

## Iron (III) Chloride as an Efficient Catalyst for Stereoselective Synthesis of Glycosyl Azides and a Cocatalyst with Cu(0) for the Subsequent Click Chemistry

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### SUPPORTING INFORMATION

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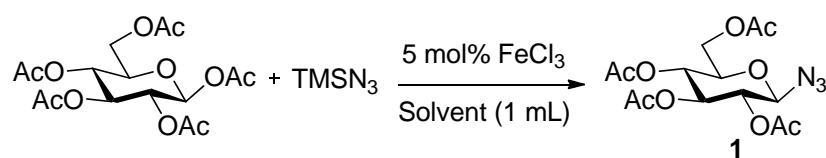
**General.** <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded on 400 MHz <sup>1</sup>H (100 MHz <sup>13</sup>C) spectrometers in deuteriochloroform with chloroform as an internal reference unless otherwise stated. Chemical shifts are reported in ppm ( $\delta$ ). Coupling constants, *J*, are reported in Hz. Electrospray ionization (ESI) mass spectra were recorded with data reported in the form m/e (intensity relative to base peak). Analytical TLC was performed on silica gel glass plates. Visualization was accomplished with UV light or with phosphomolybdic acid (PMA) and KMnO<sub>4</sub> staining agents. Column (flash) chromatography was performed using 32-63  $\mu$ m silica gels. THF and toluene were dried over Na with benzophenone-ketyl intermediate under nitrogen atmosphere and distilled before use. DMF, CH<sub>2</sub>Cl<sub>2</sub> and CHCl<sub>3</sub> were dried over CaH and distilled before use. MeOH was dried over magnesium turnings under nitrogen atmosphere and distilled before use. Oxygen was purged in CHCl<sub>3</sub> before use. All reaction products were isolated as chromatographically pure materials. All reagents and catalysts used for azido glycosylation reaction were purchased from Across Organics. The acetate

derivatives of glucose<sup>1</sup>, galactose<sup>2</sup>, mannose<sup>3</sup>, ribose, xylose<sup>4</sup>, maltose, lactose<sup>5</sup>, maltotriose<sup>6</sup> were synthesized by using 2 equiv of acetic anhydride (for each hydroxyl group) in the presence of 10 mol% of Sodium acetate under reflux condition. *N*-protected glucosamine derived substrates and propargyl-*O*-linked glucose peracetate<sup>7</sup> were synthesized by using literature methods.

**Caution:** Organic azides (e.g. TMSN<sub>3</sub>) are potentially explosive and corrosive compounds and should be handled with great care,<sup>8</sup> despite we did not sense any adverse effects during our studies at small scale reactions.

**Attention:** It has been documented in the literatures that, ionic azides react with CH<sub>2</sub>Cl<sub>2</sub> to form explosive azidochloromethane and/or diazidomethane.<sup>9</sup> It has been noted that, if these azides were formed, they would be converted to stable triazole derivatives in the subsequent azide-alkyne cycloaddition reactions. However, we did not observe these side products under our catalytic conditions.

**Table S1. Effect of solvents on the azido glycosylation of glucose  $\beta$ -peracetate.<sup>a</sup>**

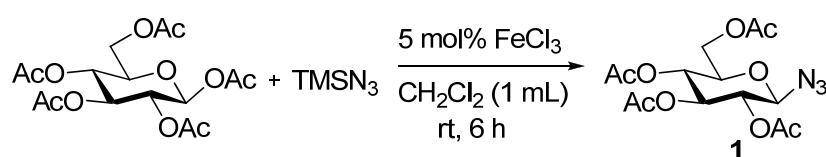


Entry	Solvent	Time (h)	Conversion (%) <sup>b</sup>	Yield (%) <sup>c</sup>
1	CH <sub>3</sub> CN	36	5	n.d.
2	CH <sub>3</sub> COCH <sub>3</sub>	36	0	--
3	CH <sub>3</sub> NO <sub>2</sub>	36	decomposed	--
4	THF	36	0	--
5	TBME	40	55	47
6	toluene	40	95	86
7 <sup>d</sup>	toluene	7	quantitative	91
8	CH <sub>2</sub> Cl <sub>2</sub>	6	quantitative	96

<sup>a</sup>Carried out by using TMSN<sub>3</sub> (0.38 mmol, 1.5 equiv), 5 mol% FeCl<sub>3</sub> catalyst in 1 mL of solvent. <sup>b</sup>Conversion of  $\beta$ -D-glucose peracetate was determined by <sup>1</sup>H NMR analysis of the crude reaction mixture. <sup>c</sup>Isolated, purified yield. <sup>d</sup>Reaction was conducted at 70 °C .

**(A) Detailed procedures for FeCl<sub>3</sub> catalyzed azido glycosylation and analytical data of azido glycosides 1-10.**

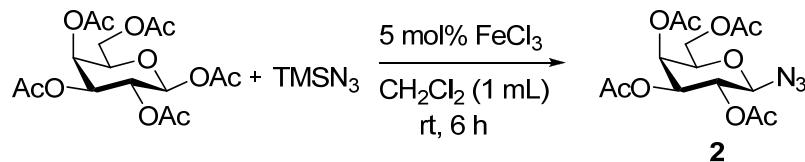
**2,3,4,6-tetra-O-acetyl-6-deoxy- $\beta$ -D-glucopyranosyl azide (1).<sup>10</sup>**



To a 10 mL, round bottomed flask was placed 5 mol% of FeCl<sub>3</sub> (2 mg, 0.0125 mmol, 0.05 equiv) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) under nitrogen atmosphere. A solution of (2S,3R,4S,5R,6R)-6-(acetoxymethyl)tetrahydro-2H-pyran-2,3,4,5-tetrayl tetraacetate (97.5 mg, 0.25 mmol, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added. After having been

stirred for 5 min at ambient temperature, a solution of trimethyl silyl azide i.e. TMSN<sub>3</sub> (50 µL, 44 mg, 0.38 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 6 h, the reaction mixture was quenched with saturated aqueous NaHCO<sub>3</sub> solution (5 mL). Two layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 × 3 mL). The combined organic layers were washed with brine (5 mL), dried (MgSO<sub>4</sub>), filtered and evaporated under reduced pressure. The crude residue was purified by column chromatography (EtOAc/hexanes, 1/2) on silica gel to afford 89 mg (96%) of 2,3,4,6-tetra-*O*-acetyl-6-deoxy-β-D-glucopyranosyl azide **1** as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 5.19 (t, *J* = 9.2, 1H), 5.07 (t, *J* = 10, 1H), 4.92 (t, *J* = 9.2, 1H), 4.63 (d, *J* = 8.8, 1H), 4.25 (dd, *J* = 12.4, 4.8, 1H), 4.14 (dd, *J* = 12.4, 2, 1H), 3.80-3.76 (m, 1H), 2.07 (s, 3H, COCH<sub>3</sub>), 2.05 (s, 3H, COCH<sub>3</sub>), 2.00 (s, 3H, COCH<sub>3</sub>), 1.98 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 170.5, 170.0, 169.2, 169.1, 87.8, 74.0, 72.5, 70.6, 67.9, 61.6, 20.6, 20.4; MS ESI (C<sub>14</sub>H<sub>19</sub>O<sub>9</sub>N<sub>3</sub>, 373): 396 (M+Na<sup>+</sup>, 100); TLC R<sub>f</sub> 0.34 (EtOAc/hexanes, 1/2); m.p. 124-125 °C (lit.<sup>11</sup> m.p. 127 °C); [α]<sub>D</sub><sup>25</sup> -30.2 (*c* 1.0, CHCl<sub>3</sub>) (lit.<sup>10</sup> [α]<sub>D</sub><sup>23</sup> -31 (*c* 1.0, CHCl<sub>3</sub>)).

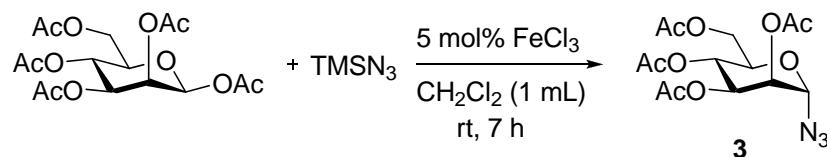
**2,3,4,6-tetra-*O*-acetyl-6-deoxy- $\beta$ -D-galactopyranosyl azide (**2**).<sup>10</sup>**



To a 10 mL, round bottomed flask was placed 5 mol% of FeCl<sub>3</sub> (2 mg, 0.0125 mmol, 0.05 equiv) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) under nitrogen atmosphere. A solution of (2*S*,3*S*,4*S*,5*R*,6*R*)-6-(acetoxymethyl)tetrahydro-2*H*-pyran-2,3,4,5-tetrayl tetraacetate (97.5 mg, 0.25 mmol, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added. After having been stirred for 5 min at ambient temperature, a solution of TMSN<sub>3</sub> (50  $\mu$ L, 44 mg, 0.38 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 6 h, the reaction mixture was quenched with saturated aqueous NaHCO<sub>3</sub> solution (5 mL). Two layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2  $\times$  3 mL). The combined organic layers were washed with brine (5 mL), dried (MgSO<sub>4</sub>), filtered and evaporated under reduced pressure. The crude residue was purified by column chromatography (EtOAc/hexanes, 1/2) on silica gel to afford 89 mg (96%) of 2,3,4,6-tetra-*O*-acetyl-6-deoxy- $\beta$ -D-galactopyranosyl azide **2** as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  5.41 (d, *J* = 2.4, 1H), 5.15 (dd, *J* = 10, 8.8, 1H), 5.02 (dd, *J* = 10.4, 3.6, 1H), 4.59 (d, *J* = 8.8, 1H), 4.18-4.14 (m, 2H), 4.00 (td, *J* = 6.8, 0.6, 1H),

2.16 (s, 3H, COCH<sub>3</sub>), 2.08 (s, 3H, COCH<sub>3</sub>), 2.05 (s, 3H, COCH<sub>3</sub>), 1.98 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 170.3, 170.0, 169.9, 169.2, 88.1, 72.7, 70.6, 67.95, 66.8, 61.2, 20.5, 20.4, 20.35; MS ESI (C<sub>14</sub>H<sub>19</sub>O<sub>9</sub>N<sub>3</sub>, 373): 396 (M+Na<sup>+</sup>, 100); TLC R<sub>f</sub> 0.35 (EtOAc/hexanes, 1/2); m.p. 95–97 °C (lit.<sup>11</sup> m.p. 94–95 °C); [α]<sub>D</sub><sup>25</sup> -15.4 (c 1.0, CHCl<sub>3</sub>) (lit.<sup>10</sup> [α]<sub>D</sub><sup>23</sup> -14 (c 1.0, CHCl<sub>3</sub>)).

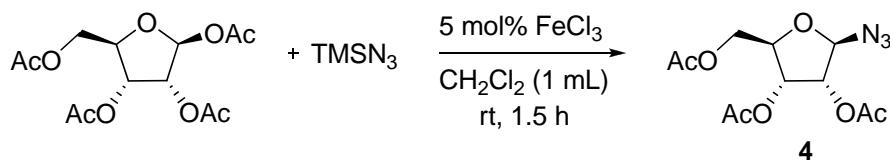
**2,3,4,6-tetra-O-acetyl-6-deoxy- $\alpha$ -D-mannopyranosyl azide (3).<sup>12</sup>**



To a 10 mL, round bottomed flask was placed 5 mol% of FeCl<sub>3</sub> (2 mg, 0.0125 mmol, 0.05 equiv) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) under nitrogen atmosphere. A solution of (3*S*,4*S*,5*S*,6*R*)-6-(acetoxymethyl)tetrahydro-2*H*-pyran-2,3,4,5-tetrayl tetraacetate (97.5 mg, 0.25 mmol, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added. After having been stirred for 5 min at ambient temperature, a solution of TMSN<sub>3</sub> (50 μL, 44 mg, 0.38 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 7 h, the reaction mixture was quenched with saturated aqueous NaHCO<sub>3</sub> solution (5 mL). Two layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 × 3 mL).

The combined organic layers were washed with brine (5 mL), dried ( $\text{MgSO}_4$ ), filtered and evaporated under reduced pressure. The crude residue was purified by column chromatography (EtOAc/hexanes, 1/2) on silica gel to afford 83 mg (89%) of 2,3,4,6-tetra-*O*-acetyl-6-deoxy- $\alpha$ -D-mannopyranosyl azide **3** as a white solid.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.37 (d,  $J = 1.5$ , 1H), 5.28-5.20 (m, 2H), 5.13 (t,  $J = 2.3$ , 1H), 4.28 (dd,  $J = 12.4$ , 5.5, 1H), 4.16-4.11 (m, 2H), 2.18 (s, 3H,  $\text{COCH}_3$ ), 2.14 (s, 3H,  $\text{COCH}_3$ ), 2.1 (s, 3H,  $\text{COCH}_3$ ), 2.05 (s, 3H,  $\text{COCH}_3$ );  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  170.5, 169.8, 169.7, 169.6, 87.4, 70.6, 69.1, 68.2, 65.6, 62.1, 20.7, 20.61, 20.6, 20.5; MS ESI ( $\text{C}_{14}\text{H}_{19}\text{O}_9\text{N}_3$ , 373): 396 ( $\text{M}+\text{Na}^+$ , 100); TLC  $R_f$  0.32 (EtOAc/hexanes, 1/2). m.p. 103-105 °C (lit.<sup>13</sup> m.p. 104.1-104.9 °C);  $[\alpha]_D^{25} +181.1$  ( $c$  1.0,  $\text{CHCl}_3$ ) (lit.<sup>13</sup>  $[\alpha]_D^{23} +184$  ( $c$  1.0,  $\text{CHCl}_3$ )).

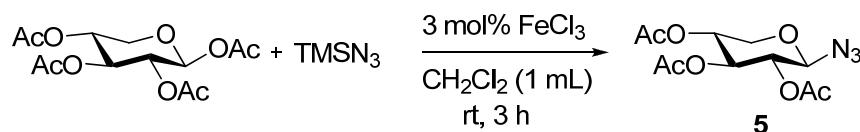
**2,3,5-tri-*O*-acetyl- $\beta$ -D-ribofuranosyl azide (4).<sup>14</sup>**



To a 10 mL, round bottomed flask was placed 5 mol% of  $\text{FeCl}_3$  (1.2 mg, 0.0075 mmol, 0.03 equiv) in anhydrous  $\text{CH}_2\text{Cl}_2$  (0.5 mL) under nitrogen atmosphere. A solution of (2*S*,3*R*,4*R*,5*R*)-5-(acetoxymethyl)tetrahydrofuran-2,3,4-triyl triacetate (79.5 mg, 0.25 mmol, 1 equiv) in  $\text{CH}_2\text{Cl}_2$  (0.25 mL) was added. After having been stirred for 5 min at

ambient temperature, a solution of TMSN<sub>3</sub> (50 µL, 44 mg, 0.38 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 1.5 h, the reaction mixture was quenched with saturated aqueous NaHCO<sub>3</sub> solution (5 mL). Two layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 × 3 mL). The combined organic layers were washed with brine (5 mL), dried (MgSO<sub>4</sub>), filtered and evaporated under reduced pressure. The crude residue was purified by column chromatography (EtOAc/hexanes, 1/2) on silica gel to afford 72 mg (96%) of 2,3,5-tri-*O*-acetyl-β-D-ribofuranosyl azide **4** as a colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 5.33 (d, *J* = 1.4, 1H), 5.30 (dd, *J* = 6.6, 5.0, 1H), 5.10 (dd, *J* = 4.8, 1.6, 1H), 4.38 (dd, *J* = 12, 3.2, 1H), 4.34-4.30 (m, 1H), 4.12 (dd, *J* = 12.2, 4.2, 1H), 2.092 (s, 3H, COCH<sub>3</sub>), 2.09 (s, 3H, COCH<sub>3</sub>), 2.04 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 170.5, 169.5, 169.3, 92.6, 79.3, 74.4, 70.4, 62.9, 20.6, 20.4, 20.3; MS ESI (C<sub>11</sub>H<sub>15</sub>O<sub>7</sub>N<sub>3</sub>, 301): 324 (M+Na<sup>+</sup>, 100); TLC R<sub>f</sub> 0.34 (EtOAc/hexanes, 1/2); [α]<sub>D</sub><sup>25</sup> +111.3 (*c* 1.0, CHCl<sub>3</sub>) (lit.<sup>14</sup> [α]<sub>D</sub><sup>23</sup> +116 (*c* 1.0, CHCl<sub>3</sub>)).

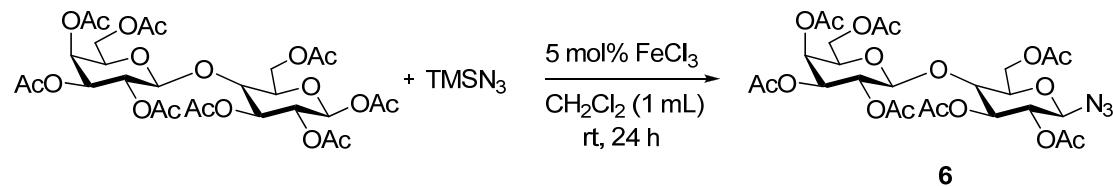
**2,3,5-tri-*O*-acetyl- $\beta$ -D-xylopyranosyl azide (**5**).<sup>10</sup>**



To a 10 mL, round bottomed flask was placed 5 mol% of FeCl<sub>3</sub> (1.2 mg, 0.0075 mmol, 0.03 equiv) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) under nitrogen atmosphere. A solution of (2*S*,3*S*,4*S*)-tetrahydro-2*H*-pyran-2,3,4,5-tetrayl tetraacetate (79.5 mg, 0.25 mmol, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added. After having been stirred for 5 min at ambient temperature, a solution of TMSN<sub>3</sub> (50  $\mu$ L, 44 mg, 0.38 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 3 h, the reaction mixture was quenched with saturated aqueous NaHCO<sub>3</sub> solution (5 mL). Two layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2  $\times$  3 mL). The combined organic layers were washed with brine (5 mL), dried (MgSO<sub>4</sub>), filtered and evaporated under reduced pressure. The crude residue was purified by column chromatography (EtOAc/hexanes, 1/2) on silica gel to afford 69 mg (92%) of 2,3,5-tri-*O*-acetyl- $\beta$ -D-xylopyranosyl azide **5** as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  5.18 (t, *J* = 8.8, 1H), 5.00-4.95 (m, 1H), 4.87 (t, *J* = 8.4, 1H), 4.63 (d, *J* = 8.0, 1H), 4.21 (dd, *J* = 12, 5.6, 1H), 3.44 (dd, *J* = 11.6, 9.2, 1H), 2.07 (s, 3H, COCH<sub>3</sub>), 2.042 (s, 3H, COCH<sub>3</sub>), 2.04 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  169.9, 169.7, 169.3, 88.3, 71.5,

70.4, 68.4, 64.3, 20.6, 20.6; MS ESI ( $C_{11}H_{15}O_7N_3$ , 301): 324 ( $M+Na^+$ , 100); TLC  $R_f$  0.34 (EtOAc/hexanes, 1/2); m.p. 84–85 °C (lit.<sup>15</sup> m.p. 87 °C);  $[\alpha]_D^{25} +88.4$  ( $c$  1.0, CHCl<sub>3</sub>) (lit.<sup>15</sup>  $[\alpha]_D^{25} +86$  ( $c$  1.0, CHCl<sub>3</sub>)).

**2,3,6-tri-O-acetyl-4-O-(2',3',4',6'-tetra-O-acetyl- $\alpha$ -D-galactopyranosyl)- $\beta$ -D-glucopyranosylazide (6).<sup>10</sup>**



To a 10 mL, round bottomed flask was placed 5 mol% of FeCl<sub>3</sub> (2 mg, 0.0125 mmol, 0.05 equiv) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) under nitrogen atmosphere. A solution of (2*S*,3*S*,4*S*,6*R*)-6-(acetoxymethyl)-5-((2*S*,3*S*,4*S*,6*R*)-3,4,5-triacetoxy-6-(acetoxymethyl)tetrahydro-2*H*-pyran-2-yloxy)tetrahydro-2*H*-pyran-2,3,4-triyl triacetate (169.5 mg, 0.25 mmol, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added. After having been stirred for 5 min at ambient temperature, a solution of TMSN<sub>3</sub> (50  $\mu$ L, 44 mg, 0.38 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 24 h, the reaction mixture was quenched with saturated aqueous NaHCO<sub>3</sub> solution (5 mL). Two layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2  $\times$  3 mL). The combined organic layers were washed with brine (5 mL), dried (MgSO<sub>4</sub>), filtered and

evaporated under reduced pressure. The crude residue was purified by column chromatography (EtOAc/hexanes, 3/2) on silica gel to afford 149 mg (90%) of 2,3,6-tri-*O*-acetyl-4-*O*-(2',3',4',6'-tetra-*O*-acetyl- $\alpha$ -D-glucopyranosyl)- $\beta$ -D-glucopyranosyl azide **6** as a amorphous solid.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.34 (d,  $J$  = 3.6, 1H), 5.20 (t,  $J$  = 9.4, 1H), 5.09 (dd,  $J$  = 10.4, 8, 1H), 4.94 (dd,  $J$  = 10.4, 3.6, 1H), 4.85 (t,  $J$  = 9.6, 1H), 4.62 (d,  $J$  = 8.8, 1H), 4.52-4.46 (m, 2H), 4.14-4.05 (m, 3H), 3.87 (t,  $J$  = 6.8, 1H), 3.81 (t,  $J$  = 9.6, 1H), 3.70-3.71 (m, 1H), 2.14 (s, 3H,  $\text{COCH}_3$ ), 2.13 (s, 3H,  $\text{COCH}_3$ ), 2.08 (s, 3H,  $\text{COCH}_3$ ), 2.06 (s, 3H,  $\text{COCH}_3$ ), 2.05 (s, 3H,  $\text{COCH}_3$ ), 2.04 (s, 3H,  $\text{COCH}_3$ ), 1.95 (s, 3H,  $\text{COCH}_3$ );  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  170.3, 170.1, 170.0, 169.6, 169.5, 169.1, 101.1, 87.7, 75.7, 74.8, 72.5, 71.0, 70.9, 70.7, 69.0, 66.6, 61.7, 60.8, 20.8, 20.7, 20.6, 20.5; MS ESI ( $\text{C}_{26}\text{H}_{35}\text{O}_{17}\text{N}_3$ , 661): 684 ( $\text{M}+\text{Na}^+$ , 100); TLC  $R_f$  0.20 (EtOAc/hexanes, 1/1);  $[\alpha]_D^{25}$  -21.2 ( $c$  1.0,  $\text{CHCl}_3$ ) (lit.<sup>10</sup>  $[\alpha]_D^{23}$  -20.4 ( $c$  1.0,  $\text{CHCl}_3$ )).

