

Supplementary Information

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Experimental Section

All reagents were purchased from commercial sources and used without further purification. Copper(I) iodide (fine grey powder), thiophenol, iodobenzene, 2-iodotoluene, 4-iodotoluene, 1,10-phenanthroline, 1,4-di-*tert*-butylbenzene, *t*BuOK were purchased from ACROS. Toluene (dried, seccoSolv®) was purchased from Merck and purged with argon for 15 min before use. All reagents were transferred to the reaction vessel (Pyrex tube with a Teflon screw cap) in a glove box. GC experiments were performed on an Agilent 6890N gas chromatograph equipped with a 30 m X 0.53 mm X 3.0 µm HP-1 capillary column and a FID detector. Elemental analysis was performed on a Thermo CHNS-O analyzer (FlashEA 1112 series). High resolution ESI-MS were recorded using a Waters LCT Premier XE with a dual ionization ESCi® in the Mass Spectrometry Facility in the Institute of Chemistry, Academia Sinica. Leucine Enkephalin [M+H]⁺ 556.277 was used as a reference standard.

Typical Procedure of Copper(I)-Catalyzed C-S Coupling Reaction: In a glove box, CuI (19.0 mg, 0.10 mmol, 10 mol%), 1,10-phenanthroline (18.0 mg, 0.10 mmol, 10 mol%), *t*BuOK (168.3 mg, 1.5 mmol), 1,4-di-*tert*-butylbenzene (19.0 mg, 0.10 mmol) and toluene (3 mL) were transferred to a Pyrex tube with a Teflon screw cap. Thiophenol (112.9 µL, 1.1 mmol) and iodobenzene (111.9 µL, 1.0 mmol) were then added to the above mixture at room temperature. The mixture was stirred and heated at 120 °C in an oil bath for 6 h. GC yield for diphenyl thioether: 87.6%. (1,4-di-*tert*-butylbenzene was used as the internal standard, correction factor: 1.12).

Copper(I)-Catalyzed C-S Coupling Reaction without Phen : In a glove box, CuI (19.0 mg, 0.10 mmol, 10 mol%), *t*BuOK (168.3 mg, 1.5 mmol), 1,4-di-*tert*-butylbenzene (19.0 mg, 0.10 mmol) and toluene (3 mL)

were transferred to a Pyrex tube with a Teflon screw cap. Thiophenol (112.9 μL , 1.1 mmol) and iodobenzene (111.9 μL , 1.0 mmol) were then added to the above mixture at room temperature. The mixture was stirred and heated at 120 $^{\circ}\text{C}$ in an oil bath for 6 h. GC yield for diphenyl thioether: 6.6%. (1,4-di-*tert*-butylbenzene was used as the internal standard, correction factor: 1.12).

Selective Copper(I)-Catalyzed C-S Coupling Reaction with Phen: In a glove box, CuI (19.0 mg, 0.10 mmol, 10 mol%), 1,10-phenanthroline (18.0 mg, 0.10 mmol, 10 mol%), *t*BuOK (168.3 mg, 1.5 mmol), 1,4-di-*tert*-butylbenzene (19.0 mg, 0.10 mmol) and toluene (3 mL) were transferred to a Pyrex tube with a Teflon screw cap. Thiophenol (102.6 μL , 1.0 mmol), 4-iodotoluene (411 mg, 1.9 mmol) and 2-iodotoluene (239 μL , 1.9 mmol) were then added to the above mixture at room temperature. The mixture was stirred and heated at 120 $^{\circ}\text{C}$ in an oil bath for 6 h. GC yield: 39% for 2-tolylphenyl thioether (**2a**) and 49.4% for 4-tolylphenyl thioether (**2b**). (1,4-di-*tert*-butylbenzene was used as the internal standard, correction factor: 1.08).

Selective Copper(I)-Catalyzed C-S Coupling Reaction without Phen: In a glove box, CuI (19.0 mg, 0.10 mmol, 10 mol%), *t*BuOK (168.3 mg, 1.5 mmol), 1,4-di-*tert*-butylbenzene (19.0 mg, 0.10 mmol) and toluene (3 mL) were transferred to a Pyrex tube with a Teflon screw cap. Thiophenol (102.6 μL , 1.0 mmol), 4-iodotoluene (436 mg, 2.0 mmol) and 2-iodotoluene (254 μL , 2.0 mmol) were then added to the above mixture at room temperature. The mixture was stirred and heated at 120 $^{\circ}\text{C}$ in an oil bath for 6 h. GC yield: 11.9% for 2-tolylphenyl thioether (**2a**) and 15.2% for 4-tolylphenyl thioether (**2b**). (1,4-di-*tert*-butylbenzene was used as the internal standard, correction factor: 1.08).

Procedure of Copper(I)-Catalyzed C-S Coupling Reaction with 10 mol% Phen for 3 hr: In a glove box, CuI (19.0 mg, 0.10 mmol, 10 mol%), 1,10-phenanthroline (18.0 mg, 0.10 mmol, 10 mol%), *t*BuOK (168.3 mg, 1.5 mmol), 1,4-di-*tert*-butylbenzene (19.0 mg, 0.10 mmol) and toluene (3 mL) were transferred to a Pyrex tube with a Teflon screw cap. Thiophenol (112.9 μ L, 1.1 mmol) and iodobenzene (111.9 μ L, 1.0 mmol) were then added to the above mixture at room temperature. The mixture was stirred and heated at 120 °C in an oil bath for 3 h. GC yield for diphenyl thioether: 35.7%. (1,4-di-*tert*-butylbenzene was used as the internal standard, correction factor: 1.12).

