## ELECTRONIC SUPPLEMENTARY INFORMATION

$\underline{\text { Table S1 Comparative torsion angles of the } \mathbf{A} \text { and } \mathbf{B} \text { anions in forms I and II. }}$

| $\mathbf{A}$ | $\mathbf{I}$ | II |
| :---: | :---: | :---: |
| $\mathrm{O}(1)-\mathrm{C}(11)-(\mathrm{C} 2)-\mathrm{C}(1)$ | $24.2(4)$ | $13.0(6)$ |
| $\mathrm{O}(3)-\mathrm{C}(12)-(\mathrm{C} 3)-\mathrm{C}(2)$ | $67.2(3)$ | $82.2(5)$ |
| $\mathrm{O}(5)-\mathrm{C}(13)-(\mathrm{C} 5)-\mathrm{C}(4)$ | $-29.7(4)$ | $-21.2(6)$ |
| $\mathrm{O}(7)-\mathrm{C}(14)-(\mathrm{C} 7)-\mathrm{C}(6)$ | $-178.8(2)$ | $-176.4(4)$ |
| $\mathbf{B}$ | $\mathbf{I}$ | II |
| $\mathrm{O}(9)-\mathrm{C}(25)-(\mathrm{C} 15)-\mathrm{C}(16)$ | $-9.1(4)$ | $-18.4(6)$ |
| $\mathrm{O}(11)-\mathrm{C}(26)-(\mathrm{C} 17)-\mathrm{C}(16)$ | $179.0(3)$ | $176.2(4)$ |
| $\mathrm{O}(13)-\mathrm{C}(27)-(\mathrm{C} 19)-\mathrm{C}(20)$ | $-168.4(3)$ | $-176.0(4)$ |
| $\mathrm{O}(15)-\mathrm{C}(28)-(\mathrm{C} 21)-\mathrm{C}(20)$ | $-175.5(3)$ | $178.6(4)$ |


(a)

(b)

Figure S1: (a) The tetrahedral hydrogen bonding interactions in ATCA, which can be reproduced by replacing the central ATCA molecule with HMTA (b).


Figure S2: The different crystal morphologies of the two forms of the molecular salt, together with their respective unit cell dimensions and face indexes.


Figure S3: The different crystal packing architectures of form I (a) and form II (b). All H atoms are omitted for clarity.


Figure S4: The crystal packing architectures of form I shown down the $c$-axis in the same viewing direction as shown in Fig. S3a. The $\mathbf{A}$ and $\mathbf{C}$ pair of anion and cation is shown in green, and the $\mathbf{B}$ and D pair in yellow. All H atoms are omitted for clarity.


Figure S5: The crystal packing architectures of form II shown down the $a$-axis in a separate viewing direction as shown in Fig. S3b. The $\mathbf{A}$ and $\mathbf{C}$ pair of anion and cation is shown in green, and the $\mathbf{B}$ and D pair in yellow. All H atoms are omitted for clarity.

## Crystal Structure Solution and Refinement:

Intensity data were collected on a Bruker APEX II CCD area detector diffractometer with graphite monochromated $\mathrm{MoK} \alpha_{1}$ radiation ( $50 \mathrm{kV}, 30 \mathrm{~mA}$ ) at 293 K . The collection method involved u-scans of width 0.5 . Data reduction was carried out using the program SAINT,$+{ }^{1}$ version 6.02 , and empirical absorption corrections done using $S A D A B S .{ }^{2}$ The crystal structure was solved by direct methods using SHELXS-97. ${ }^{3}$ Non-hydrogen atoms were first refined isotropically followed by anisotropic refinement by full matrix least-squares calculations based on $F^{2}$ using SHELXL-97. ${ }^{3}$ Hydrogen atoms were first located in the difference map then positioned geometrically and allowed to ride on their respective parent atoms. Diagrams and publication material were generated using WinGX, ${ }^{4}$ ORTEP, ${ }^{5}$ PLATON ${ }^{6}$ and DIAMOND. ${ }^{7}$

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Figure S4: The asymmetric units of salt of Form I, showing the atomic numbering scheme.
Displacement ellipsoids are shown at the $50 \%$ probability level. The H atoms on the C atoms are omitted for clarity.


Figure S5: The asymmetric units of salt of Form II, showing the atomic numbering scheme.
Displacement ellipsoids are shown at the $50 \%$ probability level. The H atoms on the C atoms are omitted for clarity.

