

Supplementary Information

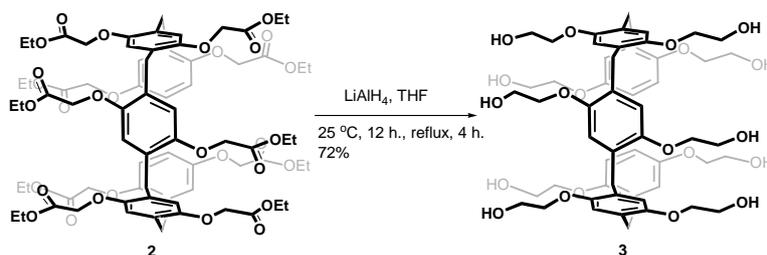
Pillar[5]arene decaamine: synthesis, encapsulation of very long linear diacids and formation of ion pair-stopped [2]rotaxanes

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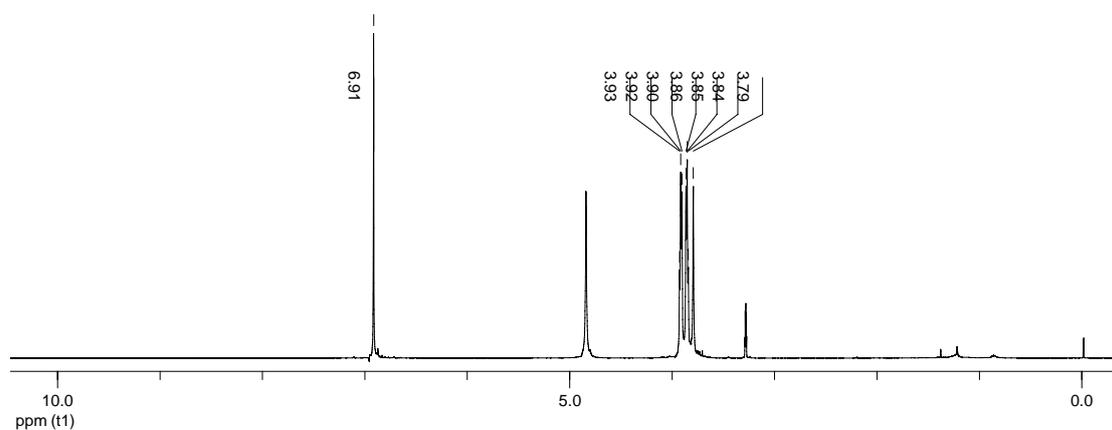
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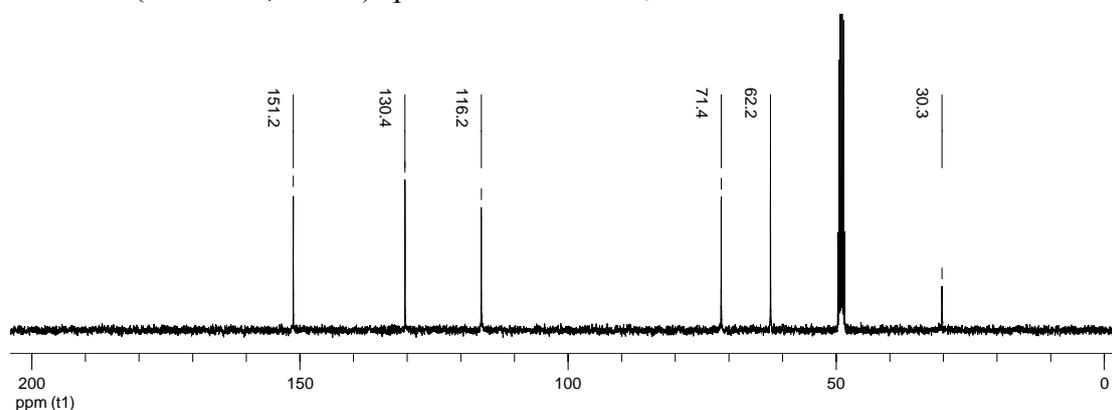
1. Synthesis and characterization of the compounds:



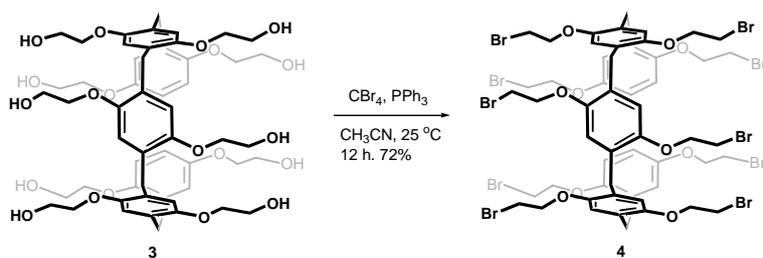
Compound 3: To a solution of compound **2**¹ (700 mg, 0.5 mmol) in anhydrous THF (30 mL) was added LiAlH_4 (380 mg, 10 mmol) at $0\text{ }^\circ\text{C}$ under nitrogen atmosphere. The reaction mixture was stirred at $25\text{ }^\circ\text{C}$ for 12 h. and then heated under reflux for 4 h. To the cooled reaction mixture was added 1.0 M aqueous NaOH dropwisely until the end of gas evolution. The resulting mixture was filtered, and the filtrate was concentrated *in vacuo* to give a white solid. The white solid was washed with water by sonication to produce compound **3** as a pure white solid (360mg, 72%). ^1H NMR ($\text{CH}_3\text{OH}-d_4$): δ 6.91 (s, 10 H), 3.92 (t, $J = 4.6$ Hz, 20 H), 3.85 (t, $J = 4.6$ Hz, 20 H), 3.79 (s, 10 H). ^{13}C NMR ($\text{CH}_3\text{OH}-d_4$): δ 151.2, 130.4, 116.2, 71.4, 62.2, 30.3. MS (ESI): m/z 1051 $[\text{M} + \text{H}]^+$, 1073 $[\text{M} + \text{Na}]^+$. HR-MS (ESI-TOF): Calcd. for $\text{C}_{55}\text{H}_{71}\text{O}_{20}$: 1051.4539. Found: 1051.4583.



