Iminium ion catalysis: the enantioselective Friedel-Crafts alkylation-acetalization cascade of naphthols with α , β -unsaturated cyclic ketones

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Supplementary Information

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General informations.

The ¹H and ¹³C NMR spectra were recorded on a Varian inova 300, at 300 MHz and 75 MHz respectively, Varian mercury 400, at 400 MHz and 100 MHz respectively, or Varian inova 600, at 600 MHz and 150 MHz respectively. The chemical shifts (δ) for ¹H and ¹³C are given in ppm relative to residual signals of the solvents (CDCl₃, DMSO- d_6). The following abbreviations are used to indicate the multiplicity: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; bs, broad signal. CDCl₃ was passed over a short pad of alumina before use. Coupling constants are given in Hz. When 2D-NMR were not performed, the carbon types were determined from DEPT ¹³C NMR experiments. NOE spectra were recorded using the DPFGSE-NOE sequence¹ using a mixing time of 2.00 s and "rsnob" 20 - 50 Hz wide selective pulses, depending on the crowding of the spectra region. High Resolution Mass spectra (HRMS) were obtained from the Department of Organic Chemistry "A. Mangini" Mass Spectroscopy facility, on a Thermo-Finnigan MAT 95 XP spectrometer. X-ray data were acquired at the Department of Physical and Inorganic Chemistry X-ray Crystallography facility, on a Bruker APEX-2 difractometer. Optical rotations were measured on a Perkin-Elmer 341 polarimeter and reported as follow: $[\alpha]_{D}^{t}$ (c in g per 100 mL, solvent). Thin Layer Chromatography (TLC) was performed on commercially available Fluka TLC plates on aluminium or PET foils with fluorescent indicator at 254 nm, using UV light as the visualizing agent and an acidic mixture of ceric ammonium molybdate or basic aqueous potassium permangante ($KMnO_4$), and heat as developing agents.

Purification of the products was carried out by flash chromatography (FC) on silica gel (Aldrich, 230-400 mesh) according to the method of Still². Organic solutions were concentrated under reduced pressure on a Büchi rotary evaporator.

Materials

All the commercially available reagents and solvents were used without any further purifications; otherwise, where necessary, they were purified as recommended³. Chiral primary amine catalysts 9-amino(9-deoxy)*epi*-quinine **A** and its *pseudo*-enantiomer 9-amino(9-deoxy)*epi*-quinidine *ent*-**A** were synthesized according to literature procedures⁴.

¹ (a) K. Stott, J. Stonehouse, J. Keeler, T.-L. Hwand and A. Shaka, *J. Am. Chem. Soc.* 1995, **117**, 4199. (b) K. Stott, J. Keeler, Q. N. Van and A. J. Shaka, *J. Magn. Resonance* 1997, **125**, 302. (c) Q. N. Van, E. M. Smith and A. J. Shaka, *J. Magn. Resonance* 1999, **141**, 191. (d) See also: Claridge, T.D.W. *High Resolution NMR Techniques in Organic Chemistry*; Pergamon: Amsterdam, 1999.

² W. C. Still, M. Kahn and A. J. Mitra, *J. Org. Chem.* 1978, **43**, 2923.

³ W. L. F. Armarego and D. D. Perrin, In *Purification of Laboratory Chemicals*, 4th ed.; Butterworth Heinemann: Oxford, 1996.

⁴ (a) S. H.; McCooey and S. J.; Connon *Org. Lett.* 2007, **9**, 599; (b) W. Chen, W. Du, Y.-Z. Duan, Y. Wu, S.-Y. Yang and Y.-C. Chen, *Angew. Chem. Int. Ed.* 2007, **46**, 7667.

All the ketones and the naphthols were purchased from Sigma-Aldrich or Alfa Aesar and used as received. Compounds $1c-d^5$ and $9a-d^6$ were prepared following the literature procedures *Tert-butyl* (3-hydroxynaphthalen-2-yl)carbamate 2d was synthesized according to the literature procedure⁷. Tert-butyl (4-hydroxynaphthalen-1-yl)carbamate 4d was synthesized as follows:

In a flame-dried flask equipped with a condenser and a magnetic stirring bar, triethylamine (0.344 ml, 1 eq.) was added to a 0.5 M solution of 4-amino-1-naphthol hydrochloride (2.56 mmol, 0.5 g, 1 eq.) in anhydrous THF (5.11 ml). The reaction mixture was left to stir at r.t. for 5 min. Then di-*tert*-butyl dicarbonate (0.558 g, 1 eq.) was added as a solid and the reaction mixture was stirred and refluxed for three days. Then it was cooled to room temperature and the solvent was removed in vacuo. The crude product was purified through flash chromatography on silica gel (Hex/EtOAc 70:30) to give a violet solid, which was crystallized from Et₂O/Hex to afford **4d** in 57% yield as a pink solid.

¹**H NMR (300 MHz, CDCl₃)**⁸: δ 1.58 (s, 9H), 6.41 (d, 1H, *J* = 7.8 Hz), 6.49 (bs, 1H), 6.71 (bs, 1H), 7.22 (d, 1H, *J* = 8.1 Hz), 7.40 (t, 1H, *J* = 8.2 Hz), 7.51 (dt, 1H, *J*_a = 8.2 Hz, *J*_b = 1.5 Hz), 7.84 (d, 1H, *J* = 8.4 Hz), 8.05 (d, 1H, *J* = 8.4 Hz).

Determination of diastereomeric ratios and enantiomeric purity.

Diastereomeric ratios was determined by ¹H NMR spectroscopy of the crude product. Enantiomeric excesses were determined, after purification, through HPLC analysis on chiral stationary phase performed on an Agilent 1100-series instrumentation using Daicel Chiralpak AD-H, Daicel Chiralpak AS-H, Daicel Chiralcel OD-H, Daicel Chiralcel OJ-H, Phenomenex Lux-Amilose 2 and Phenomenex Lux-Cellulose 2 columns. Racemic samples of compounds **3ab**, **3ac**, **5ab**, **5ac**, **10aa** were obtained performing the reaction with *p*-anisidine 30 mol% and 5-nitrosalicylic acid 60 mol% as catalyst combination. All the other racemic samples were prepared by mixing the two product antipodes obtained performing the reaction with catalyst 9-amino(9-deoxy)*epi*-quinine **A** and its *pseudo*-enantiomer 9-amino(9-deoxy)*epi*-quinidine *ent*-A separately.

⁵(a) B.-D. Chong, Y.-I. Ji, S.-S. Oh, J.-D. Yang, W. Baik and S. Koo *J. Org. Chem*. 1997, **62**, 9323; (b) Y. Ergün, N. Bayraktar, S. Patir and G. Okay, *J. Hetrocyclic Chem*. 2000, **37**, 11.

⁶ a) L. Minuti, A. Taticchi, E. Gacs-Baitz and A. Marrocchi, *Tetrahedron*, 1995, **51**, 8953; b) E. Zimmerman and V. Suryanarayan, *Eur. J. Org. Chem.* 2007, 4091.

⁷ S. Kumar, D. Hernandez, B. Hoa, Y. Lee, J.-S. Yang and A. McCurdy, *Org. Lett.*, 2008, **10**, 3761.

⁸ Bachir Latli, *J. Label Compd. Radiopharm.* 2004; **47**, 847.

Optimization data for ketone 1a and β -naphthol 2a

Table 1. Catalyst screening



Entry	1a:2a	Cat. (mol%)	Acid (mol%)	t (h)	Yield %	ee %
1	1.1:1	A (20)	TFA (40)	96	87	49
2	1.1:1	B (20)	TFA (40)	48	12.5	45
3	1.1:1	C (20)	TFA (40)	96	89	45
4	1.1:1	-	-	48	0	-
5	1.1:1	-	TFA (40)	48	0	-
	1.1:1	ent-A (20)	-	48	-	-
7	1.1:1	ent-A (20)	TFA (40)	96	42	45
8	1.1:1	ent-C (20)	TFA (40)	48	60	45
9	1.1:1	ent-A (20)	TFA (60%)	96	96	33
10	3:1	A (20)	TFA (40)	96	63	53



9-amino(9-deoxy)epiquinine





N

ent-C 9-amino(9-deoxy)epi-9-amino(9-deoxy)epidihydroquinine dihydroquinidine

H₂N

Table 2. Acid screening



Entry	Acid	Yield (%)	ee (%)
1	benzoic acid	29	32
2	<i>p</i> -nitrobenzoic acid	58	71
3	3,5-dinitrobenzoic acid	28	56
4	o-nitrobenzoic acid	82	74
5	o-hydroxybenzoic acid	45	74
6	o-fluorobenzoic acid	61	57
7	2-hydroxy-5-nitrobenzoic acid	78	82
8	1-hydroxy-2-naphtoic acid	81	76
9	2-hydroxy-3-nitrobenzoic acid	61	73
10	2-hydroxy-1-naphtoic acid	58	78
11	diphenyl hydrogen phosphate	43	19
12	(R)-BINOL-hydrogenphosphate	12	n.d.
13	p-TsOH	0	-
14	2-mercaptobenzoic acid	57	53
15	Anthranilic acid	63	0

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Table 3. Solvent screening



Entry	Solvent	t (d)	Yield (%)	ee (%)
1	Toluene	3	78	82
2	dry Toluene	3	82	84
3	Water	3	43	46
4	Et ₂ O	3	6	n.d.
5	MeOH	3	0	-
6	EtOAc	3	16	n.d.
7	THF	3	0	-
8	CHCl ₃	2	52	80
9	Hexane	2	18	52
10	dioxane	2	12	n.d.
11	MTBE	2	5	n.d.
12	HFIP	2	0	-
13	Fluorobenzene	2	46	80
14	DCM	3	47	80
15	p-Xylene	3	43	68
16	Clorobenzene	3	89	79
17	1, 2-DCE	3	24	76
18	Toluene/Brine 1:1	3	83	73

Conformational analysis and absolute configuration determination.

Compounds 10

Good crystals suitable for X-ray diffraction were obtained for compound **10de** by slow evaporation of a methanol solution. The anomalous scattering determination of the absolute configuration was possible thanks to the presence of the chlorine atom. The *S* configuration was determined for the selected crystal, and its relationship to the major enantiomer obtained with ent-A catalyst, was confirmed by means of enantioselective HPLC analysis of the very same crystal used for X-ray analysis (this was not straightforward, since the crystals were obtained from a 81% ee mixture of enantiomers). The crystal cell contained two conformations of the *S* enantiomers, that were different in the orientation of the OMe group on the naphthalene ring (see below for refinement details).



Figure S1: X-Ray structure of **10de**. Two different conformations with the same *S* absolute configuration represent the asymmetric unity.

Since the prepared compounds belong to four different classes, a straight relationship of the stereochemical course of the reaction is likely, but in principle it could not be safely assumed. However, despite many attempts, enantiopure crystals of other compounds containing a suitable heavy atom could not be obtained. In the remaining cases, the X-ray analysis could not determine the absolute configuration. For these reason we switched to a different approach based on conformational analysis and chirooptical methods.

The determination of the absolute configurations (AC) of chiral molecules using chiroptical techniques like optical rotation (OR), electronic circular dichroism (ECD), and vibrational circular dichroism (VCD) has gained feasibility and reliability because of the development of reliable methods for the prediction of these properties based on density functional theory (DFT) and on its Time-Dependent formalism (TD-DFT).⁹ In the present case the theoretical calculation of ECD spectra was

⁹ For recent examples: H. Hussain, K. Krohn, I. Ahmed, S. Draeger, B. Schulz, S. Di Pietro and G. Pescitelli *Eur. J. Org. Chem.* 2012, 1783; X-F Hou, S. Yao, A. Mándi, T. Kurtán, C-P. Tang, C-Q. Ke, X-Q. Li and Y. Ye. *Org. Lett.* 2012, **14**, 460; Y-S. Cai, T. Kurtán, Z-H. Miao, A. Mándi, I. Komáromi, H-L. Liu, J. Ding, and Y-W Guo *J. Org. Chem.* 2011, **76**, 1821; M, Woźnica, A. Butkiewicz, A. Grzywacz, P. Kowalska, M. Masnyk, K. Michalak, R. Luboradzki, F. Furche, H. Kruse, S. Grimme and J. Frelek.

selected for the absolute configuration assignment. From a conformational point of view, the rigidity of the scaffold of compounds **3**, **5**, **10** and **11** reduces the number of conformations to be considered and thus simplify the conformational analysis.¹⁰ In addition to this, the configuration assignment by chirooptical methods of a sample compound of the **10** series can provide information on the reliability of this method applied to the present compounds.

