## **Electronic Supplementary Information** (9 pages)

# Molecular dynamics simulation of stability of metal-organic frameworks against H<sub>2</sub>O using the ReaxFF reactive force field

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## **Supporting Information**

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#### S.1 Simulation details

In this work, we used molecular dynamics (MD) simulation to simulate volume changes of metal-organic frameworks (MOFs) such as isoreticular MOF-1 (IRMOF-1) and MOF-74 as a function of H<sub>2</sub>O content. Our MD simulations used a time step of 0.001 ps (picosecond) and were performed up to 2000 ps. We maintained the temperature (300 K) using a Berendsen thermostat with a damping constant of 0.001 ps. Also, the pressure (1 atm) is maintained using a Berendsen barostat with a damping constant of 5 ps. (Such MD is denote as NPT). This method leads to independent changes of lattice parameters (*a*, *b*, *c*) of the MOF structures while the lattice angles (*a*, *β*, *γ*) are fixed. To eliminate boundary effects in calculating volume changes of the MOFs, we considered an infinite three-dimensionally periodic simulation boxes, in which we used simulation boxes containing unit cells ( $1 \times 1 \times 1$ ) for IRMOF-1 and a  $1 \times 1 \times 3$  supercell for MOF-74. And, the initial structures including MOFs and H<sub>2</sub>O molecules are built using a grand-canonical Monte-Carlo simulation<sup>S1</sup> with the universal force field<sup>S2</sup> where the H<sub>2</sub>O molecules are randomly distributed in free volume of the MOFs.

To describe the dynamics of MOFs including H<sub>2</sub>O molecules, we used the reactive force field (ReaxFF)<sup>S3</sup> known to accurately simulate chemical reactions (bond breaking and formation) in large systems in contrast to classical force fields. The ReaxFF uses a general relationship between bond distance and bond order on one hand and between bond order and bond energy on the other hand that leads to proper dissociation of bonds to separated atoms.<sup>3a</sup> The connectivities between the atoms can be modified during a MD simulation allowing spontaneous bond breaking and bond formation.<sup>3e</sup> And, similar to quantum mechanical (QM) calculations, each element is represented by only one atom type in the ReaxFF. This facilitates the transferability of the

parameters to a different system and avoids the modification of atom types during a chemical reaction.

The nonbonded interactions need to be calculated between every atom pair in order to describe a system with changing connectivity. Thus, the van der Waals energy is calculated between all atom pairs using a distance-corrected Morse potential. A shielded interaction is implemented to avoid the excessive high repulsion between the bonded atom pairs. Similarly, the Coulomb energy is calculated between all atom pairs using a shielded coulomb potential. For each element, the electronegativity, hardness, and shielding parameters are optimized to reproduce the QM derived charges. To account for the polarization effect, the atomic charges are dynamically derived using the electronegativity equalization method where they are calculated at every step in contrast to classical force field using fixed charges. Generally, all parameters of the ReaxFF are developed from QM calculations. For further details about the ReaxFF method see refs. S3a and S3b.

In this work, we used reported ReaxFF parameters<sup>84</sup> for description of the ZnO-water system. The ReaxFF reproduce experimental and/or QM lattice parameters, heats of formation, and mechanical properties such as bulk modulus and elastic constants for four ZnO crystal structures (wurtzite, zinc blende, rocksalt, and caesium chloride). Also, it successfully predicts water adsorption reactions on ZnO surfaces. By MD simulations with this ReaxFF, we obtained the following lattice parameters for pure MOFs; 26.88 Å for IRMOF-1, 35.25 Å for IRMOF-10, and *a* = 26.73 Å and *c* = 6.78 Å for MOF-74. These are comparable to experimental values; 25.67 – 25.89 Å for IRMOF-1, <sup>S5</sup> 34.28 Å for IRMOF-10, <sup>S5c</sup> and *a* = 25.93 Å and *c* = 6.84 Å for MOF-74. <sup>S6</sup> The ReaxFF parameters used in this work are shown below. And this force field can be used with

both the standalone ReaxFF program (available from Prof. A. C. T. van Duin of this work on request) and with the open source LAMMPS/ReaxFF software (from Sandia).

