

Supporting information for:

Stereochemical effects of chiral monolayers on enhancing the resistance to mammalian cell adhesion

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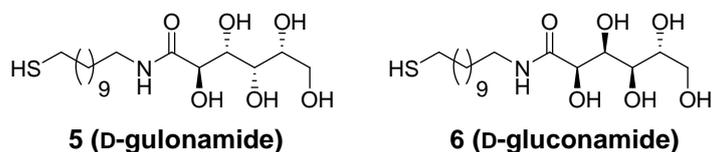
Materials and Methods

Chemicals. 1-Pentadecanethiol and chemicals used for synthesizing all other alkanethiols were purchased from Aldrich Chemicals (Milwaukee, WI) and used as received. Ethanol was used as a solvent for preparing all alkanethiol solutions and for washing the SAM modified gold substrates. A minimum amount of DMSO was mixed with ethanol for solubilizing polyol-terminated alkanethiols, which were poorly soluble in pure ethanol. Bovine serum albumin was purchased from Sigma Chemical Company (St. Louis, MO). Phosphate buffered saline, 1 × PBS (2.7 mM potassium chloride, 137 mM sodium chloride, 8 mM sodium phosphate dibasic, 1.48 mM potassium phosphate monobasic, pH 7.42) was prepared by dissolving 5 PBS tablets (SIGMA, Allentown, PA) in 1000 mL of deionized water with a resistivity of 18 MΩ cm (Millipore, Billerica, MA).

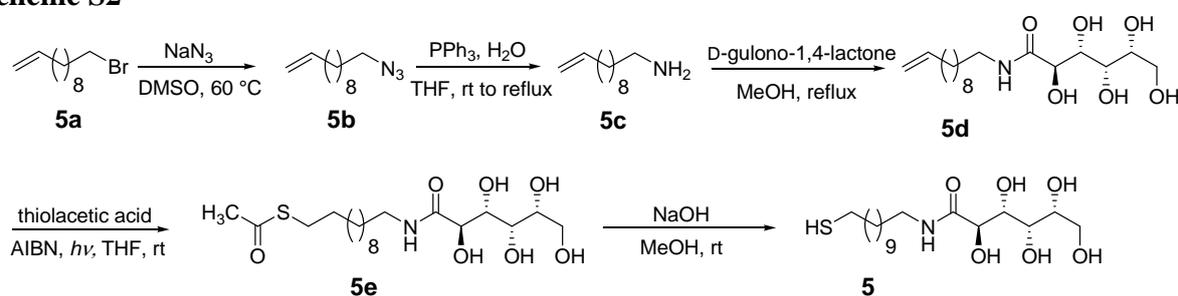
General Information for Synthesis of Alkanethiols. The processes involving reactants sensitive to moisture or air were executed under an atmosphere of argon using oven-dried glassware. Reagents and solvents were reagent grade and used as supplied unless otherwise mentioned. THF was distilled from sodium benzophenone ketyl. Solvents were removed under reduced pressure using rotary evaporator below 40 °C. EMD Silica Gel 60 F₂₅₄ precoated plates (0.25-mm thickness) were used for TLC and a solution of phosphomolybdic acid/ ceric sulfate/ sulfuric acid (10g : 1.25 g : 8% 250 mL), followed by charring at ~ 150 °C, was used for visualization. Flash column chromatography was performed using SILICYCLE, Silica-P Flash Silica Gel with 40-63μ mesh size. NMR spectra (¹H, ¹³C) were recorded on 300 MHz Bruker instrument. ¹H chemical shifts are reported in ppm relative to CDCl₃ δ 7.26 and DMSO-d₆ δ 2.50. ¹³C chemical shifts are reported relative to CDCl₃ δ 77.23 and DMSO-d₆ δ 39.51. (High

Resolution Mass Spectra) HRMS samples were analyzed by positive ion electrospray or electron impact.

Scheme S1



Scheme S2



The synthesis for **5** is shown in Scheme S2. Briefly, the alkene, **5d** was obtained by aminolysis,¹ of D-gulono-1,4-lactone in refluxing methanol. Reaction of **5d** with thioacetic acid using catalytic amount of AIBN (azobisisobutyronitrile) under UV light afforded the thioester **5e**. Base hydrolysis of the thioester **5e** with methanolic NaOH followed by acidification afforded alkanethiol **5**. Compound **6** was synthesized similarly.

Synthesis of Compound 5b. NaN₃ (1.115 g, 17.153 mmol) was added to a solution of 11-bromo-1-undecene, **5a** (2.000 g, 8.576 mmol) in DMSO (20.0 mL) and the mixture was stirred at rt for 10 min followed by heating at 60 °C for 2 h. The mixture was cooled to rt followed by addition of 20 mL of brine. The organic layer was extracted with DCM and the combined organic extracts were dried over anhydrous Na₂SO₄. Evaporation of solvent yielded the crude product as yellow oil. The crude product was purified using silica gel column (Hexanes → 5%

EtOAc in Hexanes) to obtain pure compound, **5b** (1.028 g, 61%) as a colorless oil. $R_f = 0.41$, (Hexanes). $^1\text{H NMR}$ (300 MHz, CDCl_3): δ 5.89-5.75 (m, 1H), 5.04-4.92 (m, 2H), 3.26 (t, $J_{\text{H-H}} = 6.9$ Hz, 2H), 2.04 (q, $J_{\text{H-H}} = 6.3$ Hz, 2H), 1.65-1.56 (m, 2H), 1.39-1.36 (m, 2H), 1.30 (br s, 10H). $^{13}\text{C NMR}$ (75 MHz, CDCl_3): δ 139.6, 114.5, 51.9, 34.2, 29.82, 29.8, 29.54, 29.5, 29.3, 29.2, 27.1. HRMS: found = 194.1649 [M - H], calcd for $[\text{C}_{11}\text{H}_{21}\text{N}_3 - \text{H}]$ 194.1652.

Synthesis of Compound 5c. PPh_3 (0.793 g, 3.023 mmol) was added to a solution of compound, **5b** (0.531 g, 2.748 mmol) in THF (15.0 mL) and the reaction mixture was stirred at rt for 2 h. 55 μL (~1 eq) of water was then added to the reaction mixture which was stirred at rt for another 30 min. The reaction mixture was then refluxed till no more starting material was seen on TLC. The reaction mixture was allowed to cool to rt, solvent was evaporated and the residue was directly loaded on to the silica gel column (EtOAc \rightarrow DCM \rightarrow 3:3:1 DCM : MeOH : Et_3N). Pure compound, **5c** (0.429 g, 92 %) was obtained as colorless viscous oil after evaporation of solvent. $^1\text{H NMR}$ (300 MHz, CDCl_3): δ 5.83-5.72 (m, 1H), 5.00-4.88 (m, 2H), 2.67 (t, $J_{\text{H-H}} = 6.8$ Hz, 2H), 2.05 (br s, 2H), 2.01 (q, $J_{\text{H-H}} = 6.7$ Hz, 2H), 1.45-1.32 (m, 4H), 1.26 (br s, 10H). $^{13}\text{C NMR}$ (75 MHz, CDCl_3): δ 138.5, 113.7, 41.5, 33.4, 29.2, 29.09, 29.06, 28.7, 28.5, 26.5. HRMS: found = 170.1905 [M + H] $^+$, calcd for $[\text{C}_{11}\text{H}_{23}\text{N} + \text{H}]^+$ 170.1909.

