# *In-situ* reversible conversion of porphyrin aggregate morphologies

Ming-Cheng Kuo<sup>a</sup>, Hsiao-Fan Chen<sup>a</sup>, Jing-Jong Shyue<sup>b</sup>, Dario M. Bassani<sup>c</sup>,

Ken-Tsung Wong<sup>a</sup>\*

<sup>a</sup>Department of Chemistry, National Taiwan University, Taipei 10617, Taiwan Fax +886 2 33661667; Tel: +886 2 33661665; E-mail: kenwong@ntu.edu.tw <sup>b</sup>Research Center for Applied Sciences, Academia Sinica, Taipei 115, Taiwan. <sup>c</sup>Univ. Bordeaux, ISM, CNRS UMR 5255, 351, Cours de la Libération, 33400 Talence, France.

**Electronic Supplementary Information** 

### Materials and methods:

All the starting materials were purchased from commercial sources and used without further purification. Solvents for chemical synthesis were purified by distillation. All chemical reactions were carried out under an argon or nitrogen atmosphere. <sup>1</sup>H and <sup>13</sup>C NMR spectra of compounds were collected on a 400 MHz spectrometer at room temperature. **5a**<sup>1</sup>, **1a**<sup>1</sup> and **4**<sup>2</sup> were synthesized according to the published procedures.

## **Instruments and Experimental Techniques**

## **Optical Measurements:**

UV-visible absorption spectra were recorded on a HITACHI U2800A spectrophotometer.

# **Preparation of samples:**

Solutions (0.1 mM) of compounds **1b**, **2**, or **3** were prepared by adding a pure or mixed solvent to a glass vial containing the solid sample. Dissolution of solid samples was achieved by gentle shaking of the vial. The solutions (5 or 10  $\mu$ L) were then dropcast onto the substrates using a micro-pippete and the solvent was evaporated in air.

### Scanning electron microscopy:

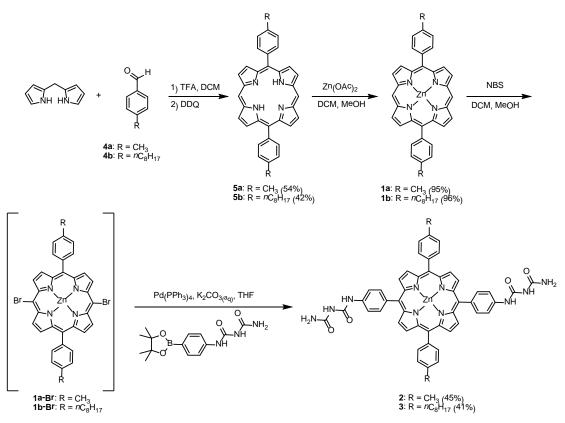
Samples were prepared by drop-casting solutions of compounds **1b**, **2**, or **3** (0.1 mM) onto a flat  $SiO_2/Si$  substrate. The samples were imaged using a FEI Nova NanoSEM 200 in low-vacuum mode with no conductive overcoat. The chamber pressure was maintained at 0.45 Torr water or THF vapor using a differential pumping system. An immersion lens was employed and the secondary electrons amplified by gas vapor and collected by an electrode mounted on the pole piece.

## Transmission electron microscopy:

Samples were dropcast onto a 200 mesh copper grid coated with formvar film stabilized with vacuum-evaporated carbon and dried under air. The samples were examined using a Hitachi H-7650 electron microscope operating at 75 kv.

## **Crystal structure determination:**

Crystallographic data were collected at 295(2) K on a Oxford Gemini A CCD diffractometer using graphite-monochromatized Cu-K $\alpha$  radiation ( $\lambda = 1.54178$  Å). Cell parameters were retrieved and refined using CrysAlis Pro software on all observed reflections. Data reduction was performed with the CrysAlis Pro software. The structures were solved and refined with SHELXL programs. The hydrogen atoms were included in calculated positions and refined using a riding mode.



Scheme S1 Synthesis of porphyrin derivatives 2, and 3.

