

Supporting Information

Chloride-mediated Chain Generation of Difluorocarbene from $\text{Me}_3\text{SiCF}_2\text{Cl}$ for Efficient Preparation of *gem*-Difluorinated Cyclopropenes and Cyclopropanes

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General remarks:

Unless otherwise mentioned, solvents and reagents were purchased from commercial sources and used as received. The solvent THF and toluene were distilled from sodium. Tetrabutylammonium fluoride (TBAF) [either as 1.0 M THF solution with 5wt% water content (for reactions in THF), or as solid TBAF hydrate (for reactions in toluene)] and tetrabutylammonium chloride (hydrate) were bought from commercial sources and were used as received. ^1H , ^{13}C and ^{19}F NMR spectra were recorded on a 400 MHz or 300 MHz NMR spectrometer. ^1H NMR chemical shifts were determined relative to internal $(\text{CH}_3)_4\text{Si}$ (TMS) at δ 0.0 or to the signal of a residual protonated solvent: CHCl_3 δ 7.26. ^{13}C NMR chemical shifts were determined relative to internal TMS at δ 0.0. ^{19}F NMR chemical shifts were determined relative to CFCl_3 at δ 0.0. Mass spectra were obtained on a mass spectrometer in the EI, ESI or MALDI mode. High-resolution mass data were recorded on a high-resolution mass spectrometer in the EI, ESI or MALDI mode.

Preparation of TMSCF_2Cl (1):

TMSCF_2Cl (1) was prepared by using literature procedures.^[1] Compound 1 was obtained as a colorless liquid, b.p. 81–82 °C. ^1H NMR: δ 0.29 (s, 9H). ^{19}F NMR: δ -62.2 (s, 2F). The characterization data were consistent with those in the literature.^[1]

Typical procedures for [2 + 1] cycloaddition between 1 and alkynes 4:

1,2-Diphenylethyne (**4f**, 0.178 g, 1.0 mmol), TMSCF_2Cl (0.317, 2.0 mmol), $n\text{Bu}_4\text{NCl}$ (0.011 g, 0.04 mmol), and toluene (3.0 mL) were added into a pressure tube at room temperature. Then the reaction mixture was heated at 110 °C for 4 h. The reaction was quenched by adding saturated Na_2CO_3 solution (5 mL), followed by extraction with Et_2O (20 mL x 2). The organic phase was separated and dried over anhydrous K_2CO_3 . After the removal of solvent under vacuum, the residue was subjected to silica gel column chromatography with petroleum ether/ Et_3N (40/1 v/v) as eluent [The column should be eluted previously with petroleum ether/ Et_3N (10/1 v/v)]. The product (3,3-difluorocycloprop-1-ene-1,2-diyl)dibenzene (**5f**) was obtained as a white solid (213 mg, yield: 93%), m.p. 58–59 °C. ^1H NMR: δ 7.86–7.73 (m, 4H), 7.61–7.45 (m, 6H). ^{19}F NMR: δ -112.7 (s, 2F). MS (EI, m/z , %): 228 (M^+ , 81.58), 178 (100.00). The characterization data were consistent with the previous report.^[2]

Characterization data for the isolated compounds (5):

1-(3,3-Difluorocycloprop-1-enyl)-4-methylbenzene (5d). Light yellow solid. m.p. 39–40 °C. IR (film): 2926, 1800, 1631, 1609, 1507, 1454, 1317, 1297, 1283, 1181, 1106, 1024, 826, 772, 503, 475 cm⁻¹. ¹H NMR: δ 7.55 (d, *J* = 8.4 Hz, 2H), 7.37 (t, *J* = 2.1 Hz, 1H), 7.28 (d, *J* = 8.7 Hz, 2H), 2.42 (s, 3H). ¹⁹F NMR: δ -106.9 (s, 2F). ¹³C NMR: δ 142.3, 133.8 (t, *J* = 10.2 Hz), 130.1, 129.8, 120.6, 112.1 (t, *J* = 12.5 Hz), 101.9 (t, *J* = 268.1 Hz), 21.6. MS (EI, *m/z*, %): 166 (M⁺, 72.08), 115 (100.00), 151 (94.84), 116 (41.65). HRMS (EI): calcd. for C₁₀H₈F₂: 166.0594; Found: 166.0591.

2-(3,3-Difluorocycloprop-1-enyl)-6-methoxynaphthalene (5e). White solid. m.p. 84–85 °C. IR (film): 3125, 2973, 1717, 1625, 1601, 1484, 1390, 1292, 1272, 1202, 1166, 1029, 1006, 902, 859, 823, 785, 738, 478 cm⁻¹. ¹H NMR: δ 8.12–7.13 (m, 7H), 3.95 (s, 3H). ¹⁹F NMR: δ -106.8 (s, 2F). ¹³C NMR: δ 159.4, 136.2, 133.9 (t, *J* = 10.2 Hz), 131.0, 130.3, 128.3, 127.7, 126.5, 118.8, 118.3, 112.2 (t, *J* = 12.5 Hz), 105.9, 102.0 (t, *J* = 269.0 Hz), 55.4. MS (EI, *m/z*, %): 232 (M⁺, 74.42), 182 (100.00). HRMS (EI): calcd. for C₁₄H₁₀OF₂: 232.0700; Found: 232.0705. Anal. Calcd. for C₁₄H₁₀OF₂: C, 72.41; H, 4.34; Found: C, 72.36; H, 4.29.

1-Butyl-3,3-difluoro-2-octylcycloprop-1-ene (5f). Colorless liquid. IR (film): 2933, 2860, 1806, 1467, 1311, 1109, 998, 809 cm⁻¹. ¹H NMR: δ 2.43–2.34 (m, 4H), 1.68–1.50 (m, 4H), 1.48–1.19 (m, 12H), 1.01–0.81 (m, 6H). ¹⁹F NMR: δ -106.6 (s, 2F). ¹³C NMR: δ 128.75 (t, *J* = 10.6 Hz), 128.68 (t, *J* = 10.8 Hz), 105.4 (t, *J* = 269.2 Hz), 31.8, 29.3, 29.11, 29.06, 28.9, 22.8, 22.6, 22.5, 22.2, 14.0, 13.6, 0.9. HRMS (EI): calcd. for C₁₅H₂₆F₂: 244.2003; Found: 244.2002.

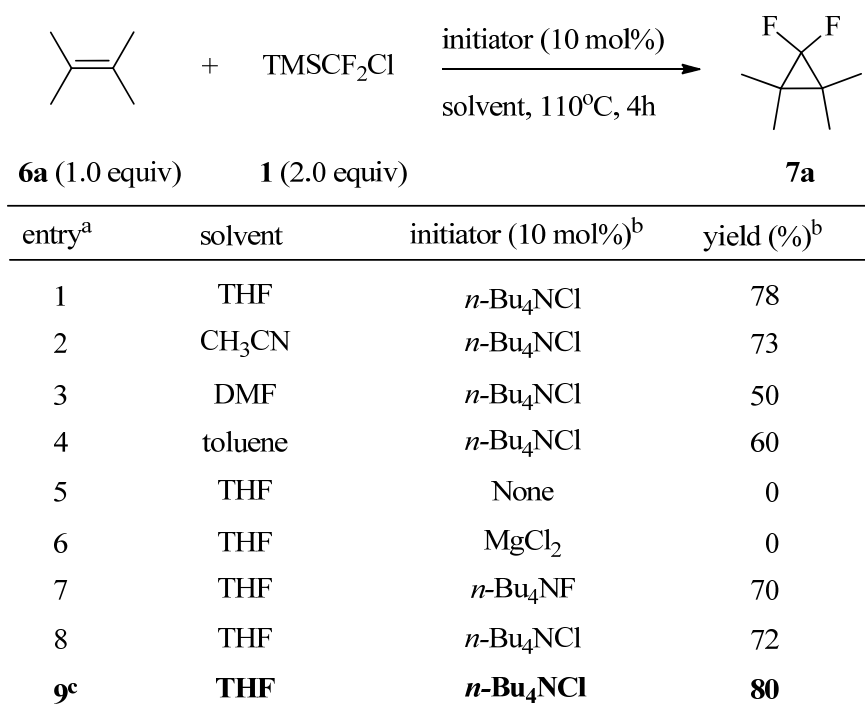
(3,3-Difluoro-2-phenylcycloprop-1-enyl)(phenyl)methyl acetate (5h). Light yellow solid. m.p. 39 °C. ¹H NMR: δ 7.52–7.37 (m, 10H), 6.89–6.84 (m, 1H), 2.20 (s, 3H). ¹⁹F NMR: δ -108.1 (s, 2F). MS (EI, *m/z*, %): 257 (M⁺ – COMe, 9.83), 105 (100.00), 43 (77.39), 258 (35.60). The characterization data were consistent with the previous report.^[3]

(3,3-Difluoro-2-phenylcycloprop-1-enyl)methyl acetate (5i). Light yellow solid. m.p. 66–67 °C. ¹H NMR: δ 7.67–7.56 (m, 2H), 7.52–7.44 (m, 3H), 5.31–5.11 (m, 2H), 2.21 (s, 3H). ¹⁹F NMR: δ -100.5 (s, 2F). MS (EI, *m/z*, %): 181 (M⁺ – COMe, 39.15), 43 (100.00), 182 (35.43). The characterization data were consistent with the previous report.^[3]

(2-Butyl-3,3-difluorocycloprop-1-enyl)benzene (**5j**). Colorless liquid. IR (film): 3064, 2962, 2935, 2876, 1888, 1797, 1467, 1450, 1320, 1283, 1006, 836, 765, 691, 612, 513, 447 cm^{-1} . ^1H NMR: δ 7.63–7.53 (m, 2H), 7.51–7.41 (m, 3H), 2.72–2.61 (m, 2H), 1.81–1.68 (m, 2H), 1.56–1.39 (m, 2H), 0.98 (t, $J = 7.5$ Hz, 3H). ^{19}F NMR: δ -109.4 (s, 2F). ^{13}C NMR: δ 130.4, 129.8, 128.9, 127.2 (t, $J = 11.2$ Hz), 125.8 (t, $J = 10.9$ Hz), 124.3, 103.7 (t, $J = 269.8$ Hz), 30.7, 23.2, 22.2, 13.6. MS (EI, m/z , %): 208 (M^+ , 2.40), 115 (100.00), 165 (58.27), 129 (56.03). HRMS (EI): calcd. for $\text{C}_{13}\text{H}_{14}\text{F}_2$: 208.1064; Found: 208.1067.

(3,3-Difluoro-2-isobutylcycloprop-1-enyl)benzene (**5k**). Colorless liquid. IR (film): 2964, 1797, 1468, 1261, 1094, 1010, 869, 795, 763, 691 cm^{-1} . ^1H NMR: δ 7.62–7.54 (m, 2H), 7.49–7.41 (m, 3H), 2.62–2.53 (m, 2H), 2.17–2.00 (m, 1H), 1.06 (d, $J = 6.0$ Hz, 6H). ^{19}F NMR: δ -108.5 (s, 2F). ^{13}C NMR: δ 130.6, 129.8, 129.1, 126.6 (t, $J = 12.6$ Hz), 126.5 (t, $J = 10.8$ Hz), 124.5, 103.7 (t, $J = 269.8$ Hz), 32.7, 27.3, 22.3. HRMS (EI): calcd. for $\text{C}_{13}\text{H}_{14}\text{F}_2$: 208.1064; Found: 208.1063.

Survey of [2 + 1] cycloaddition reaction between **1** and alkenes **6**:



^a Yields were determined by ^{19}F NMR spectroscopy using PhCF_3 as an internal standard. ^b The amount of initiator was calculated on the basis of the amount of reagent **1**. ^c 2 mol% of *n*-Bu₄NCl was used.

Typical procedure for [2 + 1] cycloaddition between **1** and alkenes **6**:

1-Bromo-4-vinylbenzene (**6f**, 0.183 g, 1.0 mmol), TMSF₂Cl (0.476, 3.0 mmol), *n*Bu₄NCl (0.022 g, 0.06 mmol), and THF (3.0 mL) were added into a pressure tube at room temperature. The reaction mixture was heated at 110 °C for 4 h. Then, the reaction was quenched by adding water (5 mL), followed by extraction with Et₂O (20 mL x 2). The organic phase was separated and dried over anhydrous MgSO₄. After the removal of solvent under vacuum, the residue was subjected to silica gel column chromatography with petroleum ether as eluent. The product 1-bromo-4-(2,2-difluorocyclopropyl) benzene (**7f**) was obtained as a colorless liquid (231 mg, yield: 99%). IR (film): 1495 147 1379 1419 1305 1232 1189 1041 1014 933 828 767 721 506 cm⁻¹. ¹H NMR: δ 7.46 (d, *J* = 8.4 Hz, 2H), 7.10 (d, *J* = 8.2 Hz, 2H), 2.78–2.62 (m, 1H), 1.91–1.76 (m, 1H), 1.65–1.51 (m, 1H). ¹⁹F NMR: δ -126.6 (dtd, *J* = 154.9 Hz, *J* = 12.8 Hz, *J* = 3.8 Hz, 1F), δ -142.7 (ddd, *J* = 154.5 Hz, *J* = 13.0 Hz, *J* = 5.1 Hz, 1F). ¹³C NMR: δ 132.7, 131.5, 129.7 (t, *J* = 1.9 Hz), 121.1, 112.2 (dd, *J* = 285.5 Hz, *J* = 286.7 Hz), 26.6 (t, *J* = 11.2 Hz), 17.1 (t, *J* = 11.4 Hz). MS (EI, *m/z*, %): 232 (M⁺, 4.87), 153 (100.00), 234 (4.76), 133 (60.86). HRMS (EI): calcd. for C₁₃H₁₄F₂: 231.9699; Found: 231.9698.

Characterization data for the isolated compounds (**7**):

(2,2-Difluoro-1-methylcyclopropyl)benzene (**7b**). Colorless liquid. ¹H NMR: δ 7.38–7.26 (m, 5H), 1.73–1.63 (m, 1H), 1.54–1.50 (m, 3H), 1.45–1.35 (m, 1H). ¹⁹F NMR: δ -131.7 (dd, *J* = 150.0 Hz, *J* = 13.2 Hz, 1F), δ -136.7 (dd, *J* = 150.0 Hz, *J* = 12.8 Hz, 1F). MS (EI, *m/z*, %): 168 (M⁺, 28.30), 153 (100.00). The characterization data were consistent with the previous report.^[4]

1-(2,2-Difluoro-1-methylcyclopropyl)-4-methoxybenzene (**7c**). Colorless liquid. ¹H NMR: δ 7.23 (d, *J* = 9.0 Hz, 2H), 6.87 (d, *J* = 8.8 Hz, 2H), 3.79 (s, 3H), 1.67–1.56 (m, 1H), 1.49 (s, 3H), 1.41–1.30 (m, 1H). ¹⁹F NMR: δ -133.1 (dd, *J* = 149.9 Hz, *J* = 13.4 Hz, 1F), δ -138.0 (dd, *J* = 149.7 Hz, *J* = 12.3 Hz, 1F). MS (EI, *m/z*, %): 198 (M⁺, 25.39), 183 (100.00). The characterization data was consistent with the previous report.^[5]

(2,2-Difluoro-1-pentylcyclopropyl)benzene (**7d**). Colorless liquid. IR (film): 3030 2931 2861 1605 1464 1447 1378 1311 1213 1151 1053 1016 960 905 700 640 cm⁻¹. ¹H NMR: δ 7.39–7.21 (m, 5H), 1.87–1.55 (m, 3H), 1.45–1.33 (m, 1H), 1.21 (s, 8H), 0.84 (t, *J* = 6.3 Hz, 3H). ¹⁹F NMR: δ -130.7 (dd, *J* = 148.3 Hz, *J* = 13.5 Hz, 1F), δ -138.1 (dd, *J* = 148.9 Hz, *J* = 12.7 Hz, 1F). ¹³C NMR: δ 129.44, 129.41, 128.3, 127.2,

125.8 (t, $J = 287.4$ Hz), 36.0 (dd, $J = 9.7$ Hz, $J = 9.0$ Hz), 34.6 (dd, $J = 5.7$ Hz, $J = 2.8$ Hz), 31.7, 29.1, 26.3 (d, $J = 2.2$ Hz), 22.6, 22.1 (t, $J = 9.7$ Hz), 14.0. MS (EI, m/z , %): 238 (M^+ , 32.23), 153 (100.00), 167 (57.92). HRMS (EI): calcd. for $C_{15}H_{20}F_2$: 238.1533; Found: 238.1531.

(2,2-Difluorocyclopropyl)benzene (**7e**). Colorless liquid. 1H NMR: δ 7.46–7.18 (m, 5H), 2.87–2.65 (m, 1H), 1.92–1.71 (m, 1H), 1.70–1.54 (m, 1H). ^{19}F NMR: δ 124.5–125.5 (m, 1F), 141.1–142.1 (m, 1F). MS (EI, m/z , %): 154 (M^+ , 100.00), 133 (73.36). The characterization data was consistent with the previous report.^[6]

4-((2,2-Difluorocyclopropyl)methyl)-1,2-dimethoxybenzene (**7g**). Colorless liquid. IR (film): 3006, 2939, 2837, 1592, 1517, 1473, 1286, 1265, 1239, 1197, 1158, 1144, 1024, 905, 801, 766 cm^{-1} . 1H NMR: δ 6.85–6.74 (m, 3H), 3.88 (s, 3H), 3.87 (s, 3H), 2.84–2.65 (m, 2H), 1.86–1.68 (m, 1H), 1.54–1.40 (m, 1H), 1.14–1.01 (m, 1H). ^{19}F NMR: δ -127.5 (dt, $J = 156.0$ Hz, $J = 13.3$ Hz, 1F), δ -142.2 (ddd, $J = 156.2$ Hz, $J = 13.4$ Hz, $J = 4.1$ Hz, 1F). ^{13}C NMR: δ 149.0, 147.7, 132.2, 120.0, 114.4 (t, $J = 283.3$ Hz), 111.5, 111.3, 55.9, 55.8, 32.4 (d, $J = 3.6$ Hz), 23.5 (t, $J = 10.2$ Hz), 16.3 (t, $J = 10.9$ Hz). MS (EI, m/z , %): 228 (M^+ , 81.80), 151 (100.00), 187 (86.62). HRMS (EI): calcd. for $C_{12}H_{14}O_2F_2$: 228.0962; Found: 228.0960.