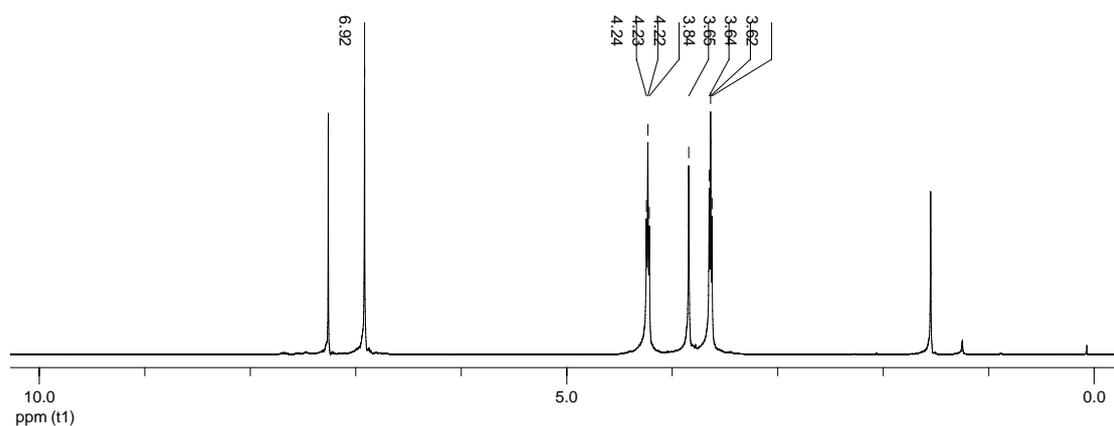
^1H NMR (400 MHz, 298 K) spectrum of **3** in $\text{CH}_3\text{OH}-d_4$.



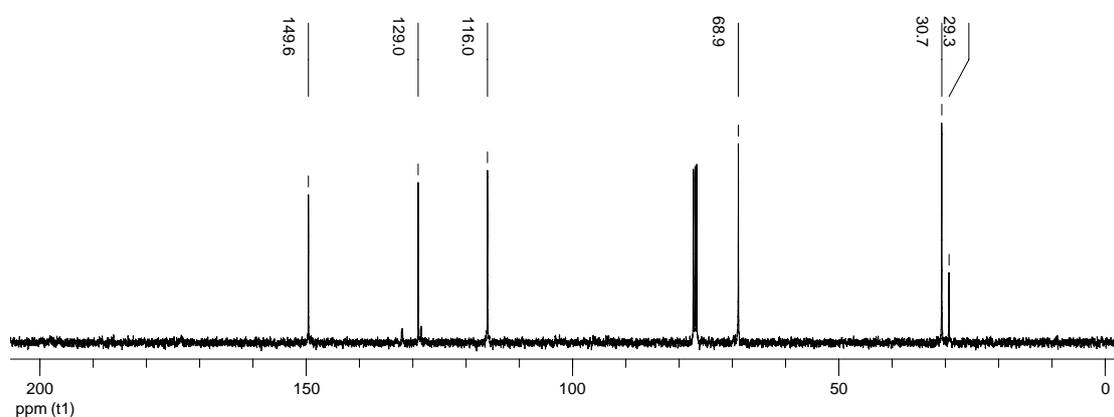
^{13}C NMR (400 MHz, 298 K) spectrum of **3** in $\text{CH}_3\text{OH}-d_4$.



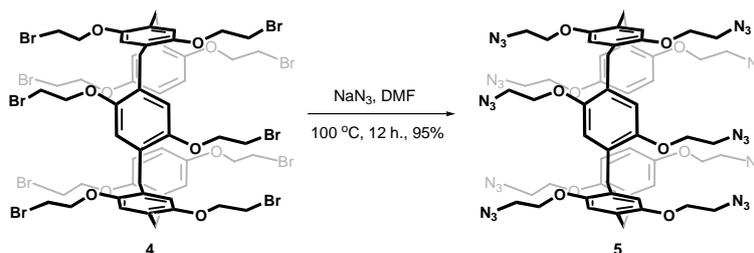
Compound 4: Carbontetrabromide (1.38 g, 4.16 mmol) was added slowly in small portions to a suspension of **3** (290 mg, 0.28 mmol) and triphenylphosphine (1.09 g, 4.16 mmol) in anhydrous acetonitrile (9 ml) at 0°C with stirring. The reaction mixture was allowed to warm to room temperature, and the resulting mixture was stirred for 12 h. under nitrogen atmosphere. Then 200 mL of water was added to the reaction mixture, whereupon a white solid was precipitated. The precipitate was collected by filtration, and thoroughly washed with methanol to give **4** as a white solid (338 mg, 72%). ^1H NMR (CDCl_3): δ 6.92 (s, 10 H), 4.23 (t, $J = 5.6$ Hz, 20 H), 3.84 (s, 10 H), 3.64 (t, $J = 5.6$ Hz, 20 H). ^{13}C NMR (CDCl_3): δ 149.6, 129.0, 116.0, 68.9, 30.7, 29.3. MS (ESI): m/z 1680 $[\text{M} + \text{H}]^+$. HR-MS (ESI-TOF): Calcd. for $\text{C}_{55}\text{H}_{61}\text{O}_{10}\text{Br}_{10}$: 1680.5996. Found: 1680.5972.



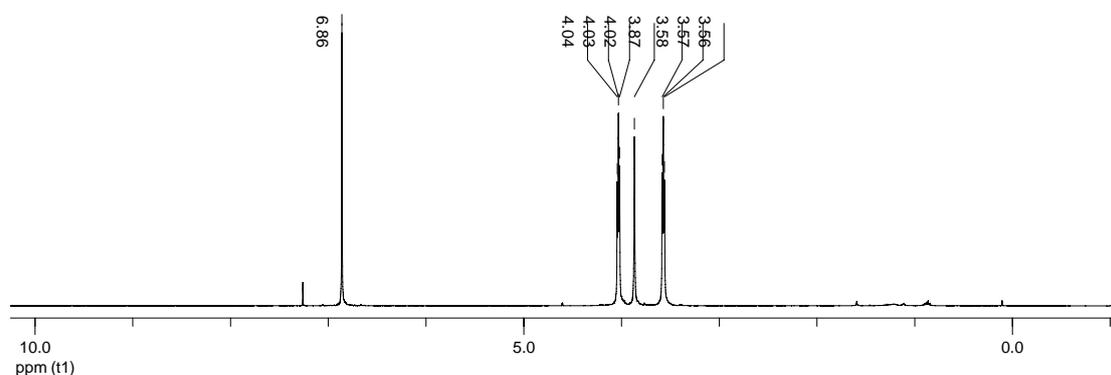
^1H NMR (400 MHz, 298 K) spectrum of **4** in CDCl_3 .



