Dimethylamine borane dehydrogenation chemistry: syntheses, X-ray and neutron diffraction studies of 18electron aminoborane and 14-electron aminoboryl complexes

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10 Supporting Information

General details of synthetic procedures and instrumentation

Materials: All reactions involving air- or moisture-sensitive ¹⁵ compounds were carried out under an inert atmosphere by using Schlenk-type glassware or in a glovebox. Solvents were dried using an MBraun SPS800 prior to use. NMR-solvents were dried over molecular sieves and degassed before use when necessary. Solid starting materials were dried on high vacuum before use

 $_{20}$ when necessary. Unless otherwise noted, all starting materials were commercially available and were used without further purification. M(IMes)_2(H)_2Cl (**1a/b**) and Na[BAr^f_4] were prepared via literature routes. $^{\rm s1,s2}$

Techniques: The following instruments were used for physical

²⁵ characterization of the compounds: IR: Nicolet Magna-IR 560; NMR: Bruker AVC500 (¹H: 500 MHz; ¹³C: 125 MHz); Bruker DRX500 (¹¹B: 160 MHz), Varian Unity500 (¹H: 500 MHz; ¹³C: 125 MHz, ¹¹B: 160 MHz), Varian Mercury VX-300 (³¹P: 122 MHz, ¹⁹F: 282 MHz, ¹¹B: 96 MHz). Details of the
³⁰ crystallographic studies are contained within the manuscript itself

and within the respective CIFs.

Syntheses and characterizing data for new compounds

- ³⁵ **2**: To a suspension of **1a** and Na[BAr^f₄] (0.25 mmol of each) in fluorobenzene (50 cm³) at -30°C was added Me₂NH[•]BH₃ (0.295g, 5 mmol) and the reaction mixture warmed to 20 °C over a period of 1 h. After stirring for a further 24 h, the resulting light yellow solution was filtered, concentrated in *vacuo* and light yellow
- ⁴⁰ crystals suitable for X-ray diffraction obtained by layering with pentane and storage at 20 °C. Isolated yield 0.33 g, 76 %. ¹H NMR (toluene- d_8 , 300 MHz, 20 °C): δ_H -15.24 (s, 2H, IrH), -6.31 (br s, 2H, BH₂), 1.49 (s, 24H, *ortho*-Me of IMes), 1.61 (s, 6H, Me of NMe₂), 2.21 (s, 12H, *para*-Me of IMes), 5.98 (s, 4H, NCH of
- ⁴⁵ IMes), 6.63 (s, 8H, *meta*-CH of IMes), 7.69 (s, 4H, *para*-CH of $[BAr_4^{f_4}]$), 8.31 (s, 8H, *para*-CH of $[BAr_4^{f_4}]$). ¹³C NMR (toluened₈, 75 MHz, 20 °C): (i) signals due to cation: $\delta_{\rm C}$ 17.1 (*ortho*-Me of IMes), 17.9 (*para*-Me of IMes), 39.6 (Me of NMe₂), 121.6 (NCH of IMes), 129.1 (*ortho*-quaternary C of IMes) 134.4 (*meta*-
- ⁵⁰ CH of IMes), 138.8 (*para*-quaternary C of IMes), carbene quaternary of IMes and N-bound aryl quaternary signals not observed; (ii) signals due to anion: $\delta_{\rm C}$ 117.6 (*para*-CH), 124.9 (q, ${}^{1}J_{\rm CF}$ = 273.6 Hz, CF₃), 129.5 (q, ${}^{2}J_{\rm CF}$ = 34.5 Hz, *meta*-quaternary

C), 135.1 (*ortho*-CH), 162.2 (q, ${}^{1}J_{CB} = 50.3$ Hz, *ipso*-quaternary 55 C). ${}^{11}B$ NMR (toluene- d_8 , 96 MHz, 20 °C): δ_B -6.1 ([BAr^f_4]⁻ anion), 36 (br, H₂BNMe₂). ${}^{19}F$ NMR (toluene- d_8 , 282 MHz, 20 °C): δ_F -61.3. IR (ν_{BH} , cm⁻¹): 2360, 2343. Elemental microanalysis: calcd for C₇₆H₇₀N₅B₂F₂₄Ir: C, 52.94; H, 4.09, N, 4.06; measd: C, 52.94; H, 3.92; N, 3.93. *Crystallographic data*:

- ⁶⁰ (X-ray) $C_{76}H_{70}N_5B_2F_{24}Ir$, $M_r = 1723.2$, monoclinic, *C1c1*, a = 22.8105(2), b = 19.0153(2), c = 19.9792(2) Å, $\beta = 117.1045(4)^{\circ}$, V = 7714.2(1) Å³, Z = 4, $\rho_c = 1.484$ Mg m⁻³, T = 150(2) K, $\lambda = 0.71073$ Å. 66010 reflections collected, 17099 independent [R(int) = 0.051], which were used in all calculations. $R_1 = 0.0360$,
- ⁶⁵ $wR_2 = 0.0813$ for observed unique reflections $[F^2 > 2\sigma(F^2)]$ and $R_1 = 0.0423$, $wR_2 = 0.0866$ for all unique reflections. Max. and min. residual electron densities 1.19 and -0.96 e Å⁻³. CSD refs: 880791 (X-ray), 880792 (neutron).
- ⁷⁰ **3**: was synthesized from **1b** as per the synthesis of **2**, and yellow crystals suitable for X-ray diffraction obtained from a fluorobenzene/pentane layering at 20 °C. Isolated yield 0.29 g, 71 %. ¹H NMR (toluene- d_8 , 300 MHz, 20 °C): $\delta_{\rm H}$ -23.59 (d, ¹ $J_{\rm RhH}$ = 43.5 Hz, 1H, RhH), 0.51 (br s, 1H, BH), 1.41 (s, 3H, NMe), 1.51
- ⁷⁵ (s, 24H, *ortho*-Me of IMes), 1.59 (s, 3H, NMe), 2.19 (s, 12H, *para*-Me of IMes), 6.01 (s, 4H, NCH of IMes), 6.65 (s, 8H, *meta*-CH of IMes), 7.68 (s, 4H, *para*-CH of [BAr^f₄]⁻), 8.30 (s, 8H, *para*-CH of [BAr^f₄]⁻). ¹³C NMR (toluene-*d*₈, 75 MHz, 20 °C): (i) signals due to cation: $δ_C$ 17.7 (*ortho*-Me of IMes), 21.8 (*para*-Me
- ⁸⁰ of IMes), 31.5 (Me of NMe₂), 41.3 (Me of NMe₂), 122.9 (NCH of IMes), 129.8 (*ortho*-quaternary C of IMes) 135.5 (*meta*-CH of IMes), 140.1 (*para*-quaternary C of IMes), 183.7 (d, ${}^{1}J_{RhC} = 46.5$ Hz, carbene quaternary C), N-bound aryl quaternary signals not observed; (ii) signals due to anion: δ_C 118.3 (*para*-CH), 125.7 (q,
- ⁸⁵ ${}^{1}J_{CF} = 273.1$ Hz, CF₃), 130.2 (q, ${}^{2}J_{CF} = 31.5$ Hz, *meta*-quaternary C), 135.9 (*ortho*-CH), 163.0 (q, ${}^{1}J_{CB} = 49.1$ Hz, *ipso*-quaternary C). ¹¹B NMR (toluene- d_8 , 96 MHz, 20 °C): $\delta_{\rm B}$ -6.0 ([BAr^f₄]⁻ anion), 53 (br, H₂BNMe₂). ¹⁹F NMR (toluene- d_8 , 282 MHz, 20 °C): $\delta_{\rm F}$ -62.2. IR ($\nu_{\rm BH}$, cm⁻¹): 2394. Elemental microanalysis: ⁹⁰ calcd for C₇₆H₆₈B₂F₂₄N₅Rh: C, 55.90; H, 4.20, N, 4.29; measd: C, 55.67; H, 3.99; N, 4.00. *Crystallographic data:* (X-ray) C₇₆H₆₈B₂F₂₄N₅Rh, $M_{\rm r} = 1631.9$, monoclinic, $P2_1/n$, a =12.7501(1), b = 14.1549(1), c = 21.1993(2) Å, $\beta = 93.0899(3)$ °, V = 3820.4(1) Å³, Z = 2, $\rho_c = 1.419$ Mg m⁻³, T = 150(2) K, $\lambda =$ 95 0.71073 Å. 61210 reflections collected, 8721 independent [R(int) = 0.024], which were used in all calculations. $R_1 = 0.0666$, $wR_2 =$ 0.1801 for observed unique reflections [$F^2 > 2\sigma(F^2)$] and $R_1 =$ 0.0824, $wR_r = 0.1922$ for all unique reflections Max, and min
- 0.0824, $wR_2 = 0.1922$ for all unique reflections. Max. and min. residual electron densities 1.40 and -0.98 e Å⁻³. CSD ref.: 880793 ¹⁰⁰ (X-ray), 880794 (neutron).



Fig. S1 Molecular structure of the cationic component of 3 as determined by X-ray diffraction (left; one disorder component ⁵ only) and Density Functional Theory (right). Counter-ion, mesityl Me groups and H atoms (except Rh- and B-bound Hs for DFT structure) omitted for clarity. Thermal ellipsoids set at the 30 % probability level for X-ray structure.

10 Details of Density Functional Theory Calculations

The DFT calculations were performed using the Amsterdam Density Functional (ADF) Package Software 2012.^{s3} Calculations were performed using the Vosko-Wilk-Nusair local density ¹⁵ approximation with exchange from Becke^{s4} and correlation corrections from Lee-Yang-Parr^{s5} (BLYP). Slater-type orbitals (STOs)^{s6} were used for the triple zeta basis set with an additional set of polarization functions (TZP). The large frozen core basis set approximation was applied with no molecular symmetry. The ²⁰ general numerical integration was 6 for geometry optimization and 4 for frequency calculation. No significant imaginary frequencies were observed for the optimized geometry of

complex **3**. See below for the run file for frequency calculation which contain coordinates for the optimized geometry of complex ²⁵ **3**.

#! /bin/sh

"\$ADFBIN/adf" <<eor ATOMS

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| Page | 8 | of | 11 | |
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| | ~ | ς. | ••• | |

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