

## Electronic Supplementary Information

### **Twist Angle Perturbation on Mixed (Phthalocyaninato)(porphyrinato) Dysprosium(III) Double-decker SMMs**

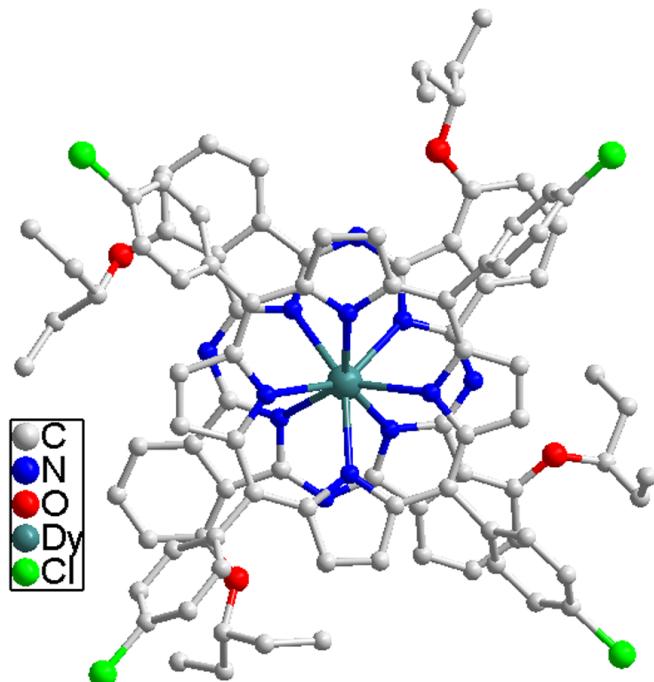
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## Experimental Section

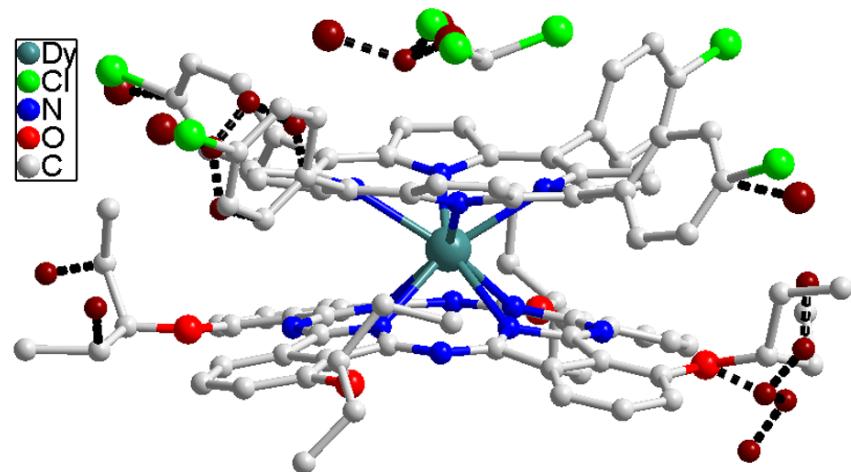
**General Remarks.** All the reagents and solvents were used as received. The compounds Dy(Pc)(TCIPP) (**1a**), Dy{Pc( $\alpha$ -OC<sub>5</sub>H<sub>11</sub>)<sub>4</sub>}(TCIPP) (**2a**), and DyH{Pc( $\alpha$ -OC<sub>5</sub>H<sub>11</sub>)<sub>4</sub>}(TCIPP) (**2b**) were prepared and isolated following the reported procedures mentioned in the text.

Electronic absorption spectra were recorded on a Hitachi U-4100 spectrophotometer. MALDI-TOF mass spectra were taken on a Bruker BIFLEX III ultrahighresolution Fourier transform ion cyclotron resonance (FT-ICR) mass spectrometer with alpha-cyano-4-hydroxycinnamic acid as matrix. Elemental analyses were performed on an Elementar Vario El III. Electrochemical measurements were carried out with a BAS CV-50W voltammetric analyser.