4-((2,2-Difluorocyclopropyl)methyl)-2-methoxyphenol (**7h**). Colorless liquid. IR (film): 3545, 3031, 2941, 2848, 1641, 1516, 1474, 1434, 1272, 1236, 1201, 1153, 1125, 1023, 904, 810, 561 cm^{-1} . 1H NMR: δ 6.88–6.71 (m, 3H), 5.51 (s, 1H), 3.89 (s, 3H), 2.81–2.63 (m, 2H), 1.85–1.66 (m, 1H), 1.54–1.39 (m, 1H), 1.14–0.99 (m, 1H). ^{19}F NMR: δ -127.4 (dt, $J = 156.4$ Hz, $J = 13.1$ Hz, 1F), δ -142.2 (ddd, $J = 156.0$ Hz, $J = 12.6$ Hz, $J = 4.3$ Hz, 1F). ^{13}C NMR: δ 146.5, 144.2, 131.6, 120.8, 114.4, 114.4 (t, $J = 282.9$ Hz), 110.7, 55.8, 32.5 (d, $J = 4.4$ Hz), 23.6 (t, $J = 11.0$ Hz), 16.3 (t, $J = 10.9$ Hz). MS (EI, m/z , %): 214 (M^+ , 77.21), 173 (100.00). HRMS (EI): calcd. for $C_{11}H_{12}O_2F_2$: 214.0805; Found: 214.0806.

4-((2,2-Difluorocyclopropyl)methyl)-1-(difluoromethoxy)-2-methoxybenzene (**7i**). Colorless liquid. IR (film): 3018, 2944, 2851, 1600, 1515, 1475, 1423, 1385, 1127, 1036, 985, 970, 904, 802 cm^{-1} . 1H NMR: δ 7.10 (d, $J = 8.4$ Hz, 1H), 6.86–6.85 (m, 3H), 3.88 (s, 3H), 2.77 (d, $J = 7.8$ Hz, 2H), 1.86–1.68 (m, 1H), 1.57–1.43 (m, 1H), 1.45–1.03 (m, 1H). ^{19}F NMR: δ -80.6 (d, $J = 75.2$ Hz, 2F), -127.6 (dt, $J = 156.4$ Hz, $J = 13.8$ Hz, 1F), -142.1 (ddd, $J = 156.8$ Hz, $J = 13.1$ Hz, $J = 3.2$ Hz, 1F). ^{13}C NMR: δ 151.1, 138.5, 122.4, 120.4, 116.2 (t, $J = 258.4$ Hz), 114.2 (t, $J = 283.3$ Hz), 112.6, 55.9, 32.6 (d, $J = 2.3$ Hz), 23.2 (t, $J = 10.9$ Hz), 16.3 (t, $J = 11.0$ Hz). MS (EI, m/z , %): 264 (M^+ , 53.95), 137 (100.00). HRMS (EI): calcd. for $C_{12}H_{12}O_2F_4$: 264.0773; Found: 264.0769.

1-(3,3-Difluoro-2-phenylcycloprop-1-enyl)-4-(2,2-difluorocyclopropyl)benzene (9). White solid. m.p. 98–99 °C. IR (film): 1784, 1611, 1466, 1268, 1237, 1192, 1043, 995, 960, 845, 831, 769, 686, 605, 544, 492 cm⁻¹. ¹H NMR: δ 7.81–7.70 (m, 4H), 7.56–7.48 (m, 3H), 7.38 (d, *J* = 8.1 Hz, 2H), 2.88–2.73 (m, 1H), 1.98–1.83 (m, 1H), 1.76–1.63 (m, 1H). ¹⁹F NMR: δ -112.7 (s, 2F), δ -126.1 (dtd, *J* = 154.4 Hz, *J* = 12.8 Hz, *J* = 3.8 Hz, 1F), δ -142.6 (ddd, *J* = 154.4 Hz, *J* = 13.6 Hz, *J* = 4.5 Hz, 1F). ¹³C NMR: δ 155.9, 147.5, 147.2, 139.3, 132.9, 131.6, 131.5, 129.4, 128.9, 123.6, 122.4, 112.1 (t, *J* = 287.4 Hz), 127.3 (t, *J* = 11.0 Hz), 17.6 (t, *J* = 10.5 Hz). MS (EI, *m/z*, %): 304 (M⁺, 46.56), 151 (100.00). HRMS (EI): calcd. for C₁₈H₁₂F₄: 304.0875; Found: 304.0872.

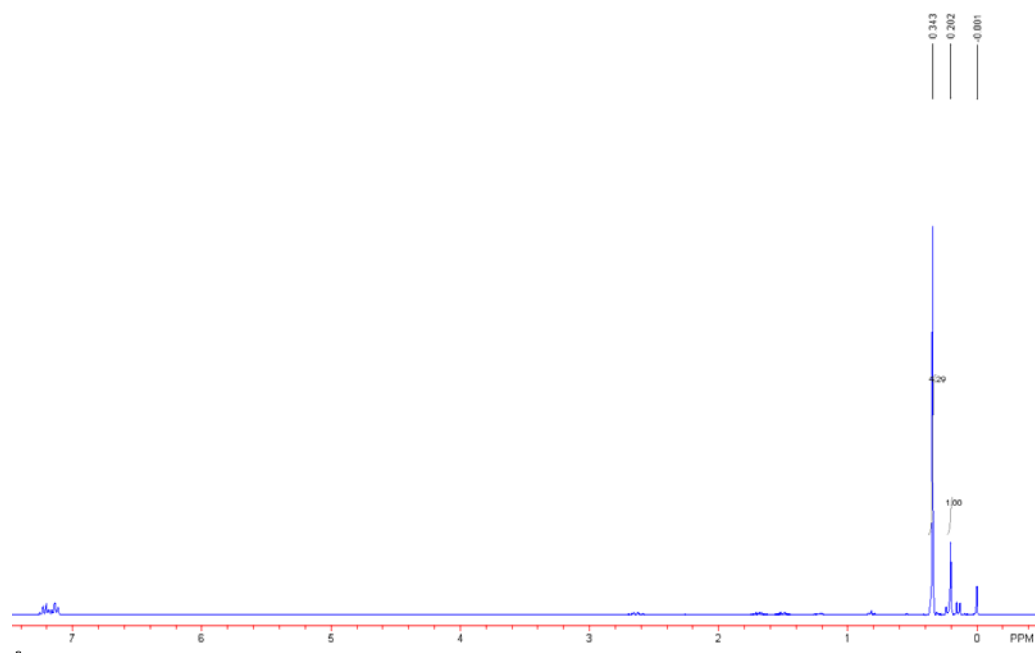
Typical procedure for difluoromethylation of phenols using compound 1:

A pressure tube was charged with 4-allyl-2-methoxyphenol (**6h**, 0.164 g, 1 mmol), aqueous KOH (25 wt.%, 0.8 mL) and toluene (2 mL). To this solution TMSCF₂Cl (0.397 g, 2.5 mmol) was added at -78 °C, then the tube was sealed. The mixture was heated to 50 °C for 4h. Then the mixture was extracted with Et₂O (20 mL x 2), and the combined organic phase was separated and dried over MgSO₄. After the removal of solvents under vacuum, the crude product was further purified by silica gel column chromatography to give product 4-allyl-1-(difluoromethoxy)-2-methoxybenzene (**10**) as a colorless liquid (175 mg, yield: 82%). IR (film): 2980 1641 1599 1511 1467 1421 1382 1274 1216 1123 919 cm⁻¹. ¹H NMR: δ 7.07 (d, *J* = 8.2 Hz, 1H), 6.80–6.25 (m, 3H), 6.02–5.87 (m, 1H), 5.14–5.10 (m, 1H), 5.09–5.06 (m, 1H), 3.86 (s, 3H), 3.36 (d, *J* = 6.9 Hz, 2H). ¹⁹F NMR: δ -80.6 (d, *J* = 56.8 Hz, 2F). ¹³C NMR: δ 150.9, 138.9, 136.9, 122.2, 120.2, 116.3 (t, *J* = 258.3 Hz), 112.9, 55.9, 39.9. MS (EI, *m/z*, %): 214 (M⁺, 100.00), 91 (55.30), 147 (53.70). HRMS (EI): calcd. for C₁₁H₁₂O₂F₂: 214.0805 ; Found: 214.0807.

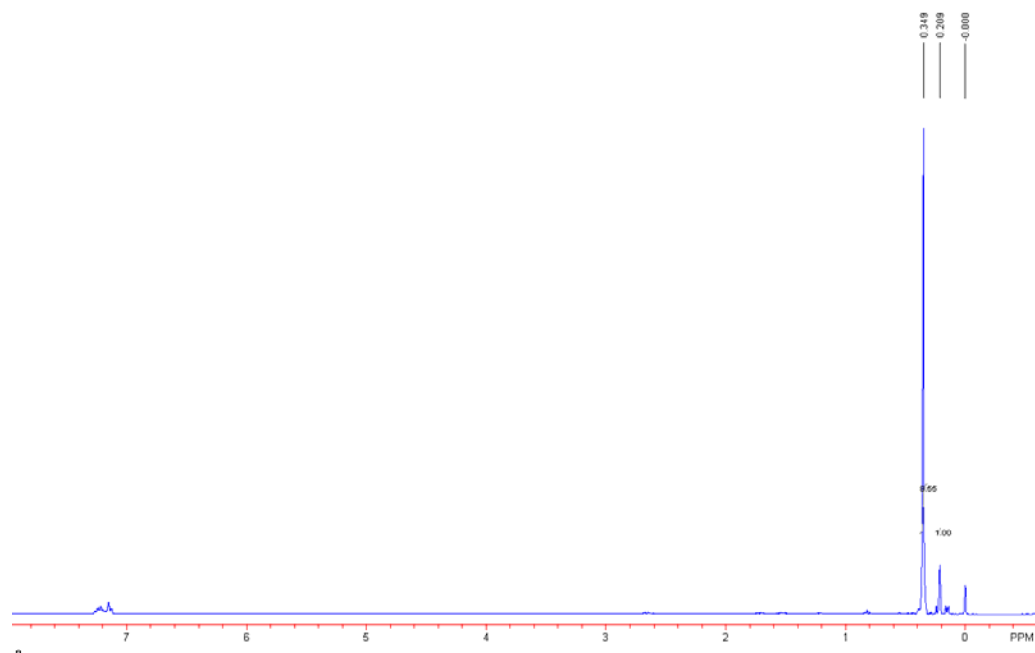
Determination of the formation of TMSCl in the [2 + 1] cycloaddition between TMSCF₂Cl (**1**) and alkene (**6b**):

Styrene **6b** (0.104 g, 1.0 mmol), TMSCF₂Cl (0.476, 3.0 mmol), and *n*Bu₄NCl (0.022 g, 0.06 mmol) (without solvent) were mixed in a pressure tube at room temperature. Then reaction mixture was heated at 110 °C for 4 h. The reaction mixture was characterized by ¹H NMR spectroscopy. (After the authentic TMSCl was added to the mixture, the peak of TMSCl was identical with that observed in the reaction mixture).

A: ^1H NMR (from the reaction mixture):



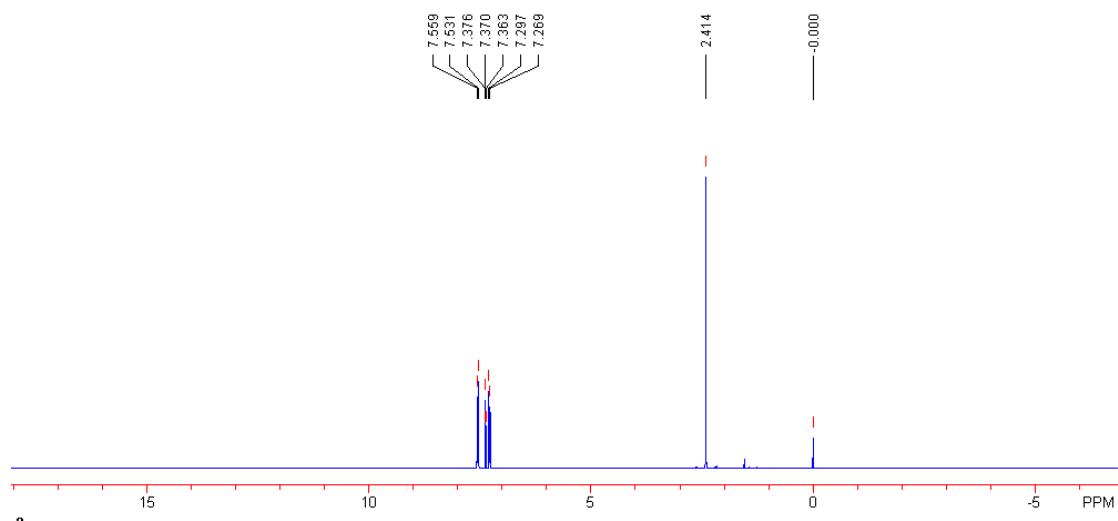
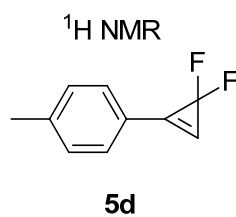
B: ^1H NMR (the authentic TMSCl was added to the reaction mixture):



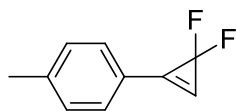
References:

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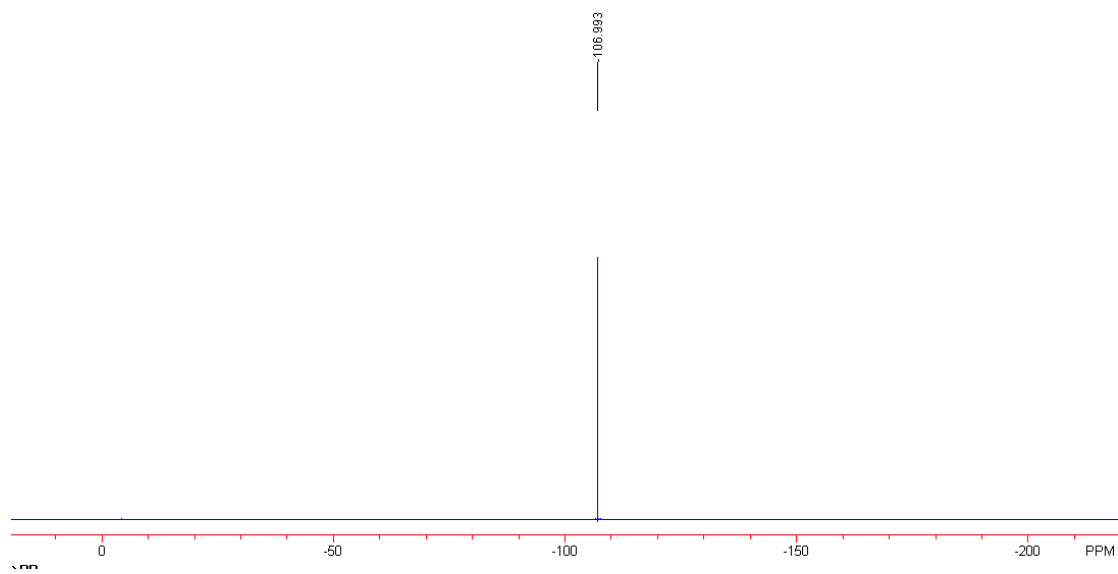
^1H , ^{19}F , and ^{13}C NMR spectra of all new products:



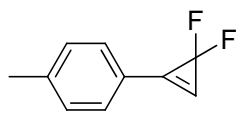
^{19}F NMR



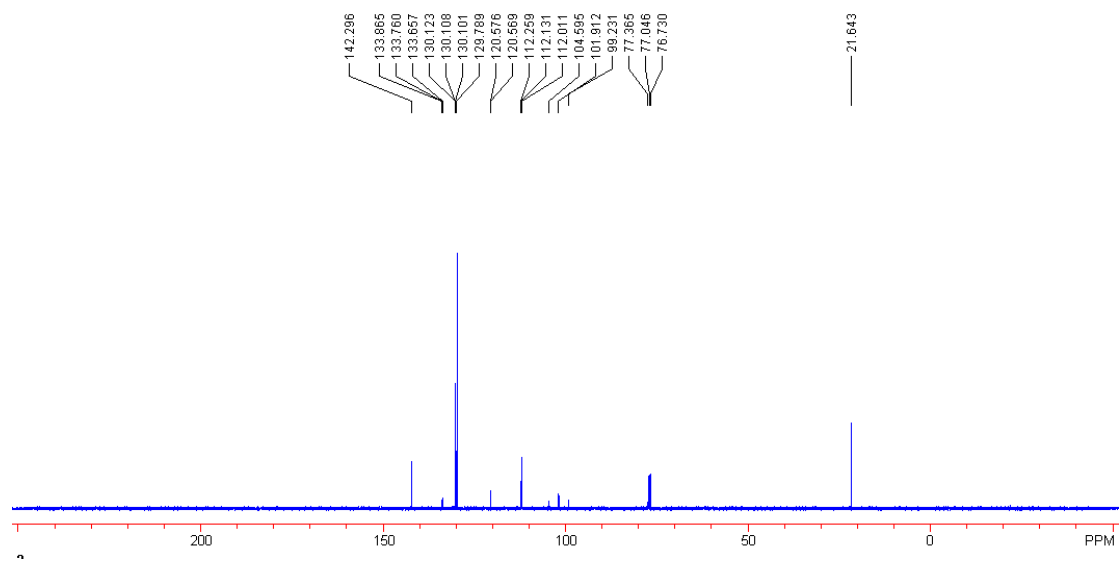
5d

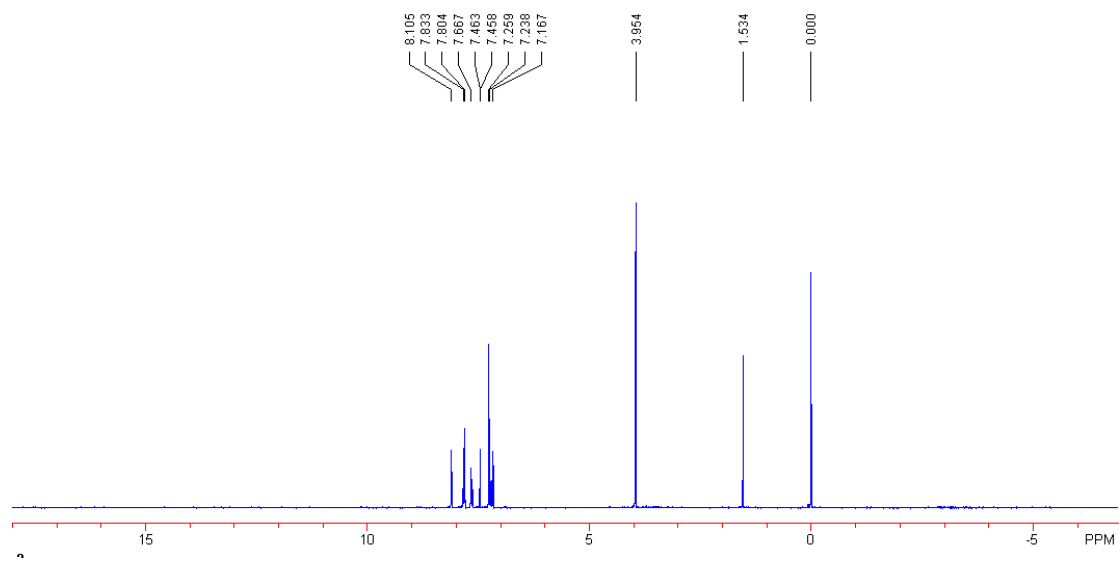
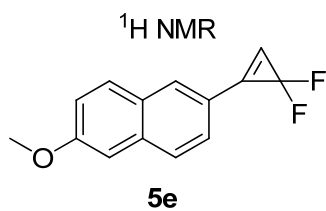


