## **Supplementary Information**

## Self-Assembled Cage as an Endo-Template for Cyclophane Synthesis

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## Contents

- Fig. S1. <sup>1</sup>H NMR (500 MHz,  $D_2O$ , 300 K) spectrum of  $1\supset 3a$ .
- Fig. S2a, b.  ${}^{13}$ C NMR (125 MHz, D<sub>2</sub>O, 300 K) spectrum of 1 $\supset$ 3a.
- Fig. S3. <sup>1</sup>H DOSY (D<sub>2</sub>O, 300 K) spectrum of 1⊃3a.
- **Fig. S4.**  ${}^{1}\text{H}{}^{-1}\text{H} \text{ COSY (D}_{2}\text{O}, 300 \text{ K}) \text{ spectrum of } 1 \supset 3a.$
- Fig. S5a, b.  ${}^{1}\text{H}{}^{-13}\text{C}$  HSQC (D<sub>2</sub>O, 300 K) spectrum of 1 $\supset$ 3a.
- Fig. S6.  ${}^{1}\text{H}{}^{-13}\text{C}$  HMBC (D<sub>2</sub>O, 300 K) spectrum of  $1 \supset 3a$ .
- Fig. S7.  ${}^{1}\text{H}{}^{-1}\text{H}$  NOESY (D<sub>2</sub>O, 300 K) spectrum of 1 $\supset$ 3a.
- Fig. S8. <sup>1</sup>H NMR (500 MHz, D<sub>2</sub>O, 300 K) spectrum of 1⊃3b.
- Fig. S9a, b.  ${}^{13}$ C NMR (125 MHz, D<sub>2</sub>O, 300 K) spectrum of  $1 \supset 3b$ .
- Fig. S10. <sup>1</sup>H DOSY (D<sub>2</sub>O, 300 K) spectrum of 1⊃3b.
- Fig. S11.  ${}^{1}\text{H}{}^{-1}\text{H} \text{ COSY (D}_{2}\text{O}, 300 \text{ K}) \text{ spectrum of } 1 \supset 3b.$
- Fig. S12a, b.  ${}^{1}\text{H}{}^{-13}\text{C}$  HSQC (D<sub>2</sub>O, 300 K) spectrum of 1 $\supset$ 3b.
- Fig. S13a, b.  ${}^{1}\text{H}{}^{-13}\text{C}$  HMBC (D<sub>2</sub>O, 300 K) spectrum of 1 $\supset$ 3b.
- Fig. S14.  ${}^{1}\text{H}{}^{-1}\text{H}$  NOESY (500 MHz, D<sub>2</sub>O, 300 K) spectrum of  $1 \supset 3b$ .
- Fig. S15. <sup>29</sup>Si NMR (60 MHz,  $D_2O$ , 300 K) spectrum of  $1 \supset 3a$ .
- Fig. S16.  ${}^{1}\text{H}{}^{-29}\text{Si}$  HMBC (D<sub>2</sub>O, 300 K) spectrum of 1 $\supset$ 3a.
- Fig. S17. <sup>29</sup>Si NMR (60 MHz, D<sub>2</sub>O, 300 K) spectrum of 1⊃3b.
- Fig. S18. <sup>1</sup>H-<sup>29</sup>Si HMBC (300 MHz, D<sub>2</sub>O, 300 K) spectrum of 1⊃3b.
- Fig. S19. LDI-MS spectra of (a) 3a and (b) 3b.
- Fig. S20. Optimized structures of (a)  $1 \supset 3a$  and (b)  $1 \supset 3b$ .
- Fig. S21. Optimized structures of (a) 3a and (b) 3b.

