Supplementary Information

Gene delivery with polycationic fullerene hexakis-adducts

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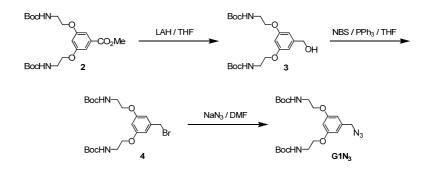
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Experimental section

General: Reagents and solvents were purchased as reagent grade and used without further purification. Compounds $1^{[1]}$ and $2^{[2]}$ were prepared according to a previously reported procedure. All reactions were performed in standard glassware under an inert Ar or N₂ atmosphere. Evaporation and concentration were done at water aspirator pressure and drying in vacuo at 10^{-2} Torr. Column chromatography: silica gel 60 (230-400 mesh, 0.040-0.063 mm) was purchased from E. Merck. Thin Layer Chromatography (TLC) was performed on glass sheets coated with silica gel 60 F₂₅₄ purchased from E. Merck, visualization by UV light. IR spectra (cm⁻¹) were measured on an ATI Mattson Genesis Series FTIR instrument. NMR spectra were recorded on a Bruker AC 300 or AC 400 with solvent peaks as reference. Elemental analysis were performed by the analytical service at the Laboratoire de Chimie de Coordination (Toulouse, France).

Preparation of compound G1N₃.



Compound 3. A 1 M LAH solution in THF (8.2 mL, 8.23 mmol) was added dropwise to a solution of **2** (3.74 g, 8.23 mmol) in dry THF (100 mL) at 0°C under argon. After 3 h, MeOH (1 mL) was slowly added then H₂O (several drops). The resulting mixture was filtered on Celite and evaporated. Column chromatography (SiO₂, CH₂Cl₂/AcOEt 1:1) gave **3** (3.52 g, 99%) as a white solid. IR (neat): 3347 (O-H), 3360 (N-H), 1689 (C=O) cm⁻¹. ¹H NMR (300 MHz, CDCl₃): 6.53 (d, J = 2 Hz, 2H), 6.37 (t, J = 2 Hz, 1H), 4.95 (broad s, 2H), 4.63 (s, 2H), 4.01 (t, J = 5 Hz, 4H), 3.52 (m, 4H), 1.45 (s, 18H). ¹³C NMR (75 MHz, CDCl₃): 160.0, 156.0, 143.9, 105.5, 100.6, 79.7, 67.3, 65.0, 40.1, 28.5. Elemental analysis (%) calcd for C₂₁H₃₄N₂O₇: C 59.14, H 8.03, N 6.57; found: C 58.56, H 8.04, N 6.31.

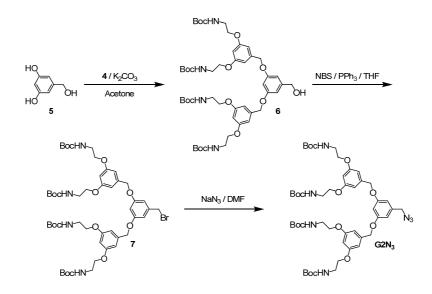
¹ J. Iehl, J.-F. Nierengarten, Chem. Eur. J. 2009, 15, 7306-7309.

² A. J. Brouwer, S. J. E. Mulders, R. M. J. Liskamp, Eur.J. Org. Chem. 2001, 1903-1915.

Compound 4. A solution of **3** (3.52 g, 8.26 mmol), PPh₃ (2.82 g, 10.08 mmol) and *N*bromosuccinimide (1.91 g, 10.07 mmol) in THF (50 mL) was stirred at room temperature. After 15 min., H₂O was added to the mixture. The resulting aqueous layer was extracted with AcOEt (3 X). The combined organic layers were dried (MgSO₄), filtered and evaporated. Column chromatography (SiO₂, cyclohexane/AcOEt 7:3) gave **4** (3.24 g, 80%) as a white solid. IR (neat): 3366 (N-H), 1677 (C=O). ¹H NMR (300 MHz, CDCl₃): 6.54 (d, J = 2 Hz, 2H), 6.38 (t, J = 2 Hz, 1H), 4.94 (broad s, 2H), 4.39 (s, 2H), 4.00 (t, J = 5 Hz, 4H), 3.52 (m, 4H), 1.45 (s, 18H). ¹³C NMR (75 MHz, CDCl₃): 159.8, 155.5, 139.9, 107.8, 101.4, 79.8, 67.3, 40.0, 33.3, 28.4. Elemental analysis (%) calcd for C₂₁H₃₃BrN₂O₆: C 51.54, H 6.80, N 5.72; found: C 51.27, H 6.59, N 5.57.

Compound G1N₃. A solution of **4** (0.92 g, 1.89 mmol) and NaN₃ (0.247 g, 3.8 mmol) in DMF (20 mL) was stirred at room temperature. After 12 h, H₂O was added to the mixture. The resulting aqueous layer was extracted with Et₂O (3 X). The combined organic layers were dried (MgSO₄), filtered and evaporated. Column chromatography (SiO₂, cyclohexane/AcOEt 7:3) gave **G1N₃** (0.69 g, 81%) as a white solid. IR (neat): 3345 (N-H), 2098 (N₃), 1680 (C=O). ¹H NMR (300 MHz, CDCl₃): 6.46 (d, J = 2 Hz, 2H), 6.41 (t, J = 2 Hz, 1H), 4.96 (broad s, 2H), 4.26 (s, 2H), 4.01 (t, J = 5 Hz, 4H), 3.52 (m, 4H), 1.45 (s, 18H). ¹³C NMR (75 MHz, CDCl₃): 160.2, 156.0, 138.0, 107.0, 101.3, 79.7, 67.5, 54.9, 40.2, 28.5. Elemental analysis (%) calcd for C₂₁H₃₃N₅O₆: C 55.86, H 7.37, N 15.51; found: C 55.50, H 7.09, N 15.11.

Preparation of compound G2N₃.



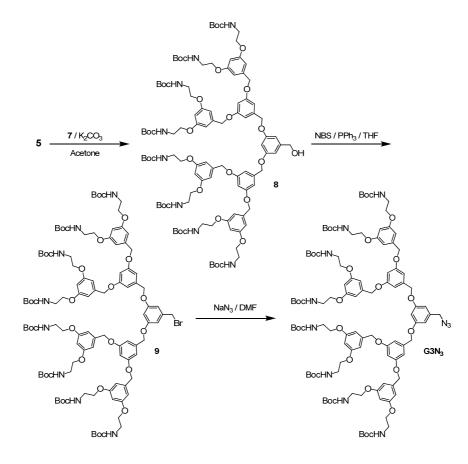
Compound 6. A mixture of **4** (3.24 g, 6.63 mmol), **5** (0.40 g, 3.87 mmol) and K₂CO₃ (1.59 g, 11.54 mmol) in acetone (55 mL) was heated under reflux. After 2 days, the mixture was filtered and evaporated. Column chromatography (SiO₂, CH₂Cl₂/AcOEt 8:2) gave **6** (2.67 g, 97%) as a white product. IR (neat): 3346 (O-H/N-H), 1686 (C=O). ¹H NMR (300 MHz, CDCl₃): 6.61 (d, J = 2 Hz, 2H), 6.56 (d, J = 2 Hz, 4H), 6.50 (t, J = 2 Hz, 1H), 6.39 (t, J = 2 Hz, 2H), 5.00 (broad s, 4H), 4.97 (s, 4H), 4.64 (d, 2H), 4.00 (t, J = 5 Hz, 8H), 3.51 (m, 8H), 1.45 (s, 36H). Elemental analysis (%) calcd for C₄₉H₇₂N₄O₁₅: C 61.49, H 7.58, N 5.85; found: C 61.60, H 7.66, N 5.70.