A preliminary conformational search on **10aa** was carried out using Monte Carlo method together with the MMFF94 molecular mechanics force field (as implemented in Titan 1.0.5, Wavefunction inc.). All the conformations within a 10 kcal/mol window were then optimized using DFT at the B3LYP/6-31G(d) level¹⁰¹¹, the harmonic vibrational frequencies of each conformation were calculated at the same level to confirm their stability (no imaginary frequencies were observed) and to evaluate the free energy of each conformation.



Figure S2: 3D view of the two most stable conformations of **10aa**, calculated at the B3LYP/6-31G(d) level. Energy differences are in kcal/mol and represent ZPE–corrected free energies in standard conditions.

After DFT minimization only two conformations were found to be enclosed in the 10 kcal/mol

J. Org. Chem. 2011, **76**, 3306. For reviews see: G. Bringmann, T. Bruhn, K. Maksimenka and Y. Hemberger. *Eur. J. Org. Chem.* 2009, 2717. T.D. Crawford; M.C. Tam and M.L. Abrams, J. Chem. Phys. A 2007, **111**, 12057. For a review on conformational analysis for the AC determination see: A. Mazzanti and D. Casarini, D. WIREs Comput. Mol. Sci. 2012, **2**, 613.

¹⁰ P.L. Polavarapu; E.A. Donahue; G. Shanmugam; G. Scalmani; E.K. Hawkins; C. Rizzo; I. Ibnusaud; G. Thomas; D. Habel and D. Sebastian; *J Phys Chem A* 2011, **115**, 5665.

¹¹ Program Gaussian 09, rev A.02. M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski and D. J. Fox, Gaussian, Inc., Wallingford CT, 2009.

window (Figure S2). These correspond to two conformational diastereoisomers due to the rotation on the indane ring that is almost perpendicular to the napthol ring in the ground states. The electronic excitation energies and rotational strengths have been calculated in the gas phase for the two conformations of **10aa** using TD-DFT with four different methods (functionals) to ascertain if different calculations provide different shapes of the simulated spectra. The simulation were performed with the hybrid functionals BH&HLYP¹² and M06-2X,¹³ the Long-range Correlated LC- ω B97XD that includes empirical dispersion,¹⁴ and CAM-B3LYP that includes long range correction using the Coulomb Attenuating Method.¹⁵. All the calculations employed the 6-311++G(2d,p) basis set, that proved to be sufficiently accurate at a reasonable computational cost.¹⁶



Figure S3: ECD simulations for the two conformations of **10aa**, obtained with the four different functionals and the same 6-311++G(2d,p) basis set.

The rotational strengths were calculated in both length and velocity representation with the resulting values being very similar. For this reason the errors due to basis set incompleteness should be considered very small, or negligible.¹⁷ All the calculations were performed supposing *S* Absolute Configuration, with the results shown in Figure S3: ECD simulations for the two conformations of 10aa, obtained with the four different functionals and the same 6-311++G(2d,p) basis set. The eight simulated spectra are divided into two set of opposite shaped spectra (reddish lines and

¹² In Gaussian 09 the BH&HLYP functional has the form: $0.5*E_x^{HF} + 0.5*E_x^{LSDA} + 0.5*\Delta E_x^{Becke88} + E_c^{LYP}$

¹³ Y. Zhao and D.G. Truhlar, *Theor. Chem. Acc.* 2008, **120**, 215.

¹⁴ J-D. Chai and M. Head-Gordon. *Phys. Chem. Chem. Phys.* 2008, **10**, 6615.

¹⁵ T. Yanai; D. Tew and N. Handy. *Chem. Phys. Lett.* 2004, *393*, 51.

 ¹⁶ G. Cera, M. Chiarucci, A. Mazzanti, M. Mancinelli and M. Bandini. *Org. Lett.* 2012, 14, 1350-1353. A. Mazzanti, T. Calbet, M. Font-Bardia, A. Moyano and R. Rios. *Org. Biomol. Chem.* 2012,10,1645-1652. S. Duce, F. Pesciaioli, L. Gramigna, L. Bernardi, A. Mazzanti and G. Bencivenni. *Adv. Synth. Cat.* 2011, 353, 860.

¹⁷ P.J. Stephens, D.M. McCann, F.J. Devlin, J.R. Cheeseman and M.J Frisch J.Am.Chem.Soc. 2004, **126**, 7514.

bluish/greenish lines), each one corresponding to one of the two conformations of **10aa**. The opposite pattern can be attributed to the excitonic coupling¹⁸ to the two chromophores, i.e. the naphtyl and the indane ring, that have opposite helicity in the two conformations (R^*M^* in conformation **a** and R^*P^* in conformation **b**). Therefore the shape of the simulated spectrum that should be compared with the experimental one strongly depends on the population ratio employed. Unfortunately, the calculated energies of the two conformations are very similar ($\Delta G^\circ = 0.12$ kcal/mol), thus a correct evaluation of the populations¹⁹ is completely unreliable. This imply that the AC assignment is unfeasible without further experimental data.



Figure S4: variable temperature spectra (¹H at 400 MHz in DMSO-d₆) of compound **10aa**. The signal of the CH of the indanone ring is showed. The asterisk marks a solvent impurity.

However, the conformational analysis of **10aa** showed that the indane ring is forced to be perpendicular to the naphtol ring by the steric hindrance exerted by the OH and the *peri* hydrogen (H-

¹⁸a) N. Harada and K. Nakanishi, Acc. Chem. Res. 1972, **5**, 257; (b) N. Harada and K. Nakanishi, Circular Dichroic Spectroscopy: Exciton Coupling in Organic Stereochemistry; University Science Books: Mill Valley, CA, 1983; (C) K. Nakanishi andN. Berova, In Circular Dichroism: Principle and Applications; Berova, N.; Nakanishi, K.; Woody, R.W. Eds; VCH: New York, 2000; Chapter 12, p 337.

¹⁹ G. Mazzeo, E. Giorgio, R. Zanasi, N. Berova and C. Rosini *J. Org. Chem.* 2010, **75**, 4600.

8) If the interconversion barrier is sufficiently high, two observable conformational diastereoisomers could be generated. This is what actually happened. ¹H NMR showed two sets of signals in a 69:31 ratio due to the frozen rotation around the C_{α} -CH bond. When the temperature is raised the two multiplets broadens and show a single peak at +120°C, when the rotation is fast in the NMR time scale, (Figure S4). An energy barrier of 17.8 kcal/mol was derived at the coalescence temperature (+90°C).

NOE spectra where then acquired to determine which conformational diastereoisomer is the more populated. In the **a** diastereoisomer the CH hydrogen of indane is close to the *peri* H-8 hydrogen of the naphthol ring (1.87 Å), whereas in the **b** conformation the CH points towards the OH and it is far from H-8. DPFGSE-NOE²⁰ spectra obtained on saturation of the two CH signals showed that the major conformation has the CH close to H-8 and the minor has the CH close to OH (Figure S5: DPFGSE-NOE spectra of 10aa (600 MHz in DMSO-d6). Bottom: control spectrum. Middle trace. NOE obtained on saturation of the major CH. Top: NOE spectrum obtained on saturation of the minor CH. (in both NOEs the small inverted peak of the second diastereoisomer was due to saturation transfer effects)..



Figure S5: DPFGSE-NOE spectra of **10aa** (600 MHz in DMSO- d_6). Bottom: control spectrum. Middle trace. NOE obtained on saturation of the major CH. Top: NOE spectrum obtained on saturation of the

²⁰ (a) K. Stott, J. Stonehouse, J. Keeler, T. L. Hwand and A. J. Shaka, *J. Am. Chem. Soc.* 1995, **117**, 4199; (b) K. Stott, J. Keeler, Q. N. Van and A. J. Shaka, *J. Magn. Resonance* 1997, **125**, 302. See also: T. D. W. Claridge, *High Resolution NMR Techniques in Organic Chemistry*; Pergamon: Amsterdam, 1999.

minor CH. (in both NOEs the small inverted peak of the second diastereoisomer was due to saturation transfer effects).

In the present case the integration of the NMR spectra and the NOE spectra provide the exact ratio of the two conformational diastereoisomers to be used in the simulation of the experimental ECD spectrum. The simulated ECD spectrum was thus obtained for each model of calculation by taking into account the 69:31 populations ratio experimentally determined by NMR. All the four simulations obtained (Figure S6. Simulations of the experimental ECD spectrum of 10aa (black traces) obtained with different methods of calculation (functionals). Each simulated spectrum (red, blue, green and purple lines) was obtained starting from the spectra obtained for the two conformations weighted using the experimental NMR data. The experimental spectrum of 10aa was obtained in acetonitrile solution (1 10-4 M, 0.2 cm path length). $\Delta \varepsilon$ are expressed in Mol L-1 cm-1. The simulated spectra were vertically scaled to match the experimental intensity, and red shifted to match the experimental peak at 238 nm.) display now a good agreement with the experimental spectrum and the best simulation was obtained with the BH&HLYP functional. Provided the experimental ratio of the two conformations,²¹ the ECD simulations showed to be able to tackle the absolute configuration of **10aa**.

²¹ D. Casarini, L. Lunazzi, M. Mancinelli, A. Mazzanti and P. Scafato *Chirality* 2009, **21**, 16.



Figure S6. Simulations of the experimental ECD spectrum of **10aa** (black traces) obtained with different methods of calculation (functionals). Each simulated spectrum (red, blue, green and purple lines) was obtained starting from the spectra obtained for the two conformations weighted using the experimental NMR data. The experimental spectrum of **10aa** was obtained in acetonitrile solution (1 10^{-4} M, 0.2 cm path length). $\Delta \varepsilon$ are expressed in Mol L⁻¹ cm⁻¹. The simulated spectra were vertically scaled to match the experimental intensity, and red shifted to match the experimental peak at 238 nm.

Compounds 11

No compound of the **11** series could be crystallized as single crystals and the assignment of **11bb** was performed by means of same chirooptical methods used for **10aa**.

As Before, also in the case of **11bb** the two conformations found by conformational analysis and subsequent DFT optimizations have the CH pointing towards the OH (**a** conformation) and the CH in position 3 (**b** conformation), respectively. In this case, however, the steric hindrance to the rotation of the indane ring is smaller and the ¹H NMR spectrum recorded at room temperature does not show the signals corresponding to the two conformational diastereoisomers. However, since calculations estimated a very similar energy they should be both populated. For this reason, variable temperature NMR spectra were recorded down to -100°C. Below -60°C the signal of the CH broadened and split at -100°C into two signals with a 90:10 ratio. Following the trend observed for **10aa**, the lower field signal (90%) can be attributed to the conformation in which the CH hydrogen is close to OH. Comfortably, this is also the lowest energy conformation, and also the evaluation of the energy difference matched well the experimental ratio (calculated $\Delta G^{\#}$: 0.56 kcal/mol; experimental: 0.75 kcal/mol at -100°C). By applying Boltzmann statistics, the ratio at room temperature is 78:22.



Figure S7: 3D view of the two stable conformations of **11bb**, calculated at the B3LYP/6-31G(d) level. Energy differences are in kcal/mol and represent ZPE–corrected free energies in standard conditions.

The rotational strengths and electronic excitation energies were calculated in the gas phase for the two conformations of **11bb** using TD-DFT and the same four different methods used for **10aa**. All the calculations employed the 6-311++G(2d,p) basis set and supposing *S* Absolute Configuration (Figure S8. TD-DFT simulations for the two conformations of **11bb**, obtained with the four functionals and the 6-311++G(2d,p) basis set.).

As for **10aa**, the opposite helicity generated by the out-of-plane disposition of the indane ring yields two set of calculated ECD spectra with opposite pattern. However, when the final simulated spectra are obtained using the population ratio determined by low-temperature NMR, the agreement with the experimental trace is very good (Figure S9), and the *S* absolute configuration can be assigned to compound **11bb**. In addition to this, the agreement of the four methods and the similarity of the spectra enhances the reliability of the assignment.



