## Supporting Information

Self-discriminating and hierarchical assembly of racemic binaphthylbisbipyridines and silver ions: from metallocycles to gel nanofibers

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## 1. General Remarks

Melting points were measured with a hot-stage XT-4 microscope and were not corrected. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were obtained using a Bruker AV600 or AV400 spectrometer at 298 K . Tetramethylsilane (TMS) was used as internal standard in ${ }^{1} \mathrm{H}$ NMR experiments, and deuterated solvents as internal standard in ${ }^{13} \mathrm{C}$ NMR experiments $\left(\mathrm{CDCl}_{3}, \delta=77.00 \mathrm{ppm} ; \mathrm{CD}_{3} \mathrm{CN}, \delta_{\mathrm{CN}}=\right.$ 117.70 ppm ). Mass spectra (ESI-MS) were recorded with a Thermo Finnigan LCQ instrument. Elemental analyses (C, H, N) were carried out with a VarioEL instrument. FTIR spectra were performed on a Bruke Vertex 70 infrared spectrometer at room temperature. TEM was performed on a JEOL JEM-1011 transmission electron microscope operated at an acceleration voltage of 100 kV . Samples were prepared by wiping a small amount of gel samples onto carbon-coated copper grid followed by naturally evaporating the solvent. SEM pictures were taken using an XL 30 ESEM FEG field emission scanning electron microscope with 20 kV operating voltage. PXRD patterns were recorded by Bruker D8 advance X-ray diffractometer with $\mathrm{CuK}_{a} 1$ radiation source operated at 1.6 kW . X-ray crystallographic analysis was performed on a Bruker SMART APEX II CCD diffractometer with graphite-monochromated $\mathrm{Mo}-\mathrm{K}_{a}$ radiation $(\lambda=0.71073 \AA)$ operated at $2.0 \mathrm{~kW}(50 \mathrm{kV}, 40 \mathrm{~mA})$. The structures were solved by direct methods using the program SHELXL-97 and refined anisotropically by full matrix least squares on $F^{2}$ values with SHELXL-97. Hydrogen atoms were located from the expected geometry and were refined only isotropically.

## 2. Synthetic procedure and characterization of compound (rac)-1



Scheme S1 The synthetic route to compound (rac)-1.
To a degassed solution of 4-bromo-2, ''-bipyridine ( $0.57 \mathrm{~g}, 2.42 \mathrm{mmol}$ ) in toluene ( 45 mL ) is added $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(0.13 \mathrm{~g}, 0.11 \mathrm{mmol})$ under a nitrogen atmosphere. After the resulting mixture is stirred for 10 min at room temperature, (rac)-3,3'-bis(pinacolboryl)-2,2'-bis(methoxymethoxy)-1,l'-binaphhyl ( $0.72 \mathrm{~g}, 1.15 \mathrm{mmol}$ ) and $1 \mathrm{M} \mathrm{Na}_{2} \mathrm{CO}_{3}(13.8 \mathrm{~mL})$ are added sequentially into the mixture. Then the mixture is refluxed for 3 days under nitrogen. After removal of the solvent, the resulting residue is extracted with dichloromethane $(50 \mathrm{~mL} \times 3)$, washed with brine $(40 \mathrm{~mL})$ and dried over anhydrous $\mathrm{MgSO}_{4}$. Evaporation of the solvent gives a residue, which is purified by flash silica gel column chromatography (dichloromethane/ethyl acetate, 2/1) to afford pure (rac)-1 in $79 \%$ yield as a white solid ( $0.62 \mathrm{~g}, 0.91 \mathrm{mmol})$. M.p. $190-192^{\circ} \mathrm{C} ; \operatorname{FTIR}\left(\mathrm{KBr}, \mathrm{cm}^{-1}\right): 1601$, $1584,1567,1543,1463,1448,1428,1385,1160,1082,1013,992,970,931,911,848,796,747$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{CN}, 600.1 \mathrm{MHz}, 4.4 \mathrm{mM}\right) \delta(\mathrm{ppm}): 8.865(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 2 \mathrm{H}), 8.848(\mathrm{~d}, J=4.8 \mathrm{~Hz}$, $2 \mathrm{H}), 8.773(\mathrm{~d}, J=4.2 \mathrm{~Hz}, 2 \mathrm{H}), 8.573(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 8.276(\mathrm{~s}, 2 \mathrm{H}), 8.126(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H})$, $7.990(\mathrm{td}, J=7.8 \mathrm{~Hz}, 1.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.812(\mathrm{dd}, J=5.4 \mathrm{~Hz}, 1.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.580(\mathrm{~m}, 2 \mathrm{H}), 7.483(\mathrm{~m}$, $2 \mathrm{H}), 7.446(\mathrm{~m}, 2 \mathrm{H}), 7.309(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.518(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.457(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H})$, $2.476(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 150.9 \mathrm{MHz}\right) \delta(\mathrm{ppm}): 156.41,155.99,151.12,149.24,148.90$, $148.13,137.05,134.18,132.98,131.08,130.73,128.23,127.19,126.51,126.36,125.63,124.86$, $123.88,121.52,121.35,99.00,56.13$; MS (ESI) $(m / z)$ Calcd. for $\mathrm{C}_{44} \mathrm{H}_{34} \mathrm{~N}_{4} \mathrm{O}_{4}: 682.3$, Found: 683.5 $([\mathrm{M}+1]+)$; Anal. Calcd. for $\mathrm{C}_{44} \mathrm{H}_{34} \mathrm{~N}_{4} \mathrm{O}_{4}: \mathrm{C}, 77.40 ; \mathrm{H}, 5.02 ; \mathrm{N}, 8.21$, Found: C, $77.36 ; \mathrm{H}, 5.08 ; \mathrm{N}$, 8.22\%.