Synthesis of Compound 5d. D-gulono-1,4-lactone (0.103 g, 0.579 mmol) was added to a solution of compound, **5c** (0.098 g, 0.579 mmol) in MeOH (5.0 mL) and the mixture was refluxed for 20 h. The mixture was allowed to cool to to rt and then in an ice bath, which led to precipitation of the crude product as a white solid. The solid was washed with ice cold methanol and any residual solvent was evaporated to yield the pure compound, **5d** (0.106 g, 53%) as a white solid. $R_f = 0.61$, (20% MeOH in DCM). $^1\text{H NMR}$ (300 MHz, DMSO-d_6): δ 7.83-7.79 (m, 1H), 5.83-5.70 (m, 1H), 5.44 (br s, 1H), 4.99-4.89 (m, 2H), 4.66 (br s, 1H), 4.46-4.37 (m, 3H),

3.88 (br s, 1H), 3.60-3.29 (m, 5H), 3.07-3.01 (m, 2H), 1.98 (q, $J_{\text{H-H}} = 6.7$ Hz, 2H), 1.38-1.31 (m, 4H), 1.22 (br s, 10 H). ^{13}C NMR (75 MHz, DMSO- d_6): δ 173.2, 138.9, 114.7, 73.3, 72.9, 71.9, 69.5, 62.5, 38.3, 33.2, 29.1, 29.0, 28.9, 28.8, 28.6, 28.3, 26.4. HRMS: found = 370.2197 [M + Na] $^+$, calcd for [C₁₇H₃₃NO₆ + Na] $^+$ 370.2206.

Synthesis of Compound 5e. Recrystallized azobis(isobutyronitrile) (AIBN) (5 mg) and thiolacetic acid (0.026 g, 0.344 mmol) were added to a solution of compound, **5d** (0.030 g, 0.086 mmol) in MeOH (6.0 mL). The reaction mixture was stirred under UV source for 18 h. The solvent was removed under vacuum to yield the crude product as white solid. The solid was washed with ice cold methanol and any residual solvent evaporated to give pure compound, **5e** (0.030 g, 82%) as a white solid. $R_f = 0.48$, (20% MeOH in DCM). ^1H NMR (300 MHz, DMSO- d_6): δ 7.84-7.80 (m, 1H), 5.47 (br s, 1H), 4.67 (br s, 1H), 4.48-4.41 (m, 3H), 3.88-3.83 (m, 1H), 3.60-3.33 (m, 5H), 3.18 (br s, 1H), 3.08-3.02 (m, 2H), 2.82-2.77 (m, 1H), 2.29 (s, 3H), 1.47-1.38 (m, 4H), 1.22 (br s, 14 H). ^{13}C NMR (75 MHz, DMSO- d_6): δ 195.7, 173.2, 73.3, 72.9, 71.8, 69.5, 68.9, 62.5, 52.5, 38.3, 35.9, 35.8, 30.6, 29.0, 28.8, 28.3, 28.1, 26.4, 25.1. HRMS: found = 446.2168 [M + Na] $^+$, calcd for [C₁₉H₃₇NO₇S + Na] $^+$ 446.2188.

Synthesis of Compound 5. Argon gas was bubbled through a solution of compound, **5e** (0.031 g, 0.072 mmol) in MeOH (2.0 mL) for 10 min followed by addition of 80.5 μL solution of 1N NaOH in MeOH. The reaction mixture was stirred at rt till TLC (20% MeOH in DCM) indicated complete consumption of the starting material. The reaction mixture was cooled to 0 °C in an ice bath and to it was added 80.5 μL solution of 1N HCl in MeOH. The mixture was stirred at 0 °C for 20 min and then kept undisturbed at 0 °C to allow precipitation of solid product. The supernatant was carefully decanted and compound, **5** (0.029 g, quantitative) was obtained as a white solid after evaporation of any residual solvent. ^1H NMR (300 MHz, DMSO- d_6): δ 7.81 (br

s, 1H), 5.46 (br s, 1H), 4.66 (br s, 1H), 4.47-4.38 (m, 3H), 3.88 (br s, 1H), 3.60-3.32 (m, 5H), 3.04 (m, 2H), 2.68-2.64 (m, 1H), 1.58-1.38 (m, 4H), 1.22 (br s, 14H). ¹³C NMR (75 MHz, DMSO-d₆): δ 173.2, 73.3, 72.8, 69.5, 62.5, 38.4, 29.1, 28.8, 28.6, 28.5, 27.7, 26.4. HRMS: found = 404.2068 [M + Na]⁺, calcd for [C₁₇H₃₅NO₆S + Na]⁺ 404.2083.

Cleaning of Glass Substrates. Substrates for the gold films were Fisher's finest premium microscope slides (Fisher Scientific, Pittsburgh, PA). Prior to gold deposition, the slides were cleaned with piranha solution. The slides were soaked in piranha solution (3 parts of 35% H₂O₂ in water and 7 parts of concentrated H₂SO₄) at 70 °C for 45 min. Warning! *Piranha solution is extremely corrosive and also has the potential for detonation if contaminated with a significant amount of oxidizable material.* After cooling, the piranha solution was poured off and the slides were rinsed 20 times with water having a resistivity of 18 MΩ cm (Millipore, Billerica, MA), followed by 10 rinses of ethanol and 10 rinses of methanol. The slides were then dried individually with a stream of nitrogen gas and stored in an 80 °C oven overnight.

Gold Deposition on Glass Substrates. Semitransparent gold films of approximately 280 Å thickness were deposited onto the piranha cleaned glass substrates with an electron beam evaporation system from Thermionics (Port Townsend, WA). A layer of titanium (approximately 70 Å thick) was applied first for adhesion of the gold film. Films were deposited at an oblique angle of 45° to the normal of the substrate. The rates of deposition were set at 0.2 Å/s for both gold and titanium. Pressure was maintained at or below 2 × 10⁻⁶ Torr throughout the deposition.

