## Sc<sub>2</sub>@ $C_{3\nu}(8)$ -C<sub>82</sub> vs. Sc<sub>2</sub>C<sub>2</sub>@ $C_{3\nu}(8)$ -C<sub>82</sub>: Drastic effect of C<sub>2</sub> capture on the redox properties of scandium metallofullerenes

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Figure S1 Mass spectrum of  $Sc_2@C_{3\nu}(8)$ -C<sub>82</sub> in a negative reflection mode.



Figure S2 CV and DPV curves of  $Sc_2@C_{3\nu}(8)$ -C<sub>82</sub>.



Figure S3 <sup>13</sup>C NMR spectrum (125 MHz) of Sc<sub>2</sub>@ $C_{3\nu}(8)$ - $C_{82}$  measured in CS<sub>2</sub>/acetone- $d_6$  capillary at 298K.

<sup>13</sup>C NMR (125 MHz, CS<sub>2</sub>/acetone-*d*<sub>6</sub> capillary, 298 K) δ 154.46 (s, 3C), 148.33 (s, 6C), 147.61 (s, 3C), 147.50 (s, 6C), 146.28 (s, 6C), 145.91 (s, 3C), 144.26 (s, 3C), 144.22 (s, 6C), 144.07 (s, 6C), 143.76 (s, 6C), 143.58 (s, 6C), 140.69 (s, 1C), 139.94 (s, 6C), 138.12 (s, 6C), 137.79 (s, 3C), 136.57 (s, 6C), 133.93 (s, 6C).



**Figure S4** <sup>45</sup>Sc NMR spectrum (145.8 MHz) of Sc<sub>2</sub>@C<sub>3ν</sub>(8)-C<sub>82</sub> in 1,2-dichlorobenzene- $d_4$  at 293 K. The chemical shift scale was calibrated using Sc<sub>2</sub>O<sub>3</sub> in HCl/D<sub>2</sub>O as external reference (0 ppm).



**Figure S5** Molecular orbital (MO) diagrams of  $C_{3\nu}(8)$ - $C_{82}(\text{left})$  and  $\text{Sc}_2@C_{3\nu}(8)$ - $C_{82}(\text{right})$ .



**Figure S6** Molecular orbital (MO) diagrams of  $C_{3\nu}(8)$ - $C_{82}(\text{left})$  and  $\text{Sc}_2C_2@C_{3\nu}(8)$ - $C_{82}(\text{right})$ .





Sc<sub>2</sub>@C<sub>3v</sub>(8)-C<sub>82</sub>

Sc<sub>2</sub>C<sub>2</sub>@C<sub>3</sub>,(8)-C<sub>82</sub>

Figure S7 LUMOs of  $Sc_2@C_{3\nu}(8)$ -C<sub>82</sub> and  $Sc_2C_2@C_{3\nu}(8)$ -C<sub>82</sub>.