Figure S8. TD-DFT simulations for the two conformations of **11bb**, obtained with the four functionals and the 6-311++G(2d,p) basis set.



Figure S9. Simulations of the experimental ECD spectrum of **11bb** (black traces) obtained with different methods of calculation. Each simulated spectrum (red, blue, green and purple lines) was obtained starting from the spectra obtained for the four conformations weighted using Boltzmann statistics at +25°C. The experimental spectrum was obtained in acetonitrile solution (1 10^{-4} M, 0.2 cm path length). $\Delta\epsilon$ are expressed in Mol L⁻¹ cm⁻¹. The simulated spectra were vertically scaled to match the experimental intensity, and red shifted to match the experimental maxima.

Compounds 5

As reported in the text, compounds **5** correspond to the hemiacetalic form. From a conformational point of view, these structures are completely blocked, and only one conformation is supposed to be populated. This correspond to the chair conformation of the cyclohexane part of the bicyclic system.

Compound **5da**, containing a phenyl group in position 4 was selected for the spectroscopic analysis because the presence of a second chromophore that could couple with the naphthalene ring to cause a stronger ECD spectrum.

The analysis of the ¹H spectrum of compound **5da** showed that the phenyl ring in position **4** occupies the equatorial position. The signal of benzylic CH shows a trans-diaxial coupling constant of 12.8 Hz and an axial-equatorial constant of 4.6 Hz. The $2R^*$, $4S^*$, $6S^*$ relative configuration can be therefore assumed in the following discussions (Figure S10. Left: 3D view of the optimized conformation of 5da, calculated at the B3LYP/6-31G(d) level.). It is worth to note that the second diastereoisomer produced by the reaction (**6da**) does not evolve to the hemiacetalic form because in this compound the phenyl ring should occupy the axial position. Although the absolute configuration of **6da** could not be established by anomalous scattering, the X-ray structure confirmed that the two chiral carbons have opposite chirality (thus $3R^*$, $5S^*$).



Figure S10. Left: 3D view of the optimized conformation of **5da**, calculated at the B3LYP/6-31G(d) level. The chair-shaped cyclohexane ring is shaped. Center: ¹H multiplet of the CH in position 4 showing the trans-diaxial and axial-equatorial coupling constants. Right: X-ray structure of **6da**.

To exclude the presence of other low-energy conformations, the conformational search by MM methods was performed as above, but no other conformations were found. As in the previous cases, DFT optimization at the B3LYP/6-31G(d) level provided the final geometry to be used in the TD-DFT simulations. The rotational strengths and electronic excitation energies were calculated in the gas phase by TD-DFT and the same four different methods used for **10aa** and **11bb**. All the calculations employed the 6-311++G(2d,p) basis set and supposed *S* Absolute Configuration at the reaction center. (thus 2R, 4S, 6S) (Figure S11). In this case the simulation of the experimental spectrum is straightforward, since no conformational averaging is needed and the four calculated spectra are very



similar. As shown in Figure S11, the experimental trace is well matched by all the simulation obtained for the 2*R*, 4*S*, 6*S* absolute configuration.

Figure S11. TD-DFT simulations for **5da**, obtained with the four different functionals and the 6-311++G(2d,p) basis set. The black line corresponds to the experimental spectrum obtained in acetonitrile solution (1 10⁻⁴ M, 0.05 cm path length). $\Delta \varepsilon$ are expressed in Mol L⁻¹ cm⁻¹. The simulated spectra were vertically scaled to match the experimental intensity, and red shifted to match the experimental maxima.

Compounds 3

Compounds **3** has the same hemiacetalic scaffold of compounds **5**, thus a second chiral carbon is generated with complete stereocontrol during the cyclization. With respect to the cyclohexane scaffold the two substituents must occupy the 1-3 diaxial positions (Figure S12). Compound **3ab** was selected for the stereochemical assignment.

As in the case of **5**, the bicyclic structure is very rigid, and only a single conformation was found by MM search. In this case, however, a second conformation with a boat-shaped cyclohexane could not be excluded in principle. When optimized by DFT, this conformation showed to be about 5 kcal/mol higher in energy with respect to the conformation found by conformational search, and it can be neglected in the ECD simulations. Compound **3ab** lacks a second chromophore on the cyclohexane ring and the experimental ECD is rather weak. However, the TD-DFT simulations performed assuming the 1*S*, 5*R* absolute configuration satisfactorily followed the experimental trend (Figure S12). It should be pointed out that the same *S* configuration at the reaction center was assigned as in the three previous cases.



Figure S12 3D view of the optimized conformation of **3ab**, calculated at the B3LYP/6-31G(d) level.



Figure S13. TD-DFT simulations for **3ab**, obtained with the four different functionals and the same 6-311++G(2d,p) basis set. The black line corresponds to the experimental spectrum obtained in acetonitrile solution (5 10^{-5} M, 0.2 cm path length). $\Delta \varepsilon$ are expressed in Mol L⁻¹ cm⁻¹. The simulated spectra were vertically scaled to match the experimental intensity, and red shifted to match the experimental maxima



Crystal data for 10de

Molecular formula: $C_{20}H_{15}ClO_4$, MW 338.77. Monoclinic, space group P2₁, a = 8.6325(13), b =8.6693(13), c = 22.017(3), $\beta = 92.146(2)$. V = 1646.6(4) Å³, T = 298(2) °K, Z = 4, $\rho_c = 1.367$ g cm⁻³, F(000) = 2762, graphite-monochromated Mo_{Ka} radiation ($\lambda = 0.71073$ Å), $\mu(Mo_{Ka}) = 0.247$ mm⁻¹, colorless sticks $(0.40 \times 0.15 \times 0.10 \text{ mm}^3)$, empirical absorption correction with SADABS (transmission factors: 0.9078 – 0.9758), 2400 frames, exposure time 20 s, $1.85 \le \theta \le 27.40$, $-11 \le h \le 11$, $-11 \le k \le 11$ 11, $-28 \le l \le 28$, 7373 reflections collected, 5444 independent ($R_{int} = 0.0277$), solution by direct methods (SHELXS97) and subsequent Fourier syntheses, full-matrix least-squares on F_0^2 (SHELX97), hydrogen atoms refined with a riding model except for the hydroxyl hydrogen that was experimentally located; data / restraints / parameters = 7373/1/441, $S(F^2) = 1.044$, R(F) = 0.0634 and $wR(F^2) =$ 0.1258 on all data, R(F) = 0.0454 and $wR(F^2) = 0.1124$ for 5444 reflections with $F_0 > 4\sigma(F_0)$, weighting scheme $w = 1/[\sigma^2(F_0^2) + (0.0646P)^2 + 0.0000P]$ where $P = (F_0^2 + 2F_c^2)/3$, largest difference peak and hole 0.186 and -0.293 e Å⁻³. Flack parameter: 0.02(6). The unit cell contains two different conformation belonging to the same chirality, that are different because of the different disposition of the OMe group. CCDC-893970 CIF file contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data request/cif.



Crystal data for 6da

Molecular formula: $C_{22}H_{20}ClO_2$, MW 316.38. Monoclinic, space group C2, a = 19.582(3), b = 6.4957(10), c = 15.898(2), $\beta = 123.3630(10)$. V = 1688.9(4) Å³, T = 298(2) °K, Z = 4, $\rho_c = 1.244$ g cm⁻³, F(000) = 672, graphite-monochromated Mo_{Ka} radiation ($\lambda = 0.71073$ Å), μ (Mo_{Ka}) = 0.078 mm⁻¹, colorless plates (0.40 × 0.40 × 0.20 mm³), empirical absorption correction with SADABS (transmission factors: 0.9845 – 0.9694), 2400 frames, exposure time 15 s, $1.53 \le \theta \le 27.49$, $-25 \le h \le 25$, $-8 \le k \le 8$, $-20 \le l \le 20$, 9738 reflections collected, 3846 independent ($R_{int} = 0.0201$), solution by direct methods (SHELXS97) and subsequent Fourier syntheses, full-matrix least-squares on F_o^2 (SHELX97), hydrogen atoms refined with a riding model except for the hydroxyl hydrogen that was experimentally located; data / restraints / parameters = 3846/1 / 221, $S(F^2) = 1.025$, R(F) = 0.0440 and $wR(F^2) = 0.0879$ on all data, R(F) = 0.0360 and $wR(F^2) = 0.827$ for 3296 reflections with $F_0 > 4\sigma(F_0)$, weighting scheme $w = 1/[\sigma^2(F_o^2) + (0.0379P)^2 + 0.3168P]$ where $P = (F_o^2 + 2F_c^2)/3$, largest difference peak and hole 0.127 and -0.125 e Å⁻³. Flack parameter: -0.6(11). CCDC-894320 CIF file contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Relative Configuration assignment of compounds ent-5da and ent-6da

The relative stereochemistry of compounds **ent-5da** and **ent-6da** was established by means of NOE experiments. It was found that in **ent-5da** the phenyl group is *trans* to the naphthalene moiety, thus adopting an equatorial position and that in compound **ent-6da** the phenyl group is *cis* to the naphthalene moiety adopting again an equatorial position. This may also explain why structure **ent-6da**, which has the *cis* relative disposition, is only obtained as the "open form". In fact, in the corresponding "closed form" the *cis*-phenyl group would give rise to a strained all-axial structure where the two aromatic moieties will be in close contact.



Compound ent-5da

Irradiation of axial proton H_a generated a visible NOE effect on the two equatorial protons of the vicinal carbons of the cyclohexane ring and on the ortho-protons of the phenyl group.







Figure S15. DPFGSE-NOE spectra obtained for *ent*-5da (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_a . Detail of the aliphatic region.



Figure S16. DPFGSE-NOE spectra obtained for *ent*-5da (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_a . Detail of the aromatic region.

Irradiation of equatorial proton H_b produced a NOE effect on all four protons of the two adjacent methylene groups on the cyclohexane ring and on the C3H of the naphthalene ring.



Figure S17. DPFGSE-NOE spectra obtained for *ent*-5da (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_b .



Figure S18. DPFGSE-NOE spectra obtained for *ent-5da* (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_b . Detail of the aliphatic region.



Figure S19. DPFGSE-NOE spectra obtained for *ent*-5da (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_b . Detail of the aromatic region.

Compound ent-6da

Irradiation of axial proton H_a at 3.13 ppm generated a strong NOE effect on H_b , thus confirming the 1,3-*cis*-diaxial relative disposition of these two protons, and additional NOE effects on the equatorial protons only of the vicinal methylene groups of the cyclohexane ring and on the ortho-protons of the phenyl group.



Figure S20. DPFGSE-NOE spectra obtained for *ent-6da* (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_a . Full spectrum.



Figure S21. DPFGSE-NOE spectra obtained for *ent-6da* (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_a . Detail of the aliphatic region.



Figure S22. DPFGSE-NOE spectra obtained for *ent-6da* (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_a . Detail of the aromatic region.

Irradiation of the axial proton H_b at 3.66 ppm produced visible NOE effect on proton H_a , on the equatorial protons of the vicinal methylene groups of the cyclohexane ring, and on OH and C3H protons of the naphthalene moiety.



Figure S23. DPFGSE-NOE spectra obtained for *ent-6da* (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_b . Full spectrum.



Figure S24. DPFGSE-NOE spectra obtained for *ent-6da* (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_b . Detail of the aliphatic region.



Figure S25. DPFGSE-NOE spectra obtained for *ent*-6da (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_b . Detail of the aromatic region.

Relative Configuration assignment of compound ent-5ca



The relative stereochemistry of compound **ent-5ca** was established by means of NOE experiments. Irradiation of axial proton H_a generated a strong NOE effect on geminal proton H_b as well as a visible NOE on H_c , on one of the two protons on C3 and on the methylene of the ethyl group. As a control, irradiation on equatorial proton H_b in addition to the strong NOE effect on geminal proton H_a , produced only an additional visible NOE effect on H_c . Neither H_a nor H_b show visible NOE effects on C5H. According to these experiments the C5 ethyl group should be in an axial position, trans with respect to the C6-Ar bond.



Figure S26. DPFGSE-NOE spectra obtained for *ent-5ca* (25°C, 600 MHz, CDCl₃) on saturation of proton H_a . Detail of the aliphatic region.