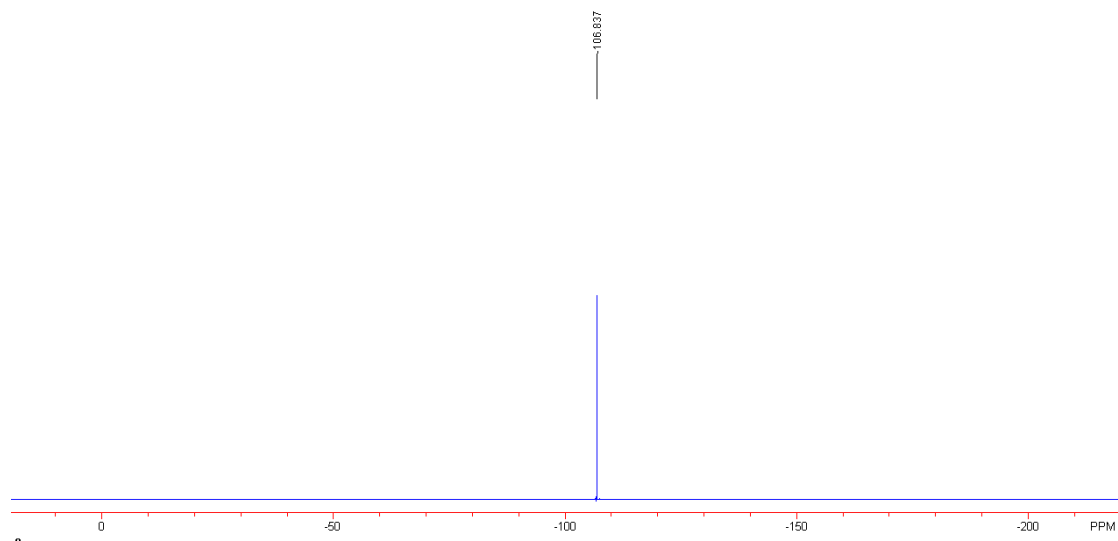
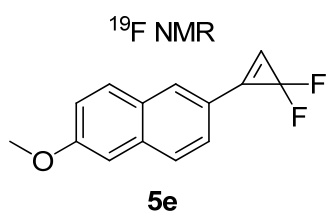
^{13}C NMR

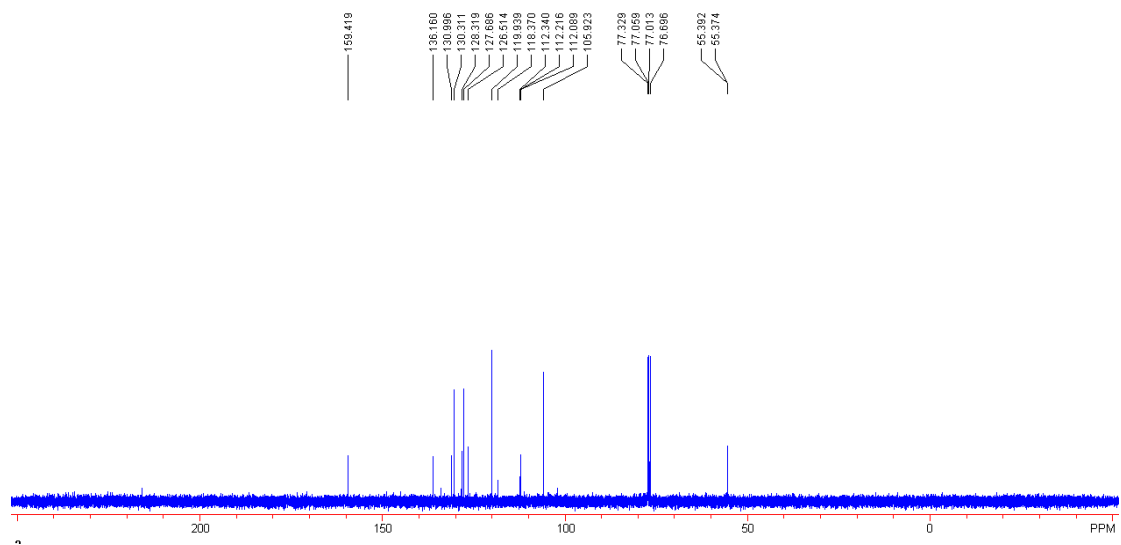
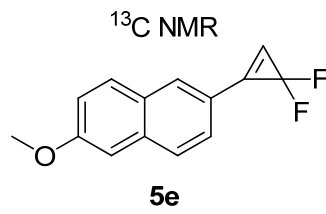


5d

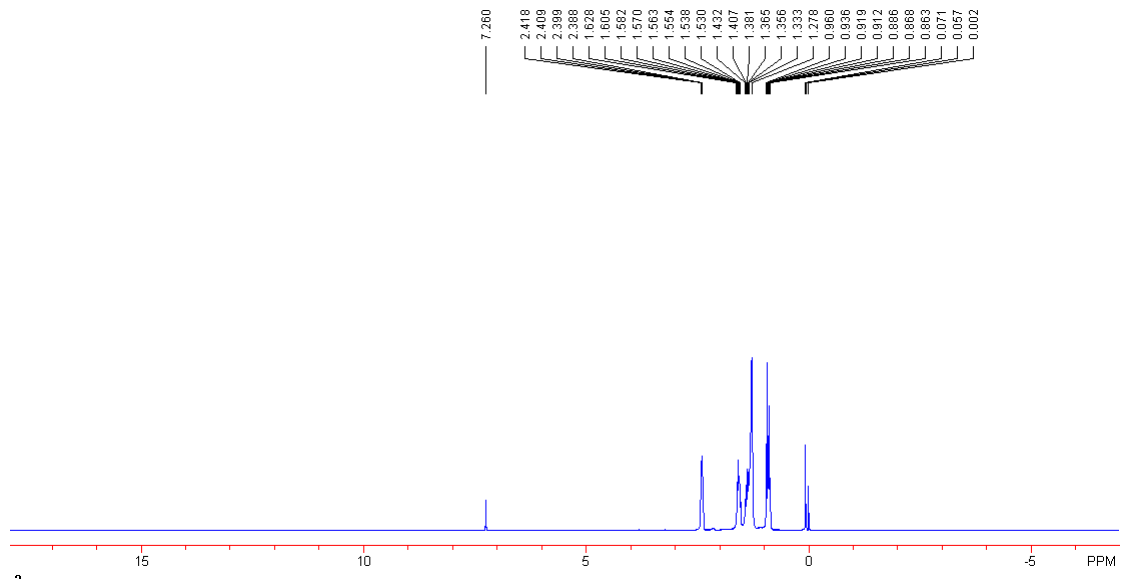
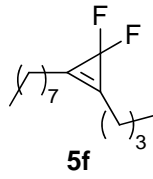




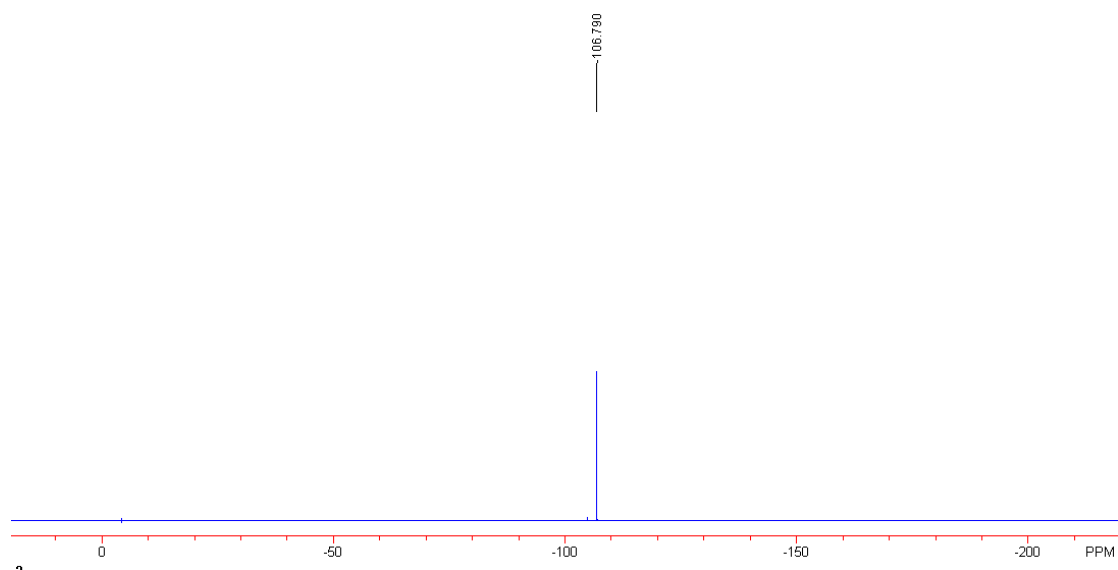
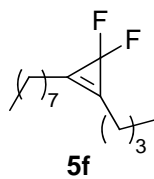




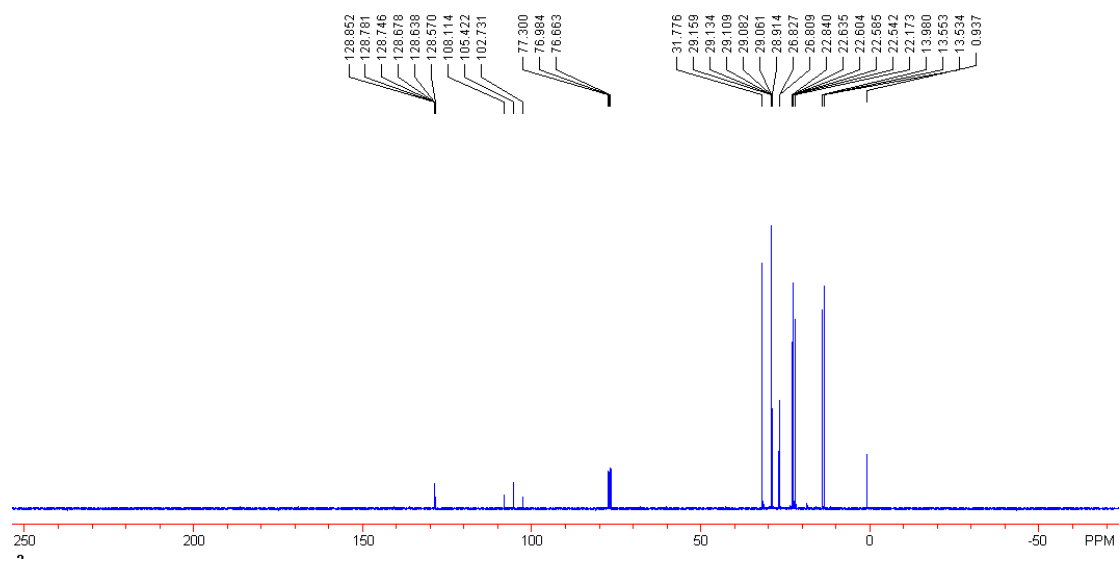
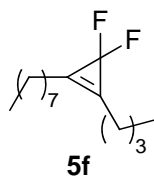
^1H NMR



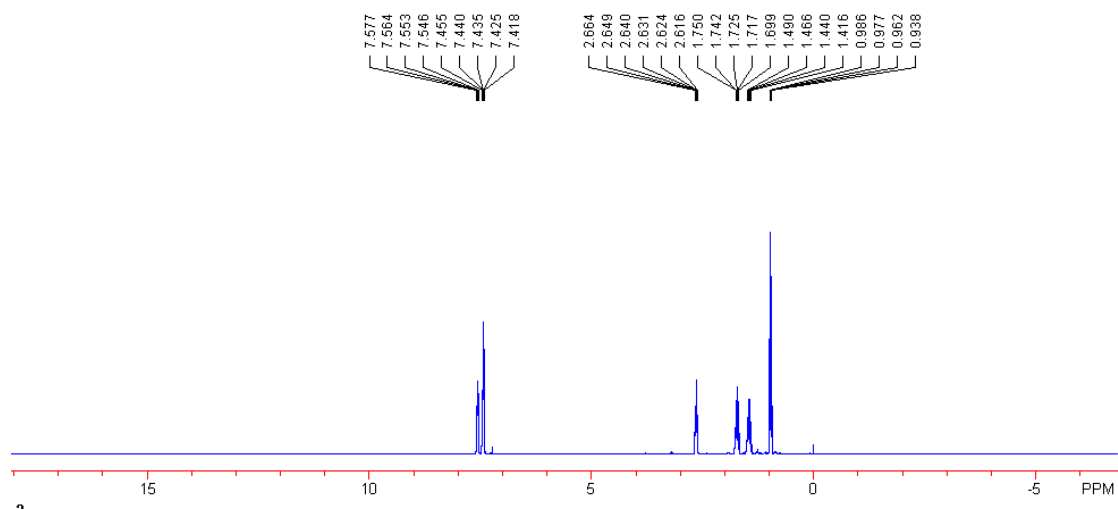
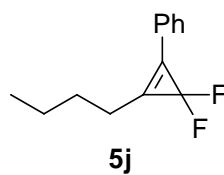
^{19}F NMR



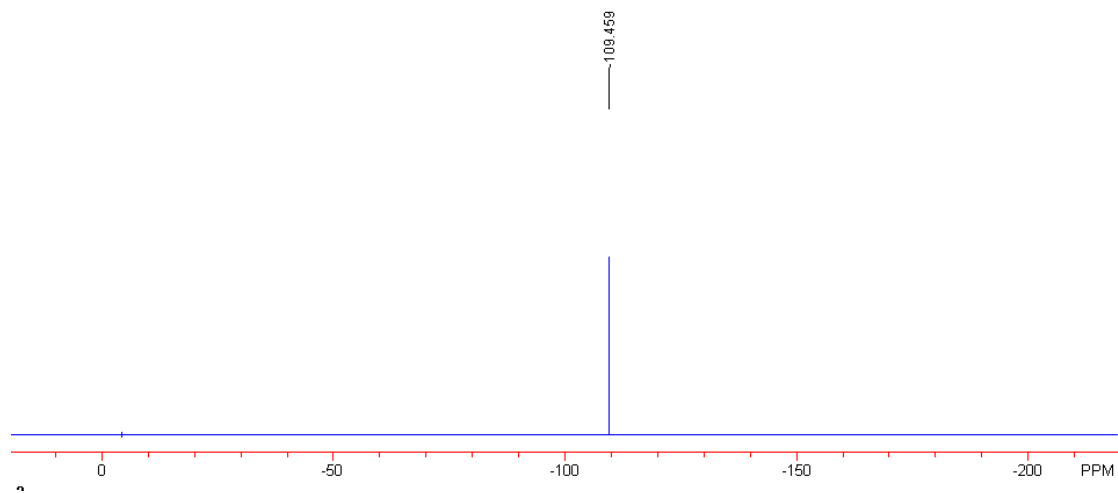
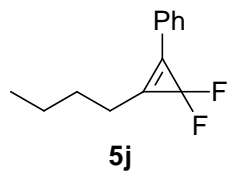
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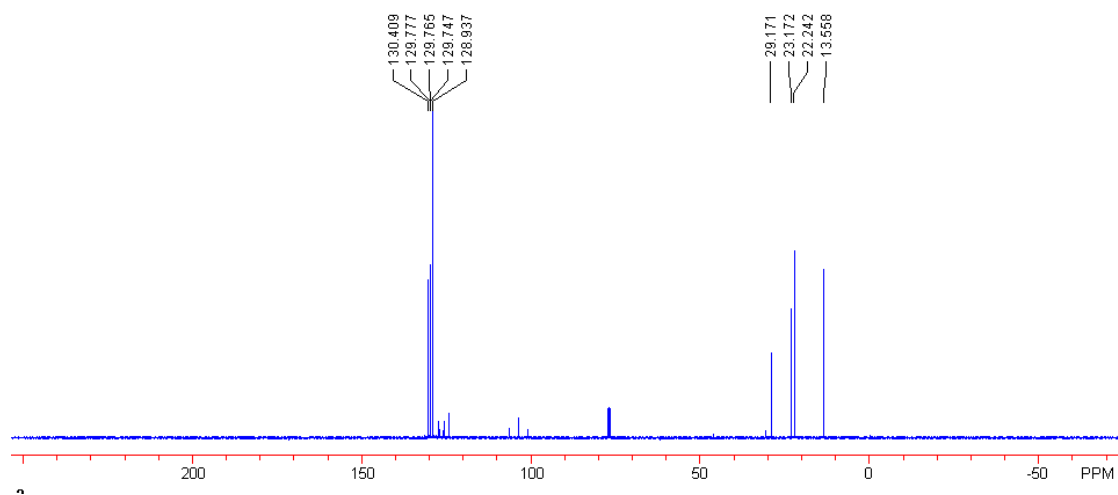
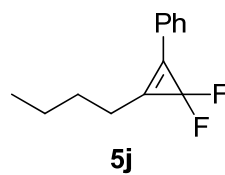
^1H NMR



