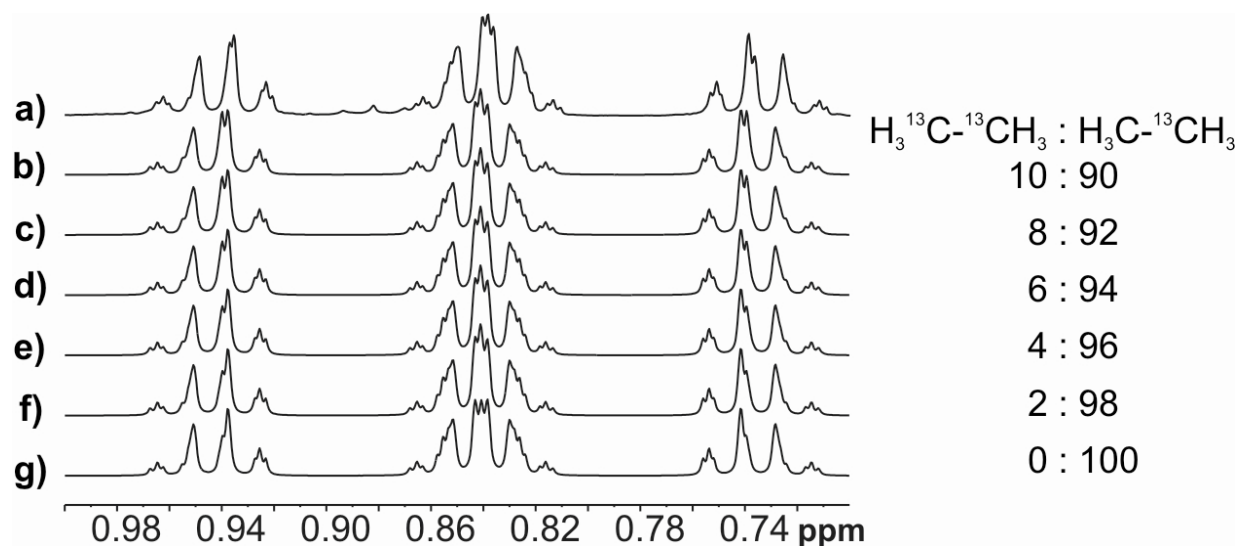


Fig. S3 a) Experimental spectrum of the resulting products of the ^{13}C -labelled cyanocuprate converted with MeI and an excess of ^{13}C -labelled MeLi in comparison with b-g) the simulated ^1H spectra of different mixtures of $\text{H}_3^{13}\text{C}-\text{CH}_3$ and $\text{H}_3\text{C}-^{13}\text{CH}_3$ (ratios given aside).

DFT functional Calculations

All calculations were done using the GAUSSIAN 03 package.² All geometry optimisations were performed with the DFT-method and the B3LYP hybrid functional, using SDD for copper, iodine and gold and 6-31+G(d) for all other atoms. Local minima have zero and transition states (TS) have one and only one imaginary frequency. The intrinsic reaction coordinate (IRC) analysis³⁻⁵ was carried out to confirm that stationary points are smoothly connected to each other. All energies used throughout are zero-point corrected and calculated for the gas-phase.