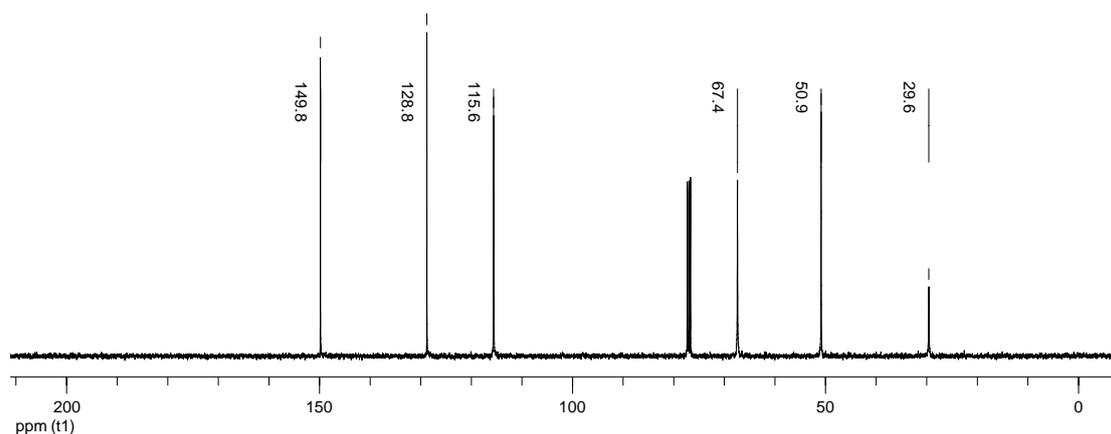
^{13}C NMR (400 MHz, 298 K) spectrum of **4** in CDCl_3 .



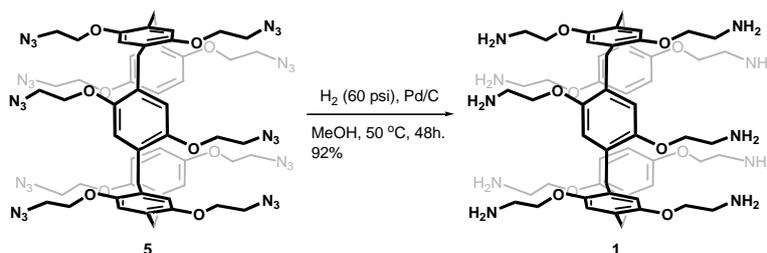
Compound 5: Sodium azide (193 mg, 3 mol) was added to the solution of **4** (100 mg, 0.06 mmol) in anhydrous DMF (5 ml). After stirring at 100 °C for 12 h. under nitrogen atmosphere, the mixture was cooled to room temperature and poured into water (80 ml). The precipitate was collected by filtration, and washed with water to yield **5** as white solid (74 mg, 95%). ¹H NMR (CDCl₃): δ 6.86 (s, 10 H), 4.03 (t, *J* = 4.8 Hz, 20 H), 3.87 (s, 10 H), 3.57 (t, *J* = 4.8 Hz, 20 H). ¹³C NMR (CDCl₃): δ 149.8, 128.8, 115.6, 67.4, 50.9, 29.6. MS (ESI): *m/z* 1323 [M + Na]⁺. HR-MS (ESI-TOF): Calcd. for C₅₅H₆₀O₁₀N₃₀Na: 1323.5006. Found: 1323.5013.



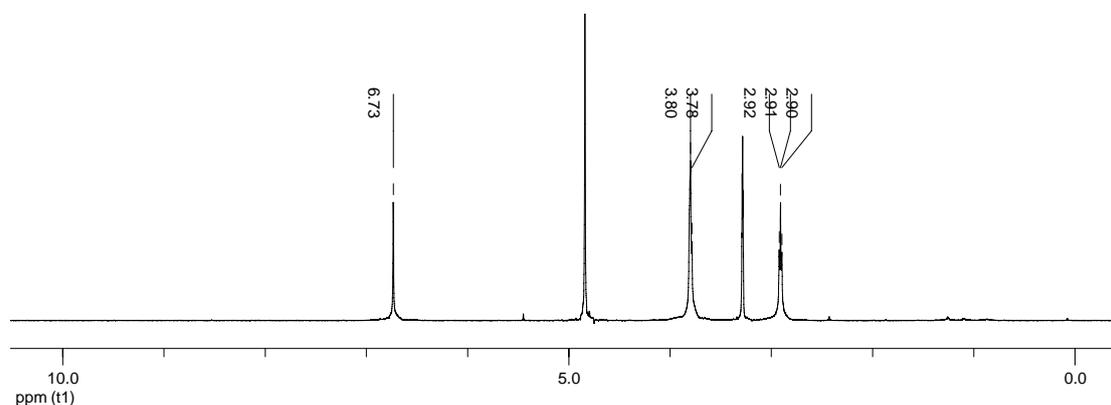
¹H NMR (400 MHz, 298 K) spectrum of **5** in CDCl₃.



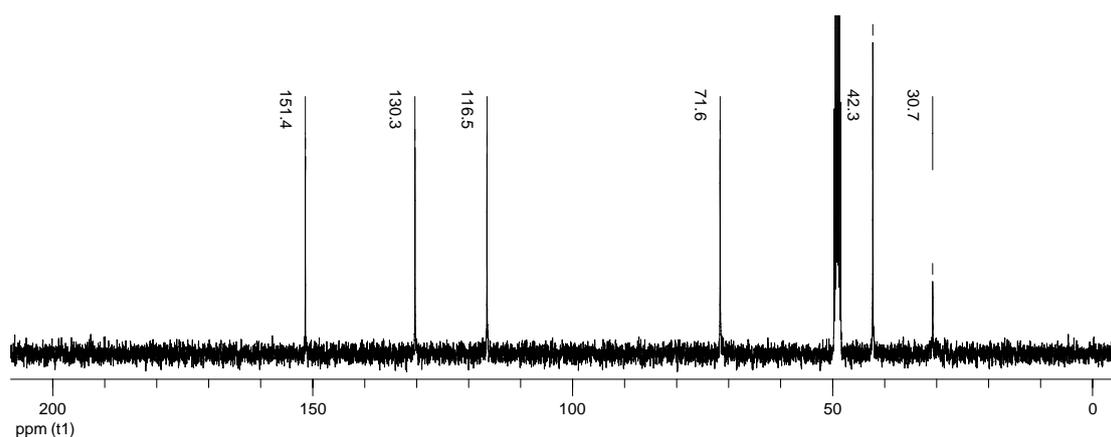
¹³C NMR (400 MHz, 298 K) spectrum of **5** in CDCl₃.