## Materials and methods:

<sup>1</sup>H, <sup>13</sup>C NMR, and 2D NMR spectra were recorded on a Bruker DRX-500 spectrometer equipped with a 5 mm BBO Z-gradient probe, a Bruker AV-500 spectrometer equipped <sup>29</sup>Si NMR and <sup>1</sup>H-<sup>29</sup>Si HMBC spectra were with a 5 mm BBI Z-gradient probe. measured on a JEOL JNM-ECA 300. TMS (CDCl<sub>3</sub> solution) in a capillary served as external standard ( $\delta$  0 ppm). IR measurements (ATR) were carried out using a DIGILAB Scimitar FTS-2000 instrument. UV-visible spectral data were recorded on LDI-TOFMS spectra were measured with Voyager a SHIMADZU UV-3150. DE-STR without matrix. Melting points were determined with a Yanaco MF-500 V micro melting point apparatus. Elemental analyses were performed on a Yanaco MT-6 at the Elemental Analysis Center of School of Science of the University of Tokyo. Solvents and reagents were purchased from TCI Co., Ltd., WAKO Pure Chemical Industries Ltd., and Sigma-Aldrich Co. Deuterated H<sub>2</sub>O was acquired from Cambridge Isotope Laboratories, Inc. and used as supplied for the complexation reactions and NMR measurements.



**Preparation of 1⊃3a.** A mixture of cage **1** (14.9 mg, 5 μmol), **2a** (6.6 mg, 15 μmol) in D<sub>2</sub>O (0.5 mL) was stirred at 100 °C for 20 h to give clear red solution. <sup>1</sup>H NMR analysis of the solution revealed the formation of **1⊃3a**. After filtration of the resulted orange solution, the resulting solution was dried by a freeze-drying equipment. **1⊃3a** was isolated as an orange powder (15.4 mg, 84% yield). <sup>1</sup>H NMR (500 MHz, D<sub>2</sub>O, 300 K): δ 8.65 (br, 12H), 8.44 (br, 12H), 8.20 (br, 12H), 8.12 (d, *J* = 7.5 Hz, 4H), 8.12 (d, *J* = 7.5 Hz, 4H), 3.73 (s, 18H), 3.73 (s, 18H), 3.73 (s, 36H), (br, 12H), (br, 12H); <sup>13</sup>C NMR (125 MHz, D<sub>2</sub>O, 300 K): δ 167.2 (*C*<sub>q</sub>), 161.4 (*C*<sub>q</sub>), 152.0 (*C*H), 146.3 (*C*<sub>q</sub>), 143.5 (*C*<sub>q</sub>), 132.4 (*C*<sub>q</sub>), 131.1 (*C*H), 130.3 (*C*<sub>q</sub>), 127.6 (*C*H), 127.2 (*C*H), 126.9 (*C*<sub>q</sub>), 124.8

(CH), 122.7 (CH), 122.0 (CH), 120.1 ( $C_q$ ), 47.5 (CH<sub>2</sub>), 46.4 (CH<sub>2</sub>), 26.1 (CH<sub>3</sub>); DOSY (m<sup>2</sup>/s): log*D* = -9.83; IR (KBr, cm<sup>-1</sup>): 2427, 2353, 2337, 1615, 1520, 1385, 1213, 1058, 859, 808, 674; m.p.: ~230 °C (decomposed); E.A. Calcd. for C<sub>122</sub>H<sub>144</sub>N<sub>42</sub>O<sub>46</sub>Pd<sub>6</sub>Si<sub>4</sub>• 23(H<sub>2</sub>O): C, 35.74; H, 4.67; N, 14.35. Found: C, 35.89; H, 4.42; N, 14.07.



**Fig. S1.** <sup>1</sup>H NMR (500 MHz, D<sub>2</sub>O, 300 K) spectrum of  $1 \supset 3a$ .

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**Fig. S2b.** <sup>13</sup>C NMR (125 MHz, D<sub>2</sub>O, 300 K) spectrum of  $1 \supset 3a$ .



**Fig. S4.**  ${}^{1}\text{H}{}^{-1}\text{H} \text{ COSY }(D_2O, 300 \text{ K}) \text{ spectrum of } 1 \supset 3a.$ 



Fig. S5a. <sup>1</sup>H-<sup>13</sup>C HSQC (D<sub>2</sub>O, 300 K) spectrum of  $1 \supset 3a$ .



<sup>1</sup>H-<sup>13</sup>C HSQC (D<sub>2</sub>O, 300 K) spectrum of  $1 \supset 3a$ . Fig. S5b.



