S 1

Contrasting Reductions of Group 14 Metal(II) Chloride Complexes: Synthesis of a β -

Diketiminato Tin(I) Dimer

Sam L. Choong, Christian Schenk, Andreas Stasch, Deepak Dange and Cameron Jones*

SUPPLEMENTARY INFORMATION

| Contents | 1. Syntheses | S1 |
|----------|--------------------------|------------|
| | 2. X-Ray Crystallograpy | S4 |
| | 3. Computational Studies | S9 |
| | 4. References | S15 |

1. Syntheses

General considerations. All manipulations were carried out using standard Schlenk and glove box techniques under an atmosphere of high purity dinitrogen. Hexane, toluene and THF were distilled over molten potassium metal, while diethyl ether was distilled over Na/K (50:50) alloy. Melting points were determined in sealed glass capillaries under dinitrogen and are uncorrected. Mass spectra were recorded at the EPSRC National Mass Spectrometric Service at Swansea University. Microanalyses were carried out the Science Centre, London Metropolitan University. IR spectra were recorded using a Nicolet 510 FT-IR spectrometer as Nujol mulls between NaCl plates. ¹H, $^{13}C{^{1}H}$ and $^{119}Sn{^{1}H}$ NMR spectra were recorded on either Bruker DXP300 or DRX400 spectrometers and were referenced to the residual solvent resonances (${}^{1}H$ and ${}^{13}C{{}^{1}H}$), or external ^{But}NacnacH,¹ [(^{But}MesNacnac)GeCl]² $(^{119}\text{Sn}\{^{1}\text{H}\}).$ ^{But}MesNacnacH,¹ SnMe₄ and $[{(^{Mes}Nacnac)Mg}_2]^3$ were prepared by variations of literature procedures. All other reagents were used as received.

Synthesis of [(^{But}MesNacnac)SnCl] 5: A 1.6 M solution of *n*-butyllithium in hexane (1.6 cm³, 2.56 mmol) was added to a solution of ^{But}MesNacnacH (0.99 g, 2.36 mmol) in diethyl ether (15 cm³) at - 78 °C. The resultant solution was warmed to room temperature and was stirred for 1 hr before being cooled again to -78 °C. The solution was then treated with suspension of SnCl₂ (0.45g, 2.37 mmol) in diethyl ether (15 cm³), whereupon the reaction solution was warmed to ambient temperature. Volatiles were then removed *in vacuo*, the residue extracted with hexane (20 cm³), and the extract filtered. The filtrate was concentrated to *ca*. 10 cm³ and stored at overnight at -30 °C yielding colourless needle-like crystals of **5** (yield 0.93g, 69%). M.p. 196 °C (sharp); ¹H NMR (300 MHz, C₆D₆, 298 K): δ 1.09 (s, 18H, C(CH₃)₃), 2.05 (s, 12H, *o*-CH₃), 2.63 (s, 6H, *p*-CH₃), 5.99 (s, 1H,

NCC*H*CN), 6.64 (s, 4H, Ar-*H*); ¹³C{¹H} NMR (400 MHz, C₆D₆, 300K): δ 20.5 (*o*-*C*H₃), 21.0 (*p*-*C*H₃), 31.3 (C(*C*H₃)₃), 42.3 (*C*(CH₃)₃), 103.7 (NC*C*HCN), 128.9, 130.1, 134.4, 143.8 (Ar-*C*), 173.7 (NCCH); ¹¹⁹Sn{¹H} NMR (149.2 MHz, C₆D₆, 300K): δ - 235.5; IR *v*/cm⁻¹ (Nujol): 1377s, 1260s, 1093s, 859w, 722w; MS (EI 70eV), *m*/*z* (%): 572.2 (M⁺, 21), 417.3 (^{But}MesNacnac⁺, 36), 202.1 (MesNCBu^{t+}, 100); anal. calc. for C₂₉H₄₁ClSnN₂: C 60.91%, H 7.23%, N 4.90%; found: C 60.18%, H 7.14%, N 5.04%.

Synthesis of [(^{But}MesNacnac)PbCl] 6: ^{*n*Bu}MesNacnacH (0.79 g, 1.88 mmol) was dissolved in THF (20 cm³) and treated with a 1.6 M solution of *n*-butyllithium in hexane (1.25 cm³, 2.00 mmol) at -80 °C. This mixture was subsequently warmed to ambient temperature and stirred for 2 hours, before being added to a suspension of PbCl₂ (0.523 g, 1.88 mmol) in THF (20 cm³) at -80 °C. Upon warming to room temperature, the solution took on a bright yellow colour. All volatiles were subsequently removed *in vacuo*, and the residue was extracted into hexane (15 cm³). The extract was filtered and the filtrate stored at -30 °C overnight to yield yellow crystals of **6** (yield 0.46 g, 42%). M.p. 209-212 °C (decomp.); ¹H NMR (300 MHz, C₆D₆, 298 K): δ 1.14 (s, 18H, C(CH₃)₃), 2.08 (s, 6H, *o*-CH₃), 2.12 (s, 6H, *o*-CH₃), 2.65 (s, 6H, *p*-CH₃), 5.71 (s, 1H, NCCHCN), 6.70 (s, 4H, Ar-H); ¹³C{¹H} NMR (400 MHz, C₆D₆, 300K): δ 19.6 (*o*-CH₃), 20.6 (*o*-CH₃), 20.9 (*p*-CH₃), 31.8 (C(CH₃)₃), 43.6 (C(CH₃)₃), 107.7 (NCCH), 130.3, 132.2, 133.9, 145.1 (Ar-*C*), 173.0 (NCCH); IR *v*/cm⁻¹ (Nujol): 1623w, 1376s, 1260s, 800s, 721m, 668w; MS (EI 70eV), *m/z* (%): 660.2 (M⁺, 10), 625.3 (M⁺-Cl, 8), 417.3 (^{But}MesNacnac⁺, 90), 202.1 (MesNCBu¹ ⁺, 100); anal. calc. for C₂₉H₄₁N₂PbCl: C 52.75%, H 6.26%, N 4.24%; found: C 52.59%, H 6.18%, N 4.14%.