# **Synthetic procedures:**

Synthesis of 5,15-Bis(4-octylphenyl)porphyrin (5b). A solution of dipyrromethane (191 mg, 1.3 mmol) and 4-octylbenzaldehyde (282 mg, 1.3 mmol) in dry  $CH_2Cl_2$  (250 mL) was stirred under N<sub>2</sub> for 15 min. TFA (55  $\mu$ L, 0.85 mmol) was added to the solution via syringe, the flask was shielded from light, and the solution was stirred for 16 h at room temperature. DDQ (0.5 g, 2.17 mmol) was added, and the solution was

stirred for an additional 2 h. The crude product was purified by flash column chromatography on silica gel (DCM) to afford pure **5b** (190 mg, 42% yield) as a purple solid. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  10.30 (s, 2H), 9.38 (d, *J* = 4.8 Hz, 4H), 9.12 (d, *J* = 4.8 Hz, 4H), 8.18 (d, *J* = 7.6 Hz, 4H), 7.62 (d, *J* = 7.6 Hz, 4H), 3.00 (t, *J* = 7.6 Hz, 4H), 1.97 (quin, *J* = 7.6 Hz, 4H) 1.65-1.01 (m, 20H), 0.99 (t, *J* = 6.8 Hz, 6H), -3.05 (s, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  147.3, 145.1, 142.4, 138.6, 134.8, 131.5, 131.1, 127.0, 119.3, 105.1, 36.0, 32.0, 31.7, 29.7, 29.6, 29.4, 22.8, 14.2; HRMS (*m*/*z*, FAB<sup>+</sup>) [M<sup>+</sup> + H] Calcd for C<sub>48</sub>H<sub>55</sub>N<sub>4</sub> 687.4427, found 687.4426.

**Synthesis of [5,15-Bis(4-octylphenyl)porphyrinato]zinc(II) (1b).** A saturated solution of Zn(OAc)<sub>2</sub> in methanol (20 mL) was added to a solution of **5b** (1.374 g, 2 mmol) in CHCl<sub>3</sub> (300 mL) and the resulting mixture was stirred for 3 h at 60 °C. Then, the organic residue was washed with water, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and evaporated. The crude product was purified by column chromatography on silica gel (DCM) to afford pure **2** (96%, 1.44 g) as a purple solid. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  10.27 (s, 2H), 9.41 (d, *J* = 4.4 Hz, 4H), 9.16 (d, *J* = 4.4 Hz, 4H), 8.16 (d, *J* = 7.6 Hz, 4H), 7.60 (d, *J* = 7.6 Hz, 4H), 3.00 (t, *J* = 7.6 Hz, 4H), 1.97 (quin, *J* = 7.6Hz, 4H), 1.64-1.41 (m, 20H), 0.98 (t, *J* = 6.8 Hz, 6H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz) δ 150.3, 149.2, 142.2, 139.8, 134.6, 132.6, 131.6, 126.7, 120.3, 106.1, 36.0, 32.0, 31.7, 29.7, 29.4, 22.8, 14.2; HRMS (*m*/*z*, FAB<sup>+</sup>) Calcd for C<sub>48</sub>H<sub>52</sub>N<sub>4</sub>Zn 748.3483, found 748.3477.

