Electronic Supplementary Information

Addition of halide to π -bond directly from aqueous NaX solution: A general strategy for installation of two different functional groups

Palash Pandit, Krishnanka S. Gayen, Saikat Khamarui, Nirbhik Chatterjee and Dilip K. Maiti*

Department of Chemistry, University of Calcutta, University College of Science, 92, A. P. C. Road, Kolkata-700009, India

> *Corresponding author. Fax: 91-33-2351 9755, Tel: 91-33-2350 1014 <u>maitidk@yahoo.com</u>

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1. Materials and Methods

All reagents were purchased from commercial suppliers and used without further purification, unless otherwise specified. Commercially supplied ethyl acetate and petroleum ether were distilled before use. CH_2Cl_2 was dried by distillation over P_2O_5 . Petroleum ether used in our experiments was in the boiling range of 60°-80° C. Column chromatography was performed on silica gel (60-120 mesh, 0.120 mm-0.250 mm). Analytical thin layer chromatography was performed on 0.25 mm extra hard silica gel plates with UV254 fluorescent indicator. Melting points are reported uncorrected. ¹H NMR and ¹³C NMR spectra (Bruker Advance 300) were recorded at ambient temperature using 300 MHz spectrometers (300 MHz for ¹H and 75 MHz for ¹³C). Chemical shift is reported in ppm from internal reference tetramethylsilane and coupling constant in Hz. Proton multiplicities are represented as s (singlet), bs (broad singlet), d (doublet), dd (double doublet), t (triplet), q (quartet), and m (multiplet). Infrared spectra were recorded on FT-IR spectrometer (Perkin Elmer Spectrum 100) as KBr pellets (solid sample) and in thin film on NaCl window (liquid sample). Optical rotation of the chiral compounds was measured in a polarimeter (Perkin Elmer 343) using standard 10 cm quartz cell in sodium-D lamp at ambient temperature. EI-MS analysis was performed in GC-MS machine (Perkin Elmer Clarus 600) using column Elite 5 MS (30 m x 0.25 mm x 0.25 μ m) with maximum temperature 300° C. HR-MS data were acquired by electron spray ionization technique on a O-tof-micro quadriple mass spectrophotometer (Bruker). Single crystal X-ray diffraction studies of the crystalline heterocyclic compound were done in Xray diffractometer (Bruker Smart Apex-II).

2. Generation of nanoreactor by self-assembled surfactant



Figure 1: Formation of nanoreactor by self-assembled aggregation

Nature uses well-defined reaction environment for the entire essential chemical reactions through construction of extremely complex assemblies in cell.¹ It provokes synthetic chemists how to use this idea to enhance the synthetic efficiency of chemical conversions in a designed self-assembled reactor of nanometer size. Above the CMC suitable surfactants can generate an organized media (AOM)² in water or organic solvents consisting of nanoreactor.³ It is an unique example of such confined environment. Inside of the aqueous micelle (Figure 1) is sufficiently lipophilic in nature and makes both organic substrates and reagents soluble. Moreover, it not only brings them in close proximity toward enhancing the feasibility of chemical transformations but also protects water-labile reagents, reaction intermediates etc. from hydrolytic decomposition.⁴ It simultaneously can exchange ionic, polar, non-polar organic and inorganic components through the interface thereby shifting the equilibrium in forward direction and making even the inaccessible chemical transformation feasible with much faster reaction rate and desired selectivities. Sphere-like assembly of amphiphilic surfactants formed on dissolving surfactants in water was confirmed by the inverted polarizing optical microscope image of its thin film (Figure 2).