**2,3,6-tri-*O*-acetyl-4-*O*-(2',3',4',6'-tetra-*O*-acetyl- $\beta$ -D-glucopyranosyl)- $\beta$ -D-glucopyranosylazide (7).<sup>10</sup>**

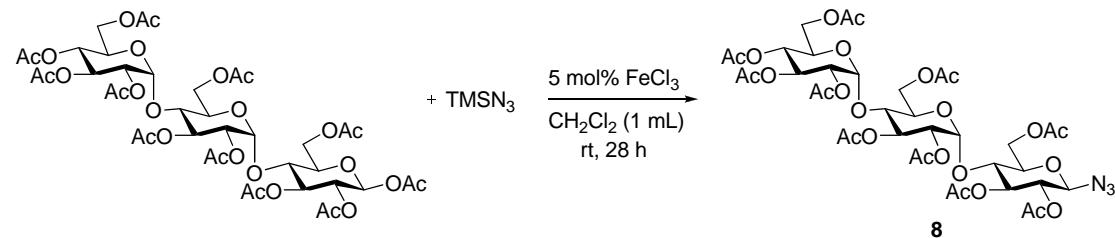


To a 10 mL, round bottomed flask was placed 5 mol% of FeCl<sub>3</sub> (2 mg, 0.0125 mmol, 0.05 equiv) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) under nitrogen atmosphere. A solution of

(*2S,3S,4S,5R,6R*)-6-(acetoxymethyl)-5-((*3S,4S,5S,6R*)-3,4,5-triacetoxy-6-(acetoxymethyl)tetrahydro-2*H*-pyran-2-yloxy)tetrahydro-2*H*-pyran-2,3,4-triyl triacetate (169.5 mg, 0.25 mmol, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added. After having been stirred for 5 min at ambient temperature, a solution of TMSN<sub>3</sub> (50 µL, 44 mg, 0.38 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 22 h, the reaction mixture was quenched with saturated aqueous NaHCO<sub>3</sub> solution (5 mL). Two layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 × 3 mL). The combined organic layers were washed with brine (5 mL), dried (MgSO<sub>4</sub>), filtered and evaporated under reduced pressure. The crude residue was purified by column chromatography (EtOAc/hexanes, 3/2) on silica gel to afford 152 mg (92%) of 2,3,6-tri-*O*-acetyl-4-*O*-(2',3',4',6'-tetra-*O*-acetyl-β-D-glucopyranosyl)-β-D-glucopyranosyl azide **7** as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 5.40 (d, *J* = 4, 1H), 5.35 (t, *J* = 10, 1H), 5.26 (t, *J* = 9.0, 1H), 5.05 (t, *J* = 9.8, 1H), 4.85 (dd, *J* = 10.8, 4.0, 1H), 4.78 (t, *J* = 8.8, 1H), 4.70 (d, *J* = 8.8, 1H), 4.51 (dd, *J* = 12, 2.4, 1H), 4.26-4.24 (m, 12.4, 2H), 4.07-4.03 (m, 2H), 4.01-3.99 (m, 1H), 3.79-3.70 (m, 1H), 2.15 (s, 3H, COCH<sub>3</sub>), 2.10 (s, 3H, COCH<sub>3</sub>), 2.05 (s, 3H, COCH<sub>3</sub>), 2.04 (s, 3H, COCH<sub>3</sub>), 2.02 (s, 3H, COCH<sub>3</sub>), 2.01 (s, 3H, COCH<sub>3</sub>), 2.00 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz,

$\text{CDCl}_3$ )  $\delta$  170.4, 170.3, 169.9, 169.8, 169.3, 169.3, 95.6, 77.3, 77.0, 76.7, 74.9, 72.3, 71.4, 69.9, 69.1, 68.5, 67.9, 62.4, 61.4, 20.7, 20.6, 20.5, 20.4; MS ESI ( $\text{C}_{26}\text{H}_{35}\text{O}_{17}\text{N}_3$ , 661): 684 ( $\text{M}+\text{Na}^+$ , 100); TLC  $R_f$  0.15 (EtOAc/hexanes, 1/1); m.p. 181–183 °C (lit.<sup>16a</sup> m.p. 180–182 °C);  $[\alpha]_D^{25}$  -32.1 (*c* 1.0,  $\text{CHCl}_3$ ) (lit.<sup>16b</sup>  $[\alpha]_D^{24}$  -31 (*c* 1.0,  $\text{CHCl}_3$ )).

**2,3,4,6-tetra-*O*-acetyl-6-deoxy- $\alpha$ -D-glucopyranosyl-(1-4)-2,3,6-tri-*O*-acetyl-6-deoxy- $\alpha$ -D-glucopyranosyl-(1-4)-2,3,6-tri-*O*-acetyl-6-deoxy- $\beta$ -D-glucopyranosyl azide (8).<sup>17</sup>**



To a 10 mL, round bottomed flask was placed 5 mol% of FeCl<sub>3</sub> (2 mg, 0.0125 mmol, 0.05 equiv) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) under nitrogen atmosphere. A solution of (2*S*,3*R*,4*S*,5*R*,6*R*)-6-(acetoxymethyl)-5-((2*R*,3*R*,4*S*,5*R*,6*R*)-3,4-diacetoxy-6-(acetoxymethyl)-5-((2*R*,3*R*,4*S*,5*R*,6*R*)-3,4,5-triacetoxy-6-(acetoxymethyl)tetrahydro-2*H*-pyran-2-yloxy)tetrahydro-2*H*-pyran-2,3,4-triyl triacetate (241.5 mg, 0.25 mmol, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added. After having been stirred for 5 min at ambient temperature, a solution of TMSN<sub>3</sub> (50 μL, 44 mg, 0.38 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 28 h, the reaction mixture was

quenched with saturated aqueous NaHCO<sub>3</sub> solution (5 mL). Two layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 × 3 mL). The combined organic layers were washed with brine (5 mL), dried (MgSO<sub>4</sub>), filtered and evaporated under reduced pressure. The crude residue was purified by column chromatography (EtOAc/hexanes, 3/2) on silica gel to afford 206 mg (87%) of 2,3,4,6-tetra-*O*-acetyl-6-deoxy- $\alpha$ -D-glucopyranosyl-(1-4)-2,3,6-tri-*O*-acetyl-6-deoxy- $\alpha$ -D-glucopyranosyl-(1-4)-2,3,6-tri-*O*-acetyl-6-deoxy- $\beta$ -D-glucopyranosyl azide **8** as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 5.37-5.32 (m, 2H), 5.32 (d, *J* = 6.4, 1H), 5.25-5.21 (m, 2H), 5.03 (t, *J* = 9.8, 1H), 4.82 (dd, *J* = 10.4, 4.0, 1H), 4.76-4.67 (m, 3H), 4.49-4.41 (m, 2H), 4.29 (dd, *J* = 12.2, 4.4, 1H), 4.21 (dd, *J* = 12.4, 3.6, 1H), 4.15 (dd, *J* = 12.4, 3.2, 1H), 4.02 (dd, *J* = 12.4, 2.0, 1H), 3.96 (d, *J* = 9.2, 1H), 3.93-3.87 (m, 3H), 3.81-3.79 (m, 1H), 2.15 (s, 3H, COCH<sub>3</sub>), 2.12 (s, 3H, COCH<sub>3</sub>), 2.06 (s, 3H, COCH<sub>3</sub>), 2.02 (s, 3H, COCH<sub>3</sub>), 2.0 (s, 6H, COCH<sub>3</sub>), 1.98 (s, 3H, COCH<sub>3</sub>), 1.97 (s, 6H, 2 × COCH<sub>3</sub>), 1.96 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 170.5, 170.4, 170.3, 170.4, 170.2, 169.7, 169.6, 169.4, 169.3, 95.8, 95.6, 87.3, 74.8, 74.0, 73.4, 72.4, 71.6, 71.4, 70.3, 70.0, 69.2, 69.0, 68.4, 67.8, 62.6, 62.2, 61.3, 20.8, 20.7, 20.5, 20.5, 20.4; MS ESI (C<sub>38</sub>H<sub>51</sub>O<sub>25</sub>N<sub>3</sub>, 949): 972 (M+Na<sup>+</sup>, 100); TLC R<sub>f</sub> 0.10 (EtOAc/hexanes, 1/1); m.p. 95-96 °C; [α]<sub>D</sub><sup>25</sup> +69.2 (*c* 1.0, CHCl<sub>3</sub>) (lit.<sup>17b</sup> [α]<sub>D</sub><sup>25</sup> +65 (*c* 1.2, CHCl<sub>3</sub>)).

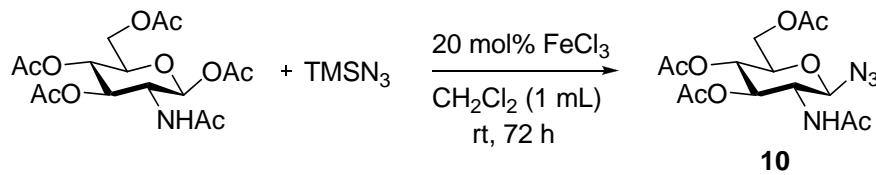
**3,4,6-tri-*O*-acetyl-2-deoxy-2-phthalimido- $\beta$ -D-glucopyranosyl azide (**9**).<sup>18</sup>**



To a 10 mL, round bottomed flask was placed 5 mol% of FeCl<sub>3</sub> (2 mg, 0.0125 mmol, 0.05 equiv) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) under nitrogen atmosphere. A solution of (2*S*,3*R*,4*R*,5*S*,6*R*)-6-(acetoxymethyl)-3-(1,3-dioxoisindolin-2-yl)tetrahydro-2*H*-pyran-2,4,5-triyl triacetate (119 mg, 0.25 mmol, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added. After having been stirred for 5 min at ambient temperature, a solution of TMSN<sub>3</sub> (50  $\mu$ L, 44 mg, 0.38 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 22 h, the reaction mixture was quenched with saturated aqueous NaHCO<sub>3</sub> solution (5 mL). Two layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2  $\times$  3 mL). The combined organic layers were washed with brine (5 mL), dried (MgSO<sub>4</sub>), filtered and evaporated under reduced pressure. The crude residue was purified by column chromatography (EtOAc/hexanes, 3/2) on silica gel to afford 105 mg (91%) of 3,4,6-tri-*O*-acetyl-2-deoxy-2-phthalimido- $\beta$ -D-glucopyranosyl azide **9** as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.87 (dd, *J* = 5.4, 3.0, 2H), 7.76 (dd, *J* =

5.5, 3.0, 2H), 5.80 (dd,  $J$  = 10.5, 9.2, 1H), 5.65 (d,  $J$  = 9.4, 1H), 5.19 (t,  $J$  = 9.7, 1H), 4.35 (dd,  $J$  = 4.7, 12.4, 1H), 4.24 (dd,  $J$  = 12.5, 2.3, 1H), 4.24 (dd,  $J$  = 10.6, 9.2, 1H), 3.99-3.96 (m, 1H), 2.12 (s, 3H, COCH<sub>3</sub>), 2.03 (s, 3H, COCH<sub>3</sub>), 1.86 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 170.6, 169.9, 169.3, 134.5, 131.2, 123.7, 73.9, 85.5, 70.4, 68.4, 61.7, 53.1, 20.6, 20.5, 20.3; MS ESI (C<sub>20</sub>H<sub>20</sub>O<sub>9</sub>N<sub>4</sub>, 460): 483 (M+Na<sup>+</sup>, 100); TLC R<sub>f</sub> 0.28 (EtOAc/hexanes, 1/1); m.p. 135-137 °C (lit.<sup>18a</sup> m.p. 135-138 °C); [α]<sub>D</sub><sup>25</sup> +39.6 (c 1.0, CHCl<sub>3</sub>) (lit.<sup>18b</sup> [α]<sub>D</sub><sup>25</sup> +38 (c 1.0, CHCl<sub>3</sub>)).

**3,4,6-tri-O-acetyl-2-N-acetyl-2-deoxy-β-D-glucopyranosyl azide (10).<sup>10</sup>**

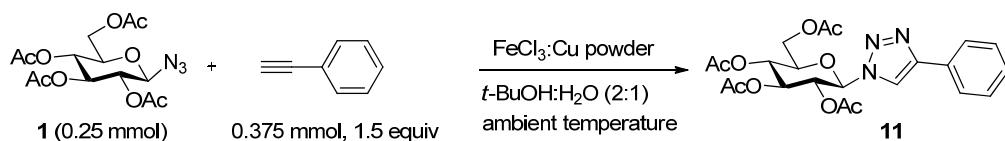


To a 10 mL, round bottomed flask was placed 5 mol% of FeCl<sub>3</sub> (8 mg, 0.05 mmol, 0.2 equiv) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) under nitrogen atmosphere. A solution of (2S,3R,4R,5S,6R)-3-acetamido-6-(acetoxymethyl)tetrahydro-2*H*-pyran-2,4,5-triyl triacetate (97 mg, 0.25 mmol, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added. After having been stirred for 5 min at ambient temperature, a solution of TMSN<sub>3</sub> (50 μL, 44 mg, 0.38 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 72 h, the

reaction mixture was quenched with saturated aqueous NaHCO<sub>3</sub> solution (5 mL). Two layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 × 3 mL). The combined organic layers were washed with brine (5 mL), dried (MgSO<sub>4</sub>), filtered and evaporated under reduced pressure. The crude residue was purified by column chromatography (EtOAc/hexanes, 3/2) on silica gel to afford 86 mg (93%) of 3,4,6-tri-*O*-acetyl-2-*N*-acetyl-2-deoxy-β-D-glucopyranosyl azide **10** as a white solid.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 6.09 (d, *J* = 8.8, 1H, NH), 5.25 (t, *J* = 10, 1H), 5.08 (t, *J* = 9.6, 1H), 4.78 (d, *J* = 9.2, 1H), 4.25 (dd, *J* = 4.8, 12.4, 1H), 4.16-4.13 (m, 1H), 3.91 (dd, *J* = 19.4, 9.3, 1H), 3.81 (m, 1H), 2.09 (s, 3H, COCH<sub>3</sub>), 2.024 (s, 3H, COCH<sub>3</sub>), 2.02 (s, 3H, COCH<sub>3</sub>), 1.97 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 170.9, 170.7, 170.6, 169.3, 88.3, 76.7, 73.9, 72.1, 68.1, 61.9, 54.0, 23.1, 20.7, 20.6, 20.5; MS ESI (C<sub>14</sub>H<sub>20</sub>O<sub>8</sub>N<sub>4</sub>, 372): 395 (M+Na<sup>+</sup>, 100); TLC R<sub>f</sub> 0.20 (EtOAc/hexanes, 1/1); m.p. 161-162 °C (lit.<sup>19</sup> m.p. 158-161 °C); [α]<sub>D</sub><sup>25</sup> -45.9 (*c* 1.0, CHCl<sub>3</sub>) (lit.<sup>10</sup> [α]<sub>D</sub><sup>23</sup> -47.7 (*c* 1.0, CHCl<sub>3</sub>)).

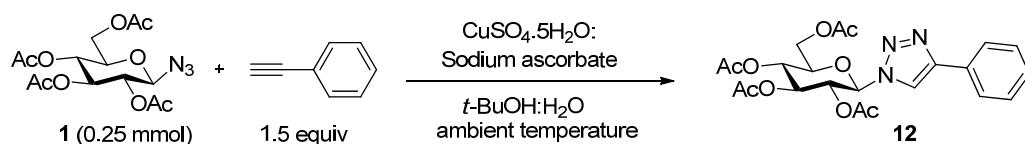
**Table S2: Optimization of FeCl<sub>3</sub>/Cu(0) ratio for cycloaddition reactions (i. e. click chemistry) between β-azido glucose and 2-phenylethyne.<sup>a</sup>**



Entry	FeCl <sub>3</sub> mol%	Cu powder mol%	FeCl <sub>3</sub> :Cu ratio	Time, h	Yield (%)
1	20	0	1:0	48	No reaction
2	0	100	0:1	48	No reaction
3	20	20	1:1	48	No reaction
4	20	30	1:1.5	48	Sluggish reaction
5	20	40	1:2	40	92
6	20	100	1:5	10	94

<sup>a</sup>Reaction details: β-azido glucose (0.25 mmol, 1 equiv), 2-phenylethyne (0.375 mmol, 1.5 equiv) in *t*-BuOH:H<sub>2</sub>O (2 mL:1 mL). <sup>b</sup>Isolated yield after purification by silica gel column chromatography.

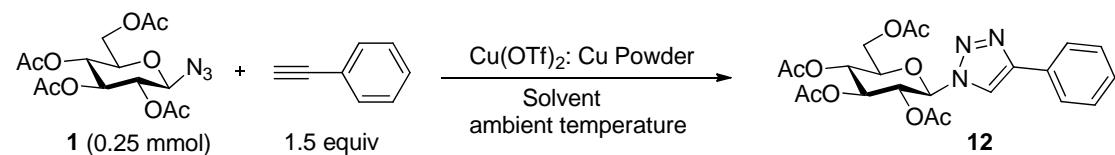
**Table S3: Control experiments for cycloaddition reactions between β-azido glucose and 2-phenylethyne<sup>a</sup> by using Sharpless and Fokin's reagents.<sup>20</sup>**



Entry	CuSO <sub>4</sub> (mol%)	sodium ascorbate (mol%)	CuSO <sub>4</sub> : sodium ascorbate ratio	Time (h)	Conversion (%) <sup>b</sup>	Yield (%) <sup>c</sup>
1	3	30	1:10	48	8	n.d.
2	20	40	1:2	20	7	n.d.
3	20	40	1:2	70	30	24 <sup>d</sup>
4	20	100	1:5	20	15	n.d.
5	20	100	1:5	70	45	40

<sup>a</sup>Reaction details: β-azido glucose (0.25 mmol, 1 equiv), 2-phenylethyne (0.375 mmol, 1.5 equiv) in *t*-BuOH:H<sub>2</sub>O (2 mL:1 mL). <sup>b</sup>Conversion of β-azido glucose **1** was determined by <sup>1</sup>H NMR analysis of the crude reaction mixture. <sup>c</sup>Isolated yield after purification by silica gel column chromatography. n.d. = not determined. <sup>d</sup>Remaining starting material was recovered.

**Table S4: Control experiments for cyclo-addition reaction between  $\beta$ -azido glucose and 2-phenylethyne<sup>a</sup> by using bimetallic catalyst Cu(OTf)<sub>2</sub>:Cu system.<sup>21</sup>**

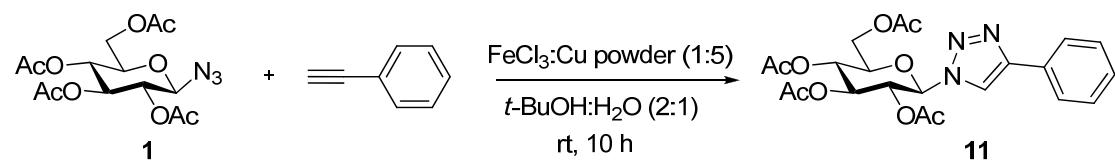


Entry	Cu(OTf) <sub>2</sub> (mol%)	Cu Powder (mol%)	Cu(OTf) <sub>2</sub> : Cu(0) ratio	Solvent	Time (h)	Conv. (%) <sup>b</sup>	Yield (%) <sup>c</sup>
1	10	10	1:1	CH <sub>3</sub> CN	40	21	16 <sup>d</sup>
2	10	50	1:5	CH <sub>3</sub> CN	12	quant.	92
3	10	10	1:1	<i>t</i> -BuOH:H <sub>2</sub> O	35	0	n.d.
4	10	50	1:5	<i>t</i> -BuOH:H <sub>2</sub> O	35	9	n.d.

<sup>a</sup>Reaction details:  $\beta$ -azido glucose (0.25 mmol, 1 equiv), 2-phenylethyne (0.375 mmol, 1.5 equiv) in CH<sub>3</sub>CN (6 mL) or *t*-BuOH:H<sub>2</sub>O (2 mL:1 mL). <sup>b</sup>Conversion of  $\beta$ -azido glucose **1** was determined by <sup>1</sup>H NMR analysis of the crude reaction mixture. <sup>c</sup>Isolated yield after purification by silica gel column chromatography. n.d. = not determined. <sup>d</sup>Remaining starting material was recovered.

**(B) Detailed procedures for FeCl<sub>3</sub>:Cu catalyzed 1,3-dipolar cycloaddition and analytical data for glycosyl 1,2,3-triazole conjugates **11-16**.**

**Phenyl 2,3,4,6-tetra-*O*-acetyl-6-deoxy- $\beta$ -D-glucopyranosyl 1,2,3-triazole (**11**).<sup>22</sup>**

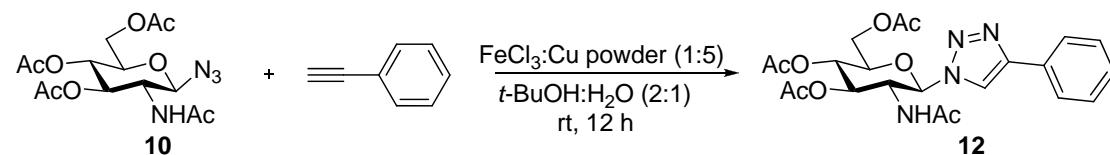


In a 10 mL, round bottomed flask were placed 20 mol% of FeCl<sub>3</sub> (8.1 mg, 0.05 mmol, 0.2 equiv) and Cu<sub>(s)</sub> powder (16 mg, 0.25 mmol, 1.0 equiv) in 1:1 mixture of *t*-BuOH:H<sub>2</sub>O (1 mL : 1 mL). The resulting mixture was stirred at ambient temperature for 5 min and a solution of 2,3,4,6-tetra-*O*-acetyl-6-deoxy- $\beta$ -D-glucopyranosyl azide

**1** (93 mg, 0.25 mmol, 1 equiv) in *t*-BuOH (1 mL) was added followed by addition of ethynylbenzene (41  $\mu$ L, 38 mg, 0.375 mmol, 1.5 equiv). The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 10 h, the organic volatiles were removed under reduced pressure and the resulting mixture was partitioned between H<sub>2</sub>O and CH<sub>2</sub>Cl<sub>2</sub> (5 mL each). The organic layer was separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2  $\times$  3 mL). The combined organic layers were washed with brine (10 mL), dried (MgSO<sub>4</sub>), filtered, passed through Celite bed to remove inorganic salts and concentrated under reduced pressure. The resulting crude residue was purified by column chromatography (EtOAc/hexanes, 1/1) on silica gel to afford 109 mg (92%) of phenyl 2,3,4,6-tetra-*O*-acetyl-6-deoxy- $\beta$ -D-glucopyranosyl 1,2,3-triazole **11** as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.0 (s, 1H), 7.83 (d, *J* = 7.2, 2H), 7.43 (t, *J* = 7.2, 2H,), 7.36 (d, *J* = 7.6, 1H), 5.93 (d, *J* = 9.2, 1H), 5.52 (t, *J* = 9.4, 1H), 5.44 (t, *J* = 9.4, 1H), 5.27 (t, *J* = 9.6, 1H), 4.33 (dd, *J* = 12.6, 5.0, 1H), 4.16 (dd, *J* = 12.6, 1.8, 1H), 4.05-4.02 (m, 1H), 2.084 (s, 3H, COCH<sub>3</sub>), 2.08 (s, 3H, COCH<sub>3</sub>), 2.04 (s, 3H, COCH<sub>3</sub>), 1.88 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  170.5, 169.9, 169.4, 169.0, 148.5, 129.9, 128.9, 128.6, 125.9, 117.7, 85.8, 75.3, 72.7, 70.2, 67.7, 61.6, 20.7, 20.6, 20.5, 20.2; MS ESI (C<sub>22</sub>H<sub>25</sub>O<sub>9</sub>N<sub>3</sub>, 475): 498 (M+Na<sup>+</sup>, 100); TLC R<sub>f</sub> 0.42 (EtOAc/hexanes, 1/1); m.p. 211-213 °C (lit.<sup>22b</sup> m.p.

213–215 °C);  $[\alpha]_D^{25} -54.5$  (*c* 1.0, CHCl<sub>3</sub>) (lit.<sup>22c</sup>  $[\alpha]_D^{20} -56.2$  (*c* 1.0, CHCl<sub>3</sub>)).

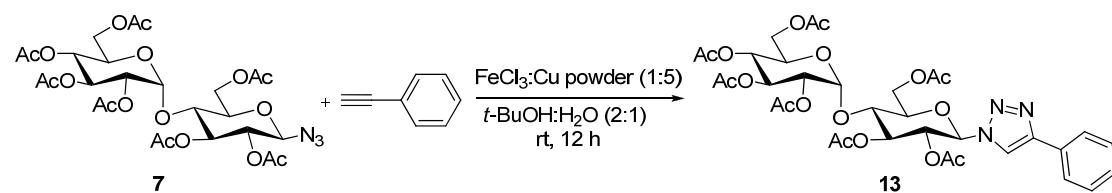
**Phenyl 3,4,6-tri-*O*-acetyl-2-*N*-acetyl-6-deoxy-β-D-glucopyranosyl 1,2,3-triazole (12).**



In a 10 mL, round bottomed flask were placed 20 mol% of FeCl<sub>3</sub> (8.1 mg, 0.05 mmol, 0.2 equiv) and Cu<sub>(s)</sub> powder (16 mg, 0.25 mmol, 1.0 equiv) in 1:1 mixture of *t*-BuOH:H<sub>2</sub>O (1 mL : 1 mL). The resulting mixture was stirred at ambient temperature for 5 min and a solution of 3,4,6-tri-*O*-acetyl-2-*N*-acetyl-2-deoxy-β-D-glucopyranosyl azide **10** (93 mg, 0.25 mmol, 1 equiv) in *t*-BuOH (1 mL) was added followed by addition of ethynylbenzene (41 μL, 38 mg, 0.375 mmol, 1.5 equiv). The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 12 h, the organic volatiles were removed under reduced pressure and the resulting mixture was partitioned between H<sub>2</sub>O and CH<sub>2</sub>Cl<sub>2</sub> (5 mL each). The organic layer was separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 × 3 mL). The combined organic layers were washed with brine (10 mL), dried (MgSO<sub>4</sub>), filtered, passed through Celite bed to remove inorganic salts and concentrated under reduced pressure. The resulting crude residue was purified by column chromatography (EtOAc/hexanes, 1/1)

on silica gel to afford 105 mg (89%) of phenyl 3,4,6-tri-*O*-acetyl-2-*N*-acetyl-6-deoxy- $\beta$ -D-glucopyranosyl 1,2,3-triazole **13** as a off white solid.  $^1\text{H}$  NMR (400 MHz, CDCl<sub>3</sub>/CD<sub>3</sub>OD)  $\delta$  8.04 (s, 1H), 7.52 (d, *J* = 7.3, 2H), 7.13 (t, *J* = 7.4, 2H), 7.05 (d, *J* = 7.2, 1H), 5.81 (d, *J* = 10, 1H), 5.19 (t, *J* = 10, 1H), 4.99 (t, *J* = 9.6, 1H), 4.36 (t, *J* = 10.2, 1H), 4.05 (dd, *J* = 12.4, 4.8, 1H), 3.89-3.82 (m, 2H), 1.79 (s, 3H), 1.787 (s, 3H, COCH<sub>3</sub>), 1.76 (s, 3H, COCH<sub>3</sub>), 1.46 (s, 3H, COCH<sub>3</sub>);  $^{13}\text{C}$  NMR (100 MHz, CDCl<sub>3</sub>/CD<sub>3</sub>OD)  $\delta$  171.5, 170.6, 170.1, 169.4, 147.6, 129.2, 128.3, 128.0, 125.2, 118.7, 85.4, 74.2, 72.0, 67.8, 61.4, 52.6, 21.4, 19.7, 19.7, 19.6; MS ESI (C<sub>22</sub>H<sub>26</sub>O<sub>8</sub>N<sub>4</sub>, 474): 497 (M+Na<sup>+</sup>, 100); TLC R<sub>f</sub> 0.40 (EtOAc/hexanes, 1/1); m.p. 228-229 °C; [α]<sub>D</sub><sup>25</sup> -62.4 (c 1.0, CHCl<sub>3</sub>).