Figure S27. DPFGSE-NOE spectra obtained for *ent-5ca* (25°C, 600 MHz, $CDCl_3$) on saturation of proton H_b . Detail of the aliphatic region.

General procedure for the Friedel-Crafts alkylation-acetalization cascade of naphthols with α , β -unsaturated cyclic ketones

In a screw-capped vial equipped with a Teflon coated magnetic stir bar an anhydrous toluene solution of 9-amino(9-deoxy)*epi*-quinine **A** (0.04 mmol, 0.2 eq., 20 mol%) and 2-hydroxy-5-nitrobenzoic acid (0.08 mmol, 0.4 eq., 40 mol%) was prepared under argon. After 5 min α , β -unsaturated ketone (0.2 mmol, 1 eq.) was added. The resulting yellow solution was stirred for further 5 minutes then β - or α -naphthol (0.22 mmol, 1.1 eq) was added and stirring was continued for 72 hours at 40 °C. the Then the septum was replaced, the vial was refilled with argon, and quickly closed with the screw-cap. The vial was placed at 40°C in a pre-heated oil bath and stirring was continued for 72 hours. Subsequently the reaction mixture was diluted with an 1:1 mixture of Et₂O/DCM, passed through a short plug of silica gel and solvent was evaporated in vacuo to give the crude product which was purified through flash chromatography on silica gel.

(15,5R)-2,3,4,5-tetrahydro-1H-1,5-methanonaphtho[2,1-b]oxocin-5-ol 3aa (Table 2, entry 1): The



reaction was performed following the general procedure on using 9-amino(9-deoxy)*epi*-quinine **A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 80:20) as a white solid in 82% yield and 84% ee. **HPLC analysis**: Phenomenex Lux-Cellulose 2 column; Hex/i-PrOH 95:5, flow rate 0.65 mL/min, T = 20 °C, λ = 230 nm, τ_{minor} = 20.0 min

 $τ_{major}$ = 30.1 min. **[α]**_D³⁰: +106.7 (*c* 3.96, CHCl₃, 84% ee). **HRMS** calculated for C₁₆H₁₆O₂: 240.11503, found 240.11537. ¹H-NMR (300 MHz, CDCl₃): δ 1.43 (tt, *J*_a = 13.7 Hz, *J*_b = 4.8 Hz, 1H), 1.55-1.66 (m, 1H), 1.74 (tt, *J*_a = 13.1 Hz, *J*_b = 3.8 Hz, 1H), 1.86 (td, *J*_a = 14.4 Hz, *J*_b = 5.5 Hz, 1H), 1.89-7.99 (m, 1H), 2.00-2.09 (m 1H), 2.11-2.20 (m, 1H), 2.19 (dd, *J*_a = 12.4 Hz, *J*_b = 2.9 Hz, 1H), 2.92 (bs, 1H), 3.86 (m, 1H), 7.07 (d, *J* = 8.2, 1H), 7.31 (td, *J*_a = 7.5 Hz, *J*_b = 1.1 Hz, 1H), 7.47 (td, *J*_a = 7.7 Hz, *J*_b = 1.5 Hz, 1H), 7.63 (d, *J* = 8.9, 1H), 7.76 (d, *J* = 8 Hz, 1H), 7.84 (d, *J* = 8.5 Hz, 1H). ¹³C-NMR (75 MHz, CDCl₃): δ 19.4 (CH₂), 29.8 (CH₂), 30.6 (CH), 36.3 (CH₂), 39.4 (CH₂), 98.3 (C), 116.3 (C), 117.7 (CH), 121.5 (CH), 123.0 (CH), 126.4 (CH), 128.1 (CH), 128.6 (CH), 128.9 (C), 131.4 (C), 153.0 (C).

(15,5R)-10-bromo-2,3,4,5-tetrahydro-1H-1,5-methanonaphtho[2,1-b]oxocin-5-ol 3ab (Table 2, entry



2): The reaction was performed following the general procedure on using 9-amino(9-deoxy)*epi*-quinine **A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 80:20) as a white solid in 58% yield and 82% ee. **HPLC analysis**:

Phenomenex Lux-Cellulose 2 column; Hex/i-PrOH 90:10, flow rate 0.65 mL/min, T = 20 °C, λ = 230 nm, τ_{minor} = 12.9 min τ_{major} = 15.1 min. [α]_D³⁰: +76.8 (*c* 0.6, CHCl₃, 82% ee). HRMS calculated for C₁₆H₁₅BrO₂: 318.02554, found 318.02526. ¹H-NMR (600 MHz, CDCl₃): δ 1.38 (tq, J_a = 14.1 Hz, J_b = 4.4 Hz, 1H), 1.581.67 (m, 1H), 1.75 (tt, $J_a = 13.2$ Hz, $J_b = 4.4$ Hz, 1H), 1.84-1.93 (m, 2H), 2.02-2.08 (m, 1H), 2.17 (d, J = 13.5 Hz, 1H), 2.20 (dd, $J_a = 12.3$ Hz, $J_b = 2.9$ Hz, 1H), 2.9 (bs, 1H), 3.8 (m, 1H), 7.08 (d, J = 9.2, 1H), 7.51-7.56 (m, 2H), 7.71 (d, J = 9.0 Hz, 1H), 7.90 (d, J = 2.0 Hz, 1H). ¹³C-NMR (150 MHz, CDCl₃): δ 19.3 (CH₂), 29.8 (CH₂), 30.6 (CH), 36.1 (CH₂), 39.3 (CH₂), 98.4 (C), 116.5 (C), 116.6 (C), 118.8 (CH), 123.3 (CH), 127.2 (CH), 129.6 (CH), 130.0 (CH), 132.1 (C), 130.5 (C), 153.3 (C).

(1S,5R)-11-methoxy-2,3,4,5-tetrahydro-1H-1,5-methanonaphtho[2,1-b]oxocin-5-ol 3ac (Table 2,

entry 3): The reaction was performed following the general procedure on using 9-amino(9-deoxy)*epi*-quinine **A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 80:20) as a white solid in 82% yield and 78% ee. **HPLC analysis**:

Phenomenex Lux-Cellulose 2 column; Hex/i-PrOH 9:1, flow rate 0.65 mL/min, T = 20 °C, λ = 230 nm, τ_{minor} = 15.97 min τ_{major} = 24.92 min. [α]_D³⁰: +105 (*c* 1.10, CHCl₃, 78% ee). HRMS calculated for C₁₇H₁₈O₃: 270.125595, found 270.12562. ¹H-NMR (400 MHz, CDCl₃): δ 1.34-1.49 (m, 1H), 1.55-1.67 (m, 1H), 1.73 (tt, J_a = 13.0 Hz, J_b = 3.8 Hz, 1H), 1.85 (dt, J_a = 13.4 Hz, J_b = 5.3 Hz, 1H), 1.90-1.99 (m, 1H), 2.0-2.09 (m, 1H), 2.11-2.22 (m, 2H), 3.06 (brs, 1H), 3.73-3.78 (m, 1H), 3.91 (s, 3H), 6.92 (d, J = 8.7 Hz, 1H), 6.99 (dd, J_a = 8.7 Hz, J_b = 2.5 Hz, 1H), 7.11 (brd, J = 2.4 Hz, 1H), 7.54 (d, J = 8.8 Hz, 1H), 7.66 (d, J = 8.8 Hz, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ 19.4 (CH₂), 29.4 (CH₂), 30.8 (CH), 36.4 (CH₂), 39.4 (CH₂), 55.3 (CH₃), 98.2 (C), 101.0 (CH), 114.7 (CH), 115.2 (CH), 115.3 (C), 124.2 (C), 127.8 (CH), 130.1 (CH), 132.7 (C), 153.6 (C), 158.4 (C).

tert-butyl



((1*R*,5*S*)-5-hydroxy-2,3,4,5-tetrahydro-1*H*-1,5-methanonaphtho[2,1-b]oxocin-7yl)carbamate 3ad (Table 2, entry 4): The reaction was performed following the general procedure on using 9-amino(9-deoxy)*epi*-quinidine *ent*-A as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/EtOAc 9:1) as a pink solid in 94% yield and 73% ee. HPLC analysis: Daicel Chiralpak AD-H column; Hex/i-PrOH 9:1, flow rate 0.7 mL/min, T = 25 °C, λ = 230 nm, τ_{minor} = 10.9 min τ_{major} = 15.3 min. [α]_p²⁹: -56.6 (c 1.39,

CHCl₃, 73% ee). **HRMS** calculated for $C_{21}H_{25}NO_4$ 355.17836, found 355.17882. ¹H-NMR (400 MHz, **CDCl₃**): δ 1.37 (tt, J_a = 13.6 Hz, J_b = 5.0 Hz, 1H), 1.58 (s, 10H), 1.72 (tt, J_a = 13.0 Hz, J_b = 3.8 Hz, 1H), 1.85 (td, J_a = 13.4 Hz, J_b = 5.2 Hz, 1H), 1.85-1.96 (m, 1H), 2.01-2.10 (m 1H), 2.14-2.22 (m, 1H) 2.21 (dd, J_a = 12.0 Hz, J_b = 3.2 Hz, 1H), 3.67 (bs, 1H), 3.78 (m, 1H), 7.27-7.41 (m, 3H) 7.66-7.77 (m, 2H), 8.41 (bs 1H). ¹³C-NMR (75 MHz, CDCl₃): δ 19.3 (CH₂), 28.4 (CH₃), 29.6 (CH₂), 30.7 (CH), 36.4 (CH₂), 39.2 (CH₂), 80.6 (C), 99.5 (C), 113.1 (CH), 116.1 (C), 121.0 (CH), 123.6 (CH), 124.6 (CH), 126.5 (C), 126.8 (C), 128.3 (CH), 129.2 (C), 143.0 (C), 152.8 (C).

(8R,13S)-8,9,10,11,12,13-hexahydro-8,13-methanonaphtho[2,1-b]oxonin-8-ol 3ba (Table 2, entry 5):



The reaction was performed following the general procedure using 9amino(9-deoxy)*epi*-quinine **A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 85:15) to give a yellow solid in 57% yield and 73% ee. **HPLC analysis**: Phenomenex Lux-Cellulose 2 column; Hex/i-PrOH 95:5, flow rate 0.65 mL/min, T = 20 °C, λ =

230 nm, $\tau_{minor} = 20.8 \text{ min } \tau_{major} = 33.1 \text{ min. } [\alpha]_D^{28}$: +20.9 (c 0.54, CHCl₃, 73% ee). HRMS calculated for C₁₇H₁₈O₂ 254.13068, found 254.13044. ¹H-NMR (300 MHz, CDCl₃): δ 1.42-1.68 (m, 4H), 2.00-2.31 (m, 5H), 2.58 (dd, $J_a = 13.8 \text{ Hz}$, $J_b = 1.7 \text{ Hz}$, 1H), 2.79 (s, 1H), 3.73 (m, 1H), 7.07 (d, J = 8.9 Hz, 1H), 7.34 (dt, $J_a = 7.5 \text{ Hz}$, $J_b = 1.1 \text{ Hz}$, 1H), 7.49 (dt, $J_a = 8.3 \text{ Hz}$, $J_b = 1.5 \text{ Hz}$, 1H), 7.65 (d, J = 8.9 Hz, 1H), 7.79 (d, J = 8 Hz, 1H), 7.79 (d, J = 8 Hz, 1H), 7.84 (d, J = 8.5 Hz, 1H). ¹³C-NMR (75 MHz, CDCl₃): δ 22.3 (CH₂), 25.1 (CH₂), 28.7 (CH), 33.8 (CH₂), 37.0 (CH₂), 43.0 (CH₂), 100.4 (C), 117.8 (C), 118.7 (CH), 122.3 (CH), 123.1 (CH), 126.3 (CH), 128.3(CH), 128.7(CH), 129.4 (C), 131.6 (C), 151.2 (C).