Microcontact Printing. Microcontact printing was done using polydimethylsiloxane (PDMS) stamps using slight modifications of literature reported procedures.² Briefly, gold slides were cut into approximately 1.0 cm × 1.0 cm pieces (gold substrates), rinsed with ethanol and then dried with a stream of nitrogen gas. PDMS stamps were dabbed with 2.0 mM solution of 1-

pentadecanethiol, dried with a stream of nitrogen gas and placed on the gold substrates to allow conformal contact for 20 seconds. The substrates were then rinsed with ethanol, dried with a stream of nitrogen gas and placed in 1.0 mM solutions of alkanethiols **3**, **4**, **5**, **6**; 0.2 mM solutions of alkanethiols **1**, **1'** and 0.2 mM solution containing 1:1 mixture of the enantiomers **1** and **1'**; 1.0 mM solutions of alkanethiols **2**, **2'** and 1.0 mM solution containing 1:1 mixture of the enantiomers **2** and **2'** for 15 h. The substrates were then taken out of the solution, rinsed with ethanol and dried with a stream of nitrogen gas before using them in mammalian cell or bacterial cell culture.

Mammalian Cell Culture. Swiss 3T3 Albino cells were cultured using general culture procedures.³ Briefly, Swiss 3T3 Albino cells purchased from ATCC (Rockville, MD) were cultured in Dulbecco's modified Eagle's medium (DMEM; pH 7.4) supplemented with 10% fetal bovine serum (FBS), 10 $\mu\text{L}/\text{mL}$ Penicillin-Streptomycin and 24 $\mu\text{L}/\text{mL}$ Nystatin in a 25 mL Falcon tissue culture flask (Becton Dickson, Franklin Lakes, NJ). All cultures were kept in an incubator at 37 °C supplemented with 5% CO₂. The media was changed every 3 d. Confluent layer of 3T3 fibroblast cells were detached from culture flask by incubating with 2 mL 0.25% trypsin/0.5 mM EDTA solution for 5-10 min followed by addition of 8 mL of serum containing media.

Mammalian Cell Culture on SAM Modified Gold Substrates. SAM modified gold substrates were placed in a fresh 25 mL Falcon tissue culture flask and incubated in 7.5 mL of serum containing media supplemented with 1.0 μL Penicillin-Streptomycin and 2.4 μL Nystatin for 20 min. 2.5 mL of the cell suspension was introduced in the flask and incubated for 24 h. Media was changed after 24 h of incubation. Pictures of cell attachment and growth on the plate

were taken daily using Cannon C-5060 Wide Zoom camera and viewed from AE31 trinocular inverted microscope. Culture media was changed every 3 d.

Surface Plasmon Resonance. All surface plasmon resonance (SPR) experiments were performed on a SPRImagerII Array System (GWC Technologies, Madison, WI). 1 μ L solutions of the alkanethiols **2**, **2'** and 1 μ L solutions containing 1:1 mixture of the enantiomers **2** and **2'** were spotted with a micropipette on SpotReady 16 chips (GWC Technologies, Madison, WI). The chips were placed in an environment saturated with vapors of ethanol to prevent evaporation of ethanol from the spotted solutions (chips were placed in a tightly covered Petri dish, containing swabs of Kimwipes (Kimberly-Clark Corporation, Roswell, GA) saturated with ethanol and placed on the bench top at room temperature). After 15 h the chips were rinsed with ethanol and blow dried with a stream of nitrogen gas. The chips were then given a final rinse with water and gently blow dried with a stream of nitrogen gas before mounting them in a standard flow cell (GWC Technologies, Madison, WI) for the experiment. Phosphate buffered saline, 1 \times PBS (2.7 mM potassium chloride, 137 mM sodium chloride, 8 mM sodium phosphate dibasic, 1.48 mM potassium phosphate monobasic, pH 7.42) was used as buffer and for preparing the protein solution. The protein solution was filtered through a 0.25- μ m filter before use.

