## **Supporting Information**

## Cuboctahedral Pd nanoparticles on WC for Enhanced Methanol Electrooxidation in Alkaline Solution

Young-Woo Lee, A-Ra Ko, Sang-Beom Han, Hyun-Su Kim, Do-Young Kim,

Si-Jin Kim, and Kyung-Won Park\*

Department of Chemical and Environmental Engineering, Soongsil University

Seoul 156743, Republic of Korea

\*Corresponding author. E-mail: kwpark@ssu.ac.kr

## Experimental

**Synthesis of WC.** Tungsten oxide hydrate were synthesized by hydrothermal method. Sodium metatungstate hydrate ( $3Na_2WO_4 \cdot 9WO_3 \cdot xH_2O$ , Aldrich) as precursor was dissolved in 5 M hydrochloric acid (Sigma) solution by vigorous stirring at an ambient room temperature or below. The resulting precursor solutions were heated to 120 °C at a heating rate of approximately 3 °C min<sup>-1</sup>. With vigorous stirring, the acidic tungstate solutions were aged at that temperature for 60 min to allow a complete precipitating reaction. The resulting precipitates were thoroughly washed with de-ionized water, filtered by using a filtration and finally dried at 70 °C overnight.

Next, Tungsten oxide obtained by hydrothermal method, was loaded into an alumina boat, which was then inserted into a vitreous silica tube inside a cylindrical furnace. A gas mixture of 20 ml min<sup>-1</sup>  $CH_4$  with 80 ml min<sup>-1</sup>  $H_2$  was passed through the quartz tube at a rate of 100 ml min<sup>-1</sup> during the entire reaction. Tungsten carbide samples studied in this work were prepared at 900 °C. After carburization, all of the samples were passivated in a air gas flow for more than 5h before being removed from silica tube.

Synthesis of cuboctahedral Pd NPs on WC support catalyst. The as-prepared WC supported Pd nanoparticles were prepared by reducing Pd salt in ethylene glycol (EG) solution. A solution of 2 mM Na<sub>2</sub>PdCl<sub>4</sub>, 10 mM NaNO<sub>3</sub>, and 20  $\mu$ M FeCl<sub>3</sub> was dissolved in 50 ml of EG with 50 mg poly(vinyl pyrrolidone) (PVP, MW = 29,000). The mixed Pd salt solution and as-prepared WC (255.4 mg) treated were added to the solution and mixed with continuous stirring at room temperature. All chemicals used were of analytical grade. The solution was raised by 5 °C min<sup>-1</sup> and was kept for 90 min at 250 °C until Na<sub>2</sub>PdCl<sub>4</sub> was completely reduced by EG. The resulting colloid solution was cooled at room temperature, washed with water and then with ethanol several times to remove ethylene glycol and excess PVP.

**Characterizations**. For the structure analysis of the catalysts, X-ray diffraction (XRD) analysis was carried out using Rigaku X-ray diffractometer with Cu K<sub> $\alpha$ </sub> ( $\lambda = 0.15418$  nm) source with a Ni filter. The source was operated at 40 kV and 100 mA. The 2 $\theta$  angular scan from 20° to 80° was explored at a scan rate of 5° min<sup>-1</sup>. For all the XRD measurement, the resolution in the scans was kept at 0.02°. The

Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2010

morphology and size distribution of the catalysts were characterized by field-emission transmission electron microscopy (FE-TEM) using a Tecnai G2 F30 system operating at 300 kV. TEM samples were prepared by placing drops of catalyst suspension dispersed in ethanol on a carbon-coated copper grid. Energy dispersive X-ray (EDX) analysis of the catalysts was performed on a field emission transmission electron microscope (FE-TEM, Tecnai G2 F30 system).

Electrochemical properties of the catalysts were measured in a three-electrode cell at 25 °C using a potentiostat (Eco Chemie, AUTOLAB). A Pt wire and Hg/HgO (in saturated 1.0 M NaOH) were used as a counter and reference electrode, respectively. The glassy carbon electrode as a working electrode was polished with 1, 0.3, and 0.05  $\mu$ m Al<sub>2</sub>O<sub>3</sub> paste and then washed in deionized water. The catalyst ink was prepared by ultrasonically dispersing catalyst powders in an appropriate amount of Millipore water. The catalyst ink was dropped onto a glassy carbon working electrode (geometrical area of 0.0706 cm<sup>2</sup>). After drying in 50 °C oven, total metal loading of cuboctahedral Pd/WC and Pd/C was 8 and 20  $\mu$ g cm<sup>-2</sup>, respectively. To compare electrochemical properties and catalytic activity of the catalysts, cyclic voltammograms (CVs) were obtained between -0.8 to +0.2 V in 1.0 M KOH and 2.0 M KOH + 2 M CH<sub>3</sub>OH, respectively.



*Figure S1.* X-ray diffraction pattern of as-prepared WO<sub>3</sub>·H<sub>2</sub>O measured in the scan range 20-60° as compared with that of the reference (JCPDS No. 43-0679).

Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2010



*Figure S2.* (A) Field-emission transmission electron microscopy (FE-TEM) and (B) high-resolution transmission electron microscopy (HR-TEM) images of as-prepared WO<sub>3</sub>·H<sub>2</sub>O particles. (C) FE-TEM and (D) HR-TEM images of as-synthesized WC particles. (E) HR-TEM image of WC containing {101} facets. The inset indicates a FFT pattern of WC. (F) The profile of the lattice fringe of WC with {101} facets.



Figure S3. Composition and weight ratio of Pd NPs and WC obtained by means of EDX analysis.