Compound 7. As described for **4** from **6** (2.68 g, 2.80 mmol), PPh₃ (0.95 g, 3.63 mmol) and *N*-bromosuccinimide (0.65 g, 3.63 mmol) in THF (25 mL). Column chromatography (SiO₂, cyclohexane/AcOEt 7:3) gave **7** (2.08 g, 73%) as a white solid. IR (neat): 3347 (N-H), 1691 (C=O). ¹H NMR (300 MHz, CDCl₃): 6.63 (d, J = 2 Hz, 2H), 6.56 (d, J = 2 Hz, 4H), 6.52 (t, J = 2 Hz, 1H), 6.40 (t, J = 2 Hz, 2H), 4.98 (broad s, 4H), 4.96 (s, 4H), 4.41 (s, 2H), 4.01 (t, J = 5 Hz, 8H), 3.52 (m, 8H), 1.45 (s, 36H). Elemental analysis (%) calcd for C₄₉H₇₁BrN₄O₁₄.0.6 CH₂Cl₂: C 55.69, H 6.81, N 5.24 ; found: C 55.76, H 6.86, N 5.25

Compound G2N₃. As described for **G1N₃** from 7 (600 mg, 0.59 mmol) and NaN₃ (76 mg, 1.17 mmol) in DMF (10 mL). Column chromatography (SiO₂, cyclohexane/AcOEt 7:3) gave **G2N₃** (521 mg, 90%) as a white solid. IR (neat): 3347 (N-H), 2100 (N₃), 1694 (C=O). ¹H NMR (300 MHz, CDCl₃): 6.55-6.57 (m,7H), 6.40 (t, J = 2 Hz, 2H), 4.99 (broad s, 4H), 4.97

(s, 4H), 4.27 (s, 2H), 4.01 (t, J = 5 Hz, 8H), 3.52 (m, 8H), 1.45 (s, 36H). ¹³C NMR (75 MHz, CDCl₃): 160.2, 160.1, 156.0, 139.3, 137.8, 107.4, 106.1, 101.9, 101.0, 79.6, 70.0, 67.4, 54.9, 40.2, 28.5. Elemental analysis (%) calcd for C₄₉H₇₁N₇O₁₄: C 59.92, H 7.29, N 9.98; found: C 59.93, H 7.34, N 9.68.

Preparation of compound G3N₃.

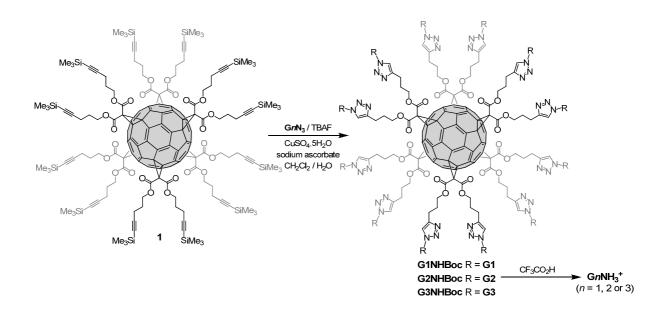


Compound 8. As described for **6** from **7** (1.47 g, 1.44 mmol), **5** (88 mg, 0.63 mmol) and K₂CO₃ (0.346 g, 2.51 mmol) in acetone (25 mL). Column chromatography (SiO₂, CH₂Cl₂/AcOEt 8:2) gave **8** (1.33 g, quantitative) as a white solid. IR (neat): 3341 (O-H/N-H), 1694 (C=O). ¹H NMR (300 MHz, CDCl₃): 6.64 (d, J = 2 Hz, 4H), 6.59 (d, J = 2 Hz, 2H), 6.55 (d, J = 2 Hz, 8H), 6.52 (t, J = 2 Hz, 2H), 6.49 (t, J = 2 Hz, 1H), 6.38 (t, J = 2 Hz, 4H), 5.04 (broad s, 8H), 4.98 (s, 4H), 4.95 (s, 8H), 4.61 (d, 2H), 3.98 (t, J = 5 Hz, 16H), 3.50 (m, 16H), 1.44 (s, 72H). ¹³C NMR (75 MHz, CDCl₃): 160.1, 156.0, 143.9, 139.6, 139.5, 106.5, 106.2, 105.9, 101.7, 101.3, 101.0, 79.7, 70.1, 70.0, 67.4, 65.1, 40.2, 28.5. Elemental analysis (%) calcd for C₁₀₅H₁₄₈N₈O₃₁. 0.5 CH₂Cl₂: C 61.49, H 7.29, N 5.44; found: C 61.47, H 7.48, N 5.57.

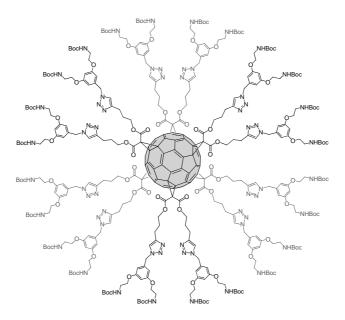
Compound 9. As described for **4** from from **8** (1.27 g, 0.63 mmol), PPh₃ (0.25 g, 0.94 mmol) and *N*-bromosuccinimide (0.17 g, 0.94 mmol) in THF (11 mL). Column chromatography (SiO₂, cyclohexane/AcOEt 1:1) gave **9** (0.87 g, 65%) as a white solid. IR (neat) 3356 (N-H), 1694 (C=O). ¹H NMR (300 MHz, CDCl₃): 6.66 (d, J = 2 Hz, 4H), 6.63 (d, J = 2 Hz, 2H), 6.56-6.52 (m, 11H), 6.39 (t, J = 2 Hz, 4H), 5.03 (broad s, 8H), 4.97 (s, 4H), 4.95 (s, 8H), 4.41 (s, 2H), 3.99 (t, J = 5 Hz, 16H), 3.51 (m, 16H), 1.44 (s, 72H). ¹³C NMR (75 MHz, CDCl₃): 160.1, 156.0, 139.9, 139.4, 139.2, 108.4, 106.6, 106.1, 102.4, 101.7, 100.9, 79.7, 70.2, 70.0, 67.4, 40.2, 33.7, 28.5. Elemental analysis (%) calcd for C₁₀₅H₁₄₇BrN₈O₃₀: C 60.60, H 7.12, N 5.38; found: C 60.23, H 7.15, N 5.21.