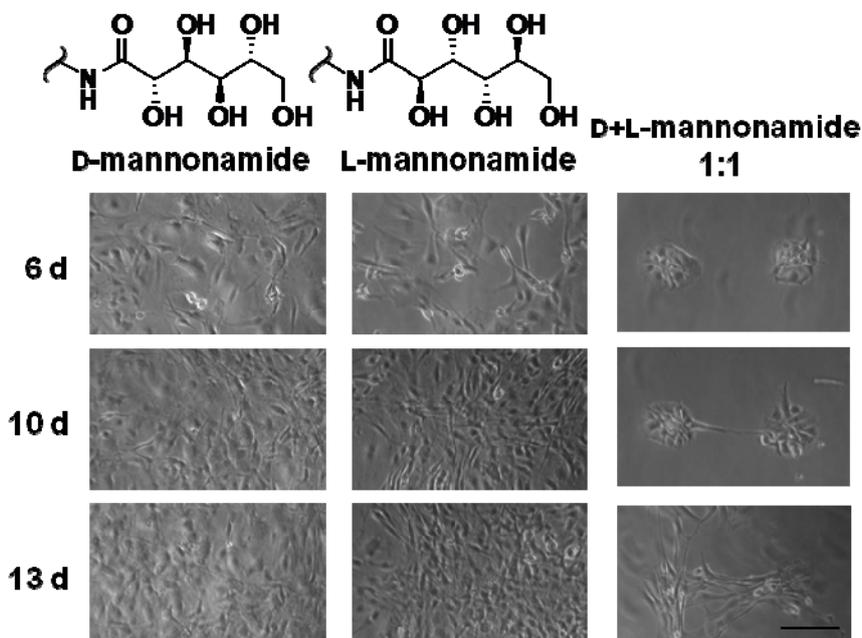
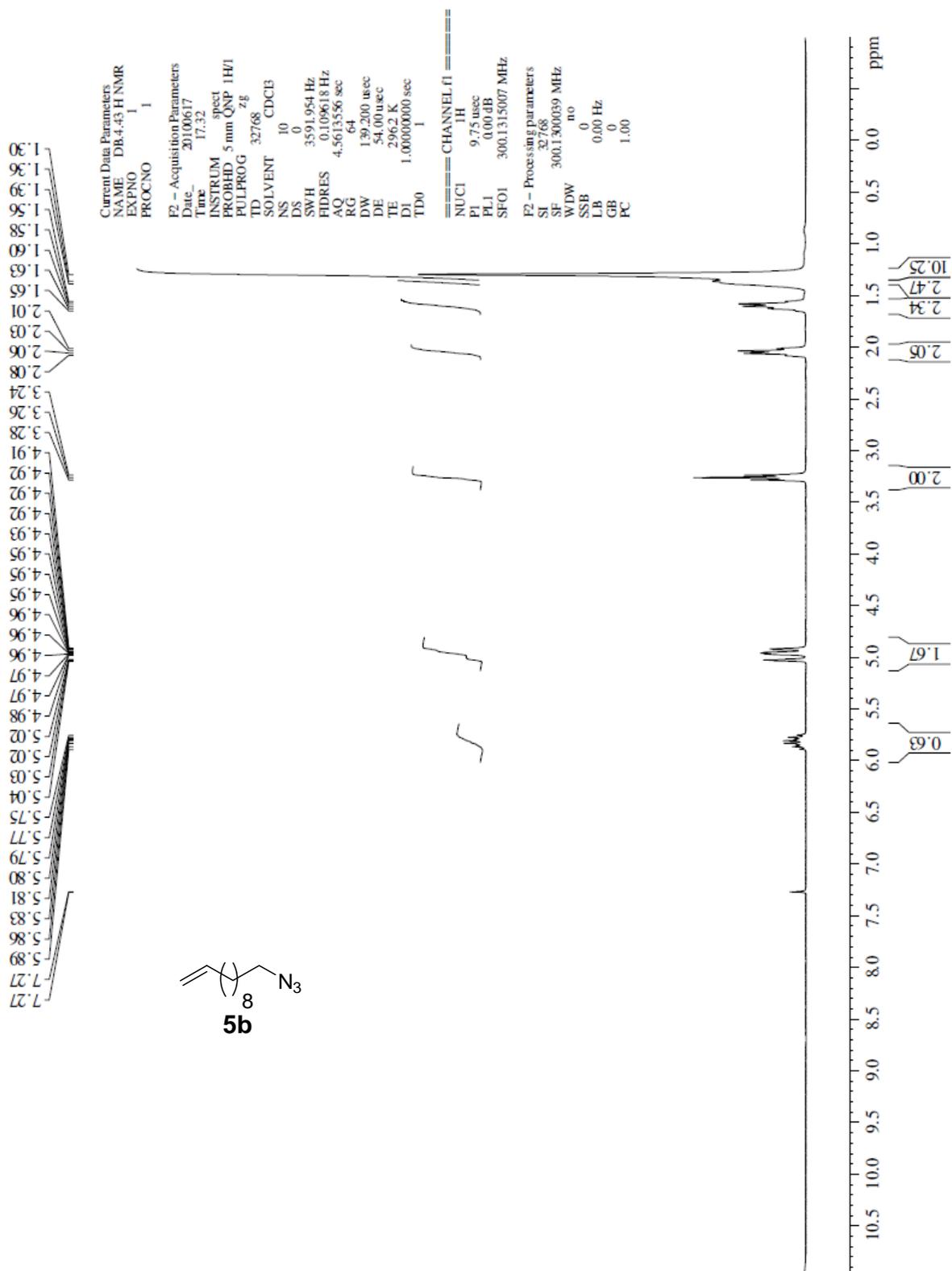
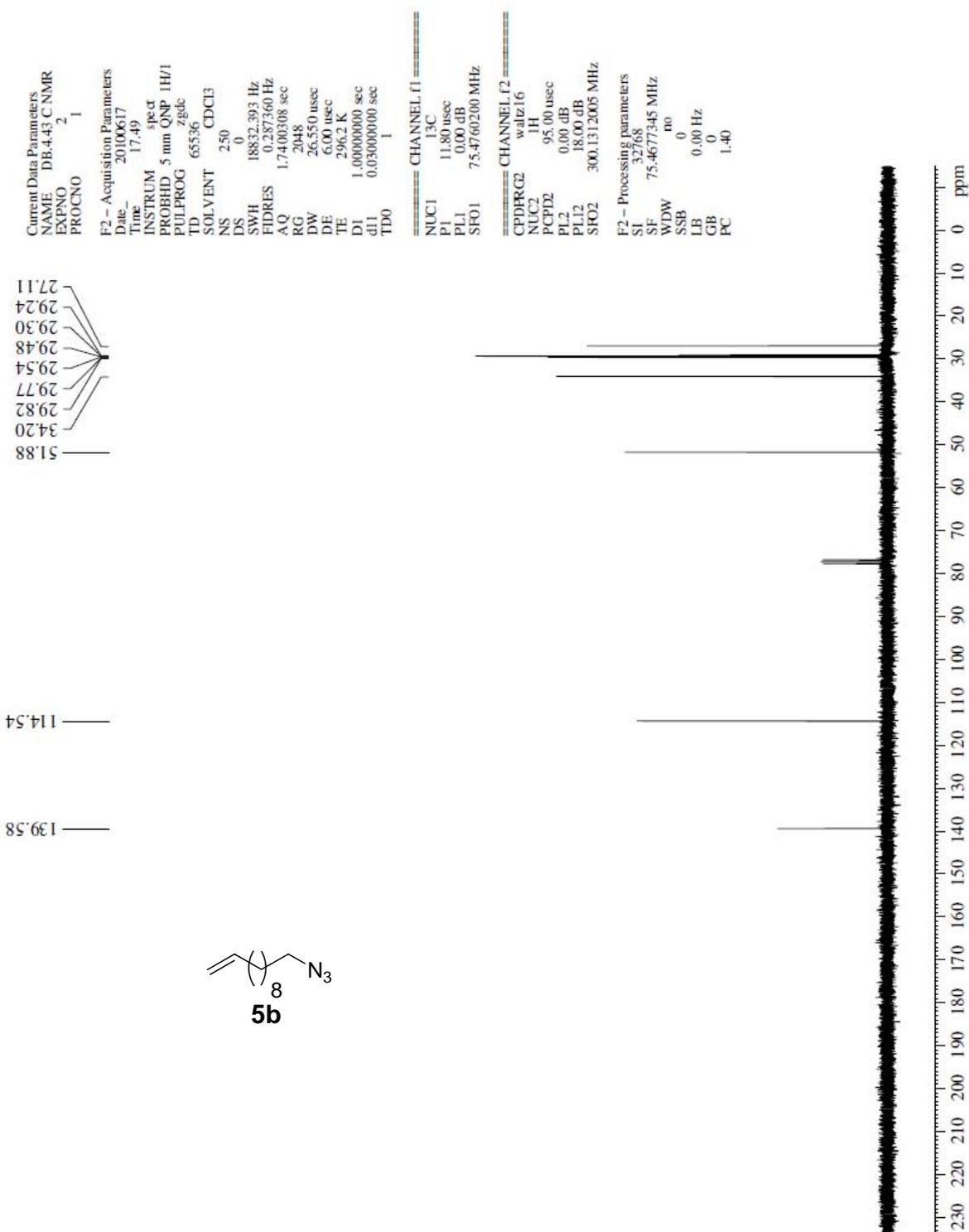