Reactive MD-force field: Han et al. ZnO/MOF force field

- 39 ! Number of general parameters
  - 50.0000 !Overcoordination parameter
  - 9.5469 !Overcoordination parameter
  - 26.5405 !Valency angle conjugation parameter
  - 1.7224 !Triple bond stabilisation parameter
  - 6.8702 !Triple bond stabilisation parameter
  - 60.4850 !C2-correction
  - 1.0588 !Undercoordination parameter
  - 4.6000 !Triple bond stabilisation parameter
  - 12.1176 !Undercoordination parameter
  - 13.3056 !Undercoordination parameter
- -70.5044 !Triple bond stabilization energy
  - 0.0000 !Lower Taper-radius
- 10.0000 !Upper Taper-radius
- 2.8793 !Not used
- 33.8667 !Valency undercoordination
- 6.0891 !Valency angle/lone pair parameter
- 1.0563 !Valency angle
- 2.0384 !Valency angle parameter
- 6.1431 !Not used
- 6.9290 !Double bond/angle parameter
- 0.3989 !Double bond/angle parameter: overcoord
- 3.9954 !Double bond/angle parameter: overcoord
- -2.4837 !Not used
- 5.7796 !Torsion/BO parameter
- 10.0000 !Torsion overcoordination
- 1.9487 !Torsion overcoordination
- -1.2327 !Conjugation 0 (not used)

2.1645 !Conjugation

1.5591 lvdWaals shielding

0.1000 !Cutoff for bond order (\*100)

2.1365 !Valency angle conjugation parameter

0.6991 !Overcoordination parameter

50.0000 !Overcoordination parameter

1.8512 !Valency/lone pair parameter

0.5000 !Not used

20.0000 !Not used

5.0000 !Molecular energy (not used)

0.0000 !Molecular energy (not used)

2.6962 !Valency angle conjugation parameter

4 ! Nr of atoms; cov.r; valency;a.m;Rvdw;Evdw;gammaEEM;cov.r2;# alfa;gammavdW;valency;Eunder;Eover;chiEEM;etaEEM;n.u. cov r3;Elp;Heat inc.;n.u.;n.u.;n.u.;n.u.

ov/un;val1;n.u.;val3,vval4

С	1.3825	4.0000	12.0000	1.9133	0.1853	0.9000	1.1359	4.0000
	9.7602	2.1346	4.0000	33.2433	79.5548	5.8678	7.0000	0.0000
	1.2104	0.0000 ^	199.0303	8.6991	34.7289	13.3894	0.8563	0.0000
	-2.8983	2.5000	1.0564	4.0000	2.9663	0.0000	0.0000	0.0000
Н	0.8930	1.0000	1.0080	1.3550	0.0930	0.8203	-0.1000	1.0000
	8.2230	33.2894	1.0000	0.0000	121.1250	3.7248	9.6093	1.0000
	-0.1000	0.0000	61.6606	3.0408	2.4197	0.0003	1.0698	0.0000
	-19.4571	4.2733	1.0338	1.0000	2.8793	0.0000	0.0000	0.0000
0	1.2450	2.0000	15.9990	2.3890	0.1000	1.0898	1.0548	6.0000
	9.7300	13.8449	4.0000	37.5000	116.0768	8.5000	8.3122	2.0000
	0.9049	0.4056	59.0626	3.5027	0.7640	0.0021	0.9745	0.0000
	-3.5500	2.9000	1.0493	4.0000	2.9225	0.0000	0.0000	0.0000
Zn	1.8862	2.0000	65.3900	1.9200	0.2998	0.4828	-1.6836	2.0000
	11.5134	18.3776	2.0000	0.0078	0.0000	2.0219	5.7915	0.0000
	-1.2000	0.0000 2	66.4838	5.3430	10.1260	0.7590	0.0000	0.0000
	-3.0614	2.1158	1.0338	6.2998	2.5791	0.0000	0.0000	0.0000
10	! Nr (	of bonds; E	Edis1;LPpe	en;n.u.;pbe	e1;pbo5;13	corr;pbo6		