Compound G3N₃. As described for **G1N₃** from **9** (871 mg, 0.42 mmol) and NaN₃ (54 mg, 0.84 mmol) in DMF (10 mL). Column chromatography (SiO₂, cyclohexane/AcOEt; 1:1) gave **G3N₃** (750 mg, 87%) as a white solid. IR (neat): 3352 (N-H), 2100 (N₃), 1694 (C=O). ¹H NMR (300 MHz, CDCl₃): 6.66 (d, J = 2 Hz, 4H), 6.56-6.53 (m, 13H), 6.39 (t, J = 2 Hz, 4H), 5.02 (broad s, 8H), 4.98 (s, 4H), 4.95 (s, 8H), 4.26 (s, 2H), 3.99 (t, J = 5 Hz, 16H), 3.51 (m, 16H), 1.45 (s, 72H). ¹³C NMR (75 MHz, CDCl₃): 160.3, 160.2, 160.1, 156.0, 139.5, 139.3, 137.8, 107.4, 106.5, 106.2, 102.1, 101.7, 101.0, 79.7, 70.2, 70.0, 67.4, 54.9, 40.2, 28.5. Elemental analysis (%) calcd for C₁₀₅H₁₄₇N₁₁O₃₀: C 61.72, H 7.25, N 7.54; found: C 61.42, H 7.43, N 7.31.

Preparation of GnNHBoc and GnNH₃⁺.



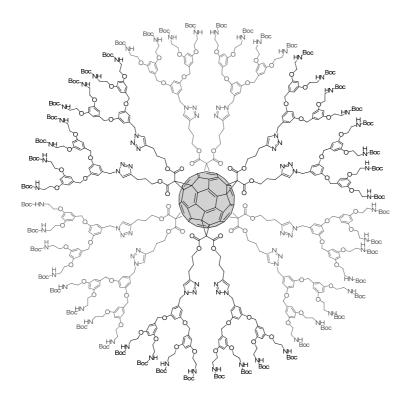
Compound G1NHBoc.



A 1 M TBAF solution in THF (0.7 mL, 0.7 mmol) was added to a mixture of **1** (51 mg, 0.02 mmol), **G1N₃** (294 mg, 0.65 mmol), CuSO₄.5H₂O (0.8 mg, 0.005 mmol) and sodium ascorbate (2.3 mg, 0.015 mmol) in CH₂Cl₂/H₂O (1:1, 2.5 mL). Column chromatography (SiO₂, CH₂Cl₂ containing 5% methanol) followed by gel permeation chromatography (Biobeads SX-1, CH₂Cl₂) gave **G1NHBoc** (109 mg, 85%) as an orange glassy product. IR (neat): 3348 (N-H), 1743 (C=O ester), 1695 (C=O Boc). UV/Vis (CH₂Cl₂): 229 (253200), 278

(121100), 318 (sh, 53600), 334 (sh, 41700). ¹H NMR (300 MHz, CDCl₃): 7.52 (s, 12H), 6.37 (m, 36H), 5.38 (broad s, 24H), 5.20 (broad s, 24H), 4.30 (m, 24H), 3.92 (m, 48H), 3.46 (m, 48H), 2.74 (m, 24H), 2.07 (m, 24H), 1.43 (s, 216H). ¹³C NMR (75 MHz, CDCl₃): 163.7, 160.2, 155.9, 146.7, 145.7, 141.2, 137.4, 121.6, 107.0, 101.0, 79.5, 69.2, 67.3, 66.4, 53.9, 45.6, 40.0, 28.5, 28.0, 22.1. Elemental analysis (%) calcd for $C_{390}H_{480}N_{60}O_{96.6}$ CH₂Cl₂: C 59.06, H 6.16, N 10.43; found: C 58.90, H 6.08, N 9.35.

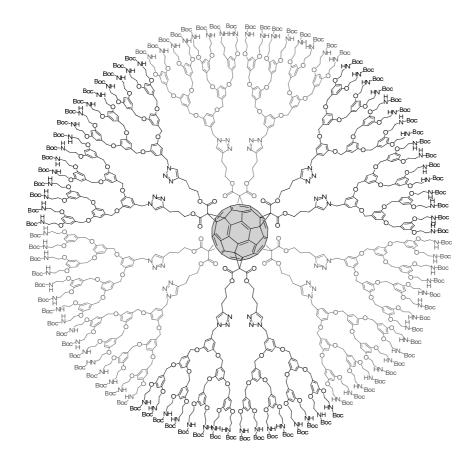
Compound G2NHBoc.



As described for **G1NHBoc** from **1** (85 mg, 0.03 mmol), **G2N₃** (550 mg, 0.04 mmol), CuSO₄.5H₂O (0.5 mg, 0.0028 mmol), sodium ascorbate (2.3 mg, 0.008 mmol) and TBAF (0.4 mL, 0.4 mmol) in CH₂Cl₂/H₂O (1:1, 3 mL). Column chromatography (SiO₂, CH₂Cl₂ containing 3% methanol) followed by gel permeation chromatography (Biobeads SX-1, CH₂Cl₂) gave **G2NHBoc** (270 mg, 68%) as an orange glassy product. IR (neat): 3350 (N-H), 1740 (C=O ester), 1694 (C=O Boc). UV/Vis (CH₂Cl₂): 230 (405500), 281 (147600), 318 (sh, 44700), 334 (sh, 35000). ¹H NMR (300 MHz, CDCl₃): 7.44 (broad s, 12H), 6.44 (m, 84H), 6.32 (m, 24H), 5.27-5.32 (m, 72H), 4.78 (s, 48H), 4.28 (m, 24H), 3.91 (m, 48H), 3.45 (m, 48H), 2.70 (m, 24H), 2.04 (m, 24H), 1.41 (s, 432H). ¹³C NMR (100 MHz, CDCl₃): 163.6, 160.2, 159.9, 156.0, 147.1, 145.8, 141.2, 139.0, 137.3, 121.6, 107.2, 106.1, 101.9, 100.7, 79.4,

69.9, 69.2, 67.2, 66.4, 53.8, 45.5, 40.0, 28.4, 28.1, 22.0. Elemental analysis (%) calcd for C₇₂₆H₉₃₆N₈₄O₁₉₂.2 CHCl₃: C 61.79, H 6.68, N 8.31; found: C 61.40, H 6.80, N 8.13.

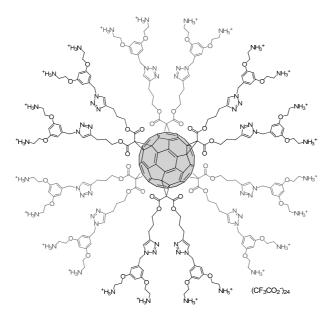
Compound G3NHBoc.