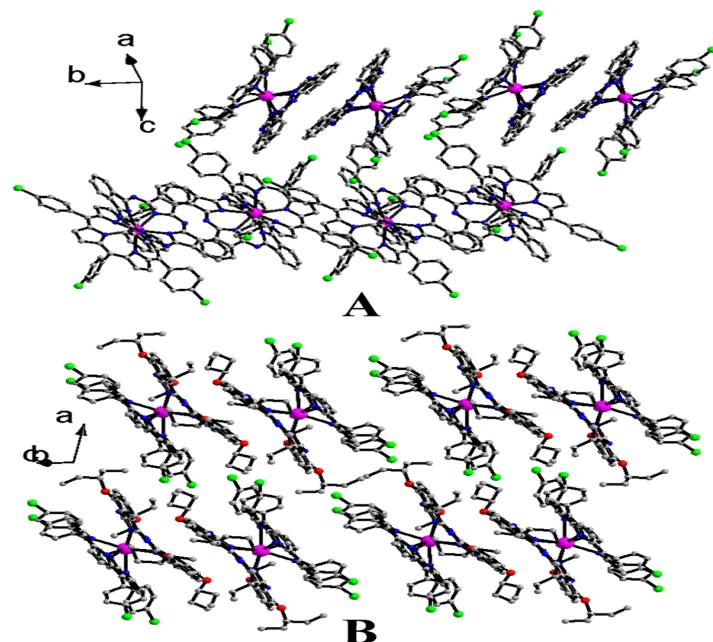
Satisfactory elemental analysis results were obtained for all the newly prepared double-decker complexes in particular **1a**, **2a**, and **2b** after repeatedly column chromatographic purification followed by recrystallization from chloroform and methanol, Table S3 (Supporting Information). The MALDI-TOF mass spectra of these complexes showed an intense cluster corresponding to the molecular ion [M+H]<sup>+</sup>, Table S3 (Supporting Information). These sandwich double-decker compounds were also characterized with electronic absorption spectroscopy, Fig. S12†. Single crystals of all three mixed ring double-decker compounds suitable for X-ray diffraction analysis were obtained by diffusion of methanol onto the solution of corresponding compounds in chloroform.



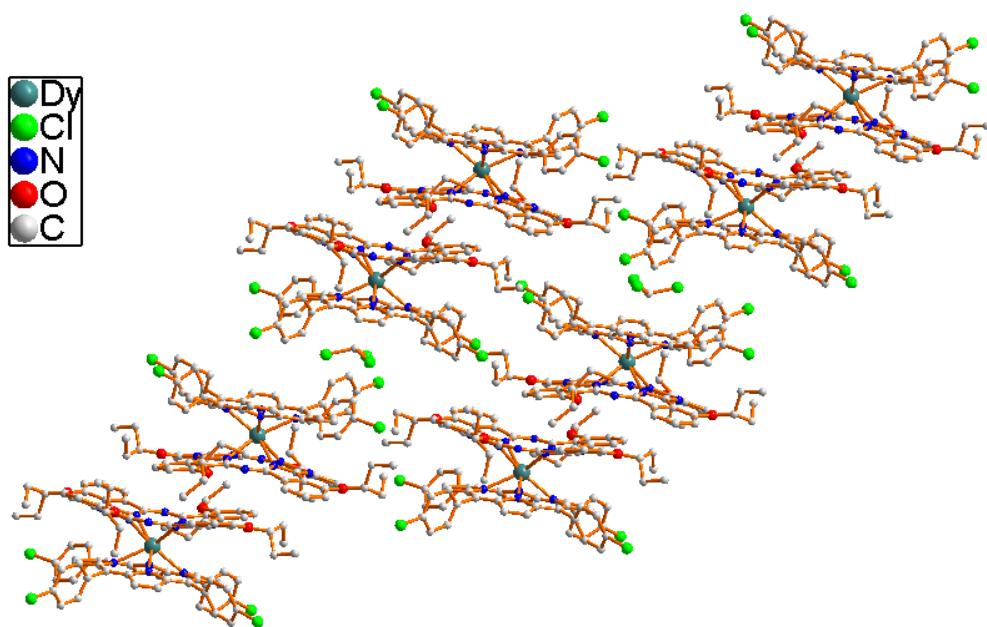
**Figure S1.** Molecular structure of  $\text{Dy}[\text{Pc}(\alpha\text{-OC}_5\text{H}_{11})_4](\text{TCIPP})$  (**2a**) in a top view with all hydrogen atoms and solvent molecules omitted for clarity.