## 3. Synthesis and characterization of complex (rac)-1•AgPF $\mathbf{6}^{2}$

A solution of $\mathrm{AgPF}_{6}(0.1 \mathrm{mmol}, 25.3 \mathrm{mg})$ and $(\mathrm{rac})-\mathbf{1}(0.1 \mathrm{mmol}, 68.3 \mathrm{mg})$ in dry $\mathrm{CH}_{3} \mathrm{CN}(20.0$ mL ) was stirred overnight in the dark, and the most part of solvents was removed under reduced pressure. The complex was precipitated with diethyl ether, filtered and dried ( $83.3 \mathrm{mg}, 89 \%$ ). Colorless rodlike single crystals were obtained by slow evaporation of a $\mathrm{CH}_{3} \mathrm{CN}$ solution. FTIR $\left(\mathrm{KBr}, \mathrm{cm}^{-1}\right): 1604,1572,1543,1479,1433,1158,1081,1016,968,920,840,793,749,558 ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{CN}, 600.1 \mathrm{MHz}, 4.4 \mathrm{mM}\right) \delta(\mathrm{ppm}): 8.798(\mathrm{~m}, 4 \mathrm{H}), 8.773(\mathrm{~s}, 2 \mathrm{H}), 8.526(\mathrm{~d}, J=7.8 \mathrm{~Hz}$, 2H), 8.337 (s, 2H), $8.140(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 8.102(\mathrm{~m}, 2 \mathrm{H}), 7.920(\mathrm{~m}, 2 \mathrm{H}), 7.597(\mathrm{~m}, 4 \mathrm{H}), 7.479$ $(\mathrm{m}, 2 \mathrm{H}), 7.335(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.544(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.481(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.514(\mathrm{~s}$, $6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{CN}, 150.9 \mathrm{MHz}, 12.6 \mathrm{mM}\right) \delta(\mathrm{ppm}): 153.51,153.24,150.55,150.12,149.79$, $148.67,138.24,133.90,132.14,131.07,130.56,128.26,127.21,126.13,125.71,125.63,125.29$, $124.85,122.26,121.93,98.70,55.37$; MS (ESI) $(\mathrm{m} / \mathrm{z}): 683.28\left([\mathbf{1}+\mathrm{H}]^{+}\right), 789.17\left([\mathbf{1} \cdot \mathrm{Ag}]^{+}\right.$and $\left.\left[\mathbf{1}_{2} \cdot \mathrm{Ag}_{2}\right]^{2+}\right)$.