Figure 2: Polarizing optical microscope image of the nanoreactor formed in aqueous CTAB (10%)

3. General procedure and characterization data of α , α -dihaloketones (3a-g)

A solution of iodosobenzenediacetate (2.0 mmol) in water (10 mL) was taken in a round-bottom flask (25 mL) and stirred for 30 min. Alkyne **1** (1 mmol) was added under vigorous stirring followed by the addition of surfactant (CTAB/aliquat - 10 mol %). Progress of the reaction was monitored by TLC and the reaction was complete after 2.5-4.5 h depending on the substrates used. The post-reaction mixture was extracted with ethyl acetate (3 × 10 mL), and the combined organic portion was washed with brine solution (2 × 10 mL), dried on activated sodium sulfate, and concentrated in a rotary evaporator under reduced pressure at room temperature. Thus, the reaction with phenyl acetylene (**1a**, 102 mg, 1.0 mmol) afforded **3a** after purification by column chromatography on silica gel (60-120 mesh) with ethyl acetate-petroleum ether (1:9, v/v) as an eluent in an isolated yield of 83% (228 mg, 0.83 mmol). The α , α -dihaloketones (**3a**-**g**) were characterized by NMR, FT-IR and Mass (EI-MS and HR-MS) spectral analysis.

3.1. Compound $3a^5$



Yield: 83 % (228 mg, 0.83 mmol). Characteristic: viscous liquid. ¹H NMR (300 MHz, CDCl₃): δ 6.72 (1H, s), 7.48-7.54 (2H, m), 7.61-7.67 (1H, m), 8.07-8.10 (2H, m). ¹³C NMR (75 MHz, CDCl₃): δ 39.6, 128.9, 129.7, 130.8, 134.4, 185.9. EI-MS (m/z): 278 (M+2), 276 (M⁺), 235, 171, 169, 145, 143, 120, 105, 90, 77. IR (neat, cm⁻¹): 1091, 1475, 1601, 2926, 3446. HR-MS (m/z) for C₈H₆Br₂O (M⁺): Calculated 275.8785, found 275.8772 (one of the peaks).

3.2. Compound 3b



Yield: 65 % (122 mg, 0.65 mmol). Characteristic: brown Oil. ¹H NMR (300 MHz, CDCl₃): δ 5.51 (1H, s), 7.36 (2H, m), 7.54 (1H, q, *J* = 8.1 Hz), 8.23 (1H, d, *J* = 7.8 Hz), 8.09 (1H, d, *J* = 7.2 Hz). ¹³C NMR (75 MHz, CDCl₃): δ 66.4, 127.8, 128.9, 129.9, 133.3, 162.3. EI-MS (m/z): 190 (M+2), 188 (M⁺), 126, 107, 105, 96, 82, 58. IR (neat, cm⁻¹): 1091, 1475, 1601, 2926, 3446. HR-MS (*m*/*z*) for C₈H₆Cl₂O (M⁺): Calculated 187.9796, found 187.9725 (one of the peaks).

3.3. Compound 3c



Yield: 80 % (291 mg, 0.80 mmol).

Characteristic: solid.

Melting Point: 83 °C.

¹H NMR (300 MHz, CDCl₃): δ 3.85 (3H, s), 5.05 (2H, s), 6.62 (1H, s), 7.08 (2H, d, J = 4.5 Hz), 7.35 (1H, m), 10.50 (1H, s).

¹³C NMR (75 MHz, CDCl₃): δ 56.1, 73.3, 108.8, 117.9, 119.0, 120.5, 124.6, 130.1, 152.5, 162.3, 190.4.

EI-MS (m/z): 365 (M+2), 363 (M⁺), 303, 137, 120, 82, 51.

IR (KBr, cm⁻¹): 1041, 1169, 1454, 1661, 1690, 2902, 3431.

HR-MS (m/z) for C₁₁H₁₀Br₂O₄ (M+): Calculated 363.8946, found 363.8920 (one of the peaks).

3.4. Compound 3d



Yield: 82 % (273 mg, 0.82 mmol).

Characteristic: yellow oil.

¹H NMR (300 MHz, CDCl₃): δ 3.03 (2H, s), 6.77 (1H, s), 7.02 (1H, d, J = 8.4 Hz), 7.11 (1H, t, J = 7.2 Hz), 7.54-7.59 (1H, m), 7.89 (1H, dd, J = 1.8, 7.8 Hz), 10.58 (1H, s).

