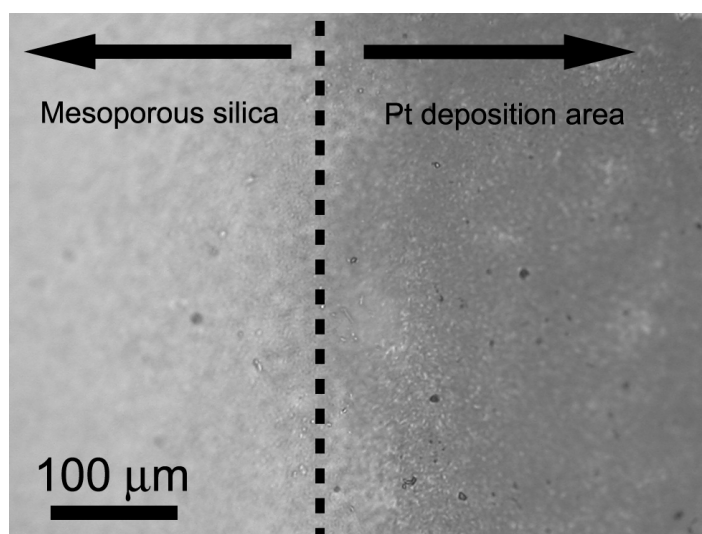


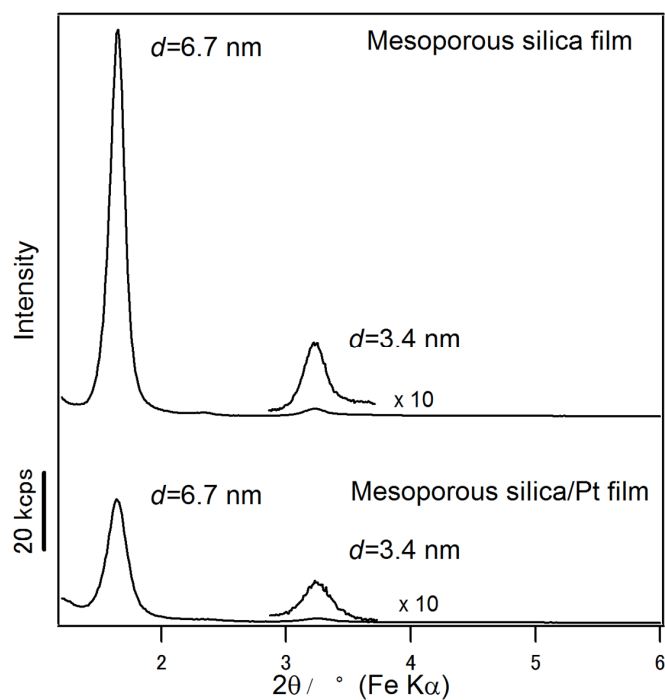
## Electronic Supplementary Information (Figure S1 to S10)

### Figure S1



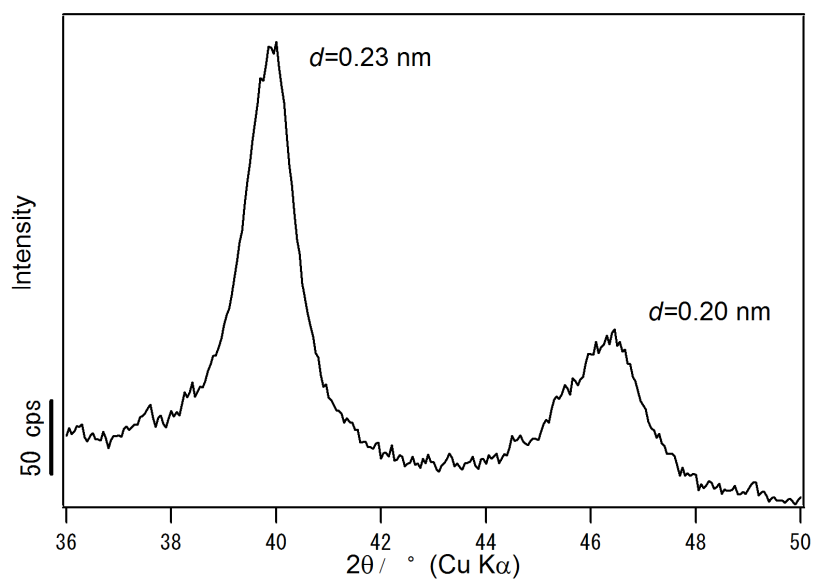
**Figure S1** Optical microscopic image of the mesoporous silica/Pt composite film before silica removal. The Pt electrodeposited area shows black color (see the right side region).

