Supplementary Information

Facile and highly atom-economic approach to biaryl-containing

medium-ring bislactones

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General

Nuclear magnetic resonance spectra were recorded on a Bruker DPX 300 at 300MHz with CDCl₃ as solvent. Chemical shifts are reported in δ (ppm) with the solvent resonance as the internal standard (chloroform, H, 7.263 ppm; C, 77.66, 77.23, 76.81 ppm). Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet and br s = broad), coupling constant in Hz, integration, and assignment. Mass spectra were recorded with a VG ZAB-HS spectrometer or a LCQ FLEET mass spectrometer (ESI). Elemental analyses were done on a Elementar Vario MICRO analyzer. Reagents: Benzene (AR grade) was dried with sodium and distilled. Acetonitrile (AR grade) was first refluxed with CaH₂ and distilled. Chloroform was also refluxed with CaH₂ and distilled. THF was dried with sodium and distilled.

Optimization for photoreaction

To make the **DOP** type [4+2] cycloadducts as the predominant product with high yields, solvent for the photo-induced reaction of **PQ** and bicyclopropylidene **1a** was optimized. As shown in Table 1, photoinduced reaction of **PQ** with **1a** in benzene gave the corresponding **DOP** type product **2a** with higher yield than the reaction in acetonitrile.

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Entry	Alkene	Solvents	DOP product and yields ^a
1	1a	PhH	2a (95%)
2		CH ₃ CN	2a (81%)
^a Isolated yield a	fter flash column chromatog	raphy.	

 Table 1
 Optimization on photocycloaddition conditions of PQ with 1a leading to DOP 2a

Photo-cycloaddition of 9, 10-phenanthrenedione (PQ) with strained alkenes 1a in anhydrous benzene. A solution of PQ (416 mg, 2 mmol) and 1a (4 mmol) in anhydrous benzene (100 mL) was photolyzed by $\lambda > 400$ nm light and monitored by TLC to reach a complete conversion of PQ. Flash column chromatograph on 300-400 mesh silica gel with petroleum ether/ethyl acetate as eluents afforded pure analytic final **DOP** product 1a.

2a, colorless oil; ¹**H NMR** (300 MHz, CDCl₃): δ 8.71 (d, J = 8.1 Hz, 2H), 8.27 (dd, J = 8.1, 1.4 Hz, 2H), 7.75-7.63 (m, 4H), 1.36-1.31(m, 4H), 0.95-0.90(m, 4H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 134.7 (2C), 132.3 (2C), 126.8 (2C), 126.5 (2C), 125.0 (2C), 122.6 (2C), 120.8 (2C), 62.0 (2C), 11.0 (4C) ppm; MS m/z (% base): 236 (14), 194 (24), 181 (73), 152 (100), 68 (55); **EA** Found: C, 83.27; H, 5.43. Calc. for C₂₀H₁₆O₂: C, 83.31; H, 5.59%.

Photooxidation of 2a in anhydrous CH₃CN. A solution of **DOP 2a** (0.02M) in anhydrous acetonitrile was photolyzed by $\lambda > 400$ nm light and purged with dry O₂. The reactions were monitored by TLC to reach a complete conversion of the **DOP**. Flash column chromatograph on 300-400 mesh silica gel with petroleum ether/ethyl acetate as eluents afforded the bislactone products **3a**.

3a, white powder; m.p. 192-194°C; ¹**H NMR** (300 MHz, CDCl₃): δ 7.58-7.36 (m, 8H), 1.78-1.69 (m, 2H), 1.44-1.35 (m, 2H), 0.99-0.90 (m, 2H), 0.28-0.19 (m,2H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 170.0 (2C), 137.9 (2C), 133.7 (2C), 131.1 (2C), 131.0 (2C), 127.8 (2C), 126.2 (2C), 63.5 (2C), 14.2 (2C), 6.4 (2C) ppm; IR (KBr) v_{max}/cm⁻¹: 1754, 1443, 1274, 1224, 1069, 756, 748; MS m/z (% base): 236 (2), 180 (100), 152 (55), 68 (16); **EA** Found: C, 75.05; H, 5.05. Calc. for C₂₀H₁₆O₄: C, 74.99; H, 5.03%.

Photooxidation of 2a in anhydrous CH₃OH. A solution of **DOP 2a** (0.02M) in anhydrous CH₃OH was photolyzed by $\lambda > 400$ nm light and purged with dry O₂. The reactions were monitored by TLC to reach a complete conversion of the **DOP**. Flash column chromatograph on 300-400 mesh silica gel with petroleum ether/ethyl acetate as eluents afforded the major product **A** and the minor bislactone product **3a**.