(8R,13S)-2-methoxy-8,9,10,11,12,13-hexahydro-8,13-methanonaphtho[2,1-b]oxonin-8-o| 3bc (Table



2, entry 6): The reaction was performed following the general procedure using 9-amino(9-deoxy)*epi*-quinine A as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 85:15) to give a yellow solid in 79% yield and 76% ee. HPLC analysis: Phenomenex Lux-Cellulose 2 column; Hex/i-PrOH 9:1, flow rate 0.7

mL/min, T = 23 °C, λ = 230 nm, τ_{minor} = 16.0 min τ_{major} = 33.7 min. [α]_D³⁰: + 41.5 (*c* 0.54, CHCl₃, 76% ee). **HRMS** calculated for C₁₈H₂₀O₃ 284.14125, found 284.14163. ¹H-NMR (400 MHz, CDCl₃): δ 1.54 (m, 4H), 2.04 (m, 2H), 2.19, (m, 3H), 2.54 (d, *J* = 14.4 Hz, 1H), 2.90 (br. s, 1H), 3.59, (m, 1H), 3.91 (s, 3H), 6.91 (d, *J* = 8.8 Hz, 1H), 7.00 (dd, *J_a* = 8.8 Hz, *J_b* = 2.6 Hz, 1H), 7.10 (d, *J* = 2.1 Hz, 1H), 7.55 (d, *J* = 8.8 Hz, 1H), 7.67 (d, *J* = 9.0 Hz, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ 22.6 (CH₂), 25.4 (CH₂), 29.1 (CH), 33.5 (CH₂), 37.3 (CH₂), 43.3 (CH₂), 55.5 (CH₃), 100.6 (C), 102.4 (CH), 114.9 (CH), 116.5 (CH), 117.2 (C), 124.9 (C), 128.3 (CH), 130.5 (CH), 152.1 (C), 158.5 (C).

(15,25,55)-2-ethyl-2,3,4,5-tetrahydro-1H-1,5-methanonaphtho[2,1-b]oxocin-5-ol 3ca (Table 2, entry



7): The reaction was performed following the general procedure using 9amino(9-deoxy)*epi*-quinidine *ent-A* as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 85:15) to give a yellow oil in 16% yield and 60% ee. **HPLC analysis**: Phenomenex Lux-Amilose 2 column; Hex/i-PrOH 85:15, flow rate 0.4 mL/min, T = 23°C, λ = 230 nm, τ_{minor} = 13.9 min τ_{major} = 15.7 min. **[\alpha]**³²: + 6.7 (*c*

0.285; CHCl₃, 60% ee). HRMS calculated for C₁₈H₂₀O₂ 268.14633, found 268.14658. ¹H-NMR (600 MHz,

CDCl₃): δ 1.11 (t, J = 7.6 Hz, 3H), 1.43 (m, 1H), 1.67, (m, 3H), 1.84 (m, 1H), 1.96 (dd, J_a = 4.0 Hz, J_b = 1.4 Hz, 1H), 2.00 (dd, J_a = 5.34 Hz, J_b = 2.5 Hz, 1H), 2.34 (dd, J_a = 12.7 Hz, J_b = 2.9 Hz, 1H), 2.88 (s, 1H), 3.69 (s, 1H), 7.07 (d, J = 9.1 Hz, 1H), 7.31 (m, 1H), 7.47 (m, 1H), 7.63, (d, J = 9.0 Hz, 1H), 7.76 (d, J = 8.23 Hz, 1H), 7.83 (d, J = 8.5 Hz, 1H). ¹³C-NMR (150 MHz, CDCl₃): δ 12.6 (CH₃), 23.1 (CH₂), 24.2 (CH₂), 30.8 (CH₂), 34.0 (CH), 35.1 (CH₂), 39.0 (CH), 99.5 (C), 117.7 (CH), 117.8 (C), 121.3 (CH), 122.9 (CH), 126.4 (CH), 128.1 (CH), 128.7 (CH), 128.9 (C), 131.3 (C), 152.9 (C).

(2R,6S)-3,4,5,6-tetrahydro-2H-2,6-methanonaphtho[1,2-b]oxocin-2-ol 5aa (Table 3, entry 1): The



reaction was performed following the general procedure using 9-amino(9-deoxy)*epi*-quinine **A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 85:15) to give a white solid in 73% yield and 96% ee. **HPLC analysis**: Phenomenex Lux-Amilose 2 column; Hex/i-PrOH 95:5, flow rate 0.65 mL/min, T = 20 °C, λ = 230 nm, τ_{minor} =

14.5 min τ_{major} = 16.1 min. $[\alpha]_D^{30}$: -28.9 (*c* 3.60, CHCl₃, 96% ee). **HRMS** calculated for C₁₆H₁₆O₂ 240.11503, found 240.11537. ¹H-NMR (600 MHz, CDCl₃): δ 1.33-1.47 (m, 1H), 1.54-1.63 (m, 1H), 1.67-1.81 (m, 2H) 1.85 (dt, 1H, J_a = 13.7 Hz, J_b = 5.5 Hz), 2.05 (m, 1H), 2.14 (dd, 1H, J_a = 12.2 Hz, J_b = 2.6 Hz), 2.18 (d, 1H, J = 13.3 Hz), 2.98 (bs, 1H), 3.28 (m, 1H), 7.11 (d, 1H, J = 8.2), 7.33 (d, 1H, J = 8.2), 7.44 (m, 2H), 7.75 (m, 1H), 8.21 (m, 1H). ¹³C-NMR (150 MHz, CDCl₃): δ 18.7 (CH₂), 31.5 (CH₂), 35.3 (CH), 36.5 (CH₂), 39.1 (CH₂), 99.1 (C), 118.5 (C), 119.4 (CH), 121.7 (CH), 123.8 (C), 125.7 (CH), 125.7 (CH), 126.1 (CH), 127.4 (CH), 133.5 (C), 150.3 (C).

(2R,6S)-8-chloro-3,4,5,6-tetrahydro-2H-2,6-methanonaphtho[1,2-b]oxocin-2-ol 5ab (Table 3, entry



2): The reaction was performed following the general procedure using 9-amino(9deoxy)*epi*-quinine **A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 80:20) to give a white solid in 91% yield and 93% ee. **HPLC analysis**: Daicel Chiralpak AD-H column; Hex/i-PrOH 95:5, flow rate 0.7 mL/min λ = 214 nm, τ_{major} = 18.27 min, τ_{minor} = 19.37 min.

[α]³⁰_D: -14.99 (*c* 0.99, CHCl₃, 93% ee). **HRMS** calculated for C₁₆H₁₅O₂Cl 274.076059, found 274.07611. ¹H-NMR (400 MHz, CDCl₃): δ 1.29-1.46 (m, 1H), 1.53-1.64 (m, 1H), 1.64-1.78 (m, 2H), 1.84 (dt, 1H, J_a = 13.6 Hz, J_b = 5.3 Hz), 2.00-2.23 (m, 3H), 3.17-3.23 (m, 1H), 3.23-3.25 (brs, 1H), 7.16 (s, 1H), 7.49 (m, 1H), 7.55 (m, 1H), 8.15 (m, 1H), 8.22 (m, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ 18.6 (CH₂), 31.3 (CH₂), 35.1 (CH), 36.2 (CH₂), 39. (CH₂), 99.3 (C), 118.9 (C), 122.0 (C), 122.1 (CH), 124.1 (CH), 124.7 (C), 125.7 (CH), 125.8 (CH), 126.7 (CH), 130.2 (C), 149.4 (C).

(2R,6S)-8-methoxy-3,4,5,6-tetrahydro-2H-2,6-methanonaphtho[1,2-b]oxocin-2-ol 5ac (Table 3, entry



3): The reaction was performed following the general procedure using 9amino(9-deoxy)*epi*-quinine **A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 70:30) to give a white solid in 95% yield and 92% ee. **HPLC analysis**: Phenomenex Lux-Cellulose 2 column; Hex/i-PrOH 87:13, flow rate 0.7 mL/min, T = 15 °C, λ = 230 nm, τ_{minor} =

10.63 min τ_{major} = 17.68 min. **HRMS** calculated for C₁₇H₁₈O₃ 270.125595, found 270.12566. ¹**H-NMR (400 MHz, CDCl₃):** δ 1.25-1.40 (m, 1H), 1.43-1.52 (m, 1H), 1.55-1.80 (m, 3H), 1.93-2.02 (m, 2H), 2.02-2.07 (m, 1H), 3.04-3.10 (m, 1H), 3.14 (brs, 1H), 3.84 (s, 3H), 6.31 (s, 1H), 7.31-7.42 (m, 2H), 8.03-8.10 (m, 2H).

¹³C-NMR (100 MHz, CDCl₃): δ 18.8 (CH₂), 31.3 (CH₂), 35.8 (CH), 36.6 (CH₂), 39.2 (CH₂), 55.7 (CH₃), 98.6 (), 104.0 (CH), 117.4 (C), 121.4 (CH), 121.6 (CH), 124.4 (C), 125.0 (CH), 125.1 (C), 125.7 (CH), 144.0 (C), 148.8 (C).

Compound **5ac** (50 mg, 0.185 mmol) has been reacted with TsCl (70.54 mg, 0.37 mmol, 2 equiv) in the presence of Et₃N (52 μ L, 0.37 mmol, 2 equiv.) and DMAP (50% mol, 0.0925 mmol, 11.30 mg) in 4 ml of



dichloromethane for 2 days. The reaction mixture was poured into water (10 ml) and extracted 2 times with 10 ml of dichloromethane. After evaporation of the solvent compound **12** was obtained in 50 yield. $[\alpha]_D^{31}$: -40.5 (*c* 0.44, CHCl₃, 92% ee). ¹H-NMR (400 MHz, CDCl₃): δ 1.51-1.71 (m, 1H), 1.73-1.90 (m, 1H), 1.96-2.17 (m, 2H), 2.27-2.53 (m, 7H), 3.41 (m, 1H), 4.02 (s, 3H), 6.67 (s, 1H), 7.33-7.39 (m, 2H), 7.39-7.47 (m, 2H), 7.75-7.82 (m, 1H), 7.82-7.88 (m, 2H), 8.15-8.22 (m, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ 21.7 (CH₃), 25.4 (CH₂), 31.8 (CH₂), 38.2 (CH), 41.1 (CH₂), 47.6 (CH₂), 55.7 (CH₃), 101.1 (CH), 121.9 (CH), 122.7 (CH), 125.3 (C), 125.7 (CH), 127.3 (CH), 128.2 (CH), 128.8 (C), 130.0 (CH), 133.7 (C), 134.1 (C), 135.9 (C), 145.5 (C), 154.6 (C), 209.9 (C).

tert-butyl



((2R,6S)-2-hydroxy-3,4,5,6-tetrahydro-2H-2,6-methanonaphtho[1,2-b]oxocin-8-yl)carbamate 5ad (Table 3, entry 4): The reaction was performed following the general procedure using 9-amino(9-deoxy)*epi*-quinine A as catalyst and the title compound was obtained in 58% yield determined by ¹H-NMR using
^{OC} CH₂Br₂ as internal standard. Compound 5ad was purified by flash chromatography on silica gel (eluent mixture Hex/Et₂O 60:40) to give a pink

solid in 87% ee. Further purification was carried out through preparative HPLC (AD-H column, Hex/i-PrOH 8:2, flow rate 20 mL/min) to give a white solid. **HPLC analysis**: Daicel Chiralpak AD-H column; Hex/i-PrOH 8:2, flow rate 1 mL/min, T = 25 °C, λ = 230 nm, τ_{major} = 8.6 min, τ_{minor} = 13.3 min. $[\alpha]_D^{30}$: -11.0 (*c* 0.82, CHCl₃, 87% ee). **HRMS** calculated for C₂₁H₂₅NO₄ 355.17836, found 355.17882. ¹H-NMR (600 MHz, CDCl₃): δ 1.42 (tt, 1H, J_a = 13.5 Hz, J_b = 4.5 Hz), 1.54 (s, 10H), 1.70 (tt, 1H, J_a = 13.5 Hz, J_b = 3.7 Hz), 1.79-1.86 (m, 1H), 1.84 (td, 1H, J_a = 13.1 Hz, J_b = 5.3 Hz), 2.01-2.06 (m, 1H), 2.11-2.19 (m, 2H), 2.96 (s, 1H), 3.28 (m, 1H), 6.56 (s, 1H), 7.39-7.54 (m, 3H), 7.80 (d, 1H, J = 8.7 Hz), 8.24 (d, 1H, J = 8.1 Hz). ¹³C-NMR (100 MHz, CDCl₃): δ 17.8 (CH₂), 27.4 (CH₃), 30.3 (CH₂), 34.3 (CH), 35.5 (CH₂), 38.1 (CH₂), 79.3 (C), 98.1 (C), 117.2 (C), 119.8 (C), 121.3 (CH), 122.4 (C), 123.0 (C), 123.9 (C), 124.2 (2×CH), 125.0 (2 CH), 146.9 (C).