## 4. Comparison of FTIR spectra of ligand (rac)-1 with its silver complex

 (rac)-1• $\mathbf{A g P F}_{6}$

Fig. S1 Comparison of FTIR spectra of complex (rac)-1• $\mathrm{AgPF}_{6}$ with ligand (rac)-1.

## 5. ${ }^{1} \mathrm{H}$ NMR spectra of complexes of (rac)-1 and $\mathrm{AgPF}_{6}$ with different metal /

## ligand ratios



Fig. S2 ${ }^{1} \mathrm{H}$ NMR spectra $\left(\mathrm{CD}_{3} \mathrm{CN}, 600.1 \mathrm{MHz}, 298 \mathrm{~K}\right)$ of complexes of $(\mathrm{rac})-\mathbf{1}$ and $\mathrm{AgPF}_{6}$ with different $\mathrm{M} / \mathrm{L}$ ratios. a) $\left.\mathrm{AgPF}_{6} /(r a c)-\mathbf{1}=1 / 1,[(r a c)-\mathbf{1}]=4.4 \mathrm{mM} ; \mathrm{b}\right) \mathrm{AgPF}_{6} /(r a c)-\mathbf{1}=1 / 2,[(r a c)-\mathbf{1}]$ $=8.8 \mathrm{mM} ; \mathrm{c}) \mathrm{AgPF}_{6} /(\mathrm{rac}) \mathbf{- 1}=2 / 1,[(r a c)-\mathbf{1}]=4.4 \mathrm{mM}$.

## 6. Mass spectrum of complex (S)-1•AgPF 6



Fig. S3 Positive ESI-MS spectrum of complex (rac)-1• $\mathrm{AgPF}_{6}$ in $\mathrm{CH}_{3} \mathrm{CN}$ (a) and mixed signals of $[1+\mathrm{Ag}]^{+}(\diamond)$ and $\left[\mathbf{1}_{2}+\mathrm{Ag}_{2}\right]^{2+}(\bigcirc)$ at $m / z=789.17(\mathrm{~b}) . c=1 \times 10^{-6} \mathrm{M}$.

## 7. Variable temperature ${ }^{1} \mathrm{H}$ NMR spectra of complex (rac)-1• $\mathrm{AgPF}_{6}$ in $\mathrm{CD}_{3} \mathrm{CN}$



Fig. S4 VT- ${ }^{1} \mathrm{HNMR}$ spectra of complex ( rac ) $-\mathbf{1} \cdot \mathrm{AgPF}_{6}\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}, c=9.0 \mathrm{mM}\right)$. Lowering the temperature led to broader signals, resulting from a slower exchange process or the formation of the aggregated species.

## 8. Gel tests

A mixture of complex $(\mathrm{rac}) \mathbf{- 1} \cdot \mathrm{AgPF}_{6}$ and the tested solvent $(1.0 \mathrm{~mL})$ were charged with a glass test tube ( $35 \mathrm{~mm} \times 15 \mathrm{~mm}$ ), which was capped and sonicated for a few minutes using sonoreactor $\left(0.26 \mathrm{Wcm}^{-1}, 40 \mathrm{kHz}\right)$, and heated in an oil bath $\left(T=80^{\circ} \mathrm{C}\right)$ until it turned into a clear solution (if soluble). After the solution was allowed to stand at room temperature $\left(25 \pm 5^{\circ} \mathrm{C}\right)$ for 5 h , the state of the mixture was evaluated by the "stable to inversion of test tube" method. The complex (rac)-1•$\cdot \mathrm{AgPF}_{6}$ can successfully gelate $\mathrm{CH}_{3} \mathrm{CN}$ after screening common solvents. The minimum gel concentration (MGC) of complex (rac)-1• $\mathrm{AgPF}_{6}$ in $\mathrm{CH}_{3} \mathrm{CN}$ is $9.4 \times 10^{-2} \mathrm{M}$ at room temperature.
9. Crystal data and structure refinement for (rac)-1 and (rac)-1•AgPF $\mathbf{6}_{6}$