Compound 1: A suspension of **5** (900 mg, 0.69 mmol) and Pd/C (10%, 150 mg) in methanol was stirred at 50 °C under hydrogen atmosphere (60 psi) for 48 h. The resulting mixture was filtered, and the filtrate was concentrated under reduced pressure. The residue was purified by crystallization from chloroform to produce **1** as white solid (655 mg, 92%). ¹H NMR (CH₃OH-*d*₄): δ 6.73 (s, 10 H), 3.80 (br, 20 H), 3.78 (s, 10 H), 2.91 (t, *J* = 5.2 Hz, 20 H). ¹³C NMR (CH₃OH-*d*₄): δ 151.4, 130.3, 116.5, 71.6, 42.3, 30.7. MS (ESI): *m/z* 1041 [M + H]⁺. HR-MS (ESI-TOF): Calcd. for C₅₅H₈₁O₁₀N₁₀: 1041.6137. Found: 1041.6153.



¹H NMR (400 MHz, 298 K) spectrum of **1** in CH₃OH-*d*₄.



¹³C NMR (400 MHz, 298 K) spectrum of **1** in CH₃OH-*d*₄.

Reference:

1. Ogoshi, T.; Hashizume, M.; Yamagishi, T.-A.; Nakamoto, Y. *Chem. Commun.* 2010, **46**, 3708-3710.

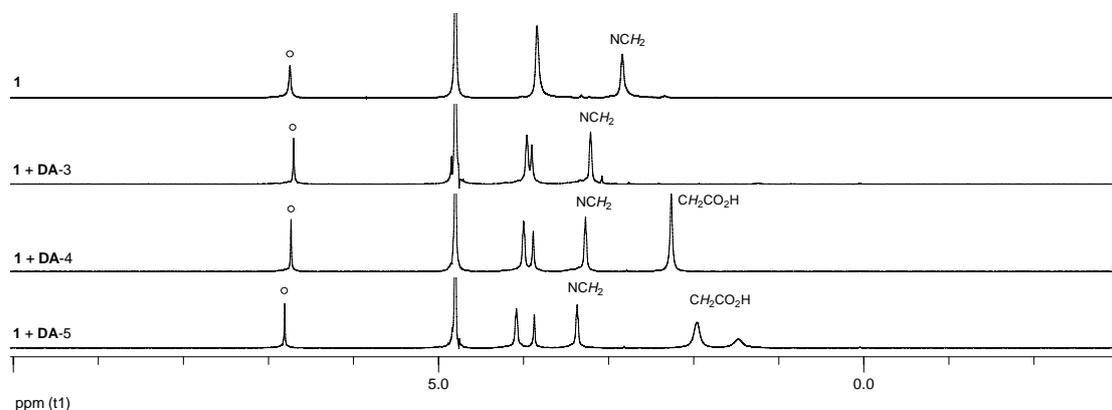
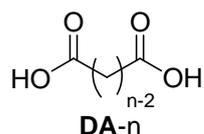


Fig SI-1: ^1H NMR spectra (400 MHz, 298 K) of **1** (2.8 mM) and its mixture with **DA-3-5** (14.0 mM) in D_2O . (o: signal of ArH on free **1**). For the acidity of CH_2 from **DA-3**, the signal of it was not observed.

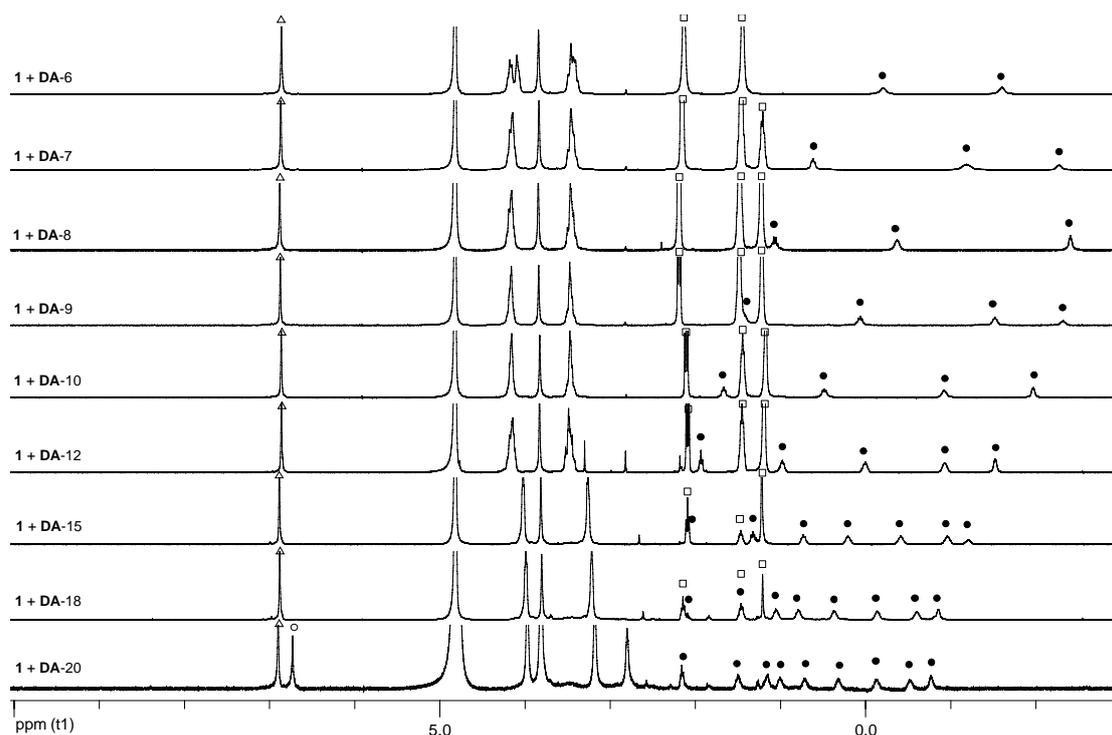


Fig SI-2: ^1H NMR spectra (400 MHz, 298 K) of **1** (2.8 mM) with **DA-n** (14.0 mM) in D_2O . (Δ : signal of ArH from complexed **1**; o: signal of ArH on free **1**; \square : signal of free **DA-n**; \bullet : signals from methylenes of encapsulated **DA-n**). For poor solubility, the longer of the diacids, the less of free diacids were observed.

