

Supplementary Information

N-Doped Porous TiO₂ with Rutile Phase and Visible Light Sensitive Photocatalytic Activity

Dong-Lin Shieh, Yi-Shiue Lin, Jia-Hau Yeh, Sian-Cong Chen, Bo-Chiuan Lin and Jong-Liang Lin*

Department of Chemistry, National Cheng Kung University, 1, Ta Hsueh Road, Tainan, Taiwan 701, ROC; Fax: 886-6-2740552; Tel: 886-6-2757575 ext. 65326; E-mail: jonglin@mail.ncku.edu.tw

Experimental section

1. Preparation of photocatalyst

The visible light active rutile powder was prepared by hydrothermal method, with an additional annealing process. A Teflon-lined stainless steel autoclave containing 0.7 g of TiN (99 %, Strem) and 8.0 ml of 5 M HNO₃ aqueous solution was kept at 100 °C for 6 h. The final N-doped rutile samples were obtained after washing with deionized water, followed by annealing at 200 °C for 2 h in air. This procedure for preparation of N-doped rutile from TiN has been repeated using TiC (99 %, Strem) and TiS₂ (99.8 %, Strem) as precursor, but only mixed phases (anatase and rutile) were obtained.

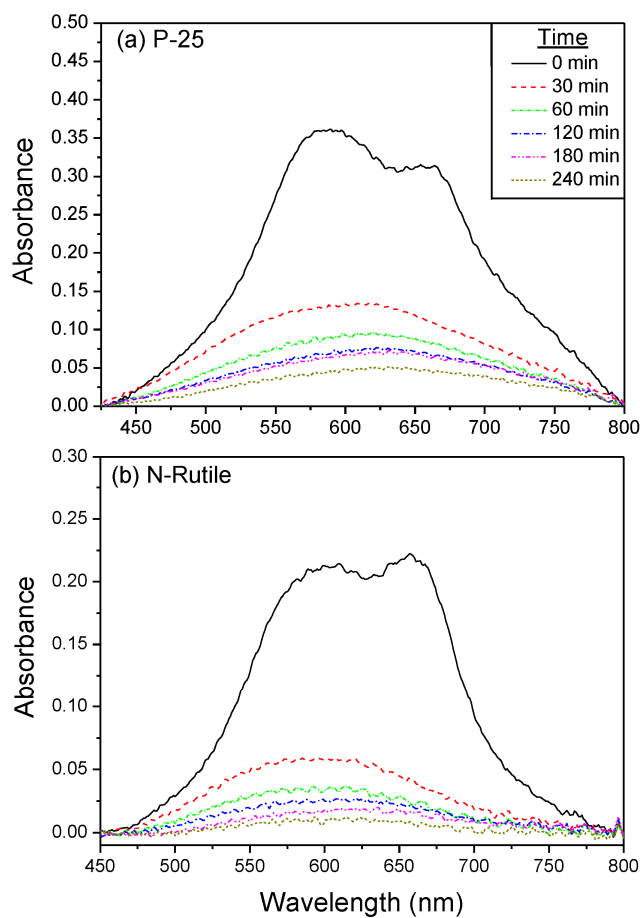
2. Characterization

The crystalline structure of the resulting TiO₂ was determined by X-ray diffraction using a Shimadzu XD-D1 diffractometer with Cu-K α radiation at 1.5406 Å. The nitrogen doping species of the rutile samples were examined by X-ray photoelectron spectroscopy using a Thermo VG-Scientific XPS spectrometer with an excitation beam of 1486.7 eV. The UV-Vis diffuse reflectance spectra were measured using a Hitachi U-3010 spectrophotometer. The BET area and pore volume of the N-doped TiO₂ were determined by N₂ physisorption using a Micrometrics ASAP 2010 instrument.