As described for **G1NHBoc** from **1** (57 mg, 0.02 mmol), **G3N₃** (550 mg, 0.27 mmol), CuSO₄.5H₂O (0.4 mg, 0.0028 mmol), sodium ascorbate (1.1 mg, 0.006 mmol) and TBAF (0.3 mL, 0.3 mmol) in CH₂Cl₂/H₂O (1:1, 1 mL). Column chromatography (SiO₂, CH₂Cl₂) followed by gel permeation chromatography (Biobeads SX-1, CH₂Cl₂) gave **G3NHBoc** (451 mg, 89%) as an orange glassy product. IR (neat) 3363 (N-H), 1745 (C=O ester), 1695 (C=O Boc). UV/Vis (CH₂Cl₂): 231 (706500), 280 (228600), 318 (sh, 43100), 334 (sh, 33100). ¹H NMR (300 MHz, CDCl₃): 7.59 (broad s, 12H), 6.51 (m, 48H), 6.44 (m, 156H), 6.30 (m, 48H), 5.30 (s, 96H), 5.28 (s, broad, 96H), 4.76 (m, 48H), 4.70 (m, 24H), 4.33 (m, 24H), 3.88 (m, 192H), 3.42 (m, 192H), 2.74 (m, 24H), 2.11 (m, 24H), 1.40 (broad s, 864H). ¹³C NMR (75 MHz, CDCl₃): 163.7, 160.3, 160.1, 159.9, 156.0, 146.9, 145.9, 141.1, 139.3, 139.0, 137.6, 121.8, 107.1, 106.4, 106.1, 101.7, 101.5, 100.6, 79.4, 69.7, 69.2, 67.2, 66.4, 53.8, 45.3, 43.5,

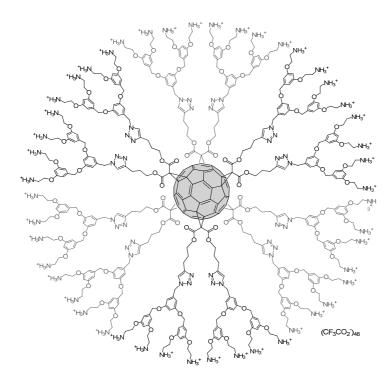
40.1, 28.5, 28.1, 22.2. Elemental analysis (%) calcd for C₁₃₉₈H₁₈₄₈N₁₃₂O₃₈₄: C 63.01, H 6.99, N 6.94; found: C 62.81, H 6.96, N 6.48.

Compound G1NH₃⁺.



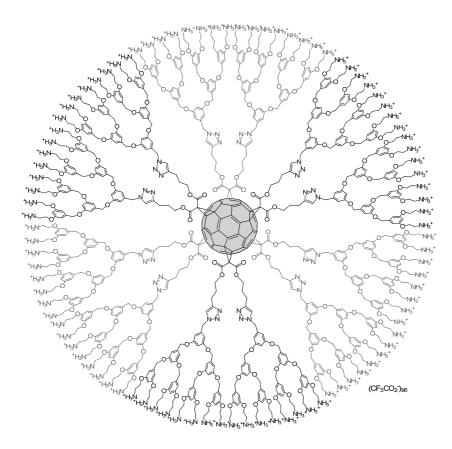
G1NHBoc (93 mg, 0.012 mmol) was dissolved in TFA (2 mL). After 4 h, the mixture was evaporated and dried under vacuum to afford **G1NH₃⁺** as its trifluoroacetate salt (98 mg, quantitative). Orange glassy product. IR (neat): 1742 (C=O ester), 1671 (C=O TFA). UV/Vis (H₂O): 201 (576000), 275 (91500), 314 (sh, 45300), 332 (sh, 36200). ¹H NMR (300 MHz, D₂O): 7.67 (m, 12H), 6.52 (m, 12H), 6.46 (m, 24H), 5.35 (m, 24H), 4.12 (m, 72H), 3.33 (m, 48H), 2.57 (m, 24H), 1.84 (m, 24H). ¹³C NMR (100 MHz, D₂O): 164.1, 162.4 (q, *J* = 36 Hz), 159.2, 146.1, 145.4, 141.3, 136.9, 123.7, 116.4 (q, *J* = 290 Hz), 107.6, 101.7, 69.2, 66.8, 64.1, 54.0, 46.3, 38.8, 27.0, 21.0.

Compound G2NH₃⁺.

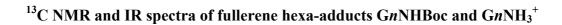


G2NHBoc (190 mg, 0.014 mmol) was dissolved in TFA (2 mL). After 4 h, the mixture was evaporated and dried under vacuum to afford **G2NH₃**⁺ as its trifluoroacetate salt (224 mg, quantitative). Orange glassy product. IR (neat): 1736 (C=O ester), 1671 (C=O TFA). UV/Vis (H₂O): 203 (1238300), 280 (136600), 318 (sh, 40100), 334 (sh, 31900). ¹H NMR (300 MHz, D₂O): 7.36 (m, 12H), 6.48-6.45 (m, 84H), 6.36 (m, 24H), 5.22 (m, 24H), 4.64 (m, 72H), 4.03 (m, 96H), 3.28 (m, 96H), 2.39 (m, 24H), 1.65 (m, 24H). ¹³C NMR (100 MHz, D₂O) δ 163.7, 162.4 (q, *J* = 34 Hz), 160.2, 159.7, 159.0, 146.5, 145.4, 141.4, 139.1, 137.4, 122.8, 116.5 (q, *J* = 290 Hz), 107.1, 106.6, 101.7, 101.2, 66.6, 63.9, 53.6, 48.9, 46.4, 38.8, 27.1, 21.1.

Compound G3NH₃⁺.



G3NHBoc (251 mg, 0.009 mmol) was dissolved in TFA (2 mL). After 4 h, the mixture was evaporated and dried under vacuum to afford **G3NH₃**⁺ as its trifluoroacetate salt (264 mg, quantitative). Orange glassy product. IR (neat) 1738 (C=O ester), 1671 (C=O TFA). UV/Vis (H₂O): 203 (2779600), 280 (247200), 318 (sh, 41400), 334 (sh, 29100). ¹H NMR (300 MHz, D₂O): 7.59 (broad s, 12H), 6.50-6.10 (m, 252H), 4.80 (broad s, 24H), 4.70 (broad s, 144H), 3.85 (m, 216H), 3.2 (m, 192H), 2.30 (m, 24H), 1.65 (m, 24H). ¹³C NMR (100 MHz, D₂O): 163.6, 162.4 (q, *J* = 35 Hz), 159.8, 159.4, 158.9, 146.4, 145.2, 141.5, 139.9, 138.9, 137.4, 122.5, 116.5 (q, *J* = 292 Hz), 106.1, 106.1, 106.1, 101.1, 101.1, 69.4, 69.0, 66.7, 65.5, 63.8, 53.4, 48.9, 38.8, 27.4, 21.0.