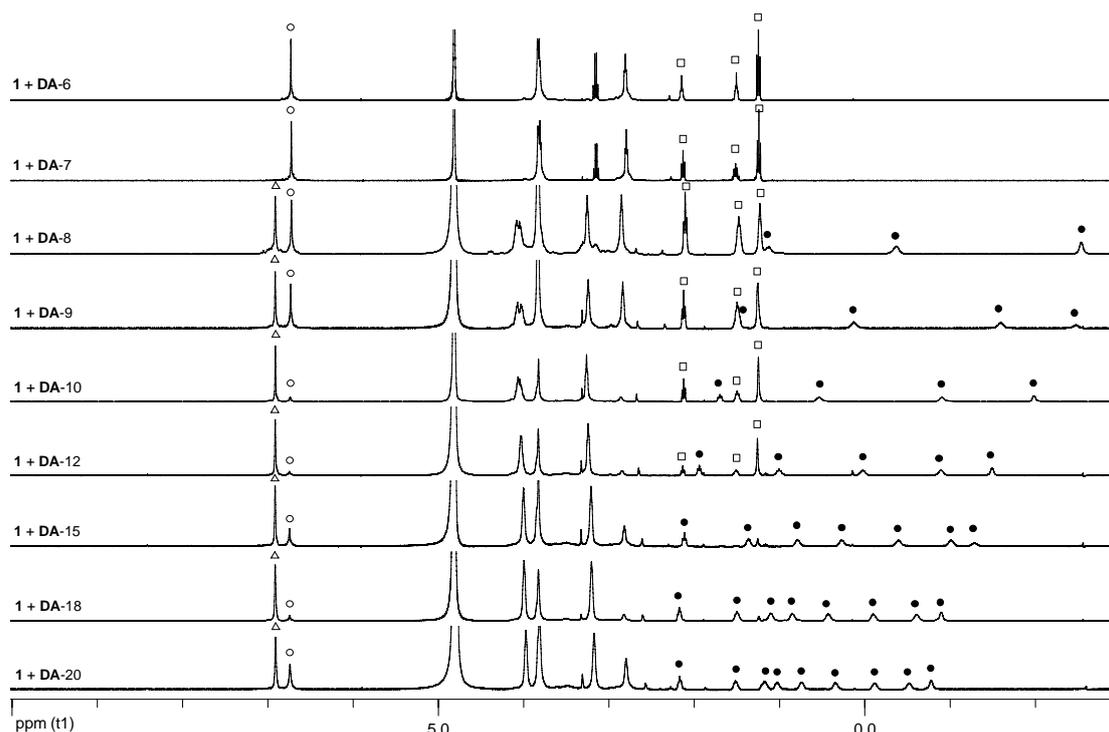


Fig SI-3: ¹H NMR spectra (400 MHz, 298 K) of **1** (2.8 mM) with **DA-n** (2.8 mM) in D₂O. (Δ: signal of ArH from complexed **1**; o: signal of ArH on free **1**; □: signal of free **DA-n**; •: signals from methylenes of encapsulated **DA-n**).

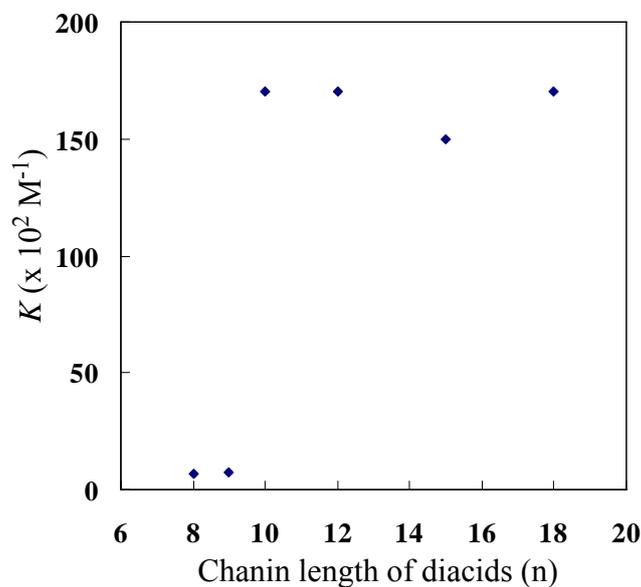


Fig SI-4: Plot of association constants (K) versus the chain length of diacids (n), indicating hydrophobic interaction contributed to the binding of **1** with **DA-n** ($n = 10, 12, 15, 18$). K was calculated from the integrated intensity of the ArH signals of **1** exhibited in Fig SI-3.

Table SI-1: The length of diacids at extended state and results of ESI-TOF analysis on diacid $\mathbf{1}$.

| Diacids (n) | Length of diacid (nm) | Formula of diacid $\mathbf{1}$ | HR-MS (ESI-TOF) ^a | Calculated mass ^b for diacid $\mathbf{1}$ |
|-------------------------------|-----------------------|--|------------------------------|--|
| Adipic acid (DA-6) | 0.85 | C ₆₁ H ₉₀ N ₁₀ O ₁₄ | 1185.6567 | 1185.6560 |
| Pimelic acid (DA-7) | 0.98 | C ₆₂ H ₉₂ N ₁₀ O ₁₄ | 1199.6754 | 1199.6716 |
| Suberic acid (DA-8) | 1.10 | C ₆₃ H ₉₄ N ₁₀ O ₁₄ | 1213.6884 | 1213.6873 |
| Azelaic acid (DA-9) | 1.23 | C ₆₄ H ₉₆ N ₁₀ O ₁₄ | 1227.7064 | 1227.7029 |
| Sebacic acid (DA-10) | 1.36 | C ₆₅ H ₉₈ N ₁₀ O ₁₄ | 1241.7223 | 1241.7186 |
| Dodecanedioic acid (DA-12) | 1.62 | C ₆₇ H ₁₀₂ N ₁₀ O ₁₄ | 1269.7525 | 1269.7499 |
| Pentadecanedioic acid (DA-15) | 1.98 | C ₇₀ H ₁₀₈ N ₁₀ O ₁₄ | 1311.7970 | 1311.7968 |
| Octadecanedioic acid (DA-18) | 2.35 | C ₇₃ H ₁₁₄ N ₁₀ O ₁₄ | 1353.8472 | 1353.8438 |
| Eicosanedioic acid (DA-20) | 2.72 | C ₇₅ H ₁₁₈ N ₁₀ O ₁₄ | 1381.8790 | 1381.8751 |

^a: negative mode; ^b: [diacid $\mathbf{1}$ - H].