**Phenyl 2,3,6-tri-*O*-acetyl-4-*O*-(2',3',4',6'-tetra-*O*-acetyl- $\beta$ -D-glucopyranosyl)- $\beta$ -D-glucopyranosyl 1,2,3-triazole (**13**).<sup>22</sup>**

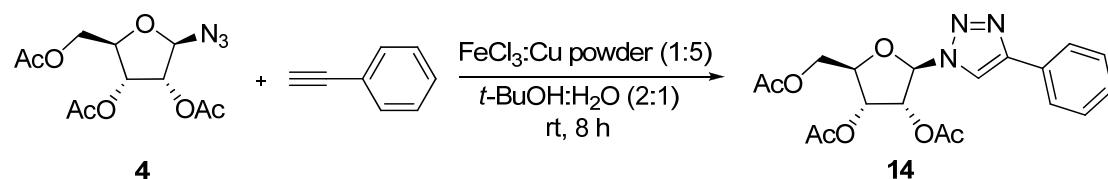


In a 10 mL, round bottomed flask were placed 20 mol% of FeCl<sub>3</sub> (8.1 mg, 0.05 mmol, 0.2 equiv) and Cu<sub>(s)</sub> powder (16 mg, 0.25 mmol, 1.0 equiv) in 1:1 mixture of *t*-BuOH:H<sub>2</sub>O (1 mL : 1 mL). The resulting mixture was stirred at ambient temperature for 5 min and a solution of 2,3,6-tri-*O*-acetyl-4-*O*-(2',3',4',6'-tetra-*O*-acetyl- $\beta$ -D-glucopyranosyl)- $\beta$ -D-glucopyranosylazide **7** (165 mg, 0.25 mmol, 1 equiv)

in *t*-BuOH (1 mL) was added followed by addition of ethynylbenzene (41  $\mu$ L, 38 mg, 0.375 mmol, 1.5 equiv). The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 12 h, the organic volatiles were removed under reduced pressure and the resulting mixture was partitioned between H<sub>2</sub>O and CH<sub>2</sub>Cl<sub>2</sub> (5 mL each). The organic layer was separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2  $\times$  3 mL). The combined organic layers were washed with brine (10 mL), dried (MgSO<sub>4</sub>), filtered, passed through Celite bed to remove inorganic salts and concentrated under reduced pressure. The resulting crude residue was purified by column chromatography (EtOAc/hexanes, 3/2) on silica gel to afford 166 mg (87%) of phenyl 2,3,6-tri-*O*-acetyl-4-*O*-(2',3',4',6'-tetra-*O*-acetyl- $\beta$ -D-glucopyranosyl)- $\beta$ -D-glucopyranosyl 1,2,3-triazole **13** as a off white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.92 (s, 1H), 7.81 (d, *J* = 7.6, 2H), 7.42 (t, *J* = 7.4, 2H), 7.34 (d, *J* = 7.4, 1H), 5.94 (d, *J* = 9.2, 1H), 5.51-5.36 (m, 4H), 5.07 (t, *J* = 9.8, 1H), 4.88 (dd, *J* = 10.8, 4, 1H), 4.49 (dd, *J* = 12.4, 2.0, 1H), 4.29-4.23 (m, 2H), 4.17 (t, *J* = 8.8, 1H), 4.08 (dd, *J* = 12.6, 1.8, 1H), 4.02-3.99 (m, 2H), 2.12 (s, 3H, COCH<sub>3</sub>), 2.10 (s, 3H, COCH<sub>3</sub>), 2.07 (s, 3H, COCH<sub>3</sub>), 2.03 (s, 3H, COCH<sub>3</sub>), 2.01 (s, 3H, COCH<sub>3</sub>), 1.85 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  170.5, 170.5, 170.3, 169.9, 169.4, 169.3, 148.4, 129.9, 128.9, 128.6, 125.9, 95.9, 85.3, 75.4, 75.2, 72.5,

70.8, 70.0, 69.2, 68.8, 67.9, 62.5, 61.5, 20.8, 20.7, 20.6, 20.6, 20.2; MS ESI ( $C_{34}H_{41}O_{17}N_3$ , 763): 786 ( $M+Na^+$ , 100); TLC  $R_f$  0.21 (EtOAc/hexanes, 1/1); m.p. 170–171 °C;  $[\alpha]_D^{25}$  -38.5 ( $c$  1.0, CHCl<sub>3</sub>).

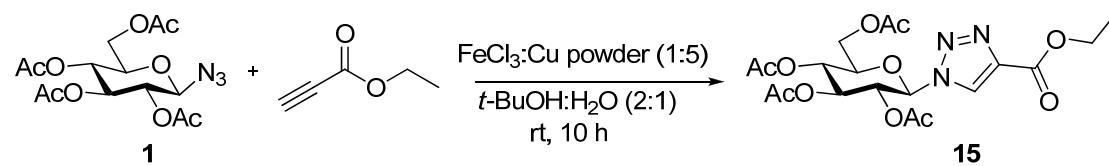
**Phenyl 2,3,5-tri-*O*-acetyl-β-D-ribofuranosyl 1,2,3-triazole (14).<sup>22</sup>**



In a 10 mL, round bottomed flask were placed 20 mol% of FeCl<sub>3</sub> (8.1 mg, 0.05 mmol, 0.2 equiv) and Cu<sub>(s)</sub> powder (16 mg, 0.25 mmol, 1.0 equiv) in 1:1 mixture of *t*-BuOH:H<sub>2</sub>O (1 mL : 1 mL). The resulting mixture was stirred at ambient temperature for 5 min and a solution of 2,3,5-tri-*O*-acetyl-β-D-ribofuranosyl azide **4** (75 mg, 0.25 mmol, 1 equiv) in *t*-BuOH (1 mL) was added followed by addition of ethynylbenzene (41 μL, 38 mg, 0.375 mmol, 1.5 equiv). The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 8 h, the organic volatiles were removed under reduced pressure and the resulting mixture was partitioned between H<sub>2</sub>O and CH<sub>2</sub>Cl<sub>2</sub> (5 mL each). The organic layer was separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 × 3 mL). The combined organic layers were washed with brine (10 mL), dried (MgSO<sub>4</sub>), filtered, passed through Celite bed to remove

inorganic salts and concentrated under reduced pressure. The resulting crude residue was purified by column chromatography (EtOAc/hexanes, 1/1) on silica gel to afford 97 mg (96%) of phenyl 2,3,5-tri-*O*-acetyl- $\beta$ -D-ribofuranosyl 1,2,3-triazole **14** as a colorless oil.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.96 (s, 1H), 7.80 (d,  $J$  = 7.2, 2H), 7.40 (t,  $J$  = 7.6, 2H), 7.32 (d,  $J$  = 7.2, 1H), 6.18 (d,  $J$  = 3.6, 1H), 5.9 (dd,  $J$  = 4.8, 4.0, 1H), 5.63 (t,  $J$  = 5.4, 1H), 4.48 (dd, 8.2, 4.2, 1H), 4.40 (dd,  $J$  = 12.4, 2.8, 1H), 4.23 (dd,  $J$  = 12.4, 4, 1H), 2.10 (s, 6H,  $2 \times \text{COCH}_3$ ), 2.02 (s, 3H,  $\text{COCH}_3$ );  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  170.2, 169.3, 169.2, 148.0, 129.9, 128.8, 125.7, 118.8, 89.9, 80.0, 74.1, 70.6, 62.7, 20.5, 20.3, 20.2; MS ESI ( $\text{C}_{19}\text{H}_{21}\text{O}_7\text{N}_3$ , 403): 426 ( $\text{M}+\text{Na}^+$ , 100); TLC  $R_f$  0.43 (EtOAc/hexanes, 1/1);  $[\alpha]_D^{25}$  +87.2 ( $c$  1.0,  $\text{CHCl}_3$ ).

**2,3,4,6-tetra-*O*-acetyl-(4'-ethylcarboxylate)- $\beta$ -D-glucopyranosyl 1,2,3-triazole (15).<sup>22</sup>**

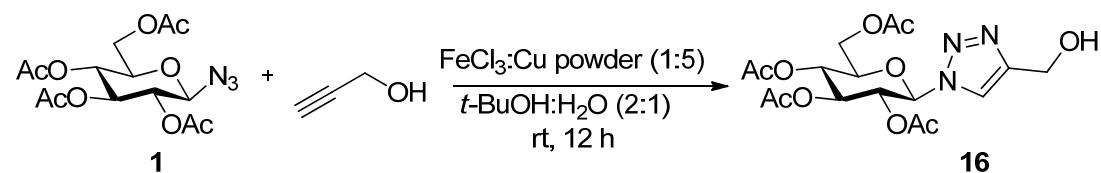


In a 10 mL, round bottomed flask, 20 mol% of  $\text{FeCl}_3$  (8.1 mg, 0.05 mmol, 0.2 equiv) and  $\text{Cu}_{(s)}$  powder (16 mg, 0.25 mmol, 1.0 equiv) were placed in 1:1 mixture of *t*-BuOH: $\text{H}_2\text{O}$  (1 mL : 1 mL). The resulting mixture was stirred at ambient temperature for 5 min and a solution of 2,3,4,6-tetra-*O*-acetyl-6-deoxy- $\beta$ -D-gluco-pyranosyl azide **1** (93 mg, 0.25 mmol, 1 equiv) in *t*-BuOH (1 mL) was added followed by addition of

ethyl propiolate (38  $\mu$ L, 37 mg, 0.375 mmol, 1.5 equiv). The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 10 h, the organic volatiles were removed under reduced pressure and the resulting mixture was partitioned between H<sub>2</sub>O and CH<sub>2</sub>Cl<sub>2</sub> (5 mL each). The organic layer was separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2  $\times$  3 mL). The combined organic layers were washed with brine (10 mL), dried (MgSO<sub>4</sub>), filtered, passed through Celite bed to remove inorganic salts and concentrated under reduced pressure. The resulting crude residue was purified by column chromatography (EtOAc/hexanes, 1/1) on silica gel to afford 112 mg (95%) of 2,3,4,6-tetra-*O*-acetyl-(4'-ethyl carboxylate)- $\beta$ -D-glucopyranosyl 1,2,3-triazole **15** as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.33 (s, 1H), 5.92 (d, *J* = 8.8, 1H), 5.46-5.37 (m, 2H), 5.24 (t, *J* = 9.6, 1H), 4.42 (q, *J* = 7.1, 2H), 4.31 (dd, *J* = 12.6, 5.0, 1H), 4.15 (dd, *J* = 12.8, 1.4, 1H), 4.03-4.0 (m, 1H), 2.08 (s, 3H, COCH<sub>3</sub>), 2.06 (s, 3H, COCH<sub>3</sub>), 2.02 (s, 3H, COCH<sub>3</sub>), 1.89 (s, 3H, COCH<sub>3</sub>), 1.41 (t, *J* = 7, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  170.4, 169.8, 169.3, 168.8, 160.1, 140.9, 126.0, 85.8, 75.3, 72.3, 70.4, 67.5, 61.5, 61.4, 20.6, 20.4, 20.0, 14.2; MS ESI (C<sub>19</sub>H<sub>25</sub>O<sub>11</sub>N<sub>3</sub>, 494): 517 (M+Na<sup>+</sup>, 100); TLC R<sub>f</sub> 0.39 (EtOAc/hexanes, 1/1); m.p. 170-171 °C (lit.<sup>22d</sup> m.p. 172-174 °C); [ $\alpha$ ]<sub>D</sub><sup>25</sup> -46.3 (*c* 1.0, CHCl<sub>3</sub>) (lit.<sup>22d</sup> [ $\alpha$ ]<sub>D</sub><sup>25</sup> -62 (*c* 0.2, CHCl<sub>3</sub>)).

**2,3,4,6-tetra-O-acetyl-(4'-hydroxymethyl)-6-deoxy- $\beta$ -D-glucopyranosyl**

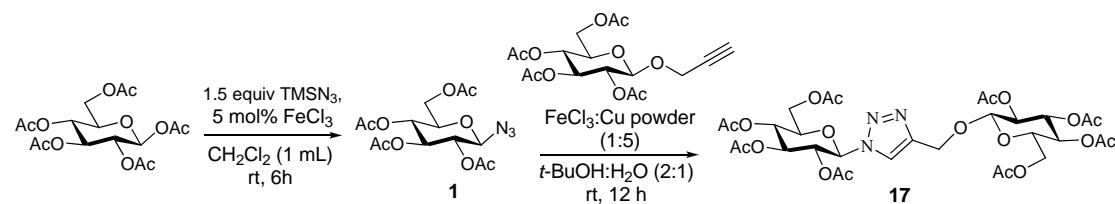
**1,2,3-triazole (16).<sup>22</sup>**



In a 10 mL, round bottomed flask were placed 20 mol% of  $\text{FeCl}_3$  (8.1 mg, 0.05 mmol, 0.2 equiv) and  $\text{Cu}_{(s)}$  powder (16 mg, 0.25 mmol, 1.0 equiv) in 1:1 mixture of  $t\text{-BuOH:H}_2\text{O}$  (1 mL : 1 mL). The resulting mixture was stirred at ambient temperature for 5 min and a solution of 2,3,4,6-tetra-O-acetyl-6-deoxy- $\beta$ -D-glucopyranosyl azide **1** (93 mg, 0.25 mmol, 1 equiv) in  $t\text{-BuOH}$  (1 mL) was added followed by addition of prop-2-yn-1-ol (22  $\mu\text{L}$ , 21 mg, 0.375 mmol, 1.5 equiv). The resulting reaction mixture was stirred at ambient temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 12 h, the organic volatiles were removed under reduced pressure and the resulting mixture was partitioned between  $\text{H}_2\text{O}$  and  $\text{CH}_2\text{Cl}_2$  (5 mL each). The organic layer was separated and the aqueous layer was extracted with  $\text{CH}_2\text{Cl}_2$  ( $2 \times 3$  mL). The combined organic layers were washed with brine (10 mL), dried ( $\text{MgSO}_4$ ), filtered, passed through Celite bed to remove inorganic salts and concentrated under reduced pressure. The resulting crude residue was purified by column chromatography (EtOAc/hexanes, 1/1) on silica

gel to afford 100 mg (93%) of 2,3,4,6-tetra-*O*-acetyl-(4'-hydroxymethyl)-6-deoxy- $\beta$ -D-glucopyranosyl 1,2,3-triazole **16** as a off white solid.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.84 (s, 1H), 5.89 (d,  $J$  = 8.4, 1H), 5.46-5.37 (m, 2H), 5.23 (t,  $J$  = 9.4, 1H), 4.74 (brs, 2H), 4.25 (dd,  $J$  = 12.6, 5, 1H), 4.10 (d,  $J$  = 12.4, 1H), 4.01-3.99 (m, 1H), 3.42 (brs, 1H), 2.021 (s, 3H,  $\text{COCH}_3$ ), 2.02 (s, 3H,  $\text{COCH}_3$ ), 1.98 (s, 3H,  $\text{COCH}_3$ ), 1.82 (s, 3H,  $\text{COCH}_3$ );  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  170.5, 169.9, 169.4, 169.0, 148.5, 120.2, 85.6, 75.0, 72.6, 70.3, 67.7, 61.5, 56.3, 20.6, 20.5, 20.4, 20.1; MS ESI ( $\text{C}_{17}\text{H}_{23}\text{O}_{10}\text{N}_3$ , 429): 452 ( $\text{M}+\text{Na}^+$ , 100); TLC  $R_f$  0.40 (EtOAc/hexanes, 1/1); m.p. 149-151 °C (lit.<sup>22d</sup> m.p. 152-154 °C);  $[\alpha]_D^{25}$  -9.1 (c 1.0,  $\text{CHCl}_3$ ) (lit.<sup>22c</sup>  $[\alpha]_D^{20}$  -6 (c 1.0,  $\text{CHCl}_3$ )).

**(C) Procedure for one pot synthesis of 2,3,4,6-tetra-*O*-acetyl-4'-*O*-(2,3,4,6-tetra-*O*-acetyl-6-deoxy-glucopyranose)-6-deoxy- $\beta$ -D-glucopyranosyl triazole (17).<sup>22</sup>**



To a 10 mL, round bottomed flask was placed 5 mol% of  $\text{FeCl}_3$  (2 mg, 0.0125 mmol, 0.05 equiv) in anhydrous  $\text{CH}_2\text{Cl}_2$  (0.5 mL) under nitrogen atmosphere. A solution of (2*S*,3*S*,4*S*,5*R*)-6-(acetoxymethyl)tetrahydro-2*H*-pyran-2,3,4,5-tetrayl tetraacetate (98 mg, 0.25 mmol, 1 equiv) in  $\text{CH}_2\text{Cl}_2$  (0.25 mL) was added. After having been stirred for 5 min, a solution of  $\text{TMSN}_3$  (50  $\mu\text{L}$ , 44 mg, 0.38 mmol, 1.5 equiv) in  $\text{CH}_2\text{Cl}_2$  (0.25 mL) was added slowly. The resulting reaction mixture was stirred at ambient

temperature and the progress of the reaction was monitored by TLC. After complete consumption of starting material in 6 h as judged by TLC, the reaction mixture was concentrated under reduced pressure and the resulting residue was dried in *vacuo*. The residue was then suspended in a mixture of *t*-BuOH:H<sub>2</sub>O (1 mL: 1 mL). After having been stirred for 5 min, a solution of 15 mol% of FeCl<sub>3</sub> (6.0 mg, 0.0375 mmol, 0.15 equiv) and (3*R*,4*S*,5*S*,6*R*)-2-(acetoxymethyl)-6-(prop-2-ynylloxy)tetrahydro-2*H*-pyran-3,4,5-triyl triacetate (145 mg, 0.375 mmol, 1.5 equiv) in *t*-BuOH (1 mL) was added followed by addition of Cu<sub>(s)</sub> powder (16 mg, 0.25 mmol, 1.0 equiv). After having been stirred for 12 h at ambient temperature, the organic volatiles were removed under reduced pressure and the resulting mixture was partitioned between H<sub>2</sub>O and CH<sub>2</sub>Cl<sub>2</sub> (5 mL each). The organic layer was separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 × 3 mL). The combined organic layers were washed with brine (10 mL), dried (MgSO<sub>4</sub>), filtered and passed through Celite bed to remove inorganic salts. The organic layer was then concentrated under reduced pressure and the resulting crude residue was purified by column chromatography (EtOAc/hexanes, 1/1) on silica gel to afford 171 mg (90%) of **17** as a white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.80 (s, 1H), 5.86 (d, *J* = 8.8, 1H), 5.42-5.39 (m, 2H), 5.23 (t, *J* = 9.6, 1H), 5.16 (t, *J* = 9.4, 1H), 5.08 (t, *J* = 9.6, 1H), 4.97 (td, *J* = 8.2, 1.2, 1H), 4.90 (d, *J* = 12.8, 1H), 4.79 (d, *J* = 12.8, 1H), 4.53 (d, *J* = 8.0, 1H), 4.31-4.25 (m, 2H), 4.19-4.10 (m,

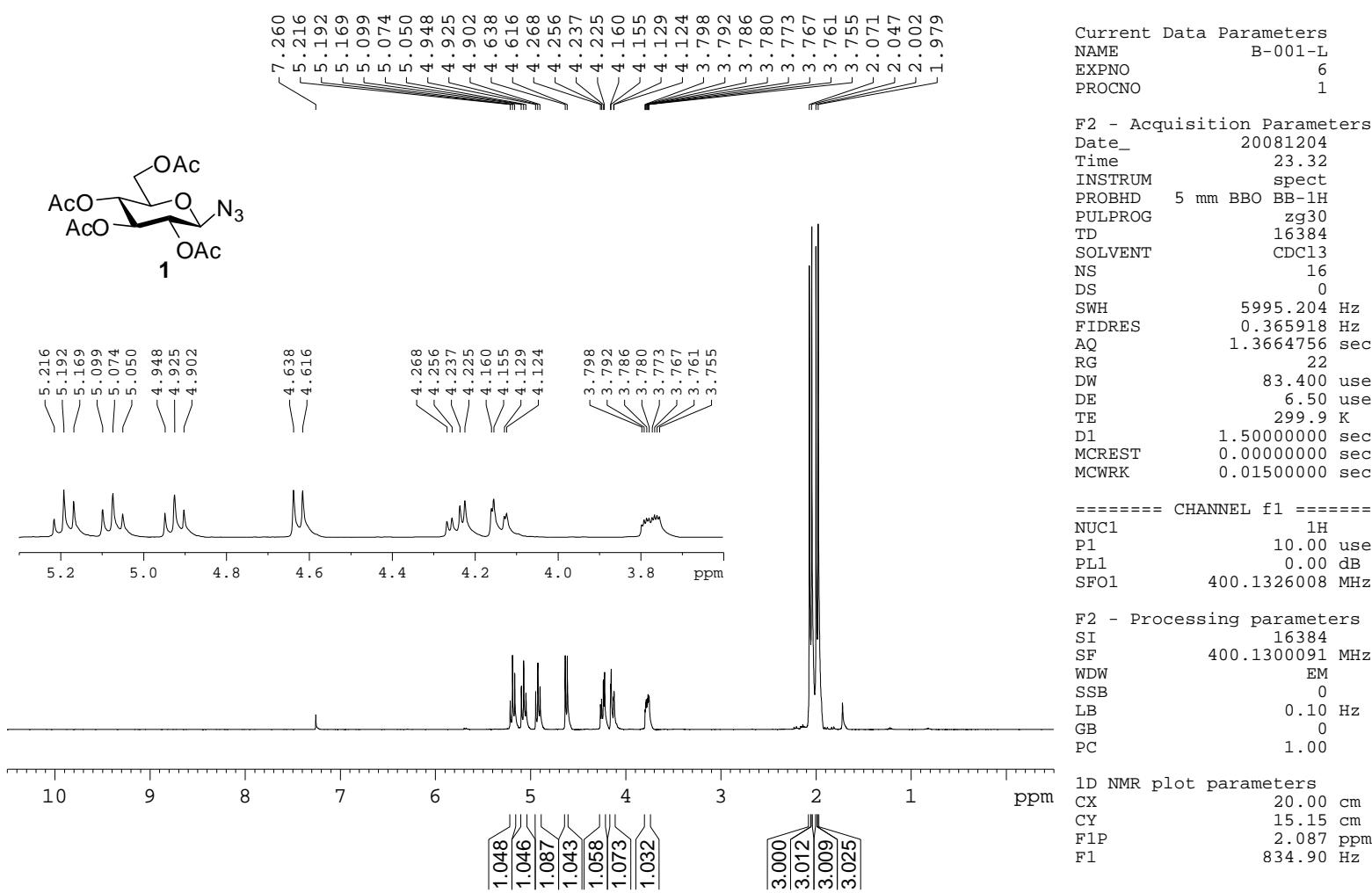
1H), 4.02-3.98 (m, 1H), 3.76-3.72 (m, 1H), 2.10 (s, 3H, COCH<sub>3</sub>), 2.06 (s, 3H, COCH<sub>3</sub>), 2.05 (s, 3H, COCH<sub>3</sub>), 2.02 (s, 3H, COCH<sub>3</sub>), 2.00 (s, 3H, COCH<sub>3</sub>), 1.96 (s, 3H, COCH<sub>3</sub>), 1.94 (s, 3H, COCH<sub>3</sub>), 1.88 (s, 3H, COCH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 170.7, 170.4, 170.2, 169.8, 169.4, 169.3, 169.0, 144.2, 121.9, 98.7, 75.2, 72.6, 72.4, 71.7, 71.0, 70.4, 68.3, 67.6, 61.8, 61.7, 61.4, 20.7, 20.6, 20.5, 20.5, 20.0; MS ESI (C<sub>30</sub>H<sub>39</sub>O<sub>19</sub>N<sub>3</sub>, 745): 768 (M+Na<sup>+</sup>, 100); TLC R<sub>f</sub> 0.33 (EtOAc/hexanes, 1/1); m.p. 198-200 °C (lit.<sup>23</sup> m.p. 199-199.5 °C); [α]<sub>D</sub><sup>25</sup> -41.6 (c 1.0, CHCl<sub>3</sub>).

