

**Electric Supplementary Information for:**  
**Morphology Control of Calcium Phosphate by Mineralization**  
**on the  $\beta$ -sheet Peptide Template**

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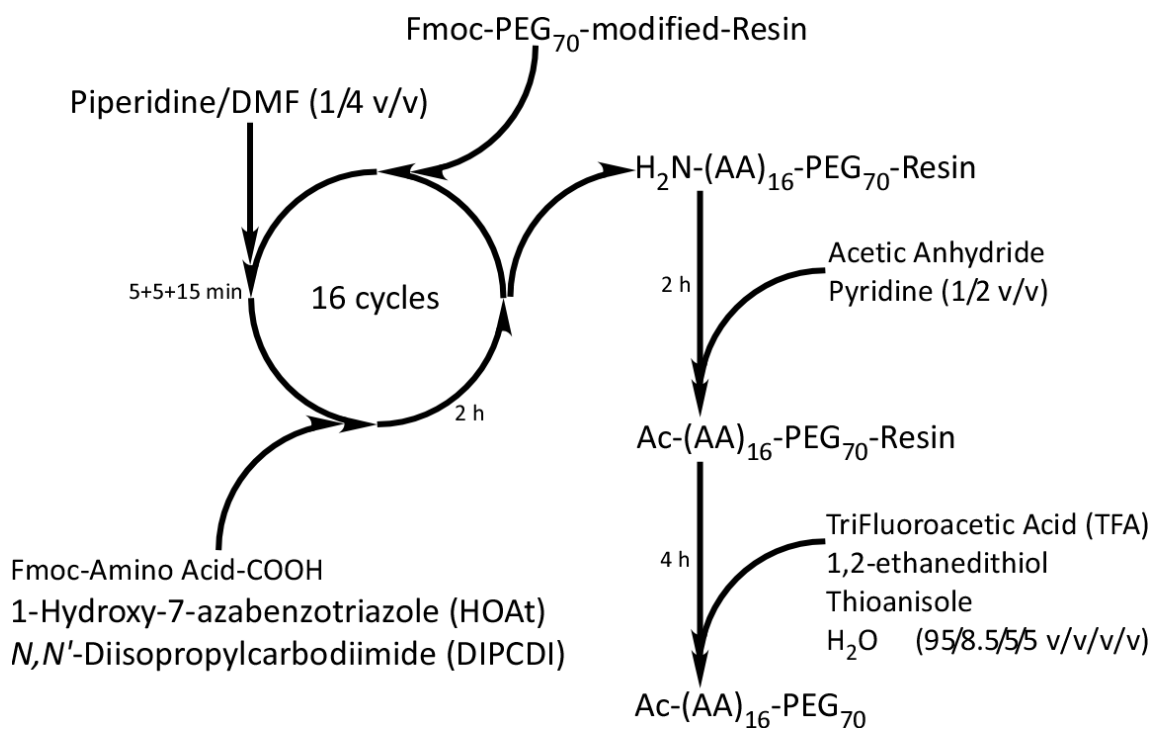
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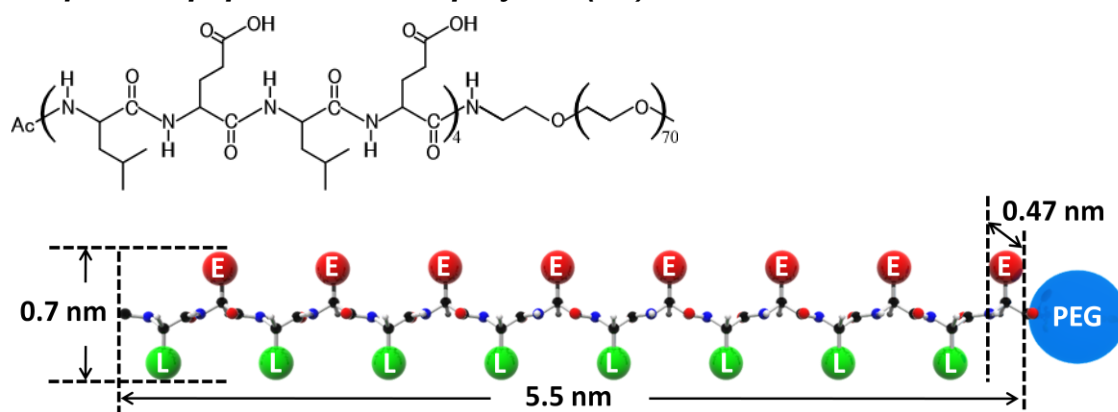
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## S1. Peptide synthesis by Fmoc solid phase peptide synthesis