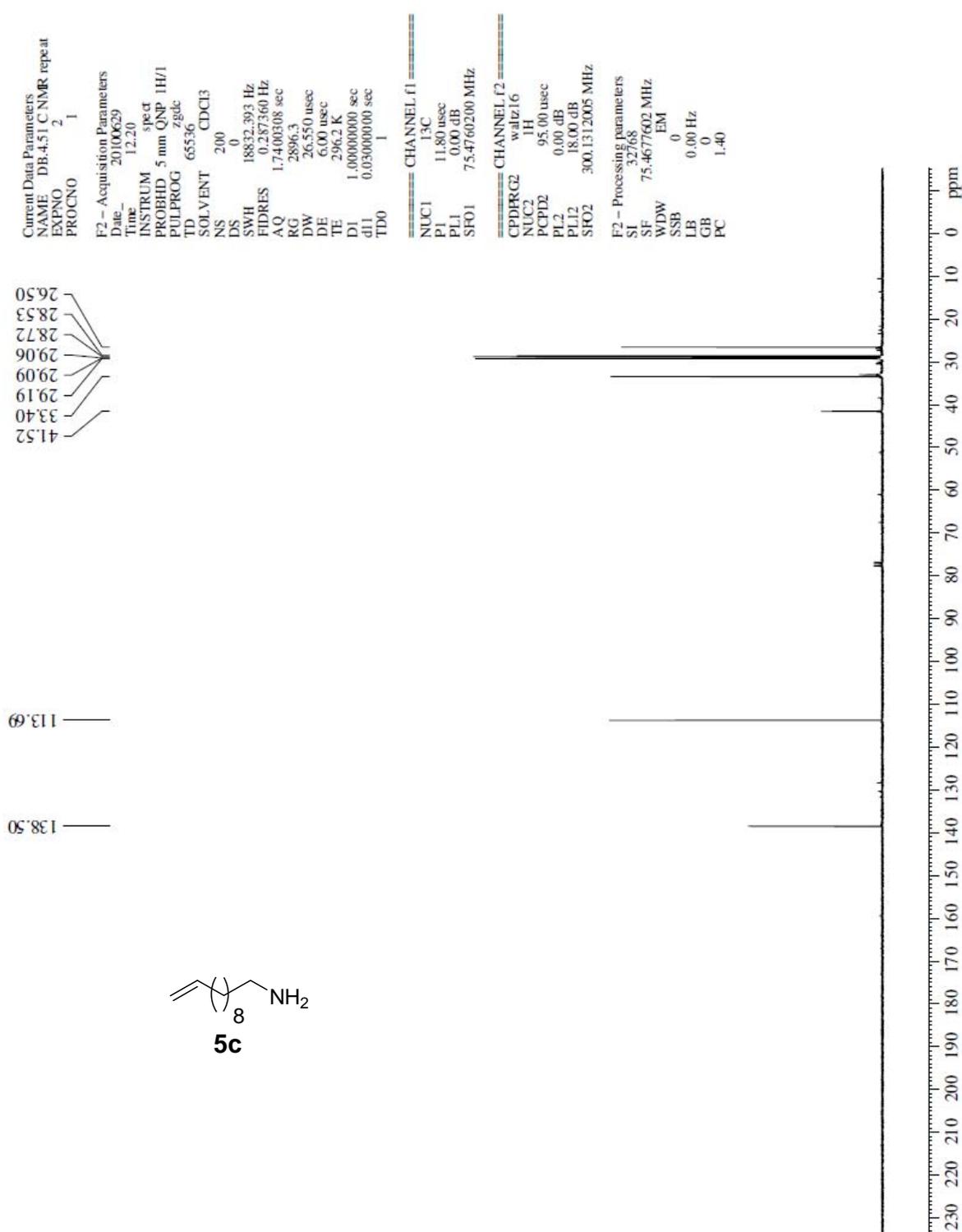
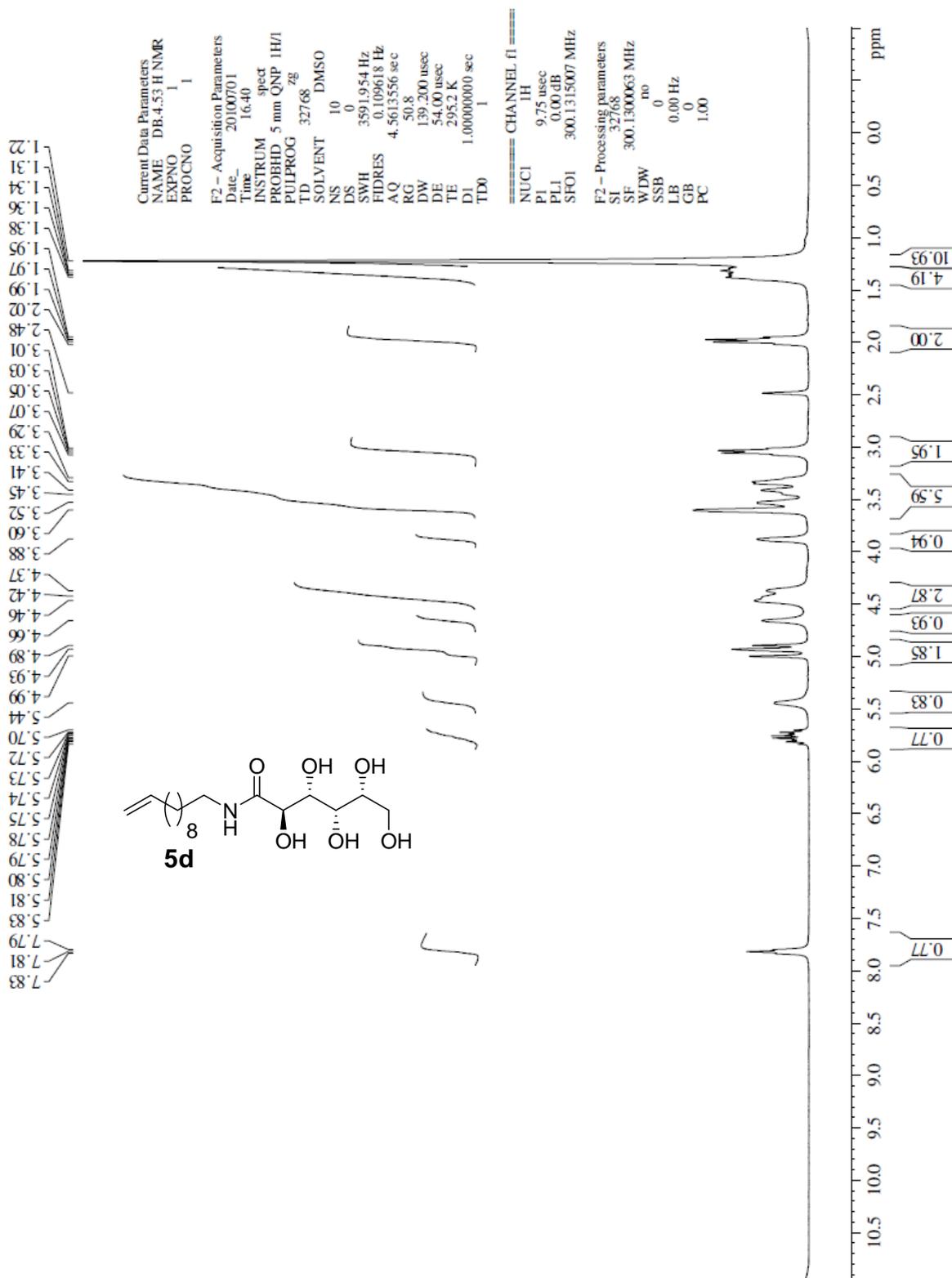