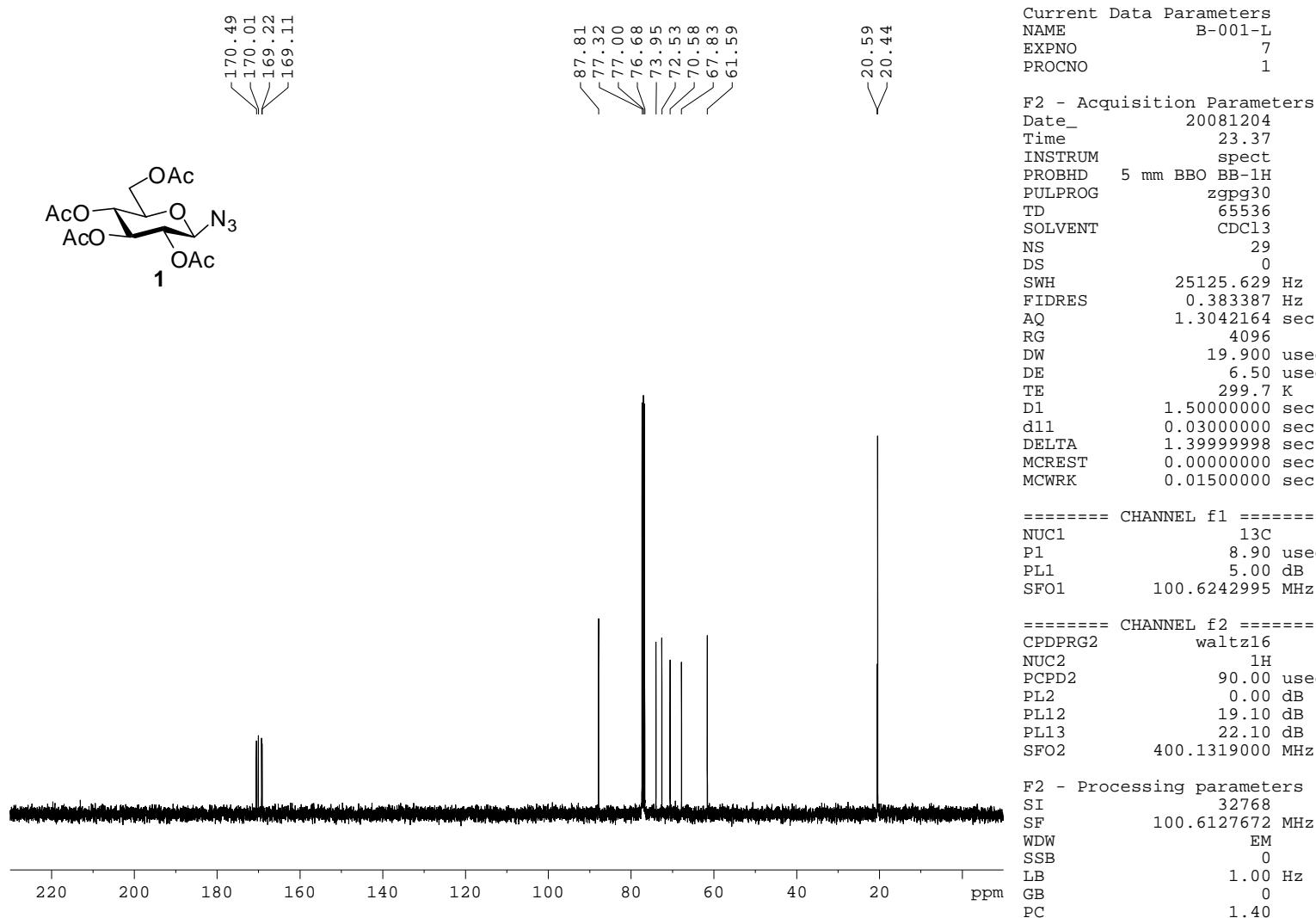
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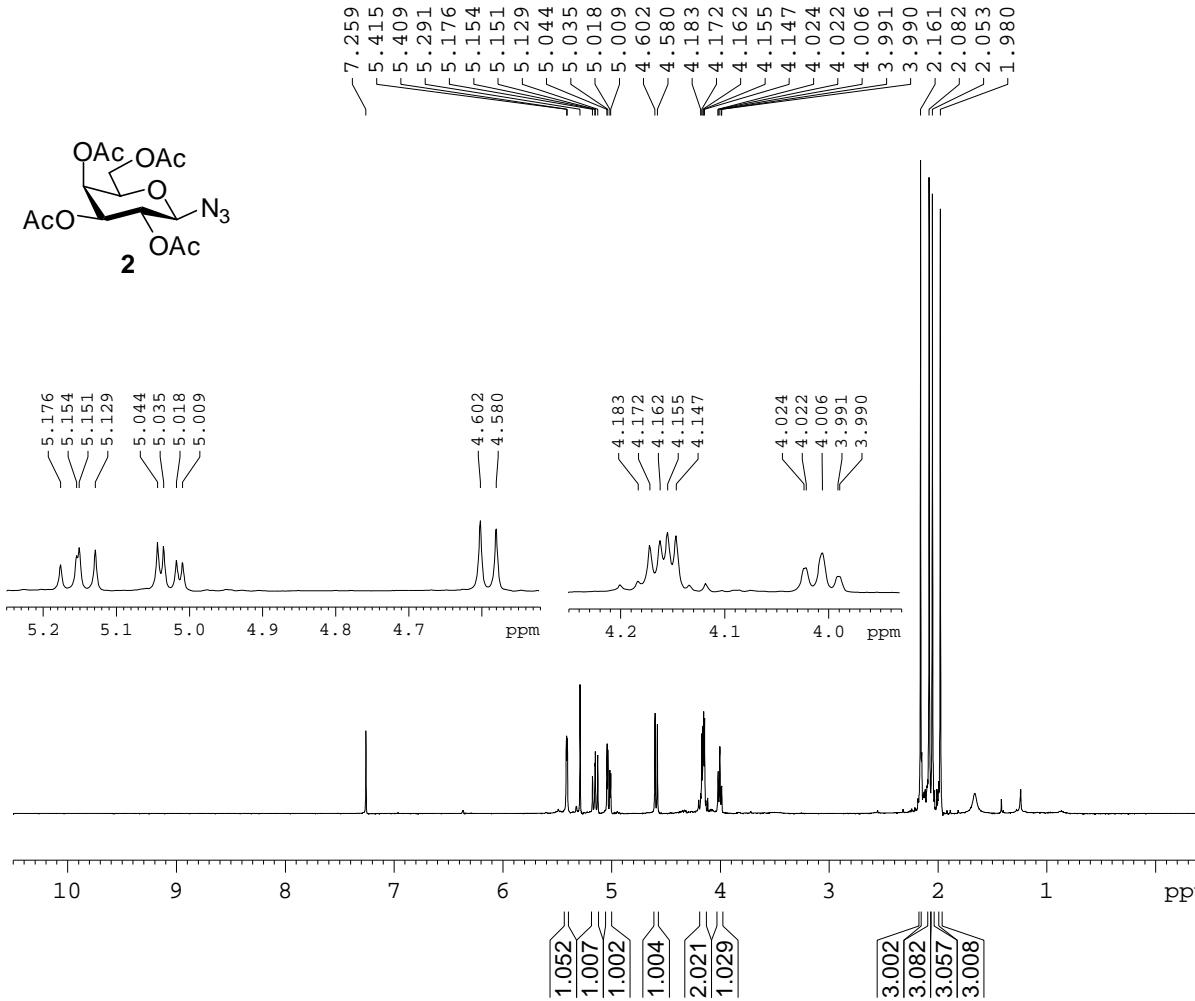
- (1) H. Bretting, G. Jacobs, I. Benecke, W.A. Koenig and J. Thiem, *Carbohydr. Res.*, 1985, **139**, 225.
  - (2) (a) G. Hodosi and P. Kovac, *Carbohydr. Res.*, 1997, **303**, 239; (b) J.-P. Utile and D. Gagnaire, *Carbohydr. Res.*, 1982, **106**, 43.
  - (3) H. Ishii, I. Kitagawa, K. Matsushita, K. Shirakawa and K. Tori, *Tetrahedron Lett.*, 1981, **22**, 1529.
  - (4) J. Xue, Y. Pan and Z. Guo, *Tetrahedron Lett.*, 2002, **43**, 1599.
  - (5) A. Yoshida, T. Hayashi, N. Takeda, S. Oida and E. Ohki, *Chem. Pharm. Bull.*, 1981, **29**, 1854.
  - (6) N. Thiebault, D. Lesur, P. Gode, V. Moreau and F. Djedaini-Pilard, *Carbohydr. Res.*, 2008, **343**, 2719.
  - (7) (a) C.O. Kappe, *Angew. Chem. Int. Ed. Engl.*, 2004, **43**, 6250; (b) D. D. Heard and R. Barker, *J. Org. Chem.*, 1968, **33**, 740.
- 8 (a) For a discussion on the hazards associated with azides, see: *Prudent Practice for*

- Handling Hazardous Chemicals in Laboratories*, National Academic Press, Washington, DC, 1983, p. 87-88; (b) For human toxicity, see: *The Merck Index*, 12<sup>th</sup> ed., Merck & Co., Rahway, NJ, 1996; p. 4818 and 8726.
- 9 (a) R. E. Conrow and W. D. Dean, *Org. Process res. Dev.*, 2008, **12**, 2285; (b) A. Hassner, M. Stern and H. E. Gottlieb, *J. Org. Chem.*, 1990, **55**, 2304.
- (10) F. D. Tropper, F. O. Andersson, S. Braun and R. Roy, *Synthesis*, 1992, 618.
- (11) H. J. Berthold, S. Franke, J. Thiem and T. Schotten, *J. Org. Chem.*, 2010, **75**, 3859.
- (12) H. Paulsen, Z. Gyorgydecik and M. Friedmann, *Chem. Ber.*, 1974, **107**, 1568.
- (13) K. Slamova, P. Marhol, K. Bezouska, V. Kren, L. Lindkvist, S. G. Hansen and H. H. Jensen, *Bioorg. Med. Chem. Lett.*, 2010, **20**, 4263.
- (14) A. Stimac and J. Kobe, *Carbohydr. Res.*, 1992, **232**, 359.
- (15) A. E. Meslouti, D. Beaupere, G. Demailly and R. Uzan, *Tetrahedron Lett.*, 1994, **35**, 3913.
- (16) (a) X. Zheng, J. Morgan, S. K. Pandey, Y. Chen, E. Tracy, H. Baumann, J. R. Missert, C. Batt, J. Jackson, D. A. Bellnier, B. W. Henderson, and R. K. Pandey, *J. Med. Chem.*, 2009, **52**, 4306; (b) M. Tamura and H. Okai, *Carbohydr. Res.*, 1984, **133**, 207.
- (17) (a) L. Marmuse, S. A. Nepogodiev and R. A. Field, *Org. Biomol. Chem.*, 2005, **3**, 2225; (b) S. A. Nepogodiev, S. Dedola, L. Marmuse, M. Y. Oliveira, and R. A. Field, *Carbohydr. Res.*, 2007, **342**, 529.
- (18) (a) A. Dan, M. Lergenmüller, M. Amano, Y. Nakahara, T. Ogawa and Y. Ito, *Chem. Eur. J.*, 1998, **4**, 2182; (b) L. Szilagyi, and Z. Gyoergydeak, *Carbohydr. Res.*, 1985, **143**, 21.
- (19) D. Macmillan, A. M. Daines, M. Bayrhuber and S. L. Flitsch, *Org. Lett.*, 2002, **4**, 1467.
- (20) (a) V. V. Rostovtsev, L. G. Green, V. V. Fokin and K. B. Sharpless, *Angew. Chem.*

- Int. Ed. Engl.*, 2002, **41**, 2596; (b) S. Dedola, D. L. Hughes, S. A. Nepogodiev, M. Rejzek and R. A. Field, *Carbohydr. Res.*, 2010, **345**, 1123.
- (21) S. K. Yousuf, S. C. Taneja and D. Mukherjee, *J. Org. Chem.*, 2010, **75**, 3097.
- (22) (a) R. Kumar, P. R. Maulik and A.K. Misra, *Glycoconj. J.*, 2008, **25**, 595; (b) L. L. Rossi and A. Basu, *Bioorg. Med. Chem. Lett.*, 2005, **15**, 3596; (c) S. Dedola, S. A. Nepogodiev, M. Rejzek, R. A. Field and D. L. Hughes, *Carbohydr. Res.*, 2010, **345**, 1123; (d) Bokor, E. Somsak, L. Docsá, T. and Gergely, P. *Bioorg. Med. Chem.*, 2010, **18**, 1171.
- (23) V. O. Rodionov, S. I. Presolski, S. Gardinier, Y. H. Lim and M. G. Finn, *J. Am. Chem. Soc.*, 2007, **129**, 12696.







Current Data Parameters  
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EXPNO 1  
PROCNO 1

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PULPROG       zg30
TD             16384
SOLVENT        CDCl3
NS              16
DS                 0
SWH            5995.204 Hz
FIDRES        0.365918 Hz
AQ             1.3664756 sec
RG                128
DW             83.400 usec
DE               6.50 usec
TE                298.0 K
D1      1.50000000 sec
MCREST        0.00000000 sec
MCWRK         0.01500000 sec

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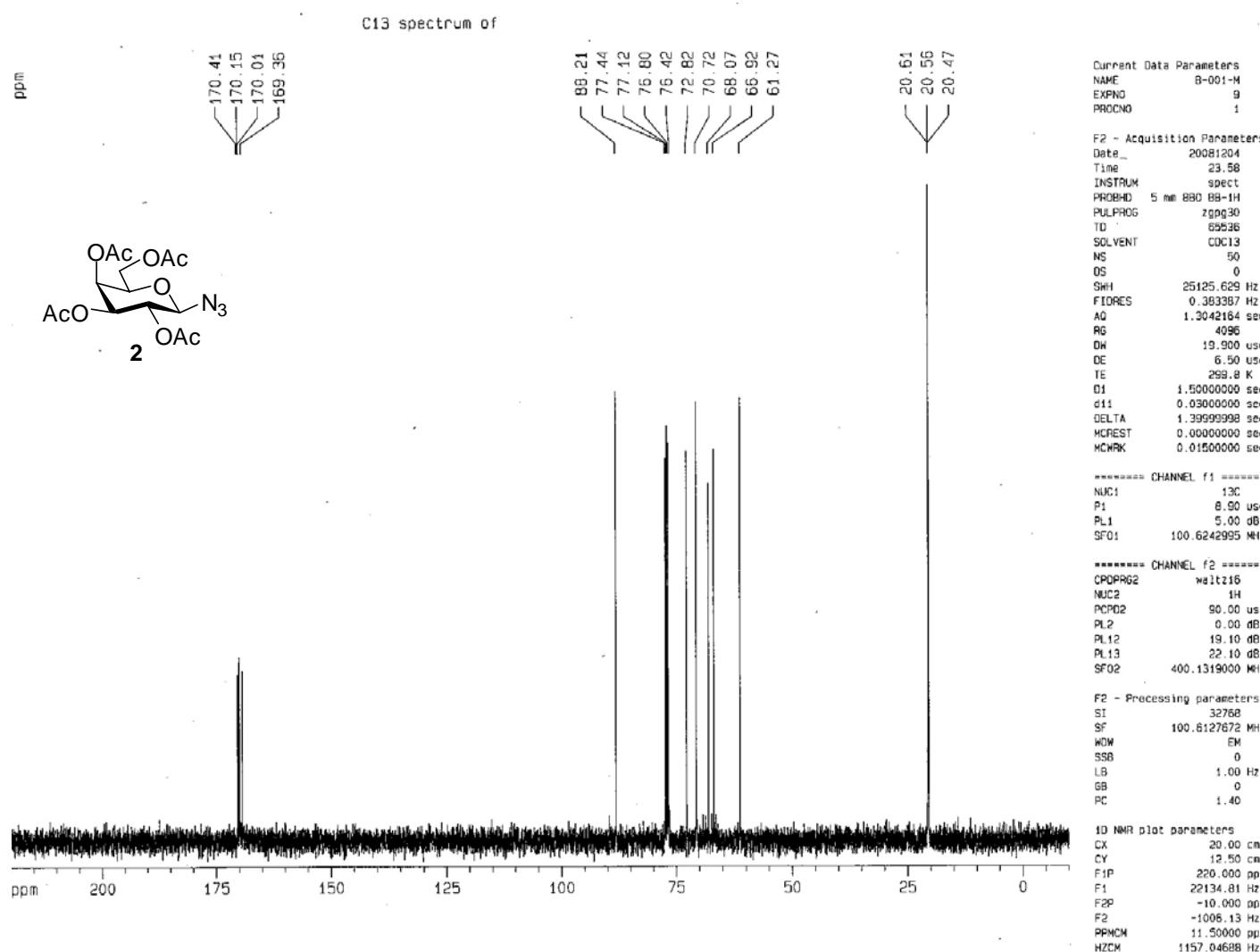
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PL1          0.00 dB  
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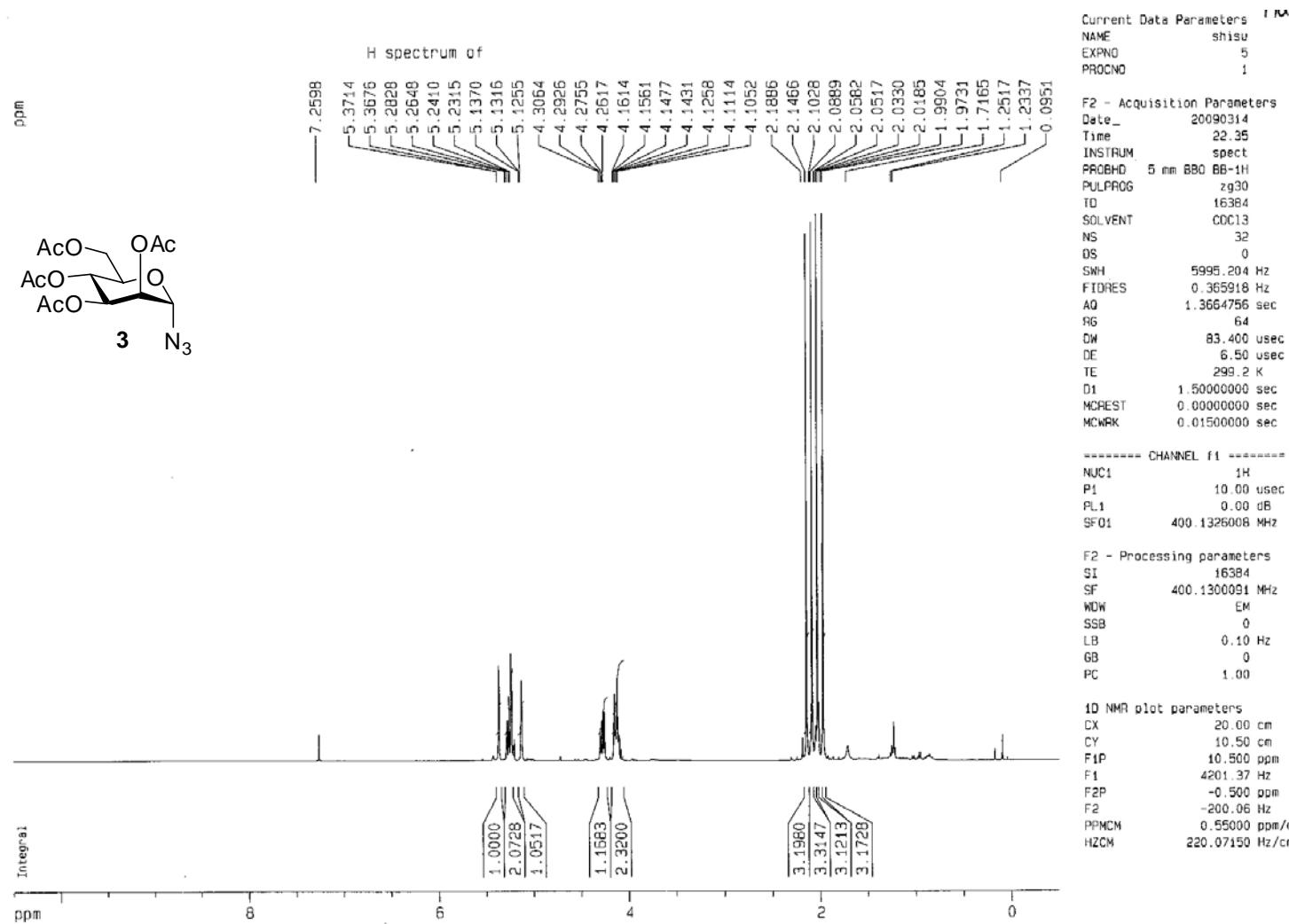
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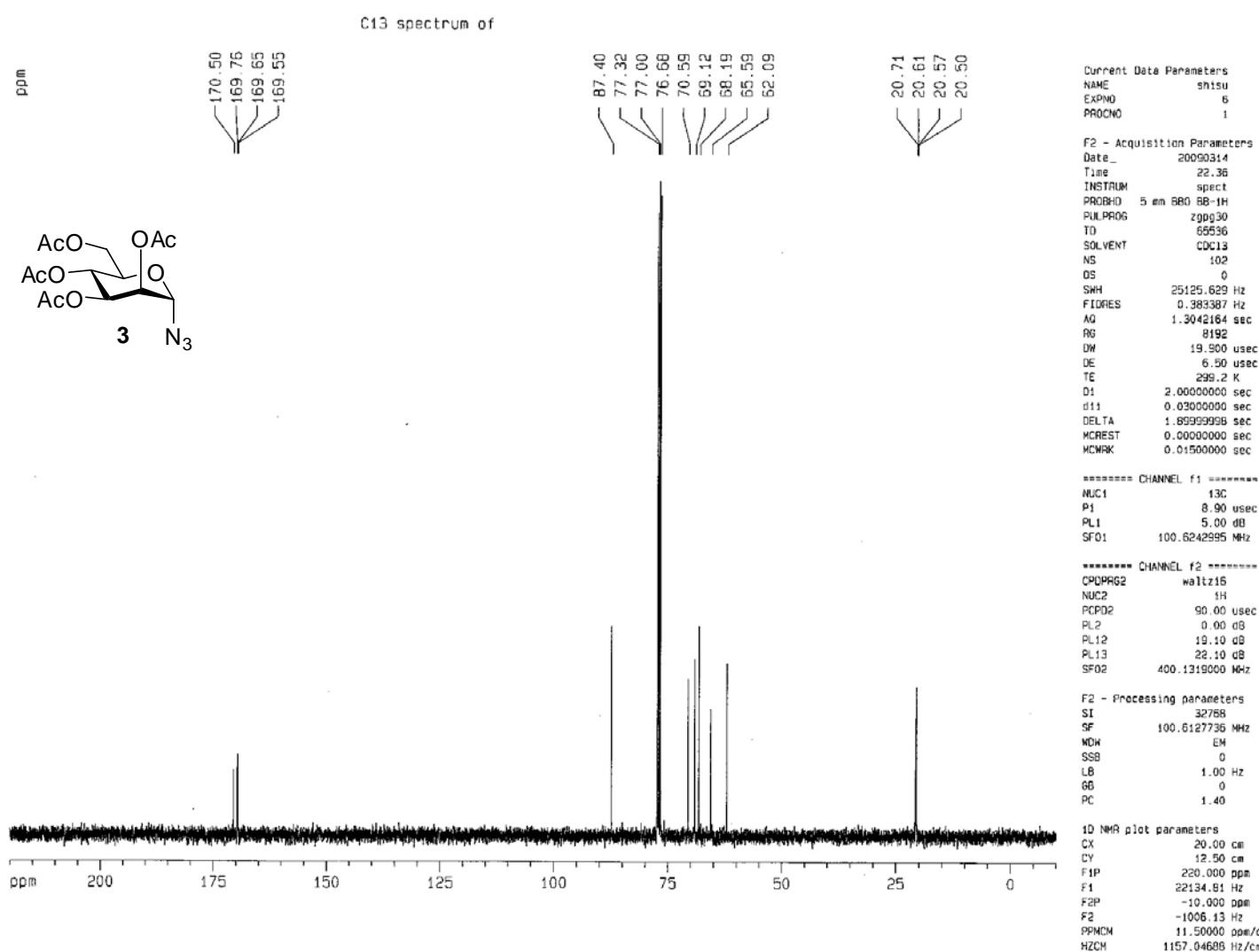
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PC           1.00