¹³C NMR (75 MHz, CDCl₃): δ 68.5, 107.2, 113.2, 120.4, 121.9, 125.6, 128.5, 135.7, 159.7, 189.6. EI-MS (m/z): 336 (M+2), 334 (M⁺), 332, 278, 276, 274, 253, 225, 211, 198, 185, 170, 118, 63. IR (neat, cm⁻¹): 1036, 1103, 1161, 1283, 1456, 1479, 1595, 1697, 2863. HR-MS (*m*/*z*) for C₁₀H₈Br₂O₃ (M⁺): Calculated 333.8840, found 333.8820 (one of the peaks).

3.5. Compound 3e



Yield: 78 % (295 mg, 0.78 mmol).

Characteristic: brown solid.

Melting Point: 78 °C.

¹H NMR (300 MHz, CDCl₃): δ 5.08 (2H, s), 6.78 (1H, s), 7.07 (1H, d, J = 9.0 Hz), 8.37 (1H, dd, J = 2.7, 9.0 Hz), 8.67 (1H, d, J = 2.7 Hz), 10.46 (1H, s).

¹³C NMR (75 MHz, CDCl₃): δ 69.2, 108.5, 113.5, 118.7, 124.6, 124.7, 125.4, 130.4, 163.2, 187.2.

EI-MS (m/z): 379 (M⁺), 341, 281, 207, 186, 171, 152, 137, 82, 80, 53.

IR (KBr, cm⁻¹): 1011, 1070, 1277, 1343, 1593, 1690, 2927, 3465.

HR-MS (m/z) for C₁₀H₇Br₂NO₅ (M⁺): Calculated 378.8691, found 378.8696 (one of the peaks).

3.6. Compound 3f



Yield: 75 % (353 mg, 0.75 mmol).

Characteristic: brown oil.

 $[\alpha]_{D}^{20}$ -13.90° (*c* 0.90, CHCl₃).

¹H NMR (300 MHz, CDCl₃): δ 1.24 (3H, s), 1.28 (3H, s), 1.35 (3H, s), 1.42 (3H, s), 3.88-3.93 (2H, m), 4.03-4.08 (2H, m), 4.27 (1H, dd, *J* = 6.0, 13.5 Hz), 4.35 (1H, s), 4.46 (1H, s), 4.55 (1H, d, *J* = 3.6 Hz), 5.83 (1H, d, *J* = 3.6 Hz), 6.60 (1H, s).

¹³C NMR (75 MHz, CDCl₃): δ 25.4, 26.2, 26.7, 26.8, 67.3, 70.8, 72.5, 81.2, 81.9, 83.0, 105.2, 106.6, 109.0, 111.9, 122.2.

EI-MS (m/z): 472 (M⁺), 281, 207, 96, 82, 58.

IR (neat, cm⁻¹): 1085, 1254, 1372, 1381, 1455, 1611, 1664, 2935, 2987.

HR-MS (m/z) for C₁₅H₂₂Br₂O₇ (M⁺): Calculated 471.9732, found 471.9722 (one of the peaks).

3.7. Compound 3g



Yield: 85 % (272 mg, 0.85 mmol).

Characteristic: viscous liquid.

¹H NMR (300 MHz, CDCl₃): δ 3.91 (2H, t, *J* = 6.9 Hz), 4.24 (2H, t, *J* = 6.6 Hz), 5.73 (1H, s), 7.29 (3H, t, *J* = 3.0 Hz), 7.48-7.51 (2H, m).

¹³C NMR (75 MHz, CDCl₃): δ 49.9, 66.7, 108.4, 126.9, 127.9, 129.0, 137.2.

EI-MS (m/z): 321 (M+2), 319 (M⁺), 264, 262, 260, 184, 183, 181, 103, 76, 74.