### Figure S2



**Figure S2** Low-angle XRD patterns of the mesoporous silica films before and after Pt electrodeposition.

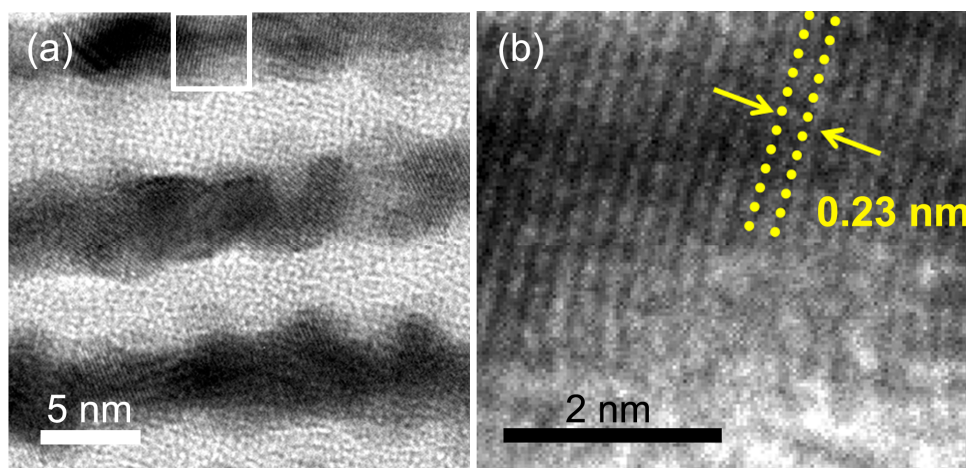
**Figure S3**



**Figure S3** Wide angle XRD pattern for the mesoporous silica/Pt composite film before silica removal.

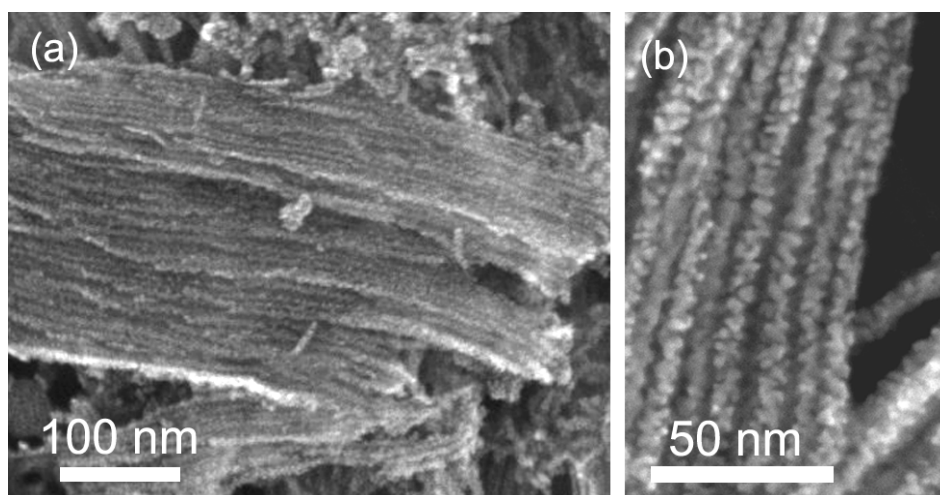
5

**Figure S4**



**Figure S4** TEM images of the mesoporous silica/Pt film before silica removal. Figure (b) is the enlarged image of the square area in Figure (a). The lattice fringes with 0.23 nm in  $d$ -spacing are attributable to the (111) plane of the Pt *fcc* structure.