 $\int_{2}^{\infty} A_{2}$, white powder; m.p. 142-144°C; ¹H NMR (300 MHz, CDCl₃): δ 7.69 (d, J = 7.9 Hz,

4H), 7.64 (d, *J* = 8.1 Hz, 2H), 7.52 (d, *J* = 7.6, 1.2 Hz, 2H), 7.39-7.32 (m, 6H), 7.29-7.23 (m, 4H), 2.79 (s, 6H), 1.46-1.38 (m, 2H), 1.09-0.76 (m, 10H), 0.16-0.07 (m, 4H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 132.9, 132.6, 132.5, 132.0, 129.2, 129.1, 127.4, 127.3, 126.6, 125.4, 124.1, 123.6, 102.0, 96.5, 77.4, 57.1, 56.9, 50.7, 13.4, 13.2, 6.3, 6.0 ppm; MS m/z (% base): 239 (100), 211 (23), 195 (91), 180 (77), 152 (47).

Investigations on active oxygen species for photooxidation step

TPP sensitized ${}^{1}O_{2}$ **generation.** A solution of **2a** (0.02M) and tetraphenylporphin (TPP, 0.002M) in anhydrous benzene was photolyzed and purged with dry O₂. The reaction was monitored by TLC and no obvious conversion of **2a** was observed extended the reaction time to 24h.

Thermal decomposition to provide ${}^{1}O_{2}$. A solution of 1,4-dimethylnaphthalene (0.02M) in anhydrous benzene was purged with dry O₂ and photolyzed by $\lambda > 400$ nm light at 0-5 °C. The reaction was monitored by TLC to reach a complete conversion of 1,4-dimethylnaphthalene. There was only one major product and the reaction mixture was evaporated at 0-5 °C to get the product which was the endoperoxide of 1,4-dimethylnaphthalene and could release the pure ${}^{1}O_{2}$ at room temperature. Then dissolved the endoperoxide (0.04M) and **1a** (0.02M) in anhydrous CH₃CN and the solution was stirred at 25 °C. The reaction was monitored by TLC and no obvious conversion of **2a** was observed extended the reaction time to 12h while the endoperoxide was converted to 1,4-dimethylnaphthalene.



Scheme 1 Thermal decomposition to provide ¹O₂

Detection of the active oxygen species by ESR. The generation of the active oxygen species under photolysis was evaluated by electron spin resonance spectroscopy (ESR) analysis using 5, 5-dimethyl-1-pyrroline N-oxide (DMPO) as spin-trapping agent. The results were shown in Figure 1.



Figure S1. ESR spectrum using DMPO (200 mM) as the spin trapper were measured after photolysis of 2a in O₂-saturated acetonitrile for A, 2min; B, 4min; C, 7min and D, 10min. The sample was oxygen-saturated.

Quenching superoxide anion radical by BQ. A solution of 2a (0.02M) and benzoquinone (BQ, 0.002M) in anhydrous acetonitrile was photolyzed and purged with dry O₂. The reaction was monitored by TLC and no obvious conversion of 2a was observed extended the reaction time to 24h.

Detection of CTC complex of 2a by UV spectrum. The absorption changement from the substrate to the ground-state CT complex with molecular oxygen can be measured by direct recording the UV spectrum of oxygen-saturated and oxygen-free solutions of **2a**. The results were shown in Figure 2.



Figure 2. UV spectrum of CTC complex of 2a.

Synthesis of DOP substrates

Photo-cycloaddition of 9, 10-phenanthrenedione (PQ) with alkenes in anhydrous benzene. A solution of PQ (416 mg, 2 mmol) and alkenes (4 mmol) in anhydrous benzene (100 mL) was photolyzed by $\lambda > 400$ nm light and monitored by TLC to reach a complete conversion of PQ. Flash column chromatograph on 300-400 mesh silica gel with petroleum ether/ethyl acetate as eluents afforded pure analytic final **DOP** products.

Photo-cycloaddition of 1,10-phenanthroline-5,6-dione (PN) with alkenes in anhydrous CH₃CN. A solution of PN (204mg, 1 mmol) and alkenes (2mmol) in anhydrous acetonitrile (50 mL) was photolyzed by $\lambda > 400$ nm light and monitored by TLC to reach a complete conversion of PN. Flash column chromatograph on 300-400 mesh silica gel with chloroform/methanol as eluents afforded pure analytic final **DOP** products.

