Supplementary Material (ESI) for Chemical Communications

Electronic Supplementary Information (ESI)

Isothermal titration calorimetry as a new tool to investigate chiral interactions at crystal surfaces†

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Isothermal titration calorimetry (ITC) experiments

The ITC measurements were performed with a VP-ITC calorimeter produced by MicroCal Inc. (Northampton, MA). The chiral ITC measurements were performed as follows: First 25 mg of each tyrosin enantiomers powders was manually grinded and suspended in 5 mL double distilled water. The reservoir cell was filled with 1.442 mL of solution of the L or D tyrosin and thermostated at 25^0C. Stock solutions of 0.5M of either L- Threonine or L Phenylalanine in distilled water at pH = 7.10 were prepared. Injection volumes of 5.0 µL were delivered during 30 min, with the time interval of 200 seconds between the injections and the reference cell was filled with water

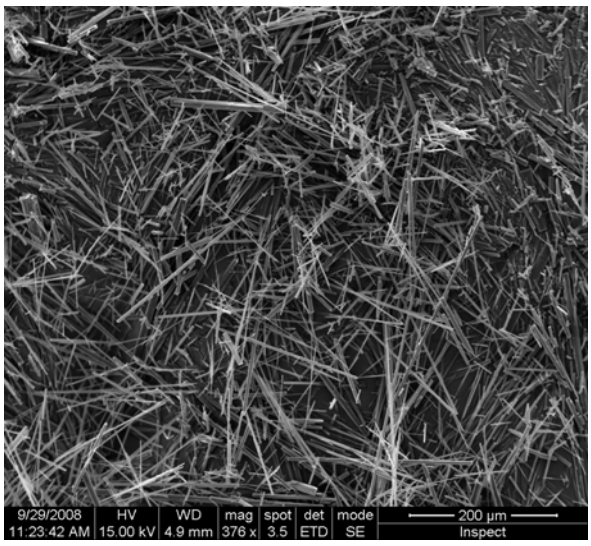


Figure S1 Scanning electron micrograph of L-tyrosine crystals crystallized from a supersaturated aqueous solution.

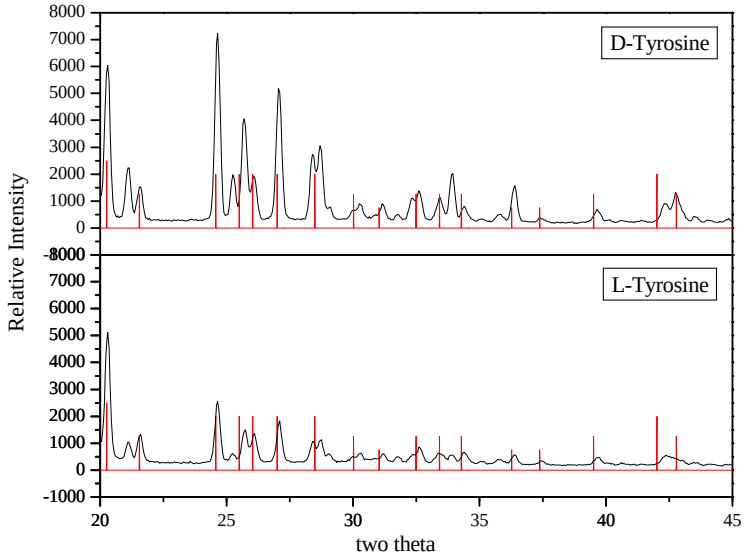
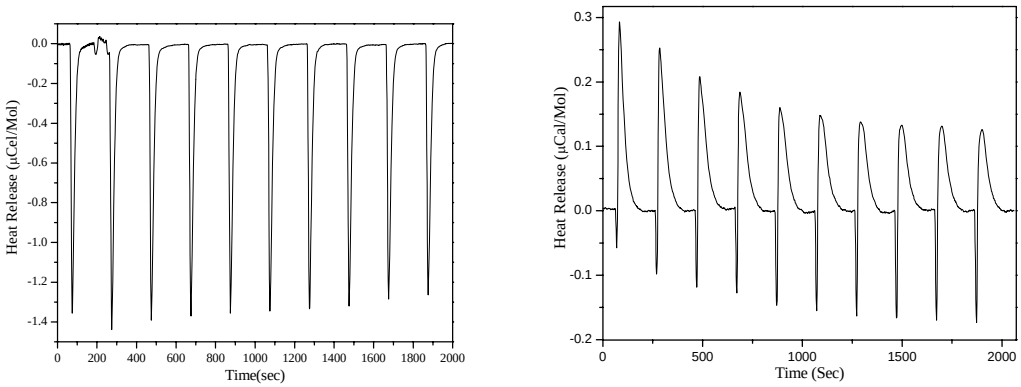


Figure S2 – X-ray diffraction pattern of L and D Tyrosin crystals



A **B**

Figure S3 ITC profile of the heat flow per injection in dilution of (**A**) L- phenylalanine **(B)** L threonine**.**