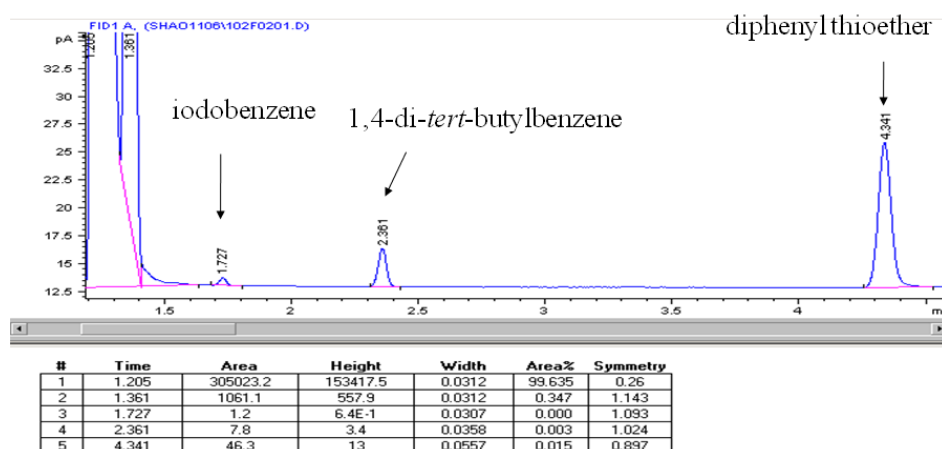
Procedure of Copper(I)-Catalyzed C-S Coupling Reaction with 20 mol% Phen for 3 hr: In a glove box, CuI (19.0 mg, 0.10 mmol, 10 mol%), 1,10-phenanthroline (36.0 mg, 0.20 mmol, 20 mol%), *t*BuOK (168.3 mg, 1.5 mmol), 1,4-di-*tert*-butylbenzene (19.0 mg, 0.10 mmol) and toluene (3 mL) were transferred to a Pyrex tube with a Teflon screw cap. Thiophenol (112.9 μ L, 1.1 mmol) and iodobenzene (111.9 μ L, 1.0 mmol) were then added to the above mixture at room temperature. The mixture was stirred and heated at 120 °C in an oil bath for 3 h. GC yield for diphenyl thioether: 63.1%. (1,4-di-*tert*-butylbenzene was used as the internal standard, correction factor: 1.12).

Procedure of Copper(I)-Catalyzed C-S Coupling Reaction with 30 mol% Phen for 3 hr: In a glove box, CuI (19.0 mg, 0.10 mmol, 10 mol%), 1,10-phenanthroline (54.0 mg, 0.30 mmol, 30 mol%), *t*BuOK (168.3 mg, 1.5 mmol), 1,4-di-*tert*-butylbenzene (19.0 mg, 0.10 mmol) and toluene (3 mL) were transferred to a Pyrex tube with a Teflon screw cap. Thiophenol (112.9 μ L, 1.1 mmol) and iodobenzene (111.9 μ L, 1.0 mmol) were then added to the above mixture at room temperature. The mixture was stirred and heated at

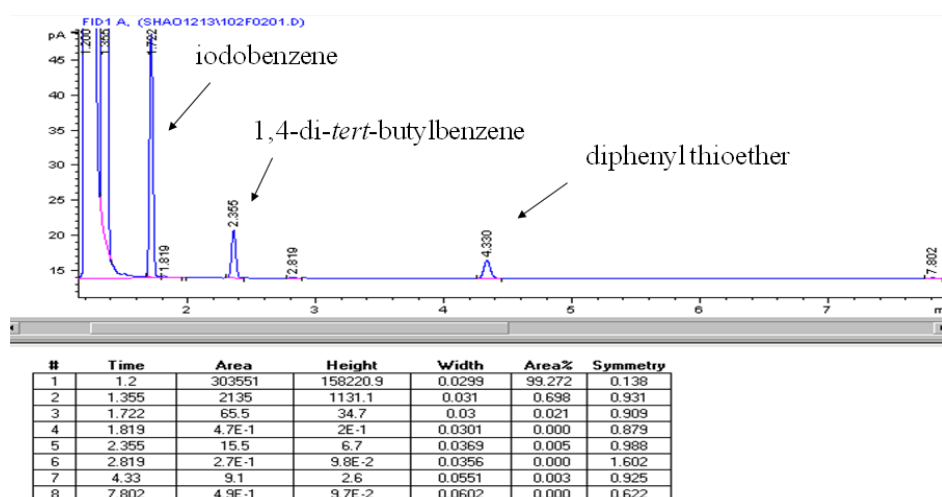
120 °C in an oil bath for 3 h. GC yield for diphenyl thioether: 67.2%. (1,4-di-*tert*-butylbenzene was used as the internal standard, correction factor: 1.12).

GC Data for Copper-Catalyzed C-S Cross-Coupling Reaction

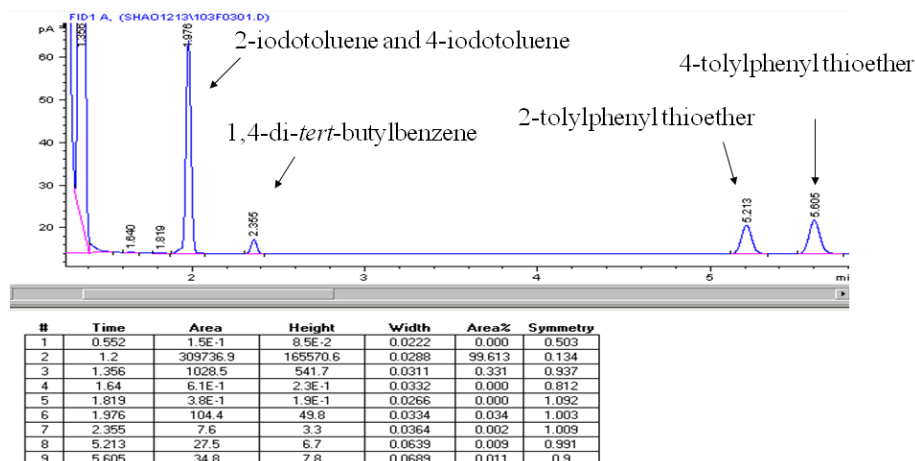
1. Copper(I)-Catalyzed C-S Coupling Reaction