2b, colorless oil; ¹**H NMR** (300 MHz, CDCl₃): δ 8.75 (dd, J = 7.9, 2.1 Hz, 2H), 8.51 (dd, J = 8.1, 0.9 Hz, 1H), 8.30 (dd, J = 8.1, 1.1 Hz, 1H), 7.84-7.65 (m, 4H), 2.65-2.54(m, 2H), 2.38-2.29(m, 2H), 2.20-2.09(m, 1H), 1.89-1.78(m, 1H), 1.36-1.31(m, 2H), 1.16-1.11(m, 2H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 134.1, 132.5, 126.9, 126.7 (2C), 126.5, 126.4, 125.0, 124.8, 122.6 (2C), 121.1, 120.7, 78.7, 61.8, 31.6 (2C), 13.0, 9.0 (2C) ppm; MS m/z (% base): 302 (4), 234 (100); **EA** Found: C, 83.40; H, 5.88. Calc. for C₂₁H₁₈O₂: C, 83.42; H, 6.00%.

2c, colorless oil; ¹**H NMR** (300 MHz, CDCl₃): δ 8.64-8.59 (m, 2H), 8.29 (dd, J= 8.1, 0.9 Hz, 1H), 8.05-8.02 (m, 1H), 7.66-7.49 (m, 4H), 2.08-2.04 (m, 2H), 1.94-1.79 (m, 3H), 1.68-1.63 (m, 2H), 1.39-1.24 (m, 3H), 1.09-0.96 (m, 4H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 127.2, 127.1, 126.8, 126.7, 125.0, 124.6, 122.7, 122.6, 121.1, 120.6, 76.1, 64.5, 30.9 (2C), 25.9, 21.5 (2C), 9.6 (2C) ppm; MS m/z (% base): 180 (92), 152 (100), 76 (42); EA Found: C, 83.70; H, 6.75. Calc. for C₂₃H₂₂O₂: C, 83.60; H, 6.71%.

2d, white powder; ¹**H NMR** (300 MHz, CDCl₃): δ 8.62 (d, J = 7.8 Hz, 2H), 8.26 (dd, J = 8.0, 1.3 Hz, 1H), 7.64-7.51 (m, 4H), 2.00-1.21 (m, 20H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 134.1, 132.5, 126.9, 126.7 (2C), 126.5, 126.4, 125.0, 124.8, 122.6 (2C), 121.1, 120.7, 78.7, 61.8, 31.6 (2C), 13.0, 9.0 (2C) ppm; MS m/z (% base): 208 (82), 180 (100), 152 (97), 104(24), 76 (31); **EA** Found: C, 83.78; H, 7.63. Calc. for C₂₆H₂₈O₂: C, 83.83; H, 7.58%.

4a, white powder; m.p. 164-166°C; ¹**H NMR** (300 MHz, CDCl₃): δ 9.03 (d, J = 3.2 Hz, 2H), 8.35 (d, J = 8.1 Hz, 2H), 7.54 (dd, J = 8.0, 4.4 Hz, 2H), 1.21-1.15 (m, 4H), 0.83-0.78 (m, 2H); ¹³C NMR (75 MHz, CDCl₃): δ 148.4, 142.5, 133.8, 128.8, 123.2, 122.8, 62.5, 11.0 ppm; MS m/z (% base): 290 (100), 262 (24), 233 (16), 182 (16), 79 (17); EA Found: C, 74.55; H, 4.89; N, 9.50. Calc. for C₁₈H₁₄N₂O₂: C, 74.47; H, 4.86; N, 9.65%.

4d, white powder; ¹**H NMR** (300 MHz, CDCl₃): δ 9.04 (dd, J = 4.3, 1.7 Hz, 2H), 8.54 (dd, J = 8.2, 1.2 Hz, 1H), 7.60 (dd, J = 8.3, 4.4 Hz, 1H), 1.96-1.22 (m, 20H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 148.0, 142.5, 130.7, 128.7, 123.8, 122.9, 79.4, 25.9, 21.6 ppm; MS m/z (% base): 374 (57.1), 212 (100), 81 (37.8); **EA** Found: C, 77.03; H, 7.11; N, 7.29. Calc. for C₂₄H₂₆N₂O₂: C, 76.98; H, 7.00; N, 7.48%.

6, white powder; ¹H NMR (300 MHz, CDCl₃): δ 8.66-8.60 (m, 2H), 8.29-8.25 (m, 1H), 8.22-8.18 (m, 1H), 8.03 (d, J = 7.5 Hz, 2H), 7.69-7.50 (m, 5H), 7.37 (t, J = 7.7 Hz, 2H), 7.05 (s, 1H), 4.74 (dd J = 11.5, 1.5 Hz, 1H), 4.49 (d, J = 11.6 Hz, 1H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 133.8, 130.3, 128.6, 127.1, 127.0, 125.7, 125.5, 122.7, 122.6, 121.2, 121.0, 88.2, 85.6, 32.2, 29.9, 29.6, 22.9, 14.3 ppm; EA Found: C, 77.48; H, 4.60. Calc. for C₂₃H₁₆O₄: C, 77.52; H, 4.53%.