- (2) Gaussian 03, R. C.; Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Montgomery, J. A., Jr., T. V.; Kudin, K. N.; Burant, J. C.; Millam, J. M.; Iyengar, S. S.; Tomasi, J.; Barone, V.; Mennucci, B.; Cossi, M.; Scalmani, G.; Rega, N.; Petersson, G. A.; Nakatsuji, H.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Klene, M.; Li, X.; Knox, J. E.; Hratchian, H. P.; Cross, J. B.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Ayala, P. Y.; Morokuma, K.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Zakrzewski, V. G.; Dapprich, S.; Daniels, A. D.; Strain, M. C.; Farkas, O.; Malick, D. K.; Rabuck, A. D.; Raghavachari, K.; Foresman, J. B.; Ortiz, J. V.; Cui, Q.; Baboul, A. G.; Clifford, S.; Cioslowski, J.; Stefanov, B. B.; Liu, G.; Liashenko, A.; Piskorz, P.; Komaromi, I.; Martin, R. L.; Fox, D. J.; Keith, T.; Al-Laham, M. A.; Peng, C. Y.; Nanayakkara, A.; Challacombe, M.; Gill, P. M. W.; Johnson, B.; Chen, W.; Wong, M. W.; Gonzalez, C.; Pople, a. J. A.; Gaussian, I., Wallingford CT, 2004.
- (3) Fukui, K. *Acc. Chem. Res.* **1981**, *14*, 363-368.
- (4) Gonzalez, C.; Schlegel, H. B. *J. Chem. Phys.* **1989**, *90*, 2154-2161.
- (5) Gonzalez, C.; Schlegel, H. B. *J. Phys. Chem.* **1990**, *94*, 5523-5527.

Energies and Cartesians Coordinates of Stationary Points

MeLi•(OMe)₂

SCF Done: E(RB+HF-LYP) = -512.564881634 A.U. after 6 cycles

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	1	0	-0.788842	1.055813	2.975777
2	6	0	-0.013699	0.350495	2.609524
3	1	0	0.940869	0.768948	2.991726
4	1	0	-0.172692	-0.571035	3.208609
5	3	0	-0.009230	0.090102	0.544882
6	8	0	-1.754444	-0.669118	-0.217108
7	8	0	1.527804	-1.051062	-0.164216
8	8	0	0.247165	1.786007	-0.598593
9	6	0	-2.644681	-1.140739	0.796707
10	1	0	-2.287121	-0.718958	1.736989
11	1	0	-2.621108	-2.239059	0.846667
12	1	0	-3.671606	-0.808215	0.585589
13	6	0	-2.062628	-1.170125	-1.507584
14	1	0	-1.352353	-0.725630	-2.209189
15	1	0	-3.085236	-0.891173	-1.802781
16	1	0	-1.970413	-2.266303	-1.537504
17	6	0	1.457705	2.433948	-0.208502
18	1	0	2.261919	1.704748	-0.322852
19	1	0	1.404413	2.750317	0.841118
20	1	0	1.650163	3.303090	-0.854469

21	6	0	-0.882101	2.649506	-0.482113
22	1	0	-0.763319	3.524916	-1.137201
23	1	0	-1.010923	2.978334	0.557745
24	1	0	-1.757457	2.074620	-0.790040
25	6	0	2.178410	-1.839037	0.833273
26	1	0	1.781452	-1.505683	1.793466
27	1	0	3.266060	-1.680660	0.795223
28	1	0	1.961940	-2.906709	0.681156
29	6	0	1.951469	-1.349209	-1.484145
30	1	0	1.423643	-0.667368	-2.155326
31	1	0	1.711429	-2.389572	-1.751174
32	1	0	3.035824	-1.196367	-1.591802

Me₂O

SCF Done: E(RB+HF-LYP) = -155.033688970 A.U. after 7 cycles

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	8	0	0.000000	0.587561	0.000000
2	6	0	1.177331	-0.195661	0.000000
3	1	0	1.233901	-0.836130	-0.895388
4	1	0	1.233901	-0.836130	0.895389
5	1	0	2.023631	0.495986	0.000000
6	6	0	-1.177331	-0.195661	0.000000
7	1	0	-1.233901	-0.836130	-0.895388
8	1	0	-2.023631	0.495986	0.000000
9	1	0	-1.233901	-0.836130	0.895388