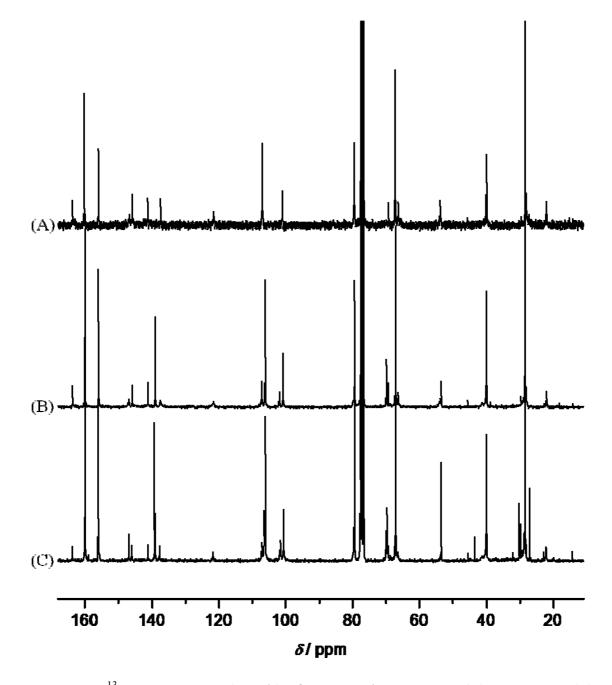


Figure S1A. ¹³C NMR spectra (CDCl₃) of compounds **G1NHBoc** (A), **G2NHBoc** (B) and **G3NHBoc** (C).

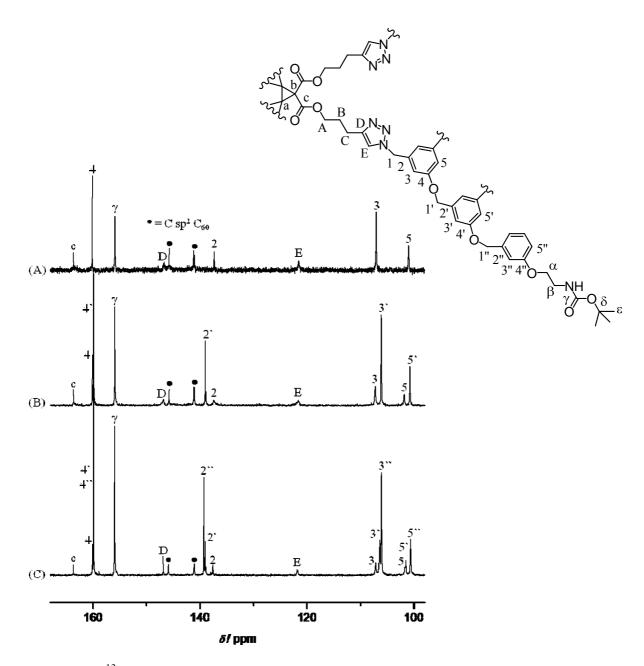


Figure S1B. ¹³C NMR spectra (CDCl₃) of compounds **G1NHBoc** (A), **G2NHBoc** (B) and **G3NHBoc** (C).

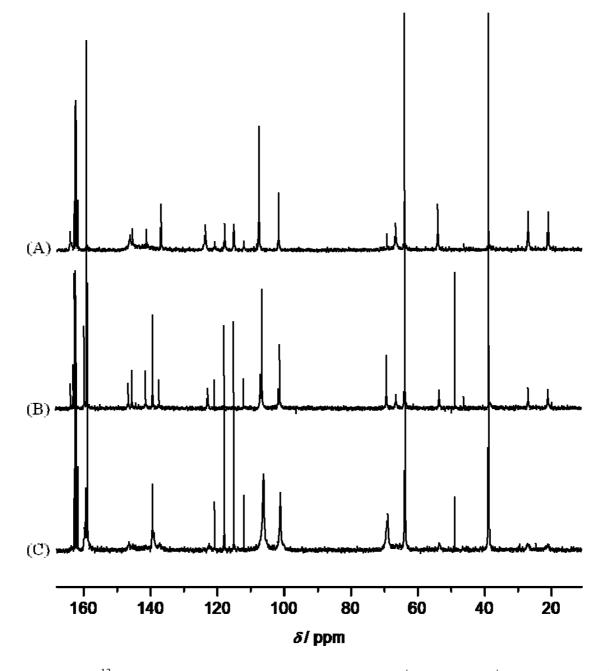


Figure S2A. ¹³C NMR spectra (D₂O) of compounds $G1NH_3^+$ (A), $G2NH_3^+$ (B) and $G3NH_3^+$ (C) (\diamond = residual CH₃OH).

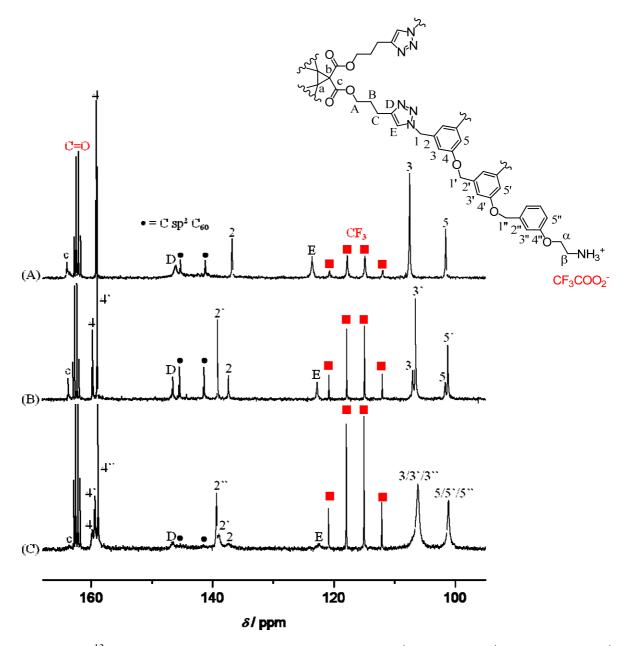


Figure S2B. ¹³C NMR spectra (D₂O) of compounds $G1NH_3^+$ (A), $G2NH_3^+$ (B) and $G3NH_3^+$ (C).