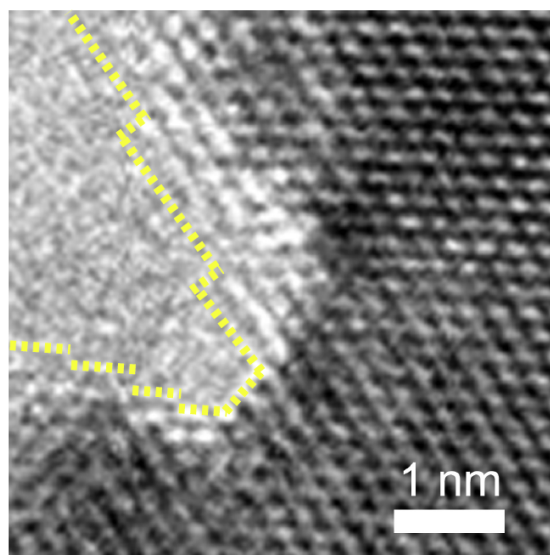
**Figure S5**



**Figure S5** (a) Low and (b) high magnified SEM images of Pt nanoworms after silica removal.

5

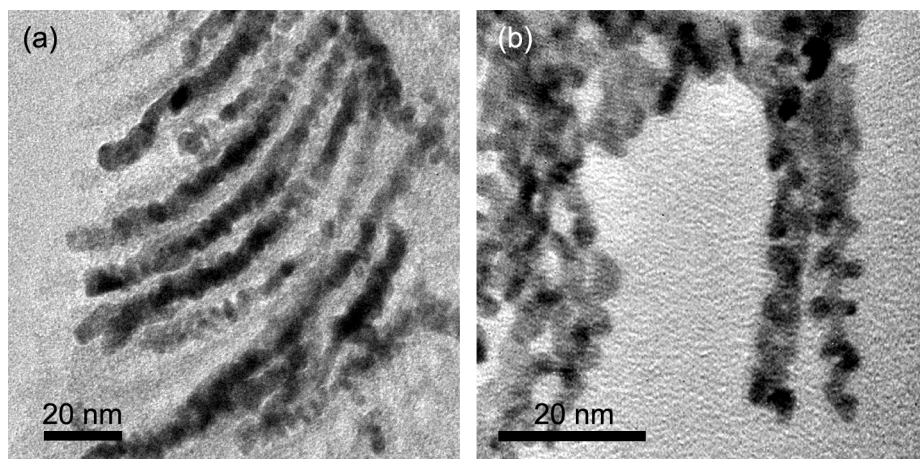
**Figure S6**



**Figure S6** Enlarged TEM image of Pt nanoworm displayed in **Figure 2b**. Rich atomic steps with a concave surface topology are indicated by yellow dot-lines.

