Supporting Information for

C–N Bond Formation via Ligand-Induced Nucleophilicity at a Coordinated Triamidoamine Ligand

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General Considerations

All reactions were performed under a nitrogen atmosphere with dry, oxygen-free solvents using an M. Braun glovebox. Celite-454 was heated to a temperature above 180 °C under dynamic vacuum for at least 8 h. Benzene- d_6 was purchased from Cambridge Isotope Laboratory, degassed, and dried over NaK alloy. Elemental analyses were performed on an Elementar microCube. NMR spectra were recorded with either a Bruker ARX or Varian Inova 500 MHz spectrometer in bezene- d_6 and are reported with reference to residual solvent resonances ($\delta = 7.16$ and 128.0) unless otherwise noted. A Varian Inova 500 spectrometer was used to record HMQC spectra with a standard pulse sequence. Infrared spectra were collected on a Bruker Alpha FT-IR with an ATR plate at a resolution of 1 cm⁻¹. Complexes 1a, $^1 1b$, $^1 3$, 2 and 5^2 were prepared according to reported procedures. All other chemicals were obtained from commercial suppliers and dried or purified by conventional means.

 then warmed sufficiently to redissolve solids, and cooled to $-30 \,^{\circ}$ C yielding 73 mg (0.113 mmol, 65% relative to carbodiimide) of colorless crystals. ¹H (500.1 MHz): 4.15 (s, N*H*, 1 H), 3.95 (septet, C*H*, 1 H), 3.39 (m, C*H*₂, 2 H), 3.22 (m, C*H*, 1 H), 3.17 (m, C*H*₂, 2 H), 3.00 (m, C*H*₂, 2 H), 2.45 (m, C*H*₂, 2 H), 2.34 (m, C*H*₂, 4 H), 1.60 (s, C(C*H*₃)₃, 9 H), 1.45 (d, CH(C*H*₃)₂, *J* = 10 Hz, 3 H), 1.40 (d, CH(C*H*₃)₂, *J* = 5 Hz, 3 H), 1.07 (d, CH(C*H*₃)₂, *J* = 5 Hz, 3 H), 1.03 (d, CH(C*H*₃)₂, *J* = 5 Hz, 3 H), 0.42 (s, C*H*₃, 9 H), 0.38 (s, C*H*₃, 9 H), 0.15 (s, C*H*₃, 9 H) 13 C{¹H} (125.8 MHz): 170.8 (s, C=N), 59.1 (s, CH₂), 57.2 (s, CH₂), 54.9 (s, CH₃), 26.2 (s, CH₃), 25.3 (s, CH₃), 23.52 (s, CH₃), 23.1 (s, CH₃), 2.7 (s, CH₃), 2.6 (s, CH₃), 2.0 (s, CH₃). IR: 3209 s (v_{NH}), 2951 m, 2506 s, 1658 s (v_{CN}), 1587 s, 1488 m, 1436 m, 1382 w, 1357 w, 1239 m, 1200 m, 1153 w, 1116 w, 1065 w, 1033 w, 1000 w, 972 m, 940 m, 906 m, 827 s, 783 s, 743 s, 680 m, 582 m, 556 w, 524 w, 442 w, 423 w. Anal. Calcd. for C₂₆H₆₃N₇Si₃Zr: C, 48.09; H, 9.78; N, 15.10. Found: C, 47.72; H, 9.65; N, 14.97.

[N,N,N,N,N,N-N(CH₂CH₂NSiMe₃)₂(CH₂CH₂NC(NⁱPrSiMe₃)(NⁱPr))]ZrNHPh

(2b). A 2 mL benzene solution of *N*,*N'*-diisopropylcarbodiimide (46 mg, 0.36 mmol) was added drop-wise to a 5 mL benzene solution of **1b** (200 mg, 0.37 mmol) and stirred 1 h. The resulting solution was filtered through Celite and lyophilized. The powder was dissolved in minimal pentane, and the solution was filtered then concentrated before cooling to -30 °C to afford 237 mg (0.35 mmol, 97%) of **2b** as a colorless microcrystalline powder. ¹H (500.1 MHz): 7.33 (t, C₆H₅, *J* = 5 Hz, 2 H), 6.95 (d, C₆H₅, *J* = 5 Hz, 2 H), 6.78 (t, C₆H₅, *J* = 5 Hz, 1 H), 6.35 (s, NH, 1 H) 3.94 (m, CH₂, 2 H), 3.48 (m, CH, 2 H) 3.27 (m, CH, 2H), 3.15 (m, CH₂, 2 H), 3.05 (m, CH₂, 2 H), 2.62 (m, CH₂, 2