Synthesis of [{(^{But}MesNacnac)Sn}₂] 7: A solution of [{(^{Mes}Nacnac)Mg}₂] (0.13 g, 0.18 mmol) in toluene (20 cm³) was added to a solution of [(^{But}MesNacnac)SnCl] (0.20 g, 0.35 mmol) in toluene (20 cm³) at -78 °C over 5 mins. This led to an immediate colour change of the reaction solution to deep green. This solution was warmed to 0 °C over 4 hrs, and stirred at that temperature for a further 2 hrs, before all volatiles were removed *in vacuo*. The residue was extracted with cold hexane (10 cm³) and the extract filtered. The filtrate was stored at -30 °C overnight, affording small dark green crystals of **7** (yield 0.11 g, 55 %). M.p. 43-45 °C (decomp.); ¹H NMR (300 MHz, C₆D₆, 298 K): δ 1.17 (s, 36H, C(CH₃)₃), 2.07 (s, 12H, *o*-CH₃), 2.35 (s, 12H, *o*-CH₃), 2.44 (s, 12H, *p*-CH₃), 5.91 (s, 2H, NCCHCN), 6.67 (s, 8H, ArH); ¹³C{¹H} NMR (400 MHz, C₆D₆, 300K): δ 21.9, 21.1 (*o*-CH₃ and *p*-CH₃), 32.0 (C(CH₃)₃), 42.7 (C(CH₃)₃), 105.0 (NCCHCN), 129.5, 129.6, 132.9, 143.0 (Ar-*C*), 177.6 (N*C*(Bu¹)CH); ¹¹⁹Sn{¹H} NMR (149.2 MHz, C₆D₆, 300K): δ 502.1; IR *v*/cm⁻¹ (Nujol): 1377s, 1260s, 866w, 722w, 683w, 666w; MS (EI 70eV), *m/z* (%): 1072.5 (M⁺, 4), 537.2

(($^{But}MesNacnac$)Sn⁺, 100), 202.1 (MesNCBu^{t +}, 81); anal. calc. for C₅₈H₈₂N₄Sn₂ C 64.94%, H 7.70%, N 5.22%; found C 64.83%, H 7.72%, N 5.28%.

Synthesis of $[{C(Bu^{t})C(H)C(Bu^{t})N(Mes)}]Ge(\mu-NMes)Mg(^{Mes}Nacnac)]$ 8: A solution of [{(^{Mes}Nacnac)Mg}₂] (0.34 g, 0.48 mmol) in toluene (20 cm³) was added over 5 min to a solution of [(^{Bur}MesNacnac)GeCl] (0.25 g, 0.48 mmol) in toluene (20 cm³) at -78 °C. Upon addition, the reaction solution immediately turned to a wine-red colour. The mixture was warmed slowly to 0 °C and stirred at that temperature for 2 hrs. All volatiles were subsequently removed in vacuo, the residue extracted with cold hexane (10 cm³), and the extract filtered. Storage of the filtrate at -30 °C overnight yielded small red crystals of 8 (yield 0.26 g, 65 %). M.p. 135-140 °C (decomp.); ¹H NMR (300 MHz, C₆D₆, 298 K): δ 0.84 (s, 9H, C(CH₃)₃), 1.47 (s, 9H, C(CH₃)₃), 1.55 (s, 6H, NC(CH₃)), 1.58 (s, 6H, Ar-CH₃), 1.85 (s, 6H, Ar-CH₃), 2.06 (s, 12H, o-CH₃), 2.19 (s, 3H, p-CH₃), 2.22 (s, 6H, Ar-CH₃), 2.30 (s, 3H, p-CH₃), 4.85 (s, 1H, NCCHCN), 6.60 (s, 2H, Ar-H), 6.64 (s, 2H, Ar-H), 6.76 (s, 4H, Ar-H), 6.97 (s, 1H, NC(Bu^t)HC(Bu^t)); ${}^{13}C{}^{1}H{}$ NMR (400 MHz, C₆D₆, 300K): δ 18.8 (NCCH₃), 19.0, 21.0, 21.1, 21.7, 23.3, 28.9 (Ar-CH₃), 29.5, 33.3 (C(CH₃)₃), 38.8, 40.4 (C(CH₃)₃), 95.3 (NCCCN), 123.7 (NCC), 125.1, 128.5, 128.6, 129.0, 129.5, 130.3, 131.1, 132.5, 133.4, 134.2, 143.1, 144.9 (Ar-C), 154.3 (NCC), 169.5 (MgN(Mes)C), GeCC not observed; IR v/cm⁻¹ (Nujol): 1377s, 1260s, 854w, 722w, 684w; MS (EI 70eV), m/z (%): 334.3 (MesNacnacH⁺, 80), 286.2 (MesNC(Bu^t)C(H)C(Bu^t)⁺, 100). Due to the extremely air sensitive nature of this compound, reproducible microanalyses could not be obtained.

Synthesis of [(^{But}MesNacnac)₂Pb] and [{(^{Mes}Nacnac)MgCl}₂{(^{But}MesNacnac)PbCl}₂]: A solution of [{(^{Mes}Nacnac)Mg}₂] (0.163 g, 0.23 mmol) dissolved in 20 cm³ of toluene was added to a solution of [(^{But}MesNacnac)PbCl] (0.300 g, 0.45 mmol) in toluene (20 cm³) at -80 °C, yielding a dark green solution. This was subsequently warmed to 0 °C, with the deposition of lead metal at *ca.* -30 °C, to give a red-orange solution. Volatiles were then removed from the reaction solution *in vacuo*, the residue extracted with cold hexane, and the extract filtered. Placing the filtrate at -30 °C overnight afforded yellow crystals of [{(^{Mes}Nacnac)MgCl}₂{(^{But}MesNacnac)PbCl}₂] (yield 0.057 g, 12 % based on [(^{But}MesNacnac)PbCl]). Placing the mother liquor at -80 °C for 2 weeks gave orange crystals of [(^{But}MesNacnac)PbCl]). Placing the mother liquor at -80 °C for 2 weeks gave orange crystals of [(^{But}MesNacnac)₂Pb] (yield 0.032 g, 7% based on [(^{But}MesNacnac)PbCl]). Data for [(^{But}MesNacnac)₂Pb]: M.p. 116-118 °C (decomp); ¹H NMR (300 MHz, C₆D₆, 298 K): δ 1.08 (s, 36H, C(CH₃)₃), 2.20 (s, 12H, *p*-CH₃), 2.23 (s, 24H, *o*-CH₃), 5.52 (s, 2H, NCCHCN), 6.64 (s, 8H, Ar-H); ¹³C ¹H} NMR (400 MHz, C₆D₆, 300K): δ 20.5, 20.7 (*o*-CH₃ and *p*-CH₃), 31.5 (C(CH₃)₃), 44.6 (C(CH₃)₃), 95.6 (NCCHCN), 128.9, 130.9, 132.1, 146.0 (Ar-C), 171.7 (NCCH); IR *v*/cm⁻¹ (Nujol): 1614m, 1376s, 1260s, 1113m, 852w, 800m; MS (EI 70eV), *m/z* (%): 1042.2 (M⁺, 6), 625.3