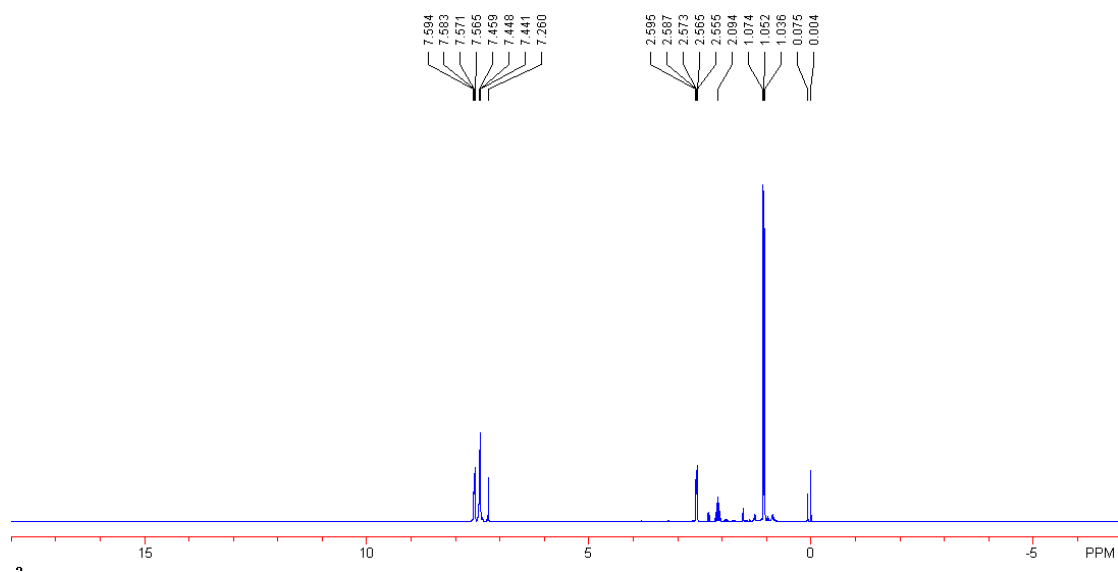
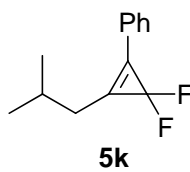
^{19}F NMR



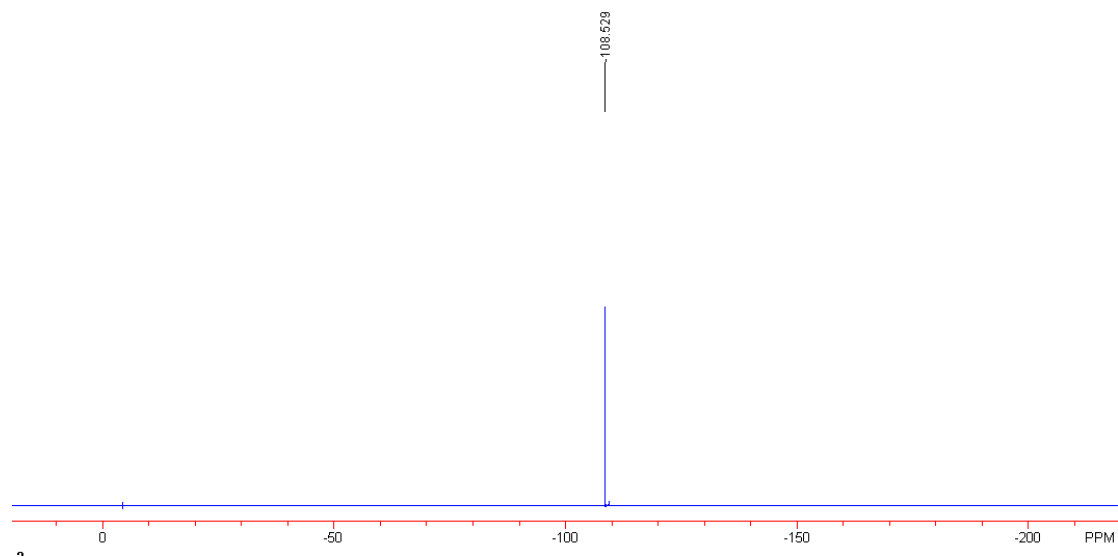
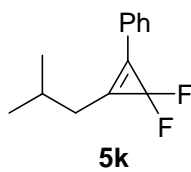
^{13}C NMR



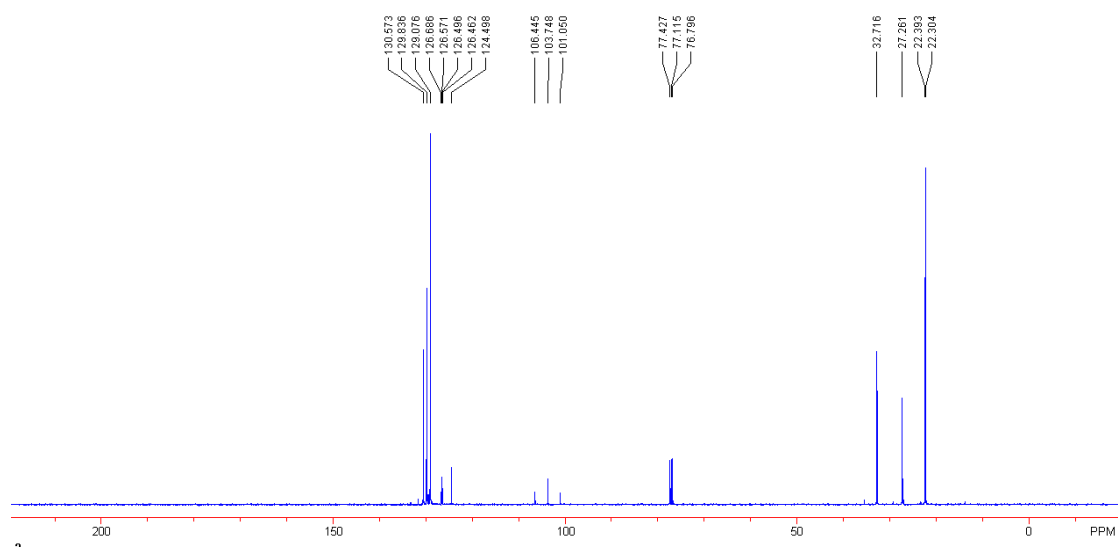
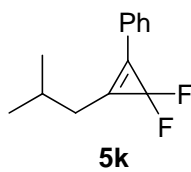
^1H NMR



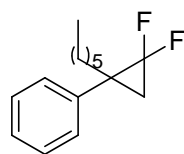
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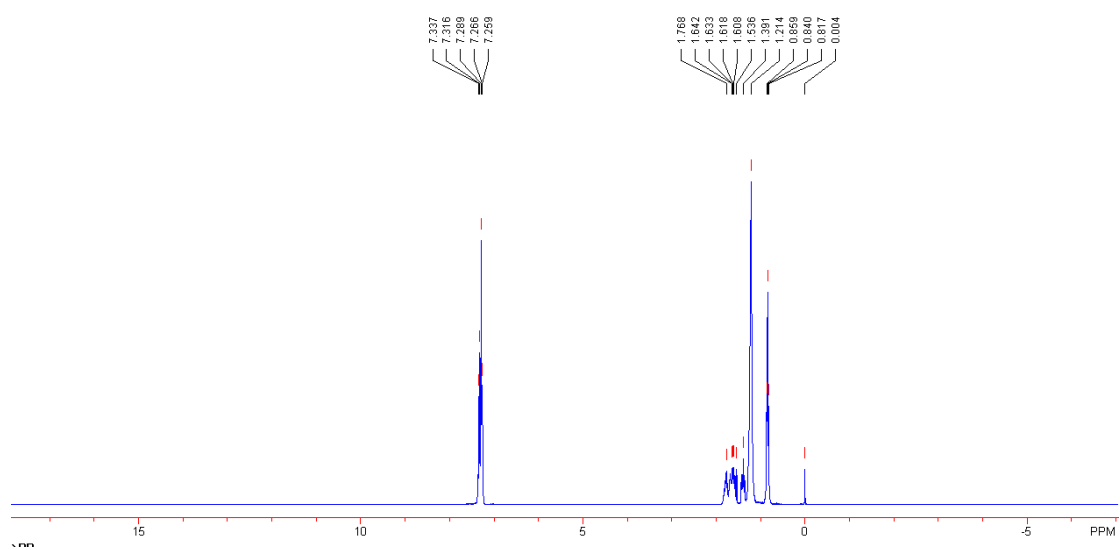
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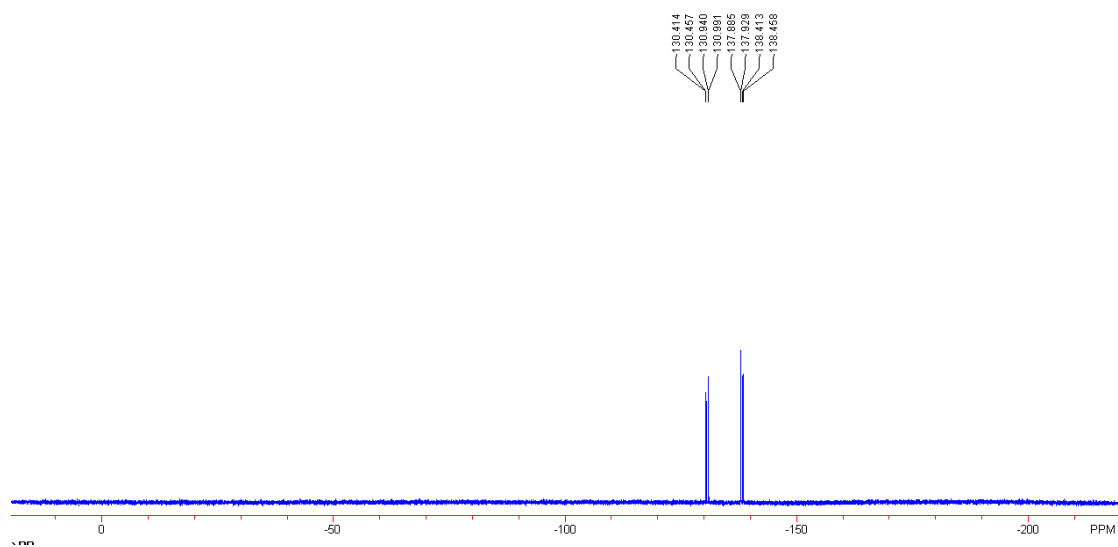
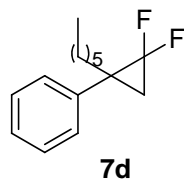
^1H NMR



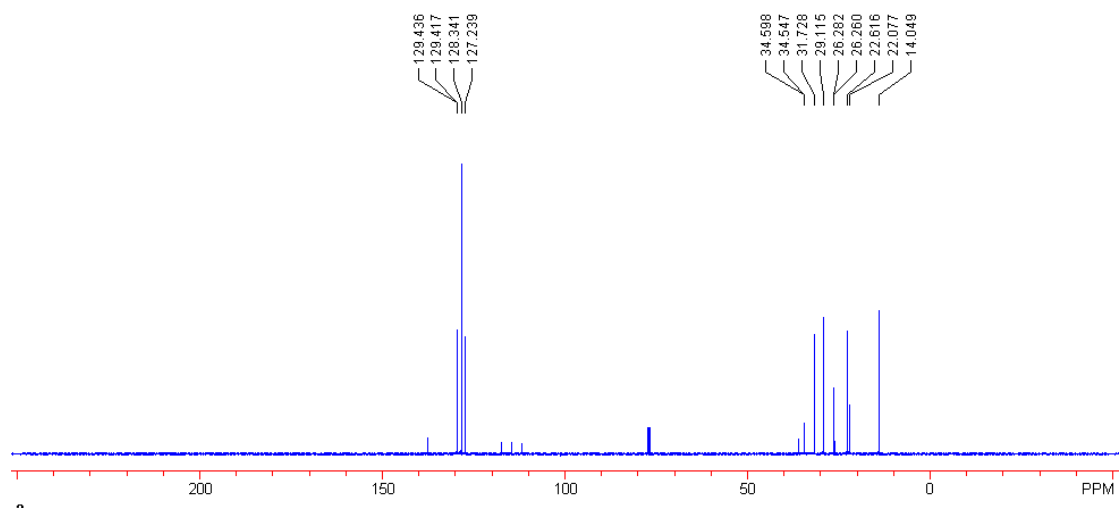
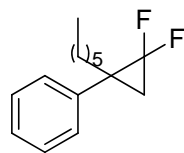
7d



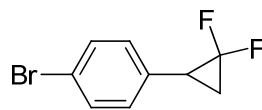
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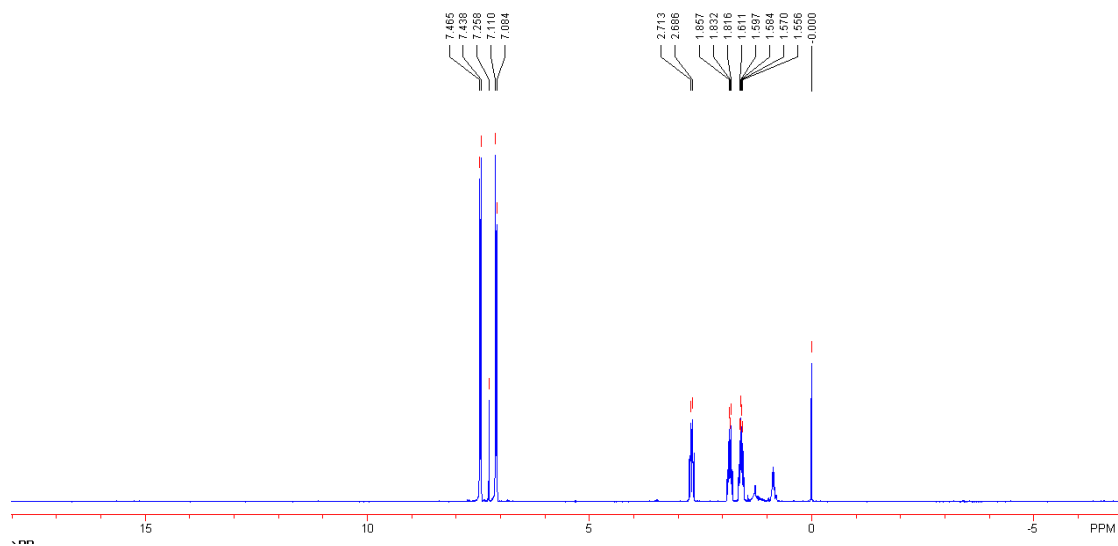
^{13}C NMR



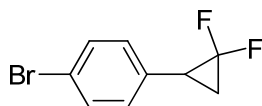
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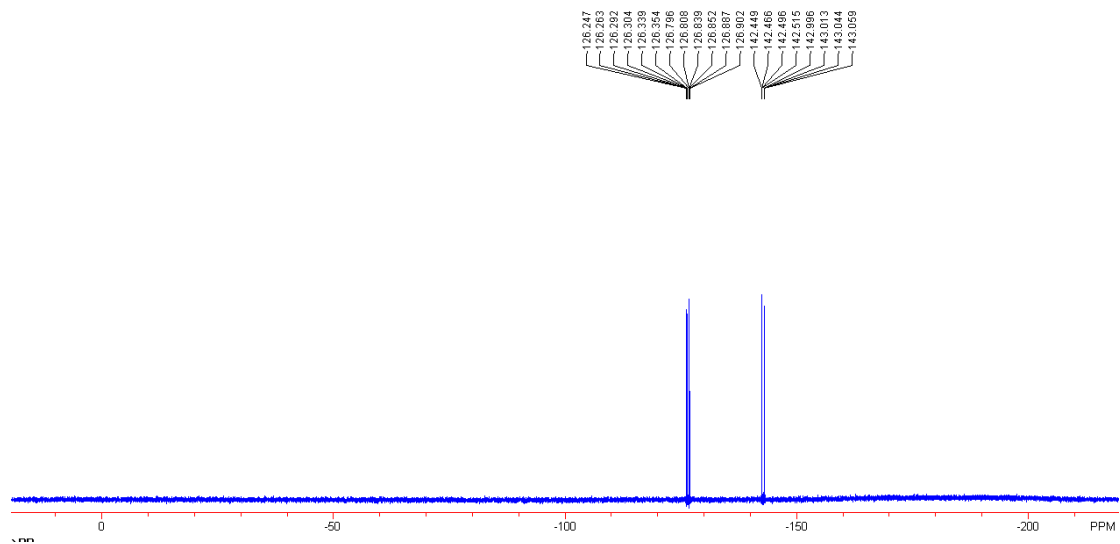
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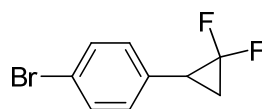
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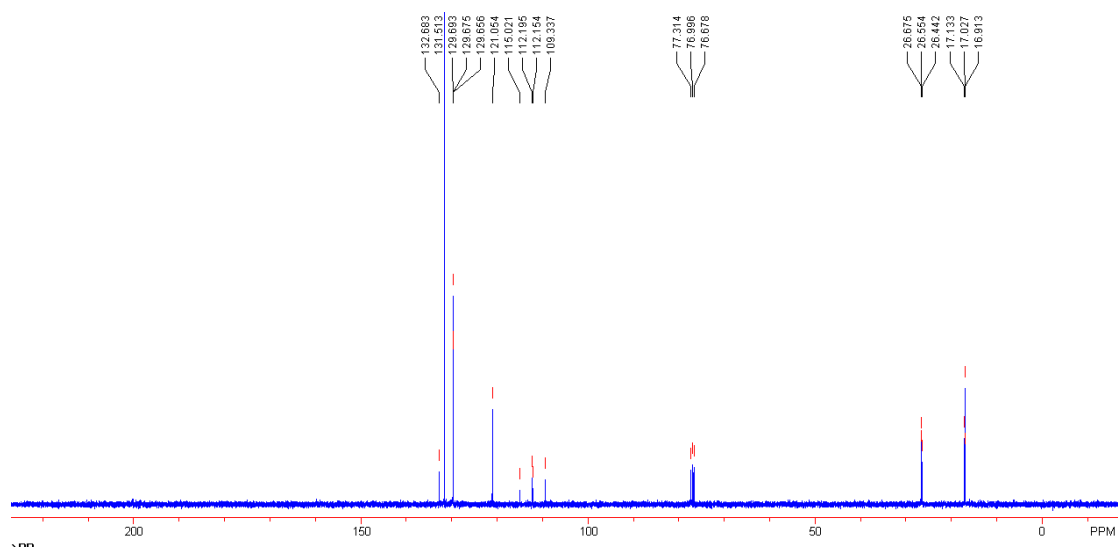
7f



