## Supporting Information

# Modular thiophene dendrons and dendrimers with peripheral 

## functional groups

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## Contents

Materials ..... S2
Instrumentation ..... S2
Synthesis ..... S2-S13
Reference ..... S13
Figure S1 ..... S14
Figure S2 ..... S15
Table S1 ..... S15
${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR Spectra ..... S16-S79
High Resolution Mass Spectrum (HRMS) ..... S80-S86

## Materials.

2-(6-bromohexyl)-thiophene, ${ }^{1}$ and 4,7-dibromo-1,10-phenanthroline ${ }^{2}$ were synthesized according to the literature procedure. 3,5,3',5'-Tetrabromo-[2,2']bithiophene was synthesized as reported earlier. ${ }^{3}$ A second generation carbazole-terminated Fréchet-type polybenzylether dendrons (4Cbz) have been synthesized in our group. ${ }^{4} \mathrm{CH}_{2} \mathrm{Cl}_{2}$ for spectroscopic measurements was spectrophotometric grade. THF was freshly distilled from sodium benzophenone ketyl. All other commercially available reagents were purchased from Aldrich and used as received. Silica gel ( $60 \AA, 32-63 \mu \mathrm{~m}$, Standard Grade) was purchased from Sorbent Technologies, Inc. (Atlanta, GA).

## Instrumentation

Nuclear magnetic resonance (NMR) spectra were recorded on a General Electric QE-300 spectrometer operating at 300 MHz for ${ }^{1} \mathrm{H}$ and 75 MHz for ${ }^{13} \mathrm{C}$ nuclei. UV-vis spectra were recorded on an Agilent 8453 UV-visible Spectrometer and fluorescence spectra on a Perkin Elmer LS 45 Luminescence Spectrometer. High resolution mass spectra were recorded on a QSTAR XL MS/MS (quadrupole/TOF hybrid) mass spectrometer (Applied Biosystems/MDS Sciex, Foster City, CA, USA) at Mass Spectrometry Center, Department of Chemistry, University of Tennessee Knoxville. Elemental analyses were carried out by Galbraith Laboratories, Inc. (Knoxville, TN).

## Synthesis.

Synthesis of 2-(6-Methoxyl-hexyl)-thiophene (1T6COMe). 70 mL of methanol was added dropwise to sodium $(2.5 \mathrm{~g}, 108.7 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$. After addition, it was warmed up to room temperature and stirred for 30 minutes. A solution of 2-(6-bromohexyl)-thiophene (14.8 g, 60 mmol ) in 30 mL of methanol was added dropwise. After addition, the reaction mixture was heated up to refluxing overnight and then cooled to room temperature. Methanol was removed by rotary evaporation. The residue was
poured into 50 g of water-ice and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 50 \mathrm{~mL})$. The extract was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The product was obtained as a colorless liquid by vacuum distillation ( $10.2 \mathrm{~g}, 85.7 \%$ ).
${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 7.11(\mathrm{dd}, \mathrm{J}=5.1 \mathrm{~Hz}, 1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{dd}, \mathrm{J}=5.1 \mathrm{~Hz}, 3.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.79$ $(\mathrm{m}, 1 \mathrm{H}), 3.38(\mathrm{t}, \mathrm{J}=6.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.34(\mathrm{~s}, 3 \mathrm{H}), 2.84(\mathrm{t}, \mathrm{J}=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.71(\mathrm{~m}, 2 \mathrm{H}), 1.59(\mathrm{~m}, 2 \mathrm{H}), 1.40$ (m, 4H). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): 145.57,126.51,123.80,122.62,72.70,58.42,31.59,29.70$, 29.41, 28.79, 25.74.

Synthesis of Tributyl-[5-(6-methoxy-hexyl)-thiophen-2-yl]-stannane (Sn1T6COMe). 25 mL of n butyl lithium (2.5 M in hexane, 62.5 mmol ) was added dropwise to a solution of 2-(6-methoxyl-hexyl)thiophene $(11.5 \mathrm{~g}, 58.0 \mathrm{mmol})$ in 50 mL of THF at $-78^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$. After 30 minutes of stirring upon addition, tributyltin chloride ( $20.5 \mathrm{~g}, 63.0 \mathrm{mmol}$ ) was added dropwise. The reaction was allowed to warm up to room temperature for 4 hours. THF was removed by rotary evaporation. The residue was poured into 100 mL of saturated $\mathrm{NaHCO}_{3}$ aqueous solution and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 100 \mathrm{~mL})$. The extract was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The product was used directly for the next reaction without further purifications. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 7.01(\mathrm{~d}, \mathrm{~J}=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{~d}, \mathrm{~J}=2.7$ $\mathrm{Hz}, 1 \mathrm{H}), 3.39(\mathrm{t}, \mathrm{J}=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 3.35(\mathrm{~s}, 3 \mathrm{H}), 2.89(\mathrm{t}, \mathrm{J}=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.73(\mathrm{~m}, 2 \mathrm{H}), 1.60(\mathrm{~m}, 8 \mathrm{H}), 1.38$ $(\mathrm{m}, 10 \mathrm{H}), 1.11(\mathrm{t}, \mathrm{J}=8.1 \mathrm{~Hz}, 6 \mathrm{H}), 0.93(\mathrm{t}, \mathrm{J}=7.2 \mathrm{~Hz}, 9 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 151.28$, $135.07,133.61,125.21,72.75,58.42,31.65,29.75,29.48,28.97,28.89,27.18,25.81,13.58,10.61$.