IR (neat, cm⁻¹): 1029, 1043, 1162, 1443, 1471, 2869, 2924.

HR-MS (m/z) for C₁₀H₁₀Br₂O₂ (M⁺): Calculated 319.9048, found 319.9120 (one of the peaks).

4. General procedure and characterization data of 1,2-halohydrins (5a-i)

A solution of iodosobenzenediacetate (2.0 mmol) in water (10 mL) was taken in a round-bottom flask (25 mL) and stirred for 30 min. Olefin **2** (1 mmol) was added under vigorous stirring followed by the addition of surfactant (CTAB/aliquat 10 mol %). Progress of the reaction was monitored by TLC and the reaction was complete after 1-2 h depending on the used substrate. The post-reaction mixture was extracted with ethyl acetate (3×10 mL), and the combined organic portion was washed with brine solution (2×10 mL), dried on activated sodium sulfate, and concentrated in a rotary evaporator under reduced pressure at room temperature. Thus, the reaction with indene (**2a**, 116 mg, 1.0 mmol) afforded **5a** after purification by column chromatography on silica gel (60-120 mesh) with ethyl acetate-petroleum ether (1:7, v/v) as an eluent in an isolated yield of 87% (184 mg, 0.87 mmol). The 1,2-bromohydrin (**5a-i**) were characterized by NMR, FT-IR and Mass (EI-MS and HR-MS) spectral analysis.

4.1. Compound $5a^6$



Yield: 87 % (184 mg, 0.87 mmol).

Characteristic: pale yellow solid.

Melting point: 119 °C.

¹H NMR (300 MHz, CDCl₃): δ 2.43 (1H, bs), 3.09-3.18 (1H, dd, J = 11.0, 16.0 Hz), 3.45-3.54 (1H, dd, J = 11.0, 23.0 Hz), 4.04-4.23 (1H, m), 5.23 (1H, d, J = 5.4 Hz), 7.12-7.22 (3H, m), 7.31-7.33 (1H, m).

¹³C NMR (75 MHz, CDCl₃): δ 40.4, 54.5, 83.4, 124.0, 124.5, 127.6, 129.0, 139.7, 141.6. EI-MS (m/z): 214 (M+2), 212 (M⁺), 196, 194, 133, 115.

IR (KBr, cm⁻¹): 1064, 1166, 1344, 1465, 2924, 3234.

HR-MS (m/z) for C₉H₉BrO (M⁺): calculated 211.9837, found 211.9830 (one of the peaks).

4.2. Compound $\mathbf{5b}^7$



Yield: 72 % (128 mg, 0.72 mmol).

Characteristic: colorless liquid. ¹H NMR (300 MHz, CDCl₃): δ 1.19 (3H, m), 1.60-1.84 (3H, m), 2.04-2.08 (1H, m), 2.24-2.30 (1H, m), 2.50 (1H, bs), 3.49-3.57 (1H, m), 3.77-3.87 (1H, m). ¹³C NMR (75 MHz,CDCl₃): δ 24.1, 26.6, 33.5, 36.2, 61.8, 75.3. EI-MS (m/z): 180 (M+2), 178 (M⁺), 137, 134, 132, 100, 99, 83, 82, 81, 57. IR (neat, cm⁻¹): 1075, 1120, 1450, 1494, 1606, 2860, 2937, 3433. HR-MS (m/z) for C₆H₁₁BrO (M⁺): Calculated 177.9993, found 177.9990 (one of the peaks).

4.3. Compound $5c^8$



Yield: 82 % (188 mg, 0.82 mmol).

Characteristic: yellow oil.

¹H NMR (300 MHz, CDCl₃): δ 3.80 (1H, dd, J = 4.5, 7.8 Hz), 3.94 (1H, dd, J = 4.8, 12.3 Hz), 4.23 (1H, d, J = 6.3 Hz), 4.94 (1H, d, J = 5.7 Hz), 7.26-7.31 (5H, m).