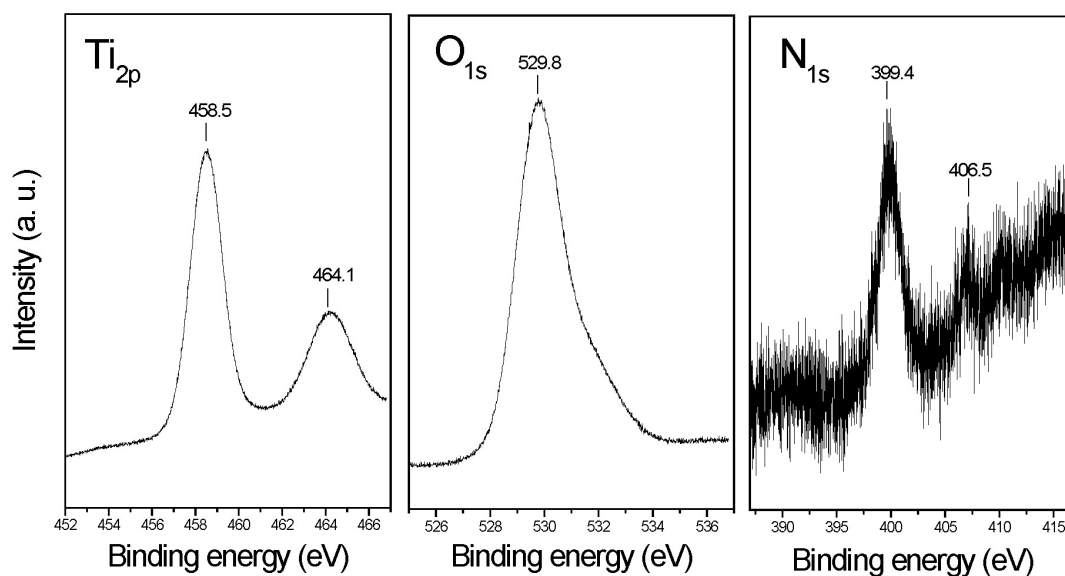
3. Photocatalytic activity measurement

In the study of methylene blue decolorization of air-TiO₂ systems, 0.5 g of TiO₂ powder was added in a 10 ml of MB aqueous solution of 500 ppm to create a submonolayer dye coverage of TiO₂ particle. The suspension was stirred and then kept in a 50 % humidity environment overnight in the dark. The MB-coated TiO₂ powder was pressed into an indented circular area ($\Phi=19.4$ mm, 0.5 mm in depth) in a

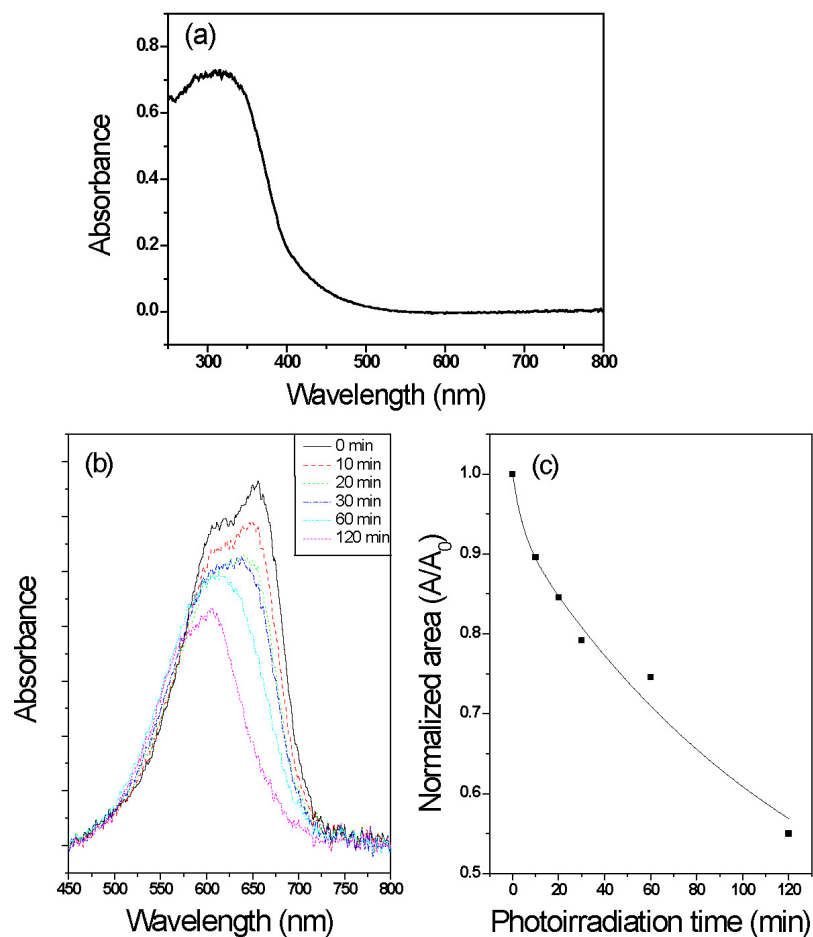
glass plate for subsequent photoirradiation and reflectance measurement. The MB coverages on Degussa-P25 ($40 \text{ m}^2 \text{ g}^{-1}$) and N-doped rutile ($77 \text{ m}^2 \text{ g}^{-1}$) samples were calculated to be 2.00×10^{12} and 1.04×10^{12} molecules/ cm^2 , respectively. An Oriel mercury lamp, with or without a filter, was used in the MB photocatalytic decolorization experiments. The light below 410 nm could be removed by the filter. The irradiation power on the TiO_2 sample at wavelengths $> 410 \text{ nm}$ was measured to be $\sim 15 \text{ mW cm}^{-2}$.



Supporting Figure 1 Change of the absorbance of methylene blue on (a) P-25 and (b) N-doped rutile with the time of photoirradiation using the full wavelength range from the light source.



Supporting Figure 2 Ti_{2p}, O_{1s} and N_{1s} X-ray photoelectron spectra of the mixed phases of anatase and rutile from TiC transformation.



Supporting Figure 3 (a) UV-Vis diffuse reflectance spectrum; (b) change of the absorption of methylene blue with photoirradiation time; (c) normalized MB peak area as a function of photoirradiation time of the mixed phases of TiO₂ (anatase + rutile) from TiC transformation.