## **Supplementary Materials**

Induced Preference for Axial Chirality in a Triarylmethylium o,o-Dimer upon Complexation with Natural  $\gamma$ -Cyclodextrin: Strong ECD Signaling and Fixation of Supramolecular Chirality to Molecular Chirality

Takanori Suzuki,\* Hitomi Tamaoki, Kazuhisa Wada, Ryo Katoono, Tatsuo Nehira, Hidetoshi Kawai, and Kenshu Fujiwara

-----

## **Experimental Details:**

• Preparation of *rac*-5,7-dihydro-5,5,7,7-tetrakis(dimethylaminophenyl)dibenzo[*c*,*e*]thiepin 2



To an aqueous solution of  $1a^{2+}(BF_4)_2$  (20.3 mg, 24.4 Me µmol) (10 mL) was added an aqueous solution of Na<sub>2</sub>S·9H<sub>2</sub>O (290 mg, 1.21 mmol) (2 mL), and the mixture was stirred for 30 min. Me The resulting suspension was extracted with AcOEt (10 mL × 3), and the combined organic layer was washed with water (10 mL × 3) and brine. After drying over Na<sub>2</sub>SO<sub>4</sub>, the solvent was removed

to give pure 2 (16.9 mg) as a white solid in quantitative yield.

decomp. 228-235 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.99 (d, J = 8.8 Hz, 2H), 7.30-7.05 (br, 4H), 7.00 (dd, J = 7.7, 1.4 Hz, 2H), 6.88 (dt, J = 7.7, 1.7 Hz, 2H), 6.81 (dt, J = 7.3, 1.7 Hz, 2H), 6.61 (d, J = 9.0 Hz, 4H), 6.46 (d, J = 8.8 Hz, 2H), 6.39 (dd, J = 7.3, 1.4 Hz, 2H), 6.35 (s, 4H), 2.90 (s, 12H), 2.83 (s, 12H) ppm; IR (KBr)  $\tilde{v}$  3037, 2880, 2794, 1608, 1559, 1513, 1474, 1439, 1352, 1208, 1183, 1160, 1142, 1058, 1004, 948, 895, 811, 776, 749, 714, 578, 567, 535 cm<sup>-1</sup>; UV-Vis (CH<sub>2</sub>Cl<sub>2</sub>):  $\lambda_{max}$  / nm ( $\epsilon$  / M<sup>-1</sup>cm<sup>-1</sup>) 274 (60200), (MeCN) : 272 (64000); LR-MS (FD) m/z = 690 (20), 689 (55), 688 (M<sup>+</sup>, bp), 656 (11), 537 (16), 536 (36), 329 (6), 328 (M<sup>2+</sup>, 11); Anal. Calcd. for C<sub>46</sub>H<sub>48</sub>N<sub>4</sub>S·0.5AcOEt: C 78.65, H 7.15, N 7.64. Found: C 78.67, H 7.85, N 7.50. • Optical resolution of **2** by HPLC

A solution of *rac*-2 in CHCl<sub>3</sub> (0.05 mL) was diluted with a solvent mixture (0.95 mL, hexane/AcOEt = 10). The whole mixture was charged on a column (Chiralpak IA) and eluted with hexane/AcOEt = 10. After 3 cycles, the enantiomers appeared as two peaks, and (*P*)-2 was collected as the first fraction: ECD (MeCN)  $\lambda_{ext}$  / nm ( $\Delta\epsilon$ / M<sup>-1</sup>cm<sup>-1</sup>) 255 (-37), 275 (+11), and 309 (-9.1). The stereochemical assignment was carried out based on the DFT calculation.

• Preparation of optically active 5,7-dihydro-5,5,7,7-tetrakis(dimethylaminophenyl) dibenzo[*c*,*e*]thiepin (*M*)-2 through treatment of  $1a^{2+}$  with Na<sub>2</sub>S in the presence of  $\gamma$ -CyD