 $(M^{+}-^{But}MesNacnac, 10)$, 417.3 ($^{But}MesNacnac^{+}$, 70), 202.1 (MesNCBu^{t+}, 100). Data for $[{(^{Mes}Nacnac)MgCl}_2{(^{But}MesNacnac)PbCl}_2]$: M.p. 170-174 °C (decomp.); Solution NMR spectroscopic data for the compound could not be reliably assigned due the poor solubility of the crystalline compound in non-coordinating solvents and the broadness and complexity of the signals in its NMR spectra. IR ν/cm^{-1} (Nujol): 1376s, 1260s, 799s, 721w, 670w, 667w; MS (EI 70eV), m/z (%): 417.3 ($^{But}MesNacnac^{+}$, 25), 334.2 ($^{Mes}NacnacH^{+}$, 32).

Synthesis of [(^{But}Nacnac)PbCl]: A 1.6 M solution of *n*-butyllithium in hexane (1.6 cm³, 2.60 mmol) was added to a solution of ^{But}NacnacH (1.19 g, 2.38 mmol) in THF (20 cm³) at -80 °C, and the reaction mixture subsequently warmed to room temperature. This solution was then slowly added to a suspension of PbCl₂ (0.662 g, 2.38 mmol) in THF (20 cm³) at -80 °C. The resultant mixture was warmed to room temperature slowly and stirred overnight. All volatiles were subsequently removed *in vacuo*, and the yellow residue was extracted into hexane (10 cm³). The extract was filtered, and the filtrate left to stand at room temperature for 2 days to give yellow crystals of [(^{But}Nacnac)PbCl] (yield 0.52 g, 30%); M.p. 218-224 °C (decomp.); ¹H NMR (400 MHz, C_6D_6 , 298 K) 0.98 (d, 6H, ${}^{3}J_{HH}$ = 6.8 Hz, CH(CH₃)₂), 1.21 (s, 18H, C(CH₃)₃), 1.22 (d, 6H, ${}^{3}J_{HH}$ = 6.8 Hz, CH(CH₃)₂), 1.32 (d, 6H, ${}^{3}J_{HH}$ = 6.8 Hz, CH(CH₃)₂), 1.46 (d, 6H, ${}^{3}J_{HH}$ = 6.8 Hz, CH(CH₃)₂), 3.13 (sept, 2H, ${}^{3}J_{HH}$ = 6.8 Hz, CH(CH₃)₂), 4.07 (sept, 2H, ${}^{3}J_{HH}$ = 6.8 Hz, CH(CH₃)₂), 5.89 (s, 1H, NCCHCN), 6.94-7.04 (m, 6H, Ar-H); ¹³C NMR (400 MHz, C₆D₆, 300K): δ 23.8 (CH(CH₃)₂), 24.2 (CH(CH₃)₂), 26.8 (CH(CH₃)₂), 27.4 (CH(CH₃)₂), 27.8 (CH(CH₃)₂), 29.5 (CH(CH₃)₂), 32.4 (C(CH₃)₃), 43.8 (C(CH₃)₃), 108.3 (NCCHCN), 123.7 (Ar-C), 125.0 (Ar-C), 126.0 (Ar-C), 143.0 (Ar-C), 143.8 (Ar-C), 143.9 (Ar-C), 173.2 (NCCH); IR v/cm⁻¹ (Nujol): 1619w, 1376s, 1260s, 799s, 721m; MS (EI 70eV), m/z (%): 744.3 (M⁺, 23), 244.1 (DipNCCBu^tH⁺, 100); anal. calc. for C₃₅H₅₃N₂PbCl: C 56.47%, H 7.18%, N 3.76%; found: C 56.58%, H 7.23%, N 3.79%. N.B. The reduction of $[(^{But}Nacnac)PbCl]$ with half an equivalent of $[\{(^{Mes}Nacnac)Mg\}_2]$ in toluene

at -80 °C led to the deposition of lead metal upon warming to room temperature, and a complex mixture of unidentified products, as determined by an ¹H NMR spectroscopic analysis of the total reaction solution.

2. X-Ray Crystallography

Crystals of $7 \cdot (\text{hexane})_{0.5}$, **8**, $[(^{\text{But}}\text{MesNacnac})_2\text{Pb}] \cdot (\text{hexane})$, $1S \cdot (\text{hexane})$, $[\{(^{\text{Mes}}\text{Nacnac})\text{MgCl}\}_2\{(^{\text{But}}\text{MesNacnac})\text{PbCl}\}_2] \cdot (\text{hexane})$, $2S \cdot (\text{hexane})$, and $[(^{\text{But}}\text{Nacnac})\text{PbCl}]$ **3S** suitable for X-ray structural determination were mounted in silicone oil. Crystallographic measurements for **7** and **8** were made using the MX1 beamline of the Australian Synchrotron ($\lambda = 0.71069 \text{ Å}$). The software package Blu-Ice⁴ was used for data acquisition, while the program XDS⁵

was employed for data reduction. Crystallographic measurements for **1S**, **2S** and **3S** were made using an Oxford Gemini Ultra diffractometer using a graphite monochromator with Mo K α radiation ($\lambda = 0.71073$ Å). All structures were solved by direct methods and refined on F² by full matrix least squares (SHELX97)⁶ using all unique data. All non-hydrogen atoms are anisotropic with hydrogen atoms included in calculated positions (riding model). Crystal data, details of data collections and refinements for all structures can be found in their CIF files and are summarized in Table S1.