8, white powder; ¹**H NMR** (300 MHz, CDCl₃): δ 8.66-8.61 (m, 2H), 8.29-8.23 (m, 2H), 7.70-7.56 (m, 4H), 5.49 (dd, J = 3.0, 1.7 Hz, 1H), 4.45 (dd, J = 11.0, 2.9 Hz, 1H), 4.36 (dd, J = 11.0, 1.8 Hz, 1H), 4.07-3.98 (m, 1H), 3.82-3.73 (m, 1H), 1.70-1.60 (m, 2H), 1.43-1.30 (m, 2H), 0.90 (t, J = 7.3 Hz, 3H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 133.4, 130.8, 127.1, 126.9, 126.8, 126.5, 126.2, 125.2, 125.1, 122.7, 122.6, 121.2, 120.7, 69.1, 66.6, 31.7, 19.3, 13.9 ppm; **EA** Found: C, 78.03; H, 6.55. Calc. for C₂₀H₂₀O₃: C, 77.90; H, 6.54%.

10, white powder; ¹**H NMR** (300 MHz, CDCl₃): δ 8.64-8.60 (m, 2H), 8.23-8.14 (m, 2H), 7.67-7.55 (m, 4H), 6.78 (t, 1H, J = 1.5 Hz), 4.59 (dd, 1H, J = 11.6, 1.9 Hz), 4.49 (dd, 1H, J = 11.7, 1.4 Hz), 2.11 (s, 3H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 169.8, 133.3, 130.7, 127.1, 127.0, 126.9, 125.9, 125.6, 125.4, 122.7, 122.6, 121.0, 120.8, 87.5, 65.3, 21.1 ppm; **EA** Found: C, 73.55; H, 4.81. Calcd. for: C₁₈H₁₄O₄ C, 73.46; H, 4.79%.

Substrate 12 was prepared according to reported method.

Photooxidation of DOPs in anhydrous CH₃CN

A solution of **DOP** (0.02M) in anhydrous acetonitrile was photolyzed by $\lambda > 400$ nm light (used $\lambda > 300$ nm light for the photooxidation of substrate **4a** and **4d**) and purged with dry O₂. The reactions were monitored by TLC to reach a complete conversion of the **DOPs**. Flash column chromatograph on 300-400 mesh silica gel with petroleum ether/ethyl acetate as eluents afforded the bislactone products.

3b, white powder; m.p. 145-147°C; ¹**H** NMR (300 MHz, CDCl₃): δ 7.74 (d, *J* = 7.5 Hz, 1H), 7.58-7.35 (m, 7H), 3.10-2.99 (m, 1H), 2.79-2.69 (m, 1H), 2.63-2.51 (m, 1H), 2.05-1.57 (m, 4H), 1.54-1.45 (m, 1H), 1.18-1.09 (m, 1H), 1.03-0.91 (m, 1H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 170.9, 168.7, 138.8, 138.7, 134.0, 132.5, 131.4, 131.3, 130.9, 130.8, 127.7, 127.6, 127.3, 125.9, 87.1, 66.0, 33.0, 30.4, 15.6, 11.2, 6.1 ppm; IR (KBr) v_{max}/cm⁻¹: 1742, 1727, 1478, 1279, 1233, 1125, 1063, 746; MS m/z (% base): 224 (1), 180 (100), 152 (60), 76 (5), 54 (11); **EA** Found: C, 75.44; H, 5.32. Calc. for C₂₁H₁₈O₄: C, 75.43; H, 5.43%.

3c, white powder; m.p. 171-172°C; ¹**H NMR** (300 MHz, CDCl₃): δ 7.60-7.39 (m, 8H), 3.43-3.38 (d, J = 13.8 Hz, 1H), 1.99-1.94 (d, J = 12.9 Hz, 1H), 1.74-0.54(m, 12H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 170.3, 169.4, 138.5, 137.6, 135.3, 134.0, 130.8, 130.7, 130.6, 130.5, 127.7, 127.4, 126.5, 126.0, 89.9, 67.3, 33.2, 33.0, 25.1, 22.5, 22.1, 11.5, 8.2 ppm; IR (KBr) v_{max} /cm⁻¹: 1743, 1726, 1445, 1288, 1234, 1125, 1070, 747; MS m/z (% base) 225 (1), 180 (100), 152 (26), 81 (20), 67 (19). EA Found: C, 76.13; H, 6.15. Calc. for C₂₃H₂₂O₄: C, 76.22; H, 6.12%.