(2S,7R)-2,3,4,5,6,7-hexahydro-2,7-methanonaphtho[1,2-b]oxonin-2-ol and 3-(1-hydroxynaphthalen-



2-yl)cycloheptanone 5ba (Table 3, entry 5): The reaction was performed following the general procedure using 9-amino(9-deoxy)*epi*-quinidine *ent-A* as catalyst. The title compound was obtained in 34% yield determined by ¹H-NMR using CH_2Br_2 as internal standard together with a 8% of Friedel-Crafts alkylation compound. Compound **5ba** was purified from crude mixture by flash chromatography on silica

gel (eluent mixture Hex/Et₂O 8:2) in 90% ee. **HPLC analysis:** Phenomenex Lux-Amilose 2 column; Hex/i-PrOH 95:5, flow rate 0.7 mL/min, T = 20 °C, λ = 230 nm, τ_{minor} = 30.3 min τ_{major} = 34.5 min. [α]³⁰: +48.0 (*c* 0.375; CHCl₃, 90% ee). **HRMS** *calcd*. for C₁₆H₁₆O₂ 240.11503, found 240.11537. ¹H NMR (300 **MHz, CDCl₃) of 5ca with traces of Friedel-Crafts alkylation compound:** δ 1.29-1.38 (m, 1H), 1.49-1.55 (m 2H), 1.61-1.71 (m, 1H), 1.83-1.93 (m 1H), 2.03-2.09 (m, 1H), 2.09-2.17 (m, 1H), 2.17-2.25 (m, 2H), 2.58 (d, 1H, *J* = 13.6 Hz), 2.99 (bs, 1H), 2.25 (m, 1H), 7.23 (d, 1H, *J* = 8.1 Hz), 7.40 (d, 1H, *J* = 8.2 Hz), 7.43-7.48 (m, 2H), 7.75-7.79 (m, 1H), 8.22-8.27 (m, 1H). ¹³C-NMR (150 MHz, CDCl₃) of 5ca with traces of Friedel-Crafts alkylation compound: δ 22.0 (CH₂), 25.0 (CH₂), 32.3 (CH), 34.3 (CH₂), 37.0 (CH₂), 42.4 (CH₂), 100.3 (C), 118.5 (C), 119.1 (CH), 120.8 (CH), 124.1 (C), 124.2 (CH), 124.8 (C), 125.3 (CH), 126.3 (CH), 132.3 (C), 147.4 (C).

(2R,5R,6R)-5-ethyl-3,4,5,6-tetrahydro-2H-2,6-methanonaphtho[1,2-b]oxocin-2-ol 5ca (Table 3, entry



6): The reaction was performed following the general procedure using 9-amino(9-deoxy)*epi*-quinine **A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 9:1) to give a yellow oil in 27%and 89% ee. **HPLC analysis**: Daicel Chiralcel OJ-H; flow rate 0.7 ml/min, Hex/i-PrOH 9:1, $\lambda = 230$ nm, 23°C, $\tau_{minor} = 24.5$ min $\tau_{major} = 32.3$ min. **[\alpha]**³¹_D: - 60.1 (*c* 0.27;

CHCl₃, 89% ee). **HRMS** calculated for $C_{18}H_{20}O_2$ 268.14633, found 268.14658. ¹H-NMR (400 MHz, **CDCl₃**): δ 1.04 (t, 3H, *J* = 7.6 Hz), 1.48-1.46 (m, 1H), 1.48-1.70 (m, 5H), 1.85-1.91 (m, 1H), 1.96-2.02 (m, 2H), 2.28 (dd, 1H, *J*_a = 12.6 Hz, *J*_b = 2.7 Hz), 2.95 (s, 1H), 3.10 (m, 1H), 7.14 (d, 1H, *J* = 8.2 Hz), 7.35 (d, 1H, *J* = 8.2 Hz), 7.41-7.49 (m, 2H), 7.74-7.79 (m, 1H), 8.20-8.25 (m, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ 12.7 (CH₃), 22.1 (CH₂), 24.2 (CH₂), 31.2 (CH₂), 34.7 (CH₂), 39.1 (CH), 41.2 (CH), 99.2 (C), 119.4 (CH), 120.2 (C), 121.6 (CH), 123.8 (C), 125.0 (CH), 125.6 (CH), 126.1 (CH), 127.4 (CH), 133.4 (C), 150.1 (C).
(2R,4S,6S)-4-phenyl-3,4,5,6-tetrahydro-2H-2,6-methanonaphtho[1,2-b]oxocin-2-ol 5da and (3S,5R)-3-(1-hydroxynaphthalen-2-yl)-5-phenylcyclohexanone 6da (Table 3, entries 7-8): The reaction was performed following the general procedure using 9-amino(9-deoxy)*epi*-quinine **A** as catalyst. Compounds 5da and 6ad were isolated after 5 days by flash chromatography on silica gel (eluent



mixture Hex/Et₂O 70:30) in 26% yield and 93% ee and 31% yield and 98% ee respectively (57% overall yield). **HPLC analysis:** for compound **5da** Daicel Chiralpak AD-H column; Hex/i-PrOH 95:5, flow rate 0.75 mL/min λ = 214 nm, τ_{major} = 32.69 min, τ_{minor} = 35.13 min; for compound **6da** Daicel Chiralpak AD-H column; Hex/i-

PrOH 80:20, flow rate 0.75 mL/min λ = 214 nm, τ_{major} = 14.39 min, τ_{minor} = 16.19 min. $[\alpha]_{p}^{30}$ for 5da: -133.8 (c 0.447; CHCl₃, 93% ee); $[\alpha]_{p}^{30}$ for 6da: +19.2 (c 0.47; CHCl₃, 98% ee). HRMS calcd. for C₂₂H₂₀O₂ 316.14633, found 316.14661. ¹H-NMR (400 MHz, CDCl₃) 5da: δ 1.92 (dt, 1H, J_a = 12.7 Hz, J_b = 3.2 Hz), 2.02-2.19 (m, 3H), 2.27 (dd, 1H, J_a = 12.3 Hz, J_b = 2.8 Hz), 2.38-2.46 (m, 1H), 2.84 (m, 1H), 3.13 (brs, 1H), 3.36-3.42 (m, 1H), 7.08-7.13 (m, 2H), 7.14-7.19 (m, 2H), 7.21-7.25 (m, 2H), 7.38 (d, 1H, J = 8.3 Hz), 7.44-7.51 (m, 2H), 7.77-7.82 (m, 1H), 8.23-8.29 (m, 1H). ¹³C-NMR (100 MHz, CDCl₃) 5da: δ 35.3 (CH), 36.1 (CH₂), 36.5 (CH), 39.6 (CH₂), 46.4 (CH₂), 99.3 (C), 118.9 (C), 119.7 (CH), 121.7 (CH), 123.8 (C), 125.2 (CH), 125.8 (CH), 126.0 (CH), 126.3 (CH), 127.0 (CH), 127.4 (CH), 128.4 (CH), 133.6 (C), 144.3 (C), 150.1 (C). ¹H-NMR (400 MHz, CDCl₃) 6da: δ 2.20-2.35 (m, 2H), 2.60-2.80 (m, 4H), 3.21 (m, 1H), 3.74 (m, 1H), 5.93 (brs, 1H), 7.21-7.31 (m, 3H), 7.32-7.42 (m, 3H), 7.43-7.54 (m, 3H), 7.81 (m, 1H), 8.01 (m, 1H). ¹³C-NMR (100 MHz, CDCl₃) 6da: δ 37.3 (CH), 39.3 (CH₂), 44.0 (CH), 47.3 (CH₂), 48.6 (CH₂), 120.1 (CH), 121.1 (CH), 123.9 (C), 124.1 (CH), 124.5 (C), 125.8 (CH), 125.9 (CH), 126.6 (CH), 126.9 (CH), 128.1 (CH), 128.8 (CH), 133.3 (CH), 143.9 (CH), 147.6 (CH), 210.8 (C).

(S)-3-(1-hydroxynaphthalen-2-yl)cyclopentanone 7 (Table 3, entry 9): The reaction was performed



following the general procedure using 9-amino(9-deoxy)*epi*-quinine **A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 70:30) to give a yellow solid in a 63% yield and 90% ee. **HPLC analysis:** Daicel Chiralpak AS-H; Hex/i-PrOH 85:15, flow rate 0.7 ml/min, $\lambda = 230$ nm, 25°C, $\tau_{major} = 23.1$ min, $\tau_{minor} = 29.2$ min. $[\alpha]_{D}^{29}$: -56.0 (*c*

0.67; CHCl₃, 90% ee). **HRMS** *calcd*. for C₁₆H₁₆O₂ 226.09938, found 226.09952. ¹H-NMR (400 MHz, **CDCl₃)**: δ 2.06-2.30 (m, 2H), 2.30-2.60 (m, 4H), 2.75 (dd, 2H, J_a = 17.8 Hz, J_b = 7.8 Hz), 3.84-3.98 (m, 1H), 5.51 (s, 1H), 7.35 (d, 1H, J = 8.6 Hz), 7.44-7.57, (m, 3H), 7.84 (d, 1H, J = 8.1 Hz), 7.99 (dm, 1H, J = 8.1 Hz). ¹³C-NMR (100 MHz, CDCl₃): δ 28.7 (CH₂), 34.2, 35.4 (CH₂), 35.9, 37.8 (CH₂), 37.9, 39.2, 43.8 (CH₂), 118.7 (CH), 120.0 (CH), 120.4 (C), 122.0 (C), 123.3 (C), 123.4 (CH), 124.1, 124.4, 124.6, 124.8 (CH), 126.5, 127.2 (CH), 132.9 (CH), 147.2 (C), 218.2 (C).

(S)-4-(1-hydroxynaphthalen-2-yl)pentan-2-one 8 (Table 3, entry 10): The reaction was performed



following the general procedure using 9-amino(9-deoxy)*epi*-quinidine *ent-A* as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hex/Et₂O 70:30) to give a yellow solid in 71% yield and 44% ee. **HPLC analysis:** Daicel Chiralpak AD-H; flow rate 0.7 ml/min, Hex/i-PrOH 96:4, λ = 230 nm, 25°C, τ_{major} = 47.7 min, τ_{minor} = 52.5 min. [α]³¹_D: + 25 (*c* 1.39; CHCl₃, 44%

ee). **HRMS** *calcd*. for C₁₆H₁₆O₂ 228.11503, found 228.11533. ¹H-NMR (400 MHz, CDCl₃): δ 1.39 (d, 3H, J = 7.4 Hz), 2.13 (s, 3H), 2.73 (dd, 1H, $J_a = 16.6$ Hz, $J_b = 8.7$ Hz), 2.89 (dd, 1H, $J_a = 16.5$ Hz, $J_b = 5.2$ Hz), 4.12 (m, 1H), 5.65 (s, 1H), 6.77 (d, 1H, $J_a = 7.6$ Hz), 7.18 (d, 1H, J = 7.5 Hz), 7.45-7.60 (m, 2H), 8.10 (d, 1H, J = 8.5 Hz), 8.26 (d, 1H, J = 8.3 Hz). ¹³C-NMR (100 MHz, CDCl₃): δ 21.4 (CH₃), 29.3 (CH), 30.4 (CH₃), 51.8 (CH₂), 108.0 (CH), 122.4 (CH), 122.5 (CH), 122.6 (C), 122.9 (CH), 124.8 (CH), 124.9 (C), 126.5 (CH), 132.1 (C), 134.3 (C), 150.1 (C), 208.5 (C).