Table S1. Summary of Crystallographic Data for Compounds 7. (hexane)_{0.5}, 8, 1S. (hexane), 2S. (hexane) and 3S.

| | $7 \cdot (\text{hexane})_{0.5}$ | 8 | 1S·(hexane) | $2S \cdot (hexane)$ | 38 |
|--------------------------------------|---------------------------------|--------------|---------------------|---------------------------------|-----------------------|
| empirical | $C_{61}H_{89}N_4Sn_2$ | C52H70GeMgN4 | $C_{64}H_{96}N_4Pb$ | $C_{110}H_{154}Cl_4Mg_2N_8Pb_2$ | $C_{35}H_{53}ClN_2Pb$ |
| formula | | | | | |
| formula weight | 1115.74 | 848.02 | 1128.64 | 2193.21 | 744.43 |
| crystal system | triclinic | monoclinic | triclinic | monoclinic | monoclinic |
| space group | <i>P</i> -1 | $P2_{1}/c$ | <i>P</i> -1 | $P2_{1}/c$ | $P2_{1}/c$ |
| <i>a</i> (Å) | 11.094(2) | 20.546(4) | 11.6557(4) | 14.0924(5) | 16.5359(4) |
| <i>b</i> (Å) | 11.584(2) | 12.171(2) | 12.1595(4) | 28.9325(9) | 10.6225(2) |
| <i>c</i> (Å) | 22.708(5) | 19.870(4) | 22.2271(8) | 27.2581(13) | 19.5757(5) |
| α (deg.) | 94.78(3) | 90 | 74.160(3) | 90 | 90 |
| β (deg) | 98.11(3) | 105.60(3) | 76.098(3) | 95.860(4) | 100.726(2) |
| γ (deg.) | 101.14(3) | 90 | 86.034(2) | 90 | 90 |
| vol (Å ³) | 2816.2(10) | 4785.8(17) | 2941.78(18) | 11055.8(7) | 3378.45(13) |
| Z | 2 | 4 | 2 | 4 | 4 |
| ρ (calcd) (g.cm ⁻ | 1.316 | 1.177 | 1.274 | 1.318 | 1.464 |
| 3) | | | | | |
| $\mu ({\rm mm}^{-1})$ | 0.927 | 0.691 | 2.908 | 3.197 | 5.097 |
| F(000) | 1166 | 1816 | 1180 | 4504 | 1504 |
| <i>T</i> (K) | 100(2) | 100(2) | 123(2) | 123(2) | 123(2) |
| reflections | 48811 | 82740 | 19383 | 65659 | 22710 |
| collected | | | | | |
| unique | 12819 | 12014 | 10856 | 19435 | 7372 |
| reflections | | | | | |
| R _{int} | 0.0178 | 0.0382 | 0.0432 | 0.1289 | 0.0277 |
| R1 indices | 0.0330 | 0.0414 | 0.0403 | 0.0585 | 0.0228 |
| [<i>I</i> >2 <i>o</i> (<i>I</i>)] | | | | | |
| wR2 indices (all | 0.0905 | 0.1078 | 0.0842 | 0.1102 | 0.0487 |
| data) | | | | | |
| CCDC No. | 859478 | 859474 | 859475 | 859477 | 859476 |



Fig. S1 Molecular structure of [(^{But}MesNacnac)₂Pb] **1S** (25% thermal ellipsoids; hydrogen atoms omitted). Selected bond lengths (Å) and angles (°): Pb(1)-N(2) 2.367(3), Pb(1)-N(4) 2.428(4), Pb(1)-N(1) 2.511(4), Pb(1)-N(3) 2.619(4), N(1)-C(1) 1.312(5), N(2)-C(3) 1.372(6), N(3)-C(30) 1.317(6), N(4)-C(32) 1.315(5), C(1)-C(2) 1.444(6), C(2)-C(3) 1.370(6), C(30)-C(31) 1.415(6), C(31)-C(32) 1.428(6), N(2)-Pb(1)-N(1) 76.77(12), N(4)-Pb(1)-N(3) 73.90(11).



Fig. S2 Molecular structure of [{(^{Mes}Nacnac)MgCl}₂{(^{But}MesNacnac)PbCl}₂] **2S** (25% thermal ellipsoids; hydrogen atoms, mesityl methyl groups, and *tert*-butyl methyl groups omitted). Selected bond lengths (Å) and angles (°): Pb(1)-N(1) 2.277(6), Pb(1)-N(2) 2.287(6), Pb(1)-Cl(1) 2.739(2), Cl(1)-Mg(1) 2.339(3), Mg(1)-N(6) 2.016(7), Mg(1)-N(5) 2.022(7), Mg(1)-Cl(2) 2.331(3), Pb(2)-N(3) 2.258(6), Pb(2)-N(4) 2.306(6), Pb(2)-Cl(3) 2.825(2), Pb(2)-Cl(2) 3.023(2), Mg(2)-N(8) 2.018(7), Mg(2)-N(7) 2.020(8), Mg(2)-Cl(4) 2.288(3), Mg(2)-Cl(3) 2.359(3), N(1)-Pb(1)-N(2) 82.2(2), Mg(1)-Cl(1)-Pb(1) 105.35(10), N(6)-Mg(1)-N(5) 94.2(3), N(3)-Pb(2)-N(4) 82.9(2), Cl(3)-Pb(2)-Cl(2) 170.41(6), Mg(1)-Cl(2)-Pb(2) 130.97(10), N(8)-Mg(2)-N(7) 4.7(3), Mg(2)-Cl(3)-Pb(2) 104.66(9).

Electronic Supplementary Material (ESI) for Chemical Communications This journal is © The Royal Society of Chemistry 2012



Fig. S3 Molecular structure of [(^{But}Nacnac)PbCl] **38** (25% thermal ellipsoids; hydrogen atoms omitted). Selected bond lengths (Å) and angles (°): Pb(1)-N(2) 2.241(2), Pb(1)-N(1) 2.309(2), Pb(1)-Cl(1) 2.5637(9), N(1)-C(1) 1.316(4), N(2)-C(3) 1.353(4), C(1)-C(2) 1.424(4), C(2)-C(3) 1.395(4), N(2)-Pb(1)-N(1) 85.52(8), N(2)-Pb(1)-Cl(1) 93.34(6), N(1)-Pb(1)-Cl(1) 94.18(6).

S 9

3. Computational Studies

For quantum-chemical calculations the Turbomole⁷ (version 5.9) program package was used, employing the Becke-Perdew 86-functional⁸ at RI-DFT⁹ level with the def2-SVP basis set¹⁰ for non-metal atoms and the def2-TZVPP basis set¹¹ for Ge and Sn atoms. The molecular orbitals of the global minimum energy structure of **7** (showing zero imaginary frequencies) in the main text were visualized with the aid of molden.¹² NBO orbital analyses were performed using the program package Gaussian09¹³ and NBO 3.0¹⁴ at the very same combination of functional and basis sets as above.