Table SI-2: Chemical shifts (δ) and chemical shift changes ($\Delta\delta$) of methylenes on free and encapsulated **DA-n**.

| Diacids (n) | δ of CH_2 on free diacids (δ_{free} , ppm) | | | δ of CH_2 on encapsulated diacids ($\delta_{\text{encapsulated}}$, ppm) | | | | | | | | | | $\Delta\delta$ ($= \delta_{\text{free}} - \delta_{\text{encapsulated}}$, ppm) | | | | | | | | | |
|--------------|---|------|--------|--|-------|-------|-------|-------|-------|-------|-------|-------|------|---|------|------|------|------|------|------|------|--|--|
| | H-2 | H-3 | H-4-10 | H-2 | H-3 | H-4 | H-5 | H-6 | H-7 | H-8 | H-9 | H-10 | H-2 | H-3 | H-4 | H-5 | H-6 | H-7 | H-8 | H-9 | H-10 | | |
| DA-6 | 2.10 | 1.41 | - | -0.25 | -1.64 | - | - | - | - | - | - | - | 2.35 | 3.05 | - | - | - | - | - | - | - | | |
| DA-7 | 2.12 | 1.43 | 1.17 | 0.58 | -1.23 | -2.32 | - | - | - | - | - | - | 1.54 | 2.66 | 3.49 | - | - | - | - | - | - | | |
| DA-8 | 2.16 | 1.45 | 1.19 | 1.03 | -0.41 | -2.46 | - | - | - | - | - | - | 1.13 | 1.86 | 3.65 | - | - | - | - | - | - | | |
| DA-9 | 2.16 | 1.45 | 1.18 | 1.40 | 0.02 | -1.57 | -2.37 | - | - | - | - | - | 0.76 | 1.43 | 2.75 | 3.55 | - | - | - | - | - | | |
| DA-10 | 2.07 | 1.41 | 1.14 | 1.63 | 0.45 | -0.97 | -2.02 | - | - | - | - | - | 0.44 | 0.96 | 2.11 | 3.16 | - | - | - | - | - | | |
| DA-12 | 2.06 | 1.42 | 1.16 | 1.90 | 0.94 | -0.04 | -0.97 | -1.57 | - | - | - | - | 0.16 | 0.48 | 1.20 | 2.13 | 2.73 | - | - | - | - | | |
| DA-15 | 2.06 | 1.43 | 1.19 | 2.06 | 1.29 | 0.69 | 0.17 | -0.45 | -1.00 | -1.26 | - | - | 0.00 | 0.14 | 0.50 | 1.02 | 1.64 | 2.19 | 2.45 | - | - | | |
| DA-18 | 2.11 | 1.43 | 1.17 | 2.11 | 1.43 | 1.02 | 0.76 | 0.34 | -0.17 | -0.64 | -0.9 | - | 0.00 | 0.00 | 0.15 | 0.41 | 0.83 | 1.34 | 1.81 | 2.07 | - | | |
| DA-20 | 2.16 | 1.50 | 1.15 | 2.16 | 1.50 | 1.15 | 1.01 | 0.72 | 0.32 | -0.12 | -0.52 | -0.77 | 0.00 | 0.00 | 0.00 | 0.14 | 0.43 | 0.83 | 1.27 | 1.67 | 1.92 | | |

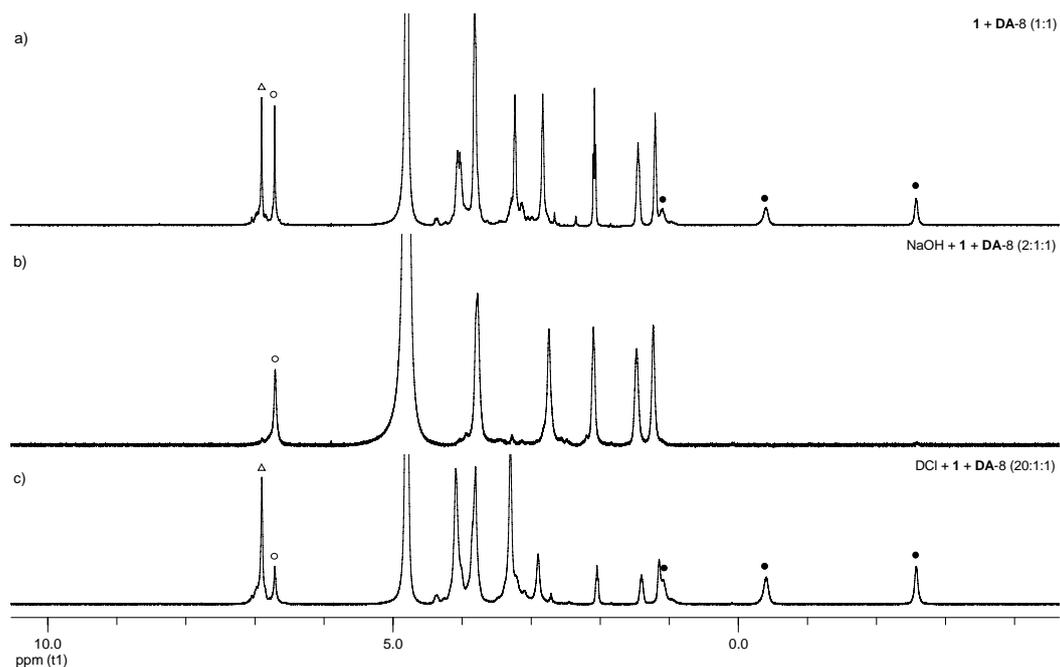


Fig SI-5: ¹H NMR spectra (400 MHz, 298 K, D₂O) of a) the mixture of **1** (2.8 mM) with DA-8 by titration with b) NaOH and c) DCl, indicating they deassembled in alkaline solution, and reassembled in acidic aqueous solution. (Δ: signal of Ar-H on complexed **1**; o: signal from Ar-H of free **1**; •: signals of methylenes on encapsulated DA-8)