CP1

SCF Done: E(RB+HF-LYP) = -1134.88630484 A.U. after 8 cycles

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	1	0	4.804038	0.179295	1.367932
2	6	0	3.861348	-0.076711	1.900031
3	1	0	4.040145	-1.076201	2.349273
4	1	0	3.807466	0.618673	2.763107
5	3	0	2.140180	-0.013532	0.679157
6	8	0	2.253065	1.574693	-0.648369
7	8	0	2.163424	-1.506307	-0.781244
8	6	0	2.673173	2.745822	0.039471
9	1	0	3.440099	2.427722	0.748458
10	1	0	1.832539	3.199665	0.584433
11	1	0	3.084852	3.481973	-0.670728
12	6	0	1.190705	1.813862	-1.562785
13	1	0	0.945336	0.852695	-2.018886
14	1	0	1.509790	2.526635	-2.341129
15	1	0	0.305325	2.199400	-1.043299
16	6	0	1.468618	-2.699367	-0.437519
17	1	0	0.474605	-2.402625	-0.101082
18	1	0	1.988569	-3.229919	0.374176
19	1	0	1.390381	-3.363111	-1.313538
20	6	0	3.503564	-1.744417	-1.187599
21	1	0	3.522905	-2.380305	-2.088552
22	1	0	4.077385	-2.220937	-0.381417
23	1	0	3.947594	-0.771813	-1.407129
24	17	0	-0.316098	0.088658	1.246509
25	29	0	-2.387269	-0.021244	0.152812

26	6	0	-2.508414	1.964525	-0.057417
27	1	0	-1.789172	2.432486	0.623350
28	1	0	-3.522166	2.326396	0.160911
29	1	0	-2.262529	2.227156	-1.096873
30	6	0	-2.481839	-2.012453	0.325698
31	1	0	-1.846155	-2.316040	1.164632
32	1	0	-2.099545	-2.454791	-0.606288
33	1	0	-3.508225	-2.367893	0.485033
34	6	0	-4.066097	-0.129793	-0.825925
35	1	0	-4.183786	0.735022	-1.483608
36	1	0	-4.848816	-0.128469	-0.058978
37	1	0	-4.104950	-1.061446	-1.396312

TS1

SCF Done: E(RB+HF-LYP) = -1134.87649990 A.U. after 9 cycles

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	29	0	-1.912119	-0.048872	-0.047769
2	6	0	-1.875615	-2.042889	-0.159832
3	1	0	-0.941936	-2.345570	-0.647468
4	1	0	-2.726301	-2.415848	-0.747854
5	1	0	-1.922060	-2.459657	0.854523
6	6	0	-2.158205	1.928632	0.084088
7	1	0	-1.259554	2.411292	-0.316338
8	1	0	-2.306623	2.217012	1.133128
9	1	0	-3.030315	2.245546	-0.505372
10	6	0	-3.775963	-0.220590	0.529497
11	1	0	-4.355299	-0.240631	-0.400306
12	1	0	-4.063061	0.639134	1.139330
13	1	0	-3.907913	-1.154621	1.080633
14	17	0	-0.068324	0.211577	-1.821935
15	6	0	-0.031477	-0.174374	2.131064
16	1	0	-0.169284	-1.235320	2.411374
17	1	0	0.822378	0.200244	2.741371
18	1	0	-0.919784	0.362316	2.500417
19	3	0	0.941240	0.016402	0.267890
20	8	0	2.299951	1.605023	0.345222
21	8	0	2.427689	-1.458728	0.054122
22	6	0	3.051572	1.828003	-0.838512
23	1	0	3.835665	2.582995	-0.660363
24	1	0	2.398039	2.153094	-1.658983
25	1	0	3.511733	0.873836	-1.104183
26	6	0	1.638600	2.777157	0.809858
27	1	0	1.027127	2.475066	1.661554
28	1	0	0.987090	3.191083	0.028820
29	1	0	2.377817	3.535829	1.115808
30	6	0	2.135350	-2.517163	-0.851370
31	1	0	1.683806	-2.058502	-1.732706
32	1	0	1.418529	-3.222045	-0.407168
33	1	0	3.059337	-3.054904	-1.120974
34	6	0	2.908128	-1.917325	1.309093
35	1	0	2.164077	-2.557792	1.802151
36	1	0	3.079818	-1.034785	1.928178
37	1	0	3.850499	-2.475571	1.180891