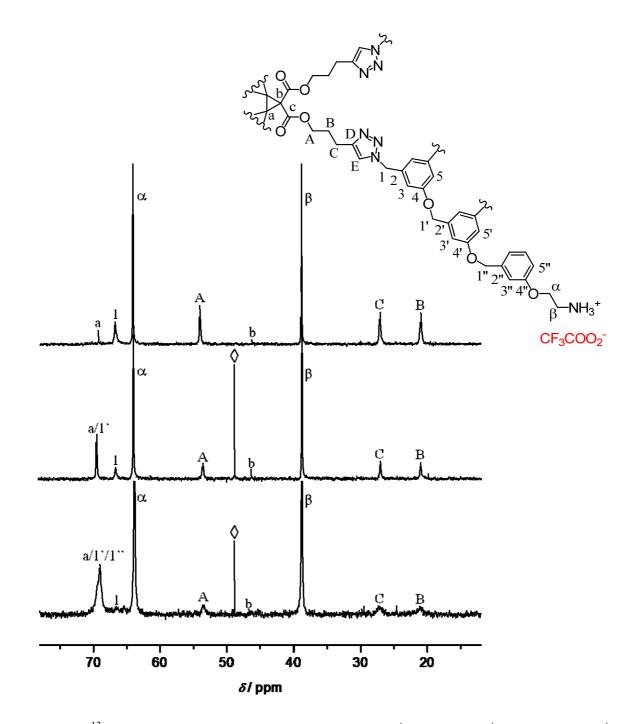


Figure S2C. ¹³C NMR spectra (D₂O) of compounds $G1NH_3^+$ (A), $G2NH_3^+$ (B) and $G3NH_3^+$ (C) (\diamond = residual CH₃OH).

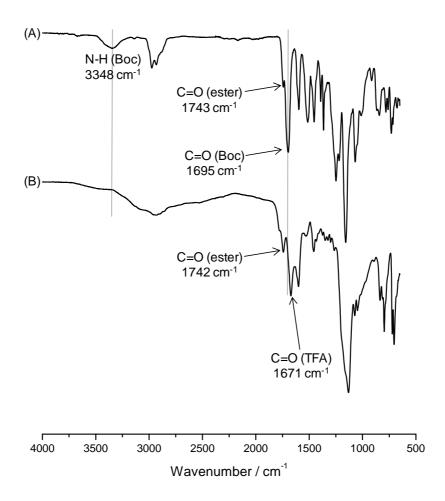


Figure S3. IR spectra of compounds G1NHBoc (A) and G1NH $_3^+$ (B).

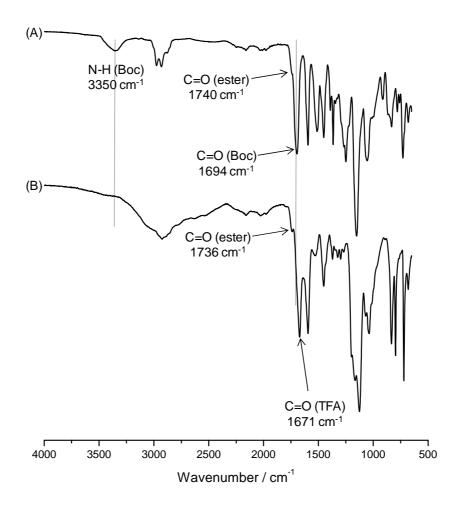


Figure S4. IR spectra of compounds G2NHBoc (A) and G2NH $_3^+$ (B).

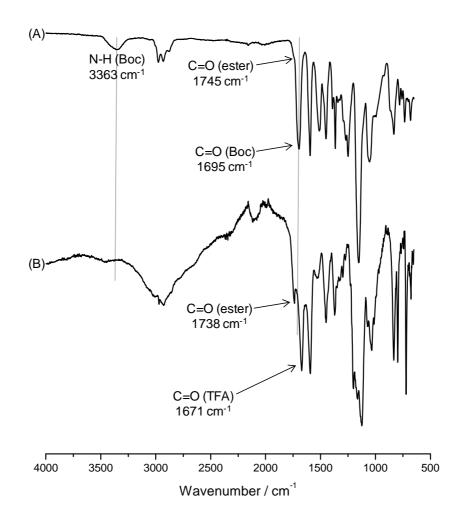


Figure S5. IR spectra of compounds G3NHBoc (A) and $G3NH_3^+$ (B).

GnNH₃⁺/pCMVLuc polyplexes

Agarose gel electrophoresis analysis.

Twenty microliters of water containing 5% glucose or 150 mM NaCl, 0.4 μ g pCMVLuc and increasing amounts of cationic vectors **GnNH**₃⁺ were subjected (30 min of complexation time) to electrophoresis in a 1% agarose gel containing 1 mM EDTA and 40 mM Tris acetate buffer and 0.5 μ g/mL ethidium bromide, for 90 min at 100 V. DNA was visualized with an UV transilluminator at 254 nm.

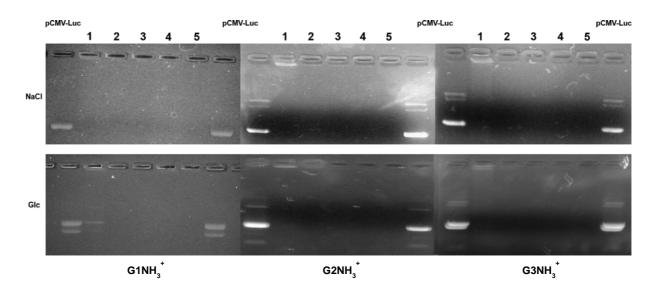


Figure S6. Agarose gels. Polyplexes were prepared at N/P = 1 to 5.

Size of particles, ς potential.

The hydrodynamic radii were determined *via* dynamic light scattering measurements using a Malvern nanoZS apparatus with the following specifications:

sampling time = 90 s; refractive index of medium (water with 5% glucose) = 1.340; refractive index of particles = 1.43; medium viscosity = 1.0140 cP temperature = 25° C. Data were analyzed using the multimodal number distribution software included with the instrument. ς potentials were measured with the same apparatus and with the following specifications: 20 measurements per sample; dielectric constant = 80; temperature = 25° C; beam mode F(Ka) = 1.5 (Smoluchowski model). Polyplexes (1 mL volume) were prepared as described for delivery experiment.

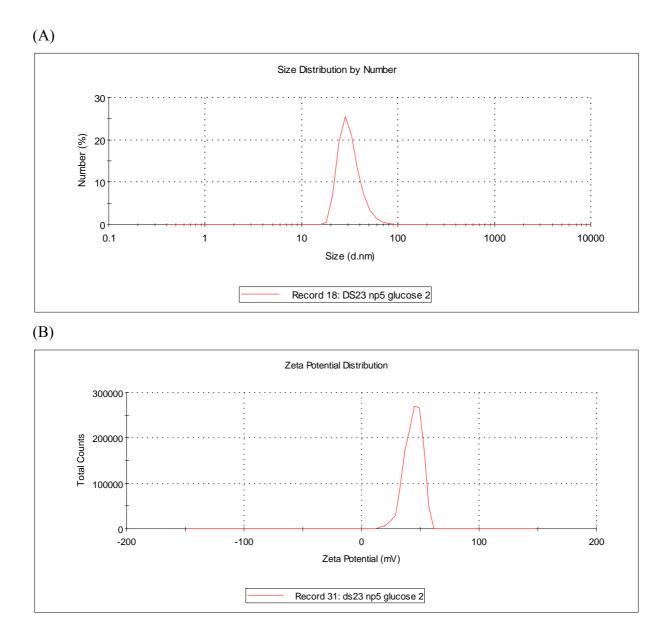


Figure S7. Size (A) and ς potential (B) of G1NH₃⁺/pCMVLuc polyplexes at N/P 5.