(S)-3-(2-hydroxynaphthalen-1-yl)-2,3-dihydro-1H-inden-1-one 10aa (Table 4, entry 1): The reaction



was performed following the general procedure on using 9-amino(9-deoxy)*epi*quinidine *ent-A* as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hexane/AcOEt 80:20) as a white solid in 63% yield and 95% ee. **HPLC analysis**: Daicel Chiralcel OD-H, Hex/i-PrOH 90:10, flow rate 0.7 ml/min, $\lambda = 254$ nm: $\tau_{major} = 16.32$ min, $\tau_{minor} = 21.18$ min. **HRMS** calculated for C₁₉H₁₄O₂ 274.09938, found 274.09907. [α]³⁰_p = -118.9 (*c* 0.45;

DMSO, 95% ee). ¹H-NMR (400 MHz, DMSO d_6): 70:30 mixture of conformational diastereosiomers A (major) and B (minor); δ 2.84 (dd, $J_a = 19.4$ Hz, $J_b = 5.0$ Hz, $1H_B$), 2.95 (dd, $J_a = 18.6$ Hz, $J_b = 3.4$ Hz, $1H_A$), 3.08 (dd, $J_a = 18.6$ Hz, $J_b = 8.2$ Hz, $1H_A$), 3.19 (dd, $J_a = 19.2$ Hz, $J_b = 8.3$ Hz, $1H_B$), 5.51 (m, $1H_A$), 5.70 (m, $1H_B$), 6.77 (d, J = 9.0 Hz, $1H_B$), 7.0 (dd, $J_a = 16.7$ Hz, $J_b = 7.7$ Hz, $2H_A$), 7.05-7.09 (m, $2H_B$), 7.12-7.16 (m, $1H_B$), 7.30 (d, J = 9.1 Hz, $1H_B$), 7.32-7.41 (m, $2H_A$), 7.43-7.52 (m, $1H_A + 1H_B$), 7.53-7.58 (m, $1H_A + 1H_B$), 7.67 (d, J = 7.7 Hz, $1H_A$), 7.71 (d, J = 8.7 Hz, $1H_A$), 7.75 (d, J = 8.9 Hz, $1H_B$), 7.77 (d, J = 8.1 Hz, $1H_B$), 7.81 (d, J = 7.6 Hz, $1H_B$), 7.84 (d, J = 8.2 Hz, $1H_A$), 8.38 (d, J = 8.7 Hz, $1H_A$). 9.43 (s, $1H_A$), 10.08 (s, $1H_B$). ¹³C-NMR (100 MHz, DMSO d_6): δ 34.55 (CH), 35.5 (CH), 42.8 (CH₂), 43.4 (CH₂), 117.6 (CH), 117.9 (C), 118.6 (CH), 119.6 (C), 122.0 (CH), 122.1 (CH), 122.2 (CH), 122.4 (CH), 122.4 (CH), 123.2 (CH), 125.3 (CH), 125.6 (CH), 131.6 (C), 133.6 (C), 134.4 (CH), 135.3 (CH), 136.3 (C), 136.6 (C), 152.6 (C), 153.6 (C), 158.8 (C), 159.6 (C), 205.2 (C), 206.3 (C).

(S)-3-(2-hydroxynaphthalen-1-yl)-6-methyl-2,3-dihydro-1H-inden-1-one 10ba (Table 4, entry 3): The



reaction was performed following the general procedure on using 9-amino(9deoxy)*epi*-quinidine *ent-A* as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hexane/AcOEt 80:20) as a white solid in 97% yield and 90% ee. **HPLC analysis**: Daicel Chiralcel OJ-H, Hex/i-PrOH 90:10, flow rate 1.0 ml/min, λ = 254 nm: τ_{minor} = 26.63 min, τ_{major} = 35.26 min. **HRMS** calculated for C₂₀H₁₆O₂ 288.11503, found 288.11535. [α]³¹_p = -288.3 (*c*

0.80; DMSO, 90% ee). ¹H-NMR (400 MHz, DMSO-*d₆*): 65:35 mixture of conformational diastereosiomers *A* (major) and *B* (minor); δ 2.35 (s, 3H_{*A*}), 2.38 (s, 3H_{*B*}), 2.83 (dd, *J_a* = 19.3 Hz, *J_b* = 4.9 Hz, 1H_{*B*}), 2.96 (dd, *J_a* = 18.4 Hz, *J_b* = 3.6 Hz, 1H_{*A*}), 3.06 (dd, *J_a* = 18.6 Hz, *J_b* = 7.9 Hz, 1H_{*A*}), 3.13-3.23 (m, 1H_{*B*}), 5.44 (m, 1H_{*A*}), 5.66 (m, 1H_{*B*}), 6.80 (d, *J* = 8.4 Hz, 1H_{*B*}), 6.91 (d, *J* = 7.8 Hz, 1H_{*A*}), 6.96 (d, *J* = 7.8 Hz, 1H_{*B*}), 7.00 (d, *J* = 8.7 Hz, 1H_{*A*}), 7.03-7.09 (m, 1H_{*B*}), 7.10-7.16 (m, 1H_{*B*</sup>), 7.26-7.40 (m, 1H_{*A*} + 2H_{*B*}), 7.47 (s, 1H_{*A*}), 7.49-7.57 (m, 1H_{*A*}), 7.60 (s, 1H_{*B*}), 7.66-7.79 (m, 1H_{*A*}), 7.80-7.86 (m, 1H_{*A*}), 8.36 (d, *J* = 8.5 Hz, 1H_{*A*}), 9.41 (brs, 1H_{*A*}), 10.06 (brs, 1H_{*B*}). ¹³C-NMR (100 MHz, DMSO-*d₆*): mixture of conformers δ 20.5 (CH₃), 20.6 (CH₃), 30.7 (CH), 34.2 (CH), 43.1 (CH₂), 43.7 (CH₂), 117.7 (C), 117.9 (CH), 118.7 (CH), 119.8 (C), 122.0 (CH), 122.1 (CH), 122.2 (CH), 122.4 (CH), 122.5 (CH), 123.1 (CH), 125.0 (CH), 125.3 (CH), 125.8 (CH), 126.7 (CH), 136.4 (CH), 136.5 (C), 136.9 (C), 153.0 (C), 153.6 (C), 156.2 (C), 157.0 (C), 205.1 (C), 206.3 (C).}

(S)-3-(6-bromo-2-hydroxynaphthalen-1-yl)-5-fluoro-2,3-dihydro-1H-inden-1-one 10cb (Table 4, entry



4): The reaction was performed following the general procedure on using 9amino(9-deoxy)*epi*-quinidine *ent*-**A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hexane/AcOEt 70:30) as a white solid in 83% yield and 90% ee. **HPLC analysis**: Daicel Chiralcel OD-H, Hex/i-PrOH 95:5, flow rate 0.7 ml/min, $\lambda = 214$ nm: $\tau_{major} = 28.23$ min, $\tau_{minor} =$ 33.48 min. **ESI-MS:** 393 (M + Na)⁺, 395 (M + Na)⁺. $[\alpha]_D^{30} = -205.4$ (*c* 0.91, DMSO, 90% ee). ¹H-NMR (400 MHz, DMSO-*d₆*): 73.5:26.5 mixture of conformational

diastereosiomers *A* (major) and *B* (minor); δ 2.82 (dd, J_a = 19.3 Hz, J_b = 5.1 Hz, 1H_B), 2.90 (dd, J_a = 18.5 Hz, J_b = 3.5 Hz, 1H_A), 3.11 (dd, J_a = 18.4 Hz, J_b = 8.1 Hz, 1H_A), 3.21 (dd, J_a = 19.2 Hz, J_b = 8.2 Hz, 1H_B), 5.45 (m, 1H_A), 5.68 (m, 1H_B), 6.70 (d, J = 9.3 Hz, 1H_B), 6.74-6.84 (m, 1H_A + 1H_B), 7.07 (d, J = 8.9 Hz, 1H_A), 7.21 (m, 1H_A), 7.26-7.38 (m, 2H_B), 7.62 (dd, J_a = 9.2 Hz, J_b = 2.2 Hz, 1H_A), 7.70-7.78 (m, 2H_A + 1H_B), 7.88 (dd, J_a = 8.6 Hz, J_b = 5.4 Hz, 1H_B), 8.06 (d, J = 2.1 Hz, 1H_B), 8.10 (d, J = 2.2 Hz, 1H_A), 8.30 (d, J = 9.4 Hz, 1H_A), 9.75 (brs, 1H_A + 1H_B). ¹³C-NMR (100 MHz, DMSO- d_6): mixture of conformers δ 34.5 (CH), 35.6 (CH), 43.2 (CH₂), 43.7 (CH₂), 111.7 (d, CH, J_{C-F} = 22.3 Hz), 111.9 (d, CH, J_{C-F} = 22.3 Hz), 114.9 (d, CH, J_{C-F} = 23.4 Hz), 115.3 (C), 115.8 (d, CH, J_{C-F} = 23.4 Hz), 117.3 (C), 119.1 (CH), 119.4 (C), 119.8 (CH), 124.5 (CH), 124.7 (CH), 125.0 (d, CH, J_{C-F} = 10.2 Hz), 126.2 (d, CH, J_{C-F} = 10.8 Hz), 128.0 (CH), 128.7

(d, CH, J_{C-F} = 34.6 Hz), 129.4 (C), 129.5 (CH), 130.2 (CH), 130.3 (d, C, J_{C-F} = 32.5 Hz), 130.9 (C), 132.2 (C), 133.1 (C), 133.4 (d, C, J_{C-F} = 1.5 Hz), 153.6 (C), 154.3 (C), 161.9 (d, C, J_{C-F} = 9.7 Hz), 162.4 (d, C, J_{C-F} = 9.6 Hz), 165.1 (C), 165.6 (C), 167.6 (C), 168.1(C), 203.1(C), 204.2(C).

(S)-5-chloro-3-(2-hydroxy-6-methoxynaphthalen-1-yl)-2,3-dihydro-1H-inden-1-one 10de (Table 4,



entry 5): The reaction was performed following the general procedure on using 9-amino(9-deoxy)*epi*-quinidine *ent*-**A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hexane/Et₂O 70:30) as a white solid in 84% yield and 81% ee. **HPLC analysis**: Daicel Chiralpak AD-H, Hex/i-PrOH 90:10, flow rate 0.75 ml/min, $\lambda = 254$ nm: $\tau_{minor} = 23.19$ min, $\tau_{mojor} = 25.66$ min. **HRMS** calculated for C₂₀H₁₅O₃Cl₁ 338.070974, found 338.07133. **[\alpha]**³⁰ determined on the product obtained in the reaction with 9-

amino(9-deoxy)*epi*-quinine **A** as catalyst = +141.5 (*c* 0.9; DMSO, 78% ee). ¹**H-NMR (400 MHz, DMSO***d*₆): 75:25 mixture of conformational diastereosiomers *A* (major) and *B* (minor); δ 2.83 (dd, *J*_a = 19.3 Hz, *J*_b = 4.9 Hz, 1H_B), 2.93 (dd, *J*_a = 18.6 Hz, *J*_b = 3.5 Hz, 1H_A), 3.10 (dd, *J*_a = 18.6 Hz, *J*_b = 8.1 Hz, 1H_A), 3.21 (dd, *J*_a = 19.4 Hz, *J*_b = 8.4 Hz, 1H_B), 3.75 (s, 3H_B), 3.86 (s, 3H_A), 5.45 (m, 1H_A), 5.68 (m, 1H_B), 6.67 (d, *J* = 9.3 Hz, 1H_B), 6.81 (dd, *J*_a = 9.3 Hz, *J*_b = 2.6 Hz, 1H_B), 6.95-7.04 (m, 2H_A + 1H_B), 7.17-7.32 (m, 2H_A + 2H_B), 7.43 (dd, *J*_a = 8.2 Hz, *J*_b = 1.9 Hz, 1H_A), 7.51 (m, 1H_B), 7.60-7.71 (m, 2H_A + 1H_B), 7.81 (d, *J* = 8.3 Hz, 1H_B), 8.26 (d, *J* = 9.5 Hz, 1H_A), 9.26 (s, 1H_A), 9.88 (s, 1H_B). ¹³C-NMR (100 MHz, DMSO-*d*₆): mixture of conformers δ 34.5 (CH), 35.6 (CH), 43.1 (CH₂), 43.6 (CH₂), 54.9 (CH), 55.1 (CH), 107.0 (CH), 108.0 (CH), 117.2 (C), 118.1 (CH), 118.4 (CH), 119.0 (CH), 119.1 (CH), 119.4 (C), 123.6 (C), 123.7 (CH), 124.0 (CH), 125.0 (CH), 125.1 (C), 125.2 (C), 135.4 (C), 139.2 (C), 140.2 (C), 151.2 (C), 152.0 (C), 154.4 (C), 154.9 (C), 160.8 (C), 161.4 (C), 203.8 (C), 204.9 (C).