H), 2.38 (m, *CH*₂, 2 H), 2.31 (m, *CH*, 2 H), 1.63 (d, *CH*(*CH*₃)₂, *J* = 5 Hz, 3 H), 1.55 (d, *CH*(*CH*₃)₂, *J* = 5 Hz, 3 H), 1.02 (d, *CH*(*CH*₃)₂, *J* = 5 Hz, 3 H), 0.98 (d, *CH*(*CH*₃)₂, *J* = 5 Hz, 3 H), 0.49 (s, *CH*₃, 9 H), 0.46 (s, *CH*₃, 9 H), 0.12 (s, *CH*₃, 9 H) 13 C{¹H} (125.8 MHz): 171.7 (s, *C*=N),128.51 (s, *CH*), 118.3 (s, *CH*), 115.7 (s, *CH*), 56.3 (s, *CH*₂), 56.0 (s, *CH*₂), 54.1 (s, *CH*), 49.0 (s, *CH*₂), 48.9 (s, *CH*₂), 48.2 (s, *CH*), 44.1(s, *CH*₂), 44.0 (s, *CH*₂), 26.0 (s, *CH*₃), 25.0 (s, *CH*₃), 24.6 (s, *CH*₃), 23.1 (s, *CH*₃), 2.2 (s, *CH*₃), 2.0 (s, *CH*₃), 1.3 (s, *CH*₃). IR: 3230 s (v_{NH}), 2948 m, 2510 s, 1646 s (v_{CN}), 1590 s, 1487 m, 1457 m, 1439 m, 1381 w, 1365 w, 1333 w, 1279 m, 1240 m, 1210 m, 1152 w, 1128 w, 1067 m, 1029 m, 1002 m, 938 s, 901 s, 828 s, 785 s, 748 s, 691s, 678 m, 659 m, 585 m, 573 m, 520 m, 481 m, 440 m. Anal. Calcd. for $C_{28}H_{59}N_7Si_3Zr$: C, 50.25; H, 8.89; N, 14.65. Found: C, 50.16; H, 9.01; N, 14.52.

CH), 44.0(s, CH₂), 43.9 (s, CH₂), 25.8 (s, CH₃), 24.9 (s, CH₃), 24.5 (s, CH₃), 23.07 (s, CH₃), 2.2 (s, CH₃), 1.9 (s, CH₃), 1.3 (s, CH₃). IR: 2952 w, 2896 w, 2859 w, 1621 w (v_{CN}), 1490 m, 1472 m, 1448 m, 1421 m, 1327 w, 1243 m, 1219 m, 1150 w, 1116 w, 1046 w, 972 m, 916 s, 828 s, 746 s, 678 m, 616 m, 546 w, 404 w. Anal. Calcd. for C₂₂H₅₃ClN₆Si₃Zr: C, 43.13; H, 8.72; N, 13.72. Found: C, 42.80; H, 8.65; N, 13.47.