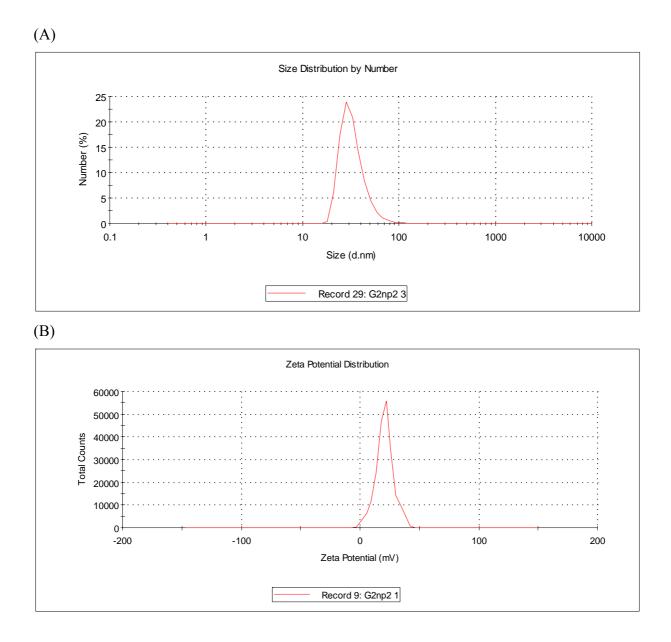


Figure S8. Size (A) and ς potential (B) of **G2NH**₃⁺/pCMVLuc polyplexes at N/P 2.

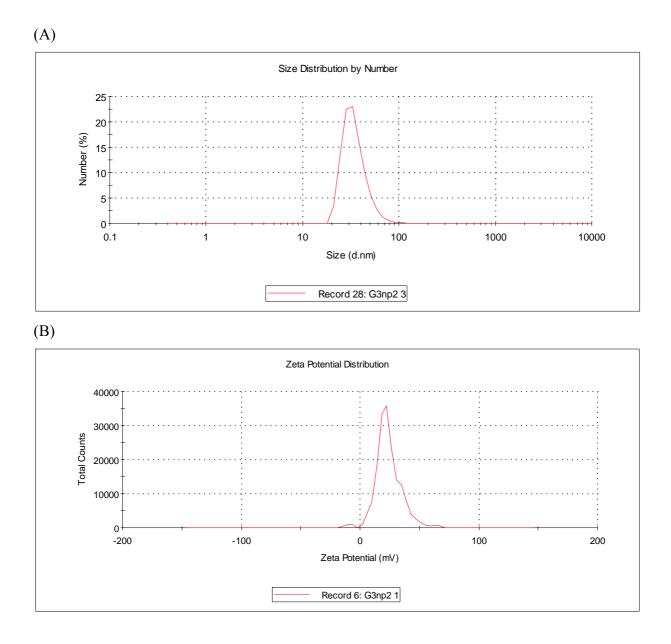


Figure S9. Size (A) and ς potential (B) of **G3NH**₃⁺/pCMVLuc polyplexes at N/P 2.

Electron microscopy analysis.

Images were taken with a TEM Philips CM12 apparatus. Polyplexes were transferred onto ultrathin carbon film grids (Ted Pella, 1822-F, formvar removed) by placing the grid on top of 10 μ L drop for 1 min. Grid with adherent particles was wicked from one side, placed on 100 μ L water drop for 30 s for washing, wicked, placed for 1 min on 60 μ L drop of freshly filtered 1.33% uranyle acetate, wicked again and air dried before imaging.

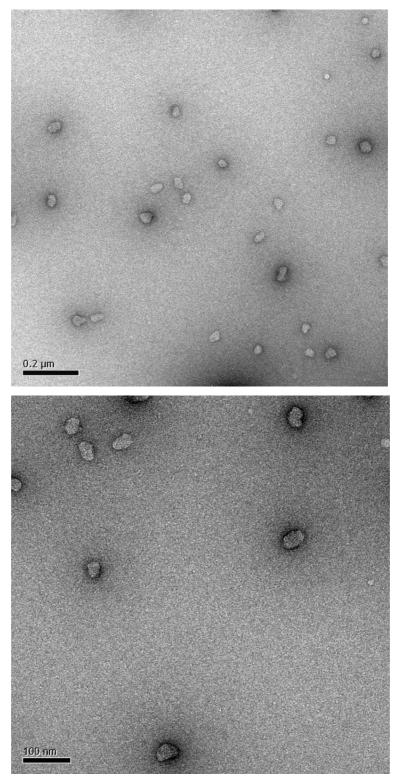


Figure S10. TEM images of $G1NH_3^+/pCMVLuc$ polyplexes at N/P 5 in water with 5% glucose.

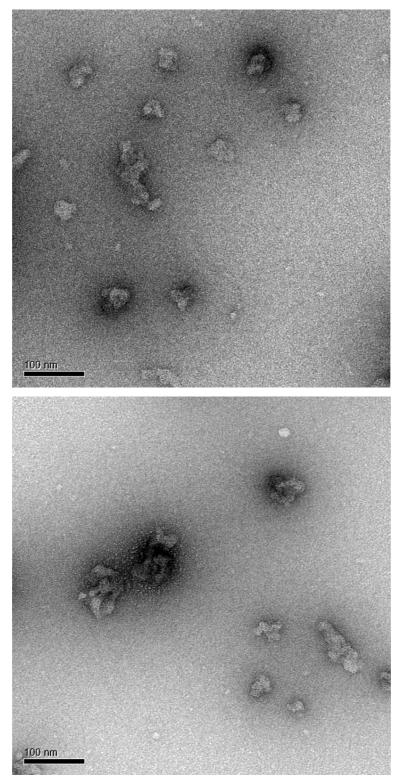


Figure S11. TEM images of $G2NH_3^+/pCMVLuc$ polyplexes at N/P 3 in water with 5% glucose.

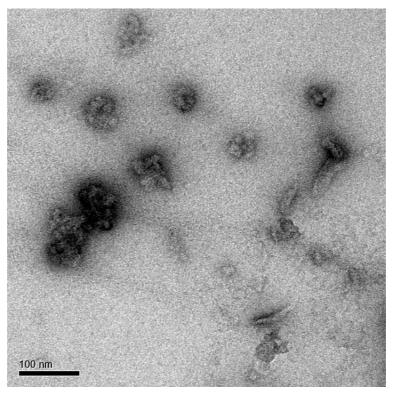


Figure S12. TEM image of $G3NH_3^+/pCMVLuc$ polyplexes at N/P 5 in water with 5% glucose.