¹³C NMR (75 MHz, CDCl₃): δ 59.5, 64.2, 76.8, 126.5, 128.4, 128.6, 140.3.

EI-MS (m/z): 232 (M+2), 230 (M⁺), 182, 184, 133, 107, 105, 79, 77.

IR (neat, cm⁻¹): 1077, 1453, 1494, 2925, 3399.

HR-MS (m/z) for C₉H₁₁BrO₂ (M⁺): Calculated 229.9942, found 229.9918 (one of the peaks).

4.4. Compound 5d



Yield: 70 % (180 mg, 0.70 mmol).

Characteristic: viscous liquid.

¹H NMR (300 MHz, CDCl₃): δ 3.11 (1H, bs), 3.52-3.62 (1H, m), 3.98 (1H, d, *J* = 3.3 Hz), 4.11-4.24 (1H, m), 4.34 (2H, d, *J* = 3.6 Hz), 6.80-7.03 (2H, m), 7.45-7.56 (1H, m), 7.80-7.82 (1H, m), 10.29 (1H, s).

¹³C NMR (75 MHz, CDCl₃): δ 50.6, 64.1, 69.3, 112.7, 121.6, 125.0, 129.9, 136.1, 159.9, 186.9.

EI-MS (m/z): 260 (M+2), 258 (M⁺), 179, 165, 161, 136, 135, 121, 107, 105, 77.

IR (neat, cm⁻¹): 1242, 1455, 1484, 1599, 1686, 2927, 3433.

HR-MS (m/z) for C₁₀H₁₁BrO₃ (M⁺): Calculated 257.9892, found 257.9881 (one of the peaks).

4.5. Compound 5e



Yield: 85 % (295 mg, 0.85 mmol).

Characteristic: yellow viscous liquid.

¹H NMR (300 MHz, CDCl₃): δ 2.68 (3H, s), 4.58-4.68 (1H, m), 4.72-4.81 (2H, m), 5.28 (1H, d, J = 10.2 Hz), 6.83 (1H, d, J = 8.7 Hz), 7.28-7.35 (6H, m), 7.38-7.52 (1H, m), 7.76-7.81 (1H, m).

¹³C NMR (75 MHz,CDCl₃): δ 32.2, 52.3, 52.5, 71.9, 114.6, 118.1, 127.8, 128.8, 129.2, 130.2, 133.3, 136.1, 139.1, 156.1, 197.8.

EI-MS (m/z): 348 (M⁺), 252, 250, 203, 120, 105, 91, 79, 77.

IR (neat, cm⁻¹): 1066, 1149, 1225, 1395, 1454, 1481, 1588, 1678, 2924, 3032, 3065, 3338. HR-MS (m/z) for C₁₇H₁₇BrO₃ (M⁺): Calculated 348.0361, found 348.0345 (one of the peaks).

4.6. Compound 5f



Yield: 83 % (267 mg, 0.83 mmol).

Characteristic: yellowish oil.

¹H NMR (300 MHz, CDCl₃): δ 2.68 (3H, m), 3.87-3.97 (2H, m), 4.29-4.41 (2H, m), 4.48-4.56 (1H, m), 7.49-7.55 (2H, m), 7.58-7.63 (2H, m), 7.76-7.82 (1H, m), 8.29-8.33 (1H, m).

¹³C NMR (75 MHz,CDCl₃): δ 31.0, 32.7, 48.4, 123.3, 124.7, 125.1, 127.0, 127.9, 128.0, 128.3, 128.6, 136.5, 153.9, 200.5.

EI-MS (m/z): 322 (M⁺), 282, 281, 207, 133, 115, 105, 82, 57.

IR (neat, cm⁻¹): 1074, 1244, 1272, 1337, 1360, 1681, 3058, 3436.

HR-MS (m/z) for C₁₅H₁₅BrO₃ (M⁺): Calculated 322.0205, found 322.0299 (one of the peaks).