# Synthesis

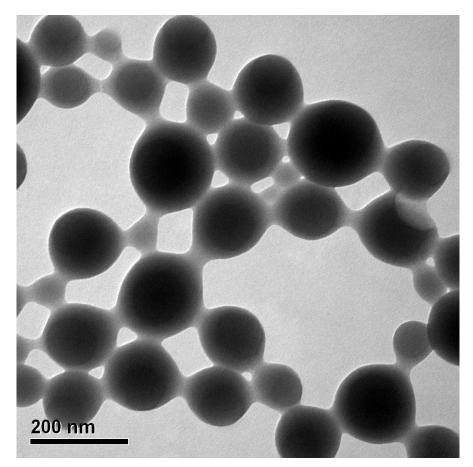
[5,15-Bis(benzenebiuret)-10,20-Bis(4-methylphenyl)porphyrinato]zinc(II) (2). A mixture of **1a** (673 mg, 1.21 mmol), NBS (476 mg, 2.67 mmol) was dissolved in DCM/MeOH = 9/1 and stirred at 0 °C for 10 min. The crude product (**1a-Br**) was concentrated and washed by MeOH and acetone, which was used directly for the next step without further purification. A mixture of Pd(PPh<sub>3</sub>)<sub>4</sub> (116 mg, 0.1 mmol), K<sub>2</sub>CO<sub>3</sub> (3.75 ml, 2M, 7.5 mmol) **1a-Br** (711 mg, 1.0 mmol), 4-pinacolatoboronic ester-benzenebiuret (**4**) (763 mg, 2.5 mmol) in degassed THF (40 mL) was refluxed for two days. The reaction was quenched by adding water and extracted with DCM and THF. The combined organic solution was washed with brine and dried over MgSO<sub>4</sub>. The crude product was purified by column chromatography on silica gel (THF/DCM = 1/2) to afford pure **2** (45%, 450 mg) as a purple solid. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 400 MHz)  $\delta$  10.34 (s, 2H), 9.05 (s, 2H), 8.79 (d, *J* = 4.8 Hz, 4H), 8.77 (d, *J* = 4.8 Hz, 4H), 8.09 (d, *J* = 8.4 Hz, 4H), 8.04 (d, *J* = 8.0 Hz, 4H), 7.85 (d, *J* = 8.4 Hz, 4H), 7.58 (d, *J* = 8.0 Hz, 4H), 2.66 (s, 6H); <sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 400 MHz)  $\delta$  155.6, 152.3, 149.4, 149.3, 139.9, 137.6, 137.4, 136.6, 134.1, 131.5, 127.2, 120.3, 119.8,

of

117.2, 21.0; HRMS (m/z, FAB<sup>+</sup>) Calcd for C<sub>50</sub>H<sub>48</sub>N<sub>10</sub>O<sub>4</sub>Zn 906.2369, found 906.2380. [5,15-Bis(benzenebiuret)-10,20-Bis(4-octylphenyl) **Synthesis** of porphyrinato]zinc(II) (3). A mixture of 1b (87 mg, 0.13 mmol), NBS (51 mg, 0.29 mmol) was dissolved in DCM/MeOH = 9/1 and stirred at 0 °C for 10 min. The crude product (1b-Br) was concentrated and washed by MeOH and acetone, which was applied directly for the next step without further purification. A mixture of  $Pd(PPh_3)_4$ (40 mg, 0.035 mmol), K<sub>2</sub>CO<sub>3</sub> (828 mg, 6 mmol) **1b-Br** (100 mg, 0.11 mmol), 4-pinacolatoboronic ester-benzenebiuret (4) (305 mg, 1 mmol) in degassed THF (10 mL) and water (3 mL) was refluxed for two days. The reaction was quenched by adding water and extracted with THF. The combined organic solution was washed with brine and dried over MgSO<sub>4</sub>. The crude product was purified by column chromatography on silica gel (THF/DCM = 1/2) to afford pure 3 (41%, 55 mg) as a purple solid. <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz)  $\delta$  10.35 (s, 2H), 9.05 (s, 2H), 8.78(d, J = 4.8 Hz, 4H), 8.75 (d, J = 4.8 Hz, 4H), 8.09 (d, J = 8.0 Hz, 4H), 8.03 (d, J = 8.0 Hz, 4H), 7.86 (d, J = 8.0 Hz, 4H), 7.54 (d, J = 8.0 Hz, 4H), 2.89 (t, J = 7.2 Hz, 4H), 1.85 (quin, J = 7.2 Hz 4H), 1.50-1.32 (m, 20H), 0.90 (t, J = 7.2 Hz 6H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) & 156.3, 153.0, 150.1 150.0, 142.1, 140.7, 138.3, 138.1, 135.4, 134.8, 132.2, 132.1, 127.1, 121.0, 120.5, 117.8, 35.8, 32.0, 31.7, 29.6, 29.4, 22.8, 14.7; HRMS  $(m/z, FAB^+)$  Calcd for C<sub>64</sub>H<sub>66</sub>N<sub>10</sub>O<sub>4</sub>Zn 1102.4560, found 1102.4563.

