## Supporting Information Large-Scale Synthesis of Water-Soluble Nanowires as Versatile Templates for Nanotubes

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## **Experimental**

Synthesis of Na<sub>2</sub>SiF<sub>6</sub> nanowire

Na<sub>2</sub>SiF<sub>6</sub> nanowires were prepared by mixing 2 mmol Na<sub>2</sub>SiO<sub>3</sub>, 15mmol NaF, and 0.5g PAA (( $C_3H_4O_2$ )<sub>n</sub>) in ethylene glycol and stirring for 1 h at 120 °C. The white precipitates were centrifugalized, washed with ethanol and finally dried at 80°C in air. *Synthesis of Pt, CdS and SnO<sub>2</sub> nanotubes* 

For synthesis of Pt nanotubes, 10 mg  $Na_2SiF_6$  nanowires were dispersed in 20 ml ethanol solution of  $H_2PtCl_6$  (4 mM). The solution was vigorously stirred and followed by dropwise adding  $NaBH_4$  ethanol solution (30 mL, 10mM). The black precipitates were centrifugalized, washed with ethanol and finally rinsed with water to obtain pure Pt nanotubes.

For synthesis of CdS nanotubes, 10 mg  $Na_2SiF_6$  nanowires were added into 30 ml tetrahydrofuran (THF) solution of sulphur powder and CdCl<sub>2</sub> (1 mmol), which were

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sonicated for 0.5 h. And then, 30 ml THF solution of NaBH<sub>4</sub> (20 mM) was dropwise added. The yellow precipitates were centrifugalized, washed with ethanol and finally rinsed with water to obtain pure CdS nanotubes.

For synthesis of  $SnO_2$  nanotubes, 10 mg  $Na_2SiF_6$  nanowires were dispersed in 20 ml ethanol solution of  $SnCl_4$  (5 mM). The solution was vigorously stirred and followed by dropwise adding  $NaBH_4$  ethanol solution (30 mL, 10 mM). The white precipitates were centrifugalized, washed with ethanol and finally rinsed with water to obtain pure  $SnO_2$  nanotubes.

## Characterization of products

The obtained products were characterized by field emission scanning electron microscopy (FESEM, Hitachi S-4800), and high-resolution transmission electron microscopy (HRTEM, JEOL JEM-2010F) with energy-dispersive X-ray spectrometer (EDX). The infrared (IR) spectra were measured with a Nicolet Nexus FTIR 670 spectrophotometer.



Figure S1 SEM (a) and XRD pattern (b) of  $Na_2SiF_6$  particles synthesized by using acetic acid as a surfactant instead of PAA



Figure S2 FTIR spectrum of as-synthesized Na<sub>2</sub>SiF<sub>6</sub> nanowires



Figure S3 (a) SEM image and (b) corresponding EDX pattern of the  $Na_2SiF_6$  nanowires, (c) SEM image and (d) corresponding EDX pattern of the  $Na_2SiF_6$ -Pt core-shell nanowires, (e) SEM image and (f) corresponding EDX pattern of the Pt nanotubes.



Figure S4 HRTEM image of Pt nanotube.



Figure S5 TEM image and corresponding EDX mapping of an individual CdS (a) and  $SnO_2$  nanotube (b).