3d, white powder; m.p. 176-180°C; ¹**H NMR** (300 MHz, CDCl₃): δ 7.72-7.68 (dd, J = 7.5, 0.8 Hz, 2H), 7.56-7.50 (td, J = 7.5, 1.3 Hz, 2H), 7.45-7.36 (m, 4H), 2.76-2.71 (d, J = 13.1 Hz, 2H), 2.76-2.71 (d, J = 12.2 Hz, 2H), 1.77-1.18 (m, 16H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 169.5 (2C), 139.6 (2C), 134.6 (2C), 130.9 (4C), 127.4 (2C), 127.2 (2C), 94.5 (2C), 33.6 (2C), 31.1 (2C), 25.3 (2C), 23.0 (4C) ppm; IR (KBr) ν_{max} /cm⁻¹: 1732, 1450, 1275, 1236, 1074, 749; MS m/z (% base): 180 (100), 162 (28), 152 (14), 91 (59), 79 (71); EA Found: C, 77.24; H, 7.00. Calc. for C₂₆H₂₈O₄: C, 77.20; H, 6.98%.

5a, white powder; m.p. 254-256°C; ¹**H** NMR (300 MHz, CDCl₃): δ 8.93 (d, J = 3.6 Hz, 2H), 7.83 (d, J = 7.5 Hz, 2H), 7.41 (dd, J = 7.5, 4.8 Hz, 2H), 1.78-1.65 (m, 2H), 1.45-1.33 (m, 2H), 1.03-0.94 (m, 2H), 0.33-0.24 (m, 2H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 168.6, 153.6, 152.0, 134.4, 130.3, 123.0, 64.5, 14.1, 6.7 ppm; IR (KBr) v_{max}/cm⁻¹: 1759, 1741, 1417, 1278, 1236, 1131, 1079, 768; MS m/z (% base): 322 (0.14, M⁺), 266 (2), 210 (48), 182 (100), 154 (65), 127 (31); **EA** Found: C, 67.09; H, 4.27; N, 8.74. Calc. for C₁₈H₁₄N₂O₄: C, 67.07; H, 4.38; N, 8.69%.

5d, white powder; m.p. 180-183°C; ¹**H NMR** (300 MHz, CDCl₃): δ 8.90 (dd, J = 4.8, 1.4 Hz, 2H), 8.13 (dd, J = 7.8, 1.5 Hz, 2H), 7.40 (dd, J = 7.8, 4.8 Hz, 2H), 2.44-2.34 (m, 4H), 1.72-1.14 (m, 16H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 167.9, 155.0, 151.7, 136.2, 130.0, 123.0, 96.3, 32.8, 31.1, 25.0, 22.7, 22.3 ppm; IR (KBr) v_{max} /cm⁻¹: 1730, 1557, 1416, 1305, 1274, 1080, 763; MS m/z (% base): 210 (13), 182 (100), 162 (42), 154 (32), 91 (74), 79 (73); **EA** Found: C, 71.11; H, 6.58; N, 6.69. Calc. for C₂₄H₂₆N₂O₄: C, 70.92; H, 6.45; N, 6.89%.

7, white powder; ¹H NMR (300 MHz, CDCl₃) of the predominant diastereoisomer: δ 8.14 (d, *J* = 7.5 Hz, 2H), 7.68-7.38 (m, 12H), 5.46 (t, *J* = 10.1 Hz, 1H), 4.30 (dd, *J* = 11.1, 2.7 Hz, 1H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 169.2, 166.8, 163.9, 137.4, 137.2, 134.4, 134.3, 133.3, 133.2, 133.1, 132.9, 132.3, 131.7, 131.5, 131.2, 131.1, 130.4, 128.8, 128.4, 128.2, 128.1, 127.9, 127.8, 127.7, 126.6, 126.3, 126.2,

90.8, 89.8, 64.7, 64.4 ppm; IR (KBr) v_{max} /cm⁻¹: 1754, 1733, 1438, 1235, 1015, 750, 707; MS m/z (% base): 180 (70), 152 (8), 105 (100), 77 (15); **EA** Found: C, 71.05; H, 4.40. Calc. for C₂₃H₁₆O₆: C, 71.13; H, 4.15%.

9, white powder; ¹**H NMR** (300 MHz, CDCl₃) of the predominant diastereoisomer: δ 7.59-7.39 (m, 8H), 6.29 (dd, J = 8.8, 2.7 Hz, 1H), 5.16 (t, J = 10.0 Hz, 1H), 4.12 (dd, J = 11.2, 2.5 Hz, 1H), 3.96-3.87 (m, 1H), 3.75-3.67 (m, 1H), 1.72-1.62 (m, 2H), 1.50-1.42 (m, 2H), 0.98 (t, J = 7.3 Hz, 3H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 169.3, 169.0, 137.5, 137.0, 133.9, 133.4, 133.3, 133.0, 132.6, 131.5, 131.3, 131.2, 130.9, 130.8, 128.0, 127.8, 127.7, 127.4, 126.2, 126.0, 99.6, 98.9, 70.4, 70.3, 67.6, 65.7, 31.5, 31.4, 19.2, 19.1, 13.9 ppm; IR (KBr) v_{max}/cm^{-1} : 1755, 1733, 1438, 1257, 1235, 1014, 751, 707; MS m/z (% base): 196 (100), 180 (58), 168 (43), 152 (18), 139 (43); EA Found: C, 70.66; H, 6.05. Calc. for C₂₀H₂₀O₅: C, 70.57; H, 5.92%.