### References

- A. Osuka, S. Marumo, N. Mataga, S. Taniguchi, T. Okada, I. Yamazaki, Y. Nishimura, T. Ohno and K. Nozaki, J. Am. Chem. Soc. 1996, 118, 155.
- 2 F.-C. Fang, C.-C. Chu, C.-H. Huang, G. Raffy, A. Del Guerzo, K.-T. Wong and D. M. Bassani, *Chem. Commun.* 2008, 6369.



**Figure. S1** TEM of **2** prepared in THF/MeOH (10:1, v/v), showing hollow sphere structures.

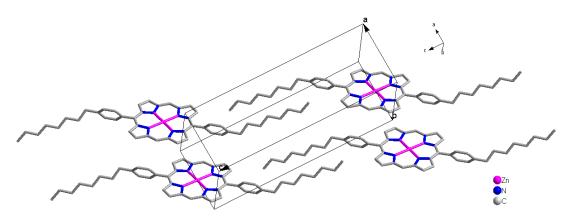
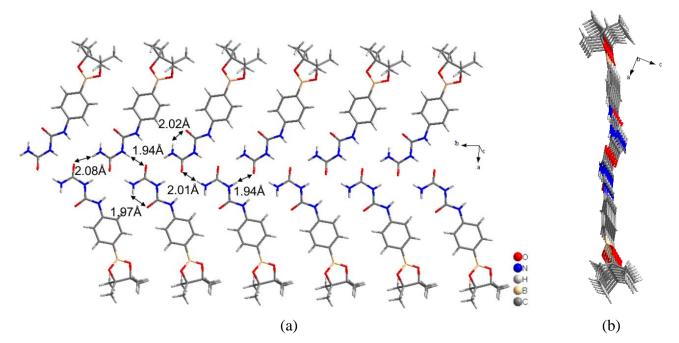


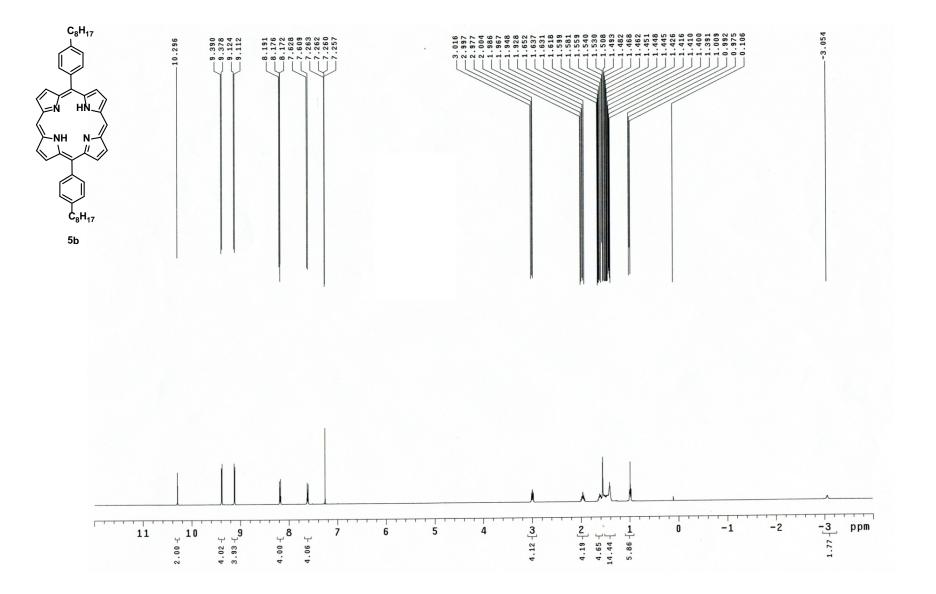
Figure. S2 Crystal packing of compound 1b in a unit cell. The hydrogen atoms are omitted for clarity.

