

Electronic Supplementary Information

The first three-fold interpenetrated framework with two different four-connected uniform nets: 6^6 dia net and new chiral 8^6 mdf net

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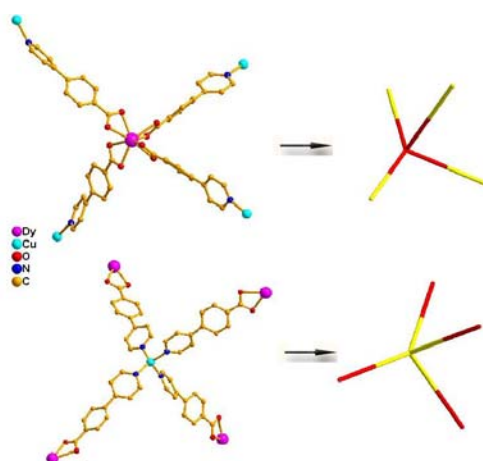


Figure S1. Each Dy^{3+} and Cu^+ ion could be considered as a four-connected node.

Optical properties: The photoluminescence property of **1**, the L^- ligand and **2** has been explored at room temperature in the solid state under excitation of 405 nm, 329nm and 484nm, respectively.

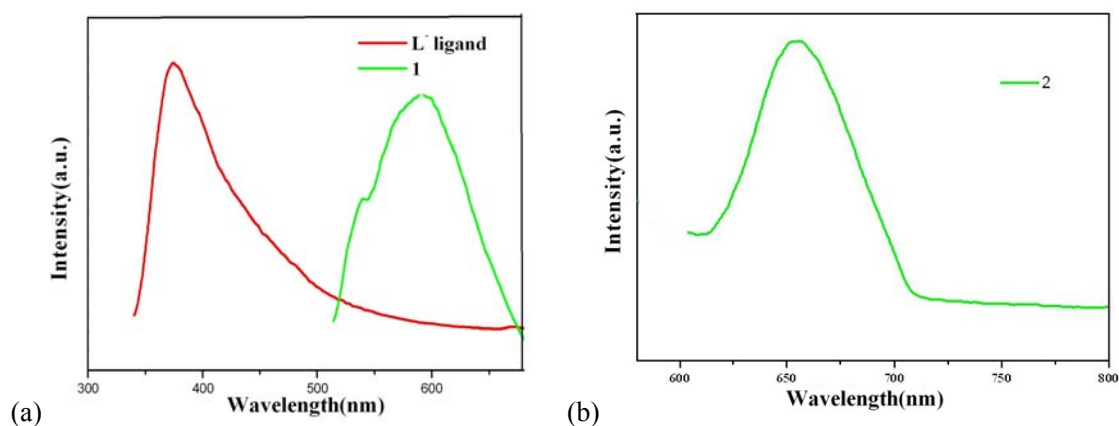


Fig. S2 (a) Emission spectrum of **1** and the L^- ligand. (b) Emission spectrum of **2**

Thermal Analysis: The thermal analysis was performed on Netzsch STA449C under the flowing nitrogen atmosphere. **1** and **2** was heated from room temperature to 1000 °C at a heating rate of 10 °C/min.

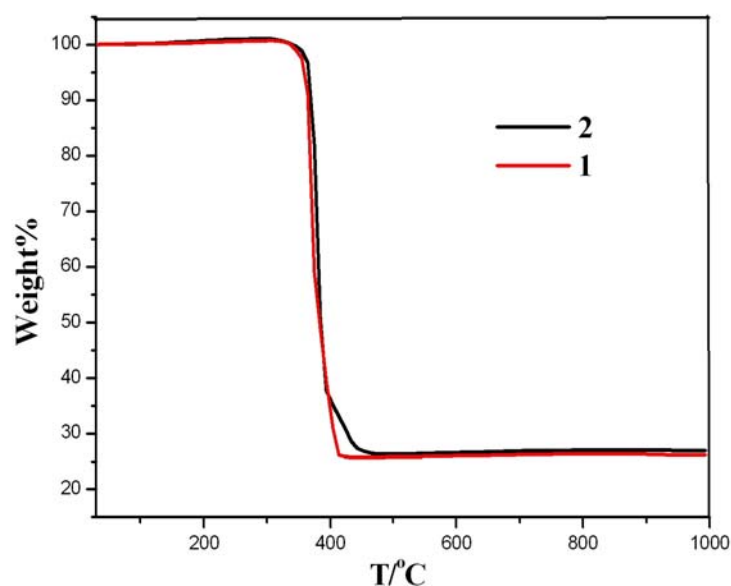


Figure S3. The TGA diagram of **1** and **2**.