 Table S2. Cartesian coordinates for the calculated structure of 7.

| Sn | 0.40700 | -1.35000 | -0.65900 |
|----|----------|----------|----------|
| Ν | 0.49300 | -2.66500 | 1.27700 |
| Ν | -1.82700 | -1.96300 | -0.60100 |
| Н | 3.08500 | -1.18600 | 0.40300 |
| С | -0.51100 | -2.53900 | 2.17300 |
| С | 1.54300 | -3.62600 | 1.30400 |
| С | -2.50000 | -2.13700 | 0.54400 |
| С | -2.26200 | -2.14800 | -1.95500 |
| С | 3.24700 | -1.71900 | 1.37100 |
| С | -0.30800 | -2.76900 | 3.72300 |
| С | -1.82100 | -2.12400 | 1.80800 |
| С | 1.27100 | -5.02400 | 1.24400 |
| С | 2.90200 | -3.18600 | 1.27400 |
| С | -4.05300 | -2.40900 | 0.58100 |
| С | -2.08400 | -3.41700 | -2.58800 |
| С | -2.66600 | -1.03000 | -2.73800 |
| Н | 4.32000 | -1.58100 | 1.62500 |
| Η | 2.63900 | -1.19000 | 2.14000 |
| С | -1.34600 | -3.76400 | 4.30200 |
| С | 1.09400 | -3.26500 | 4.13900 |
| С | -0.50500 | -1.38300 | 4.40400 |
| Н | -2.48200 | -1.99700 | 2.66800 |
| С | 2.34800 | -5.93000 | 1.15100 |
| С | -0.13000 | -5.59400 | 1.29100 |
| С | 3.93700 | -4.13100 | 1.18800 |
| С | -4.64400 | -2.32200 | 2.01100 |
| С | -4.33900 | -3.84200 | 0.06400 |
| С | -4.84700 | -1.39400 | -0.27900 |
| С | -2.43900 | -3.56200 | -3.94000 |
| С | -1.44800 | -4.57200 | -1.85500 |
| С | -3.00300 | -1.23200 | -4.09300 |
| С | -2.74500 | 0.37000 | -2.17400 |
| Η | -1.25800 | -4.76400 | 3.82500 |
| Η | -2.39300 | -3.41800 | 4.17300 |
| Н | -1.17100 | -3.89300 | 5.39400 |
| Η | 1.15100 | -3.26400 | 5.25100 |
| Η | 1.90100 | -2.60200 | 3.76600 |
| Н | 1.31000 | -4.29500 | 3.79300 |
| Η | -1.53200 | -0.98300 | 4.26700 |
| Н | 0.21100 | -0.63100 | 4.00000 |
| Н | -0.32400 | -1.47500 | 5.49900 |
| Н | 2.11900 | -7.01000 | 1.09900 |
| С | 3.68900 | -5.51700 | 1.11900 |
| Н | -0.25500 | -6.39500 | 0.52900 |

| Η | -0.90700 | -4.82600 | 1.11400 |
|----------|----------------------|--------------------------|-------------|
| Η | -0.34000 | -6.06200 | 2.28100 |
| Н | 4.98000 | -3.76900 | 1.18500 |
| Η | -5.74300 | -2.48000 | 1.95000 |
| Η | -4.48100 | -1.32400 | 2.47400 |
| Н | -4.24000 | -3.10000 | 2.69200 |
| Н | -3.77500 | -4.60200 | 0.65000 |
| Н | -4.08400 | -3.95600 | -1.01000 |
| Η | -5.42400 | -4.06500 | 0.17500 |
| Η | -4.63600 | -1.50400 | -1.36000 |
| Η | -4.62200 | -0.34800 | 0.02200 |
| Η | -5.93700 | -1.56800 | -0.13400 |
| Н | -2.30700 | -4.55100 | -4.41300 |
| С | -2.92200 | -2.48800 | -4.71500 |
| Н | -2.00600 | -4.86100 | -0.94000 |
| Η | -0.41600 | -4.30400 | -1.52600 |
| Н | -1.37800 | -5.46600 | -2.51000 |
| Η | -3.33400 | -0.35900 | -4.68500 |
| Н | -3.79700 | 0.70700 | -2.04200 |
| Н | -2.26500 | 1.09500 | -2.86600 |
| Н | -2.25200 | 0.45900 | -1.18400 |
| С | 4.82600 | -6.50800 | 0.99600 |
| С | -3.32200 | -2.68800 | -6.16000 |
| Н | 4.48000 | -7.55000 | 1.16600 |
| Н | 5.63600 | -6.29400 | 1.72900 |
| Н | 5.29000 | -6.47300 | -0.01700 |
| Н | -2.57600 | -3.30300 | -6.71100 |
| Н | -3.42700 | -1.71900 | -6.69300 |
| Н | -4.29900 | -3.21900 | -6.24100 |
| Sn | 0.73100 | 1.29300 | 1.16100 |
| N | -0.72300 | 3.08800 | 0.85100 |
| N | 2 03300 | 2 47900 | -0 42800 |
| Н | -1 87100 | 0 60000 | 2 36400 |
| С | -0.07500 | 4 26800 | 1 06000 |
| Č | -2 10800 | 2 98000 | 0.56900 |
| č | 2 11200 | 3 79800 | -0 27400 |
| Č | 2 80400 | 1 63400 | -1 27900 |
| C | -2 42200 | 1 53100 | 2 64500 |
| C | -0.69800 | 5 45400 | 1 90000 |
| C | 1 26100 | 4 48400 | 0.66500 |
| C | -2 96800 | 2 22800 | 1 42300 |
| c | -2 65100 | 3 59000 | -0.60000 |
| C | 3 08300 | 4 76100 | -1.05800 |
| C | 1 03600 | 1 00000 | -0.80400 |
| c | 2 29100 | 1.07700 | -2 54000 |
| н | -3 24100 | 1 22300 | 3 33100 |
| н | -3.24100 -1.71200 | 2 17000 | 3 21500 |
| C | -2 19000 | 5 30800 | 2 27300 |
| C | 0.00500 | 5.30000 | 3 24100 |
| C | 0.09500 | 6 82600 | 1 10700 |
| ч | -0.55900 | 5 46000 | 0.04000 |
| Γ | 1.04000 | 2 17400 | 1 1 2 0 0 0 |
| C | 4.34400 | 2.1/400 | 0.84200 |
| C | -4.03000 | <i>J</i> . <i>J</i> 0200 | -0.84200 |
| C | -1.70400 | 4.31/00 | -1.38300 |
| | 4.02700 | J.03200 5 16600 | -1./0000 |
| C | 4.02/00 | J.40000 4 11500 | -0.04400 |
| | 2.78300 1 71000 | 4.11300 | -2.13000 |
| | 4./4800 | 0.20700 | -1.02800 |
| C | 4.00400 | 1.50800 | 0.53800 |
| C | 3.04300 | 0.52800 | -3.32300 |
| U | 0.96800 | 1./5500 | -3.03600 |
| H | -2.39500 | 4.37000 | 2.82600 |
| Н | -2.86100 | 5.34200 | 1.39200 |