11, white powder; ¹H NMR (300 MHz, CDCl₃) of the predominant diastereoisomer: δ 7.57-7.40 (m, 8H), 7.27 (dd, 1H, J = 9.2, 2.9 Hz), 5.26 (dd, 1H, J = 11.1, 9.4 Hz), 4.15 (dd, 1H, J = 11.0, 2.8 Hz), 2.20 (s, 3H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 169.2, 168.3, 166.8, 137.4, 137.3, 133.3, 133.2, 132.9, 132.3, 131.7, 131.5, 131.3, 131.1, 130.9, 130.6, 128.2, 127.9, 127.8, 127.7, 126.7, 126.3, 126.1, 90.3, 89.3, 64.5, 64.3, 20.8, 20.7 ppm; IR (KBr) cm⁻¹: 1758, 1277, 1204, 1024, 755; MS m/z (% base); EA Found: C, 66.29; H, 4.19. Calcd. for: C₁₈H₁₄O₆ C, 66.26; H, 4.32%.

13, white powder; ¹**H NMR** (300 MHz, CDCl₃): δ 7.87 (dd, J = 7.8, 1.4 Hz, 1H), 7.73 (dd, J = 7.8, 1.3 Hz, 1H), 7.60 (tt, J = 7.6, 1.6 Hz, 2H), 7.47 (tt, J = 7.6, 1.2 Hz, 2H), 7.38-7.32 (m, 2H), 5.89 (q, J = 5.7 Hz, 1H), 1.72 (d, J = 5.9 Hz, 3H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 169.0, 167.9, 140.6, 140.5, 132.7, 132.4, 132.3, 131.7, 129.9, 129.2, 129.1, 127.9, 127.8, 127.7, 98.2, 29.8, 20.7 ppm; MS m/z (% base): 224 (3), 180 (100), 152 (37), 76 (2); EA Found: C, 71.56; H, 4.61. Calc. for C₁₆H₁₂O₄: C, 71.64; H, 4.51%.

Substrate 14 was also prepared according to reported method, but only PQ obtained for 14 under photooxidation.

Sequential photocycloadditions and photooxidations of PQ or PN with aryl substituted alkenes

One-pot photoreaction of PQ with styrene. A solution of **PQ** (416 mg, 2 mmol) and styrene (4 mmol) in anhydrous acetonitrile (100 mL) was photolyzed by $\lambda > 400$ nm light and purged with dry O₂. The reactions were monitored by TLC to reach a complete conversion of **PQ** and **DOP** product. Flash column chromatograph on 300-400 mesh silica gel with petroleum ether/ethyl acetate as eluents afforded the bislactone with a yield of 67%.

Sequential photocycloadditions and photooxidations of PQ or PN with aryl substituted alkenes. A solution of PQ (416 mg, 2 mmol) and aryl substituted alkenes (4 mmol) in 100 mL anhydrous benzene (PN (420 mg, 2 mmol) and 1e (4 mmol) in 100 mL anhydrous acetonitrile) was photolyzed by $\lambda > 400$ nm light and monitored by TLC to reach a complete conversion of PQ (or PN). Then evaporation to remove benzene and the residue were dissolved by 100 mL anhydrous CH₃CN. The solution was photolyzed and purged with dry O₂. The reactions were monitored by TLC to reach a complete conversion of **DOP** products. Flash column chromatograph on 300-400 mesh silica gel with petroleum ether/ethyl acetate as eluents afforded the bislactone products as a mixture of diastereoisomers.

3e, white powder; ¹**H NMR** (300 MHz, CDCl₃) of the predominant diastereoisomer: δ 7.66-7.33 (m, 13H), 6.48 (dd, J = 10.7, 2.9 Hz, 1H), 5.33 (t, J = 11.1 Hz, 1H), 4.14 (dd, J = 12.5, 3.0 Hz, 1H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 169.5, 169.0, 137.4, 137.3, 133.8, 133.7, 131.3, 131.2, 130.9, 129.3, 129.1, 127.8, 127.3, 126.8, 126.4, 126.2, 77.2, 69.1 ppm; IR (KBr) v_{max} /cm⁻¹: 1742, 1599, 1280, 1238, 1128, 754; MS m/z (% base): 238 (35), 180 (100), 152 (49), 91 (4), 76 (4); **EA** Found: C, 76.70; H, 4.66. Calc. for C₂₂H₁₆O₄: C, 76.73; H, 4.68%.