There is one step weight loss from 360 to 440 °C. Assuming that the residue corresponds to Ln_2O_3 and CuO the observed weight (26.3% for **1**, 26.8% for **2**) is in good agreement with the calculated value (26.1% for **1**, 26.4% for **2**).

X-ray powder diffraction: X-ray powder diffraction experiments were performed on a PANalytical X'pert PRO diffractometer with $\text{Cu K}\alpha$ radiation (40 kV, 40 mA) was used to identify the crystal structure of the sample. The measurement was conducted in the continuous scanning mode. The 2θ scanning range was from 5° to 50° in steps of 0.02° with a collection time of 12 s per step.

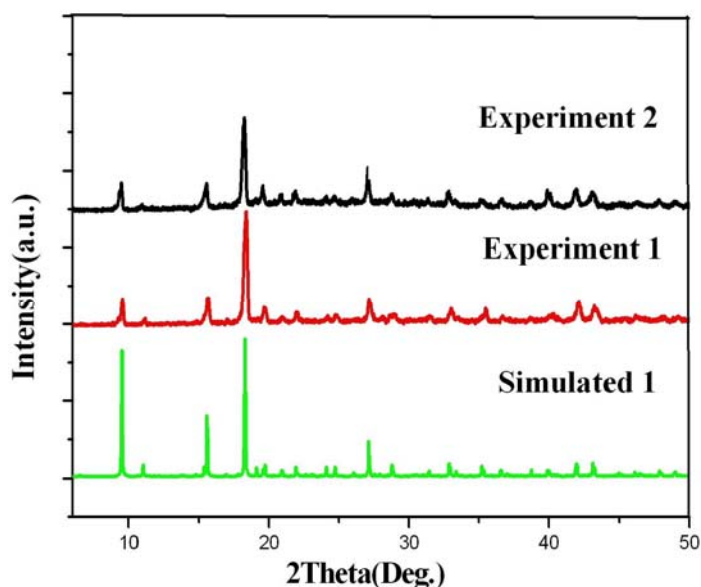


Figure S4. Simulated PXR D patterns of **1** and experimental PXR D patterns of **1** and **2**.

IR spectroscopy: The IR spectra of **1** and **2** are similar. The characteristic features of the L^- ligand dominate the IR spectrum. The strong vibrations appearing at 1589 and 1418 cm^{-1} correspond to the asymmetric and symmetric stretching vibrations of the carboxylate group, respectively. The absence of strong bands in the range 1690–1730 cm^{-1} indicate IN ligands are deprotonated

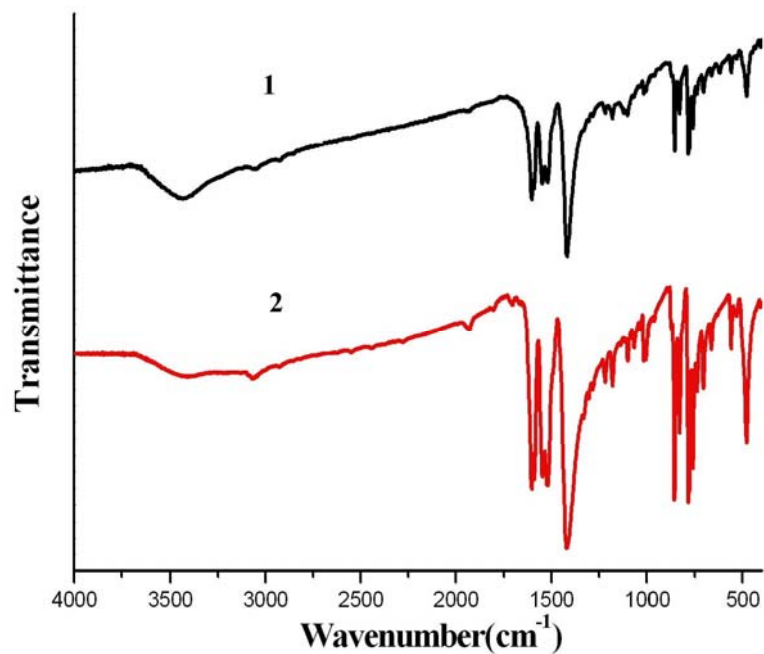


Figure S5. IR spectra of **1** and **2**.