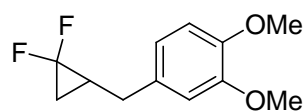
^{13}C NMR



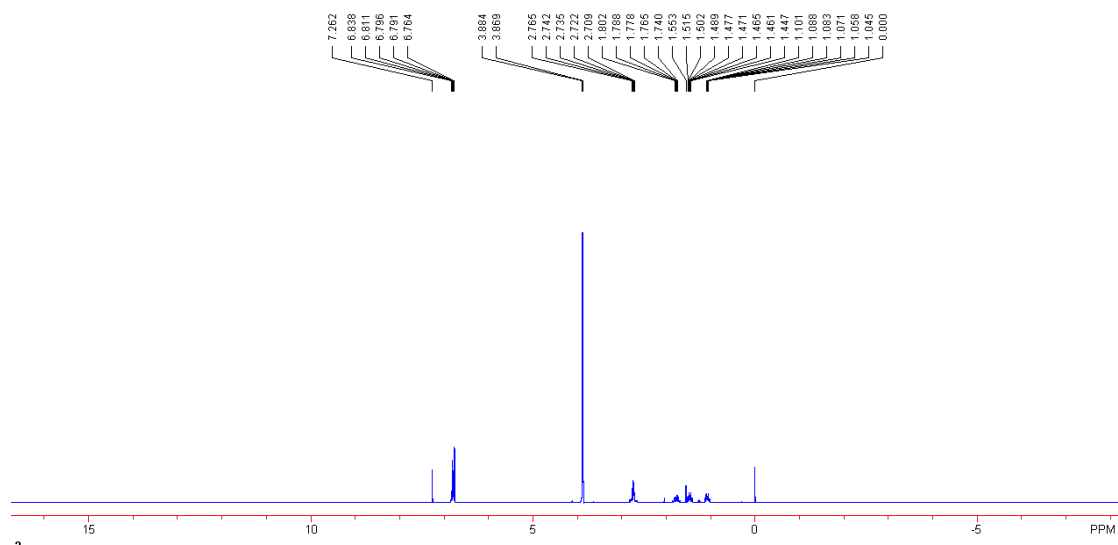
7f



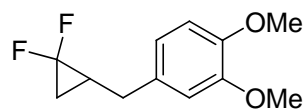
$^1\text{H NMR}$



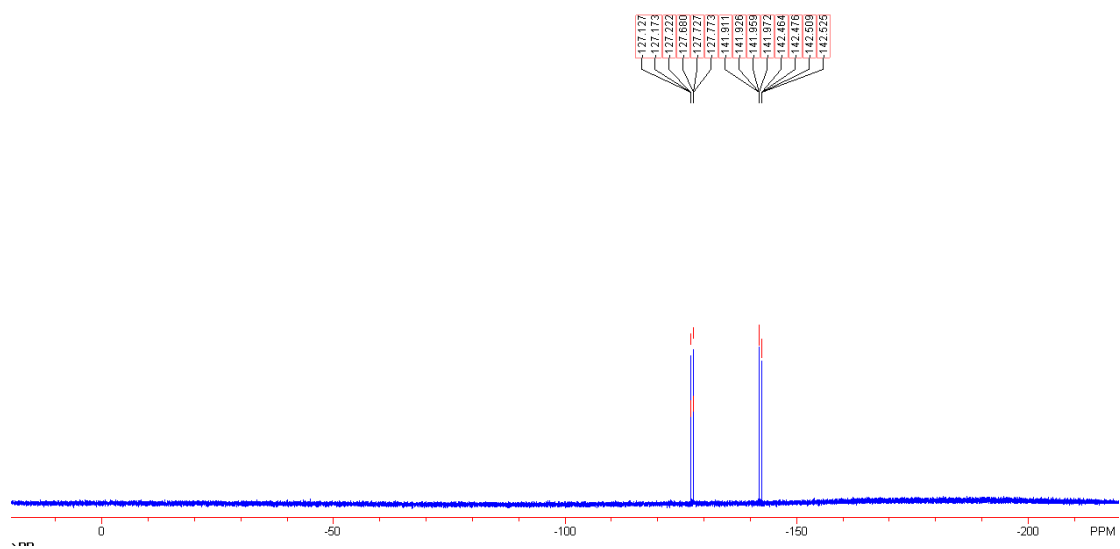
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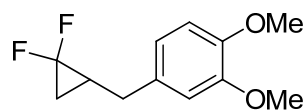
^{19}F NMR



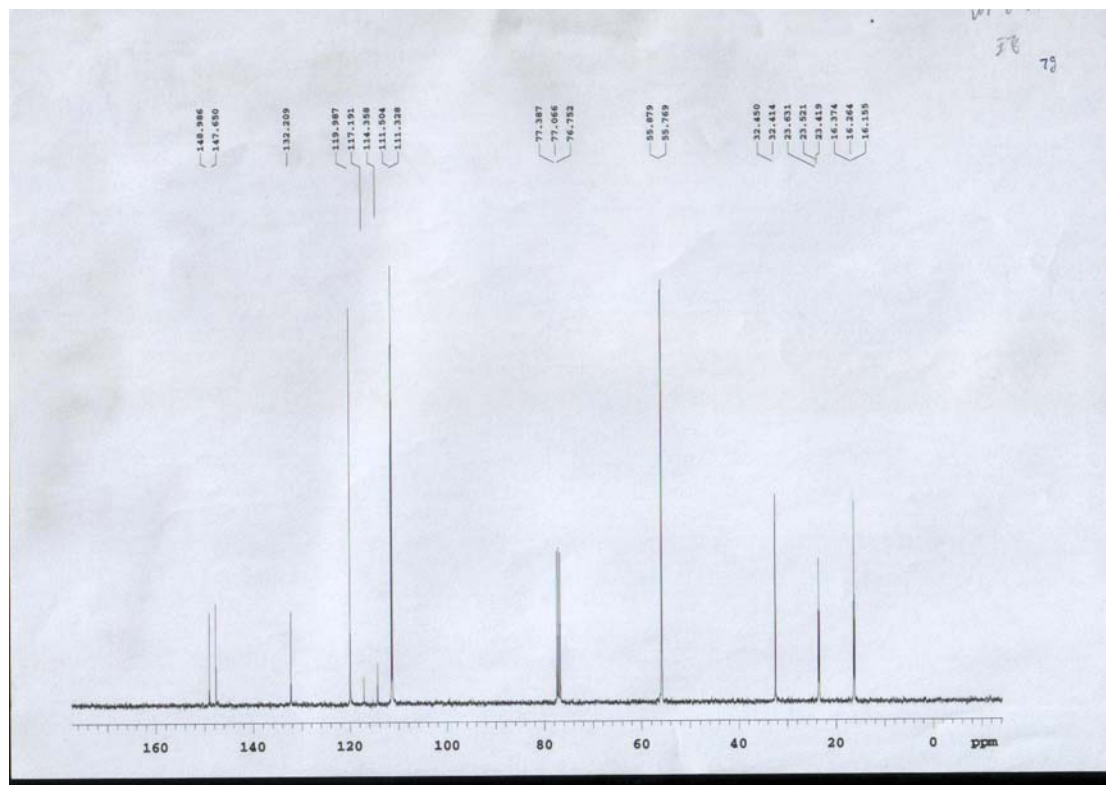
7g



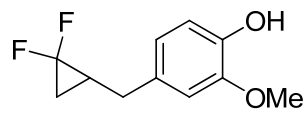
^{13}C NMR



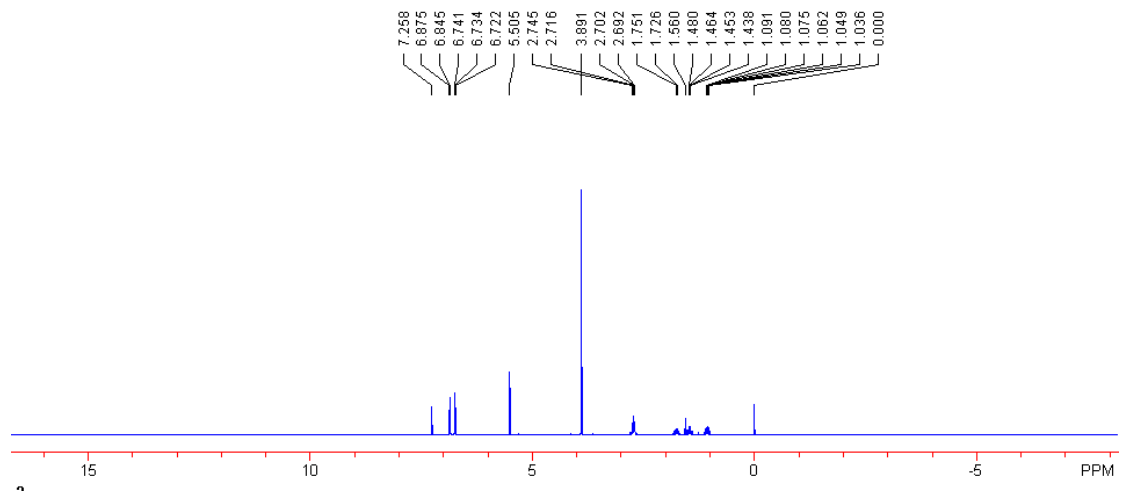
7g



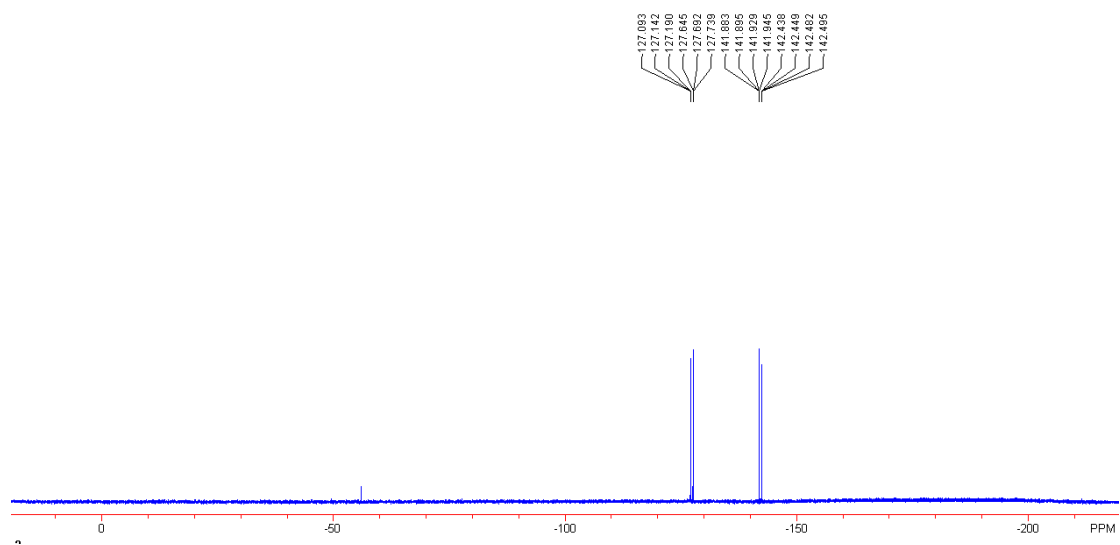
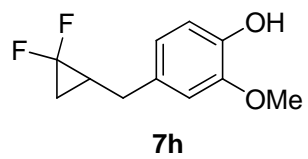
^1H NMR



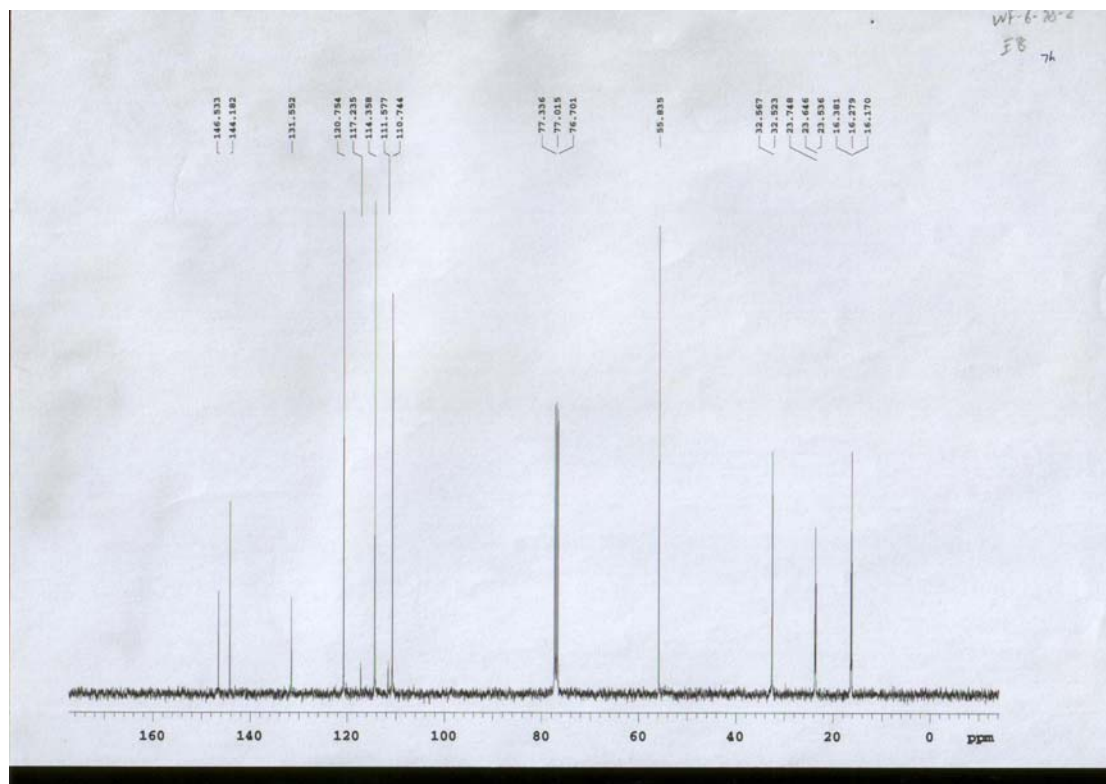
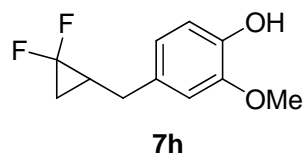
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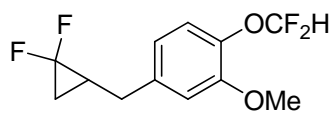
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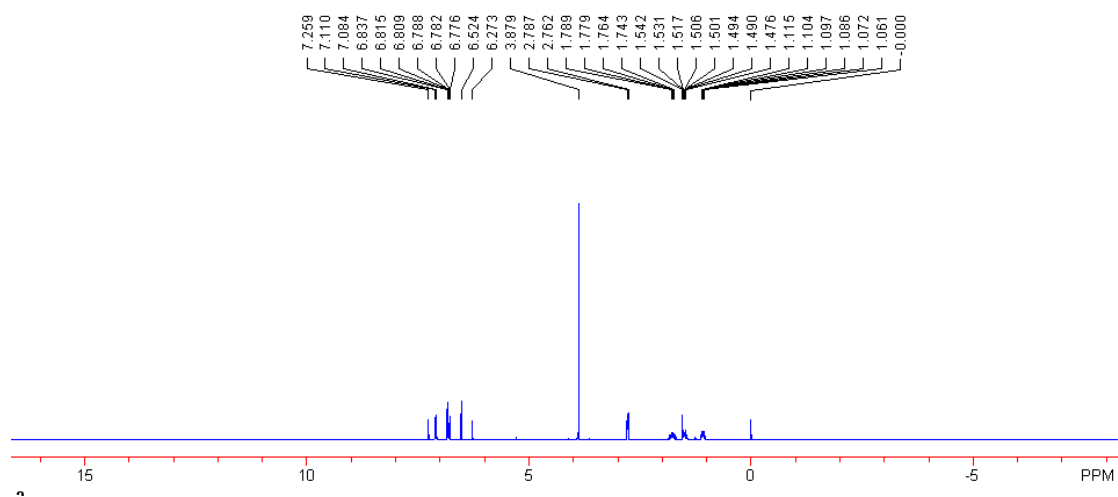
^{13}C NMR



