

Supporting information for:

Desulfurization of N,N-dimethylthioformamide by hydrosilane with the help of an iron complex. Isolation and characterization of an iron-carbene complex as an intermediate of C=S double bond cleavage.

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1. General Remarks. NMR spectra (¹H, ¹³C, ²⁹Si) were recorded on a JEOL JNM-AL400 spectrometer. The residual peaks of the solvent were used as the reference for ¹H NMR spectra. For ¹³C NMR spectra, solvent signals were used as the chemical-shift reference. Peak positions of the ²⁹Si NMR spectra were referenced to external tetramethylsilane (δ = 0 ppm). IR spectra were recorded on a Perkin Elmer FTIR-Spectrum One.

2. Spectroscopic characterization

Me₂NC(S)Ph: Anal. Calc. for C₉H₁₁NS: C, 46.56; H, 8.79; N, 13.57. Found: C, 46.95; H, 8.85; N, 13.32.

Me₂NC(S)Me: Anal. Calc. for C₄H₉NS: C, 65.41; H, 6.71; N, 8.48. Found: C, 65.13; H, 6.73; N, 8.25.

Cp(CO)Fe(C(H)NMe₂)(SSiEt₃) (3): ¹H NMR (399.65 MHz, +20 °C, in C₆D₆): 0.92 (m, 6H, SiCH₂CH₃), 1.24 (m, 9H, SiCH₂CH₃), 2.42 (s, 3H, NCH₃), 3.20 (s, 3H NCH₃), 4.49 (s, 5H C₅H₅), 11.99 (s, 1H Fe=CH). ¹³C{¹H} NMR (100.40 MHz, +20 °C, in C₆D₆): 8.74 (s, SiCH₂CH₃), 9.59 (s, SiCH₂CH₃), 44.4 (s, NCH₃), 53.6 (s, NCH₃), 86.0 (s, C₅H₅), 223.77 (s, CO), 256.0 (s, Fe=C). ²⁹Si

NMR (79.30 MHz, +20 °C, in C₆D₆): 22.6 (s). IR (cm⁻¹, in toluene): ν (CO): 1932. Sufficient element analysis data could not be obtained because of its instability.

Cp(CO)Fe(C(H)NMe₂)(SSiPh₃) (5): ¹H NMR (399.65 MHz, +20 °C, in C₆D₆): 2.17 (s, 3H, NCH₃), 3.05 (s, 3H NCH₃), 4.19 (s, 5H C₅H₅), 7.21-8.03 (m, 15H, Ph), 11.74 (s, 1H Fe=CH). ¹³C{¹H} NMR (100.40 MHz, +20 °C, in C₆D₆): 44.4 (s, NCH₃), 53.4 (s, NCH₃), 85.7 (s, C₅H₅), 127.62 (s, Ph), 128.8 (s, Ph), 136.5 (s, Ph), 140.8 (s, Ph), 222.9 (s, CO), 255.8 (s, Fe=C). ²⁹Si NMR (79.30 MHz, +20 °C, in C₆D₆): 1.8 (s). IR (cm⁻¹, in toluene): ν (CO): 1937. Anal. Calc. for C₂₇H₂₇FeNO₂Si: C, 65.18; H, 5.47; N, 2.82. Found: C, 65.04; H, 5.41; N, 2.75.

3. X-ray Crystal structure determination of 5: Dark-red crystals of **5** suitable for an X-ray diffraction study were obtained through crystallization from CH₂Cl₂/hexane. The single crystal was mounted in a glass capillary. Data for **5** were collected at 20 °C on Rigaku/MSC Mercury CCD area-detector diffractometer equipped with monochromated MoKα radiation ($\lambda = 0.71070 \text{ \AA}$). Calculations for **5** were performed with the teXane crystallographic software package of Molecular Structure Corporation. Crystal Data: C₂₂H₂₇FeNO₂Si, $M = 497.51$, red block, $0.45 \times 0.30 \times 0.30 \text{ mm}^3$, monoclinic, space group P2₁/c (No. 14), $a = 16.037(2) \text{ \AA}$, $b = 9.3525(11) \text{ \AA}$, $c = 16.402(3) \text{ \AA}$, $\beta = 101.572(3)^\circ$, $V = 2410.0(5) \text{ \AA}^3$, $Z = 4$, $\mu(\text{MoK}\alpha) = 7.813 \text{ mm}^{-1}$, $D_{\text{calc}} = 1.371 \text{ g/cm}^3$, 16617 reflections collected, 5411 ($I > 3\sigma I$) unique reflections were used in all calculations, number of variables = 373, $R = 0.0328$, $R_w = 0.1051$, and goodness of fit = 1.007.

Table S1 Crystal data and structure refinement for **5**.

5	
Empirical formula	C ₂₂ H ₂₇ FeNOSSi
Formula weight	497.51
Crystal system	Monoclinic
Crystal size(mm ³)	0.45 × 0.30 × 0.30
Space group	P2 ₁ /c (No. 14)
<i>a</i> , Å	16.037(2)
<i>b</i> , Å	9.3525(11)
<i>c</i> , Å	16.402(3)
β , deg	101.572(3)
<i>V</i> , Å ³	2410.0(5)
Z	4
μ , mm ⁻¹	7.813
<i>F</i> (000)	1040.00
<i>D</i> _{calcd} , g cm ⁻³	1.371
No. of unique reflections	5411
No. of used reflections	16617
No. of variables	373
^a <i>R</i>	0.0328
^b <i>R</i> _w	0.1051
<i>Goodness-of-fit</i>	1.007

^a*R* = $\Sigma|Fo| - |Fc| / \Sigma|Fo|$.

^b*R*_w = $[\sum w(|Fo| - |Fc|)^2 / \sum w|Fo|^2]^{0.5}$

Weighting scheme [$\sigma(F_o)^2$]⁻¹

Table S2 Selected bond distances (\AA), bond angles ($^\circ$), and torsion angles ($^\circ$) for **5**.

Fe-S	2.3371(3)
Fe-C(3)	1.8891(5)
S-Si	2.1016(5)
N-C(1)	1.466(2)
N-C(2)	1.473(2)
N-C(3)	1.306(2)
S-Fe-C(3)	96.42(5)
C(3)-Fe-C(4)	97.34(7)
S-Fe-C(4)	87.39(5)
Fe-S-Si	115.482(19)
Fe-C(3)-N	135.96(12)
C(1)-N-C(2)	113.75(13)
C(1)-N-C(3)	123.56(13)
C(2)-N-C(3)	122.54(14)
C(1)-N-C(3)-Fe	-6.2(2)
C(2)-N-C(3)-Fe	169.06(12)