CP2

SCF Done: E(RB+HF-LYP) = -1134.92455705 A.U. after 9 cycles

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	29	0	-2.718215	-0.044376	0.157558
2	6	0	-4.677411	-0.045688	-0.156602
3	1	0	-4.844226	-0.426396	-1.175666
4	1	0	-5.123333	0.953938	-0.070361
5	1	0	-5.168481	-0.729386	0.549349
6	6	0	-2.718315	1.946924	-0.015239
7	1	0	-3.311718	2.255555	-0.888098
8	1	0	-1.709918	2.378315	-0.094234
9	1	0	-3.202419	2.355873	0.885130
10	6	0	-2.767937	-2.037366	0.284900
11	1	0	-2.690778	-2.296255	1.352180
12	1	0	-1.905392	-2.480806	-0.235088
13	1	0	-3.691370	-2.473378	-0.119113
14	6	0	-0.736513	-0.079921	0.479057
15	1	0	-0.313808	-0.099494	-0.537655
16	1	0	-0.433371	-0.976384	1.036543
17	1	0	-0.410818	0.825336	1.008716
18	3	0	1.686165	-0.070277	0.402464
19	17	0	3.239728	-0.487828	1.972954
20	8	0	2.115479	1.718699	-0.472935
21	8	0	2.101195	-1.363654	-1.144514
22	6	0	2.737187	2.689557	0.366741
23	1	0	3.413413	3.325056	-0.225844
24	1	0	3.296894	2.131749	1.120392
25	1	0	1.978617	3.320754	0.852146
26	6	0	1.308124	2.293236	-1.493808
27	1	0	0.900334	1.470561	-2.084252
28	1	0	1.916669	2.946669	-2.138216
29	1	0	0.478740	2.869577	-1.062714
30	6	0	1.656211	-2.690734	-0.866438
31	1	0	0.592698	-2.630116	-0.628754
32	1	0	2.204408	-3.107413	-0.010688
33	1	0	1.800669	-3.331003	-1.749929
34	6	0	3.481741	-1.319105	-1.497634
35	1	0	4.103299	-1.675446	-0.666863
36	1	0	3.722816	-0.273486	-1.699925
37	1	0	3.659525	-1.924788	-2.400136

LiCl•(OMe₂)₃

SCF Done: E(RB+HF-LYP) = -932.976896813 A.U. after 12 cycles

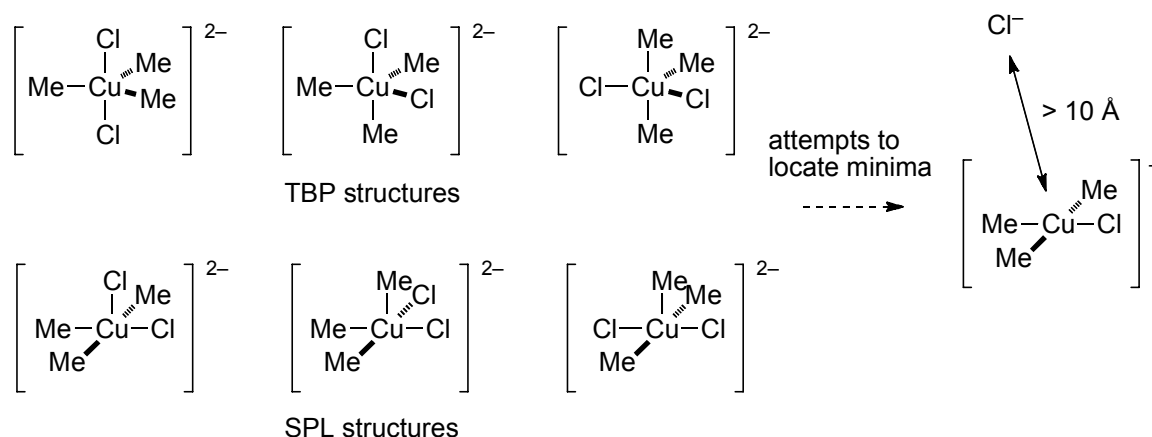
Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	17	0	0.003482	0.882891	2.259162
2	8	0	-1.647813	-0.998481	-0.201492
3	8	0	1.644403	-1.003460	-0.202886
4	8	0	0.002712	1.588489	-1.130556
5	6	0	-2.451683	-1.403286	0.912393
6	1	0	-2.062692	-0.872869	1.783543
7	1	0	-2.375413	-2.489781	1.060491
8	1	0	-3.501979	-1.131109	0.737709
9	6	0	-2.039008	-1.604305	-1.424306
10	1	0	-3.086056	-1.364672	-1.660325
11	1	0	-1.925119	-2.697254	-1.375177
12	1	0	-1.393211	-1.204210	-2.209961
13	6	0	2.445779	-1.414073	0.910691
14	1	0	2.060479	-0.881330	1.782054