Synthesis of 5,5'’-Bis-(6-methoxy-hexyl)-[2,2';3',2’’]terthiophene (3T6COMe). In a one-necked flask was charged with 27.5 g of tributyl-[5-(6-methoxy-hexyl)-thiophen-2-yl]-stannane ( 56.4 mmol ), 4.6 g of 2,3-dibromothiophene $(19.0 \mathrm{mmol}), 1.4 \mathrm{~g}$ of $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$, and 100 mL of DMF. After three freeze-thaw cycles, the mixture was heated up to $110^{\circ} \mathrm{C}$ for 20 hours under nitrogen. After cooling to room temperature, DMF was removed by vacuum distillation and the residue was poured into water, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and washed thoroughly with NaF solution to remove tributyltin chloride. The
organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was purified by column chromatography eluting with $3: 1$ hexane/diethyl ether to give the yellow oil product $(7.9 \mathrm{~g}, 87.2 \%) .{ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 7.20(\mathrm{~d}, \mathrm{~J}=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.11(\mathrm{~d}, \mathrm{~J}=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H})$, $6.85(\mathrm{~d}, \mathrm{~J}=3.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.65(\mathrm{~m}, 2 \mathrm{H}), 3.36(\mathrm{t}, \mathrm{J}=6.0 \mathrm{~Hz}, 4 \mathrm{H}), 3.23(\mathrm{~s}, 6 \mathrm{H}), 2.78(\mathrm{~m}, 4 \mathrm{H}), 1.67(\mathrm{~m}, 4 \mathrm{H})$, $1.56(\mathrm{~m}, 4 \mathrm{H}), 1.37(\mathrm{~m}, 8 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 147.43,145.91,135.14,132.46,132.18$, $131.43,129.74,127.75,126.04,124.32,124.18,124.13,72.95,58.70,31.65,31.60,30.20,30.14,29.68$, 29.04, 26.00.

## Synthesis of [5,5’’-Bis-(6-methoxy-hexyl)-[2,2';3',2’’]terthiophen-5'-yl]-tributyl-stannane

(Sn3T6COMe). 4.0 mL of n-butyl lithium ( 2.5 M in hexane, 10.0 mmol ) was added dropwise to a solution of $5,5^{\prime}$ '-bis-(6-methoxy-hexyl)-[2, $\left.2^{\prime} ; 3^{\prime}, 2^{\prime ’}\right]$ terthiophene (3T6COMe, $4.1 \mathrm{~g}, 8.6 \mathrm{mmol}$ ) in 30 mL of THF at $-78^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$. After 30 minutes of stirring upon addition, tributyltin chloride ( 3.3 g , $10.1 \mathrm{mmol})$ was added dropwise. The reaction was allowed to warm up to room temperature for 4 hours. THF was removed by rotary evaporation. The residue was poured into 40 mL of saturated $\mathrm{NaHCO}_{3}$ aqueous solution extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 50 \mathrm{~mL})$. The extract was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The product was used directly for the next reaction without further purifications. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 7.10(\mathrm{~s}, 1 \mathrm{H}), 6.90(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{~d}, \mathrm{~J}=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.65(\mathrm{~m}, 2 \mathrm{H}), 3.36$ (t, J= $6.9 \mathrm{~Hz}, 4 \mathrm{H}), 3.32(\mathrm{~s}, 6 \mathrm{H}), 2.78(\mathrm{~m}, 4 \mathrm{H}), 1.67(\mathrm{~m}, 4 \mathrm{H}), 1.56(\mathrm{~m}, 10 \mathrm{H}), 1.37(\mathrm{~m}, 14 \mathrm{H}), 1.11(\mathrm{~m}, 6 \mathrm{H})$, 0.91 (dt, 9H, J=7.2 Hz).

Synthesis of 2,3-Di(5,5'-di(6-methoxy-hexyl)-[2,2';3',2']terthiophen-5'-yl)thiophene (7T6COMe). In a one-necked flask was charged with tributyl [5,5"-bis-(6-methoxy-hexyl)[2, 2'; $\left.3^{\prime}, 2^{\prime} ’\right]$ terthiophene-5'-yl]stannane (Sn3T6COMe, $2.3 \mathrm{~g}, 3.0 \mathrm{mmol}$ ), 2,3-dibromothiophene (242 $\mathrm{mg}, 1.0 \mathrm{mmol}), 75 \mathrm{mg}$ of $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$, and 50 mL of DMF. After three freeze-thaw cycles, the mixture was heated up to $110^{\circ} \mathrm{C}$ for 20 hours under nitrogen. After cooling to room temperature, the reaction
mixture was poured into water, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and washed thoroughly with NaF solution to remove tributyltin chloride. The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was purified by column chromatography eluting with $1: 1$ hexane/diethyl ether to give the yellow oil product ( $860 \mathrm{mg}, 83.2 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 7.27(\mathrm{~d}, \mathrm{~J}=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{~s}, 1 \mathrm{H}), 7.18(\mathrm{~s}, 1 \mathrm{H})$, $7.17(\mathrm{~d}, \mathrm{~J}=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~d}, \mathrm{~J}=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{~d}, \mathrm{~J}=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.88(\mathrm{~d}, \mathrm{~J}=3.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.86$ $(\mathrm{d}, \mathrm{J}=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.66(\mathrm{~m}, 4 \mathrm{H}), 3.36(\mathrm{t}, \mathrm{J}=6.3 \mathrm{~Hz}, 8 \mathrm{H}), 3.32(\mathrm{~s}, 12 \mathrm{H}), 2.77(\mathrm{t}, \mathrm{J}=7.5 \mathrm{~Hz}, 8 \mathrm{H}), 1.67(\mathrm{~m}$, $8 \mathrm{H}), 1.57(\mathrm{~m}, 8 \mathrm{H}), 1.38(\mathrm{~m}, 16 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 147.37,147.16,146.00,145.83$, $135.03,134.76,134.45,132.83,132.54,132.16,131.90,131.79,131.74,131.61,131.36,131.03$, $130.40,129.62,129.25,127.53,127.40,126.30,126.11,124.84,124.15,124.13,124.01,72.74,58.49$, $31.47,31.39,30.01,29.98,29.50,28.86,28.84,25.81$.