| Н | -2.46600 | 6.15300 | 2.94200 |
|---|----------|----------|----------|
| Η | -0.31300 | 6.27000 | 3.90800 |
| Η | 1.17600 | 5.67600 | 3.08900 |
| Η | 0.00400 | 4.50000 | 3.76800 |
| Η | -1.05400 | 6.84000 | 0.21200 |
| Η | 0.52100 | 7.11200 | 1.03200 |
| Η | -0.99700 | 7.62300 | 1.82500 |
| Н | -5.00300 | 1.62700 | 1.83700 |
| С | -4.91000 | 2.81400 | 0.01800 |
| Η | -4.44500 | 3.98900 | -1.74600 |
| Η | -1.48300 | 5.33700 | -1.23700 |
| Η | -2.27500 | 4.43300 | -2.56500 |
| Η | -0.81200 | 3.76900 | -1.74100 |
| Η | 2.88700 | 6.53200 | -2.34300 |
| Η | 1.60400 | 6.42900 | -1.08600 |
| Η | 1.53800 | 5.35700 | -2.52200 |
| Η | 4.63400 | 4.72900 | 0.52500 |
| Η | 3.48400 | 6.09700 | 0.69000 |
| Η | 4.73000 | 6.13000 | -0.59600 |
| Η | 4.56700 | 4.92600 | -2.62800 |
| Η | 3.40200 | 3.60000 | -2.92700 |
| Η | 4.70900 | 3.38900 | -1.72200 |
| Η | 5.70700 | -0.19800 | -1.25700 |
| С | 4.28000 | -0.19000 | -2.89600 |
| Η | 5.39300 | 0.80000 | 0.86800 |
| Η | 3.82200 | 1.55600 | 1.32900 |
| Η | 5.06900 | 2.52100 | 0.49600 |
| Η | 2.63800 | 0.01700 | -4.30200 |
| Η | 0.16200 | 1.57500 | -2.28800 |
| Η | 0.67300 | 1.26700 | -3.99000 |
| Η | 0.99900 | 2.85400 | -3.20900 |
| С | -6.39600 | 2.73000 | -0.26000 |
| С | 5.08200 | -1.13400 | -3.76400 |
| Η | -6.68600 | 1.72100 | -0.63600 |
| Η | -6.71100 | 3.47200 | -1.02400 |
| Н | -6.99500 | 2.91300 | 0.66000 |
| Н | 5.92300 | -0.60600 | -4.27200 |
| Н | 4.45300 | -1.59400 | -4.55600 |
| Η | 5.53100 | -1.95800 | -3.16600 |

Table S3. Cartesian coordinates for the calculated structure of 8.

Ge -1.13900 -0.44900 -0.74800 N -0.46500 -1.25300 -2.53200 C -2.55000 -1.85500 -0.93100 N 0.25300 -0.51700 0.51400 C -1.39800 -2.03500 -3.09000 C 0.79300 -0.85000 -3.08000 C -2.51800 -2.33900 -2.22600 $C \quad \text{-}3.81700 \quad \text{-}1.98800 \quad \text{-}0.07100$ Mg 0.19400 1.24900 1.42900 С 1.26100 -1.48200 0.70300 С -1.48300 -2.52500 -4.56200 С 0.94500 0.44900 -3.63400 С 1.90300 -1.73300 -2.98800 Н -3.36700 -2.89500 -2.66100 C -5.04300 -1.39100 -0.81400 C -4.08700 -3.49000 0.21300 C -3.66100 -1.24400 1.27300 N 0.12400 3.29300 1.12000 N 0.23500 1.35600 3.48700