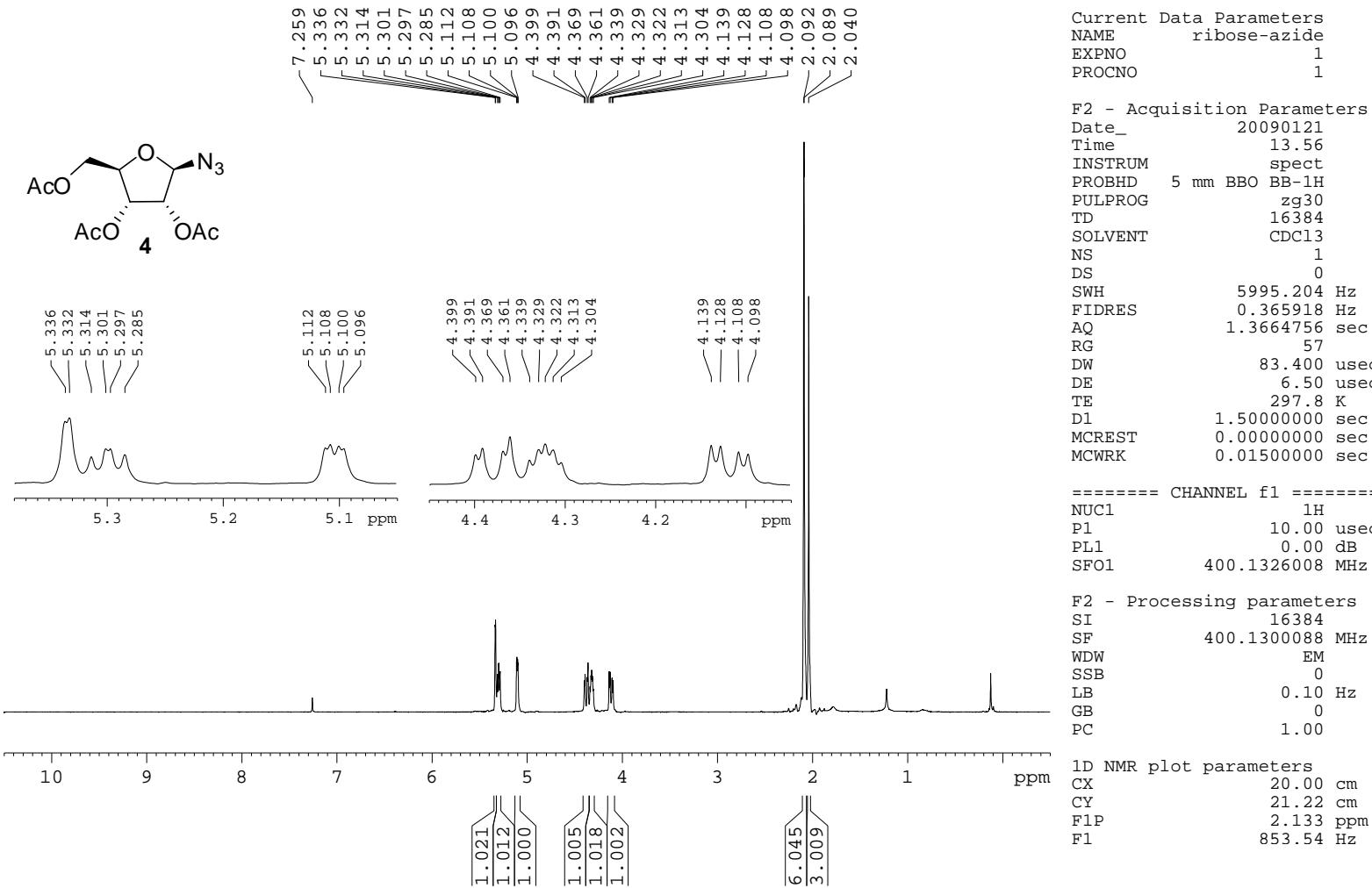
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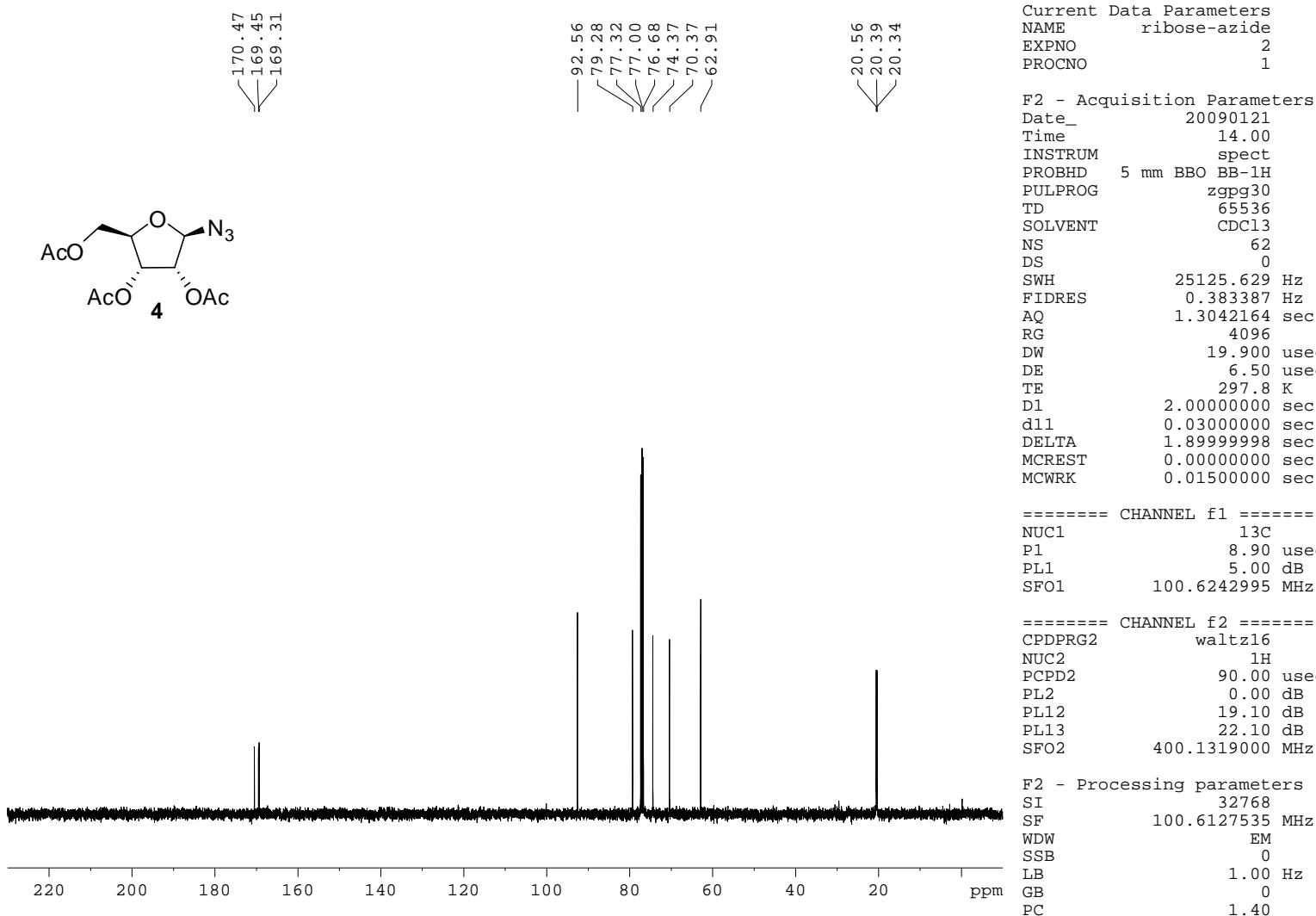
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CY 10.50 cm  
F1P 2.240 ppm  
F1 896.10 Hz

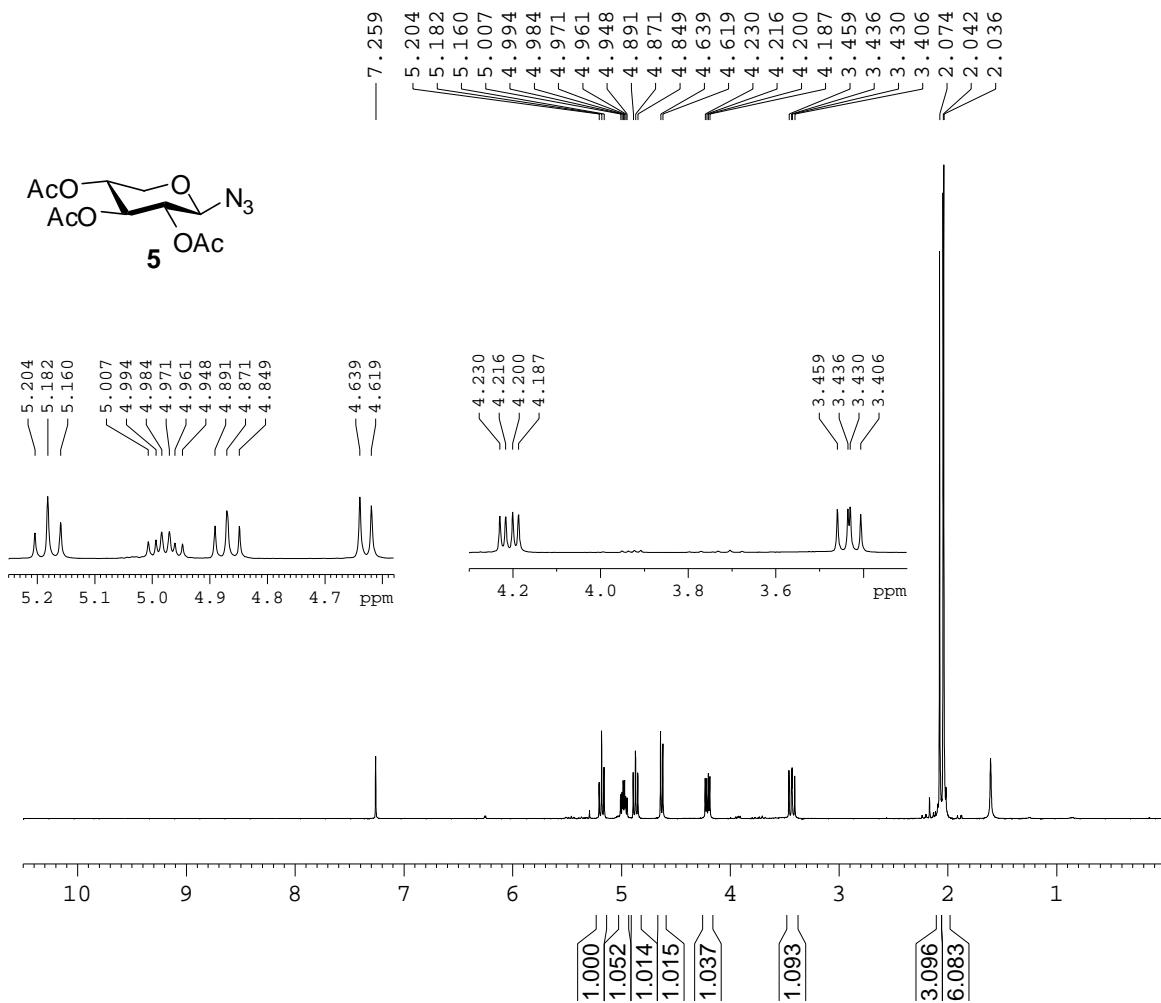












Current Data Parameters  
NAME xylose azide  
EXPNO 1  
PROCNO 1

```

F2 - Acquisition Parameters
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Time           20.32
INSTRUM       spect
PROBHD        5 mm BBO BB-1H
PULPROG      zg30
TD             16384
SOLVENT        CDC13
NS              16
DS                 0
SWH            5995.204 Hz
FIDRES       0.365918 Hz
AQ            1.3664756 sec
RG             203.2
DW             83.400 usec
DE               6.50 usec
TE             298.9 K
D1           1.50000000 sec
MCREST        0.00000000 sec
MCWRK        0.01500000 sec

```

===== CHANNEL f1 =====  
NUC1 1H  
P1 10.00 used  
PL1 0.00 dB  
SFO1 400.1326008 MHz

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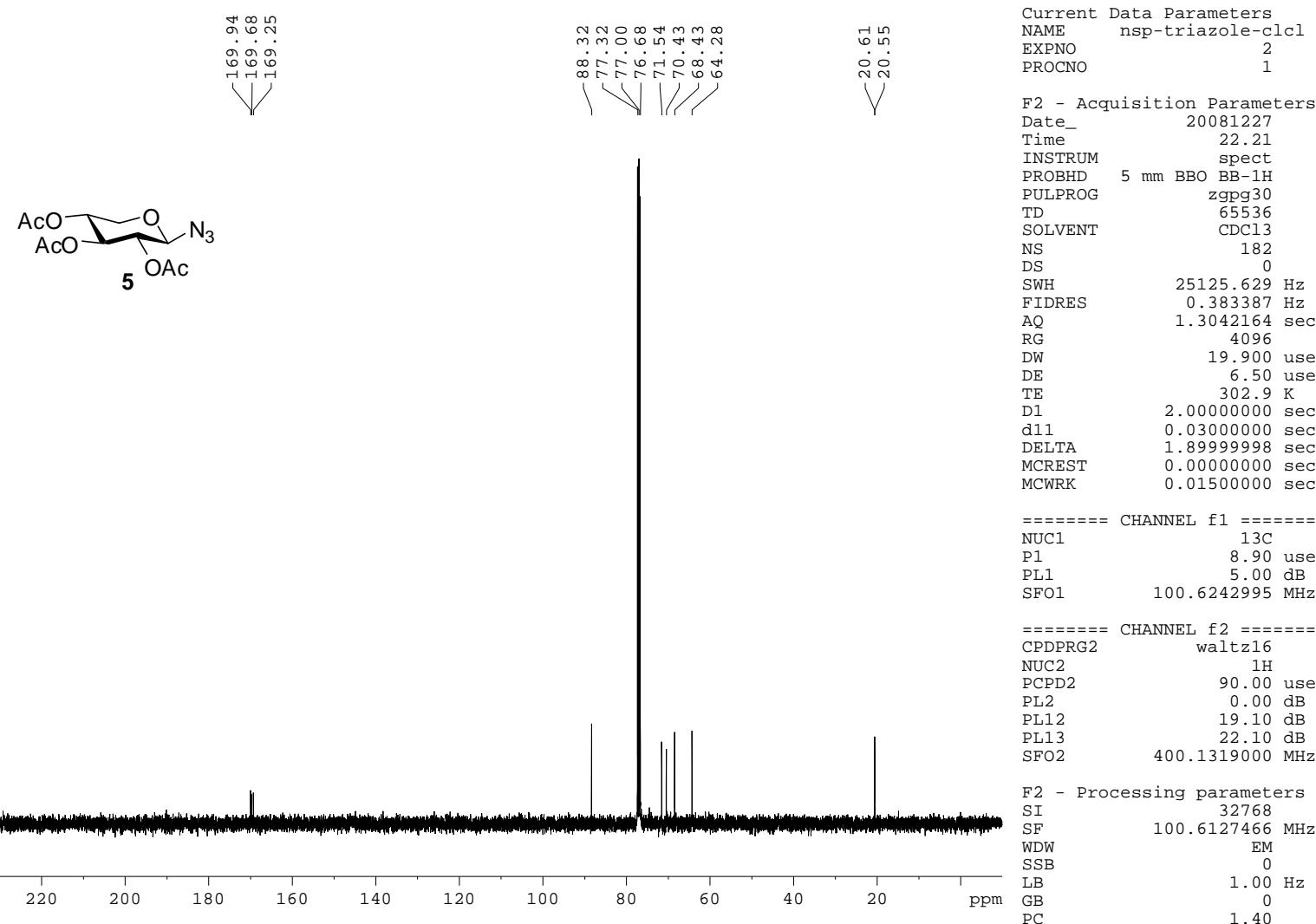
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WDW          EM
SSB          0
LB           0.10 Hz
GB          0
PC          1.00

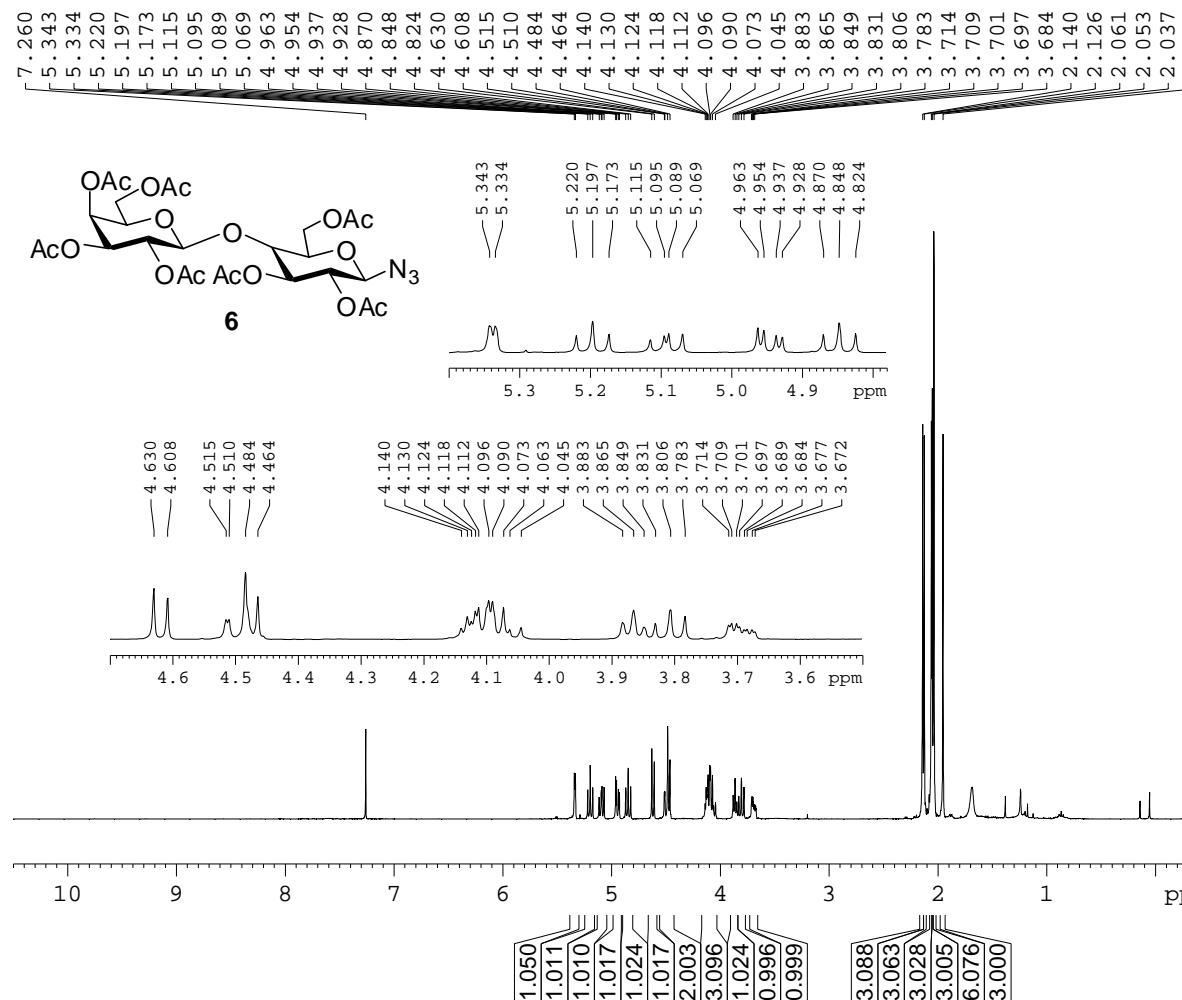
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```

ppm 1D NMR plot parameters
CX           20.00 cm
CY           10.50 cm
F1P          5.373 ppm
F1           2150.09 Hz

```





Current Data Parameters  
NAME Lact azide-pure  
EXPNO 1  
PROCNO 1

```

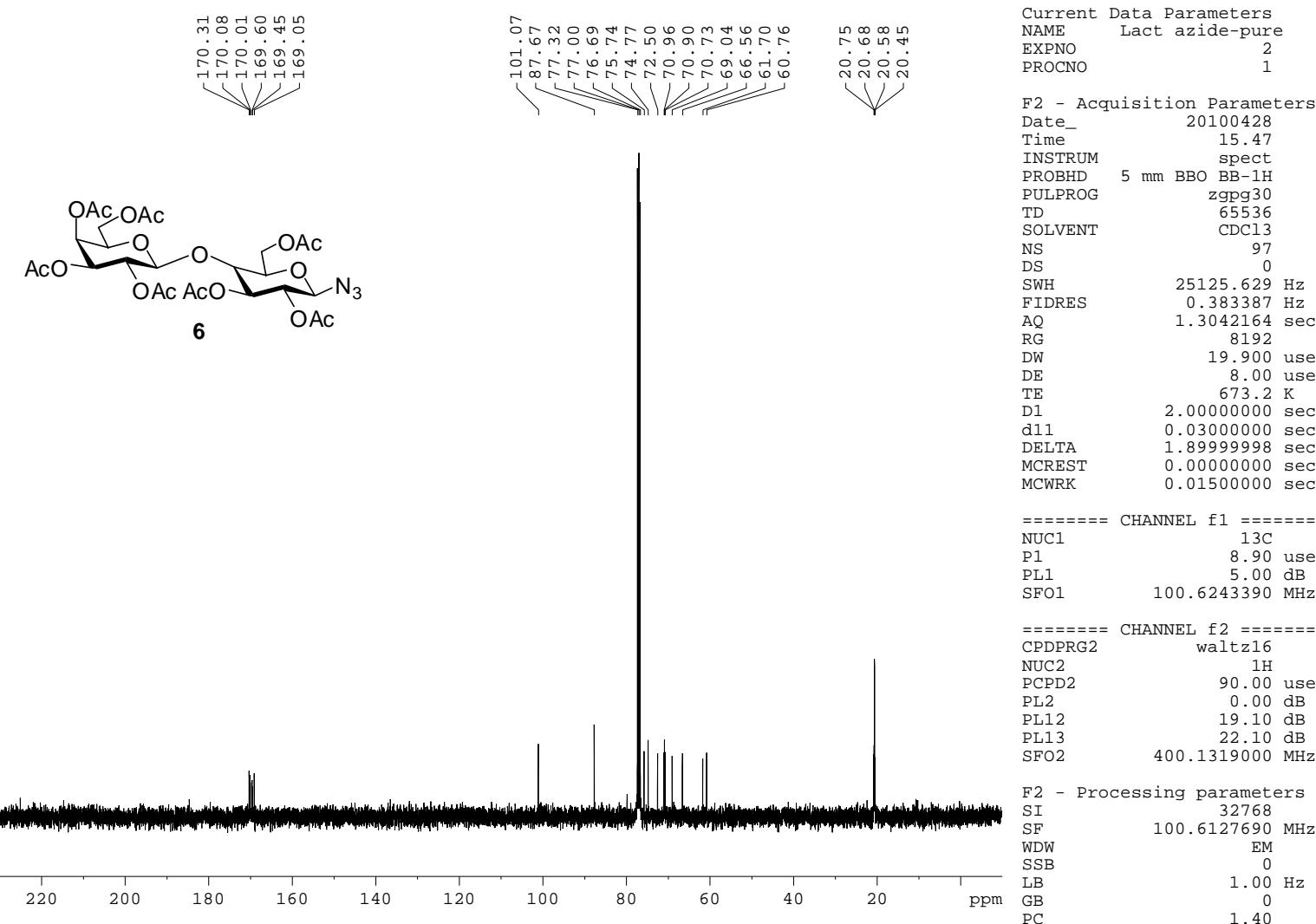
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INSTRUM       spect
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PULPROG       zg30
TD             16384
SOLVENT        CDC13
NS              16
DS               0
SWH             5995.204 Hz
FIDRES        0.365918 Hz
AQ            1.3664756 sec
RG              114
DW             83.400 usec
DE              8.00 usec
TE              673.2 K
D1      1.50000000 sec
MCREST        0.00000000 sec
MCWRK         0.01500000 sec