## Synthesis of 2,3-Di(5,5'-di(6-bromohexyl)-[2,2';3',2']terthiophen-5'-yl)thiophene (7T6CBr). An

 emulsion of 2,3-di(5,5"-di(6-methoxy-hexyl)-[2,2';3', $\left.2^{\prime \prime}\right]$ terthiophen- $\left.5^{\prime}-\mathrm{yl}\right)$ thiophene (7T6COMe, 0.5 g , 0.48 mmol ) in 20 mL HBr aqueous solution ( $47.0 \sim 49.0 \mathrm{wt} \%$ ) was heated up to refluxing for 20 hours. After cooling to room temperature, the reaction mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was neutralized with aqueous $\mathrm{NaHCO}_{3}$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The residue was purified by column chromatography eluting with $1: 1$ hexane $/ \mathrm{CHCl}_{3}$ to give the yellow oil product ( $360 \mathrm{mg}, 60.5 \%$ ). ${ }^{1} \mathrm{H}^{\mathrm{NMR}}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 7.29(\mathrm{~d}, \mathrm{~J}=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{~s}, 1 \mathrm{H}), 7.18(\mathrm{~s}, 1 \mathrm{H}), 7.17(\mathrm{~d}, \mathrm{~J}=5.1 \mathrm{~Hz}$, $1 \mathrm{H}), 6.95(\mathrm{~d}, \mathrm{~J}=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{~d}, \mathrm{~J}=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.88(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H})$, $6.66(\mathrm{~m}, 4 \mathrm{H}), 3.40(\mathrm{t}, \mathrm{J}=6.9 \mathrm{~Hz}, 8 \mathrm{H}), 2.78(\mathrm{t}, \mathrm{J}=7.5 \mathrm{~Hz}, 8 \mathrm{H}), 1.86$ (pentet, $\mathrm{J}=6.9 \mathrm{~Hz}, 8 \mathrm{H}$ ), 1.67 (pentet, $\mathrm{J}=7.2 \mathrm{~Hz}, 8 \mathrm{H}), 1.43(\mathrm{~m}, 16 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 147.18,146.97,145.81,145.64,135.16$, $134.91,134.59,132.86,132.66,132.31,132.04,131.84,131.78,131.63,131.42,131.08,130.45$, $129.68,129.29,127.61,127.46,126.38,126.19,124.93,124.32,124.28,124.13,33.88,32.63,31.33$, 31.26, 29.98, 29.94, 28.11, 27.84.Synthesis of 14T6COMe. In a one-necked flask was charged with tributyl [5,5''-bis-(6-methoxy-hexyl)-[2, $\left.2^{\prime} ; 3^{\prime}, 2^{\prime ’}\right]$ terthiophene-5'-yl]stannane ( $2.3 \mathrm{~g}, 3.0 \mathrm{mmol}$ ), 3,5,3',5'-tetrabromo-[2, $\left.2^{\prime}\right]$ bithiophene ( $241 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), 75 mg of $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$, and 50 mL of DMF. After three freeze-thaw cycles, the mixture was heated up to $110^{\circ} \mathrm{C}$ for 20 hours under nitrogen. After cooling to room temperature, the reaction mixture was poured into water, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and washed thoroughly with NaF solution to remove tributyltin chloride. The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was purified by column chromatography eluting with 1:2 hexane/diethyl ether to give the yellow oil product ( $260 \mathrm{mg}, 25.2 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 7.22(\mathrm{~s}, 1 \mathrm{H}), 7.19(\mathrm{~s}, 1 \mathrm{H}), 7.18(\mathrm{~s}, 1 \mathrm{H})$, $7.16(\mathrm{~s}, 1 \mathrm{H}), 7.12(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.07(\mathrm{~d}, \mathrm{~J}=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.95(\mathrm{~m}, 4 \mathrm{H}), 6.89(\mathrm{~m}, 4 \mathrm{H}), 6.67(\mathrm{~m}, 8 \mathrm{H})$, $3.37(\mathrm{~m}, 16 \mathrm{H}), 3.34(\mathrm{~s}, 24 \mathrm{H}), 2.77(\mathrm{~m}, 16 \mathrm{H}), 1.67(\mathrm{~m}, 16 \mathrm{H}), 1.58(\mathrm{~m}, 16 \mathrm{H}), 1.39(\mathrm{~m}, 32 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 147.41,147.30,147.21,146.15,146.03,145.88,137.81,135.41,134.59,134.55$, $134.49,134.30,134.19,133.80,133.31,132.27,132.03,131.99,131.92,131.89,131.69,131.14$, $130.74,130.24,129.50,128.62,127.44,127.36,126.59,126.44,126.36,126.16,125.99,124.23$, $124.18,124.14,124.07,124.00,72.69,58.43,31.44,31.34,29.97,29.93,29.46,28.80,25.77$.

Synthesis of 14 T 6 CBr . A solution of $14 \mathrm{~T} 6 \mathrm{COMe}(100 \mathrm{mg}, 0.05 \mathrm{mmol})$ in a mixture of acetic acid $(30 \mathrm{~mL})$ and HBr aqueous solution $(47.0 \sim 49.0 \mathrm{wt} \%, 10 \mathrm{~mL})$ was heated up to refluxing overnight. After cooling to room temperature, the reaction mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was neutralized with aqueous $\mathrm{NaHCO}_{3}$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The residue was purified by column chromatography eluting with $2: 1$ hexane $/ \mathrm{CHCl}_{3}$ to give the yellow oil product ( $50 \mathrm{mg}, 42.0 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 7.22(\mathrm{~s}, 1 \mathrm{H}), 7.20(\mathrm{~s}, 1 \mathrm{H}), 7.18(\mathrm{~s}, 1 \mathrm{H}), 7.17(\mathrm{~s}, 1 \mathrm{H}), 7.13(\mathrm{~d}, \mathrm{~J}=3.9 \mathrm{~Hz}$, $1 \mathrm{H}), 7.09(\mathrm{~d}, \mathrm{~J}=4.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.96(\mathrm{~m}, 4 \mathrm{H}), 6.89(\mathrm{~m}, 4 \mathrm{H}), 6.67(\mathrm{~m}, 8 \mathrm{H}), 3.41(\mathrm{~m}, 16 \mathrm{H}), 2.78(\mathrm{~m}, 16 \mathrm{H})$, $1.87(\mathrm{~m}, 16 \mathrm{H}), 1.68(\mathrm{~m}, 16 \mathrm{H}), 1.44(\mathrm{~m}, 32 \mathrm{H})$.