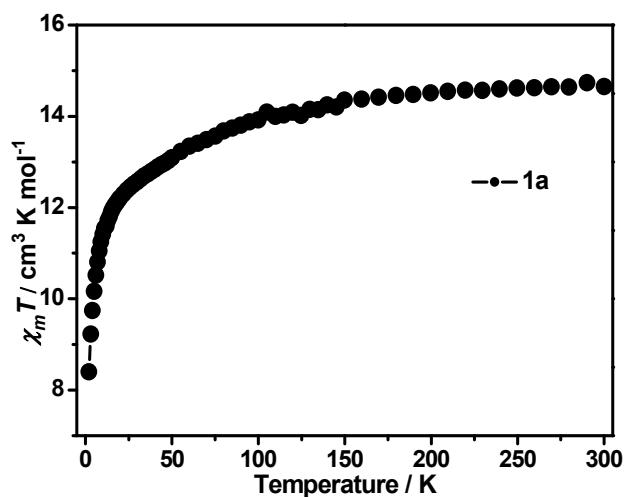
**Figure S2.** Molecular structure of  $\text{DyH}[\text{Pc}(\alpha\text{-OC}_5\text{H}_{11})_4](\text{TCIPP})$  (**2b**) in a side view with all disorder atoms drawn in dark red colour and corresponding bond drawn in black dash line (all hydrogen atoms omitted for clarity).



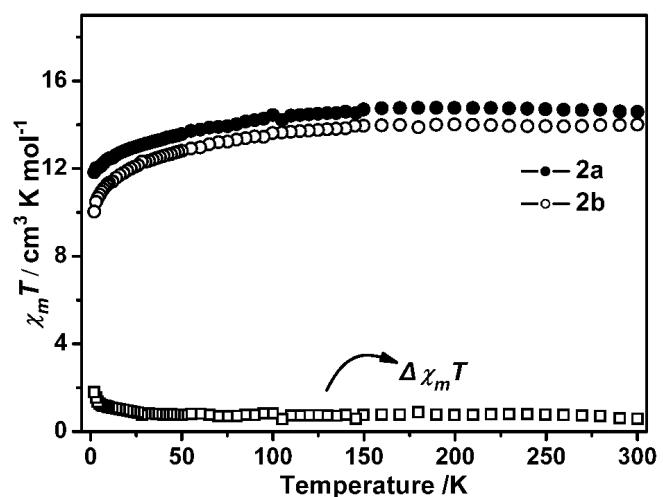
**Figure S3.** Molecular packing in single crystals of **1a** (A) and **2a** (B) with hydrogen atoms and solvent molecules omitted for clarity.



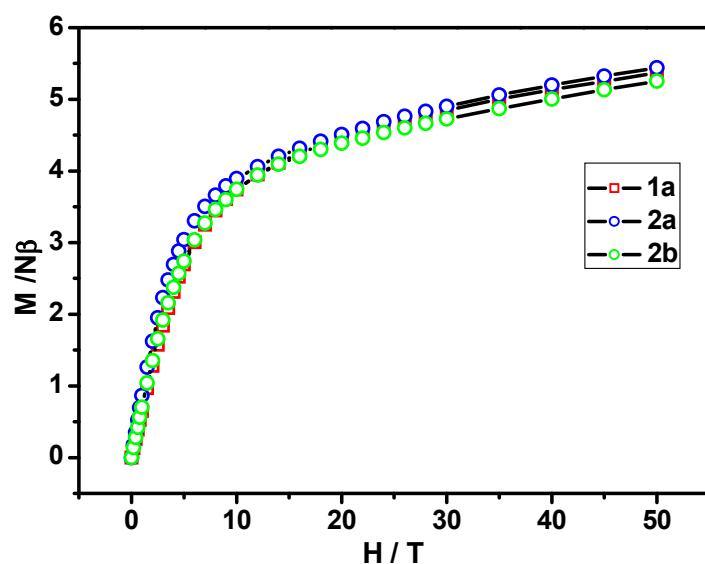
**Figure S4.** Molecular packing in single crystal of  $\text{DyH}[\text{Pc}(\alpha\text{-OC}_5\text{H}_{11})_4](\text{TCIPP})$  (**2b**) (B) with hydrogen atoms and solvent [molecules](#) omitted for clarity.



