Synthesis and Electric Properties of Dicobalt Silicide Nanobelts

Yongquan Qu,^a Jingwei Bai,^b Lei Liao,^a Rui Cheng,^b Yung-Chen Lin,^b Yu Huang,^{b,c} Ting Guo,^d Xiangfeng Duan^{a,c,*}

^a Department of Chemistry and Biochemistry, University of California, Los Angeles, CA 90095 Department of Materials Science and Engineering, University of California, Los Angeles, CA 90095 California Nanosystems Institute, University of California, Los Angeles, CA 90095 Department of Chemistry, University of California, Davis, CA 95616

*To whom correspondence should be addressed. E-mail: xduan@chem.ucla.edu

- 1. Experimental.
- 2. Fig. S1. SEM and TEM images of the nanostructures obtained at various temperatures of CoCl₂ precursor.
- 3. Fig. S2. AFM image of the Co₂Si nanobelt used in the electrical and MR measurements.
- 4. Fig. S3. TEM image showing the oxide layer on a Co₂Si nanobelt.
- 5. Fig. S3. Magnetoresistance curve of Co_2Si nanobelt in the range of \pm 5 T at 1.5 K.

1. Experimental

Synthesis of Dicobalt Silicide Nanobelts: Silicon (100) pieces with native silicon oxide layers were used as the growth substrate. The Si pieces were washed with Piranha [1:4 H_2O_2 (30%): H_2SO_4 (concentrated)], rinsed with deionized water, acetone and isopropanol, and dried with N_2 gas blow. The cleaned growth substrates were then placed at the center of a single-zone horizontal tube furnace and ca. 50 mg anhydrous $CoCl_2$ in a quartz boat was placed upstream of the growth substrate. Initially, the system was evacuated to less than 30 mTorr. Ar (99.997%) gas at a flow rate of 200 sccm was fed through the quartz tube as the carrier gas, and the total pressure was maintained at ~ 1 psi above the ambient pressure. Temperature of the growth substrate was fixed at 750 °C and temperature of CoCl₂ precursor controlled between 600 °C and 630 °C.

Device Fabrication and Low Temperature Measurements: The Co₂Si nanobelts were transferred from the growth substrate onto a Si substrate with a 300 nm silicon oxide layer via contact printing dry transfer process. The contact electrodes were defined using e-beam lithography followed by Ti/Au (50nm/50nm) metallization. The low-temperature measurement was carried out with Oxford cryogenic system.

Characterization: The as-grown samples were inspected by scanning electron microscope (SEM) (JEOL 6700) with an electron acceleration voltage of 10 keV. Transmission electron microscope (TEM) imaging was conducted on a Phillips CM120 with a 120 kV operation voltage. The HRTEM, SAED and EDX spectroscopy were collected on FEI TITAN with a 300 kV operation voltage.

2. SEM and TEM images of the nanostructures obtained at various temperatures of CoCl₂ precursor.



Fig. S1. SEM and TEM images of the nanostructures obtained at various temperatures of CoCl₂ precursor: (A), (B) 600 °C; (C), (D) 615 °C.

3. AFM image of the Co₂Si nanobelt used in electrical and MR measurements.



Fig. S2. AFM image of the Co₂Si nanobelt used in electrical and MR measurements shows an overall thickness of 25 nm (including surface oxide layers). The image area is $2 \ \mu m \times 1 \ \mu m$.

4. TEM image showing the oxide layer on a Co2Si nanobelt.



Fig. S3: TEM image of Co₂Si nanobelt with 7 nm silicon oxide shell.

5. Magnetoresistance curve of the Co_2Si nanobelt in the range of ± 5 T at 1.5 K.



Fig. S4. Magnetoresistance curve of the Co₂Si nanobelt in the range of \pm 5 T at 1.5 K.