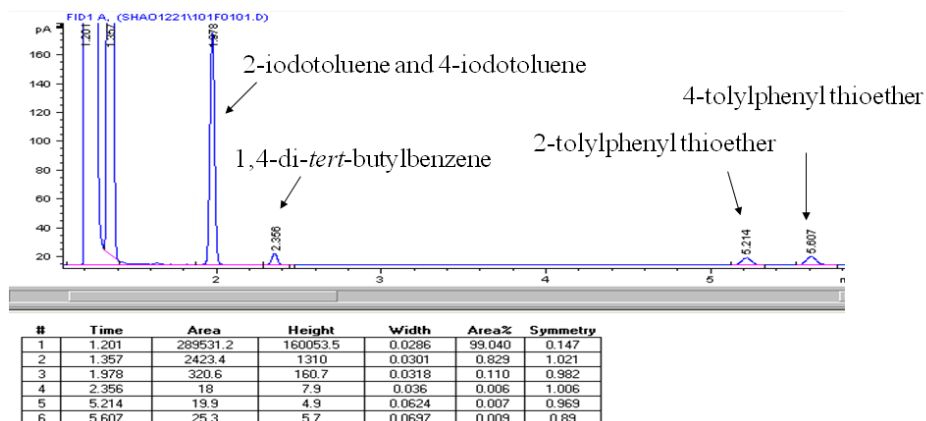
2. Copper(I)-Catalyzed C-S Coupling Reaction without Phen



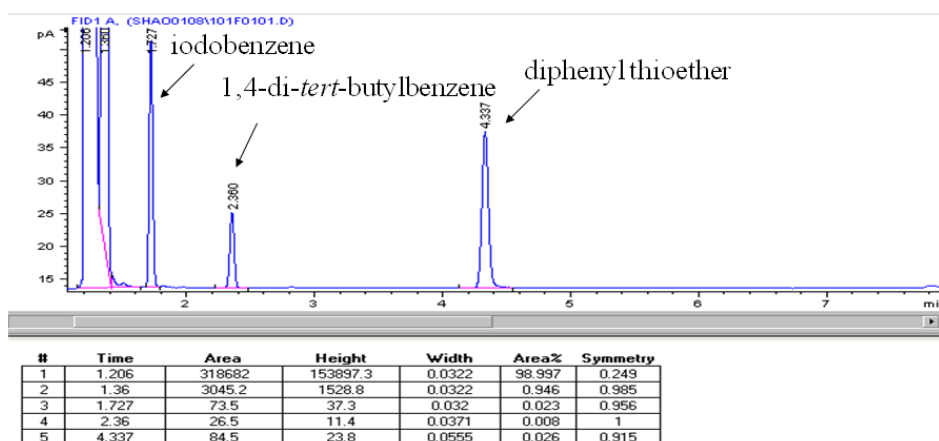
3. Selective Copper(I)-Catalyzed C-S Coupling Reaction with Phen:



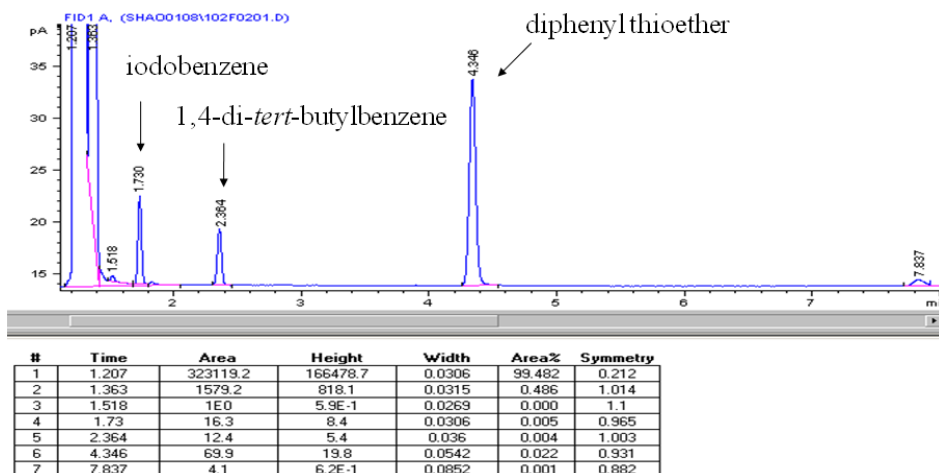
4. Selective Copper(I)-Catalyzed C-S Coupling Reaction without Phen:



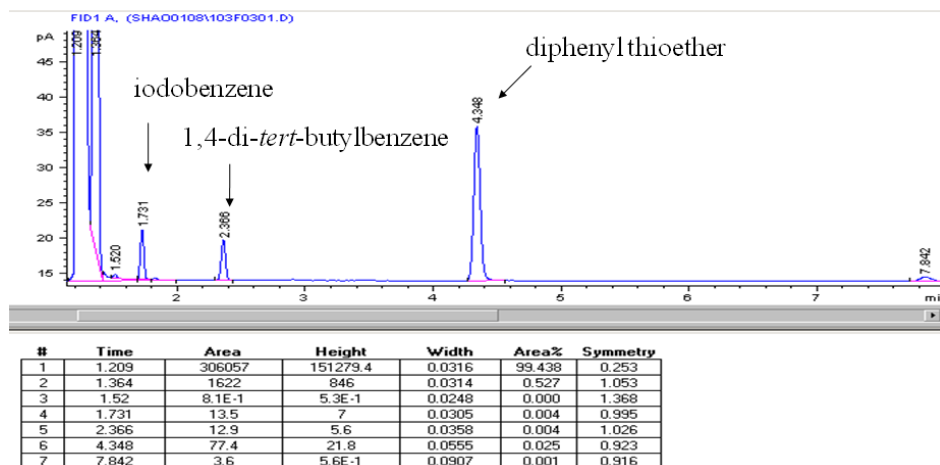
5. Copper(I)-Catalyzed C-S Coupling Reaction with 10 mol% Phen for 3 hr:



6. Copper(I)-Catalyzed C-S Coupling Reaction with 20 mol% Phen for 3 hr:

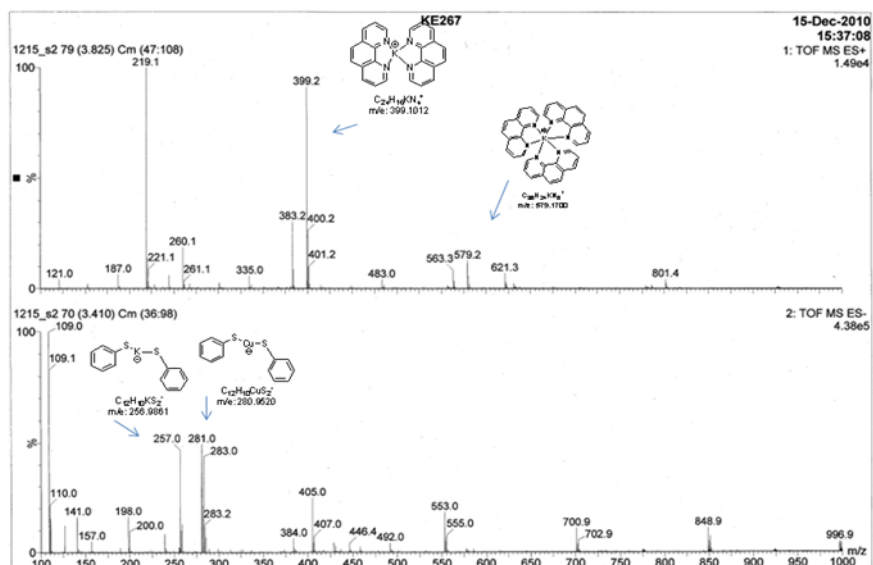


7. Copper(I)-Catalyzed C-S Coupling Reaction with 30 mol% Phen for 3 hr:



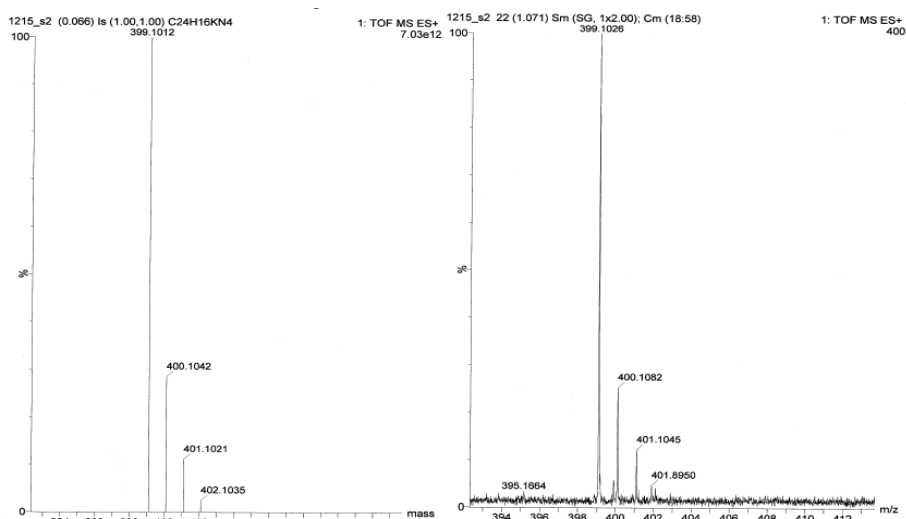
In situ ESI-MS Spectra and Isotope Distribution of Intermediates

- Experimental and theoretical isotopic distributions of $[\text{K}(\text{phen})_2]^+$, $[\text{K}(\text{phen})_3]^+$, $[\text{K}(\text{SPh})_2]^-$ and $[\text{Cu}(\text{SPh})_2]^-$ (from the ESI-MS spectra of the reaction solution taken during the reaction of thiophenol and iodobenzene with *t*BuOK in the presence of CuI and phen in toluene at room temperature).



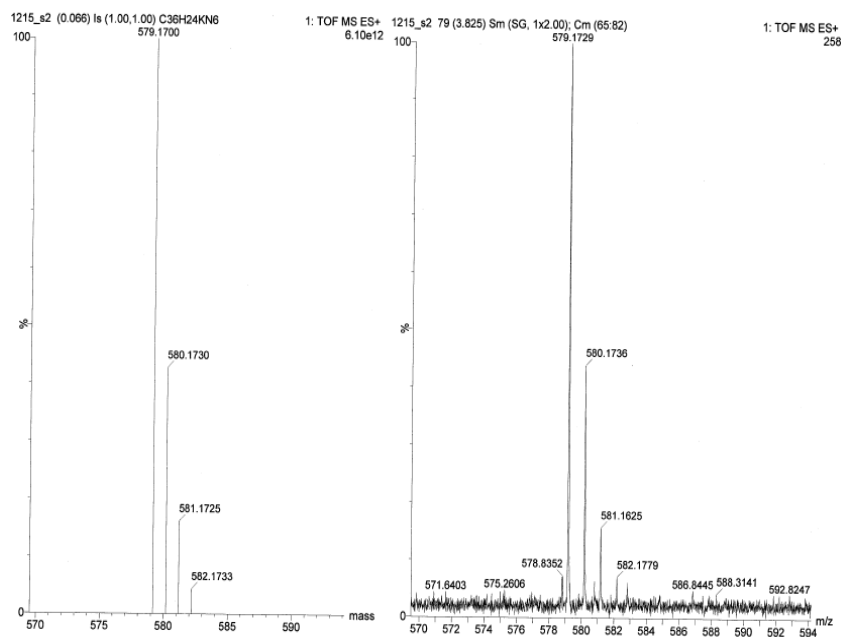
Theoretical isotopic distributions of $[\text{K}(\text{phen})_2]^+$