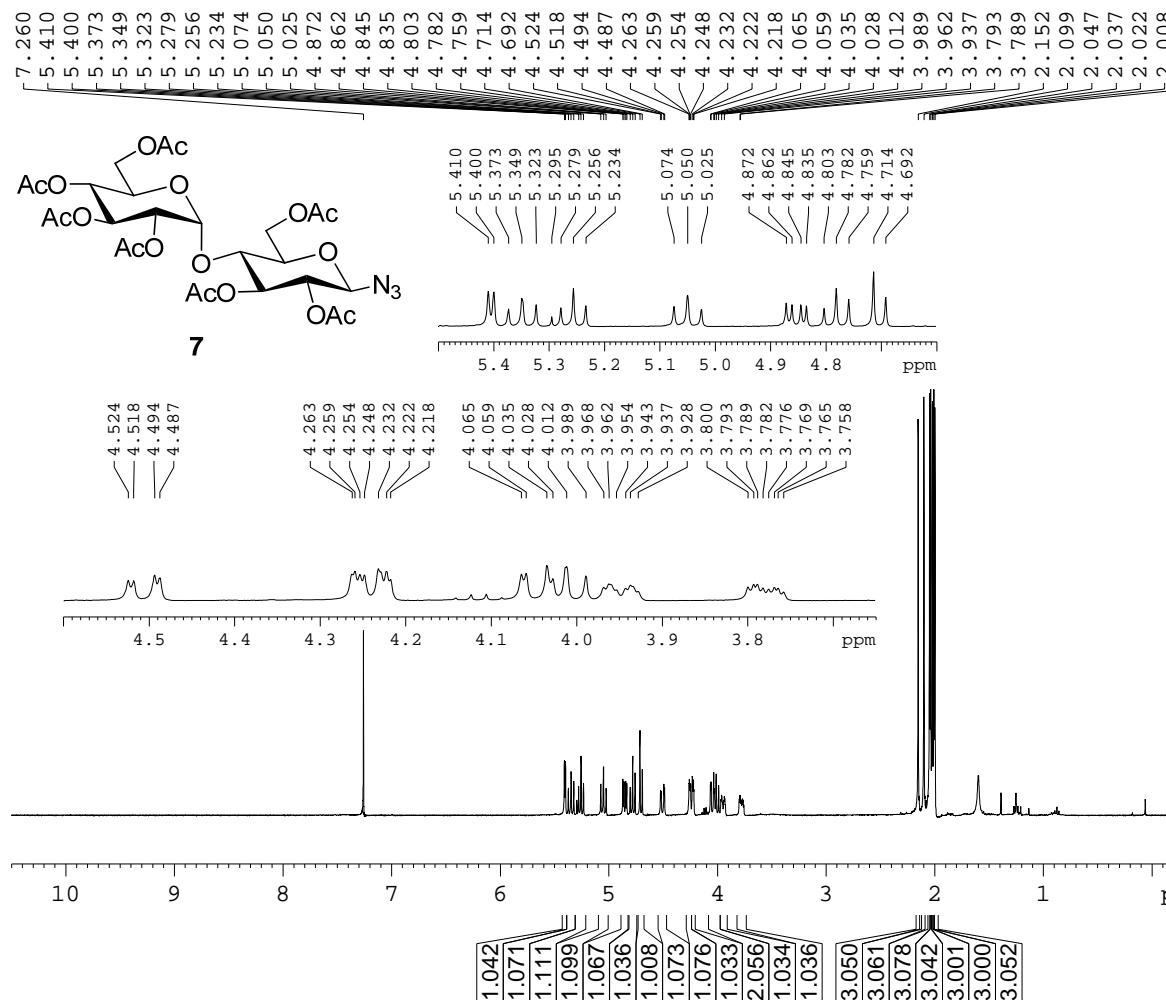
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===== CHANNEL f1 =====  
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P1 10.00 use  
PL1 0.00 dB  
SEQ1 400.1326008 MHz

F2 - Processing parameters  
SI 16384  
SF 400.1300091 MH  
WDW EM  
SSB 0  
LB 0.10 Hz  
GB 0  
PC 1.00

n 1D NMR plot parameters  
CX 20.00 cm  
CY 11.96 cm  
F1P 2.194 ppm  
F1 877.95 Hz





Current Data Parameters  
NAME Maltose azide-pure  
EXPNO 1  
PROCNO 1

```

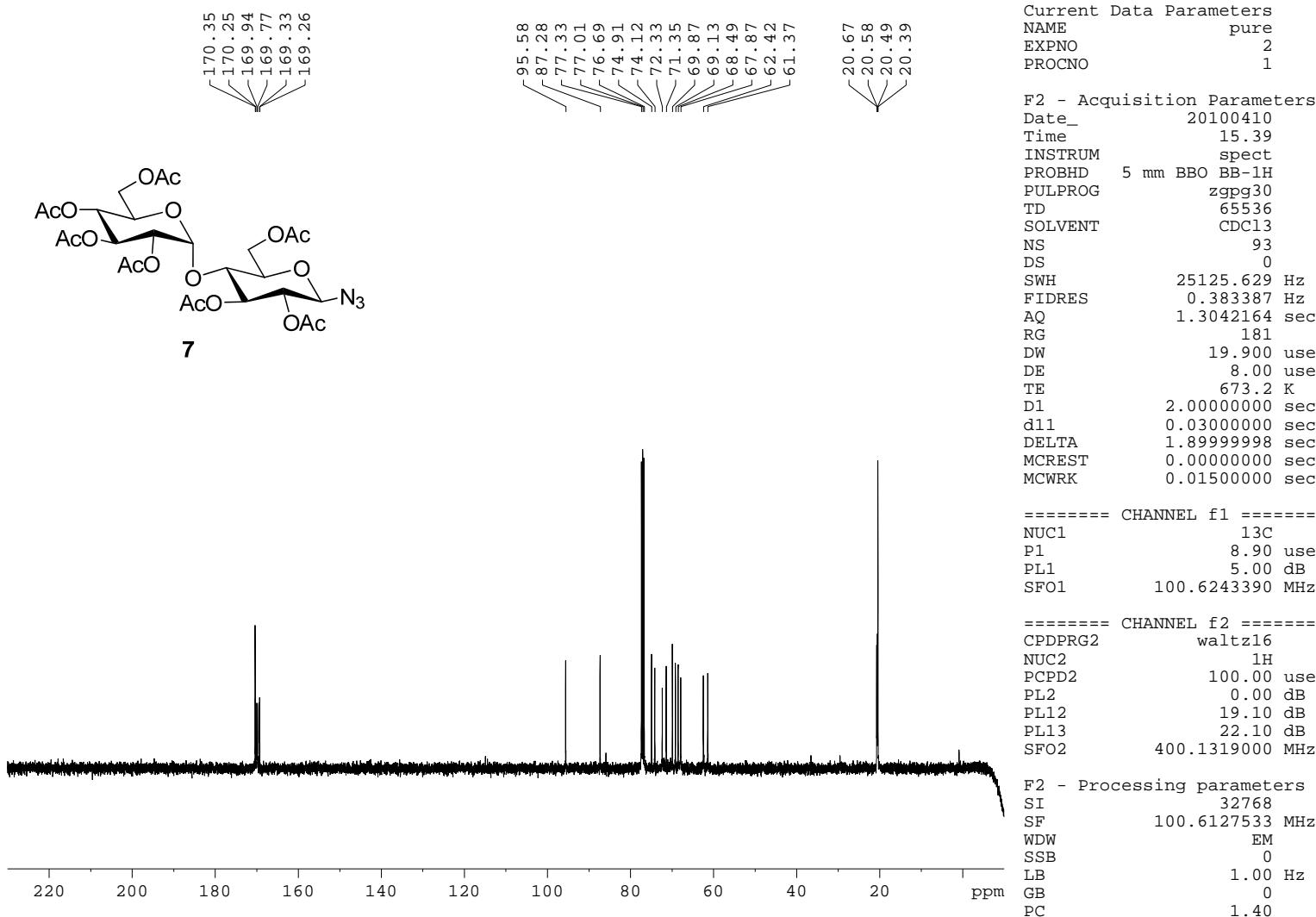
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INSTRUM       spect
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PULPROG      zg30
TD             16384
SOLVENT        CDC13
NS              16
DS               0
SWH            5995.204 Hz
FIDRES       0.365918 Hz
AQ            1.3664756 sec
RG             322.5
DW             83.400 usec
DE              6.50 usec
TE              297.8 K
D1          1.50000000 sec
MCREST        0.00000000 sec
MCWRK        0.01500000 sec

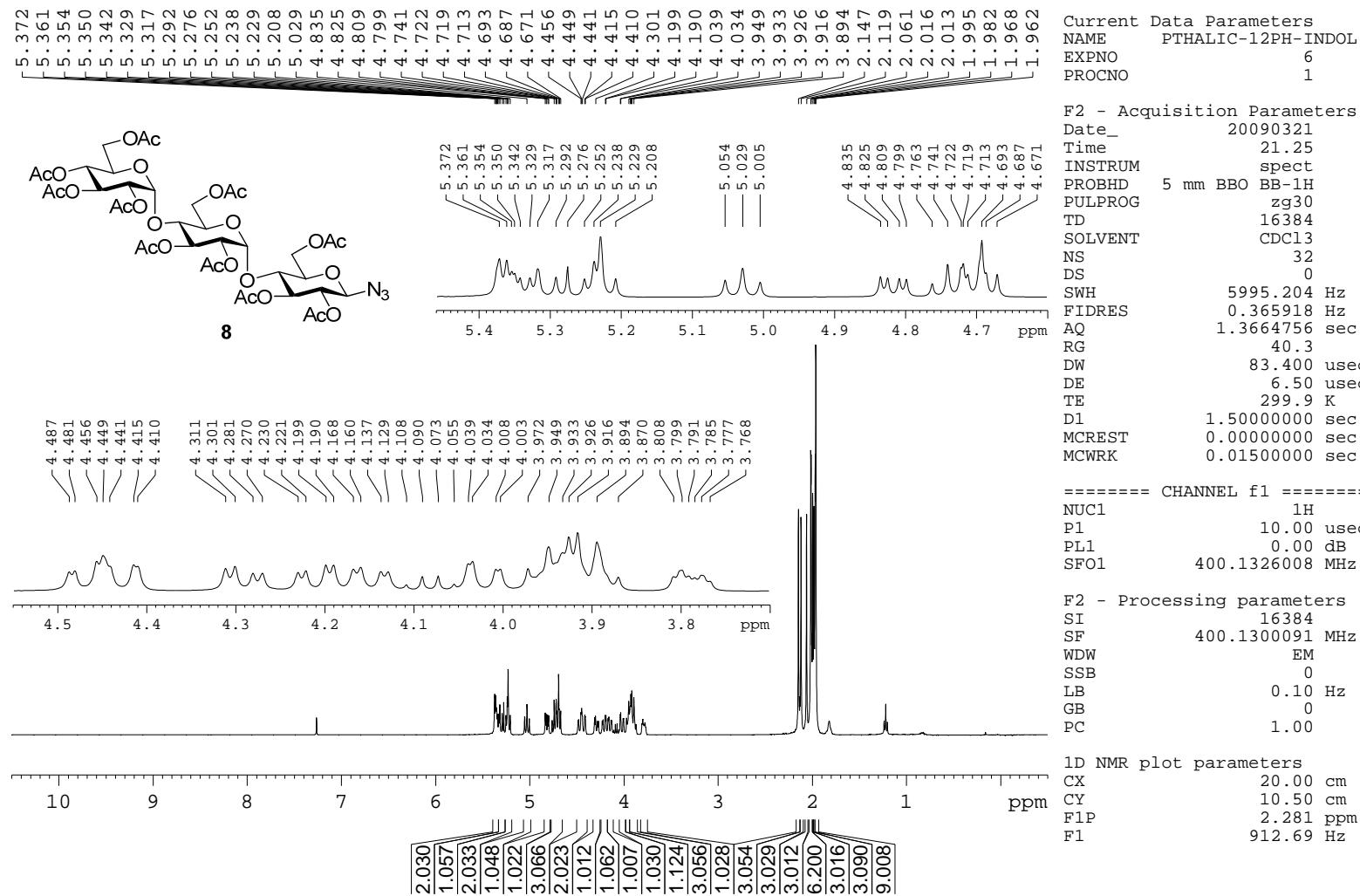
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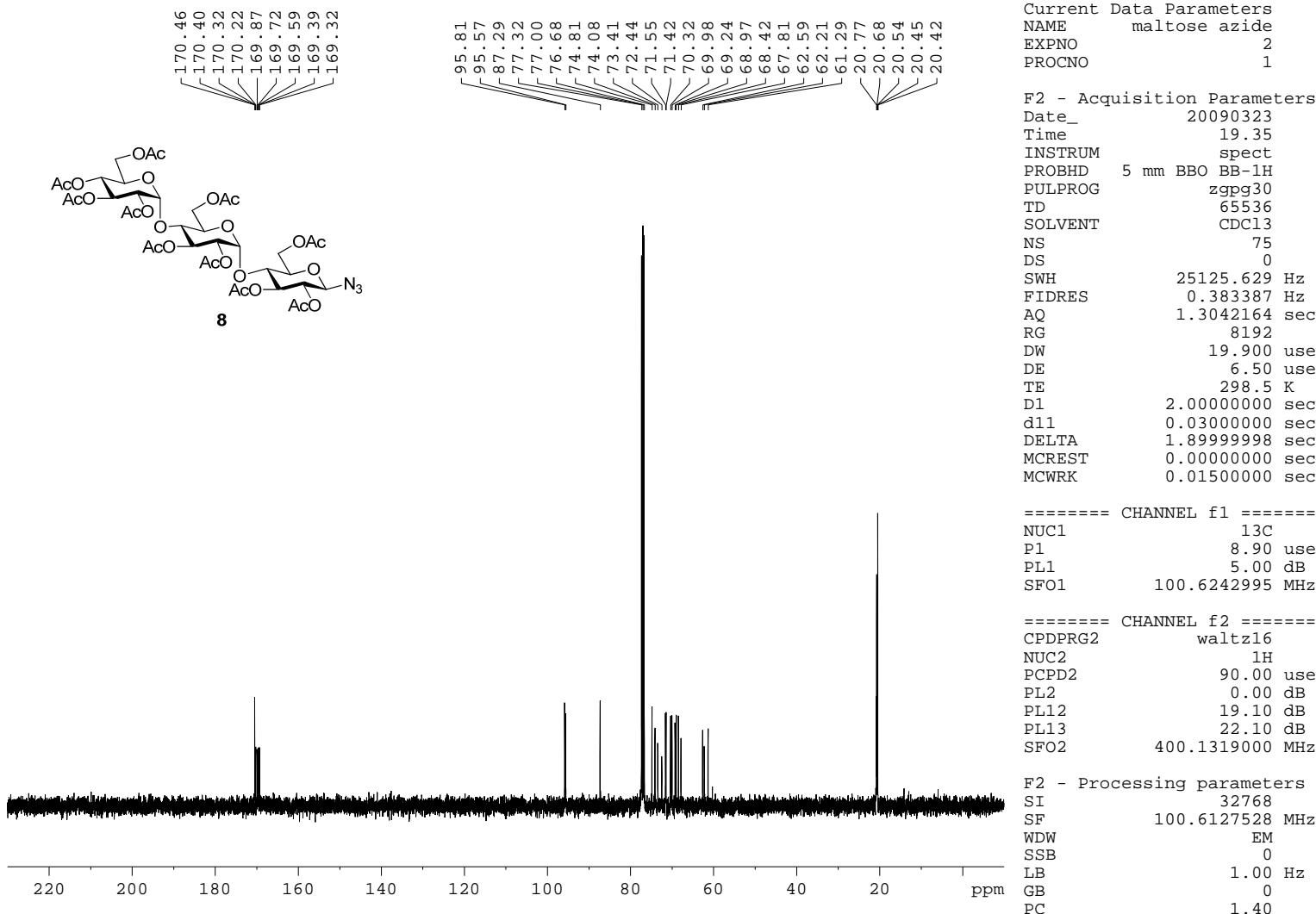
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PL1 3.00 dB  
SFO1 400.1326008 MHz

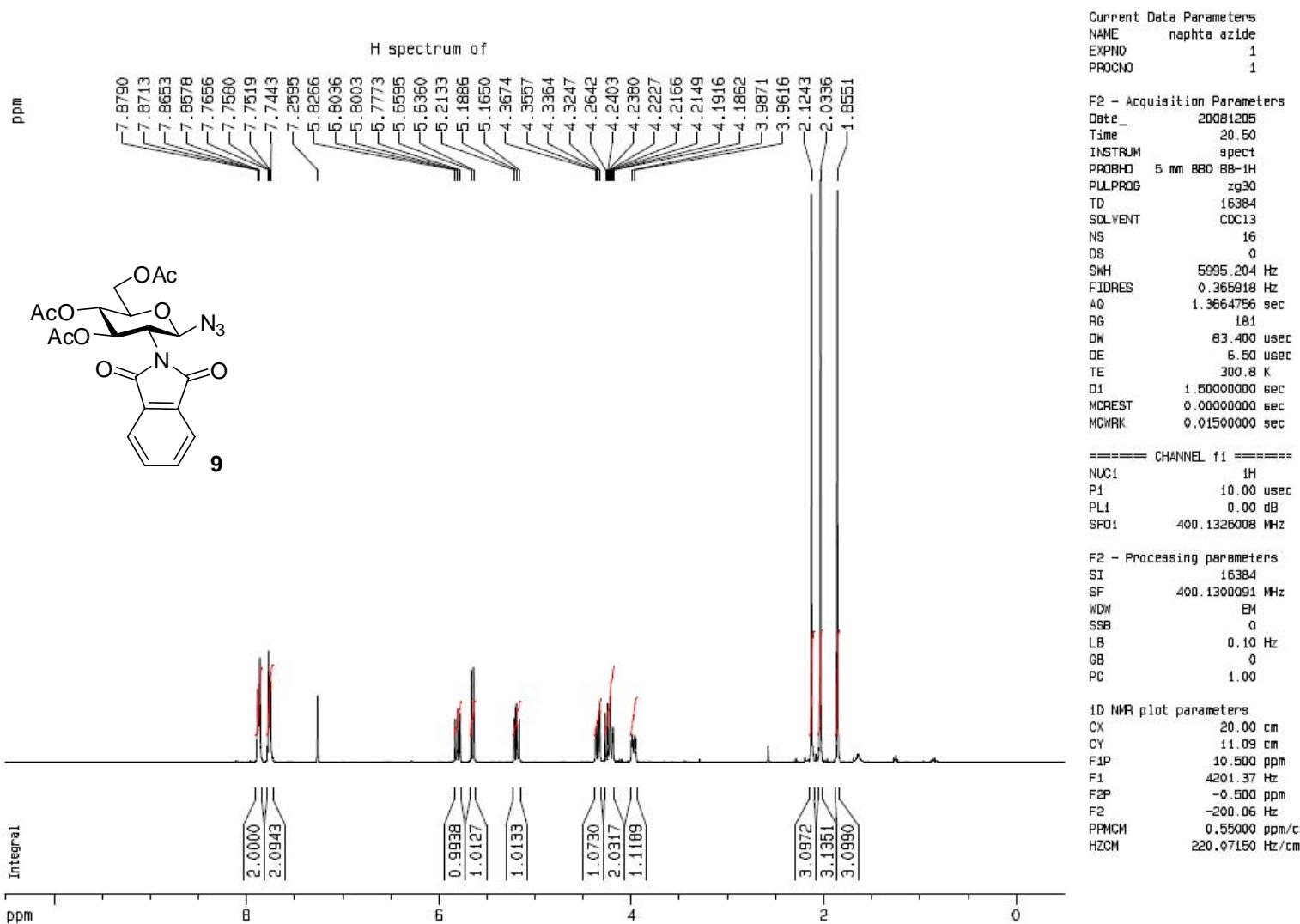
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SF 400.1300095 MHz  
WDW EM  
SSB 0  
LB 0.10 Hz  
GB 0  
PC 1.00

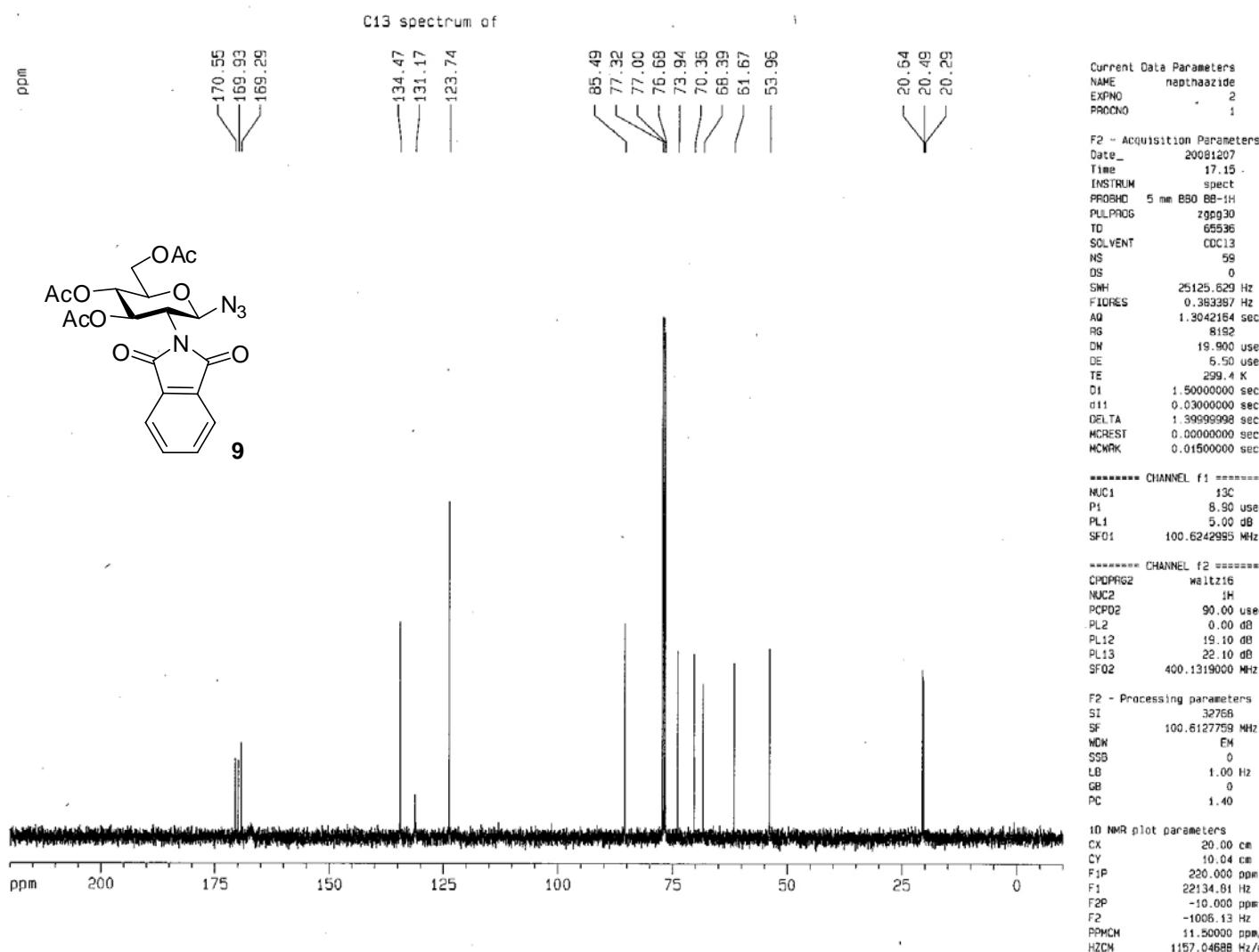
m 1D NMR plot parameters  
CX 20.00 cm  
CY 10.42 cm  
F1P 5.482 ppm  
F1 2193.37 Hz

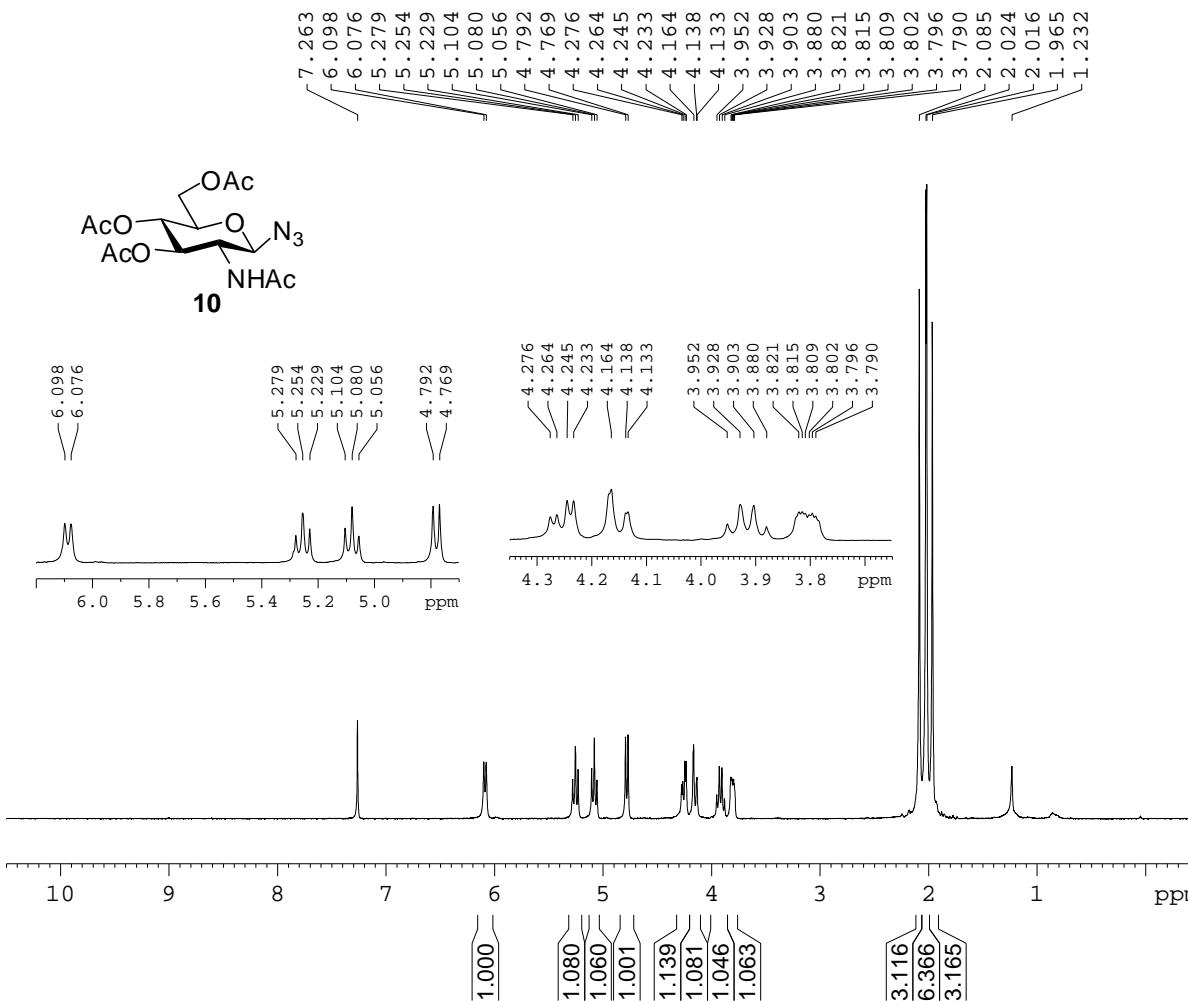










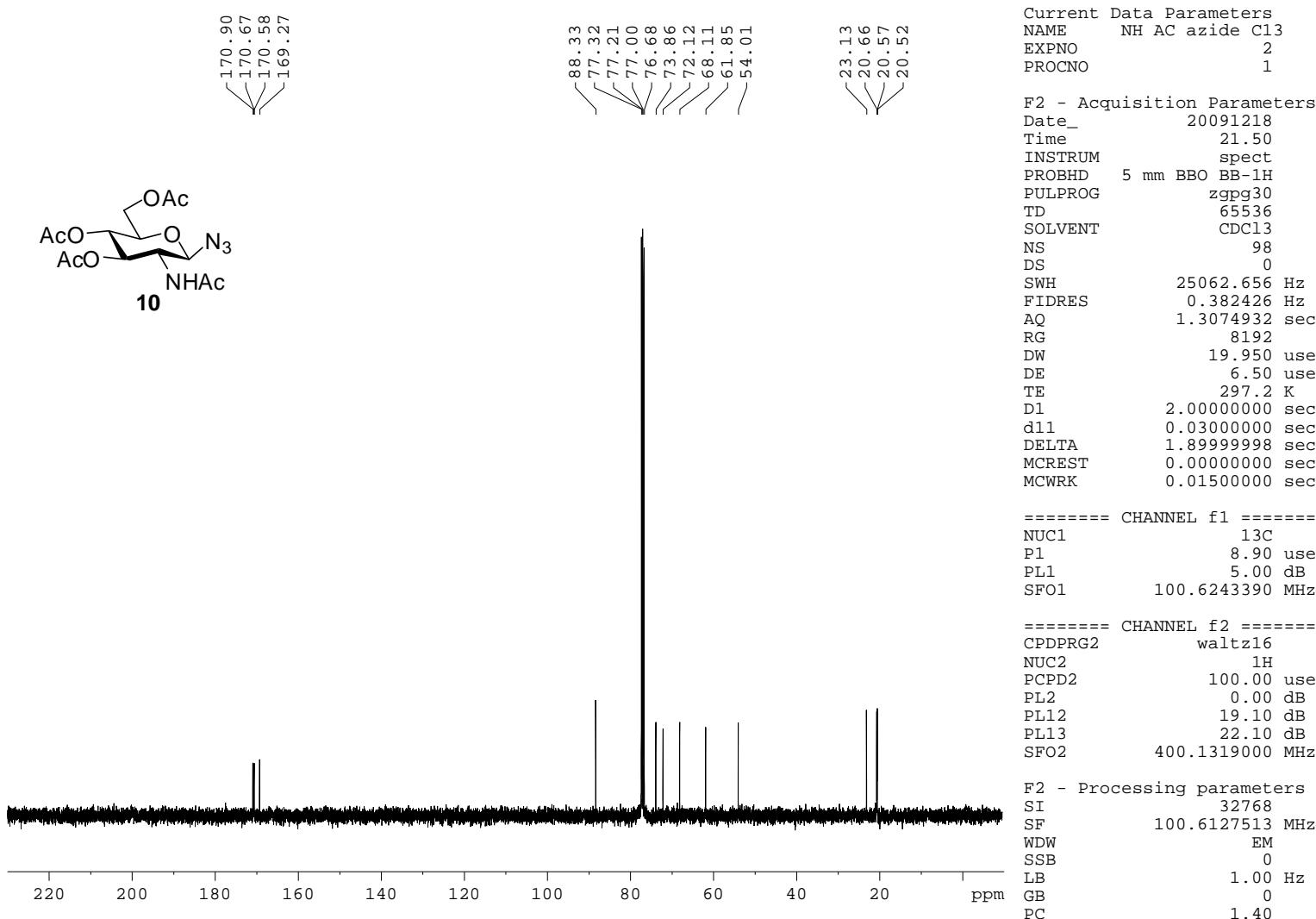


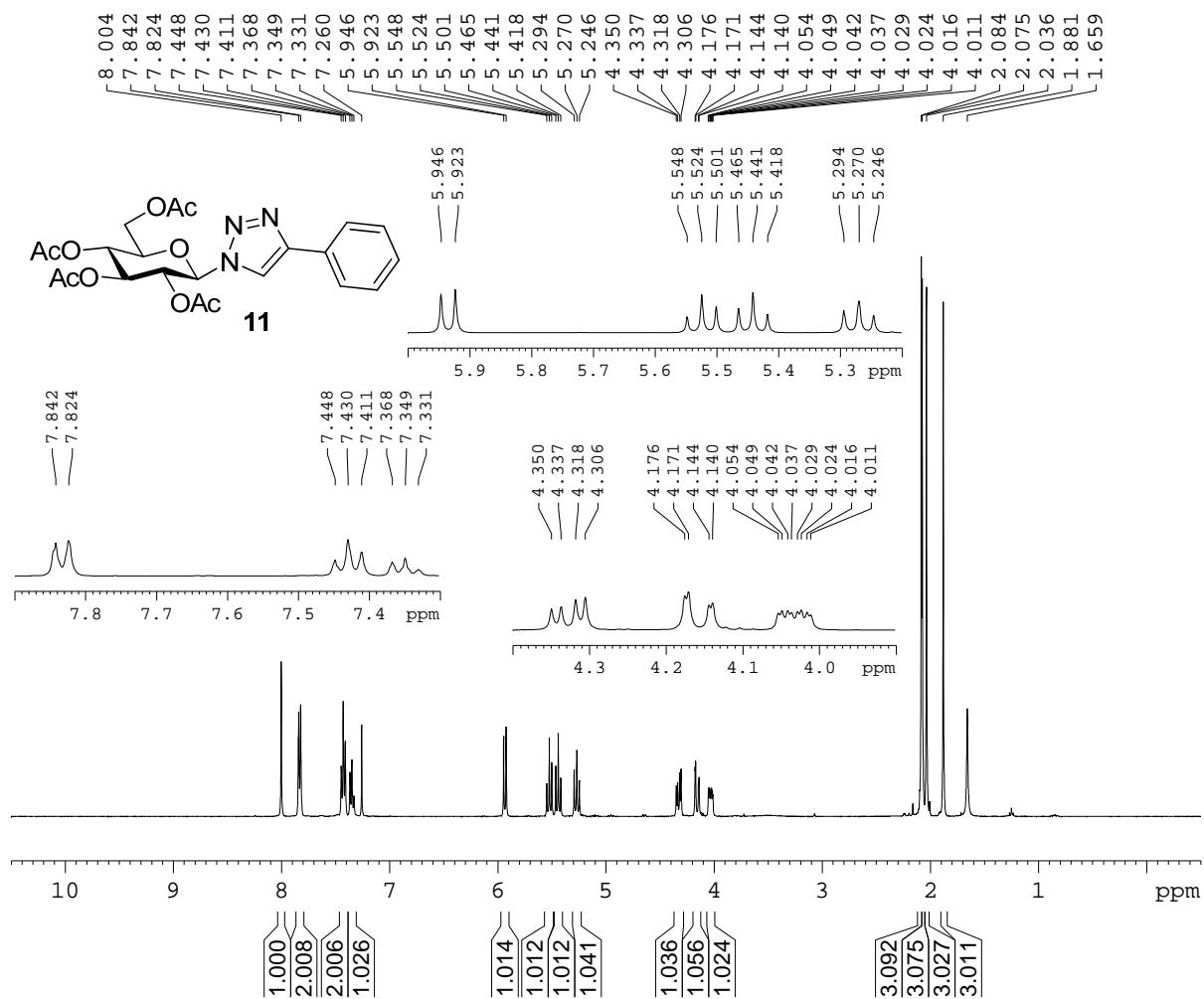
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EXPNO				1
PROCNO				1
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Time			21.49	
INSTRUM			spect	
PROBHD	5 mm	BBO	BB-1H	
PULPROG			zg30	
TD			16384	
SOLVENT			CDCl3	
NS			1	
DS			0	
SWH		5995.204	Hz	
FIDRES		0.365918	Hz	
AQ		1.3664756	sec	
RG		90.5		
DW		83.400	usec	
DE		6.50	usec	
TE		297.2	K	
D1		1.5000000	sec	
MCREST		0.0000000	sec	
MCWRK		0.0150000	sec	