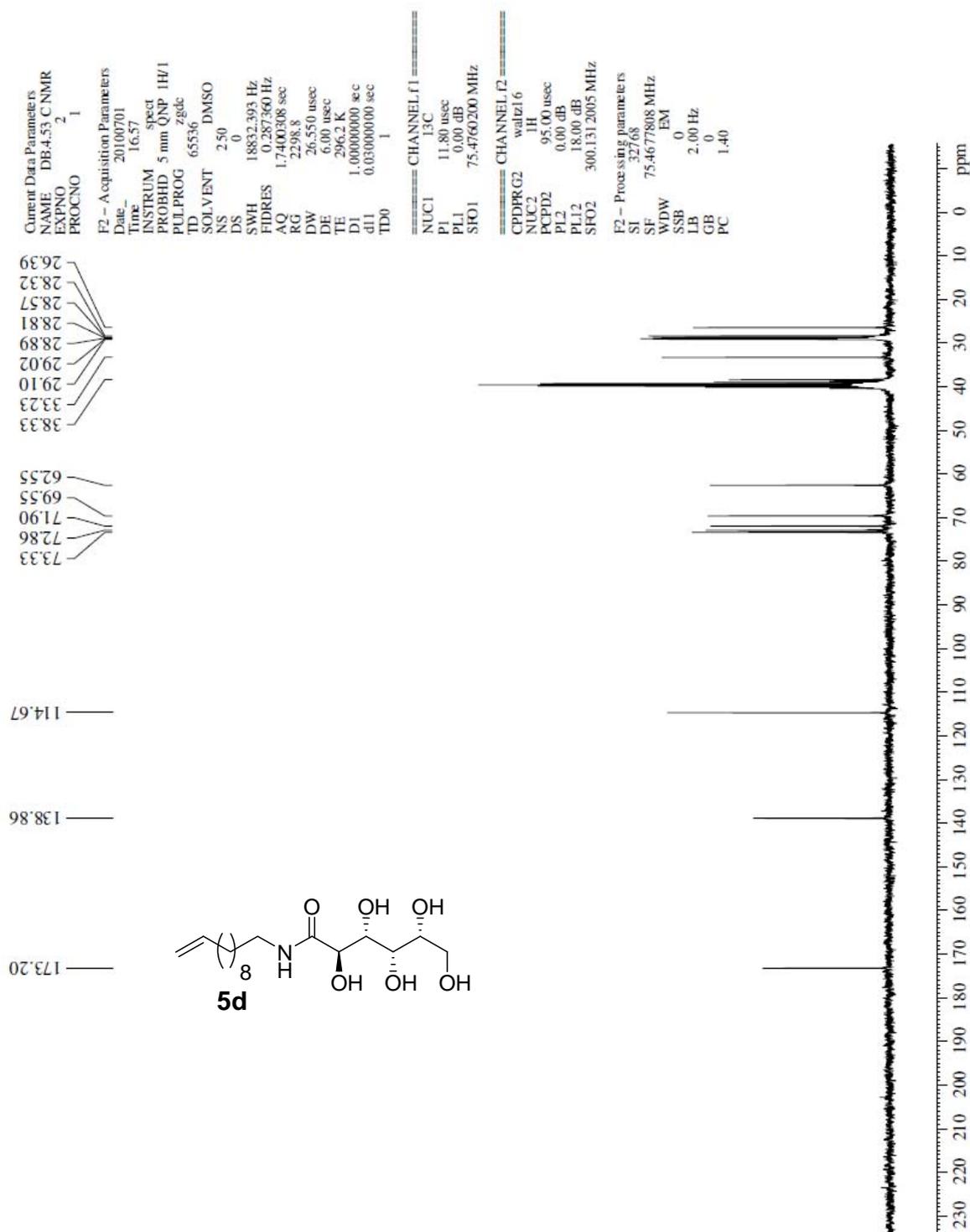
Figure S1. Optical micrographs for the adhesion of Swiss albino 3T3 fibroblast on circular patterns (135 μm in diameter) of pentadecanethiolate, surrounded by aldonamide-terminated SAMs on gold films. The aldonamide-terminated SAMs (terminal groups shown above the cell pictures) consist of D- or L-mannonamide-terminated alkanethiols, or a racemic mixture of D- and L-mannonamide-terminated alkanethiols (prepared from a solution containing 1:1 mixture of the enantiomers). The number of days the substrates were in the culture is indicated to the left. Scale bar = 152 μm .