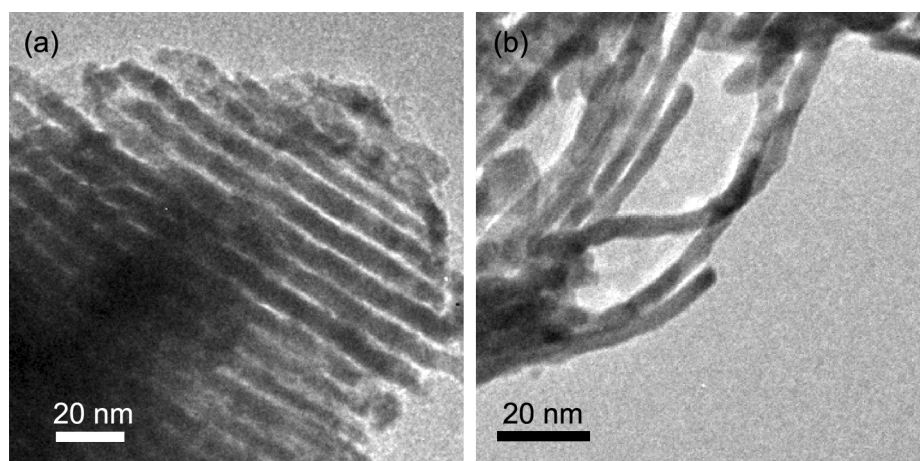
10

**Figure S7**



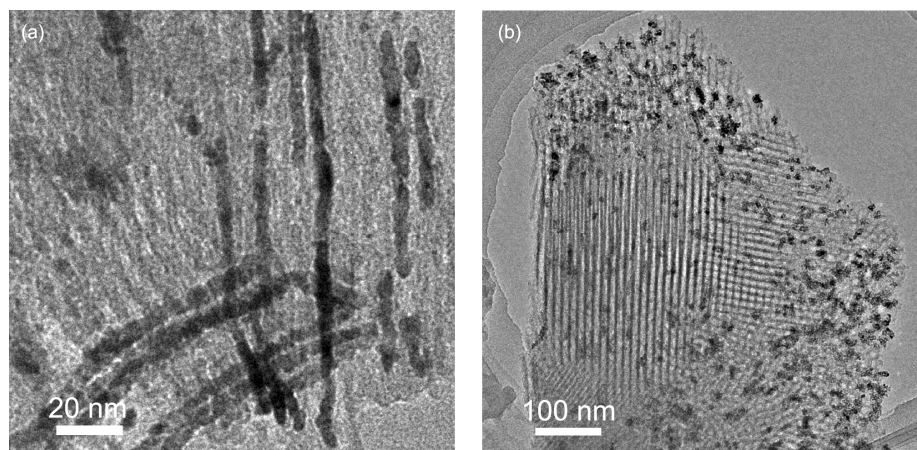
**Figure S7** TEM images of Pt nanoworms after sonicated strongly and annealed at 150 °C for 24 h. ((a) Pt nanoworms encapsulated by rigid silica walls (before silica removal), (b) Pt nanoworms after silica removal.)

**Figure S8**



**Figure S8** TEM image of (a) mesoporous silica/Pt composite and (b) Pt nanowires after removal of silica. The Pt nanowires are deposited from  $\text{H}_2\text{PtCl}_6 \cdot 6\text{H}_2\text{O}$  (50 wt%) aqueous solution without surfactants at -0.042 V vs. Ag/AgCl.

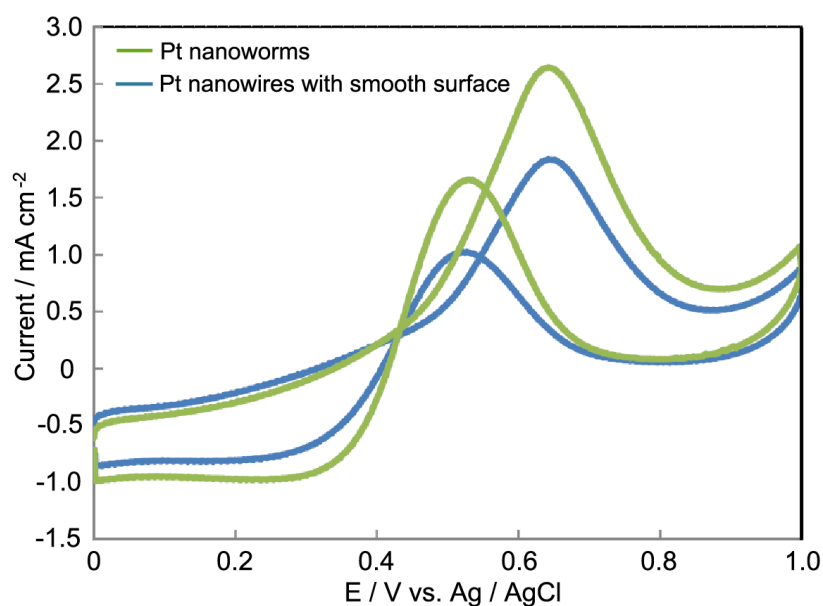
**Figure S9**



**Figure S9** TEM images of mesoporous silica/Pt composite film prepared with triblock copolymer (P123) instead of  $C_{16}EO_8$ .

5

**Figure S10**



**Figure S10** CV curves of methanol oxidation reaction. Cyclic voltammetry (CV) experiments were performed by using a conventional three-electrode cell, including an Ag/AgCl (saturated KCl) electrode as a reference electrode, a platinum wire as a counter electrode, and a modified glassy carbon electrode (GCE) (3 mm in diameter) as a working electrode. The bumpy or smooth nanowire suspension was dropped on the surface of the modified GCE and dried at room temperature. Then, methanol electro-oxidation measurements were performed in a solution of 0.5 M  $H_2SO_4$  that contained 15 0.5 M methanol at a scan rate of 50 mV/s.