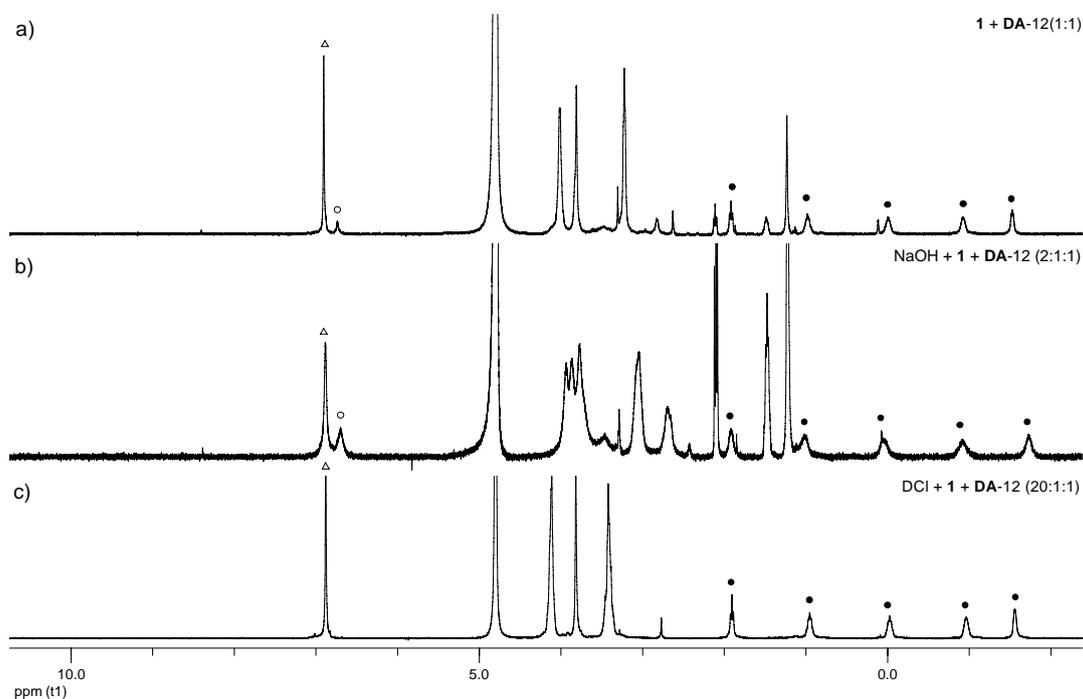


Fig SI-6: ¹H NMR spectra (400 MHz, 298 K, D₂O) of a) the mixture of **1** (2.8 mM) with DA-12 by titration with b) NaOH and c) DCl. (Δ: signal of Ar-H on complexed **1**; o: signal from Ar-H of free **1**; •: signals of methylenes on encapsulated DA-12).

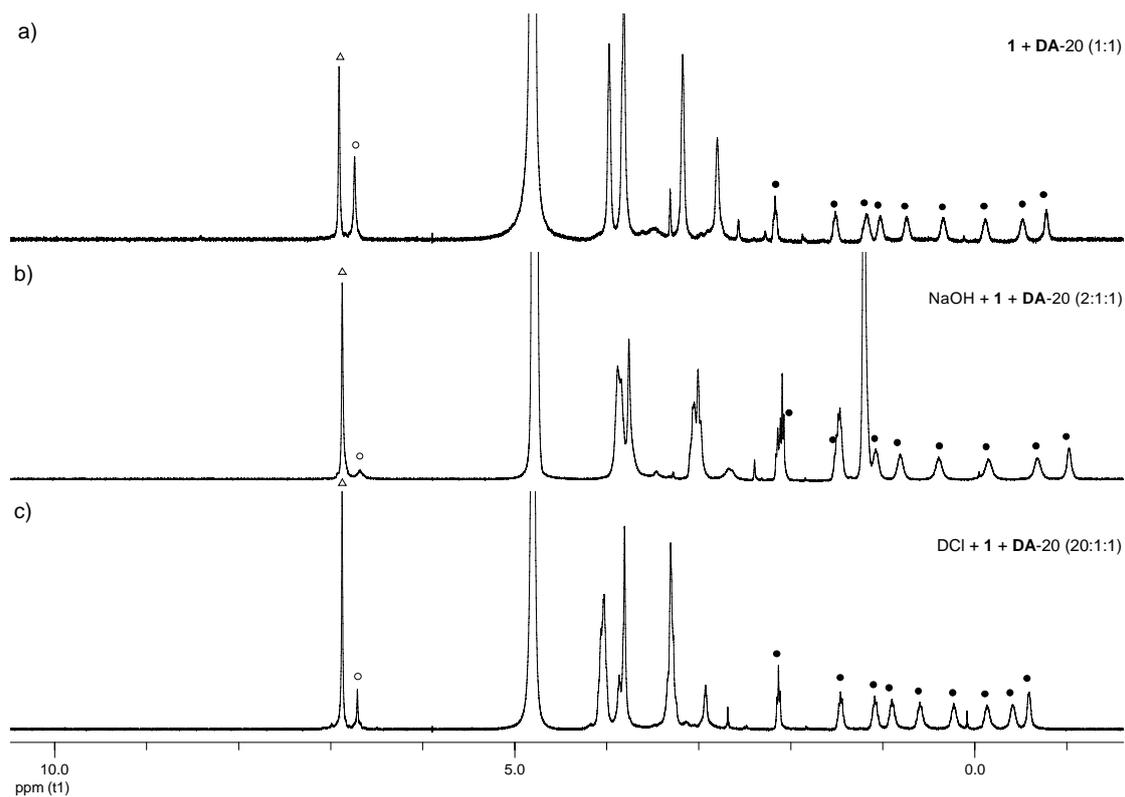


Fig SI-7: ¹H NMR spectra (400 MHz, 298 K, D₂O) of a) the mixture of **1** (2.8 mM) with DA-20 by titration with b) NaOH and c) DCl. (Δ: signal of Ar-H on complexed **1**; o: signal from Ar-H of free **1**; ●: signals of methylenes on encapsulated DA-20).