15	1	0	3.497806	-1.148794	0.735870
16	1	0	2.362289	-2.500090	1.058429
17	6	0	2.030754	-1.611779	-1.426034
18	1	0	1.388735	-1.205392	-2.211564
19	1	0	1.907302	-2.703723	-1.377776
20	1	0	3.079919	-1.381067	-1.661447
21	6	0	-1.180235	2.377438	-0.968739
22	1	0	-1.203114	3.181427	-1.717796
23	1	0	-1.221711	2.798075	0.043600
24	1	0	-2.031639	1.710886	-1.120218
25	6	0	1.188673	2.373200	-0.969820
26	1	0	1.214233	3.176377	-1.719654
27	1	0	2.037528	1.703306	-1.120875
28	1	0	1.232061	2.794529	0.042131
29	3	0	0.000437	0.070331	0.199855

Attempted Study of Intramolecular Isomerization of Organocopper(III)

Complex

In order to probe the possibility of the pseudorotation mechanism for the isomerization of organocopper(III) complexes, geometry optimization of dianionic Cu(III) complex $[\text{Me}_3\text{CuCl}_2]^{2-}$ was carried out starting from a series of possible pentacoordinate (i.e., trigonal bipyramidal (TBP) and square pyramidal (SPL)) structures as shown below. However, none of such structures was located as a stationary point, and all the attempts uniformly led to dissociation into a square-planar $[\text{Me}_3\text{CuCl}]^-$ complex and a chloride anion (closest distance $> 10 \text{ \AA}$). In addition, inclusion of a lithium cation into the below computational model did not give a pentacoordinate Cu(III) complex as well. Thus, geometry optimization led to the formation of a complex between $[\text{Me}_3\text{CuCl}]^-$ and LiCl , where the Cl atom of the former and the Li atom of the latter electrostatically interact with each other (i.e., the Cl atom of LiCl does not interact with the Cu atom). Note also that attempts to locate a pentacoordinate Cu(III) complex of $[\text{Me}_4\text{CuCl}]^-$ led to dissociation into square-planar $[\text{Me}_4\text{Cu}]^-$ and a chloride anion. Because none of pentacoordinate Cu(III) complexes existed as a local minimum, we concluded that the classical pseudorotation mechanism is unlikely to operate in the organocopper(III) reactions.



We also examined the possibility of unimolecular isomerization of a tetracoordinate Cu(III) complex. Thus, trigonal pyramidal (TP) and tetrahedral (Th) structures of $[\text{Me}_3\text{CuCl}]^-$ were calculated assuming that they are either intermediates or transition states for the *cis/trans* isomerization of the square-planar structure. However, the attempts to locate them as local minima and as TSs uniformly led to the square-planar structure and the TS for reductive elimination of ethane, respectively.