Synthesis of 3TCbz-2. To a solution of $\mathbf{4 C b z}(1.27 \mathrm{~g}, 1.0 \mathrm{mmol})$ in 1-methylpyrrolidone (NMP, 20 $\mathrm{mL})$, To the suspension of $\mathrm{NaH}(60 \%$ dispersion in mineral oil, $80 \mathrm{mg}, 2.0 \mathrm{mmol})$ in 1methylpyrrolidone (NMP, 40 mL ), 4Cbz ( $1.27 \mathrm{~g}, 1.0 \mathrm{mmol}$ ) was added in portions, and after stirring for 30 min at room temperature, $\mathbf{3 T 6 C B r}(230 \mathrm{mg}, 0.4 \mathrm{mmol})$ in $\mathrm{NMP}(1 \mathrm{~mL})$ was added dropwise to the reaction mixture and was heated up to $160{ }^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$ for 2 days. After cooling to room temperature, NMP was removed under vacuum and water ( $\sim 100 \mathrm{~mL}$ ) was added. The mixture was neutralized with dilute hydrochloride, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried over $\mathrm{MgSO}_{4}$, and the solvent evaporated. The residue was purified by column chromatography eluting with $1: 4$ hexane/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to give the yellow solid product ( $360 \mathrm{mg}, 30.5 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 8.08(\mathrm{~d}, \mathrm{~J}=7.8 \mathrm{~Hz}, 16 \mathrm{H}), 7.46 \sim 7.37(\mathrm{~m}$, $32 \mathrm{H}), 7.23 \sim 7.18(\mathrm{~m}, 16 \mathrm{H}), 7.13(\mathrm{~d}, \mathrm{~J}=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.05(\mathrm{~d}, \mathrm{~J}=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.89(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H})$, $6.82(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.62 \sim 6.60(\mathrm{~m}, 2 \mathrm{H}), 6.57(\mathrm{~d}, \mathrm{~J}=1.2 \mathrm{~Hz}, 4 \mathrm{H}), 6.53 \sim 6.45(\mathrm{~m}, 10 \mathrm{H}), 6.29(\mathrm{t}, \mathrm{J}=2.1$ $\mathrm{Hz}, 4 \mathrm{H}), 4.89(\mathrm{~s}, 8 \mathrm{H}), 4.39(\mathrm{~s}, 4 \mathrm{H}), 4.33(\mathrm{t}, \mathrm{J}=7.2 \mathrm{~Hz}, 16 \mathrm{H}), 3.85(\mathrm{t}, \mathrm{J}=6.3 \mathrm{~Hz}, 16 \mathrm{H}), 3.40(\mathrm{t}, \mathrm{J}=6.3 \mathrm{~Hz}$, $4 \mathrm{H}), 2.71(\mathrm{t}, \mathrm{J}=7.2 \mathrm{~Hz}, 4 \mathrm{H}), 2.02(\mathrm{~m}, 16 \mathrm{H}), 1.80(\mathrm{~m}, 16 \mathrm{H}), 1.56(\mathrm{~m}, 8 \mathrm{H}), 1.33(\mathrm{~m}, 8 \mathrm{H})$. Anal. Calcd $\mathrm{C}_{194} \mathrm{H}_{188} \mathrm{~N}_{8} \mathrm{O}_{14} \mathrm{~S}_{3}: \mathrm{C}, 78.94 ; \mathrm{H}, 6.42$; N, 3.80; O, 7.59; S, 3.26; Found: C, 76.21; H, 6.41; N, 3.45; O, 7.82; S, 3.22.

Synthesis of 5,5'’-Bis-(6-bromohexyl)-[2,2';3',2'’]terthiophene (3T6CBr). An emulsion of 5,5'’-bis-(6-methoxy-hexyl)-[2, $\left.2^{\prime} ; 3^{\prime}, 2^{\prime} ’\right] t e r t h i o p h e n e ~(3 T 6 C O M e, ~ 3.0 ~ g, ~ 6.3 ~ m m o l) ~ i n ~ 40 ~ m L ~ H B r ~ a q u e o u s ~$ solution ( $47.0 \sim 49.0 \mathrm{wt} \%$ ) was heated up to refluxing for 20 hours. After cooling to room temperature, the reaction mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was neutralized with aqueous $\mathrm{NaHCO}_{3}$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The residue was purified by column chromatography eluting with $1: 1$ hexane $/ \mathrm{CHCl}_{3}$ to give the yellow solid product $(2.85 \mathrm{~g}, 78.7 \%) .{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right)$ : $7.23(\mathrm{~d}, \mathrm{~J}=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.17(\mathrm{~d}, \mathrm{~J}=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.01(\mathrm{~d}, \mathrm{~J}=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{~d}, \mathrm{~J}=3.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.74$ $(\mathrm{d}, \mathrm{J}=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.72(\mathrm{~d}, \mathrm{~J}=3.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.43(\mathrm{t}, \mathrm{J}=6.9 \mathrm{~Hz}, 4 \mathrm{H}), 2.83(\mathrm{t}, \mathrm{J}=7.5 \mathrm{~Hz}, 4 \mathrm{H}), 1.90$ (pentet, $\mathrm{J}=6.9 \mathrm{~Hz}, 4 \mathrm{H}), 1.73($ pentet, $\mathrm{J}=7.5 \mathrm{~Hz}, 4 \mathrm{H}), 1.48(\mathrm{~m}, 8 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 146.49$,