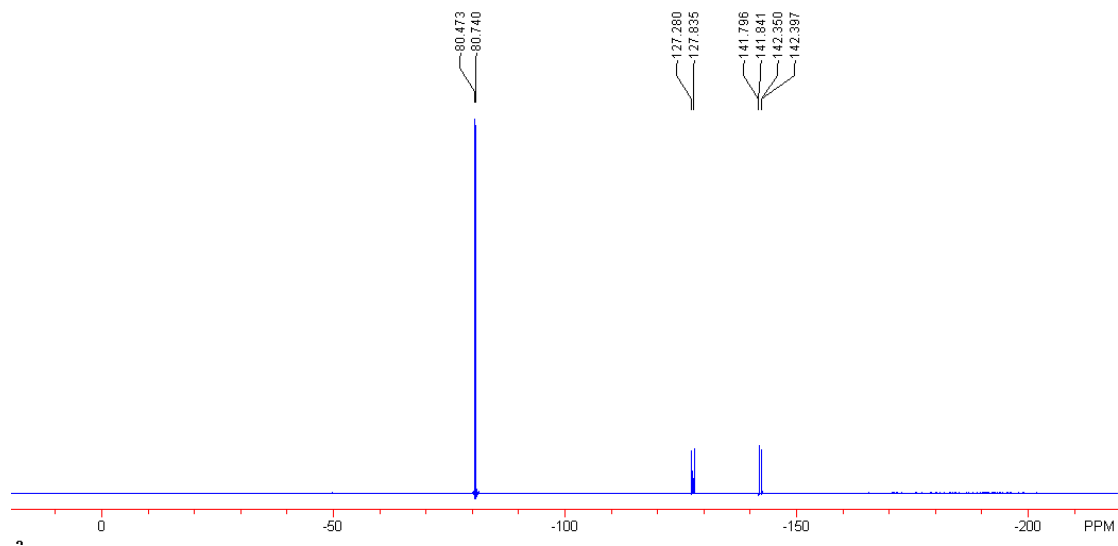
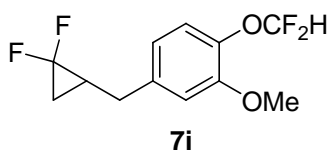
^1H NMR



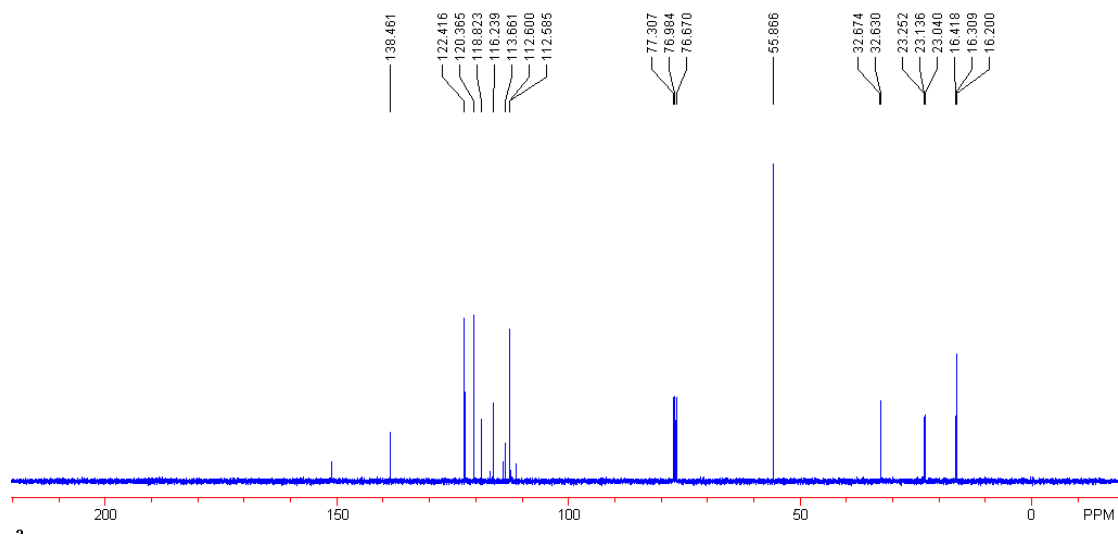
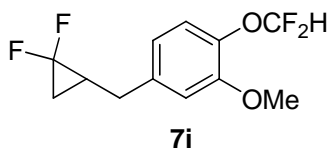
7i



^{19}F NMR



^{13}C NMR



Summary of Computational results:

Density functional theory (DFT) calculations were conducted at the B3LYP/6-311+G** level of theory using Gaussian 09.^{c1-c3} Tetramethylethylene (**6a**) was chosen as the model substrate. The self-consistent reaction field (SCRF) method, polarizable continuum model (PCM), using the integral equation formalism variant was applied to include the solvent effects in THF and toluene.^{c4}

- C1. Gaussian 09, Revision A.02, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2009.
- C2. (a) A. D. Becke, *J. Chem. Phys.*, 1993, **98**, 5648. (b) C. Lee, W. Yang, and R. G. Parr, *Phys. Rev. B.*, 1988, **37**, 785.
- C3. (a) A. D. McLean and G. S. Chandler, *J. Chem. Phys.*, 1980, **72**, 5639. (b) K. Raghavachari, J. S. Binkley, R. Seeger, and J. A. Pople, *J. Chem. Phys.*, 1980, **72**, 650.
- C4. (a) S. Miertuš, E. Scrocco, J. Tomasi, *Chem. Phys.*, 1981, **55**, 117. (b) S. Miertuš, J. Tomasi, *Chem. Phys.*, 1982, **65**, 239.

I. Reactions in THF:

Me₃SiCF₂Cl (**1**) in THF

Zero-point correction= 0.122363 (Hartree/Particle)
Sum of electronic and zero-point Energies= -1107.309943
Sum of electronic and thermal Enthalpies= -1107.297143
Sum of electronic and thermal Free Energies= -1107.347711

H	0.48086500	-2.31773300	-1.22972800
C	0.48497700	-2.29285000	-0.13727300
H	1.52378700	-2.27161900	0.20242900
H	0.03414100	-3.22307200	0.22193700
Si	-0.47886500	-0.82797500	0.52580500
C	-0.42312200	-0.71884800	2.39874100
H	0.60508800	-0.70679300	2.76951700
H	-0.92706500	-1.58697600	2.83461900
H	-0.92938200	0.17978000	2.76061100
C	-2.22771900	-0.73411200	-0.14364600
H	-2.24309900	-0.76079100	-1.23583500
H	-2.73387000	0.17722400	0.18556600
H	-2.80635600	-1.58814800	0.22183000
C	0.45488400	0.77949900	-0.09740100
F	1.75616400	0.81135300	0.32656100
F	-0.11477700	1.92489900	0.38888600
Cl	0.49398300	0.94587100	-1.91046000

Cl⁻ in THF

Zero-point correction= 0.000000 (Hartree/Particle)
Sum of electronic and zero-point Energies= -460.400964
Sum of electronic and thermal Enthalpies= -460.398603
Sum of electronic and thermal Free Energies= -460.415986

Me₃SiCF₂Cl^{•-} radical anion in THF

Zero-point correction= 0.120071 (Hartree/Particle)
Sum of electronic and zero-point Energies= -1107.333686
Sum of electronic and thermal Enthalpies= -1107.320707
Sum of electronic and thermal Free Energies= -1107.372317

H	0.47462600	-2.31576100	-1.24358900
C	0.49858800	-2.29316900	-0.14721600
H	1.54367900	-2.24803900	0.18553000
H	0.06453700	-3.22866500	0.22230100
Si	-0.49371300	-0.85351100	0.53699400
C	-0.42005200	-0.71730500	2.40984100
H	0.61526500	-0.71572200	2.77376400
H	-0.93204300	-1.58583000	2.84227200
H	-0.92523200	0.18928300	2.76675500
C	-2.23739700	-0.72211300	-0.14799700
H	-2.24570300	-0.76085800	-1.24399600
H	-2.72734300	0.20401900	0.17977600
H	-2.82561200	-1.56742300	0.22613300
C	0.44361300	0.76150900	-0.08957300
F	1.74304100	0.80841200	0.34451200
F	-0.13443500	1.91348800	0.37606900
Cl	0.50781500	0.92139200	-1.90941600

Cl[•] radical in THF

Zero-point correction= 0.000000 (Hartree/Particle)
Sum of electronic and zero-point Energies= -460.167804
Sum of electronic and thermal Enthalpies= -460.165444
Sum of electronic and thermal Free Energies= -460.183481

Cl-Me₃Si-CF₂Cl transition state (10) in THF

Imaginary Frequency: -94.7684 cm⁻¹

Zero-point correction= 0.120259 (Hartree/Particle)
Sum of electronic and zero-point Energies= -1567.676866
Sum of electronic and thermal Enthalpies= -1567.661365
Sum of electronic and thermal Free Energies= -1567.721954

Si	1.15911400	-0.02329500	-0.00124100
C	0.72768600	-0.88891200	1.60837400
H	1.20464800	-0.37396900	2.44743900
H	1.10875100	-1.91381300	1.58665400
H	-0.34802100	-0.92005700	1.77639000
C	1.23681200	1.85797500	-0.00414400

H	1.77548400	2.20143400	0.88381100
H	0.25358900	2.32634400	-0.01536500
H	1.79387900	2.19754600	-0.88214400
C	0.72726300	-0.89214500	-1.60894300
H	1.20373500	-0.37883900	-2.44928800
H	-0.34857500	-0.92353900	-1.77622200
H	1.10824400	-1.91703300	-1.58545000
C	-1.69237200	0.56918400	0.00072000
F	-2.22052500	1.25228300	1.07027400
F	-2.22429100	1.25423300	-1.06578100
Cl	-2.97363200	-1.17154400	0.00130300
Cl	3.36390400	-0.40205100	-0.00024100

CF₂Cl (11) in THF

Zero-point correction=	0.007271 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-698.172663
Sum of electronic and thermal Enthalpies=	-698.166895
Sum of electronic and thermal Free Energies=	-698.201733

C	0.33558700	-0.58246500	-0.66386300
F	1.47638900	-0.46006800	0.03060100
F	-0.33900600	-1.51141400	0.02972500
Cl	-0.84662900	1.46921600	0.30990300

Me₃SiCl in THF

Zero-point correction=	0.111924(Hartree/Particle)
Sum of electronic and zero-point Energies=	-869.509012
Sum of electronic and thermal Enthalpies=	-869.499433
Sum of electronic and thermal Free Energies=	-869.541605

Si	-0.34476300	-0.00000500	-0.12221600
C	-0.24226500	0.00004900	-1.99141500
H	0.28137500	-0.88621900	-2.35874000
H	0.28138400	0.88633200	-2.35869200
H	-1.25039600	0.00006600	-2.41861500
C	-1.14073000	1.55573500	0.54946500
H	-0.62155200	2.45024300	0.19582800
H	-1.13367600	1.56318800	1.64244600
H	-2.18229200	1.61410500	0.21660300
C	-1.14074100	-1.55577800	0.54937400
H	-0.62157000	-2.45027000	0.19568400
H	-2.18230300	-1.61412000	0.21651000
H	-1.13368500	-1.56329400	1.64234800
Cl	1.66990300	-0.00003300	0.59017900

CF₂ in THF

Zero-point correction=	0.006636 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-237.769038
Sum of electronic and thermal Enthalpies=	-237.765086

Sum of electronic and thermal Free Energies= -237.793100

C	0.00000000	-0.02370300	-0.59687900
F	0.00000000	1.04201600	0.15867400
F	0.00000000	-1.02631300	0.24033000

Tetramethylethylene (6a) in THF

Zero-point correction= 0.162432(Hartree/Particle)

Sum of electronic and zero-point Energies= -235.762198

Sum of electronic and thermal Enthalpies= -235.752470

Sum of electronic and thermal Free Energies= -235.795929

C	-0.03319300	-0.01783900	0.67525000
C	0.02888600	0.01839200	-0.67053300
C	-0.10772100	1.20794000	1.55965600
H	-0.10911300	2.15220900	1.01921700
H	0.73823000	1.22401400	2.25750200
H	-1.01353300	1.17561900	2.17726400
C	-0.03717900	-1.29018600	1.49439400
H	-0.94565500	-1.34505500	2.10644100
H	0.80576600	-1.29242200	2.19601600
H	0.02032400	-2.20244400	0.90456900
C	0.10329700	-1.20738800	-1.55493900
H	-0.74219400	-1.22289000	-2.25335400
H	0.10368800	-2.15168100	-1.01454100
H	1.00954000	-1.17562000	-2.17194500
C	0.03299100	1.29074500	-1.48966500
H	-0.02363000	2.20302500	-0.89978900
H	-0.81037700	1.29344500	-2.19078100
H	0.94112400	1.34513400	-2.10226100

Tetramethylethylene + CF₂ Transition State in THF

Imaginary Frequency: -239.4254 cm⁻¹

Zero-point correction= 0.171842 (Hartree/Particle)

Sum of electronic and zero-point Energies= -473.515205

Sum of electronic and thermal Enthalpies= -473.503031

Sum of electronic and thermal Free Energies= -473.551513

C	-1.52510000	-1.27378400	0.12170400
C	0.56601600	0.12879300	0.02486700
C	-0.75073800	0.53300400	-0.16634000
F	-1.13411500	-1.75766200	1.32882100
F	-0.97966100	-2.14921100	-0.76224900
C	-1.27441000	0.82276000	-1.56411900
H	-0.94749300	1.81742200	-1.88511300
H	-2.36527900	0.80966600	-1.55779900
H	-0.93777900	0.09693400	-2.30247800
C	-1.48932900	1.31164400	0.91085000
H	-1.32375800	0.91494300	1.91118800
H	-2.56206200	1.28140900	0.71474600

H	-1.16992100	2.35896500	0.89838100
C	1.17966500	0.02722700	1.38799800
H	2.25995300	0.17693100	1.33795700
H	1.01286700	-0.98669900	1.78544200
H	0.76449000	0.72906900	2.10907100
C	1.38830400	-0.44973900	-1.08663100
H	2.45419600	-0.34571100	-0.87754900
H	1.17909500	-0.00518900	-2.05878300
H	1.17728600	-1.52756500	-1.17085400

2,2-difluorotetramethylcyclopropane in THF

Zero-point correction=	0.176072 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-473.597274
Sum of electronic and thermal Enthalpies=	-473.585633
Sum of electronic and thermal Free Energies=	-473.632045

C	0.82630000	0.24794100	0.20519800
C	-0.57477000	-0.41038400	0.53510200
C	-0.26939800	0.17460200	-0.79641100
F	-0.14741900	-0.63586900	-1.89484900
F	-0.90359400	1.31807000	-1.20767900
C	-0.67008600	-1.92240000	0.66966600
H	-0.06620100	-2.45821600	-0.06061000
H	-1.70822100	-2.24086700	0.53702500
H	-0.35195900	-2.23113100	1.66981200
C	-1.54113600	0.32234200	1.45361900
H	-1.26753100	0.15451600	2.49945300
H	-2.55403100	-0.06446100	1.30794900
H	-1.57198800	1.39627900	1.27760200
C	1.17163900	1.59740100	0.81541400
H	1.97212700	2.07035100	0.23890100
H	1.53162800	1.46676100	1.84014800
H	0.32981200	2.28726300	0.83471700
C	2.04328500	-0.64735900	0.02965500
H	2.82111500	-0.10971700	-0.52050900
H	1.82861700	-1.56366100	-0.51758800
H	2.45397900	-0.91983600	1.00634400

Tetramethylethylene + CF₂Cl Transition State in THF

Imaginary Frequency: -309.1547 cm ⁻¹	
Zero-point correction=	0.171350 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-933.902494
Sum of electronic and thermal Enthalpies=	-933.888219
Sum of electronic and thermal Free Energies=	-933.943209

C	0.29037600	-1.03784300	-0.33439000
F	1.38210800	-1.02147900	-1.13865300
F	0.77300900	-1.46923000	0.85694500
Cl	-0.81528300	-3.07230600	-1.12800700
C	-0.40882700	0.75568700	-0.14835300

C	-0.98745400	0.93604500	-1.54923600
H	-0.22078400	0.92952100	-2.32381700
H	-1.53075300	1.88764700	-1.60963400
H	-1.67693900	0.11848500	-1.76811900
C	-1.51962600	0.48418000	0.86219200
H	-1.13802000	0.17814100	1.83612900
H	-2.15921600	-0.32111400	0.49551200
H	-2.12696100	1.38790300	0.99938600
C	0.71540900	1.50261700	0.23783100
C	1.74490300	1.96724500	-0.75854900
H	2.24407300	2.87805700	-0.40244300
H	1.32981600	2.18741600	-1.74344600
H	2.54007100	1.21242700	-0.90939000
C	1.22556400	1.50309100	1.65480300
H	1.74911400	2.44213300	1.87809000
H	1.95492900	0.68892000	1.82747900
H	0.43904800	1.38338600	2.40149100

F⁻ in THF

Zero-point correction=	0.000000 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-100.007894
Sum of electronic and thermal Enthalpies=	-100.005533
Sum of electronic and thermal Free Energies=	-100.022052

(F)Me₃SiCF₂Cl (12) in THF

Zero-point correction=	0.123062 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-1207.338006
Sum of electronic and thermal Enthalpies=	-1207.323542
Sum of electronic and thermal Free Energies=	-1207.377806

Si	-0.15262100	-1.03582800	0.03391800
C	-0.21079200	1.11291500	0.16075700
F	-1.27527600	1.70970400	-0.48674400
F	-0.29048700	1.62692600	1.44046900
Cl	1.30917700	2.00372800	-0.57705900
C	-0.10523200	-0.93219400	-1.87130900
H	0.86520000	-1.29100400	-2.23094000
H	-0.26915300	0.06416200	-2.28389800
H	-0.86196500	-1.60842400	-2.28081600
C	-1.84629000	-1.04993200	0.91890600
H	-1.82578000	-0.46124400	1.84062900
H	-2.14294600	-2.07126500	1.16699800
H	-2.62192400	-0.61667100	0.27860800
C	1.43456700	-1.07299000	1.09232700
H	2.28249800	-1.36769600	0.46444100
H	1.32996000	-1.84284500	1.86269200
H	1.68825300	-0.12813300	1.57532900
F	-0.10444200	-2.83868200	-0.08305700