4.7. Compound 5g



Yield: 86 % (319 mg, 0.86 mmol).

Characteristic: viscous liquid.

 $[\alpha]_{D}^{20}$ -40.23° (*c* 0.85, CHCl₃).

¹H NMR (300 MHz, CDCl₃): δ 1.33 (3H, s), 1.51 (3H, s), 3.62-3.70 (2H, m), 4.10 (1H, d, J = 2.7 Hz), 4.38-4.46 (3H, m), 4.69-4.73 (2H, m), 5.98 (1H, d, J = 3.9 Hz), 7.33-7.38 (5H, m).

¹³C NMR (75 MHz,CDCl₃): δ 26.4, 26.8, 34.7, 49.5, 71.8, 81.4, 82.1, 82.2, 104.7, 112.2, 128.0, 128.3, 128.7, 136.5.

EI-MS (m/z): 374 (M+2), 372 (M⁺), 251, 249, 209, 207, 127, 91, 71, 69.

IR (neat, cm⁻¹): 1075, 1106, 1163, 1215, 1374, 1388, 1455, 2931, 2988.

HR-MS (m/z) for C₁₆H₂₁BrO₅ (M⁺): Calculated 372.0572, found 372.0555 (one of the peaks).

4.8. Compound 5h⁸



Yield: 80 % (208 mg, 0.80 mmol). Characteristic: violet solid.

Melting point: 95 °C.

¹H NMR (300 MHz, CDCl₃): δ 1.93 (1H, bs), 3.10-3.27 (1H, m), 3.46-3.55 (1H, m) 4.09-4.23 (1H, m), 5.32 (1H, d, J = 6.3 Hz), 7.11-7.41 (4H, m).

¹³C NMR (75 MHz,CDCl₃): δ 30.1, 42.3, 85.0, 123.8, 124.3, 127.5, 128.8, 140.9, 142.0.

EI-MS (m/z): 261 (M+1), 260 (M⁺), 243, 242, 133, 116, 115, 105, 79.

IR (KBr, cm⁻¹): 1065, 1113, 1342, 1460, 2925, 3324.

HR-MS (m/z) for C₉H₉IO (M⁺): Calculated 259.9698, found 259.9690 (one of the peaks).

4.9. Compound 5i



Yield: 71 % (119 mg, 0.71 mmol).

Characteristic: solid.

Melting point: 101 °C.

¹H NMR (300 MHz, CDCl₃): δ 3.02 (1H, dd, *J* = 7.2, 15.9 Hz), 3.44 (1H, dd, *J* = 7.2, 15.9 Hz), 4.18-4.25 (1H, m), 5.22 (1H, s), 7.13-7.40 (5H, m).

¹³C NMR (75 MHz,CDCl₃): δ 39.8, 64.6, 83.0, 124.1, 124.6, 127.6, 129.0, 139.0, 141.4.

EI-MS (m/z): 170 (M+2), 168 (M⁺), 150, 133, 132, 131, 115, 105, 103, 77, 57.

IR (KBr, cm⁻¹): 1066, 1348, 1461, 2369, 3215, 3749.

HR-MS (m/z) for C₉H₉ClO (M⁺): Calculated 168.0342, found 168.0340 (one of the peaks).

5. General procedure and characterization data of 1,2-haloacetates (7a-f)

A mixture of alkene 2 (1.0 mmol), anhydrous MgSO₄ (0.5 g) and dry dichloromethane (15 mL) were taken together in a round-bottom flask (25 mL) and content was cooled to 0° C. PhI(OAc)₂ (644 mg, 2.0 mmol) and surfactant (CTAB/aliquat 10 mole%), NaBr (1.0 mm) were added successively. Then the content was allowed to attain room temperature. The reaction was complete after 1.5-2.5 h. The post-reaction mixture was extracted with dichloromethane (3 × 10 mL) and the combined organic portion was washed with brine solution (2 × 10 mL), dried on activated sodium sulfate, and concentrated in a rotary evaporator under reduced pressure at room temperature. Thus, the reaction with indene (**2a**, 116 mg, 1.0 mmol) afforded **7a** after purification by column chromatography on silica gel (60-120 mesh) with ethyl acetate-petroleum ether (1:8, v/v) as an eluent in an isolated yield of 75% (190 mg, 0.75 mmol). The 1,2-haloacetate (**7a-f**) were characterized by NMR, FT-IR and Mass (EI-MS and HR-MS) spectral analysis.