Experimental isotopic distributions of $[\text{K}(\text{phen})_2]^+$



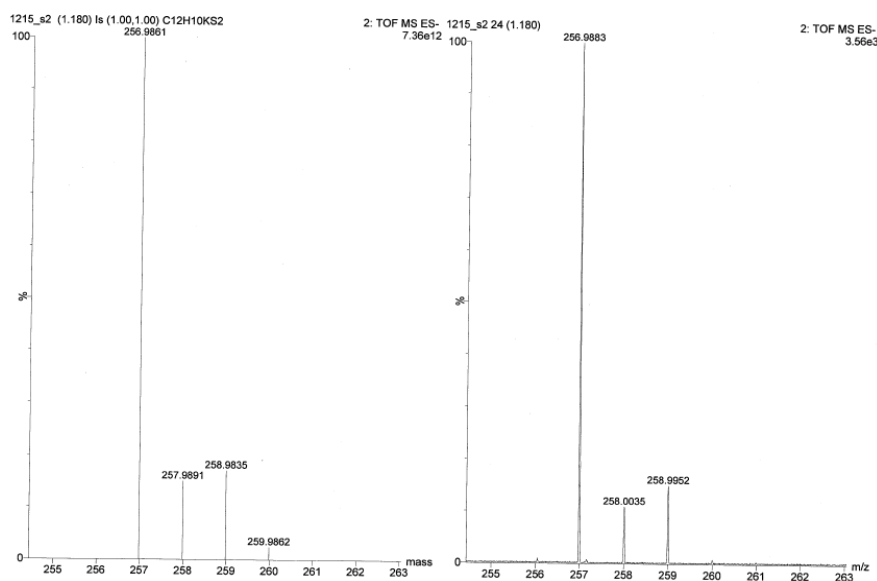
Theoretical isotopic distributions of $[\text{K}(\text{phen})_3]^+$

Experimental isotopic distributions of $[\text{K}(\text{phen})_3]^+$



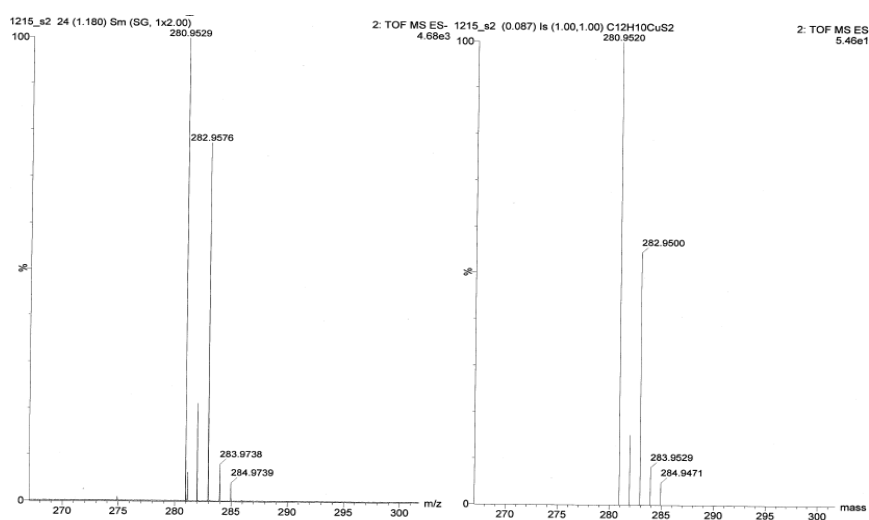
Theoretical isotopic distributions of $[\text{K}(\text{SPh})_2]^-$