```

===== CHANNEL f1 =====
NUC1          1H
P1           10.00 use
PL1          0.00 dB
SFO1        400.1326008 MHz
SI            16384
SF           400.1300074 MHz
WDW           EM
SSB             0
LB            0.10 Hz
GB             0
PC           1.00

```





Current Data Parameters  
NAME glucose triazole pu  
EXPNO 1  
PROCNO 1

## F2 - Acquisition Parameters

```

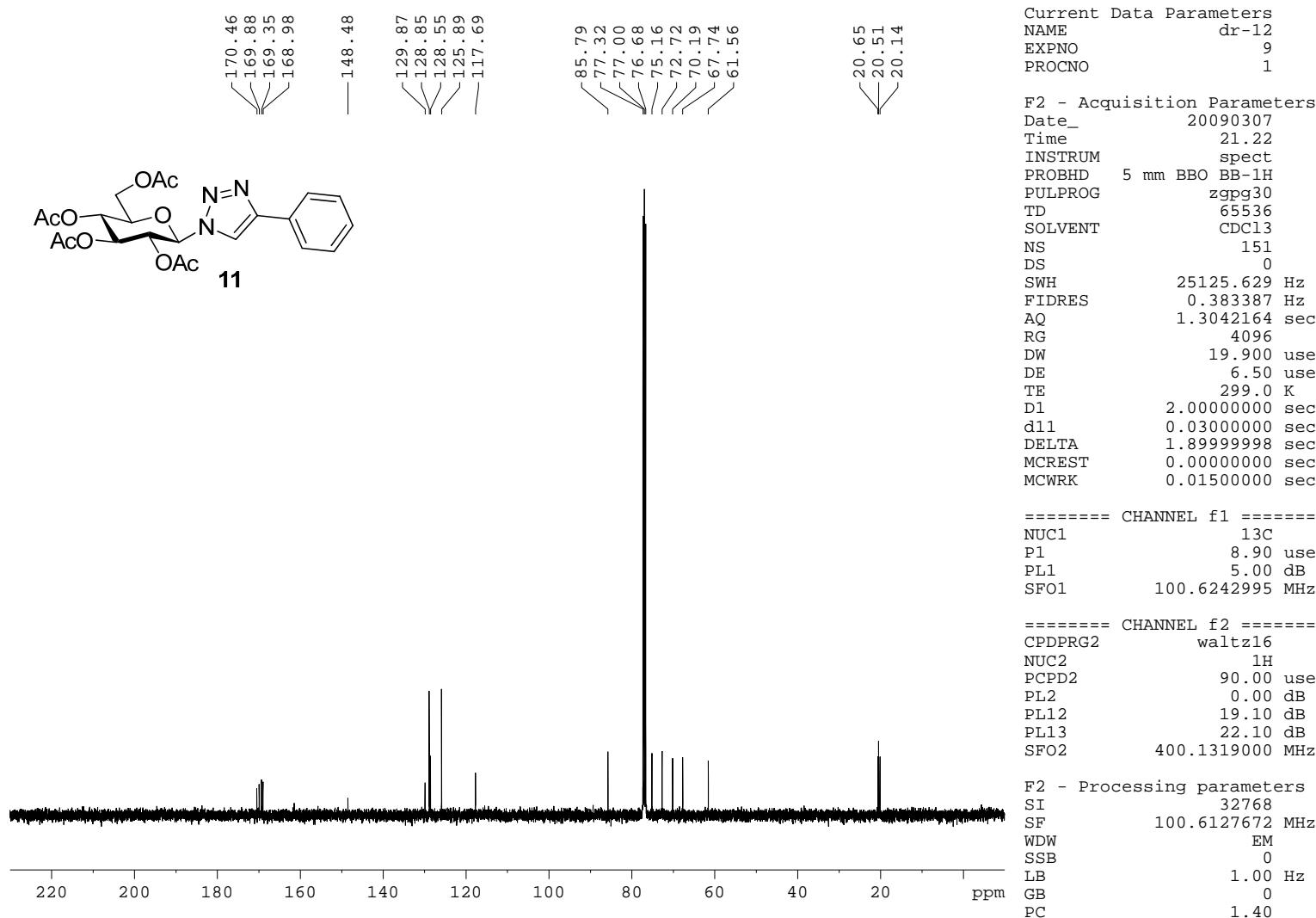
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Time       20.58
INSTRUM   spect
PROBHD    5 mm BBO BB-1H
PULPROG   zg30
TD        16384
SOLVENT   CDC13
NS         16
DS         0
SWH       5995.204 Hz
FIDRES   0.365918 Hz
AQ        1.3664756 sec
RG        161.3
DW        83.400 usec
DE        6.50  usec
TE        297.6 K
D1        1.50000000 sec
MCREST   0.00000000 sec
MCWRK    0.01500000 sec

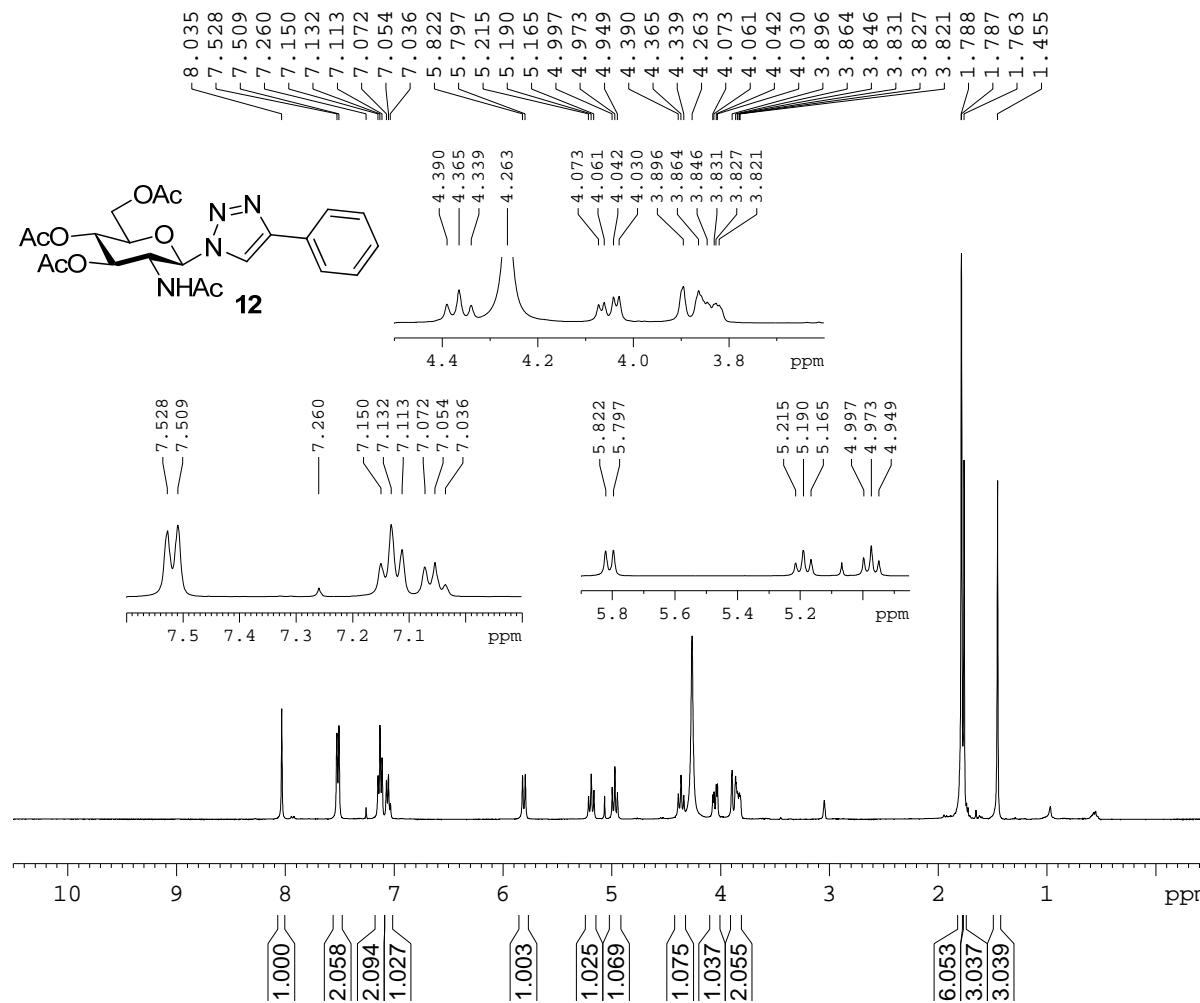
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===== CHANNEL f1 =====  
NUC1 1H  
P1 10.00 usec  
PL1 0.00 dB  
SFO1 400.1326008 MHz

F2 - Processing parameters  
SI 16384  
SF 400.1300095 MHz  
WDW EM  
SSB 0  
LB 0.10 Hz  
GB 0  
PC 1.00

n 1D NMR plot parameters  
CX 20.00 cm  
CY 10.50 cm  
F1P 2.155 ppm  
F1 862.43 Hz





Current Data Parameters

NAME NHAc  
EXPNO 1  
PROCNO 1

F2 - Acquisition Parameters

Date\_ 20090405  
Time 17.49  
INSTRUM spect  
PROBHD 5 mm BBO BB-1H  
PULPROG zg30  
TD 16384  
SOLVENT MeOD  
NS 16  
DS 0  
SWH 5995.204 Hz  
FIDRES 0.365918 Hz  
AQ 1.3664756 sec  
RG 57  
DW 83.400 usec  
DE 6.50 usec  
TE 298.0 K  
D1 1.5000000 sec  
MCREST 0.0000000 sec  
MCWRK 0.0150000 sec

===== CHANNEL f1 =====

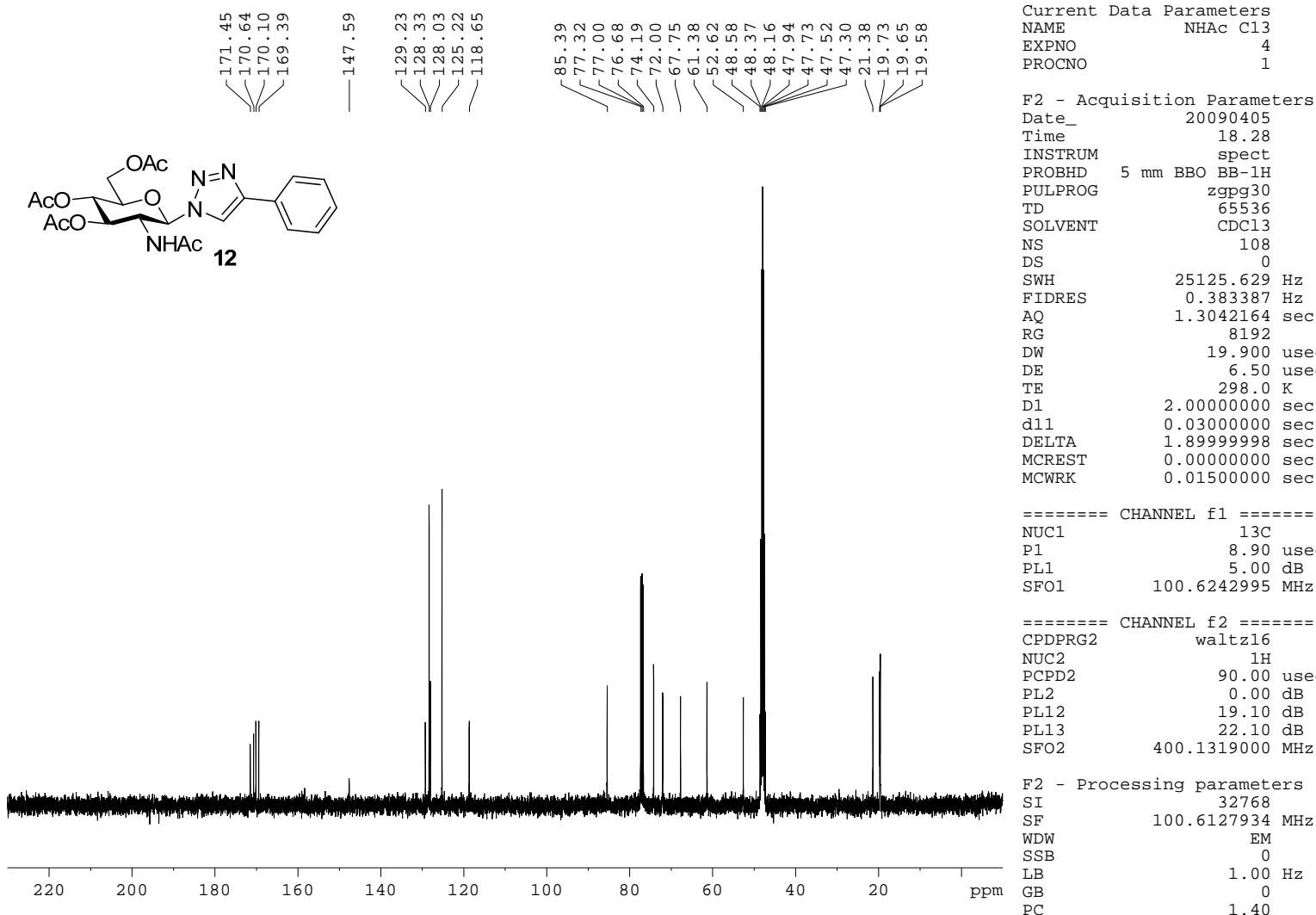
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PL1 0.00 dB  
SFO1 400.1326008 MHz