pbe2;pbo3;pbo4;n.u.;pbo1;pbo2;ovcorr

1 1 158.2004 99.1897 78.0000 -0.7738 -0.4550 1.0000 37.6117 0.4147

		0.4590	-0.1000	9.16	628	1.000	0 -0	.0777	6.7	268	1.0	000	0.0000
1	2	169.4760	0.0000	0.0000		-0.608	33 (	).0000 1.000		000	6.0	000	0.7652
		5.2290	1.0000	0.0	000	1.00	00 -0	0.0500	6.9	136	0.0	000	0.0000
2	2	153.3934	0.0000	0.0	000	-0.460	00 0	0.0000		000	6.0	000	0.7300
		6.2500	1.0000	.0000 0.0000 1.00		1.00	00 -0	0.0790	6.0	552	0.0	000	0.0000
1	3	158.6946	107.4583	23.3	136	-0.424	-0	.1743	1.00	000	10.8	209	1.0000
		0.5322	-0.3113	7.00	000	1.000	00 -0	.1447	5.2	450	0.0	000	0.0000
3	3	142.2858	145.0000	50.8	293	0.25	06 -0	.1000	1.0	000	29.7	503	0.6051
		0.3451	-0.1055	9.00	000	1.000	00 -0	.1225	5.5	000	1.0	000	0.0000
2	3	160.0000	0.0000	0.0	000	-0.572	25 (	0.0000	1.0	000	6.0	000	0.5626
		1.1150	1.0000	0.0	000	1.00	00 -0	0.0920	4.2	790	0.0	000	0.0000
1	4	0.0000	0.0000	0.0	000	0.00	00 -0	0.5000	1.0	000	36.0	0000	0.0082
		1.7973	-0.2500	20.00	000	1.000	00 -0	.2578	6.5	219	0.0	000	0.0000
2	4	0.0000	0.0000	0.0	000	0.00	00 -0	0.5000	1.0	000	50.0	0000	0.5000
		0.5000	-0.5000	30.00	000	1.000	00 -0	.2000	8.0	000	0.0	000	0.0000
3	4	159.9755	0.0000	0.0	000	-0.454	18 -0	.5000	0.0	000	35.0	000	0.0375
		1.3099	-0.5000	25.0	000	1.000	00 -0	.4787	4.6	717	0.0	000	0.0000
4	4	38.4643	0.0000	0.0	000	-0.69	44 -0	0.2000	0.0	000	16.0	000	0.2129
	0.5059 -0.2000		15.00	000	1.000	00 -0	.0814	6.0	333	0.0	000	0.0000	
7		! Nr of off	-diagonal	terms;	Ediss	;Ro;ga	amma	rsigma	;rpi;rp	i2			
1	2	0.1219	1.4000	9.8	442	1.12	03 -	1.0000	-1.0	000			
2	3	0.0283	1.2885	10.9	190	0.92	15 -	1.0000	-1.0	000			
2	4	0.1059	1.8290	9.7	818	0.95	98 -	1.0000	-1.0	000			
1	3	0.1131	1.8523	9.8	442	1.27	75	1.1342	1.(	0621			
1	4	0.3000	1.6692	11.1	307	0.01	- 00	1.0000	-1.0	000			
2	4	0.0987	1.8227	12.0	654	0.10	- 00	1.0000	-1.0	000			
3	4	0.2744	2.1414	9.7	703	1.98	04 -	1.0000	-1.0	000			
24		! Nr of an	gles;at1;a	t2;at3; <sup>-</sup>	Theta	o,o;ka	kb;pv	1;pv2					
1	1	1 59.0	573 30.7	029	0.76	06 (	0.0000	0.7	180	6.29	933	1.12	44
1	1	2 65.7	758 14.5	234	6.24	81 (	0.0000	0.5	665	0.0	000	1.62	55
2	1	2 70.2	607 25.2	202	3.73	12 (	0.0000	0.0	050	0.0	000	2.75	00
1	2	2 0.0	0.0 0.0	000	6.00	00 C	.0000	0.0	000	0.00	000	1.040	00
1	2	1 0.0	000 3.4	110	7.73	50 C	.0000	0.0	000	0.00	000	1.040	00
2	2	2 0.0	000 27.9	213	5.86	35 (	0.0000	0.0	000	0.00	000	1.040	00
1	1	3 49.6	811 7.1	713	4.38	89 (	0.0000	0.7	171	10.20	661	1.04	63