F-Me₃Si-CF₂Cl transition state (13) in THF

Imaginary Frequency: -69.3061cm^{-1}

Zero-point correction= 0.121351 (Hartree/Particle)
Sum of electronic and zero-point Energies= -1207.334778
Sum of electronic and thermal Enthalpies= -1207.319931
Sum of electronic and thermal Free Energies= -1207.377438

Si	1.49369100	-0.10134100	-0.00029000
C	1.52306600	1.78777300	-0.00684800
H	2.55829500	2.14106500	0.03230700
H	1.05871900	2.19050800	-0.91050900
H	0.98701900	2.20154800	0.85046000
C	1.04218900	-0.98795300	-1.59703500
H	1.64087200	-1.90002000	-1.68753900
H	-0.01308900	-1.25622100	-1.63802300
H	1.27165000	-0.35570400	-2.46103700
C	1.04347400	-0.97141600	1.60594600
H	1.64462600	-1.88071600	1.70704600
H	1.26930400	-0.32930600	2.46358700
H	-0.01102500	-1.24278300	1.64810200
F	3.17988500	-0.38998700	-0.00179800
C	-1.21555200	0.55937700	-0.00290700
F	-1.72212300	1.25448200	1.06855200
F	-1.73273600	1.25592200	-1.06807200
Cl	-2.52884400	-1.14523000	0.00237900

Me₃SiF in THF

Zero-point correction= 0.112714 (Hartree/Particle)
Sum of electronic and zero-point Energies= -509.168757
Sum of electronic and thermal Enthalpies= -509.159471
Sum of electronic and thermal Free Energies= -509.200458

Si	-0.00281200	0.00524100	-0.00252600
C	-1.74514100	-0.08811100	0.67122300
H	-1.74465300	-0.09804500	1.76485800
H	-2.25478500	-0.99062000	0.32195400
H	-2.33151100	0.77622900	0.34375200
C	0.94962300	1.46702800	0.67129800
H	1.98541100	1.45905800	0.32019800
H	0.95986100	1.46119400	1.76494700
H	0.49219700	2.40651200	0.34522200
C	0.05071800	-0.08737600	-1.86957500
H	1.08153700	-0.10078900	-2.23478700
H	-0.45050600	0.77968900	-2.31130900
H	-0.45244300	-0.98737400	-2.23421600
F	0.77962400	-1.35271500	0.55128200

CF₃⁻ in THF

Zero-point correction= 0.008780(Hartree/Particle)
Sum of electronic and zero-point Energies= -337.816683

Sum of electronic and thermal Enthalpies= -337.811945
Sum of electronic and thermal Free Energies= -337.842957

C	-0.25523700	0.44258400	0.18056700
F	-0.14796500	0.25740900	-1.24021900
F	1.11898500	0.25721700	0.55337100
F	-0.78305600	-0.84039200	0.55391700

II. Reactions in Toluene:

Me₃SiCF₂Cl (1) in Toluene

Zero-point correction= 0.122535 (Hartree/Particle)
Sum of electronic and zero-point Energies= -1107.308261
Sum of electronic and thermal Enthalpies= -1107.295463
Sum of electronic and thermal Free Energies= -1107.346135

Si	-0.17675200	0.00000000	1.07209300
C	-1.98296000	0.00000000	1.58571500
H	-2.06133400	0.00000000	2.67747200
H	-2.50459100	-0.88539200	1.21318900
H	-2.50459100	0.88539200	1.21318900
C	0.71263500	1.56453800	1.59977800
H	1.74980000	1.57373300	1.25614600
H	0.72020200	1.63903000	2.69179100
H	0.21472500	2.45580000	1.20886300
C	0.71263500	-1.56453800	1.59977800
H	1.74980000	-1.57373300	1.25614600
H	0.21472500	-2.45580000	1.20886300
H	0.72020200	-1.63903000	2.69179100
C	-0.20656200	0.00000000	-0.88527700
F	-0.86667100	1.08942200	-1.38267600
F	-0.86667100	-1.08942200	-1.38267600
Cl	1.44983200	0.00000000	-1.63725600

Cl[•] in Toluene

Zero-point correction= 0.000000 (Hartree/Particle)
Sum of electronic and zero-point Energies= -460.368569
Sum of electronic and thermal Enthalpies= -460.366209
Sum of electronic and thermal Free Energies= -460.383592

Me₃SiCF₂Cl^{•-} radical anion in Toluene

Zero-point correction= 0.120501(Hartree/Particle)
Sum of electronic and zero-point Energies= -1107.317603
Sum of electronic and thermal Enthalpies= -1107.304687
Sum of electronic and thermal Free Energies= -1107.355996

Si	-0.17102800	0.00000000	1.11468500
C	-1.99240500	0.00000000	1.58299000
H	-2.08316200	0.00000000	2.67412700
H	-2.50280900	-0.89283900	1.19933000

H	-2.50280900	0.89283900	1.19933000
C	0.71928300	1.58224700	1.59444700
H	1.76086800	1.58030300	1.25041200
H	0.72247500	1.67624900	2.68490200
H	0.20959000	2.46219600	1.17959600
C	0.71928300	-1.58224700	1.59444700
H	1.76086800	-1.58030300	1.25041200
H	0.20959000	-2.46219600	1.17959600
H	0.72247500	-1.67624900	2.68490200
C	-0.21017600	0.00000000	-0.85490400
F	-0.86929800	1.08945700	-1.35915500
F	-0.86929800	-1.08945700	-1.35915500
Cl	1.45097800	0.00000000	-1.62903400

Cl• radical in Toluene

Zero-point correction=	0.000000 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-460.167407
Sum of electronic and thermal Enthalpies=	-460.165046
Sum of electronic and thermal Free Energies=	-460.183084

Cl-Me₃Si-CF₂Cl transition state (10) in Toluene

Imaginary Frequency: - 78.3556 cm⁻¹

Zero-point correction=	0.120607(Hartree/Particle)
Sum of electronic and zero-point Energies=	-1567.654733
Sum of electronic and thermal Enthalpies=	-1567.639459
Sum of electronic and thermal Free Energies=	-1567.698028

Si	1.20647700	-0.03160400	-0.00756100
C	0.73654900	-0.87998900	1.59932500
H	1.18551700	-0.35238600	2.44605900
H	1.12426000	-1.90269900	1.60004900
H	-0.34399000	-0.91297400	1.73127200
C	1.25954900	1.84955700	-0.03572000
H	1.66918800	2.22385600	0.90715500
H	0.27300900	2.28645100	-0.18523400
H	1.92007200	2.18359700	-0.84111200
C	0.74099400	-0.89224700	-1.60806900
H	1.23519800	-0.40404000	-2.45318400
H	-0.33725000	-0.87626400	-1.76409100
H	1.07648800	-1.93283100	-1.58017700
C	-1.74463100	0.57791600	0.03410500
F	-2.26588400	1.18765100	1.15772300
F	-2.31698600	1.32123800	-0.97950100
Cl	-2.95751700	-1.17221900	-0.04675400
Cl	3.39465100	-0.41921200	0.01186100

CF₂Cl (11) in Toluene

Zero-point correction=	0.007371(Hartree/Particle)
Sum of electronic and zero-point Energies=	-698.145448

Sum of electronic and thermal Enthalpies= -698.139903
Sum of electronic and thermal Free Energies= -698.173818

C	0.70947000	0.51317700	-0.06334000
F	0.45308000	0.43179100	-1.40183300
F	-0.15954900	1.49000300	0.32822100
Cl	-0.51790300	-1.35023900	0.64448900

Me₃SiCl in Toluene

Zero-point correction= 0.112040(Hartree/Particle)
Sum of electronic and zero-point Energies= -869.507332
Sum of electronic and thermal Enthalpies= -869.497757
Sum of electronic and thermal Free Energies= -869.539939

Si	0.33436700	0.00000000	-0.11856200
C	0.24299800	0.00000000	-1.98978700
H	1.25158600	0.00000000	-2.41615400
H	-0.28053100	0.88522500	-2.35984300
H	-0.28053100	-0.88522500	-2.35984300
C	1.14057900	1.55414700	0.54853700
H	2.18214500	1.61162000	0.21514600
H	1.13460800	1.56426000	1.64152500
H	0.62308600	2.45034500	0.19665300
C	1.14057900	-1.55414700	0.54853700
H	2.18214500	-1.61162000	0.21514600
H	0.62308600	-2.45034500	0.19665300
H	1.13460800	-1.56426000	1.64152500
Cl	-1.66979100	0.00000000	0.58998400

CF₂ in Toluene

Zero-point correction= 0.006685(Hartree/Particle)
Sum of electronic and zero-point Energies= -237.768415
Sum of electronic and thermal Enthalpies= -237.764467
Sum of electronic and thermal Free Energies= -237.792478

C	0.25511300	-0.41167400	0.34255600
F	0.21035700	-0.37527300	-0.96315800
F	-0.38684500	0.66007200	0.72622700

Tetramethylethylene (6a) in Toluene

Zero-point correction= 0.162535(Hartree/Particle)
Sum of electronic and zero-point Energies= -235.761657
Sum of electronic and thermal Enthalpies= -235.751897
Sum of electronic and thermal Free Energies= -235.796100

C	-0.03320100	-0.01773700	0.67501900
C	0.02889100	0.01829800	-0.67030700
C	-0.10742800	1.20790600	1.55917900
H	-0.10926900	2.15202500	1.01856600
H	0.73911400	1.22481900	2.25631500

H	-1.01307600	1.17608400	2.17706300
C	-0.03752700	-1.28973800	1.49420000
H	-0.94689100	-1.34564700	2.10486200
H	0.80479000	-1.29199500	2.19658400
H	0.02125100	-2.20190100	0.90445600
C	0.10286700	-1.20736300	-1.55447200
H	-0.74280200	-1.22316200	-2.25268300
H	0.10276500	-2.15152500	-1.01392700
H	1.00935100	-1.17663000	-2.17118800
C	0.03348000	1.29030600	-1.48947000
H	-0.02326000	2.20252800	-0.89961500
H	-0.80983900	1.29363300	-2.19065400
H	0.94203400	1.34509900	-2.10142700

Tetramethylethylene + CF₂ Transition State in Toluene

Imaginary Frequency: - 243.5704cm⁻¹

Zero-point correction= 0.171450 (Hartree/Particle)

Sum of electronic and zero-point Energies= -473.523489

Sum of electronic and thermal Enthalpies= -473.511095

Sum of electronic and thermal Free Energies= -473.560793

C	-1.53674400	-1.25696900	0.14671100
C	0.56133000	0.11516100	0.03195400
C	-0.75715100	0.52097100	-0.16015900
F	-1.14537600	-1.74723700	1.35315500
F	-1.02108900	-2.15134500	-0.73945200
C	-1.27029800	0.80734700	-1.56320600
H	-0.94662200	1.80559400	-1.87727600
H	-2.36108200	0.78710900	-1.56728800
H	-0.92068200	0.08695400	-2.30104800
C	-1.49124900	1.32755600	0.90014100
H	-1.32429700	0.96118200	1.91156800
H	-2.56457600	1.29165500	0.70764100
H	-1.17051900	2.37376300	0.85587700
C	1.18696300	0.01604200	1.38863100
H	2.23559300	0.32256700	1.34540900
H	1.17936100	-1.03393000	1.72102300
H	0.68621200	0.60547600	2.15320100
C	1.38462900	-0.45170700	-1.08584200
H	2.44655900	-0.43313700	-0.83543100
H	1.24215000	0.07061200	-2.03223000
H	1.10911700	-1.50445700	-1.25427000

2,2-difluorotetramethylcyclopropane in Toluene

Zero-point correction= 0.176228(Hartree/Particle)

Sum of electronic and zero-point Energies= -473.595796

Sum of electronic and thermal Enthalpies= -473.584158

Sum of electronic and thermal Free Energies= -473.630576

C	0.82623400	0.24776900	0.20564900
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C	-0.57438100	-0.41032100	0.53563900
C	-0.26977400	0.17479100	-0.79700900
F	-0.14643600	-0.63691800	-1.89110600
F	-0.90252800	1.31753700	-1.20377600
C	-0.66975100	-1.92232300	0.66982100
H	-0.06777200	-2.45722700	-0.06260400
H	-1.70791500	-2.24092700	0.53749000
H	-0.35078500	-2.23331200	1.66921800
C	-1.54100600	0.32242400	1.45380600
H	-1.26724100	0.15834000	2.50039900
H	-2.55357000	-0.06591400	1.30969900
H	-1.57477200	1.39556900	1.27392100
C	1.17164000	1.59741400	0.81533300
H	1.97305000	2.06972400	0.23952000
H	1.52996000	1.46918300	1.84116700
H	0.33040000	2.28799200	0.83071200
C	2.04306500	-0.64760400	0.02971800
H	2.82216100	-0.10899700	-0.51775500
H	1.82855200	-1.56088800	-0.52251800
H	2.45303700	-0.92468700	1.00563500

Tetramethylethylene + CF₂Cl Transition State in Toluene

Imaginary Frequency: -310.9030 cm⁻¹

Zero-point correction= 0.171410(Hartree/Particle)

Sum of electronic and zero-point Energies= -933.876305

Sum of electronic and thermal Enthalpies= -933.861993

Sum of electronic and thermal Free Energies= -933.917240

C	0.29311900	-1.03599600	-0.34393100
F	1.31944800	-1.03322100	-1.23031700
F	0.87042600	-1.43066400	0.81766300
Cl	-0.85117900	-3.09968300	-0.99483400
C	-0.40766400	0.75645400	-0.15262100
C	-0.97959800	0.94322100	-1.55542000
H	-0.20925700	0.93964900	-2.32635700
H	-1.52220700	1.89531900	-1.61423600
H	-1.66839500	0.12693300	-1.78109000
C	-1.52331900	0.47966300	0.85115600
H	-1.14649500	0.16960900	1.82568900
H	-2.16043100	-0.32446100	0.47766900
H	-2.13202000	1.38233300	0.98921800
C	0.71434400	1.50200900	0.24251800
C	1.74817700	1.97196200	-0.74689900
H	2.24630100	2.88041900	-0.38344000
H	1.33715100	2.19807300	-1.73217100
H	2.54356000	1.21761900	-0.89894900
C	1.21790000	1.49552000	1.66182500
H	1.74028900	2.43348600	1.89221100
H	1.94652400	0.68056500	1.83393600
H	0.42788400	1.37212100	2.40420300

F⁻ in Toluene

Zero-point correction= 0.000000 (Hartree/Particle)
Sum of electronic and zero-point Energies= -99.968381
Sum of electronic and thermal Enthalpies= -99.966020
Sum of electronic and thermal Free Energies= -99.982539

(F)Me₃SiCF₂Cl (12) in Toluene

Zero-point correction= 0.123402 (Hartree/Particle)
Sum of electronic and zero-point Energies= -1207.316052
Sum of electronic and thermal Enthalpies= -1207.301704
Sum of electronic and thermal Free Energies= -1207.355543

Si	-0.16149300	-1.05371700	0.03398700
C	-0.19041100	1.10749400	0.15194200
F	-1.25270300	1.71647200	-0.48735700
F	-0.25057800	1.62833900	1.43000000
Cl	1.33680500	1.97538300	-0.60665800
C	-0.11418100	-0.92921800	-1.87094000
H	0.91802600	-1.05339800	-2.21693400
H	-0.48283800	0.01872600	-2.26652700
H	-0.69841000	-1.74696200	-2.30144500
C	-1.85458000	-1.03419900	0.92330700
H	-1.82576800	-0.42781400	1.83332300
H	-2.15542500	-2.05005900	1.18942600
H	-2.62797000	-0.60618000	0.27669300
C	1.42460000	-1.09426200	1.09660200
H	2.27492300	-1.37249300	0.46427600
H	1.32233700	-1.88038900	1.85070200
H	1.67045000	-0.15526300	1.59500200
F	-0.13003800	-2.84193000	-0.07414600

F-Me₃Si-CF₂Cl⁺ transition state (13) in Toluene

Imaginary Frequency: -58.9705cm⁻¹

Zero-point correction= 0.121684 (Hartree/Particle)
Sum of electronic and zero-point Energies= -1207.312082
Sum of electronic and thermal Enthalpies= -1207.297377
Sum of electronic and thermal Free Energies= -1207.353942

Si	1.55292100	-0.10821700	0.00425400
C	1.54270900	1.77909500	-0.03602300
H	2.56898000	2.14967900	-0.12877100
H	0.95909200	2.15802500	-0.87757700
H	1.10793500	2.19499800	0.87648300
C	1.07836900	-1.00095200	-1.57832800
H	1.61766100	-1.95210200	-1.63715100
H	0.00779200	-1.19994200	-1.62410600
H	1.36045500	-0.40531600	-2.45274900
C	1.05965600	-0.95809900	1.60673500
H	1.63298500	-1.88484500	1.71399100

H	1.29410200	-0.32216400	2.46657800
H	-0.00326200	-1.19623600	1.63393500
F	3.22950200	-0.38856200	0.02703500
C	-1.29746400	0.57903900	0.01739600
F	-1.83614500	1.14895400	1.14961100
F	-1.81503800	1.38102200	-0.97583800
Cl	-2.57082700	-1.14437800	-0.18115500

Me₃SiF in Toluene

Zero-point correction=	0.112870(Hartree/Particle)
Sum of electronic and zero-point Energies=	-509.166917
Sum of electronic and thermal Enthalpies=	-509.157635
Sum of electronic and thermal Free Energies=	-509.198633