| Η | 2.78200 | 1.04100 | 0.77200 |
|-----------|-------------|----------|----------|
| Н | 2.35700 | 0.55700 | -0.90700 |
| С | 2.62700 | -1.12400 | 0.45600 |
| С | 1.00300 | -2.80000 | 1.20600 |
| С | -1.66400 | -4.06700 | -4.55800 |
| С | -2.73600 | -1.86600 | -5.21100 |
| С | -0.27800 | -2.19400 | -5.46900 |
| Č | 2 21600 | 0.83400 | -4 10400 |
| c | -0.22500 | 1 39800 | -3 75400 |
| C | 3 15100 | -1 29400 | -3 47000 |
| C | 1 75500 | 2 12200 | -3.47000 |
| с u | 5.05500 | -3.12300 | -2.41700 |
| п | -3.93300 | -1.43900 | -0.1/800 |
| Н | -5.25300 | -1.92900 | -1./6500 |
| Н | -4.8/100 | -0.32000 | -1.06100 |
| Н | -3.24500 | -3.95400 | 0.77000 |
| Н | -4.22800 | -4.06300 | -0.73000 |
| Н | -5.01000 | -3.60500 | 0.82500 |
| Η | -2.77500 | -1.60100 | 1.83800 |
| Η | -4.56200 | -1.38900 | 1.91100 |
| Η | -3.54100 | -0.15000 | 1.10400 |
| С | 0.24900 | 4.15800 | 2.15100 |
| С | -0.13200 | 3.84300 | -0.18300 |
| Ċ | 0.39300 | 2.51900 | 4.14000 |
| Č | 0.04400 | 0 15100 | 4 24200 |
| c | 2 96400 | 0.26800 | -0.02100 |
| C | 2.50400 | 2 05000 | 0.66500 |
| C | 2.05900 | -2.03900 | 1 20200 |
| C | 2.00000 | -3.70200 | 1.59500 |
| | -0.39800 | -3.20100 | 1.38000 |
| Н | -0.//900 | -4.5/300 | -4.11300 |
| Н | -1.///00 | -4.43300 | -5.60300 |
| Н | -2.56100 | -4.39100 | -3.98900 |
| Η | -3.67800 | -2.13600 | -4.68900 |
| Н | -2.82700 | -2.19800 | -6.27000 |
| Η | -2.65000 | -0.75700 | -5.20800 |
| Η | -0.10600 | -1.10300 | -5.56900 |
| Н | -0.48600 | -2.59600 | -6.48600 |
| Н | 0.66600 | -2.65300 | -5.11300 |
| Н | 2.32900 | 1.84000 | -4.54500 |
| С | 3.33500 | -0.01800 | -4.03500 |
| Н | -0.52800 | 1.81000 | -2.76600 |
| Н | -1 12600 | 0.89500 | -4 17000 |
| Н | 0.02900 | 2 25500 | -4 41200 |
| н | 4 01200 | -1 98100 | -3 39400 |
| н | 1 20700 | -3 79900 | -3.11400 |
| н | 1 1 1 0 0 0 | -3 11600 | -1.46100 |
| и П | 2 74800 | -3.11000 | 2 22100 |
| II C | 2.74800 | -3.37000 | -2.22100 |
| C | 0.15000 | 5.05500 | 1.8//00 |
| C | 0.44000 | 3./9000 | 3.50/00 |
| C | -1.4/300 | 3.9/400 | -0.63600 |
| C | 0.94500 | 4.25200 | -1.01800 |
| С | 0.51300 | 2.50900 | 5.65600 |
| С | -1.27200 | -0.21500 | 4.63800 |
| С | 1.14400 | -0.69100 | 4.55000 |
| Н | 4.03900 | 0.35500 | -0.28800 |
| Н | 4.70100 | -1.75400 | 0.45600 |
| С | 3.40700 | -3.36500 | 1.12400 |
| Н | 1.83300 | -4.71100 | 1.78100 |
| Н | -1.09400 | -3.11000 | 0.71500 |
| Н | -0.79700 | -2.53400 | 2.37800 |
| Н | -0 43300 | -4 24800 | 1 95200 |
| \hat{C} | 4 68100 | 0.41000 | -4 57800 |
| Ч | 0.80300 | 5 08/00 | 1 12000 |
| и Ц | 0.07500 | 6 2/200 | 2 80200 |
| 11 | 0.50100 | 0.24000 | 2.00300 |

| Н | -0.84900 | 5.90800 | 1.45800 |
|---|----------|----------|----------|
| Н | 0.57400 | 4.63700 | 4.19500 |
| С | -1.71200 | 4.53700 | -1.90500 |
| С | -2.62500 | 3.50700 | 0.22400 |
| С | 0.65200 | 4.81400 | -2.27500 |
| С | 2.38200 | 4.08500 | -0.57600 |
| Н | -0.41400 | 2.11600 | 6.12700 |
| Н | 0.70600 | 3.52500 | 6.05500 |
| Н | 1.33400 | 1.83500 | 5.98400 |
| С | -1.46000 | -1.42100 | 5.34100 |
| С | -2.44900 | 0.67500 | 4.30800 |
| С | 0.90400 | -1.88400 | 5.25900 |
| С | 2.54600 | -0.33400 | 4.11100 |
| С | 4.52300 | -4.36900 | 1.32700 |
| Н | 5.51800 | -0.01900 | -3.98400 |
| Н | 4.79100 | 1.51600 | -4.57800 |
| Н | 4.82200 | 0.06700 | -5.63100 |
| Н | -2.75600 | 4.63800 | -2.24900 |
| С | -0.66800 | 4.97800 | -2.73800 |
| H | -3.60000 | 3.78800 | -0.22800 |
| Н | -2.58500 | 3.92900 | 1.25200 |
| Н | -2.61700 | 2.39500 | 0.32600 |
| Н | 1.49200 | 5.13800 | -2.91500 |
| Н | 3.07600 | 4.57400 | -1.29200 |
| Н | 2.66500 | 3.01000 | -0.51900 |
| Н | 2.57300 | 4.51400 | 0.43200 |
| Н | -2.48400 | -1.70000 | 5.64700 |
| С | -0.38700 | -2.27300 | 5.66300 |
| H | -3.40700 | 0.19800 | 4.60500 |
| Н | -2.50200 | 0.89500 | 3.21800 |
| Н | -2.38900 | 1.66300 | 4.82000 |
| Н | 1.76300 | -2.53400 | 5.50200 |
| Н | 3.30000 | -0.93200 | 4.66600 |
| Н | 2.77700 | 0.74300 | 4.26300 |
| Н | 2.69200 | -0.55700 | 3.02700 |
| Н | 5.52000 | -3.91200 | 1.14800 |
| Н | 4.42800 | -5.23900 | 0.63600 |
| Н | 4.52300 | -4.78300 | 2.36100 |
| С | -0.95000 | 5.62800 | -4.07600 |
| Ċ | -0.61600 | -3.58200 | 6.38900 |
| Ĥ | -0.91200 | 6.74000 | -4.00300 |
| Н | -1.95800 | 5.35800 | -4.45700 |
| Н | -0.20200 | 5.32900 | -4.84400 |
| Н | -0.72200 | -4.42900 | 5.67200 |
| Н | -1.54400 | -3.55400 | 7.00100 |
| Н | 0.23200 | -3.83100 | 7.06300 |
| | | | |