Synthesis of 5,5'-Bis-(6-carbazol-9-yl-hexyl)-[2,2';3',2']terthiophene (3TCbz-1). To the suspension of $\mathrm{NaH}(60 \%$ dispersion in mineral oil, $400 \mathrm{mg}, 10.0 \mathrm{mmol}$ ) in dry THF ( 50 mL ), carbazole $(1.5 \mathrm{~g}, 8.97 \mathrm{mmol})$ was added in portions, and after stirring for 30 min at room temperature, $5,5^{\prime}$ '-bis-(6-bromohexyl)-[2, $\left.2^{\prime} ; 3^{\prime}, 2^{\prime ’}\right]$ terthiophene ( $3 \mathbf{T 6 C B r}, 1.5 \mathrm{~g}, 2.61 \mathrm{mmol}$ ) in dry THF ( 10 mL ) was added dropwise to the reaction mixture and refluxed overnight. After cooling to room temperature, THF was removed by rotary evaporation and water ( $\sim 100 \mathrm{~mL}$ ) was added. The mixture was neutralized with dilute hydrochloride, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried over $\mathrm{MgSO}_{4}$, and the solvent evaporated. The residue was purified by column chromatography eluting with $2: 1$ hexane/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to give the yellow solid product ( $1.65 \mathrm{~g}, 84.6 \%) .{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 8.11(\mathrm{~d}, \mathrm{~J}=7.8 \mathrm{~Hz}, 4 \mathrm{H}), 7.43(\mathrm{~m}, 8 \mathrm{H}), 7.22(\mathrm{~m}$, $5 \mathrm{H}), 7.11(\mathrm{~d}, \mathrm{~J}=4.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.85(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.62(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H})$, $6.59(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.28(\mathrm{t}, \mathrm{J}=7.2 \mathrm{~Hz}, 4 \mathrm{H}), 2.70(\mathrm{t}, \mathrm{J}=7.2 \mathrm{~Hz}, 4 \mathrm{H}), 1.85(\mathrm{~m}, 4 \mathrm{H}), 1.59(\mathrm{~m}, 4 \mathrm{H})$, $1.37(\mathrm{~m}, 8 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 146.93,145.43,140.28,134.98,132.29,131.96,131.16$, $129.53,127.59,125.89,125.51,125.54,124.18,124.02,122.68,120.25,118.65,108.55,42.74,31.19$, 31.15, 29.86, 29.82, 28.71, 28.63, 26.82. HRMS m/z Calcd for $\mathrm{C}_{48} \mathrm{H}_{46} \mathrm{~N}_{2} \mathrm{~S}_{3}\left(\mathrm{M}^{+}\right) 746.2823$, Found 746.2784. Anal. Calcd $\mathrm{C}_{48} \mathrm{H}_{46} \mathrm{~N}_{2} \mathrm{~S}_{3}: \mathrm{C}, 77.17 ; \mathrm{H}, 6.21 ; \mathrm{N}, 3.75 ; \mathrm{S}, 12.88$; Found: C, 77.67; H, 6.22; N, 3.60; S, 11.84 .

Synthesis of 6T6COMe. 0.5 mL of $\mathrm{n}-\mathrm{BuLi}(2.5 \mathrm{M}$ in hexane, 1.25 mmol$)$ was added dropwise to a solution of $5,5^{\prime}$ '-bis-(6-methoxy-hexyl)-[2, $\left.2^{\prime} ; 3^{\prime}, 2^{\prime} ’\right]$ terthiophene (3T6COMe, $480 \mathrm{mg}, 1.01 \mathrm{mmol}$ ) in dry THF at $-78^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$. After 30 min stirring upon addition, the mixture was transferred to another flask charged with $\mathrm{CuCl}_{2}(0.41 \mathrm{~g}, 3.0 \mathrm{mmol})$ in 10 mL of dry THF at $-78^{\circ} \mathrm{C}$. The reaction mixture was warmed up to room temperature and reacted overnight. After normal workup, the residue was purified
by column chromatography eluting with $1: 1$ hexane/ ether to give the yellow oil product ( 310 mg , $65.2 \%) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 7.16(\mathrm{~s}, 2 \mathrm{H}), 6.95(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 2 \mathrm{H})$, $6.67(\mathrm{~d}, \mathrm{~J}=3.6 \mathrm{~Hz}, 4 \mathrm{H}), 3.37(\mathrm{t}, \mathrm{J}=6.3 \mathrm{~Hz}, 8 \mathrm{H}), 3.33(\mathrm{~s}, 12 \mathrm{H}), 2.78(\mathrm{~m}, 8 \mathrm{H}), 1.68(\mathrm{~m}, 8 \mathrm{H}), 1.58(\mathrm{~m}, 8 \mathrm{H})$, $1.39(\mathrm{~m}, 16 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 147.24,146.01,134.27,134.24,132.18,131.96,130.64$, $127.31,126.32,126.15,124.16,124.02,72.65,58.39,31.40,31.30,29.93,29.89,29.43,28.77,25.75$. HRMS $m / z$ Calcd for $\mathrm{C}_{52} \mathrm{H}_{70} \mathrm{O}_{4} \mathrm{~S}_{6}\left(\mathrm{M}^{+}\right) 950.3598$, Found 950.3662 .