Si	0.00052100	-0.00058500	-0.00006300
C	-1.74444000	-0.08742000	0.67050400
H	-1.74595800	-0.09839100	1.76411000
H	-2.25558700	-0.98952600	0.32245200
H	-2.33098100	0.77706100	0.34341800
C	0.94863200	1.46676400	0.67053600
H	1.98484400	1.46024500	0.32070800
H	0.96077500	1.46220100	1.76415100
H	0.49125300	2.40647000	0.34475600
C	0.05034200	-0.08671800	-1.86855100
H	1.08049600	-0.10082400	-2.23557800
H	-0.45082500	0.78029200	-2.31065800
H	-0.45183000	-0.98646900	-2.23508800
F	0.77988000	-1.35317800	0.55162200

CF₃⁻ in Toluene

Zero-point correction=	0.008823 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-337.788257
Sum of electronic and thermal Enthalpies=	-337.783522
Sum of electronic and thermal Free Energies=	-337.814535

C	-0.25589100	0.44379600	0.18102800
F	-0.14807100	0.25737600	-1.24173200
F	1.12018700	0.25707700	0.55397700
F	-0.78349800	-0.84143200	0.55436400

III. Reactions in gas phase:

Me₃SiCF₂Cl (1) in Gas Phase

Zero-point correction=	0.122786 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-1107.306093
Sum of electronic and thermal Enthalpies=	-1107.293311
Sum of electronic and thermal Free Energies=	-1107.343841

Si	-0.08952100	0.15507900	1.06437100
C	-0.99412800	1.71829700	1.58391600

H	-0.48573100	2.61385200	1.21717400
H	-1.04287000	1.78484300	2.67532900
H	-2.01838300	1.73120500	1.20232800
C	1.70873800	0.16967300	1.60059100
H	2.23129300	1.04751000	1.21133900
H	2.23862400	-0.72097500	1.25420600
H	1.77705900	0.19763000	2.69263500
C	-0.99561500	-1.39792200	1.60183700
H	-0.48024200	-2.29961500	1.26232900
H	-2.01436700	-1.42094800	1.20589000
H	-1.06104200	-1.43911800	2.69362700
C	-0.10305800	0.17830400	-0.88876300
F	0.52879100	1.28357500	-1.38313600
F	-1.37611500	0.22326400	-1.38158400
Cl	0.70389500	-1.27152000	-1.62951600

Cl⁻ in Gas Phase

Zero-point correction=	0.000000 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-460.303727
Sum of electronic and thermal Enthalpies=	-460.301367
Sum of electronic and thermal Free Energies=	-460.318750

Me₃SiCF₂Cl^{•-} radical anion in Gas Phase

Zero-point correction=	0.121072(Hartree/Particle)
Sum of electronic and zero-point Energies=	-1107.287392
Sum of electronic and thermal Enthalpies=	-1107.274490
Sum of electronic and thermal Free Energies=	-1107.325891

Si	-0.08722500	0.15049500	1.10607000
C	-0.99818300	1.72614200	1.58041700
H	-0.47878500	2.61518800	1.20199800
H	-1.04833500	1.80246600	2.67096200
H	-2.02418800	1.72477200	1.19249300
C	1.72451000	0.17336800	1.59848600
H	2.22923700	1.06166700	1.19841200
H	2.25000000	-0.72118100	1.24451700
H	1.80242200	0.20275500	2.68941500
C	-1.00677900	-1.41076500	1.59876800
H	-0.48404000	-2.31000900	1.25284300
H	-2.02507400	-1.41472300	1.19015000
H	-1.07977800	-1.45954900	2.68924800
C	-0.10460400	0.18091700	-0.86334600
F	0.52881500	1.28267300	-1.36950400
F	-1.37544000	0.22487700	-1.36802100
Cl	0.70477500	-1.27595700	-1.63033700

Cl[•] radical in Gas Phase

Zero-point correction=	0.000000 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-460.166883

Sum of electronic and thermal Enthalpies= -460.164522
Sum of electronic and thermal Free Energies= -460.182560

(CD)Me₃SiCF₂Cl in Gas Phase

Zero-point correction= 0.123119(Hartree/Particle)
Sum of electronic and zero-point Energies= -1567.634720
Sum of electronic and thermal Enthalpies= -1567.619708
Sum of electronic and thermal Free Energies= -1567.676718

Si	0.16867800	0.54540100	0.77371500
Cl	0.22566700	0.27086800	4.40370400
C	0.18288300	0.65509900	-1.20322500
F	1.31926400	1.24347500	-1.69316700
F	0.13215200	-0.57620500	-1.80231600
Cl	-1.22139300	1.61423200	-1.91203300
C	0.30998100	2.31942200	1.35620800
H	1.22777000	2.78897700	0.98964100
H	0.33330700	2.29255400	2.45024000
H	-0.54206400	2.92164800	1.02887900
C	1.68071000	-0.49926400	1.14761100
H	2.59863400	-0.02822900	0.78364700
H	1.60194600	-1.49561100	0.70269500
H	1.73272800	-0.59567100	2.23709200
C	-1.45068200	-0.29204000	1.20080900
H	-1.46595400	-0.41160400	2.28889500
H	-1.52861300	-1.27658500	0.72983500
H	-2.31092400	0.30871400	0.89304200

Cl-Me₃Si-CF₂Cl transition state (10) in Gas Phase

Imaginary Frequency: - 62.1942cm⁻¹
Zero-point correction= 0.120457(Hartree/Particle)
Sum of electronic and zero-point Energies= -1567.613120
Sum of electronic and thermal Enthalpies= -1567.597661
Sum of electronic and thermal Free Energies= -1567.658494

Si	1.27910800	-0.04485300	-0.00669700
C	0.74875300	-0.86812300	1.59215200
H	1.12628600	-0.31093100	2.45473600
H	1.15801900	-1.88141700	1.63960500
H	-0.33871300	-0.91952000	1.64548100
C	1.31316500	1.83585700	-0.04652900
H	1.72455400	2.22462100	0.88995300
H	0.31084500	2.23887700	-0.18681300
H	1.95819400	2.17692100	-0.86183900
C	0.73623900	-0.90226100	-1.58131200
H	1.10799000	-0.36514000	-2.45911500
H	-0.35266600	-0.95301600	-1.62227300
H	1.14017900	-1.91832100	-1.61103700
C	-1.81142500	0.63134700	0.04916100

F	-2.39916100	1.13636600	1.19491300
F	-2.40209700	1.40622100	-0.93324100
Cl	-2.89030300	-1.18195100	-0.17524600
Cl	3.44672600	-0.45087500	0.00424600

CF₂Cl[•] (11) in Gas Phase

Zero-point correction=	0.007443(Hartree/Particle)
Sum of electronic and zero-point Energies=	-698.092687
Sum of electronic and thermal Enthalpies=	-698.087236
Sum of electronic and thermal Free Energies=	-698.120823

C	0.69822000	0.48502800	-0.04999600
F	0.44951400	0.42179900	-1.40474400
F	-0.16664100	1.48710700	0.33559600
Cl	-0.49599600	-1.30920200	0.62668100

Me₃SiCl in Gas Phase

Zero-point correction=	0.112227(Hartree/Particle)
Sum of electronic and zero-point Energies=	-869.505232
Sum of electronic and thermal Enthalpies=	-869.495680
Sum of electronic and thermal Free Energies=	-869.537819

Si	-0.32209600	-0.00000500	-0.11454700
C	-0.24427600	0.00004900	-1.98833700
H	0.27884000	-0.88409900	-2.36174300
H	0.27884500	0.88421600	-2.36169300
H	-1.25373000	0.00006400	-2.41306500
C	-1.14053200	1.55208200	0.54761800
H	-0.62491200	2.45051600	0.19865000
H	-1.13701800	1.56440000	1.64063800
H	-2.18171700	1.60837500	0.21259500
C	-1.14054400	-1.55212500	0.54752800
H	-0.62493200	-2.45054300	0.19850600
H	-2.18172900	-1.60838900	0.21250200
H	-1.13702800	-1.56450600	1.64054700
Cl	1.66951800	-0.00003400	0.58956000

CF₂ in Gas Phase

Zero-point correction=	0.006749 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-237.767613
Sum of electronic and thermal Enthalpies=	-237.763667
Sum of electronic and thermal Free Energies=	-237.791677

C	0.25559400	-0.41245000	0.34319300
F	0.21005700	-0.37478200	-0.96330300
F	-0.38702600	0.66035600	0.72573500

Tetramethylethylene (6a) in Gas Phase

Zero-point correction= 0.162751 (Hartree/Particle)
Sum of electronic and zero-point Energies= -235.760940
Sum of electronic and thermal Enthalpies= -235.751233
Sum of electronic and thermal Free Energies= -235.794532

C	-0.03312900	-0.01792900	0.67477200
C	0.02913000	0.01838900	-0.66994400
C	-0.10135000	1.20786600	1.55873800
H	-0.04004100	2.15194900	1.02150500
H	0.71411900	1.19079600	2.29172400
H	-1.03393300	1.21196400	2.13632100
C	-0.04393100	-1.29021100	1.49291900
H	-0.92672500	-1.31432400	2.14310200
H	0.82830800	-1.32584100	2.15711000
H	-0.04727400	-2.20262000	0.90059900
C	0.11101600	-1.20685000	-1.55344900
H	-0.76020800	-1.25837100	-2.21795200
H	0.17008700	-2.14787600	-1.01070300
H	0.99107000	-1.14496900	-2.20489700
C	0.02577800	1.29027900	-1.48873100
H	-0.08735600	2.19928000	-0.90203300
H	-0.78948500	1.26589300	-2.22176800
H	0.95517500	1.37757700	-2.06481200

Tetramethylethylene + CF₂ Transition State in Gas Phase

Imaginary Frequency: - 260.4629cm⁻¹

Zero-point correction= 0.171837(Hartree/Particle)
Sum of electronic and zero-point Energies= -473.521195
Sum of electronic and thermal Enthalpies= -473.508925
Sum of electronic and thermal Free Energies= -473.558167

C	-1.51957500	-1.26035100	0.09921000
C	0.56795300	0.12470100	0.02244900
C	-0.75571300	0.51821000	-0.16710900
F	-1.16732800	-1.75220300	1.31279900
F	-0.97859700	-2.14432400	-0.77531100
C	-1.27806000	0.83259200	-1.56124500
H	-0.95055900	1.83334900	-1.86307300
H	-2.36882800	0.81708300	-1.55431800
H	-0.94336900	0.12026700	-2.31349900
C	-1.48886200	1.29865200	0.91401100
H	-1.31173000	0.90940500	1.91536600
H	-2.56326800	1.25938900	0.72911200
H	-1.17479200	2.34781100	0.89074700
C	1.17895600	0.02542900	1.38817700
H	2.26630400	0.10876500	1.33336800
H	0.95220700	-0.96036500	1.82561000
H	0.81069600	0.77677500	2.08577300
C	1.39494300	-0.45400000	-1.08531400
H	2.45716500	-0.26729600	-0.91254400
H	1.13398600	-0.07310800	-2.07135800

H 1.26069900 -1.54757500 -1.11373900

2,2-difluorotetramethylcyclopropane in Gas Phase

Zero-point correction= 0.176401(Hartree/Particle)
Sum of electronic and zero-point Energies= -473.593860
Sum of electronic and thermal Enthalpies= -473.582224
Sum of electronic and thermal Free Energies= -473.628650

C	0.82601800	0.24761000	0.20575200
C	-0.57411200	-0.41021600	0.53558200
C	-0.27025100	0.17517800	-0.79858600
F	-0.14485400	-0.63800200	-1.88737700
F	-0.90124600	1.31708500	-1.19978900
C	-0.66875300	-1.92208100	0.66959500
H	-0.07174200	-2.45561100	-0.06784200
H	-1.70743500	-2.24111500	0.54188300
H	-0.34469800	-2.23595400	1.66671900
C	-1.53992800	0.32317200	1.45378300
H	-1.26301600	0.16735900	2.50106300
H	-2.55175400	-0.06939700	1.31558200
H	-1.57949300	1.39477500	1.26612300
C	1.17056300	1.59699700	0.81596000
H	1.97565200	2.06788100	0.24398600
H	1.52300700	1.47164200	1.84444600
H	0.33039100	2.28897100	0.82355300
C	2.04216900	-0.64835000	0.02985500
H	2.82403100	-0.10852000	-0.51259600
H	1.82773600	-1.55720900	-0.52965600
H	2.44988100	-0.93259000	1.00492300

Tetramethylethylene + CF₂Cl Transition State in Gas Phase

Imaginary Frequency: -309.6489 cm⁻¹

Zero-point correction= 0.171280(Hartree/Particle)
Sum of electronic and zero-point Energies= -933.829448
Sum of electronic and thermal Enthalpies= -933.815106
Sum of electronic and thermal Free Energies= -933.869467

C	0.29311900	-1.03599600	-0.34393100
F	1.31944800	-1.03322100	-1.23031700
F	0.87042600	-1.43066400	0.81766300
Cl	-0.85117900	-3.09968300	-0.99483400
C	-0.40766400	0.75645400	-0.15262100
C	-0.97959800	0.94322100	-1.55542000
H	-0.20925700	0.93964900	-2.32635700
H	-1.52220700	1.89531900	-1.61423600
H	-1.66839500	0.12693300	-1.78109000
C	-1.52331900	0.47966300	0.85115600
H	-1.14649500	0.16960900	1.82568900
H	-2.16043100	-0.32446100	0.47766900
H	-2.13202000	1.38233300	0.98921800

C	0.71434400	1.50200900	0.24251800
C	1.74817700	1.97196200	-0.74689900
H	2.24630100	2.88041900	-0.38344000
H	1.33715100	2.19807300	-1.73217100
H	2.54356000	1.21761900	-0.89894900
C	1.21790000	1.49552000	1.66182500
H	1.74028900	2.43348600	1.89221100
H	1.94652400	0.68056500	1.83393600
H	0.42788400	1.37212100	2.40420300

F⁻ in Gas Phase

Zero-point correction=	0.000000 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-99.888693
Sum of electronic and thermal Enthalpies=	-99.886333
Sum of electronic and thermal Free Energies=	-99.902852

(F)Me₃SiCF₂Cl (12) in Gas Phase

Zero-point correction=	0.123145(Hartree/Particle)
Sum of electronic and zero-point Energies=	-1207.274408
Sum of electronic and thermal Enthalpies=	-1207.259756
Sum of electronic and thermal Free Energies=	-1207.315456

Si	-0.16163800	-1.06806900	0.03488200
C	-0.19707600	1.10850100	0.14567200
F	-1.26014400	1.71505800	-0.49432100
F	-0.26059100	1.62570000	1.42455700
Cl	1.33151900	1.98707000	-0.61131900
C	-0.11173400	-0.94473800	-1.87106500
H	0.92849500	-1.00785900	-2.20968600
H	-0.52964900	-0.01714600	-2.26680700
H	-0.64540500	-1.79678200	-2.30106300
C	-1.85946900	-1.03070100	0.91958200
H	-1.77952700	-0.54773400	1.89881200
H	-2.22930600	-2.04909400	1.05951700
H	-2.59726500	-0.46394300	0.34275700
C	1.42866200	-1.08649800	1.09556600
H	2.26914300	-1.40264200	0.46771200
H	1.31723500	-1.84540200	1.87637900
H	1.69041500	-0.13297100	1.55612300
F	-0.13091800	-2.84222100	-0.04604600

F-Me₃Si-CF₂Cl⁻ transition state (13) in Gas Phase

Imaginary Frequency: -54.4719 cm ⁻¹	
Zero-point correction=	0.121315 (Hartree/Particle)
Sum of electronic and zero-point Energies=	-1207.269141
Sum of electronic and thermal Enthalpies=	-1207.254131
Sum of electronic and thermal Free Energies=	-1207.312675

Si	1.56979800	-0.18555700	-0.06388000
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C	-1.44328300	0.73528400	0.06240800
F	-1.88375600	1.54211300	1.09589000
F	-2.06749800	1.31102000	-1.02603500
Cl	-2.68436000	-0.98885100	0.35271500
C	1.06982400	-1.05884600	1.51736000
H	1.43848000	-2.09005300	1.50296000
H	-0.01594900	-1.07707300	1.62179000
H	1.50321300	-0.56013100	2.39064100
C	1.66782200	1.69331700	0.04648300
H	1.54799400	2.15015000	-0.94041300
H	2.64821700	1.98416200	0.44010400
H	0.88655500	2.09501000	0.69196100
C	0.86265500	-0.93605500	-1.63160600
H	-0.21399700	-1.08173600	-1.54193900
H	1.34170100	-1.90294600	-1.82069000
H	1.05514700	-0.29133300	-2.49519100
F	3.20499500	-0.58199800	-0.20307600

Me₃SiF in Gas Phase

Zero-point correction=	0.113077(Hartree/Particle)
Sum of electronic and zero-point Energies=	-509.164502
Sum of electronic and thermal Enthalpies=	-509.155228
Sum of electronic and thermal Free Energies=	-509.196223

Si	0.00495200	-0.00810000	0.00264600
C	-1.74329300	-0.08648700	0.66978000
H	-1.74670500	-0.09805900	1.76339200
H	-2.25712500	-0.98777700	0.32360800
H	-2.32940500	0.77851600	0.34307200
C	0.94727600	1.46628000	0.66992700
H	1.98425300	1.46187000	0.32238600
H	0.96041500	1.46339400	1.76354800
H	0.48967300	2.40588300	0.34392800
C	0.04997700	-0.08617800	-1.86774500
H	1.07926800	-0.10048900	-2.23722600
H	-0.45161400	0.78064200	-2.30988400
H	-0.45093700	-0.98576600	-2.23643100
F	0.78038500	-1.35380800	0.55131900

CF₃⁻ in Gas Phase

Zero-point correction=	0.008835(Hartree/Particle)
Sum of electronic and zero-point Energies=	-337.731374
Sum of electronic and thermal Enthalpies=	-337.726635
Sum of electronic and thermal Free Energies=	-337.757663

C	-0.25645100	0.44449300	0.18139400
F	-0.14814800	0.25790000	-1.24409600
F	1.12311300	0.25798700	0.55484600
F	-0.78578700	-0.84356200	0.55549200