5e, white powder; ¹**H NMR** (300 MHz, CDCl₃) of the predominant diastereoisomer: δ 8.97 (s, 2H), 7.86-7.83 (m, 2H), 7.58-7.40 (m, 7H), 6.50 (d, *J* = 9.5 Hz, 1H), 5.37 (t, *J* = 11.1 Hz, 1H), 4.19 (d, *J* = 10.6 Hz, 1H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 167.9, 167.5, 155.6, 152.9, 152.8, 151.9, 134.3, 134.2, 133.2, 133.1, 130.2, 130.1, 129.7, 129.2 (2C), 128.4, 127.3, 123.1, 77.9, 69.3 ppm; IR (KBr) v_{max}/cm⁻¹: 1754, 1741, 1562, 1413, 1271, 1235, 1082, 769, 704; MS m/z (% base): 240 (16), 210 (56), 182 (100), 154 (55), 127 (26); **EA** Found: C, 69.32; H, 4.18; N, 7.94. Calc. for C₂₀H₁₄N₂O₄: C, 69.36; H, 4.07; N, 8.09%.

3f, white powder; ¹**H NMR** (300 MHz, CDCl₃) of the predominant diastereoisomer: δ 7.66-7.39 (m, 10H), 7.00-6.94 (m, 2H), 6.42 (dd, J = 10.7, 2.9 Hz, 1H), 5.32 (t, J = 11.6 Hz, 1H), 4.09 (dd, J = 11.5, 3.1 Hz, 1H), 3.85 (s, 3H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 169.5, 169.0, 160.4, 137.4, 137.2, 133.8, 133.7, 131.3, 131.1, 130.9, 130.8, 128.8, 128.4, 127.8, 126.4, 126.2, 125.7, 114.4, 77.0, 69.0, 55.5 ppm; IR (KBr) ν_{max}/cm^{-1} : 1742, 1516, 1276, 1237, 1133, 752; MS m/z (% base): 180 (100), 152 (55), 135 (23), 121 (15); **EA** Found: C, 73.77; H, 4.63. Calc. for C₂₃H₁₈O₅: C, 73.79; H, 4.85%.

3g, white powder; ¹**H NMR** (300 MHz, CDCl₃) of the predominant diastereoisomer: δ 7.80-7.20 (m, 12H), 6.82 (d, *J* = 9.4 Hz, 1H), 5.14 (t, *J* = 10.7 Hz, 1H), 4.33 (d, *J* = 10.6 Hz, 1H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 169.7, 168.4, 137.5, 137.4, 133.7, 133.6, 133.3, 131.4, 131.3, 131.1, 131.0, 130.6, 128.8, 128.3, 128.0, 127.9, 126.3, 126.2, 122.5, 76.3, 67.6 ppm; IR (KBr) v_{max}/cm⁻¹: 1746, 1476, 1274, 1232, 1068, 748; MS m/z (% base): 224 (4), 180 (100), 152 (76), 76 (3); **EA** Found: C, 62.39; H, 3.62; Br, 18.77. Calc. for C₂₂H₁₅BrO₄: C, 62.43; H, 3.57; Br, 18.88%.

3h, white powder; ¹**H NMR** (300 MHz, CDCl₃) of the predominant diastereoisomer: δ 8.67 (br s, 2H), 7.62-7.34 (m, 10H), 6.47 (dd, J = 10.5, 2.3 Hz, 1H), 5.22 (t, J = 12.2 Hz, 1H), 4.15 (dd, J = 11.5, 2.7 Hz, 1H) ppm; ¹³C NMR (75 MHz, CDCl₃): δ 169.2, 168.5, 150.4, 142.6, 137.2, 133.3, 133.1, 131.2, 131.1, 131.0, 127.9, 126.2, 126.1, 121.5, 75.4, 68.2 ppm; IR (KBr) v_{max} /cm⁻¹: 1745, 1440, 1285 1240, 1133, 753; MS m/z (% base): 315 (2), 238 (5), 208 (72), 180 (68), 152 (100), 76 (14); **EA** Found: C, 72.97; H, 4.44; N, 4.09. Calc. for C₂₁H₁₅NO₄: C, 73.03; H, 4.38; N, 4.06%.