5.1. Compound 7a



Yield: 75 % (190 mg, 0.75 mmol). Characteristic: colorless liquid. ¹H NMR (300 MHz, CDCl₃): δ 2.04 (3H, s), 3.20 (1H, dd, J = 4.2, 17.1 Hz), 3.65 (1H, dd, J = 6.6, 17.1 Hz), 4.41-4.46 (1H, m), 6.27 (1H, d, J = 3.6 Hz), 7.19-7.42 (4H, m). ¹³C NMR (75 MHz, CDCl₃): δ 21.0, 41.4, 49.9, 83.9, 124.8, 125.8, 127.6, 129.7, 138.4, 141.2, 170.3. EI-MS (m/z): 256 (M+2), 254 (M⁺), 211, 195, 193, 173, 132, 131, 115, 103, 77. IR (neat, cm⁻¹): 1018, 1048, 1227, 1371, 1740, 2926. HR-MS (*m*/*z*) for C₁₁H₁₁BrO₂ (M⁺): Calculated 253.9942, found 253.9930 (one of the peaks).

5.2. Compound $7b^8$



Yield: 71 % (171 mg, 0.71 mmol).

Characteristic: yellow oil.

¹H NMR (300 MHz, CDCl₃): δ 2.14 (3H, s), 3.56-3.76 (2H, m), 5.98 (1H, dd, *J* = 4.8, 7.8 Hz), 7.31-7.36 (5H, m).

¹³C NMR (75 MHz,CDCl₃): δ 20.9, 34.2, 74.8, 126.5, 128.7, 128.8, 137.7, 169.8.

EI-MS (m/z): 242 (M⁺), 184, 182, 164, 162, 120, 107, 103, 79, 77.

IR (neat, cm⁻¹): 1074, 1237, 1494, 1747, 2926.

HR-MS (m/z) for C₁₀H₁₁BrO₂ (M⁺): Calculated 241.9942, found 241.9935 (one of the peaks).

5.3. Compound 7c



Yield: 79 % (248 mg, 0.79 mmol).

Characteristic: yellow oil.

¹H NMR (300 MHz, CDCl₃): δ 1.32 (3H, t, *J* = 6.9 Hz), 2.04 (3H, s), 4.24-4.37 (2H, m), 4.51 (1H, d, *J* = 9.3 Hz), 6.15 (1H, d, *J* = 10.2 Hz), 7.32-7.42 (5H, m).

¹³C NMR (75 MHz,CDCl₃): δ 13.9, 20.7, 46.4, 62.3, 75.5, 127.9, 128.5, 129.1, 136.0, 167.6, 168.6.

EI-MS (m/z): 314 (M⁺), 254, 252, 193, 149, 147, 131, 120, 107, 103, 91, 77.

IR (neat, cm⁻¹): 1020, 1145, 1454, 1737, 2853, 2925.

HR-MS (m/z) for C₁₃H₁₅BrO₄ (M⁺): Calculated 314.0154, found 314.0151 (one of the peaks).

5.4. Compound 7d



Yield: 84 % (243 mg, 0.84 mmol). Characteristic: viscous liquid.

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¹H NMR (300 MHz, CDCl₃): δ 2.13 (3H, s), 3.42-3.48 (2H, m), 5.88 (1H, dd, J = 6.0, 7.5 Hz), 7.36 (5H, s). ¹³C NMR (75 MHz,CDCl₃): δ 7.6, 20.9, 75.1, 126.3, 128.6, 128.7, 138.4, 169.6.