| Compounds | (rac)-1 | ( rac ) $\mathbf{1} \cdot \mathrm{AgPF}_{6}$ |
| :---: | :---: | :---: |
| Empirical formula | $\mathrm{C}_{44} \mathrm{H}_{34} \mathrm{~N}_{4} \mathrm{O}_{4}$ | $\mathrm{C}_{88} \mathrm{H}_{68} \mathrm{~N}_{8} \mathrm{O}_{8} \mathrm{Ag}_{2} \mathrm{P}_{2} \mathrm{~F}_{12}$ |
| Formula weight | 682.75 | 1871.18 |
| Temperature (K) | 185(2) | 185(2) |
| Crystal system, Space group | Monoclinic, $P 2_{1} / \mathrm{c}$ | Triclinic, $P-1$ |
| Unit cell dimensions | $\begin{aligned} a & =21.4815(13) \AA \\ b & =12.1504(7) \AA \\ c & =13.4906(8) \AA \\ \alpha & =90^{\circ} \\ \beta & =99.2200(10)^{\circ} \\ \gamma & =90^{\circ} \end{aligned}$ | $\begin{aligned} & a=8.6460(8) \AA \\ & b=13.9854(13) \AA \\ & c=18.4584(17) \AA \\ & \alpha=89.7640(10)^{\circ} \\ & \beta=84.3230(10)^{\circ} \\ & \gamma=74.2640(10)^{\circ} \end{aligned}$ |
| Volume ( $\AA^{3}$ ) | 3475.7(4) | 2137.2(3) |
| Z, Calculated density ( $\mathrm{Mg} / \mathrm{m}^{3}$ ) | 4, 1.305 | 1,1.454 |
| Absorption coefficient ( $\mathrm{mm}^{-1}$ ) | 0.085 | 0.582 |
| $F(000)$ | 1432 | 948 |
| Crystal size (mm) | $0.30 \times 0.25 \times 0.21$ | $0.38 \times 0.22 \times 0.14$ |
| $\theta$ range for data collection ( ${ }^{\circ}$ ) | 0.96 to 26.10 | 1.11 to 25.09 |
| Limiting indices | $\begin{aligned} & -20 \leq \mathrm{h} \leq 26, \\ & -15 \leq \mathrm{k} \leq 12, \\ & -16 \leq 1 \leq 16 \end{aligned}$ | $\begin{aligned} & -9 \leq h \leq 10, \\ & -16 \leq k \leq 16, \\ & -21 \leq l \leq 22 \end{aligned}$ |
| Reflections collected / unique | $19104 / 6900\left[R_{\text {int }}=0.0571\right]$ | $16807 / 7519$ [ $\left.R_{\text {int }}=0.0343\right]$ |
| Completeness to $\theta$ | 26.10, 99.7\% | 25.09, 98.8\% |
| Absorption correction | Semi-empirical from <br> equivalents | Semi-empirical from <br> equivalents |
| Max. and min. transmission | 0.978 and 0.970 | 0.9230 and 0.8093 |
| Refinement method | Full-matrix least-squares on $F^{2}$ | Full-matrix least-squares on $F^{2}$ |
| Data / restraints / parameters | 6900/0/469 | 7519/14/570 |
| Goodness-of-fit on $F^{2}$ | 0.993 | 0.964 |
| Final $R$ indices $[I>2 \sigma(\mathrm{I})]$ | $R_{1}=0.0458, w R_{2}=0.0928$ | $R_{1}=0.0693, w R_{2}=0.2194$ |
| $R$ indices (all data) | $R_{1}=0.0987, w R_{2}=0.1201$ | $R_{1}=0.0882, w R_{2}=0.2372$ |
| Largest diff. peak and hole (e. $\AA^{-3}$ ) | 0.205 and -0.216 | 2.061 and -1.038 |
| CCDC | 787118 | 774705 |

## 10. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of (rac)-1 and (rac)-1• $\mathrm{AgPF}_{6}$

## 1H NMR spectrum of (rac)-1 in CD3CN (600.1 MHz, 4.4 mM$)$



13C NMR spectrum of (rac)-1 in CDCl3 (150.9 MHz, 298K)



13C NMR spectrum of (rac)-1AgPF6 in CD3CN (150.9 MHz, $12.6 \mathbf{m M}$ )