Synthesis of 6T6CBr. An emulsion of 6T6COMe ( $1.0 \mathrm{~g}, 1.05 \mathrm{mmol}$ ) in 40 mL HBr aqueous solution ( $47.0 \sim 49.0 \mathrm{wt} \%$ ) was heated up to refluxing for 20 hours. After cooling to room temperature, the reaction mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was neutralized with aqueous $\mathrm{NaHCO}_{3}$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The residue was purified by column chromatography eluting with 2:1 hexane/ $\mathrm{CHCl}_{3}$ to give the yellow oil product $(0.5 \mathrm{~g}, 41.5 \%) .{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 7.16(\mathrm{~s}$, $2 \mathrm{H}), 6.95(\mathrm{~d}, \mathrm{~J}=3.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{~d}, \mathrm{~J}=3.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.68(\mathrm{~m}, 4 \mathrm{H}), 3.41(\mathrm{t}, \mathrm{J}=6.3 \mathrm{~Hz}, 8 \mathrm{H}), 2.79(\mathrm{~m}$, $8 \mathrm{H}), 1.87$ (pentet, $\mathrm{J}=6.9 \mathrm{~Hz}, 8 \mathrm{H}), 1.68(\mathrm{~m}, 8 \mathrm{H}), 1.45(\mathrm{~m}, 16 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 147.16$, 145.91, 134.47, 134.38, 132.27, 132.14, 130.77, 127.47, 126.46, 126.31, 124.37, 124.21, 33.86, 32.62, 31.33, 31.24, 29.95, 28.10, 27.83.


3TCbz-1


Synthesis of 6TCbz-1. n-BuLi ( 2.5 M in hexane, $0.8 \mathrm{~mL}, 2.0 \mathrm{mmol}$ ) was added to a $0{ }^{\circ} \mathrm{C}$ THF solution of 3TCbz-1 (1.49 g, 2.0 mmol$)$, and the mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 0.5 h . This reaction mixture was then transferred at $0{ }^{\circ} \mathrm{C}$ into a refluxing THF solution containing $\mathrm{Fe}(\mathrm{acac})_{3}(0.71 \mathrm{~g}, 2.0$ mmol ) and allowed to reflux overnight. The mixture was filtered, and the salts were washed with THF.

The filtrate was concentrated under reduced pressure to yield a red oil, which was purified by silica column chromatography (hexane $/ \mathrm{CH}_{2} \mathrm{Cl}_{2}, 3: 2, \mathrm{v} / \mathrm{v}$ ) to afford the desired product as a yellow solid ( 0.68 , $45.6 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 8.09(\mathrm{~d}, \mathrm{~J}=7.8 \mathrm{~Hz}, 8 \mathrm{H}), 7.41(\mathrm{~m}, 16 \mathrm{H}), 7.22(\mathrm{~m}, 8 \mathrm{H}), 7.13(\mathrm{~s}$, 2H), $6.92(\mathrm{dJ}=3.3 \mathrm{~Hz}, 2 \mathrm{H}),, 6.86(\mathrm{~d}, \mathrm{~J}=3.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.60(\mathrm{~m}, 4 \mathrm{H}), 4.28(\mathrm{t}, \mathrm{J}=7.2 \mathrm{~Hz}, 8 \mathrm{H}), 2.70(\mathrm{~m}$, $8 \mathrm{H}), 1.86(\mathrm{~m}, 8 \mathrm{H}), 1.58(\mathrm{~m}, 8 \mathrm{H}), 1.36(\mathrm{~m}, 16 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 147.08,145.83,140.26$, $134.28,132.20,131.98,130.66,127.39,126.41,126.23,125.49,124.27,124.11,122.67,120.22$, $118.64,108.52,42.73,31.21,31.12,29.86,29.83,28.72,28.63,26.81$. HRMS $\mathrm{m} / \mathrm{z}$ Calcd for $\mathrm{C}_{96} \mathrm{H}_{90} \mathrm{~N}_{4} \mathrm{~S}_{6}\left(\mathrm{M}^{+}\right)$1490.5490, Found 1490.5696. Anal. Calcd $\mathrm{C}_{96} \mathrm{H}_{90} \mathrm{~N}_{4} \mathrm{~S}_{6}$ : C, 77.27; H, 6.08; S, 12.89; Found: C, 77.11; H, 6.25; S, 11.37.


Synthesis of 6TCbz-1. To the suspension of $\mathrm{NaH}(60 \%$ dispersion in mineral oil, $40 \mathrm{mg}, 1.0 \mathrm{mmol})$ in dry THF $(10 \mathrm{~mL})$, carbazole $(0.15 \mathrm{~g}, 0.90 \mathrm{mmol})$ was added in portions, and after stirring for 30 min at room temperature, $\mathbf{6 T 6 C B r}(175 \mathrm{mg}, 0.15 \mathrm{mmol})$ in dry THF $(1 \mathrm{~mL})$ was added dropwise to the reaction mixture and refluxed overnight. After cooling to room temperature, THF was removed by rotary evaporation and water ( $\sim 50 \mathrm{~mL}$ ) was added. The mixture was neutralized with dilute hydrochloride, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried over $\mathrm{MgSO}_{4}$, and the solvent evaporated. The residue was purified by column chromatography eluting with $3: 2$ hexane/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to give the yellow solid product ( $80.5 \mathrm{mg}, 36.0 \%$ ).