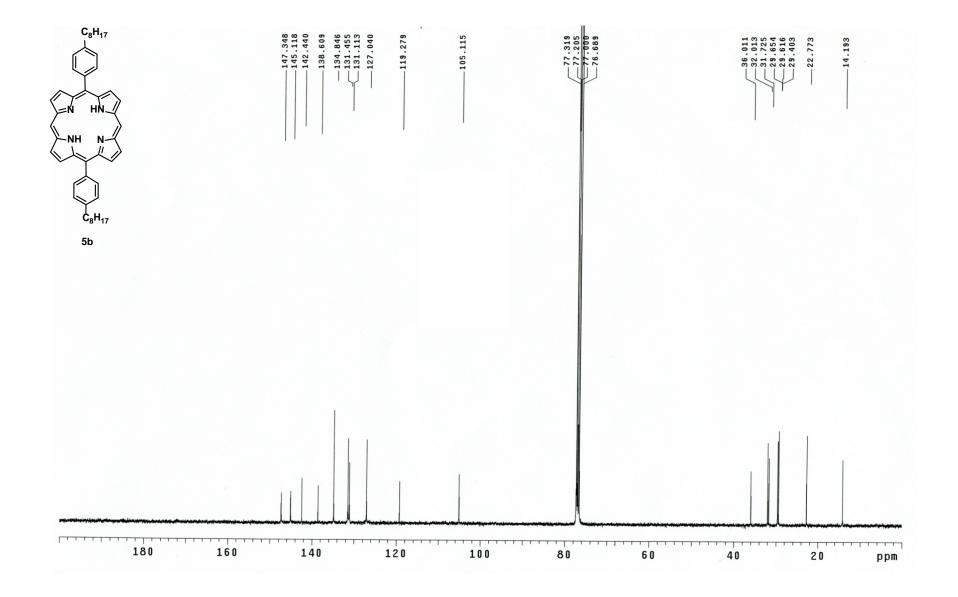


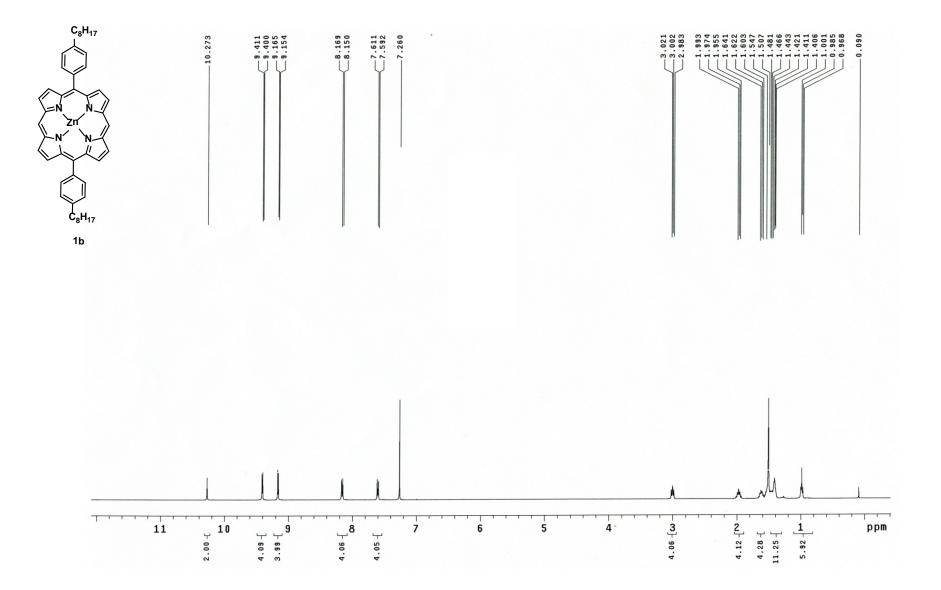
**Figure. S3** Crystal packing of 4-pinacolatoboronic ester-benzenebiuret (4). The hydrogen atoms are omitted for clarity. (a) Top view of *ab*-plane. (b) Side view from *b*-axis.