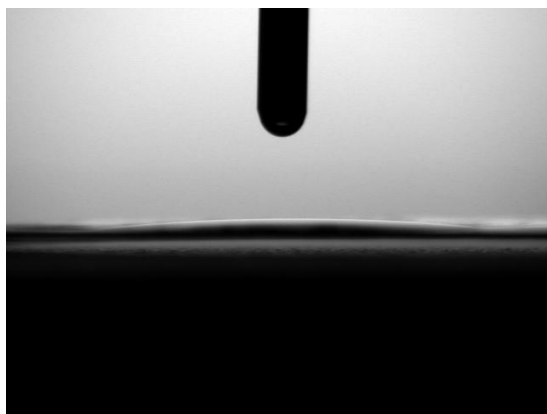


## S2. $\beta$ -sheet peptide – PEG copolymer (LE)<sub>8</sub>-PEG<sub>70</sub> model



### S3. Contact angle measurement

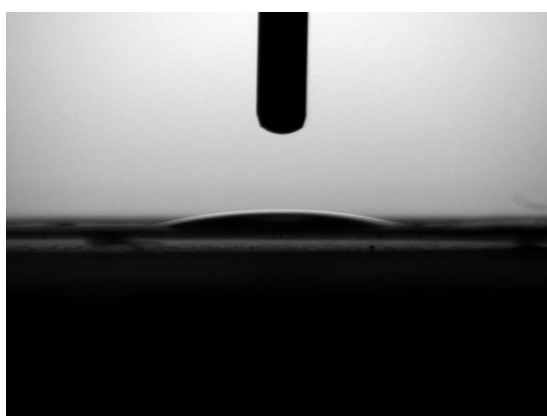
**Mica surface**



**(LE)<sub>8</sub>-PEG<sub>70</sub> LB monolayer  
(Leucine is surface side)**



**(LE)<sub>8</sub>-PEG<sub>70</sub> LB monolayer  
(Glutamic acid is surface side)**



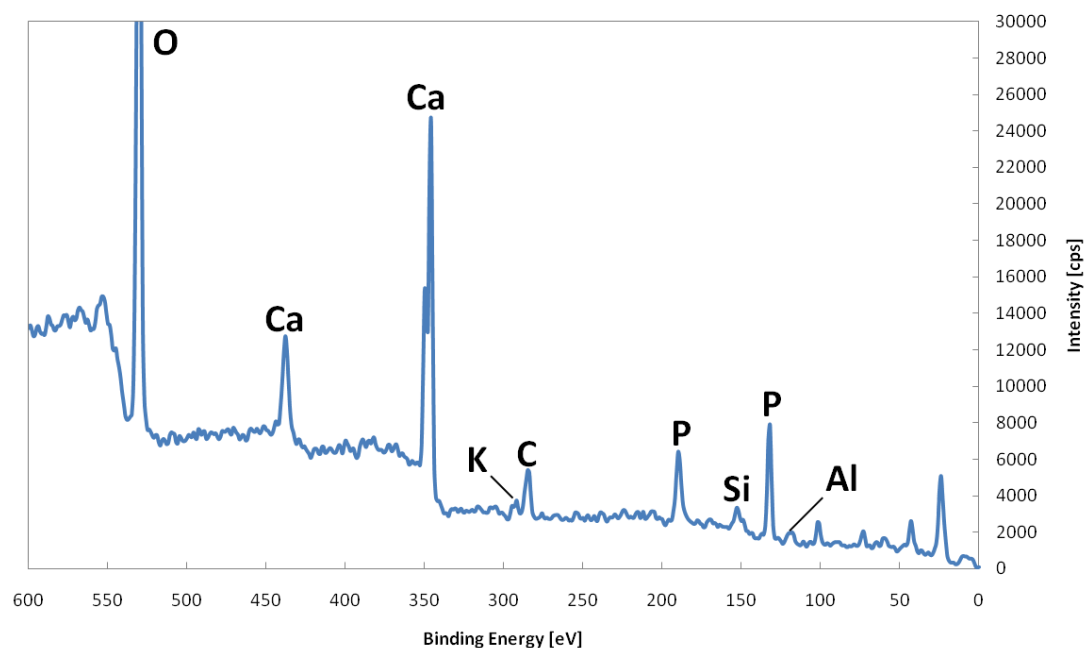
**Adsorption monolayer**



Surface	Contact angle [deg]
Mica surface	8.1
(LE) <sub>8</sub> -PEG <sub>70</sub> LB monolayer (Leucine is surface side)	43.5
(LE) <sub>8</sub> -PEG <sub>70</sub> LB monolayer (Glutamic acid is surface side)	13.9
Adsorption monolayer	17.6

The Langmuir-Blodgett method can control a molecular direction precisely. The two types of the (LE)<sub>8</sub>-PEG<sub>70</sub> monolayer of the leucine surface (hydrophobic side) and the glutamic acid surface (hydrophilic side) were prepared, and an adsorption film was compared with the LB films. A contact angle value of the adsorption film was similar to the glutamic acid surface one.

#### S4. XPS spectrum of precipitate on peptide scaffold



Mica :  $\text{KAl}_2(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_2$

Element	Atom [%]
K	15.0
Al	43.9
Si	41.1

CaP

Element	Atom [%]
Ca	54.2
P	45.8

Element	Binding Energy
C	284(1.00) $1\text{S}_{1/2}$
O	532(2.93) $1\text{S}_{1/2}$
Al	118(.753) $2\text{S}_{1/2}$
Si	149(.955) $2\text{S}_{1/2}$
P	189(1.18) $2\text{S}_{1/2}$ 135(.789) $2\text{P}_{3/2}$
K	294(2.62) $2\text{P}_{3/2}$
Ca	438(2.59) $2\text{S}_{1/2}$ 350(1.72) $2\text{P}_{1/2}$ 347(3.35) $2\text{P}_{3/2}$

**S5. AFM image of calcium phosphate nanofiber**