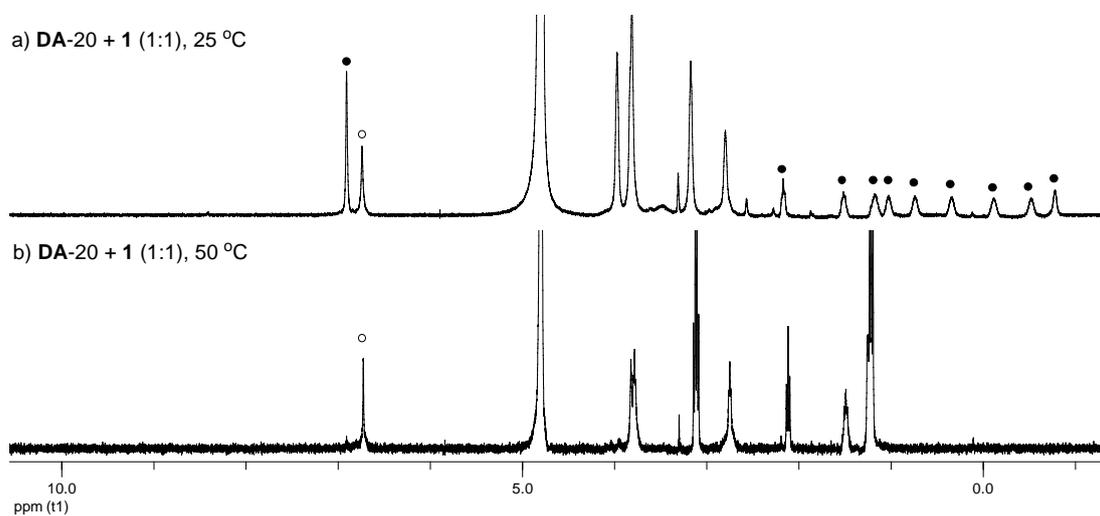


Fig SI-8: VT ¹H NMR spectra (400 MHz, D₂O) of **1** (2.8 mM) with DA-20, indicating the complex of them deassembled at 50 °C. (●: signal from Ar-H and methylene of the complex; o: signal of Ar-H on free **1**.)

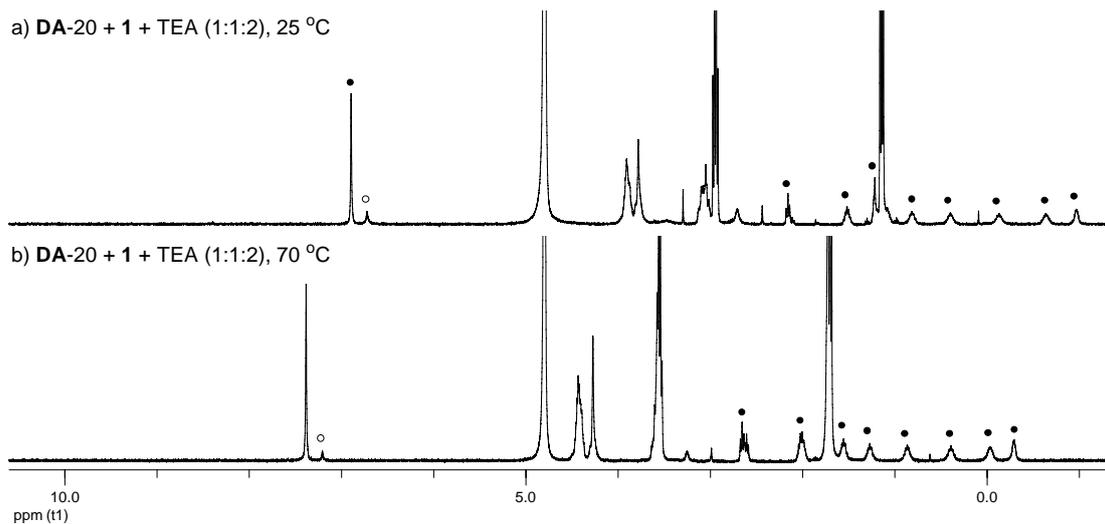


Fig SI-9: VT ^1H NMR spectra (400 MHz, D_2O) of **1** (2.8 mM) with DA-20 and TEA. (●: signal from Ar-*H* and methylene of the complex; ○: signal of Ar-*H* on free **1**.)

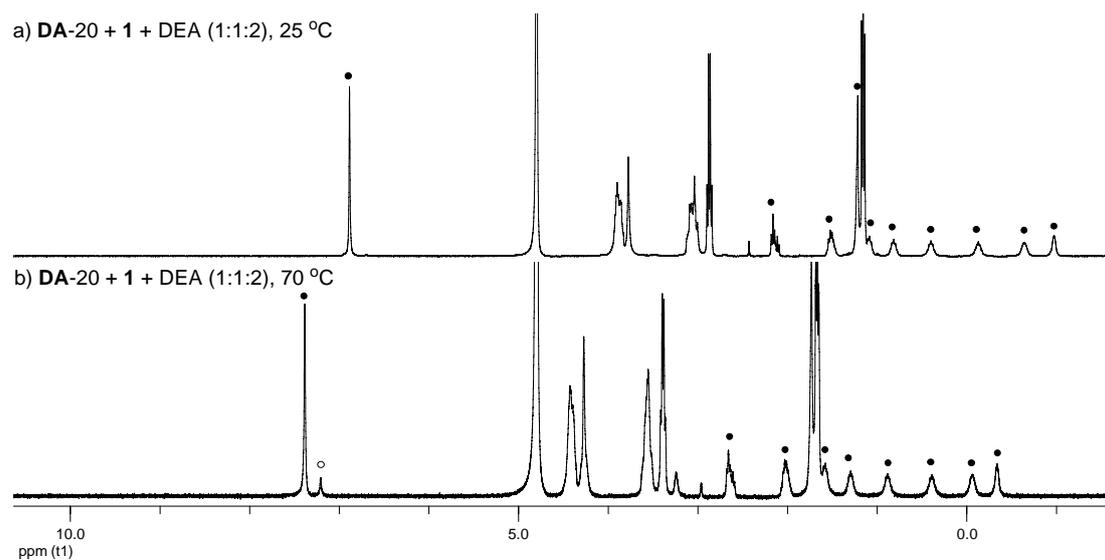


Fig SI-10: VT ^1H NMR spectra (400 MHz, D_2O) of **1** (2.8 mM) with DA-20 and DEA. (●: signal from Ar-*H* and methylene of the complex; ○: signal of Ar-*H* on free **1**.)