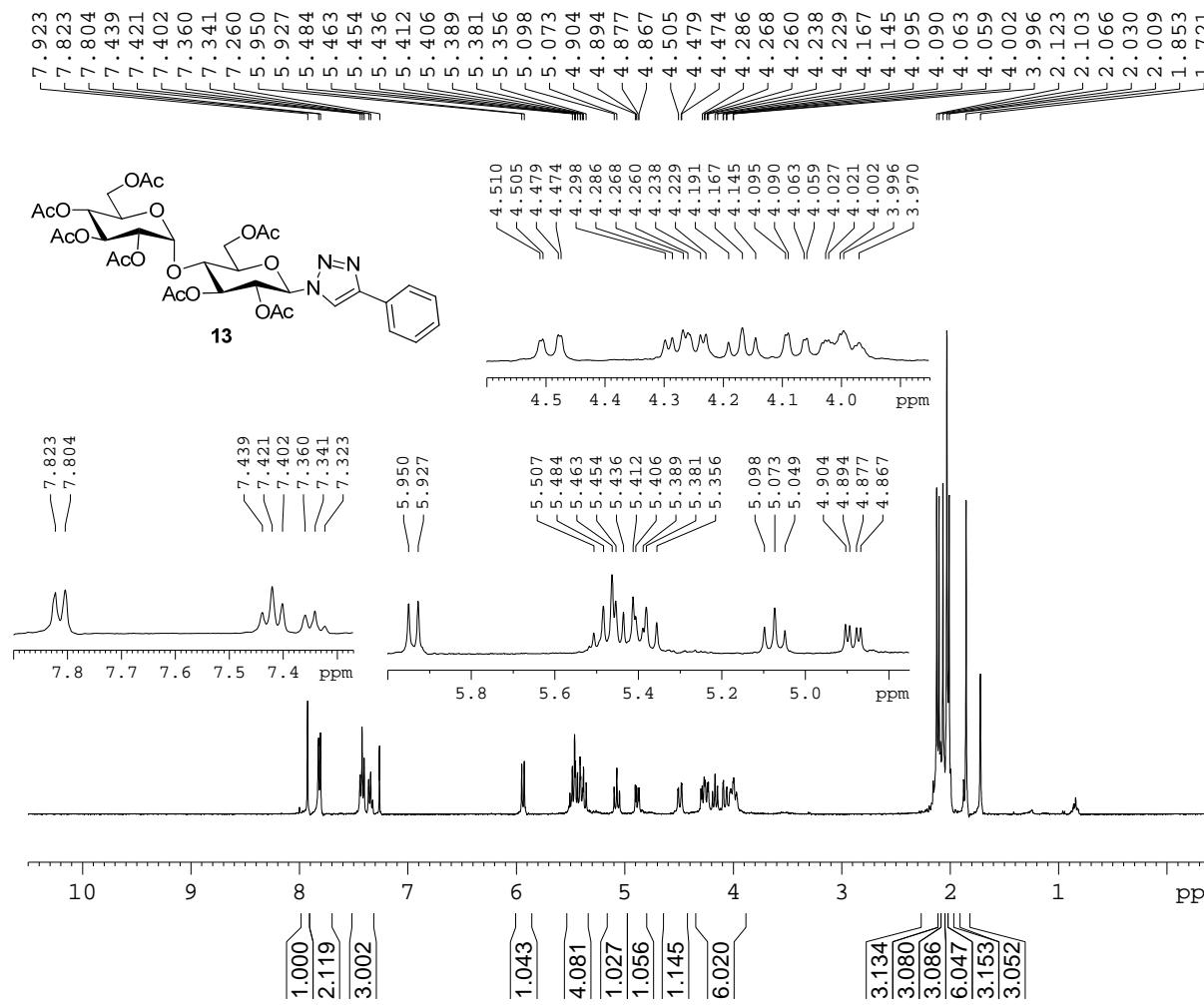
F2 - Processing parameters

SI 16384  
SF 400.1301123 MHz  
WDW EM  
SSB 0  
LB 0.10 Hz  
GB 0  
PC 1.00

1D NMR plot parameters

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CY 10.50 cm  
F1P 5.882 ppm  
F1 2353.48 Hz



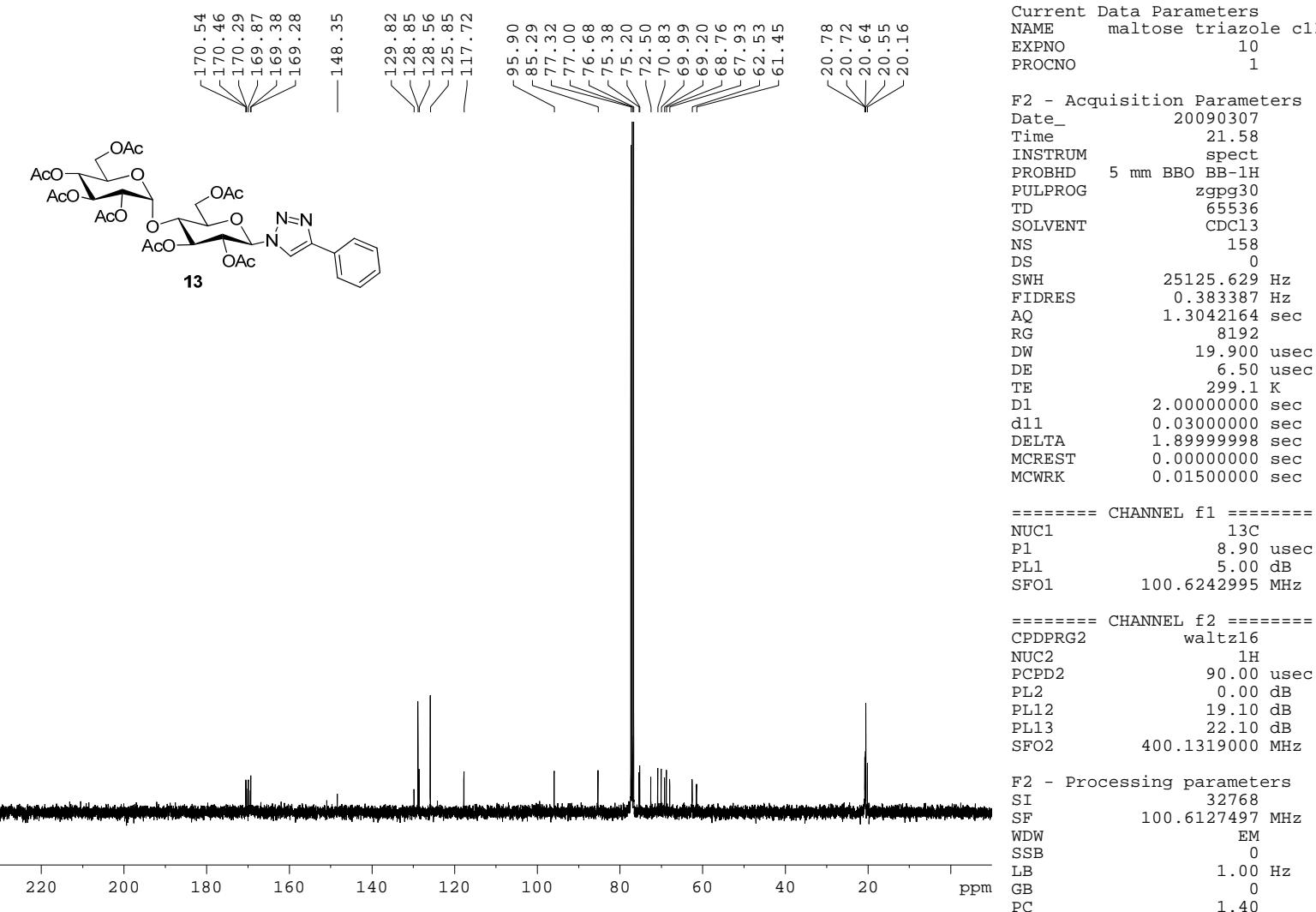


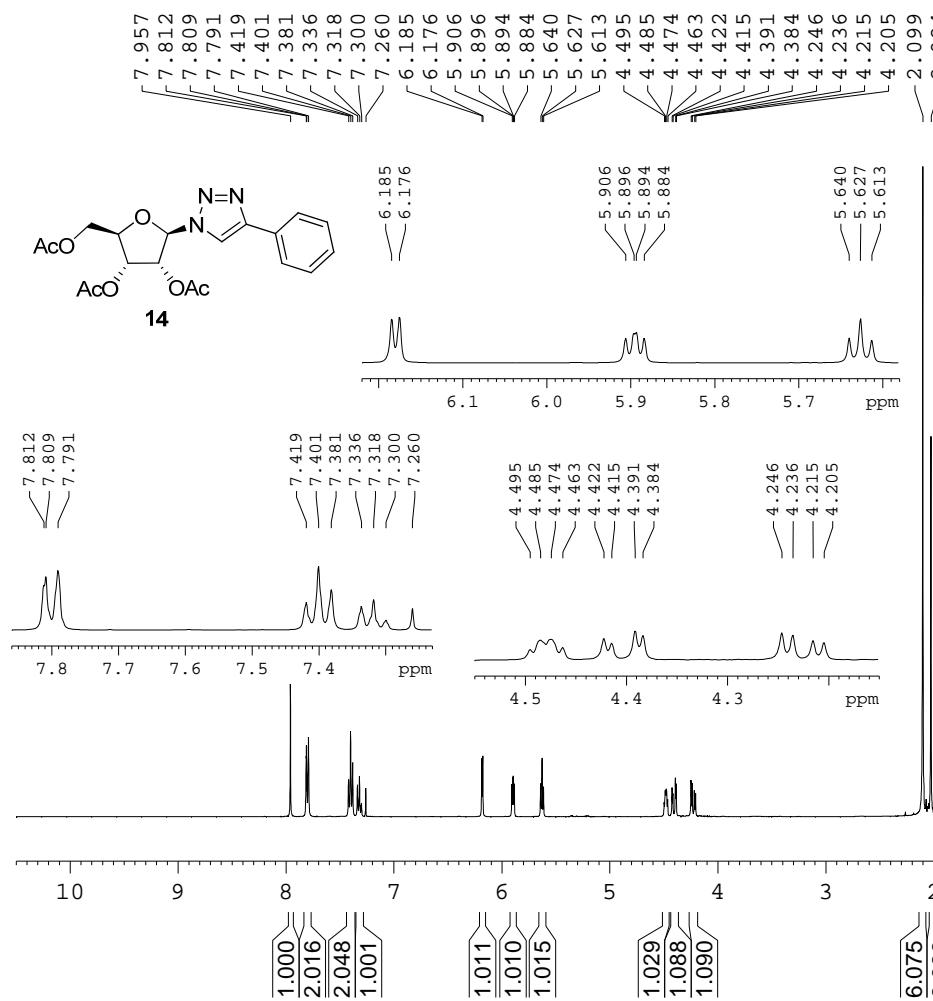
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NAME      maltose triazole cl
EXPNO          9
PROCNO          1
Date_   20090307
Time    21.56
INSTRUM   spect
PROBHD   5 mm BBO BB-1H
PULPROG   zg30
TD        16384
SOLVENT   CDCl3
NS           1
DS           0
SWH       5995.204 Hz
FIDRES   0.365918 Hz
AQ        1.3664756 sec
RG          114
DW        83.400 usec
DE         6.50  usec
TE        299.1 K
D1        1.50000000 sec
MCREST   0.00000000 sec
MCWRK    0.01500000 sec

=====
CHANNEL f1 ======
NUC1          1H
P1        10.00 usec
PL1          0.00 dB
SFO1    400.1326008 MHz
SI        16384
SF        400.1300089 MHz
WDW          EM
SSB            0
LB        0.10 Hz
GB            0
PC        1.00

```





Current Data Parameters  
NAME ribose triazol  
EXPNO 1  
PROCNO 1

```

F2 - Acquisition Parameters
Date_           20090314
Time            16.19
INSTRUM        spect
PROBHD         5 mm BBO BB-1H
PULPROG        zg30
TD              16384
SOLVENT         CDCl3
NS              16
DS              0
SWH             5995.204 Hz
FIDRES         0.365918 Hz
AQ              1.3664756 sec
RG              40.3
DW              83.400 usec
DE              6.50 usec
TE              299.9 K
D1              1.50000000 sec
MCREST         0.00000000 sec
MCWRK         0.01500000 sec

```

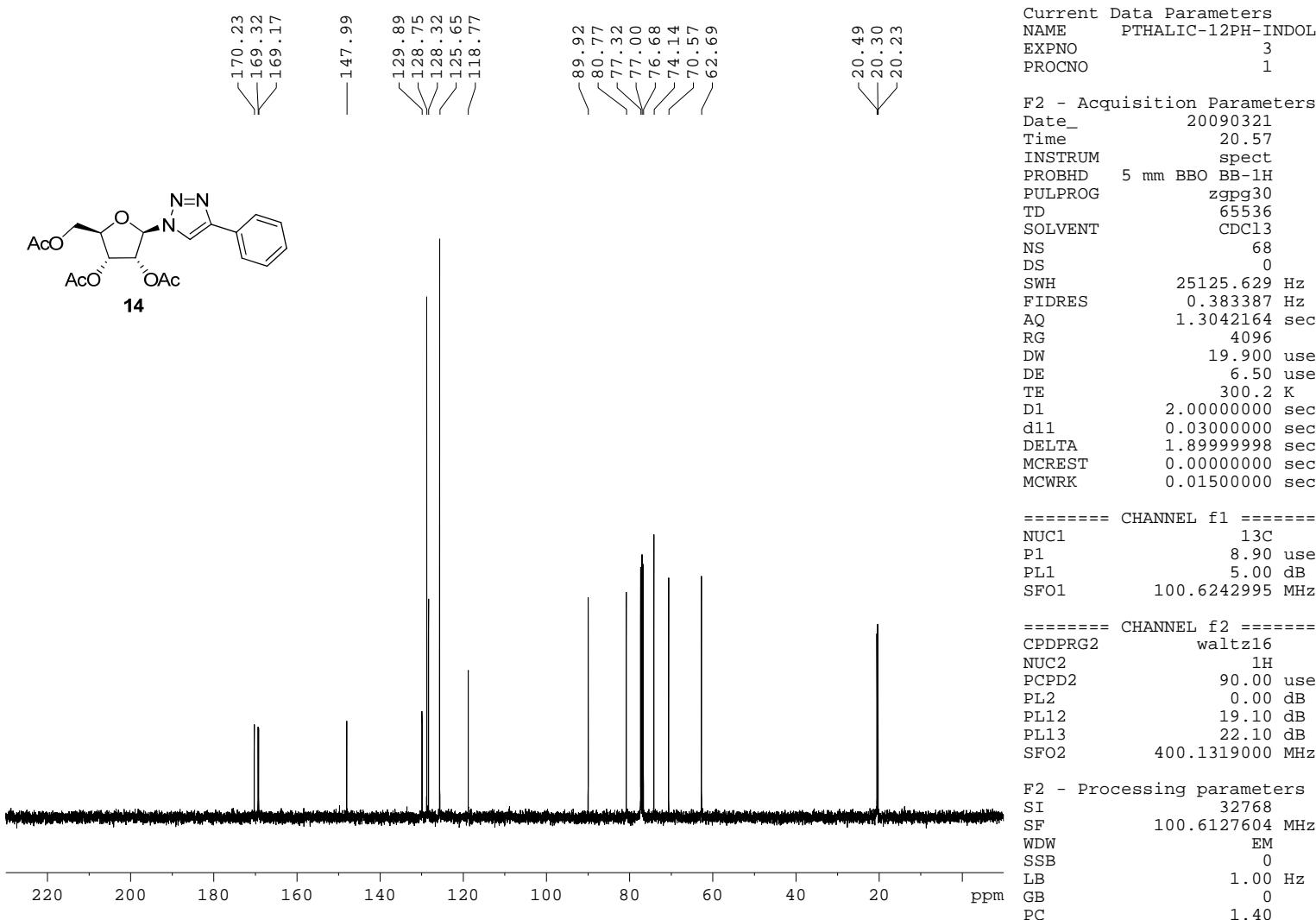
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P1 10.00 use  
PL1 0.00 dB  
SFO1 400.1326008 MHz

F2 - Processing parameters  
SI 16384  
SF 400.1300088 MH  
WDW EM  
SSB 0  
LB 0.10 Hz  
GB 0  
PC 1 00

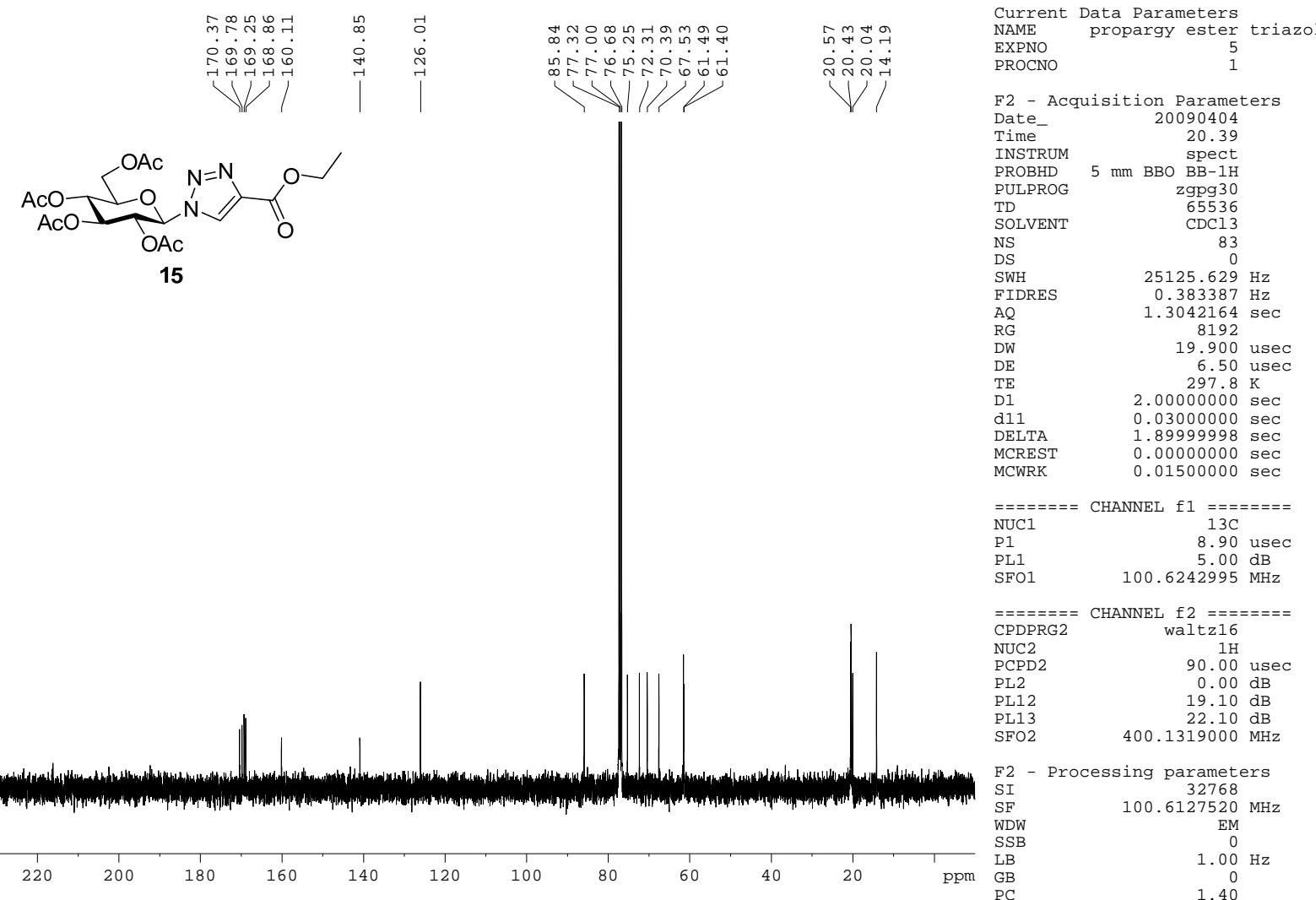
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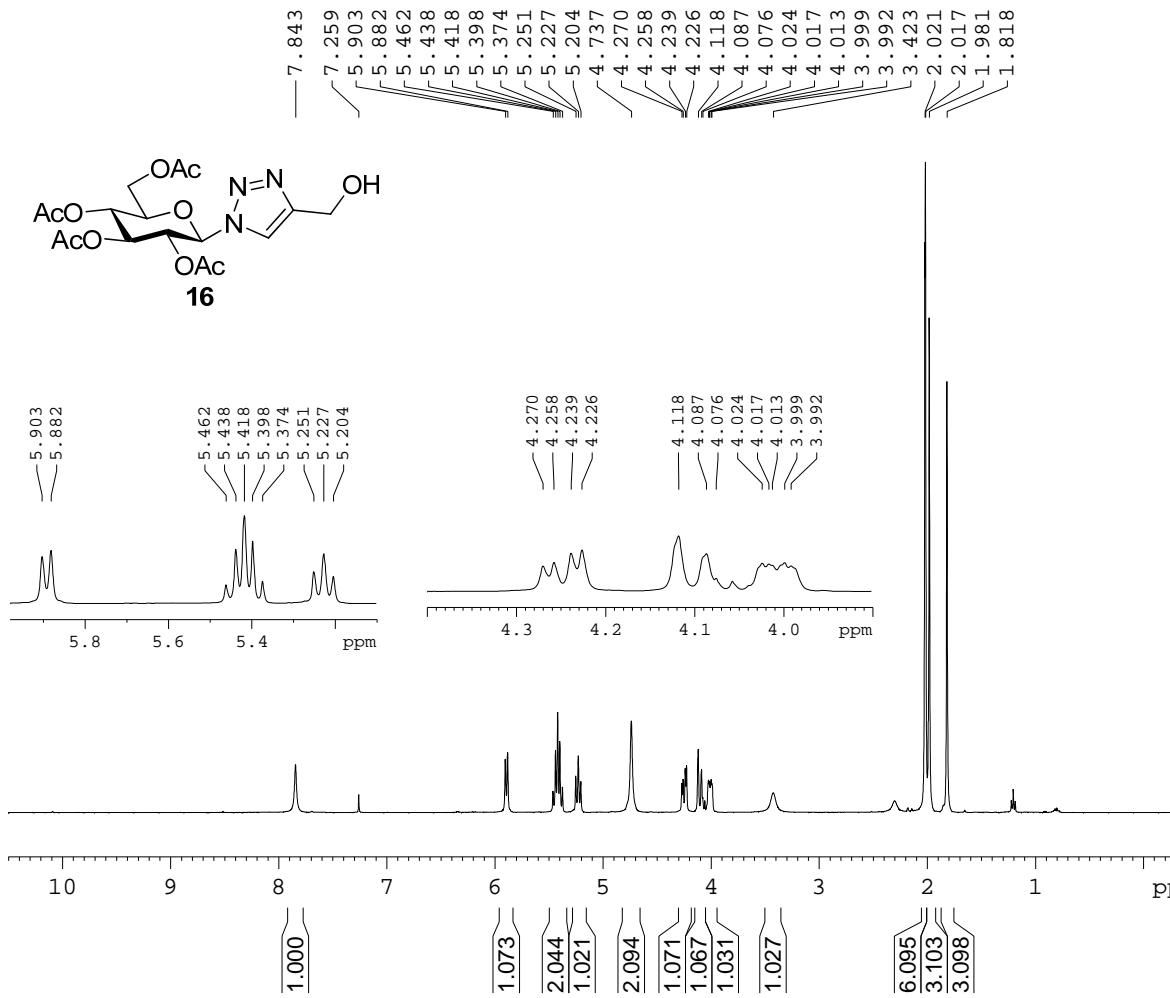
    1D NMR plot parameters
ppm   CX      20.00 cm
      CY      10.50 cm
      F1P     8.128 ppm
      F1      3252.12 Hz

```









Current Data Parameters  
NAME -propargy triazole  
EXPNO 1  
PROCNO 1

F2 - Acquisition Parameters  
Date\_ 20090326  
Time 20.50  
INSTRUM spect  
PROBHD 5 mm BBO BB-1H  
PULPROG zg30  
TD 16384  
SOLVENT CDCl<sub>3</sub>  
NS 16  
DS 0  
SWH 5995.204 Hz  
FIDRES 0.365918 Hz  
AQ 1.3664756 sec  
RG 45.3  
DW 83.400 usec  
DE 6.50 usec  
TE 298.7 K  
D1 1.5000000 sec  
MCREST 0.0000000 sec  
MCWRK 0.0150000 sec

===== CHANNEL f1 =====  
NUC1 1H  
P1 10.00 usec  
PL1 0.00 dB  
SFO1 400.1326008 MHz

F2 - Processing parameters  
SI 16384  
SF 400.1300095 MHz  
WDW EM  
SSB 0  
LB 0.10 Hz  
GB 0  
PC 1.00

1D NMR plot parameters  
CX 20.00 cm  
CY 10.50 cm  
F1P 6.065 ppm  
F1 2426.88 Hz