## Synthesis of 4,7-Bis-[5,5'-bis-(6-methoxy-hexyl)-[2,2';3',2'']terthiophen-5'-yl]-1,10-phenanthro-

line (Phen3T). To a round-bottomed flask was charged 0.23 g of tributyl [5,5''-bis-(6-methoxy-hexyl)-[2,2';3',2'']terthiophene-5'-yl]stannane (Sn3T6COMe, 0.3 mmol ), 33.8 mg of 4,7 -dibromo-1,10-
phenanthroline $(0.1 \mathrm{mmol}), 75 \mathrm{mg}$ of $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$, and 10 mL of DMF. After three freeze-thaw cycles, the mixture was heated to $110{ }^{\circ} \mathrm{C}$ for 20 hours under nitrogen. After cooling to room temperature, the reaction mixture was poured into water, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and washed thoroughly with NaF solution to remove tributyltin chloride. The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was purified by flash column using $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{CH}_{3} \mathrm{OH}(20: 1)$ as eluent. 96 mg of the pure product was obtained (yield: $85 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 9.22(\mathrm{~d}, \mathrm{~J}=4.8 \mathrm{~Hz}, 2 \mathrm{H}), 8.46(\mathrm{~s}, 2 \mathrm{H}), 7.73$ (d, $J=4.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.42(\mathrm{~s}, 2 \mathrm{H}), 7.05(\mathrm{~d}, J=3.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.98(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.71(\mathrm{~m}, 4 \mathrm{H}), 3.36(\mathrm{~m}$, $20 \mathrm{H}), 2.81(\mathrm{t}, J=7.2 \mathrm{~Hz}, 8 \mathrm{H}), 1.69(\mathrm{~m}, 8 \mathrm{H}), 1.58(\mathrm{~m}, 8 \mathrm{H}), 1.40(\mathrm{~m}, 16 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta$ $149.4,147.5,146.6,146.0,139.7,135.9,133.8,133.7,132.2,131.2,131.1,127.6,126.3,125.5,124.1$, $123.9,123.7,123.1,72.4,58.1,31.1,31.0,29.7,29.6,29.2,28.5,25.5$.

Synthesis of 4,7-Bis-[5,5"'-bis-(6-bromo-hexyl)-[2,2';3',2']terthiophen-5'-yl]-[1,10]phenanthroline (Phen3TBr). An emulsion of Phen3T ( $226 \mathrm{mg}, 0.2 \mathrm{mmol}$ ) in 20 mL HBr aqueous solution ( $47.0 \sim 49.0 \mathrm{wt} \%$ ) was heated to reflux for 20 hours. After cooling to room temperature, the reaction mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The extract was neutralized with aqueous $\mathrm{NaHCO}_{3}$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The residue was purified by column chromatography eluting with $15: 1$ $\mathrm{CHCl}_{3} / \mathrm{EtOH}$ to give the yellow oil product $(108 \mathrm{mg}, 41 \%) .{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 9.20(\mathrm{~d}, \mathrm{~J}$ $=5.1 \mathrm{~Hz}, 2 \mathrm{H}), 8.44(\mathrm{~s}, 2 \mathrm{H}), 7.71(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.42(\mathrm{~s}, 2 \mathrm{H}), 7.04(\mathrm{~d}, J=3.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.97(\mathrm{~d}, J=3.6$ $\mathrm{Hz}, 2 \mathrm{H}), 6.71(\mathrm{~m}, 4 \mathrm{H}), 3.40(\mathrm{t}, J=6.6 \mathrm{~Hz}, 4 \mathrm{H}), 3.39(\mathrm{t}, J=6.6 \mathrm{~Hz}, 4 \mathrm{H}), 2.80(\mathrm{t}, J=7.5 \mathrm{~Hz}, 8 \mathrm{H}), 1.84(\mathrm{~m}$, $8 \mathrm{H}), 1.69(\mathrm{~m}, 8 \mathrm{H}), 1.43(\mathrm{~m}, 16 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 149.8,147.7,147.0,146.1,140.1$, $136.4,134.2,133.9,132.5,131.5,131.4,127.9,126.6,125.9,124.5,124.3,124.1,123.5,33.8,32.6$, 31.3, 31.2, 29.9 (2), 29.6, 28.0, 27.8.

Synthesis of Sn3TCbz-1. About 0.5 mL of n-butyllithium ( 2.5 M in hexane, 1.3 mmol ) was added dropwise to a solution of $\mathbf{3 T C b z - 1}(0.86 \mathrm{~g}, 1.2 \mathrm{mmol})$ in 20 mL of THF at $-78{ }^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$. After 30
minutes, 0.82 g of tributyltin chloride $(0.42 \mathrm{~g}, 1.3 \mathrm{mmol})$ in 5.0 mL of anhydrous THF was added to the solution. The reaction mixture was then warmed to room temperature, stirred for another 3 hours, and poured into saturated $\mathrm{NaHCO}_{3}$ solution. The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried over $\mathrm{MgSO}_{4}$, and the solvent evaporated. The residue was used for the next reaction without further purification. ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 8.09(\mathrm{~d}, J=7.8 \mathrm{~Hz} 4 \mathrm{H}$, ), $7.43(\mathrm{~m}, 8 \mathrm{H}), 7.22(\mathrm{~m}, 4 \mathrm{H}), 7.09(\mathrm{~s}, 1 \mathrm{H}), 6.88(\mathrm{~d}, J$ $=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.84(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.59(\mathrm{~m}, 2 \mathrm{H}), 4.27(\mathrm{t}, J=7.2 \mathrm{~Hz}, 4 \mathrm{H}), 2.70(\mathrm{~m}, 4 \mathrm{H}), 1.85(\mathrm{~m}$, $4 \mathrm{H}), 1.57(\mathrm{~m}, 10 \mathrm{H}), 1.37(\mathrm{~m}, 14 \mathrm{H}), 1.11(\mathrm{~m}, 6 \mathrm{H}), 0.90(\mathrm{t}, J=7.2 \mathrm{~Hz}, 9 \mathrm{H})$.