Experimental isotopic distributions of $[\text{K}(\text{SPh})_2]^-$

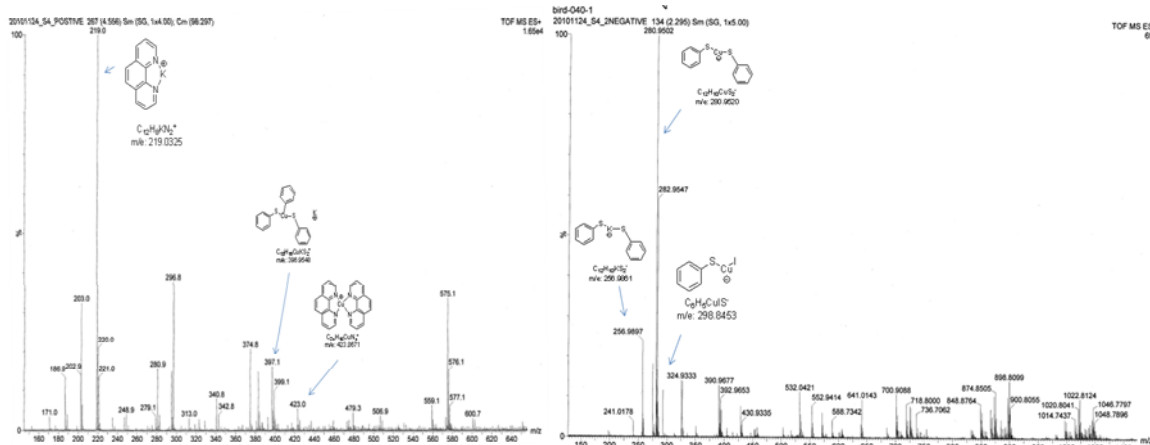


Theoretical isotopic distributions of $[\text{Cu}(\text{SPh})_2]^-$

Experimental isotopic distributions of $[\text{Cu}(\text{SPh})_2]^-$

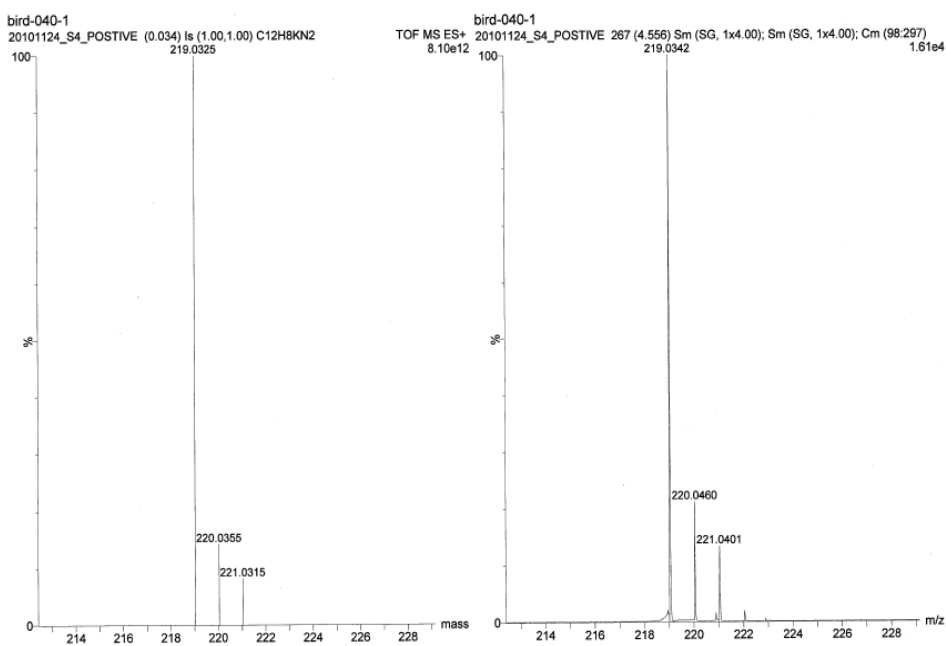


2. Experimental and theoretical isotopic distributions of $[\text{K}(\text{phen})]^+$, $[\text{Cu}(\text{SPh})_2(\text{phenyl})\text{K}]^+$, $[\text{Cu}(\text{phen})_2]^+$, $[\text{K}(\text{SPh})_2]^-$, $[\text{Cu}(\text{SPh})_2]^-$ and $[\text{Cu}(\text{SPh})\text{I}]^-$ (from the ESI-MS spectra of the reaction solution during the reaction of thiophenol and iodobenzene with *t*BuOK in the presence of CuI and phen in toluene at 120 °C).



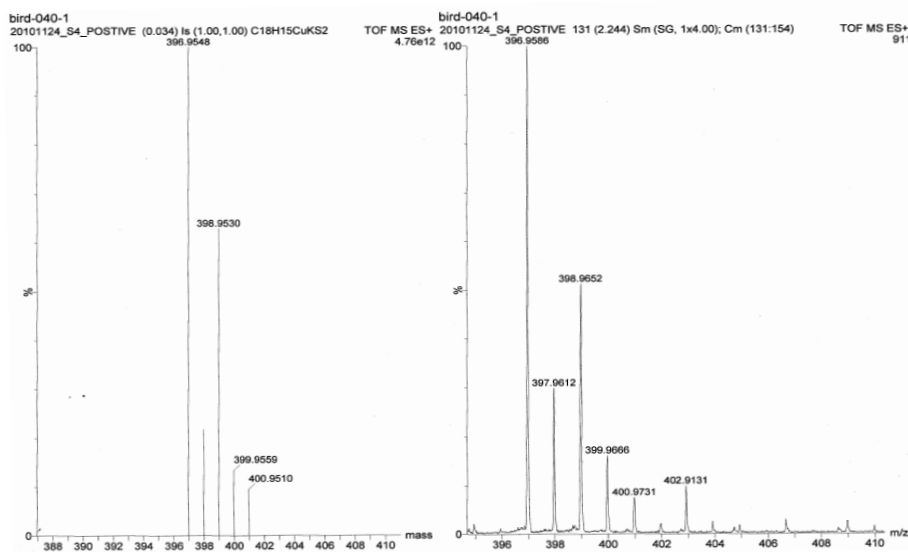
Theoretical isotopic distributions of $[\text{K}(\text{phen})]^+$