EI-MS (m/z): 289 (M⁺), 184, 169, 162, 142, 120, 104, 91, 77.

IR (neat, cm⁻¹): 1049, 1240, 1453, 1494, 1722, 2926.

HR-MS (m/z) for C₁₀H₁₁IO₂ (M⁺): Calculated 289.9804, found 289.9810 (one of the peaks).

5.5. Compound $7e^9$



Yield: 71 % (140 mg, 0.71 mmol).

Characteristic: viscous liquid.

¹H NMR (300 MHz, CDCl₃): δ 2.07 (3H, s), 3.62-3.69 (2H, m), 5.89 (1H, dd, *J* = 4.8, 7.8 Hz), 7.26 (5H, s).

¹³C NMR (75 MHz,CDCl₃): δ 20.9, 46.4, 75.0, 126.6, 128.6, 128.8, 137.2, 170.0.

EI-MS (m/z): 200 (M+2), 198 (M⁺), 163, 162, 140, 139, 138, 122, 120, 121, 107, 103, 79, 77.

IR (neat, cm⁻¹): 1024, 1229, 1373, 1403, 1455, 1746, 2855, 2928.

HR-MS (m/z) for C₁₀H₁₁ClO₂ (M⁺): Calculated 198.0448, found 198.0344 (one of the peaks).

5.6. Compound 7f



Yield: 76 % (332 mg, 0.76 mmol).

Characteristic: viscous liquid.

¹H NMR (300 MHz, CDCl₃): δ 1.31 (3H, s), 1.33 (3H, s), 1.41 (3H, s), 1.48 (3H, s), 2.10 (3H, s), 3.50-3.60 (1H, m), 3.71-3.79 (1H, m), 3.93-4.41 (7H, m), 5.43 (1H, t, *J* = 3.6 Hz), 5.06-5.10 (1H, m), 5.86 (1H, d, *J* = 3.6 Hz).

¹³C NMR (75 MHz,CDCl₃): δ 20.8, 26.7, 26.8, 29.8, 30.2, 30.7, 30.8, 46.9, 53.3, 64.4, 67.4, 67.5, 69.0, 69.1, 69.2, 69.4, 70.8, 71.1, 71.3, 71.5, 72.2, 72.3, 81.2, 82.4, 82.5, 82.9, 83.0, 83.7, 105.2, 109.0, 11.95, 169.9, 170.1.

EI-MS (m/z): 440 (M+2), 438 (M⁺), 385, 357, 199, 201, 127, 101, 72, 51.

IR (neat, cm⁻¹): 1078, 1216, 1256, 1373, 1456, 1618, 1720, 2935, 2981, 3450.

HR-MS (m/z) for C₁₇H₂₇BrO₈ (M⁺): Calculated 438.0889, found 438.0852 (one of the peaks).

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6. ¹H and ¹³C spectra of the compounds (3a-g, 5a-i and 7a-f)

















S-20











383 362 334 260 -1.513-1.335693 507 168 Br но BnO **O SI Figure 27.** ¹H NMR spectrum of compound 5g. .1 9 8 7 6 5 4 3 2 ppm 128.37 104.71 26.89 26.40 112.23 49.58 34.75 36.59 71.84 28 42 44 60 28 Br Ю BnO Ό **SI Figure 28.** ¹³C NMR spectrum of compound **5**g. 60 50 40 30 20 ppm 70 80 160 150 140 130 120 110 100 90 170 200 100 180

















7. Summary of data CCDC 799249 (compound 5a)



. Chemical formula and formula weight (M): C ₉ H ₉ BrO and 213.07		
Monoclinic		
a 12.286(8), b 4.978(3), c 14.499(9),		
90.00, 106.853(8), 90.00, 848.7(9)		
296 (2)		
P21/n		
4		
1420		
0.0528		