Synthesis of Phen3TCbz-1. To the suspension of NaH ( $60 \%$ dispersion in mineral oil, $40 \mathrm{mg}, 1.0$ mmol ) in dry THF ( 10 mL ), carbazole ( $0.15 \mathrm{~g}, 0.90 \mathrm{mmol}$ ) was added in portions, and after stirring for 30 min at room temperature, $\mathbf{P h e n} 3 T B r(133 \mathrm{mg}, 0.10 \mathrm{mmol})$ in dry THF $(1 \mathrm{~mL})$ was added dropwise to the reaction mixture and refluxed overnight. After cooling to room temperature, THF was removed by rotary evaporation and water ( $\sim 50 \mathrm{~mL}$ ) was added. The mixture was neutralized with dilute hydrochloride, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, dried over $\mathrm{MgSO}_{4}$, and the solvent evaporated. The residue was purified by column chromatography eluting with $20: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{CH}_{3} \mathrm{OH}$ to give the product as yellow oil (128 mg, 77\%). ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \delta\right): 9.22(\mathrm{~d}, J=4.8 \mathrm{~Hz}, 2 \mathrm{H}), 8.44(\mathrm{~s}, 2 \mathrm{H}), 8.09(\mathrm{~m}, 8 \mathrm{H})$, $7.72(\mathrm{~d}, J=4.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.49 \sim 7.35(\mathrm{~m}, 18 \mathrm{H}), 7.25 \sim 7.18(\mathrm{~m}, 8 \mathrm{H}), 7.05(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.97(\mathrm{~d}, J=3.9$
$\mathrm{Hz}, 2 \mathrm{H}), 6.65(\mathrm{~m}, 4 \mathrm{H}), 4.27(\mathrm{t}, J=7.2 \mathrm{~Hz}, 4 \mathrm{H}), 4.25(\mathrm{t}, J=7.2 \mathrm{~Hz}, 4 \mathrm{H}), 2.72(\mathrm{t}, J=7.5 \mathrm{~Hz}, 8 \mathrm{H}), 1.84(\mathrm{~m}$, $8 \mathrm{H}), 1.61(\mathrm{~m}, 8 \mathrm{H}), 1.36(\mathrm{~m}, 16 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right): \delta 149.76,147.65,146.90,146.10$, $140.27,140.09,136.42,134.20,133.94,132.50,131.52,131.44,127.89,126.57,125.92,125.49$, $124.45,124.24,124.09,123.50,122.69,120.26,118.65,108.52,42.81,31.25,31.18,29.93,29.87$, 28.76, 28.66, 26.87.


Synthesis of Phen3TCbz-1 (Route a). To a round-bottomed flask was charged 0.31 g of Sn3TCbz-1 ( 0.3 mmol ), 33.8 mg of 4,7-dibromo-1,10-phenanthroline ( 0.1 mmol ), 75 mg of $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$, and 10 mL of DMF. After three freeze-thaw cycles, the mixture was heated to $110^{\circ} \mathrm{C}$ for 20 hours under nitrogen. After cooling to room temperature, the reaction mixture was poured into water, extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and washed thoroughly with NaF solution to remove tributyltin chloride. The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was purified by flash column using $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{CH}_{3} \mathrm{OH}(20: 1)$ as eluent and 109 mg of pure product was obtained (yield: $65 \%$ ).

## Reference:

(1) Michalitsch, R.; ElKassmi, A.; Yassar, A.; Garnier, F. J. Heterocycl. Chem. 2001, 38, 649-653.
(2) Graf, G. I.; Hastreiter, D.; da Silva, L. E.; Rebelo, R. A.; Montalban, A. G.; McKillop, A. Tetrahedron, 2002, 58, 9095-9100.
(3) Xia, C.; Fan, X.; Locklin, J.; Advincula, R. C. Org. Lett. 2002, 4, 2067-2070.
(4) (a) Taranekar, P.; Park, J-Y.; Patton, D.; Fulghum, T.; Ramon, G. J.; Advincula, R. C. Adv. Mater. 2006, 18, 2461-2465. (b) Fréchet, J. M. J. Science 1994, 263, 1710-1715. (c) Taranekar, P.; Fulghum, T.; Patton, D.; Ponnapati, R.; Clyde, G.; Advincula, R. C. J. Am. Chem. Soc. 2007, 129, 12537-12548.

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Figure S1. UV-vis spectra of dendritic molecules.


Figure S2. Emission spectra of dendritic molecules 3T6COMe, 6T6COMe, and 4Cbz.

Table S1. Extinction coefficients, absorption, and fluorescence maxima of thiophene dendrons and dendrimers ${ }^{a}$

| compd | $\lambda^{\text {abs }}{ }_{\max }(\mathrm{nm})(\varepsilon)$ | $\lambda^{\mathrm{fl}}{ }_{\text {max }}(\mathrm{nm})$ |
| :---: | :---: | :---: |
| 3T6COMe | $245(10418), 270 \operatorname{sh}(9131), 315(8213)$ | $452^{b}$ |
| 6T6COMe | $250(15204), 299(17207), 393(13181)$ | $483,505 \mathrm{sh}^{c}$ |
| 3TCbz-1 | $237(56453), 264(34209), 295(23162), 320(9573), 331(9751), 346(8167)$ | $353,370 \mathrm{sh}, 449^{d}$ |
| 6TCbz-1 | $238(131902), 264(78078), 295(57571), 332(22529), 346(22421), 393(15335)$ | $353,483,505 \mathrm{sh}^{d}$ |
| 4Cbz | 237(109958), 264(58601), 295(39474), 332(9218), 346(10490) | $354,370 \mathrm{sh}^{d}$ |
| 3TCbz-2 | 237(179065), 264(103393), 295(71400), 331(24037), 346(23935) | $354,370 \mathrm{sh}, 450^{d}$ |

${ }^{a}$ For the absorption and fluorescence, the maxima (or shoulders) of the relevant bands are given. ${ }^{b}$ $\lambda_{\mathrm{ex}}=315 \mathrm{~nm} .{ }^{c} \lambda_{\mathrm{ex}}=395 \mathrm{~nm} .{ }^{d} \lambda_{\mathrm{ex}}=295 \mathrm{~nm}$.




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+TOF
$\mathrm{a}=3.56059492627340870 \mathrm{e}-004, \mathrm{t} 0=7.04662692797355700 \mathrm{e}+001$