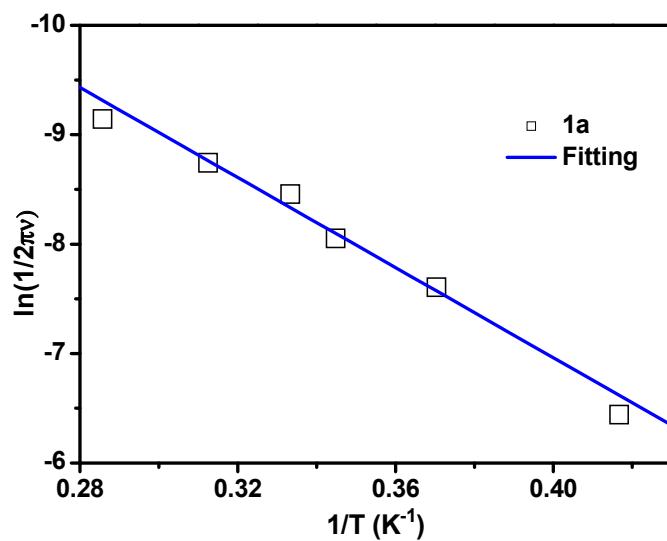
**Figure S5.** Temperature dependence of  $\chi_m T$  for **1a**.



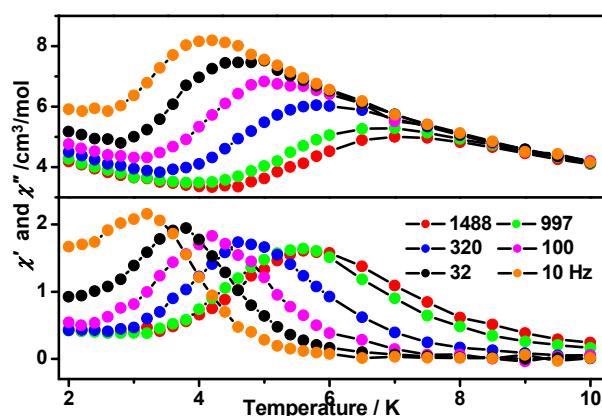
**Figure S6.** Temperature dependence of  $\chi_m T$  for **2a** and **2b** as well as  $\Delta\chi_m T$  ( $\Delta\chi_m T = \chi_m T_{2a} - \chi_m T_{2b}$ ).



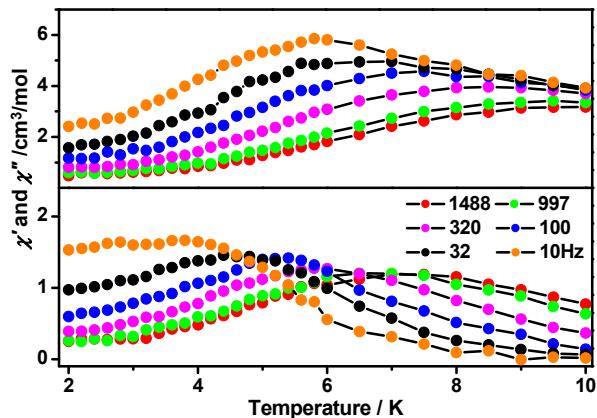
**Figure S7.** The  $M$  vs  $H/T$  curves for **1a**, **2a**, and **2b** at 2.0 K.



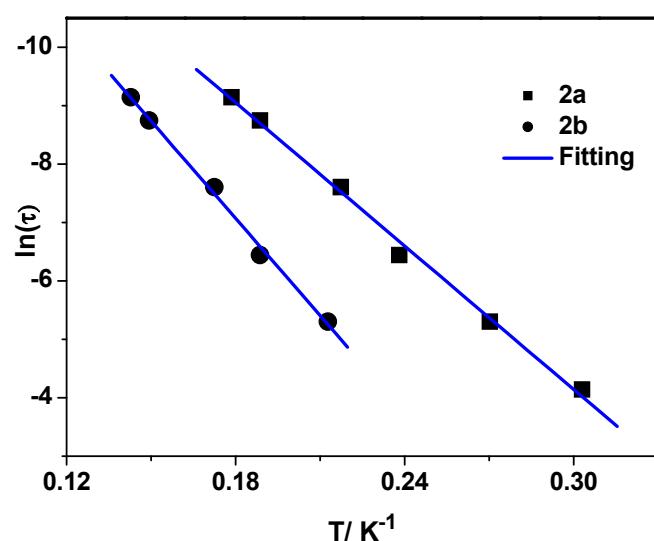
**Figure S8.** The plot of  $\ln(\tau)$  vs.  $1/T$  for **1a** under zero applied field.