**Fig. S7.** <sup>1</sup>H-<sup>1</sup>H NOESY (D<sub>2</sub>O, 300 K) spectrum of  $1\supset$ 3a.



**Preparation of 1⊃3b.** A mixture of cage **1** (14.9 mg, 5 μmol), **2b** (6.6 mg, 15 μmol) in D<sub>2</sub>O (0.5 mL) was stirred at 100 °C for 20 h to give clear orange solution. <sup>1</sup>H NMR analysis of the solution revealed the formation of **1⊃3b**. After filtration of the resulted orange solution, the resulting solution was dried by a freeze-drying equipment. **1⊃3b** was isolated as an orange powder (15.3 mg, 83% yield). <sup>1</sup>H NMR (500 MHz, D<sub>2</sub>O, 300 K): δ 8.65 (br, 12H), 8.44 (br, 12H), 8.20 (br, 12H), 8.12 (d, *J* = 7.5 Hz, 4H), 8.12 (d, *J* = 7.5 Hz, 4H), 3.73 (s, 18H), 3.73 (s, 18H), 3.73 (s, 36H), (br, 12H), (br, 12H); <sup>13</sup>C NMR (125 MHz, D<sub>2</sub>O, 300 K): δ 167.1 (*C*<sub>q</sub>), 161.6 (*C*<sub>q</sub>), 161.1 (*C*<sub>q</sub>), 152.1 (*C*H), 147.5 (*C*<sub>q</sub>), 144.3 (*C*<sub>q</sub>), 133.2 (*C*<sub>q</sub>), 132.3 (*C*H), 129.5 (*C*<sub>q</sub>), 127.6 (*C*<sub>q</sub>), 127.2 (*C*H), 125.4 (*C*H), 125.0 (*C*H), 122.4 (*C*H), 122.2 (*C*H), 120.4 (*C*<sub>q</sub>), 47.6 (*C*H<sub>2</sub>), 46.5 (*C*H<sub>2</sub>), 26.1 (*C*H<sub>3</sub>); DOSY (m<sup>2</sup>/s): log*D* = −9.81; IR (KBr, cm<sup>-1</sup>): 3403, 3086, 1607, 1575, 1550, 1521, 1384, 1100, 1087, 1059, 808, 523, 434; m.p.: ~230 °C(decomposed).



**Fig. S9a.** <sup>13</sup>C NMR (125 MHz, D<sub>2</sub>O, 300 K) spectrum of  $1 \supset 3b$ .



**Fig. S9b.** <sup>13</sup>C NMR (125 MHz,  $D_2O$ , 300 K) spectrum of  $1 \supset 3b$ .



**Fig. S10.** <sup>1</sup>H DOSY (D<sub>2</sub>O, 300 K) spectrum of  $1 \supset 3b$ .



**Fig. S12a.**  ${}^{1}\text{H}{}^{-13}\text{C}$  HSQC (D<sub>2</sub>O, 300 K) spectrum of  $1\supset 3b$ .



**Fig. S12b.**  ${}^{1}\text{H}{}^{-13}\text{C}$  HSQC (D<sub>2</sub>O, 300 K) spectrum of  $1\supset 3b$ .



**Fig. S13a.** <sup>1</sup>H-<sup>13</sup>C HMBC (D<sub>2</sub>O, 300 K) spectrum of **1⊃3b**.



**Fig. S14.**  ${}^{1}$ H- ${}^{1}$ H NOESY (D<sub>2</sub>O, 300 K) spectrum of **1** $\supset$ **3b**.



**Fig. S15.** <sup>29</sup>Si NMR (60 MHz, D<sub>2</sub>O, 300 K) spectrum of 1⊃3a.



**Fig. S16.** <sup>1</sup>H-<sup>29</sup>Si HMBC (D<sub>2</sub>O, 300 K) spectrum of 1⊃**3a**.



**Fig. S17.** <sup>29</sup>Si NMR (60 MHz, D<sub>2</sub>O, 300 K) spectrum of 1⊃3b.



**Fig. S18.** <sup>1</sup>H-<sup>29</sup>Si HMBC (D<sub>2</sub>O, 300 K) spectrum of 1⊃**3**b.







**Fig. S20.** Optimized structures (MM) of (a)  $1 \supset 3a$  and (b)  $1 \supset 3b$ .



**Fig. S21.** Optimized structures (MM) focusing on (a) **3a** and (b) **3b** within the cavity of **1**. Hydrogen atoms of **3a** and **3b** were omitted for clarity.