3	1	3	77	.7473	40.17	718	2.9802 -	25.30	63	1.6170 -	46.1315	2.2503
2	1	3	65	.0000	13.88	315	5.0583	0.0	000	0.4985	0.0000	1.4900
1	3	1	73	.5312	44.72	275	0.7354	0.0	000	3.0000	0.0000	1.0684
1	3	3	79	.4761	36.37	<b>'</b> 01	1.8943	0.0	000	0.7351	67.6777	3.0000
3	3	3	80	.7324	30.45	554	0.9953	0.0	000	1.6310	50.0000	1.0783
1	3	2	70	.1880	20.95	562	0.3864	0.0	000	0.0050	0.0000	1.6924
2	3	3	75	.6935	50.00	000	2.0000	0.0	000	1.0000	0.0000	1.1680
2	3	2	85	.8000	9.84	53	2.2720	0.0	000	2.8635	0.0000	1.5800
1	2	3	0	.0000	25.00	000	3.0000	0.0	000	1.0000	0.0000	1.0400
1	2	5	0	.0000	0.00	19 (	6.0000	0.0	000	0.0000	0.0000	1.0400
3	2	3	0	.0000	15.00	000	2.8900	0.0	000	0.0000	0.0000	2.8774
2	2	3	0	.0000	8.57	44	3.0000	0.0	000	0.0000	0.0000	1.0421
2	3	4	77	.5446	9.90	16	2.3157	0.0	000	0.4543	0.0000	2.3770
3	4	3	10	.8790	38.99	915	0.7072	0.0	000	2.0000	0.0000	2.6162
4	3	4	37	.5284	32.35	525	0.2657	0.0	000	0.4403	0.0000	1.1000
3	4	4	16	.9624	30.32	<u>2</u> 41	0.2697	0.0	000	2.0000	0.0000	3.0708
3	3	4	60	.0000	20.00	000	0.5000	0.0	000	1.0000	0.0000	2.0000
21	<pre>1 ! Nr of torsions;at1;at2;at3;at4;;V1;V2;V3;V2(BO);vconj;n.u;n</pre>											
1	1	1	1	-0.250	0 11	.5822	0.187	′9 -4	1.7057	-2.204	7 0.0000	0.0000
1	1	1	2	-0.250	0 31	.2596	0.170	)9 -4	1.6391	-1.900	0.0000	0.0000
2	1	1	2	-0.177	0 30	.0252	0.434		5.0019	-2.069	0.0000	0.0000
1	1	1	3	-0.709	8 22	2.2951	0.006	60 -2	2.5000	-2.168	0.0000	0.0000
2	1	1	3	-0.356	8 22	2.6472	0.604	-4	1.0088	-1.000	0.0000	0.0000
3	1	1	3	-0.052	8 6	.8150	0.749	8 -5	5.0913	-1.000	0.0000	0.0000
1	1	3	1	2.000	)7 2	5.5641	-0.060	)8 -2	2.6456	-1.176	6 0.0000	0.0000
1	1	3	2	-1.195	3 42	2.1545	-1.000	0 -8	8.0821	-1.000	0 0.0000	0.0000
2	1	3	1	-0.928	4 34	.3952	0.728	35 -2	2.5440	-2.464	1 0.0000	0.0000
2	1	3	2	-2.500	0 79	.6980	1.000	0 -3	3.5697	-2.750	0.0000	0.0000
1	1	3	3	-0.017	'9 5	.0603	-0.189	4 -2	.5000	-2.039	9 0.0000	0.0000
2	1	3	3	-0.558	3 80	0.0000	1.000	)0 -4	1.4000	-3.000	0.0000	0.0000
3	1	3	1	-2.500	0 76	6.0427	-0.014	1 -3	.7586	-2.900	0 0.0000	0.0000
3	1	3	2	0.034	45 78	8.9586	-0.681	0 -4	4.1777	-3.000	0.0000	0.0000
3	1	3	3	-2.500	0 66	6.3525	0.398	36 -3	3.0293	-3.000	0.0000	0.0000
1	3	3	1	2.500	00 -0	.5332	1.000	0 -3	3.5096	-2.900	0.0000	0.0000
1	3	3	2	-2.500	0 3	.3219	0.718	-5	5.2021	-2.933	0.0000	0.0000

				2.2000 0.220	0 1.0000	-2.0109	-1.0000	0.0000	0.0000
1	3	3	3	0.0531 -17.398	3 1.0000	-2.5000	-2.1584	0.0000	0.0000
2	3	3	3	0.4723 -12.414	4 -1.0000	-2.5000	-1.0000	0.0000	0.0000
3	3	3	3	-2.5000 -25.0000	1.0000	-2.5000	-1.0000	0.0000	0.0000
1 ! Nr of hydrogen bonds;at1;at2;at3;Rhb;Dehb;vhb1									

3 2 3 2.1200 -3.5800 1.4500 19.5000

## S.2 MD snapshots of IRMOF-10



*Figure S1.* MD snapshots of IRMOF-10 with 16  $H_2O$  molecules (3.5 wt%  $H_2O$ ) at 0 ps (initial structure) (a), 20 ps (b), and 2000 ps (c). In Fig. b, the yellow atoms indicate OH (bonded to Zn cation of the SBU) and H (bonded to the organic linker) atoms dissociated from  $H_2O$ .

### S.2 References

S1. Using the Cerius2 software (Version 4.10, Accelrys Inc., San Diego)

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