(6). A 1.5 mL benzene solution of *N*,*N'*-diisopropylcarbodiimide (30 mg, 0.24 mmol) was added to 2 mL benzene solution of **5** (105 mg, 0.23 mmol) at -30 °C, and the resulting solution was stirred for 30 min while gradually warming to ambient temperature. The solution was lyophilized yielding 85 mg (0.14 mmol, 63%) of product. Colorless crystals were obtained from a cooled ether solution. ¹H (500.1 MHz): 3.85 (m, *CH*(CH₃)₂, 1H), 3.46 (m, *CH*₂, 2H), 3.27 (m, *CH*₂, 2H), 3.21 (m, *CH*(CH₃)₂, 1H), 3.1 (m, *CH*₂, 1H), 3.02 (m, *CH*₂, 1H), 2.38 (m, *CH*₂, 2H), 2.25 (m, *CH*₂, 2H), 2.19 (m, *CH*₂, 2H), 1.41 (d, *CH*(CH₃)₂, *J* = 6.5 Hz, 3H), 1.37 (d, *CH*(CH₃)₂, *J* = 6.5 Hz, 3H), 1.08 (d, *CH*(CH₃)₂, *J* = 6.5 Hz, 3H), 1.03 (d, *CH*(CH₃)₂, *J* = 6.5 Hz, 3H), 0.64 (s, *CH*₃, 3H), 0.39 (s, *CH*₃, 3H), 0.37 (s, *CH*₃, 3H), 0.15 (s, *CH*₃, 3H). ¹³C{¹H} (125.8 MHz): 171 (s, *C*=N), 59.1 (s, *CH*₂), 56.8 (s, *CH*₂), 49.5 (s, *CH*), 47.7 (s, *CH*₂), 47.5 (s, *CH*₃), 1.72 (s, *CH*₃), 1.59 (s, *CH*₃), 1.81 (s, *CH*₃), 1.72 (s, *CH*₃), 1.59 (s, *CH*₃). 1R: 2958 w, 2842 w, 2170 m, 1489 m, 1438 m, 1243 m, 1201 m, 1078 m, 903 m, 825 m, 736 m, 579 m, 425 m. Anal. Calcd. for C₂₃H₅₆N₆Si₃Zr: C, 46.65; H, 9.53; N, 14.19. Found: C, 45.98; H, 9.55; N, 13.97

X-Ray structure determinations. X-Ray diffraction data were collected on a Bruker APEX 2 CCD platform diffractometer (MoK α , $\lambda = 0.71073$ Å) at 125 K. Suitable crystals of each complex **2a** and **4** were mounted in a nylon loop with Paratone-N cryoprotectant oil. The structures were solved using direct methods and standard difference map techniques and were refined by full-matrix least squares procedures on F^2 with SHELXTL (version 6.14).³ All non-hydrogen atoms were refined anisotropically. Hydrogen atoms on carbon were included in calculated positions and were refined using a riding model. The hydrogen atom on nitrogen atom N(5) of **2a**, H(5), was located in the Fourier difference map and refined freely.

compounds	2a	4
formula	$C_{26}H_{63}N_7Si_3Zr$	C ₂₂ H ₅₃ ClN ₆ Si ₃ Zr
MW	649.32	612.64
crystal system	triclinic	triclinic
a/Å	10.3.406(5)	9.517(3)
b/Å	10.3645(5)	10.057(4)
$c/{ m \AA}$	17.2962(8)	19.313(6)
$\alpha/^{\circ}$	88.5340(10)	76.739(5)
β/°	87.1460(10)	80.800(3)
$\gamma^{\prime \circ}$	77.7520(10)	62.440(3)
$V/\text{\AA}^3$	1809.07(15)	1592.0(10)
space group	P-1	P-1
Ζ	2	2
θ range/°	2.01 to 31.95	2.17 to 27.88
m/mm^{-1}	0.428	0.563
N	30146	7558
N_{ind}	11517	7558
$R_{\rm int}$	0.0245	0.0000
$R_1^{a} (I > 2s(I))$	0.0329	0.0260
$wR_2^b (I > 2s(I))$	0.0804	0.0616
$\Delta \rho_{max}; \Delta \rho_{min}/e \ {\rm \AA}^3$	1.013; -0.463	0.815; -0.293
GoF on R ₁	1.034	1.048

Table SI-1. Crystal data and structure refinement parameters for 2a and 4.

 ${}^{a}R_{1} = ||F_{o}| - |F_{c}|| / S|F_{o}|$ ${}^{b}wR_{2} = \{S[w(F_{o}^{2} - F_{c}^{2})^{2}] / S[w(F_{o}^{2})^{2}]\}^{1/2}$

References

1. A. J. Roering, A. F. Maddox, L. T. Elrod, S. M. Chan, M. B. Ghebreab, K. L. Donovan, J. J. Davidson, R. P. Hughes, T. Shalumova, S. N. MacMillan, J. M. Tanski and R. Waterman, *Organometallics*, 2009, **29**, 573–581.

2. R. Waterman, Organometallics, 2007, 27, 2492-2494.

3. Sheldrick, G. M. Acta Cryst. 2008, A64, 112–122