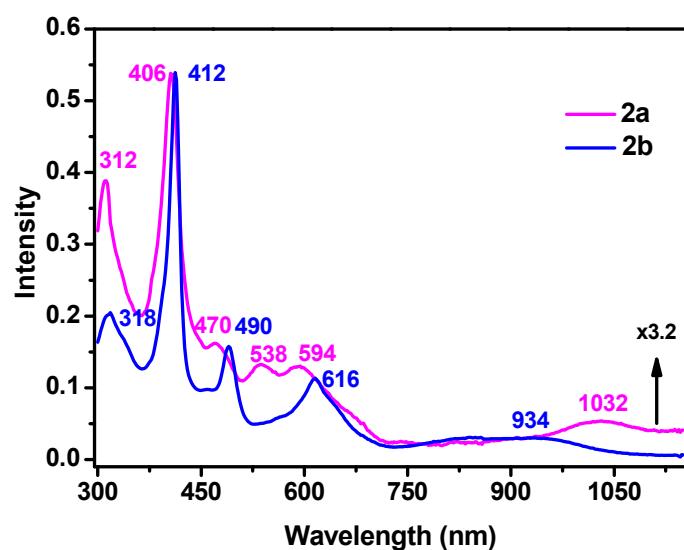
**Figure S9.** Temperature dependence of the in-phase ( $\chi'$ ) and out-of-phase ( $\chi''$ ) ac susceptibility of **2a** (top and bottom).



**Figure S10.** Temperature dependence of the in-phase ( $\chi'$ ) and out-of-phase ( $\chi''$ ) ac susceptibility of **2b** (top and bottom).



**Figure S11.** The plot of  $\ln(\tau)$  vs.  $1/T$  for **2a** and **2b** under 2000 Oe.



**Figure S12.** Electronic absorption spectra of double-decker **2a** and **2b** in  $\text{CHCl}_3$ .

**Table S1.** Crystallographic data and selected structural information for **1a**, **2a**, and **2b**.

	<b>1a</b>	<b>2a</b>	<b>2b</b>
Molecular formula	C <sub>79</sub> H <sub>54</sub> Cl <sub>10</sub> DyN <sub>12</sub> O <sub>5</sub>	C <sub>97</sub> H <sub>81</sub> Cl <sub>7</sub> DyN <sub>12</sub> O <sub>4</sub>	C <sub>97</sub> H <sub>82</sub> Cl <sub>7</sub> DyN <sub>12</sub> O <sub>4</sub>
<i>M</i>	1768.38	1889.39	1890.44
Temperature /K	293	120	150
Crystal system	Monoclinic	Triclinic	Triclinic
Space group	<i>C</i> 2/ <i>c</i>	<i>P</i> - <i>I</i>	<i>P</i> - <i>I</i>
<i>a</i> /Å	38.436(5)	14.0680(5)	14.0969(4)
<i>b</i> /Å	19.024(2)	17.6605(7)	17.6251(5)
<i>c</i> /Å	26.273(3)	18.6659(7)	18.9482(5)
$\alpha^{\circ}$	90	112.931(4)	111.959(2)
$\beta^{\circ}$	127.986(3)	96.166(3)	97.403(2)
$\gamma^{\circ}$	90	97.426(3)	97.046(2)
<i>V</i> /Å <sup>3</sup>	15141	4171.5(3)	4254.5(2)
<i>Z</i>	8	2	2
<i>F</i> (000)	7040	1928	1928
<i>D</i> <sub>c</sub> /Mg m <sup>-3</sup>	1.551	1.504	1.475
Data collection range/ <sup>o</sup>	1.56 to 25.00	2.90 to 25.00	2.98 to 25.00
Reflections collected	36641	29700	30362
Independent reflections	13255	14580	14890
<i>R</i> <sub>1</sub> [ <i>I</i> >2σ( <i>I</i> )]	0.0487	0.0668	0.0510
<i>wR</i> <sub>2</sub> [ <i>I</i> >2σ( <i>I</i> )]	0.1398	0.1387	0.1167
<i>R</i> <sub>1</sub> [ <i>for all</i> ]	0.0634	0.0969	0.0672
<i>wR</i> <sub>2</sub> [ <i>for all</i> ]	0.1506	0.1527	0.1238
Goodness of fit	1.092	1.025	1.029
separation of Dy and two N <sub>4</sub> planes (Å)	2.76	2.794	2.794
separation of Dy and N <sub>4</sub> planes of P <sub>c'</sub> (Å)	1.483	1.484	1.484
skew angle φ (°)	43.60	38.02	38.02