To a solution of  $1a^{2+}(BF_4)_2$  (20.4 mg, 24.6 µmol) in H<sub>2</sub>O (4 mL) was added  $\gamma$ -CyD (64.7 mg, 49.9 µmol) at 24 °C, and the mixture was stirred for 75 min. Then the mixture was cooled down to 0 °C, and further stirred for 30 min. To the solution was added Na<sub>2</sub>S·9H<sub>2</sub>O (110 mg, 459 µmol) at 0 °C, and the mixture was stirred for 3.5 h. The resulting suspension was allowed to warm up to 25 °C, and stirred for 16 h. After addition of 5 mL of saturated NaHCO<sub>3</sub> aq., the solution was cooled to 0 °C. The mixture was extracted with precooled AcOEt, and the organic layer was washed with precooled water and brine. The organic layer was cooled to -78 °C. To precooled MeCN in a cuvette were added a few drops of the above prepared solution of 2, and the mixture was subjected to the measurement. The concentration was determined on the basis of UV-Vis spectrometry. According to the ECD spectrum [ $\lambda_{ext}$  / nm ( $\Delta\epsilon$ / M<sup>-1</sup>cm<sup>-1</sup>) 255 (+9.3)], optical purity of the sample was determined to be 25%ee.

Electronic Supplementary Material (ESI) for Chemical Communications This journal is The Royal Society of Chemistry 2012

**Supplementary Figures:** 





**Fig. S1** Two views of molecular geometry determined by X-ray analysis on  $rac-1a^{2+} \cdot (SbCl_6)_2$  (ref. 6a). The counter anions are omitted for clarity. The dication has a rough dimension of W 15 × D 7 × H 10 Å.



**Fig. S2** (a) Changes in the UV-Vis spectrum of  $1a^{2+} (BF_4)_2 (3.4 \times 10^{-6} \text{ M})$  in H<sub>2</sub>O upon addition of  $\gamma$ -CyD (0.2 - 6.0 equiv.) at 25 °C. (b) The isotherm plot of  $\Delta$ Abs at 662 nm.



**Fig. S3** Job plot for complexation between  $\mathbf{1a}^{2+} \cdot (BF_4^{-})_2$  and  $\gamma$ -CyD in H<sub>2</sub>O monitored by ECD.  $\chi = \text{mol fraction of } \mathbf{1a}^{2+} \cdot (BF_4^{-})_2$ ,  $[\mathbf{1a}^{2+} \cdot (BF_4^{-})_2] / [\mathbf{1a}^{2+} \cdot (BF_4^{-})_2] + [\gamma$ -CyD].



**Fig. S4** Changes in the (a) UV-Vis and (b) ECD spectra of MG salt  $3 \cdot (BF_4^-)$  ( $1.9 \times 10^{-5}$  M) in H<sub>2</sub>O upon addition of  $\gamma$ -CyD (100 - 500 equiv.) at 25 °C.



**Fig. S5** (a) <sup>1</sup>H NMR spectrum (600 MHz) of  $1a^{2+} (BF_4)_2 (1.0 \times 10^{-3} \text{ M})$  in the presence of  $\gamma$ -CyD (1.0 × 10<sup>-3</sup> M), and (b) its partial region, measured in D<sub>2</sub>O at 65 °C.



**Fig. S6** Three partial regions of the ROESY spectrum (600 MHz) of  $1a^{2+} (BF_4)_2 (3.3 \times 10^{-4} \text{ M})$  in the presence of  $\gamma$ -CyD (3.3 × 10<sup>-4</sup> M), measured in D<sub>2</sub>O at 65 °C.



**Fig. S7** Changes in the ECD spectrum of  $1a^{2+} (BF_4^{-})_2 (1.0 \times 10^{-5} \text{ M})$  upon addition of  $\gamma$ -CyD (10 - 40 equiv.), measured in H<sub>2</sub>O at 65 °C. The orange (30 equiv.) and red (40 equiv.) lines are overlapped to indicate that  $\gamma$ -CyD (4.0  $\times 10^{-4}$  M) was enough to completely bias the equilibrium between free  $1a^{2+}$  and  $1a^{2+}@\gamma$ -CyD in favor of the complexed state.



**Fig. S8** ECD spectra of optically resolved **2** (1st fraction, *P*-helicity) (red line) and as-prepared **2** from  $1a^{2+}$  and Na<sub>2</sub>S in the presence of  $\gamma$ -CyD (25%ee, *M*-helicity) (blue line), measured in MeCN.



**Fig. S9** Calculated CD spectrum of (*P*)-2 by TD-DFT method (B3LYP/cc-pVDZ) including a PCM solvent effect. The molecular geometry was obtained after optimization by the DFT method (B3LYP/6-31G(d)) with a PCM solvent effect. The resultant rotational strengths were converted into a single CD curve by using a sum of Gaussian functions with a  $\sigma$  value of 1800 cm<sup>-1</sup>.