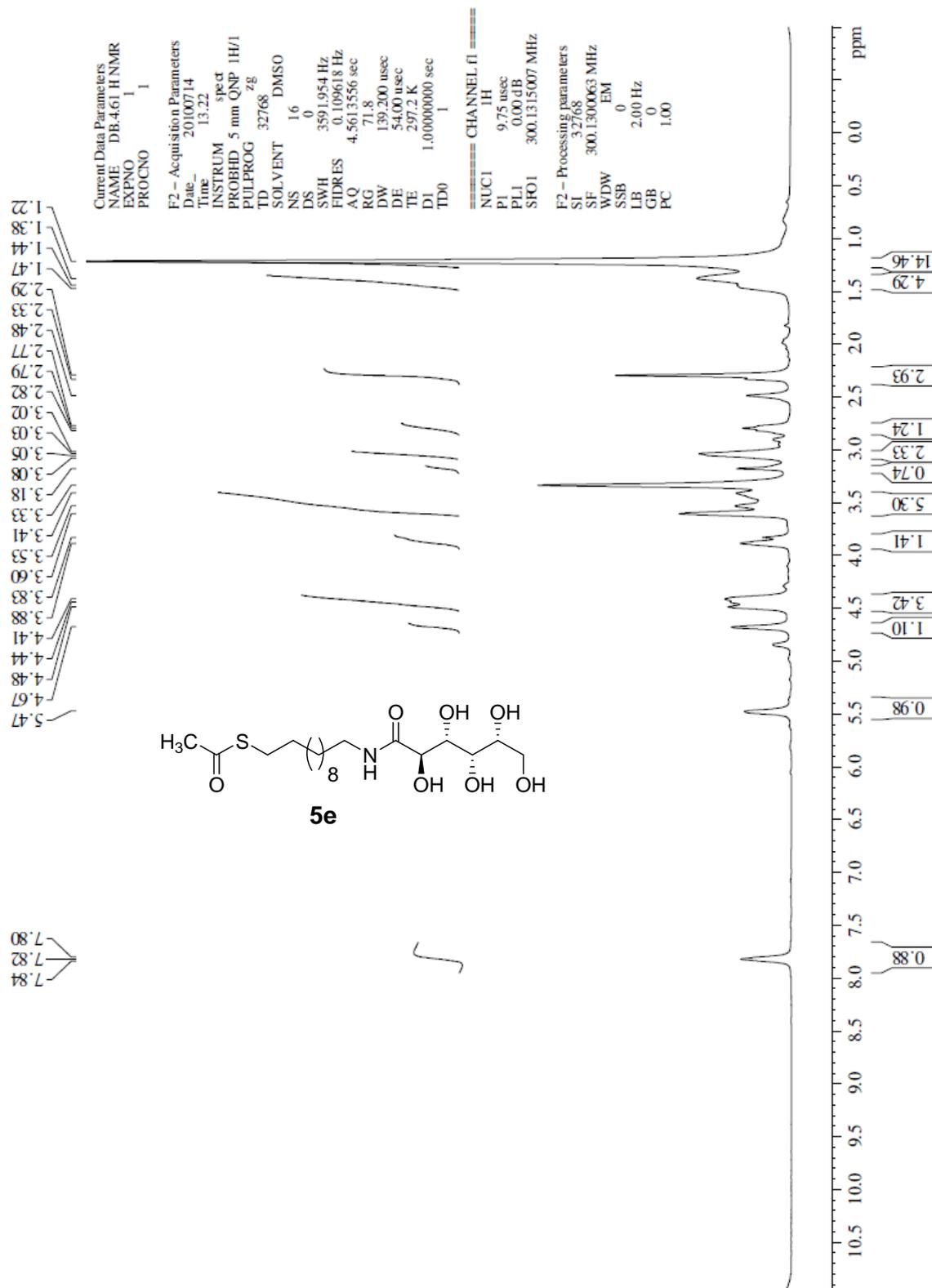


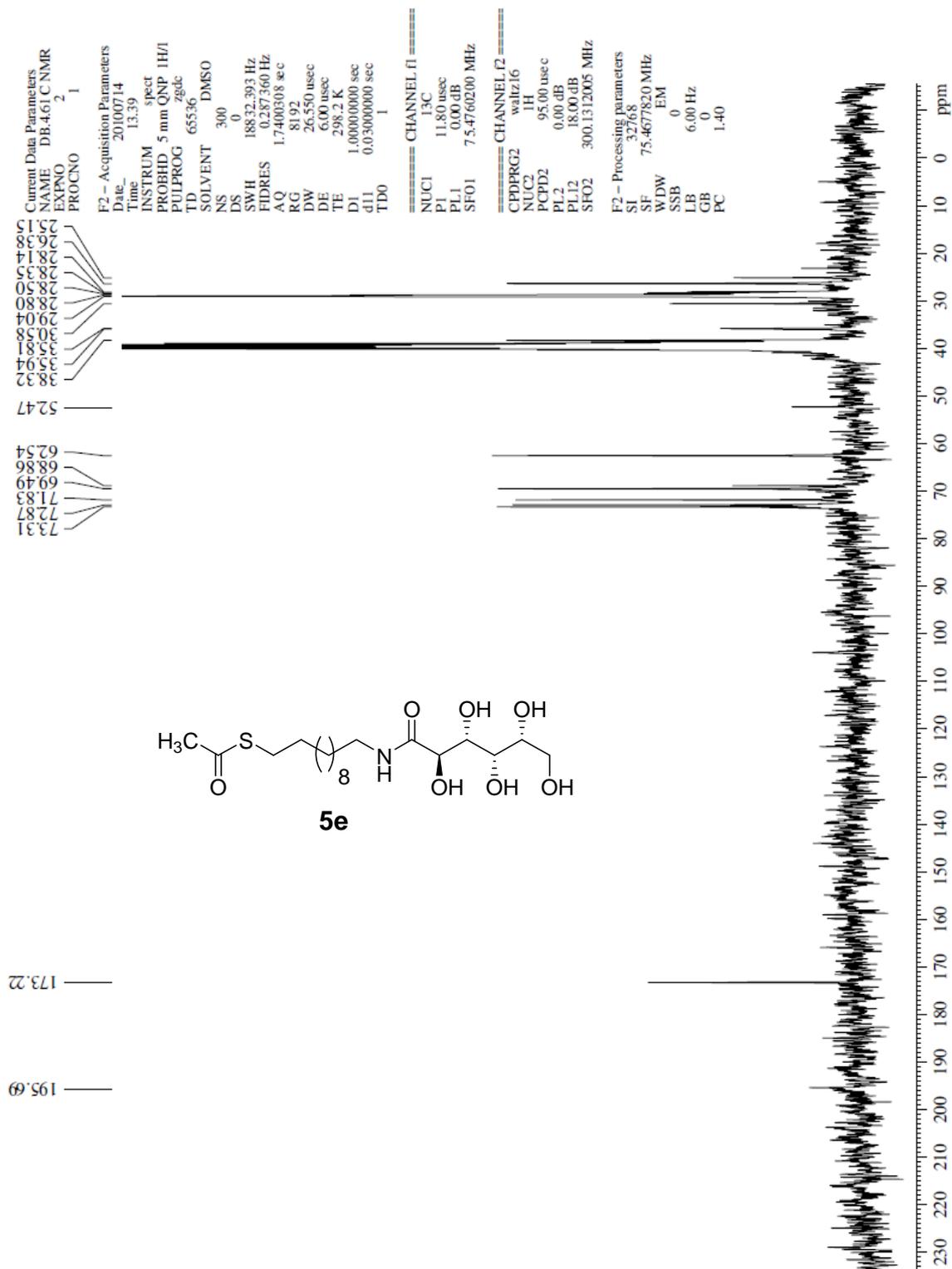


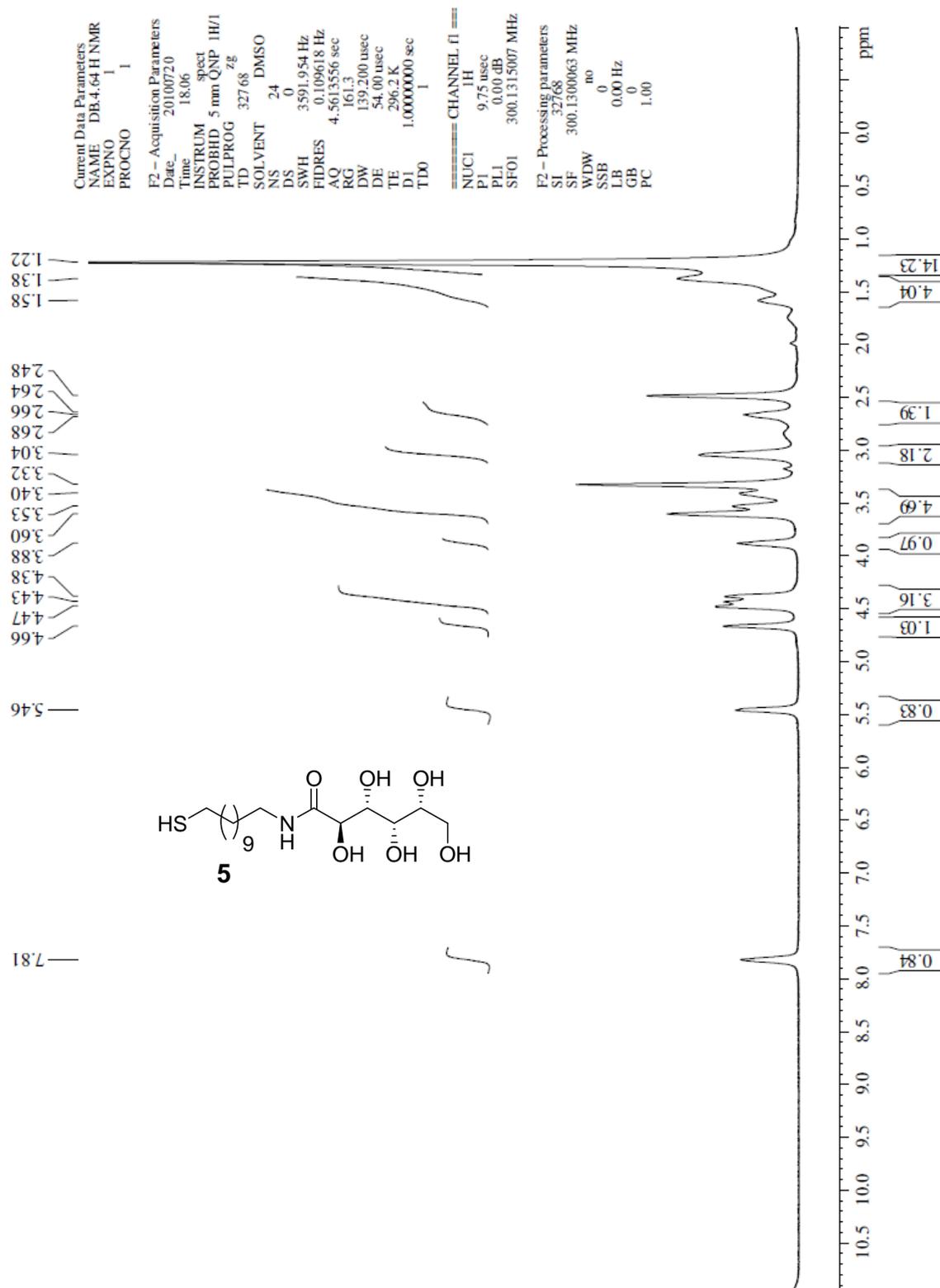