Table S4. Cartesian coordinates for the calculated structure of [(^{But}MesNacnac)Sn]·

| Sn | -0.00300 | -1.32100 | 0.18400 |
|----|----------|----------|----------|
| Ν | -1.56800 | 0.37900 | 0.04700 |
| Ν | 1.56600 | 0.37600 | 0.04100 |
| Η | -1.82000 | -0.52700 | 2.63000 |
| С | -1.31900 | 1.69400 | 0.02100 |
| С | -2.80300 | -0.33200 | -0.02000 |
| С | 1.32100 | 1.69000 | 0.01500 |
| С | 2.80100 | -0.33700 | -0.02100 |
| С | -2.91100 | -0.32600 | 2.53200 |
| С | -2.43900 | 2.80800 | -0.02400 |
| С | 0.00200 | 2.23100 | 0.04000 |

| С | -3.43400 | -0.76400 | 1.18200 |
|--------|--------------------|----------|----------|
| С | -3.29500 | -0.76700 | -1.28300 |
| С | 2.44300 | 2.80200 | -0.03800 |
| С | 3.43000 | -0.76300 | 1.18400 |
| С | 3.29200 | -0.78500 | -1.28100 |
| Н | -3.43800 | -0.85500 | 3.35400 |
| Н | -3.04500 | 0.76900 | 2.69300 |
| С | -3.90300 | 2.31200 | -0.06400 |
| Ċ | -2.30800 | 3.69400 | 1.24600 |
| Ċ | -2 23600 | 3 68300 | -1 29100 |
| Н | 0.00300 | 3 32300 | 0.03700 |
| C | -4 57500 | -1 58300 | 1 08900 |
| C | -4 44000 | -1 58600 | -1 31700 |
| C | -2 61500 | -0 34300 | -2 56300 |
| c | 2 23400 | 3 67400 | -1 30800 |
| C | 2.23400 | 3 60300 | 1 22000 |
| C | 3 90600 | 2 30300 | -0.08700 |
| C | 1 56000 | 1 58600 | 1 00800 |
| C | 2 00200 | -1.38000 | 2 52000 |
| C | 2.90300 | -0.32100 | 2.33000 |
| C | 4.43400 | -1.00700 | -1.50800 |
| C | 2.60900 | -0.3/300 | -2.56400 |
| Н | -4.18300 | 1./1500 | 0.82600 |
| Н | -4.13000 | 1./0/00 | -0.96500 |
| Н | -4.56800 | 3.20400 | -0.08700 |
| Н | -3.09/00 | 4.4/900 | 1.24000 |
| Н | -1.32600 | 4.20600 | 1.31700 |
| Н | -2.43900 | 3.08800 | 2.17000 |
| Η | -2.30400 | 3.06800 | -2.21600 |
| Η | -1.25700 | 4.20600 | -1.30600 |
| Η | -3.03300 | 4.46000 | -1.34200 |
| Η | -5.06900 | -1.90400 | 2.02300 |
| С | -5.10000 | -2.00700 | -0.14700 |
| Η | -4.82500 | -1.91100 | -2.30000 |
| Η | -2.70300 | 0.75500 | -2.73900 |
| Н | -3.05600 | -0.86100 | -3.44100 |
| Η | -1.52500 | -0.57100 | -2.53500 |
| Н | 3.03300 | 4.44800 | -1.36700 |
| Η | 1.25700 | 4.19900 | -1.31700 |
| Η | 2.29300 | 3.05500 | -2.23000 |
| Н | 2.45800 | 3.09000 | 2.15400 |
| Н | 1.34000 | 4.20700 | 1.30400 |
| Н | 3.11200 | 4.47700 | 1.21500 |
| Н | 4.57300 | 3.19400 | -0.11700 |
| Н | 4.12600 | 1.69500 | -0.98700 |
| Н | 4.19100 | 1.70900 | 0.80400 |
| Н | 5 05700 | -1 90800 | 2 03400 |
| C | 5 09700 | -2.01600 | -0 13500 |
| н | 3 43200 | -0.84300 | 3 35600 |
| н | 1 81200 | -0 52800 | 2 62900 |
| н | 3.03000 | 0.52000 | 2.62700 |
| н | 4 81400 | -1 94700 | -2 28800 |
| н | 1 51800 | -0.60200 | -2.28800 |
| и П | 2 0/000 | 0.00200 | -2.33200 |
| н Ц | 2.04200 | 0.70000 | -3.43/00 |
| C | 2.09000 6.20400 | 2 02200 | -2.75000 |
| | -0.30400 | -2.92200 | -0.21100 |
| U | 0.34000 | -2.8/000 | -0.19/00 |
| H U | -0.00800 | -3.99000 | -0.08500 |
| H | -0.82900 | -2.84000 | -1.18/00 |
| H | -/.03/00 | -2.69200 | 0.59300 |
| H | /.26500 | -2.25300 | -0.21500 |
| H | 6.35600 | -3.50900 | -1.11000 |
| H | 6.41900 | -3.54700 | 0.68700 |

4. References

- 1. P.H.M. Budzelaar, A.B. van Oort and A.G. Orpen, Eur. J. Inorg. Chem. 1998, 1485.
- 2. S.L. Choong, W.D. Woodul, C. Schenk, A. Stasch, A.F. Richards and C. Jones, *Organometallics*, 2011, **30**, 5543.
- 3. S.J. Bonyhady, C. Jones, S. Nembenna, A. Stasch, A.J. Edwards and G.J. McIntyre, *Chem. Eur. J.*, 2010, **16**, 938
- T.M. McPhillips, S.E. McPhillips, H.J. Chiu, A.E. Cohen, A.M. Deacon, P.J. Ellis, E. Garman, A. Gonzalez, N.K. Sauter, R.P. Phizackerley, S.M. Soltis, and P. Kuhn, J. Synchrotron Rad., 2002, 9, 401.
- 5. W. Kabsch, J. Appl. Cryst., 1993, 26, 795.
- 6. G.M. Sheldrick, *SHELX-97*, University of Göttingen, 1997.
- 7. O. Treutler and R. Ahlrichs, J. Chem. Phys., 1995, 102, 346.
- 8. (a) J.P. Perdew, *Phys. Rev. B* 1986, **33**, 8822; (b) A.D. Becke, *Phys. Rev. A*, 1988, **38**, 3098.
- 9. K. Eichkorn, O. Treutler, H. Öhm, M. Häser and R. Ahlrichs, *Chem. Phys. Lett.*, 1995, **240**, 283.
- 10. A. Schäfer, H. Horn and R. Ahlrichs, J. Chem. Phys., 1992, 97, 2571.
- 11. A. Schaefer, C. Huber and R. Ahlrichs, J. Chem. Phys., 1994, 100, 5829.
- 12. G. Schaftenaar and J.H. Noordik, Mol. Design, 2000, 14, 123.
- 13. M. J. Frisch et al., Gaussian 09, Wallingford CT, 2009.
- 14. A. E. Reed, L. A. Curtiss and F. Weinhold, *Chem. Rev.*, 1988, **88**, 899.