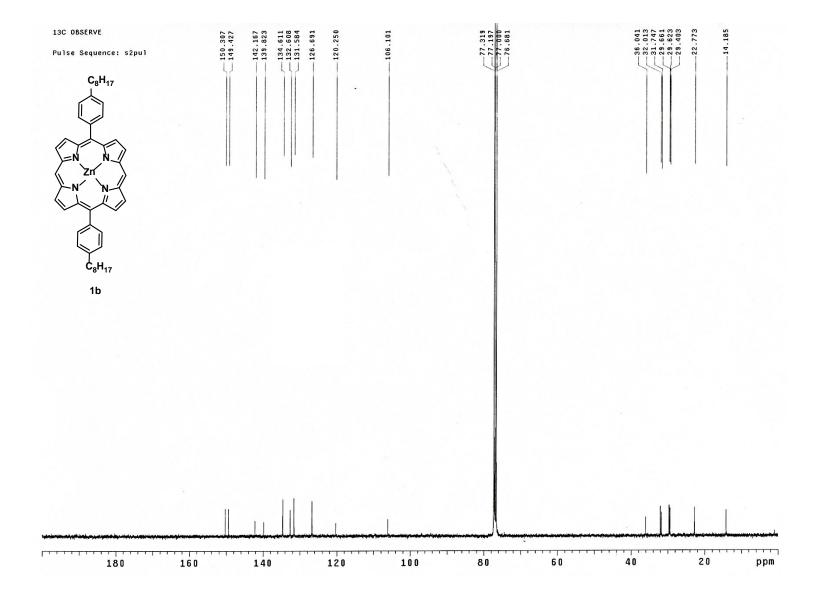
Table S1. The crysta	l data of <b>1b</b> and <b>4</b>
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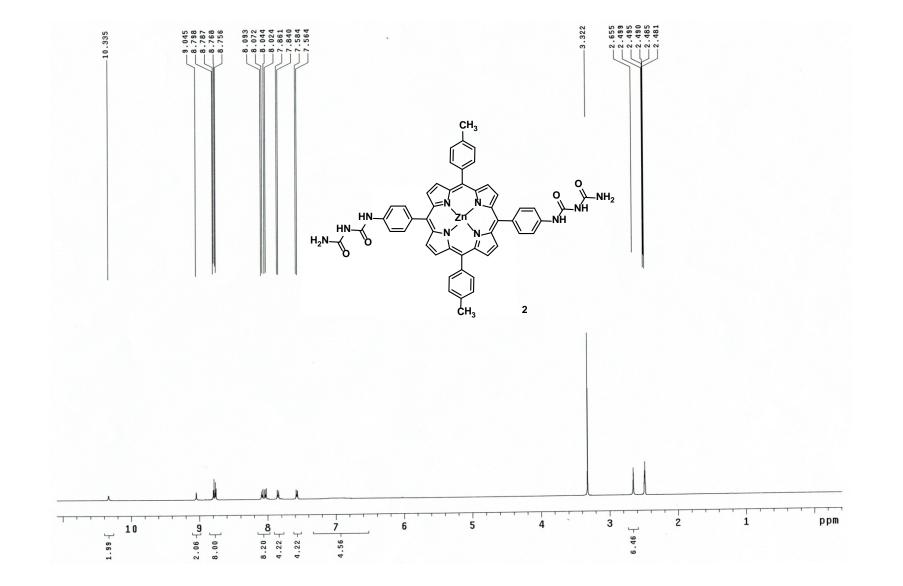
	1b	4	
Empirical formula	$C_{48}H_{52}N_4Zn$	$C_{14}H_{20}BN_{3}O_{4}$	
Formula weight	750.31	305.14	
Crystal dimensions/mm <sup>3</sup>	0.30×0.15×0.10	0.25×0.20×0.15	
Crystal system	Triclinic	Orthorhombic	
Space group	P-1	Pca2(1)	
a/Å	8.5569(2)	19.8063(6)	
b/Å	9.8055(2)	7.1749(3)	
c/Å	23.4743(6)	22.0931(8)	
α (deg)	81.298(2)	90.00	
β (deg)	89.186(2)	90.00	
γ (deg)	89.346(2)	90.00	
Cell volume/ $Å^3$	1946.65(8)	3139.6(2)	
Z	2	8	
Density (calc)/g cm <sup>-3</sup>	1.280	1.291	
F(000)	796	1296	
Temperature/K	200(2)	150(2)	
Wavelength/Å	1.54178	1.54178	
No. of reflns collected	12010	9810	
No. of indep reflns $(R_{int})$	12010(0.0000)	4131(0.0395)	
$R(F)$ , $wR2$ [I>2 $\sigma$ (I)]	0.0473, 0.1254	0.0544, 0.1423	
R(F), $wR2$ (all data)	0.0631, 0.1387	0.0581, 0.1564	