Experimental isotopic distributions of $[\text{K}(\text{phen})]^+$



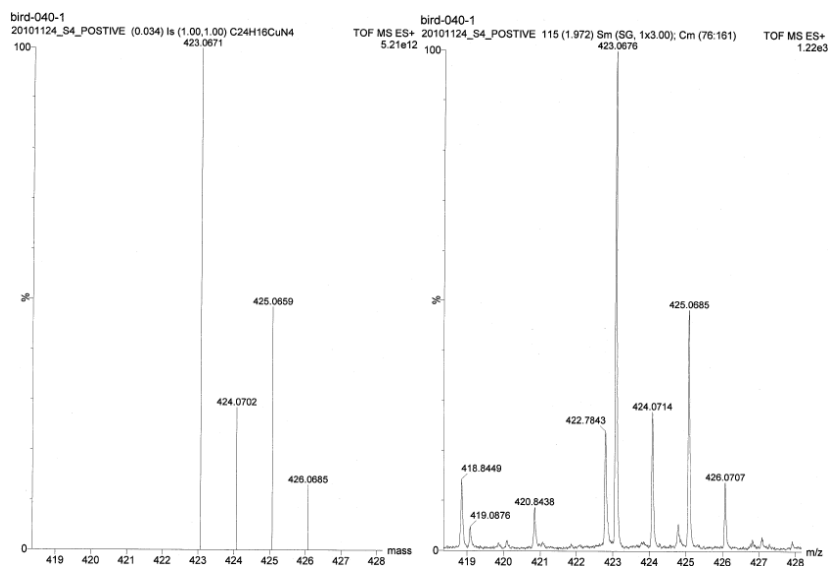
Theoretical isotopic distributions of $[\text{Cu}(\text{SPh})_2(\text{Ph})\text{K}]^+$

Experimental isotopic distributions of $[\text{Cu}(\text{SPh})_2(\text{Ph})\text{K}]^+$



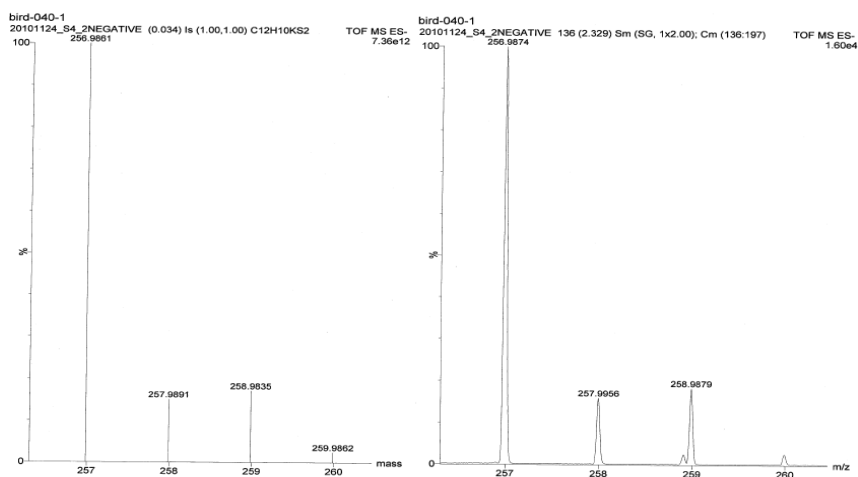
Theoretical isotopic distributions of $[\text{Cu}(\text{phen})_2]^+$

Experimental isotopic distributions of $[\text{Cu}(\text{phen})_2]^+$



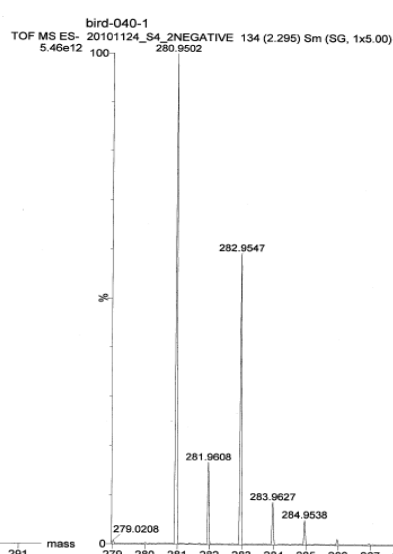
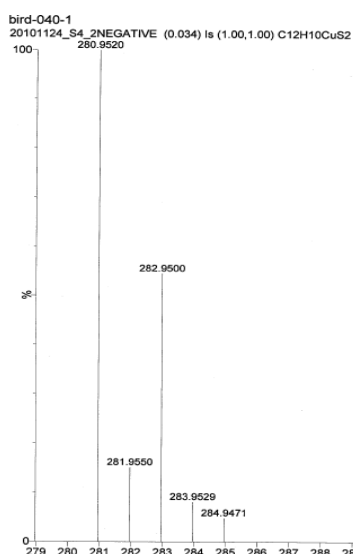
Theoretical isotopic distributions of $[\text{K}(\text{SPh})_2]^-$

Experimental isotopic distributions of $[\text{K}(\text{SPh})_2]^-$



Theoretical isotopic distributions of $[\text{Cu}(\text{SPh})_2]^-$

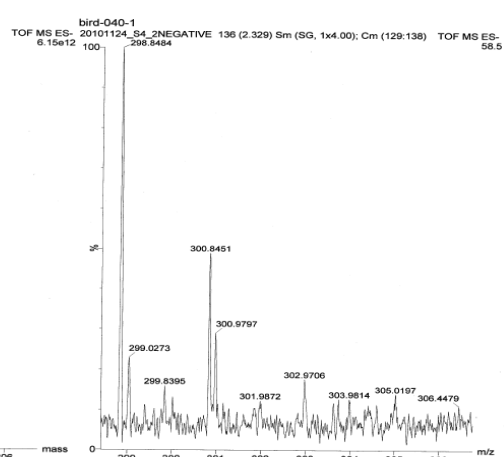
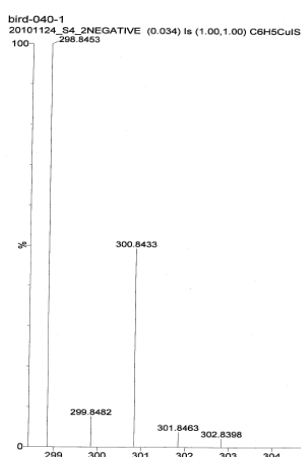
Experimental isotopic distributions of $[\text{Cu}(\text{SPh})_2]^-$