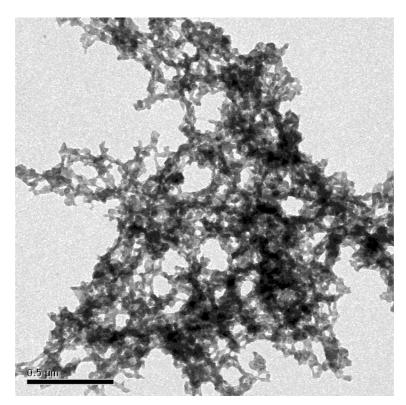


Figure S13. TEM image of G1NH₃⁺/pCMVLuc polyplexes at N/P 5 in water with 150 mM NaCl.

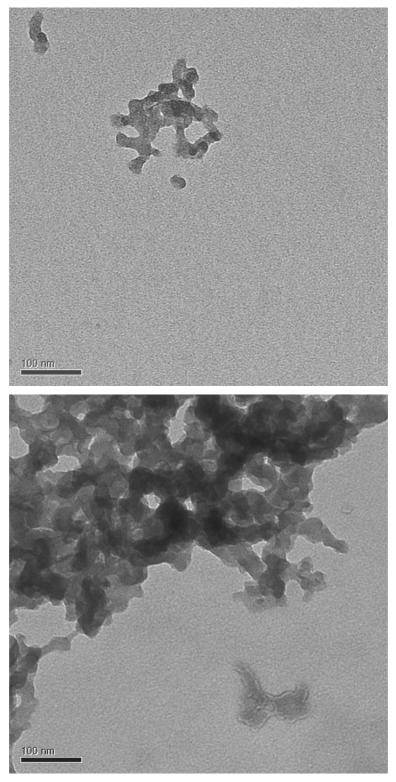


Figure S14. TEM images of **G2NH₃**⁺/pCMVLuc polyplexes at N/P 4 in water with 150 mM NaCl.

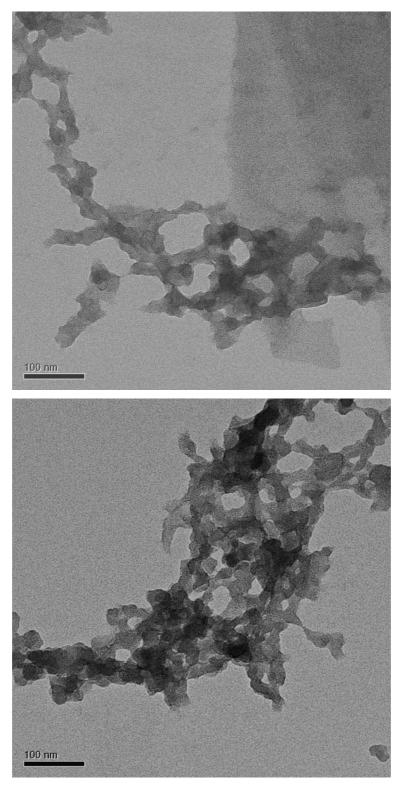


Figure S15. TEM images of G3NH₃⁺/pCMVLuc polyplexes at N/P 6 in water with 150 mM NaCl.

Gene transfer experiments.

Cell culture.

HeLa cells were grown in Eagle's MEM supplemented with 10% FBS, L-glutamine (2 mM), penicillin (100 units/mL) and streptomycin (100 μ g/mL). Cells were maintained at 37°C in a 5% CO₂ humidified atmosphere and all experiments were done in triplicates. The day before experiment, cells were seeded in 24-multiwell plates at 50.10³ cells/well in fresh complete medium (1 mL).

Polyplexes formation for pCMVLuc delivery.

The procedure is for a 24-multiwell plates experiment. Typically, an aqueous solution of the $GnNH_3^+$ (n = 1, 2 or 3) (volume depending on the desired N/P ratio) is diluted up to 50 µL in water containing 5% glucose or 150 mM NaCl. The solution is vortexed and left for 10 min. Separatly, an aqueous solution of pCMVLuc (corresponding to 2 µg of pCMVLuc) is diluted up to 50 µL in water containing 5% glucose or 150 mM NaCl. Then, the solution is vortexed and left for 10 min, after which the G₁, G₂ or G₃ solution is added to the pCMVLuc solution, and vigorously mixed (15 s). Finally, the polyplexes are incubated for 30 min at room temperature and added in each well by dilution with the cell medium without serum (1 mL). Four hours later, each well is completed with serum (0.1 mL). The gene expression profiles are analyzed 24 h after addition of polyplexes.

Quantification of the luciferase gene expression.

Luciferase gene expression was determined 24 h after delivery with a commercial kit, using manufacturer's protocol (Luciferase Assay System, Promega, Charbonnières, France). The luminescence was measured from 2 μ L of lysate during 1 s with a luminometer (Centro LB960 XS; Berthold, Thoiry, France). Luciferase activity is expressed as the mean of light units integrated over 10 s (RLU) and normalized per mg of cell protein by using the BCA assay (Pierce, Brebières, France). The errors bars represent standard deviation derived from triplicate experiments.

JetSITM-ENDO is a transfection Reagent (Polyplus transfection), which is used according to manufacturer's instructions.

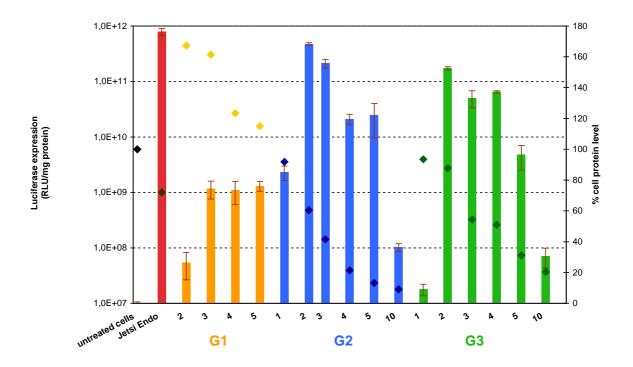


Figure S16. Delivery experiments of pCMVLuc at different N/P (Polyplexes prepared in 5% glucose solution).

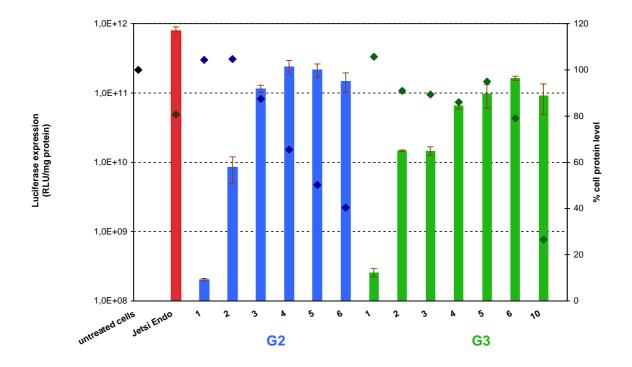


Figure S17. Delivery experiments of pCMVLuc at different N/P (Polyplexes prepared in 150 mM NaCl solution). N.B. $G1NH_3^+$ has no efficiency under these conditions.