3i, white powder; ¹**H NMR** (300 MHz, CDCl₃): δ 7.78 (d, J = 7.3 Hz, 1H), 7.63-7.26 (m, 17H), 5.48 (d, J = 12.3 Hz, 1H), 4.95 (d, J = 12.5 Hz, 1H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 170.0, 167.1, 142.0, 141.1, 138.3, 138.2, 133.9, 133.4, 131.3, 131.0, 130.9, 130.8, 128.5, 128.3, 127.9, 127.8, 127.6, 126.9, 126.5, 125.9, 89.5, 69.7 ppm; IR (KBr) v_{max}/cm⁻¹: 1732, 1721, 1451, 1278, 1269, 1135, 758, 700; MS m/z (% base): 196 (5), 180 (100), 167 (96), 152 (64), 76 (4); **EA** Found: C, 80.06; H, 4.75. Calc. for C₂₈H₂₀O₄: C, 79.98; H, 4.79%.

3j, white powder; ¹**H NMR** (300 MHz, CDCl₃) of the predominant diastereoisomer: δ 7.65-7.33 (m, 13H), 6.06 (d, J = 9.3 Hz, 1H), 5.64-5.53 (m, 1H), 1.27 (d, J = 6.3 Hz, 3H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 169.7, 169.0, 137.3, 137.2, 134.7, 134.2, 133.9, 131.3, 131.2, 130.8, 130.7, 129.5, 129.1, 128.3, 127.9, 127.8, 126.4, 126.3, 82.1, 76.0, 15.8 ppm; IR (KBr) v_{max}/cm^{-1} : 1754, 1734, 1445, 1275, 1234, 1121, 1062, 746; MS m/z (% base): 208 (100), 180 (15), 152 (75), 76 (10); **EA** Found: C, 76.91; H, 4.96. Calc. for: C₂₃H₁₈O₄: C, 77.08; H, 5.06%.

31, white powder; all the characterization data below are of the mixture of diastereoisomers; ¹**H NMR** (300 MHz, CDCl₃): δ 8.16 (d, J = 7.2 Hz, 1H), 7.70-7.32 (m, 15H), 6.43 (d, J = 1.9 Hz, 1H), 6.03 (d, J = 3.7 Hz, 1H), 5.63 (t, J = 9.6 Hz, 1H), 5.26-5.07 (m, 5H), 4.46-4.09 (m, 6H), 2.12 (s, 3H), 2.10 (s, 3H), 2.07 (s, 3H), 2.04 (s, 3H), 1.99 (s, 3H), 1.86 (s, 3H) ppm; ¹³**C NMR** (75 MHz, CDCl₃): δ 170.8, 170.7, 169.8 (2C), 169.5 (2C), 168.3, 168.2, 166.3, 166.0, 140.8, 140.4, 139.7, 137.4, 134.0, 133.8, 133.1, 132.7, 132.5, 132.4, 132.2, 131.7, 131.6, 131.0, 130.7, 130.4, 128.4, 128.3, 128.0 (2C), 126.8, 126.5, 125.9, 125.1 97.2, 95.0, 73.8, 72.3, 71.2, 70.9, 70.1, 69.2, 67.6, 67.4, 61.6, 61.3, 20.9, 20.8 (2C), 20.7, 20.6 (2C) ppm; IR (KBr) v_{max}/cm^{-1} : 1752, 1370, 1285, 1226, 1017; MS m/z (% base): 224 (0.5), 180 (100), 152 (31), 43 (85); **EA** Found: C, 60.74; H, 4.80. Calc. for C₂₆H₂₄O₁₁: C, 60.94; H, 4.72%.

Crystal structure of compound 5e:



Crystal structure of compound 3j:



Crystal structure of compound 7:

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Copies of ¹H NMR and ¹³C NMR spectra of all the new compounds











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5a-¹³C NMR(CDCl₃)



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5d-¹H NMR(CDCl₃) 1.714 1.688 1.681 1.691 1.6931 8.907 8.903 8.903 8.893 8.893 8.892 8.145 8.145 8.145 8.145 8.145 8.145 7.422 7.422 7.406 7.7306 7.7306 2.446 2.394 2.349 2.349 Mint ٦ ٢ <u>ו</u> 12/78 \cap 4.00 2.25 2.19 200 1.99 峢 WA PPM 10 0 8 ... 5d-¹³C NMR(CDCl₃) -136.152 77.649 25.109 O 0 200 50 0 PPM 150 100











⁹⁻¹³C NMR(CDCl₃)























⁵e-13C NMR(CDCl₃)







200

150

100



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3g-<sup>13</sup>C NMR(CDCl<sub>3</sub>)
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³h-¹³C NMR(CDCl₃)







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3i-¹H NMR(CDCl₃)









3j from photoreaction of PQ with 1j -¹H NMR(CDCl₃)

3j from photoreaction of PQ with 1j -¹³C NMR(CDCl₃)



3j from photoreaction of PQ with 1k -¹H NMR(CDCl₃)



3j from photoreaction of PQ with 1k -¹³C NMR(CDCl₃)



15.772







