

## **Supplementary Information**

### **The Crystal Structure of L-Arginine**

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## Solid-State $^{13}\text{C}$ NMR Spectroscopy

Solid-state  $^{13}\text{C}$  NMR data was acquired on a Chemagnetics Infinity Plus spectrometer, operating at a  $^{13}\text{C}$  Larmor frequency of 75.48 MHz and with magic-angle spinning at 12 kHz. The spectra were acquired with ramped  $^1\text{H} \rightarrow ^{13}\text{C}$  cross-polarization<sup>1</sup> using a contact time of 2 ms, TPPM  $^1\text{H}$  decoupling<sup>2</sup> during acquisition, and a recycle delay of 3 s. The FID was acquired over 17.2 hours with 20480 acquisitions co-added.

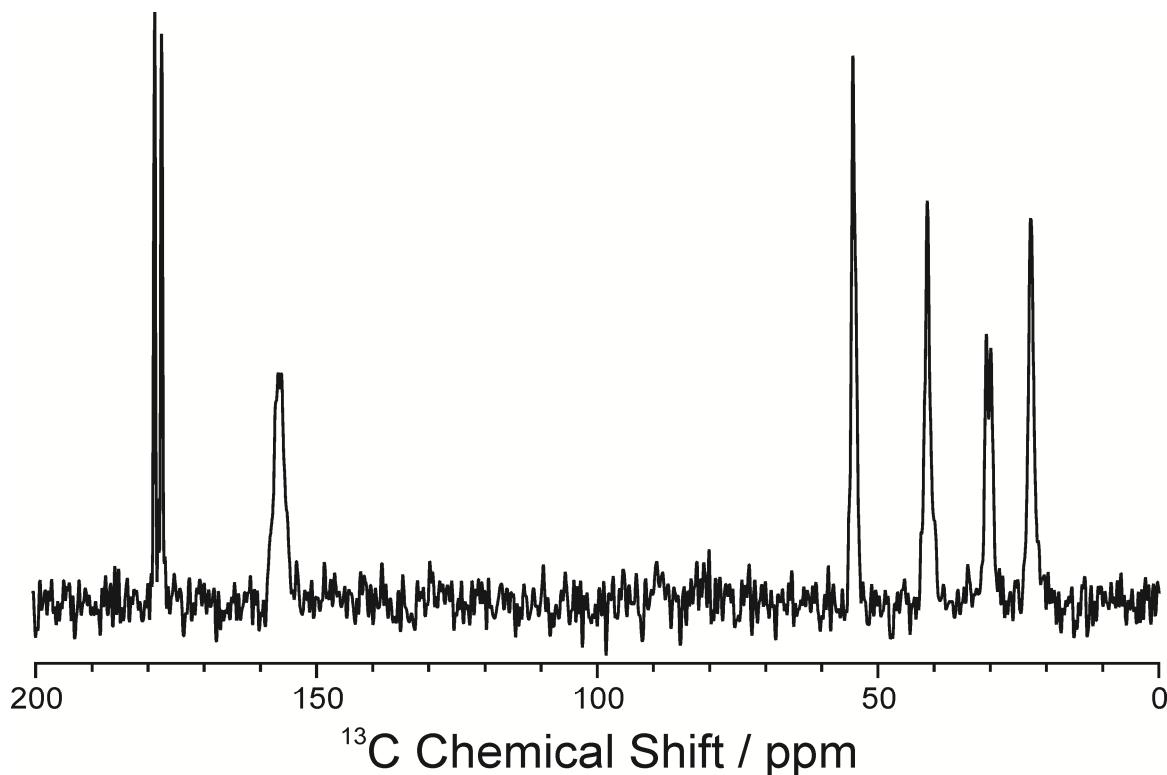


Figure S1 High-resolution solid-state  $^{13}\text{C}$  NMR spectrum of L-arginine. For the carboxylate (~178 ppm) and  $\text{C}_\beta$  (~30 ppm) environments, two peaks (in approximately 1:1 intensity ratio) are observed, suggesting that there are two crystallographically independent molecules in the asymmetric unit.

## Comparison of Crystal Structures of L-Arginine and L-Arginine Dihydrate

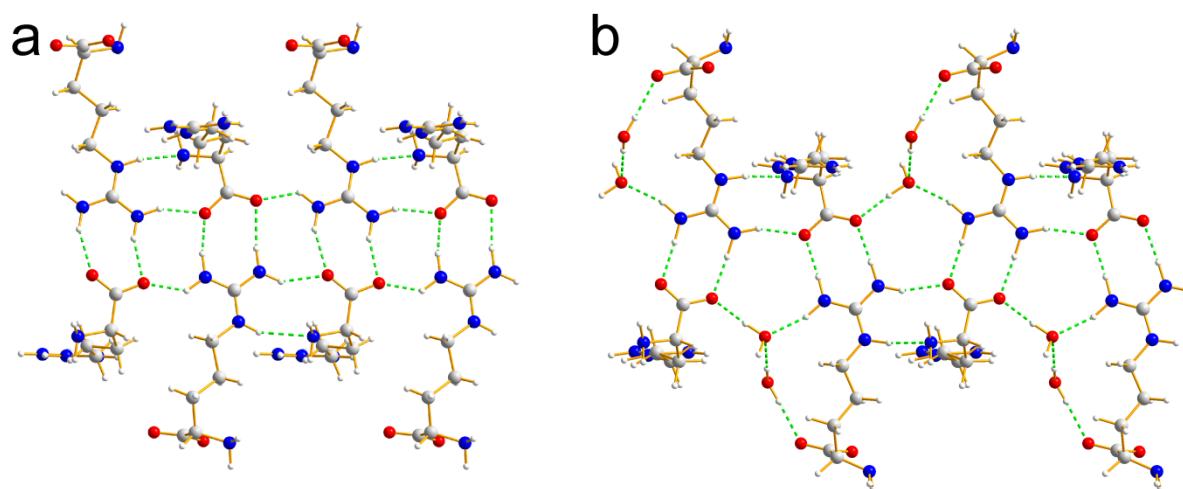


Figure S2 The hydrogen-bonded ribbons in (a) L-arginine viewed perpendicular to the (041) plane and (b) L-arginine dihydrate<sup>3</sup> viewed perpendicular to the (301) plane.

## References

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2. A. E. Bennett, C. M. Rienstra, M. Auger, K. V. Lakshmi and R. G. Griffin, 1995, **103**, 6951-6958.
3. M. S. Lehmann, J. J. Verbist, W. C. Hamilton and T. F. Koetzle, *J. Chem. Soc. Perk. Trans. 2*, 1973, 133-137.