**Table S2.** Selected bond distances ( $\text{\AA}$ ) and bond angles ( $^{\circ}$ ) for **1a**, **2a**, and **2b**.

bond and angle	<b>1a</b>	<b>2a</b>	<b>2b</b>
N(1)-Dy(1)	2.468(4)	2.441(5)	2.479(4)
N(3)-Dy(1)	2.466(4)	2.450(5)	2.494(4)
N(5)-Dy(1)	2.461(4)	2.467(5)	2.453(4)
N(7)-Dy(1)	2.463(4)	2.462(5)	2.459(4)
N(9)-Dy(1)	2.420(4)	2.454(5)	2.444(3)
N(10)-Dy(1)	2.429(4)	2.442(5)	2.446(4)
N(11)-Dy(1)	2.427(4)	2.444(5)	2.450(4)
N(12)-Dy(1)	2.438(4)	2.456(5)	2.439(4)
shortest Dy...Dy ions distance	8.928	8.937	9.077
N(1)-Dy(1)-N(10)	144.30(12)	137.92(17)	137.40(12)
N(1)-Dy(1)-N(11)	142.11(12)	148.44(17)	149.38(13)
N(10)-Dy(1)-N(11)	73.31(13)	72.84(17)	72.63(12)
N(1)-Dy(1)-N(3)	68.81(12)	69.98(17)	69.02(12)
N(10)-Dy(1)-N(3)	141.16(12)	148.07(17)	148.31(12)
N(11)-Dy(1)-N(3)	79.83(12)	83.67(17)	85.43(12)
N(1)-Dy(1)-N(9)	82.23(11)	77.64(17)	77.59(12)
N(10)-Dy(1)-N(9)	74.19(12)	73.51(17)	73.72(12)
N(11)-Dy(1)-N(9)	115.44(12)	114.30(17)	114.66(12)
N(3)-Dy(1)-N(9)	144.02(12)	137.34(17)	137.48(12)
N(1)-Dy(1)-N(12)	79.47(12)	83.52(17)	84.71(14)
N(10)-Dy(1)-N(12)	117.65(12)	115.08(17)	115.28(13)
N(11)-Dy(1)-N(12)	74.61(12)	73.62(17)	73.45(13)
N(3)-Dy(1)-N(12)	80.24(12)	77.03(17)	78.30(13)
N(9)-Dy(1)-N(12)	73.69(12)	72.33(17)	73.02(12)
N(1)-Dy(1)-N(7)	68.24(12)	69.03(17)	68.27(13)
N(10)-Dy(1)-N(7)	81.70(12)	78.84(17)	77.92(12)
N(11)-Dy(1)-N(7)	143.89(12)	137.81(17)	137.97(13)
N(3)-Dy(1)-N(7)	106.12(12)	106.80(17)	106.21(12)
N(9)-Dy(1)-N(7)	80.98(12)	85.83(17)	84.01(12)
N(12)-Dy(1)-N(7)	141.22(12)	148.13(17)	147.82(13)
N(1)-Dy(1)-N(5)	105.89(12)	107.10(16)	106.10(12)
N(10)-Dy(1)-N(5)	79.30(12)	84.80(17)	84.67(12)
N(11)-Dy(1)-N(5)	80.88(12)	77.77(17)	78.13(12)
N(3)-Dy(1)-N(5)	69.03(12)	69.02(17)	68.31(12)
N(9)-Dy(1)-N(5)	142.35(13)	149.62(18)	149.08(12)
N(12)-Dy(1)-N(5)	143.51(13)	137.45(18)	137.41(12)
N(7)-Dy(1)-N(5)	69.01(12)	68.99(17)	69.78(12)

**Table S3.** Analytical and mass spectroscopic data for the (**1a**, **2a**, and **2b**).

Molecular formula	$[M+H]^+$ ( <i>m/z</i> )	computed/found(%)		
		C	H	N
<b>2a</b> C <sub>96</sub> H <sub>80</sub> Cl <sub>4</sub> DyN <sub>12</sub> O <sub>4</sub> •CHCl <sub>3</sub>	1771.23	61.66/62.03	4.32/4.64	8.90/8.56
<b>2b</b> C <sub>96</sub> H <sub>81</sub> Cl <sub>4</sub> DyN <sub>12</sub> O <sub>4</sub> •CHCl <sub>3</sub>	1772.06	61.63/61.64	4.37/4.81	8.89/8.59