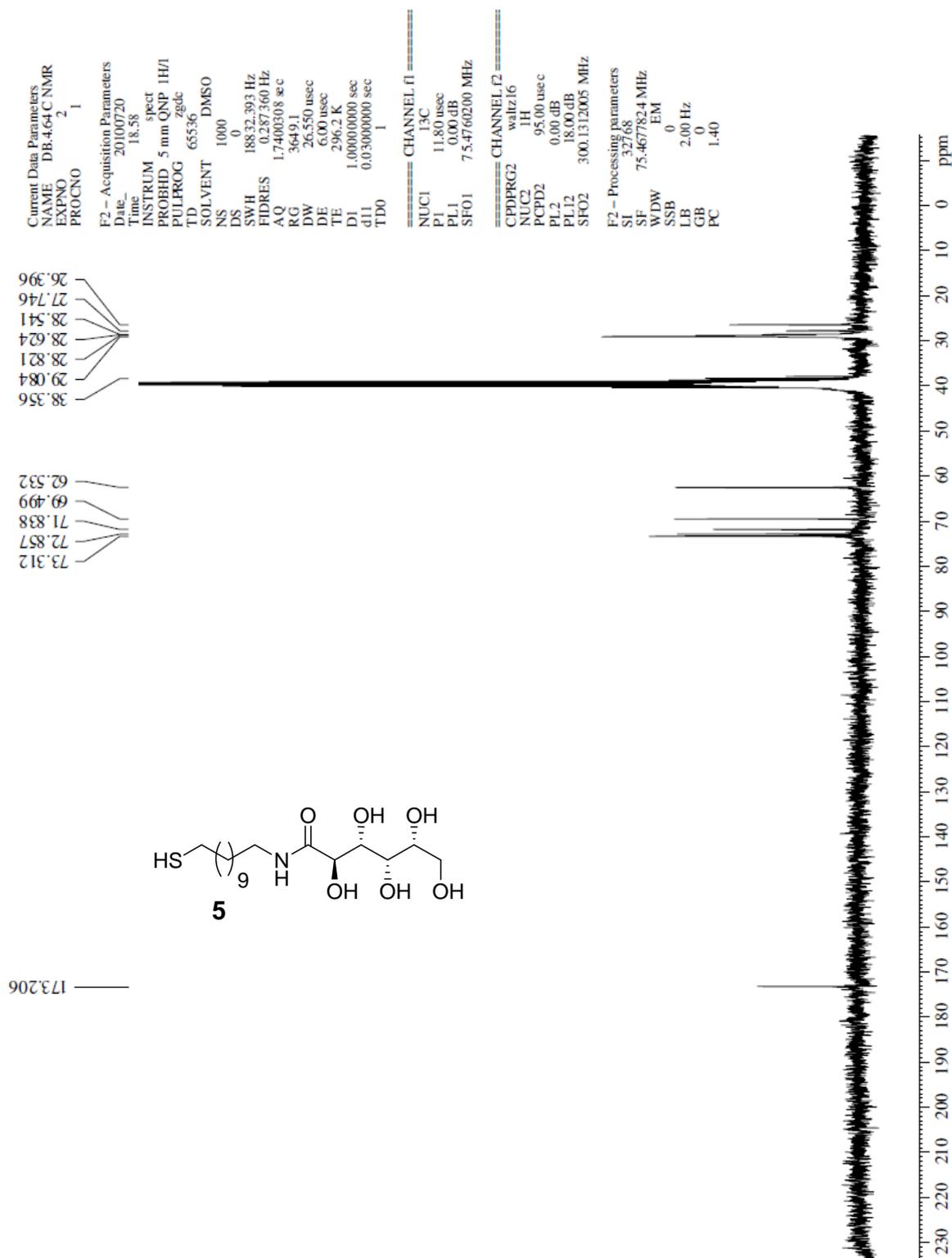












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