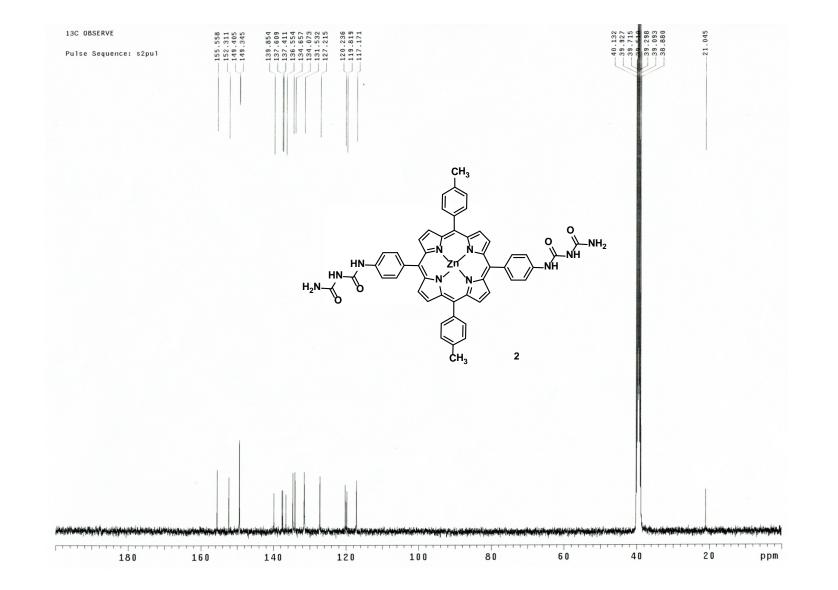


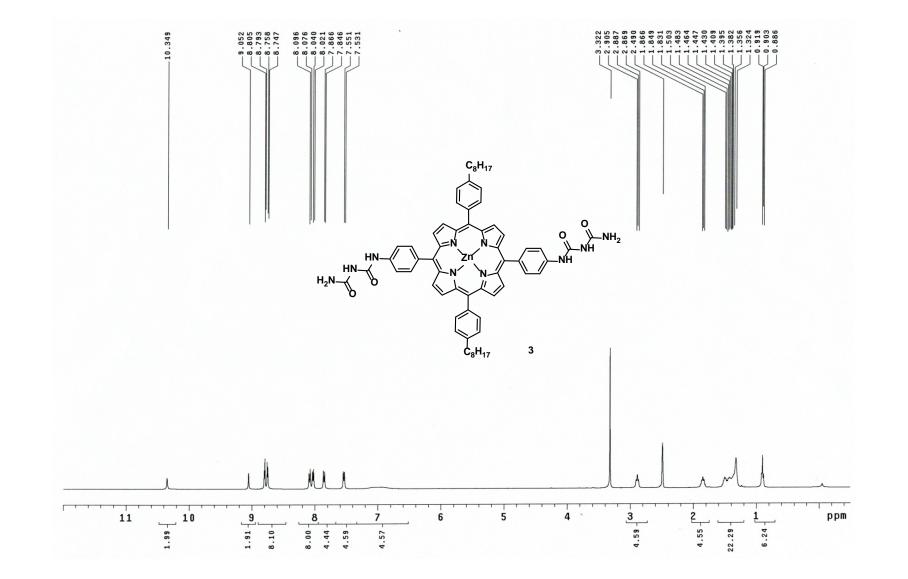






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