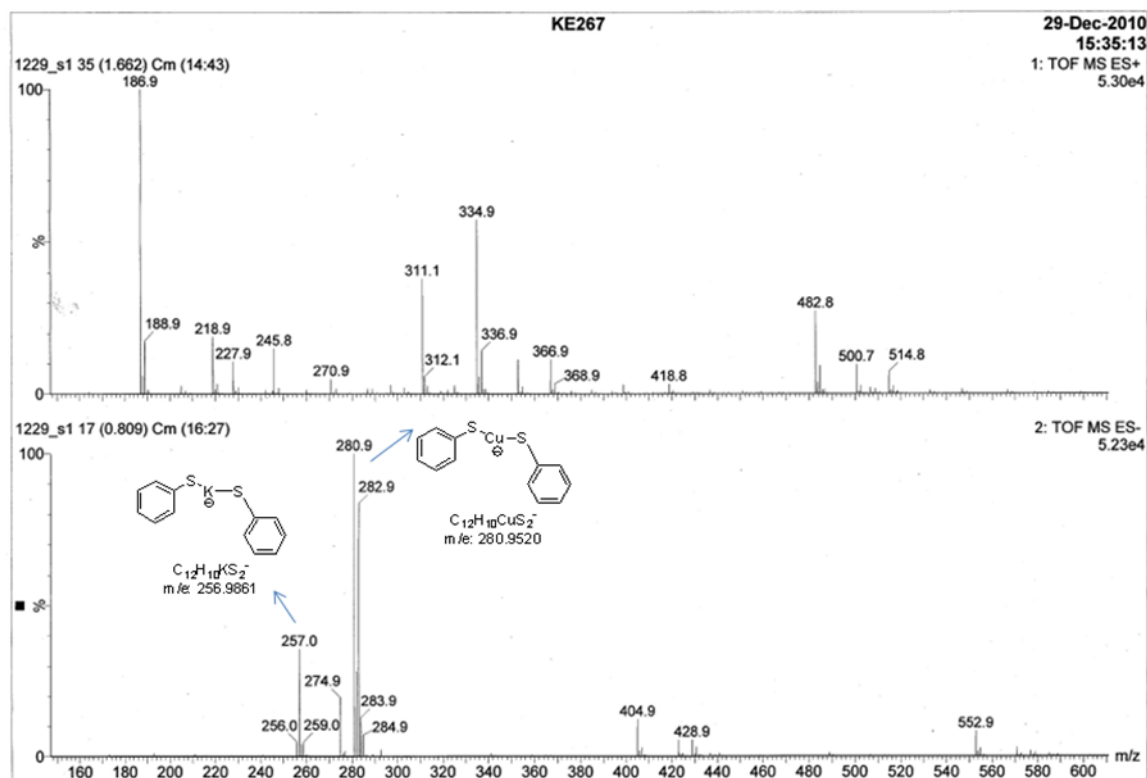
TOF MS ES-
694

Theoretical isotopic distributions of $[\text{Cu}(\text{SPh})\text{I}]^-$

Experimental isotopic distributions of $[\text{Cu}(\text{SPh})\text{I}]^-$

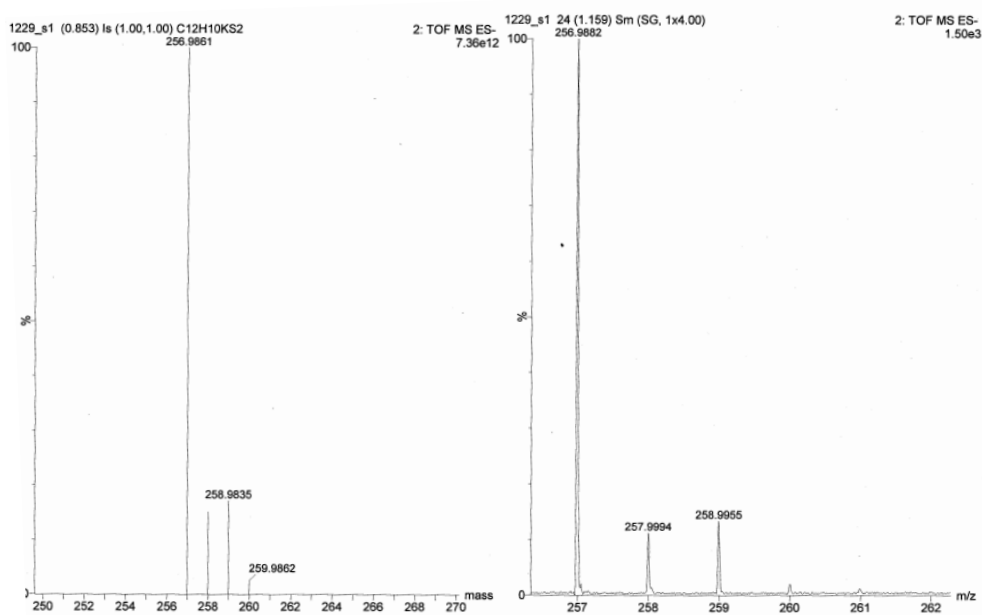


3. Experimental and theoretical isotopic distributions of $[\text{K}(\text{SPh})_2]^-$ and $[\text{Cu}(\text{SPh})_2]^-$ (from the ESI-MS spectra of the reaction solution during the reaction of thiophenol and iodobenzene with *t*BuOK in the presence of CuI in toluene at 120 °C).



Theoretical isotopic distributions of $[\text{K}(\text{SPh})_2]^-$

Experimental isotopic distributions of $[\text{K}(\text{SPh})_2]^-$



Theoretical isotopic distributions of $[\text{Cu}(\text{SPh})_2]^-$

Experimental isotopic distributions of $[\text{Cu}(\text{SPh})_2]^-$