(S)-3-(4-chloro-1-hydroxynaphthalen-2-yl)-5-fluoro-2,3-dihydro-1H-inden-1-one 11bb (Table 4, entry



6): The reaction was performed following the general procedure on using 9amino(9-deoxy)*epi*-quinidine *ent*-**A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hexane/Et₂O 60:40) as a white solid in 60% yield and 90% ee. **HPLC analysis**: Daicel Chiralcel OD-H, Hex/i-PrOH 95:5, flow rate 0.75 ml/min, λ = 214 nm: τ_{minor} = 21.63 min, τ_{major} = 24.11 min. **ESI-MS:** 327 (M + 1)⁺, 329 (M + 1)⁺, 349 (M +

Na)⁺, 351 (M + Na)⁺. $[\alpha]_{D}^{30}$ = -119.6 (*c* 0.78; DMSO, 90% ee). ¹H-NMR (400 MHz, DMSO-*d₆*): δ 2.78 (dd, *J_a* = 9.0 Hz, *J_b* = 4.0 Hz, 1H), 3.24 (dd, *J_a* = 19 Hz, *J_b* = 8.2 Hz, 1H), 5.12 (m, 1H), 7.09 (dd, *J_a* = 8.9 Hz, *J_b* = 2.1 Hz, 1H), 7.24-7.34 (m, 2H), 7.55-7.68 (m, 2H), 7.80 (dd, *J_a* = 8.5 Hz, *J_b* = 5.3 Hz, 1H), 8.06 (m, 1H), 8.30 (m, 1H), 9.83 (s, 1H). ¹³C-NMR (100 MHz, DMSO-*d₆*): δ 38.4 (CH), 44.7 (CH₂), 119.9 (d, CH, *J_{C-F}* =

22.1 Hz), 115.9 (d, CH, J_{C-F} = 23.7 Hz), 121.4 (C), 122.8 (CH), 123.6 (CH), 124.7 (C), 125.4 (d, CH, J_{C-F} = 10.7 Hz), 126.1 (CH), 126.6 (CH), 126.7 (C), 127.3 (CH), 129.8 (C), 133.3 (d, C, J_{C-F} = 1.3 Hz), 149.5 (C), 161.0 (d, C, J_{C-F} = 9.7 Hz), 165.2 (C), 167.8 (C), 203.5 (C).

(*S*)-3-(1-hydroxy-4-methoxynaphthalen-2-yl)-6-methyl-2,3-dihydro-1*H*-inden-1-one 11cc (Table 4, entry 7): The reaction was performed following the general procedure on using 9-amino(9-deoxy)*epi*-



quinidine **ent-A** as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hexane/Et₂O 75:25) as a white solid in 92% yield and 94% ee. **HPLC analysis**: Daicel Chiralpak AD-H, Hex/i-PrOH 90:10, flow rate 0.75 ml/min, $\lambda = 214$ nm: $\tau_{major} = 18.33$ min, $\tau_{minor} = 20.09$ min. **HRMS** calculated for C₂₁H₁₈O₃ 318.12559, found 318.12592. ¹H-NMR (400 MHz, DMSO-d₆): mixture of conformers δ 2.78

(dd, $J_a = 18.8$ Hz, $J_b = 3.9$ Hz, 1H), 3.18 (dd, $J_a = 19.0$ Hz, $J_b = 8.0$ Hz, 1H), 3.75 (s, 3H), 5.11 (m, 1H), 6.45 (s, 1H), 7.17 (d, J = 7.8 Hz, 1H), 7.42-7.54 (m, 4H), 8.04 (m, 1H), 8.16 (m, 1H), 8.85 (s, 1H). ¹³C-NMR (100 MHz, DMSO- d_6): δ 20.6 (CH₃), 38.1 (CH), 44.9 (CH₂), 55.5 (CH₃), 104.3 (CH), 121.3 (CH), 122.1 (CH), 122.5 (CH), 124.4 (C), 124.5 (C), 125.0 (CH), 125.7 (CH), 126.3 (CH), 126.7 (C), 136.0 (CH), 136.6 (C), 137.0 (C), 143.0 (C), 148.6 (C), 155.7 (C), 205.6 (C).

(S)-5-chloro-3-(1-hydroxy-4-methoxynaphthalen-2-yl)-2,3-dihydro-1H-inden-1-one 11dc (Table 4



entry 8): The reaction was performed following the general procedure on using 9-amino(9-deoxy)*epi*-quinidine *ent-A* as catalyst. The title compound was isolated by flash chromatography on silica gel (eluent mixture Hexane/Et₂O 70:30) as a white solid in 83% yield and 90% ee. **HPLC analysis**: Daicel Chiralcel OD-H, Hex/i-PrOH 95:5, flow rate 0.75 ml/min, $\lambda = 214$ nm: $\tau_{minor} = 32.18$ min, $\tau_{maior} = 38.27$ min. **ESI-MS:** 361 (M + Na)⁺, 363 (M + Na)⁺.

¹H-NMR (400 MHz, DMSO-*d_b*): δ 2.90 (dd, J_a = 19.0 Hz, J_b = 4.0 Hz, 1H), 3.20 (dd, J_a = 19.0 Hz, J_b = 8.1 Hz, 1H), 3.81 (s, 3H), 5.11 (m, 1H), 6.59 (s, 1H), 7.40-7.58 (m, 3H), 7.72 (d, J = 8.2 Hz, 1H), 8.06 (d, J = 8.2 Hz, 1H), 8.15 (d, J = 8.3 Hz, 1H), 8.88 (s, 1H). ¹³C-NMR (100 MHz, DMSO-*d_b*): δ 44.3 (CH₂), 38.1 (CH), 55.6 (CH₃), 104.8 (CH), 121.4 (CH), 122.1 (CH), 123.8 (C), 124.4 (CH), 124.7 (C), 125.2 (CH), 126.2 (CH), 126.8 (C), 128.0 (CH), 135.2 (C), 139.7 (C), 143.1 (C), 148.7 (C), 160.3 (C), 204.4 (C).

¹H- and ¹³C-NMR-Traces





















Standard Proton Prameters - i300@fci.unibo.it

File: home/bartoli/vnmrsys/data/Giorgio/GB1601-GB1700/GB1648_IIspot.fid











S56















Std Proton parameters

220

File: home/bartoli/Giorgio/GB1601-GB1700/GB1634_1H_DMSO.fid



200 180 160 140 120 100 80 60 40 20 0 ppm





HPLC traces

Data File D:\HPCHEM\2\DATA\GIORGIO\GB4347 3.D Sample Name: gb4347 Racemic Sample OD-H, 95:5 n-Hexane:isopropanol, 0.650 ml/min Injection Date : 25/02/02 23.29.53 : gb4347 : giorgio Sample Name Location : Vial 1 Acq. Operator : D:\HPCHEM\2\METHODS\PAOLO.M Acq. Method : 25/02/02 23.07.05 by giorgio Last changed (modified after loading) Analysis Method : D:\HPCHEM\2\METHODS\PAOLO.M Last changed : 03/06/02 5.55.04 by giorgio DAD1 A, Sig=254,100 Ref=360,100 (GIORGIO\GB4347_3.D) mAU 25-525 60 30.988 50 OH 40 rac-3aa 30 20 10 0 5 10 15 20 25 30 35 40 min Ó Area Percent Report Sorted By : Signal Multiplier : 1.0000 Dilution : 1.0000 Signal 1: DAD1 A, Sig=254,100 Ref=360,100 Peak RetTime Type Width Area Height Area # [min] [min] [mAU*s] [mAU] olo 1 25.525 PP 0.7811 3094.95581 62.13245 51.4726 2 30.988 BB 0.9192 2917.87183 50.12292 48.5274 6012.82764 112.25537 Totals : Results obtained with enhanced integrator!

*** End of Report ***











```
Electronic Supplementary Material (ESI) for Chemical Communications
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```

```
Data File:
              S:\4\DATA\EP\GB_1591_R000000.D
Sample name: GB_1591_r
Sample Info: Lux-Cellulose 2, 0.7 ml/min, Hex/i-PrOH 9:1, 230 nm, 20
               °C.
                                                                               ->
 Acquired by: EP
                                            on:
                                                       04/05/2012 18.18.03
      Method: S:\4\METHODS\RGM001-C2-A90-B10-0.75-20C-S20.M
    Location: Vial 1
                                   Volume:
                                                      5uL
      Column: Lux Cellulose-2
                                                   250 mm x 4.6 mm
        Flow: 0.700 mL/min
                                    Temp.:
                                                 20°C
```



0.000


Electronic Supplementary Material (ESI) for Chemical Communications This journal is $\ensuremath{\mathbb{C}}$ The Royal Society of Chemistry 2012

```
Data File:
             S:\4\DATA\EP\EP_285_286_R038.D
Sample name: EP_285_286_R
Sample Info: AD-H, 0.7 ml/min, 9:1 Hex/i-PrOH, 230 nm, 25°
              x rac rac prima prova
                                                                           ->
                                                    17/07/2012 18.10.21
Acquired by: EP
                                         on:
      Method: S:\4\METHODS\RGM001-C2-A90-B10-0.75-20C-S20.M
   Location: Vial 41
                                 Volume:
                                                 10 \mathrm{uL}
      Column: Lux 5u Amylose-2
                                                250 mm x 4.6 mm
                                              25°C
        Flow: 0.700 mL/min
                                  Temp.:
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```
Data File:
              S:\4\DATA\EP\EP_286_000001.D
Sample name: EP_286_
Sample Info: AD-H 0.7 ml-min, Hex/i-PrOH 9:1 230 nm, 25°C.
              3 (Boc-amino) -2-naphthol + cicloesenone cat QD-NH2
                                                                            ->
 Acquired by: EP
                                                 28/06/2012 14.13.30
                                           on:
      Method: S:\4\METHODS\RGM001-C2-A90-B10-0.75-20C-S20.M
    Location: Vial 41
                                  Volume:
                                                    5uL
                                                 250 mm x 4.6 mm
      Column: Lux 5u Amylose-2
        Flow: 0.700 mL/min
                                   Temp.: 25.01°C
```





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```
Data File: T:\2\DATA\EP\EP_273_000009.D
Sample name: EP_273_
Sample Info: Lux cellulose 2, 0.4, 85:15, 230, not controlled (33°C)
.
Acquired by: EP on: 16/07/2012 15.26.07
Method: S:\4\METHODS\RGM001-C2-A90-B10-0.75-20C-S20.M
Location: Vial 1 Volume: Injecti->
Column: Lux Cellulose-2 250 mm x 4.6 mm
Flow: 0.400 mL/min Temp.: 34.68°C
```



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S80

0.000













Data File: S:\4\DATA\EP\GB1487000000.D Sample name: GB1487 Lux-Cellulose 2, 0.7 ml/min, Hex/i-PrOH 87:13, 230 nm, Sample Info: 15°C. -> Acquired by: EP 07/05/2012 15.11.13 on: Method: S:\4\METHODS\RGM001-C2-A90-B10-0.75-20C-S20.M Location: Vial 1 Volume: 5uL Column: Lux Cellulose-2 250 mm x 4.6 mm Flow: 0.700 mL/min Temp.: 14.99°C DAD1 D, Sig=230,16 Ref=360,100 (S:\4\DATA\EP\GB1487000000.D) mAU – 527 9 140 120 HO 17.825 100 80 rac-5ac 60 40 20 0 10 18 20 min 12 14 16 Area Percent Report Signal: DAD1 D, Sig=230,16 Ref=360,100 # Time Type Width Height Area Area % Amount Compound Nam - - - ------0.317 154.231 3207.071 0.537 92.849 3220.300 49.897 10.527 BB 1 0.000 3220.300 17.825 BB 0.537 92.849 50.103 2 0.000 _ _ _ _ _ --------------0.000

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```
Data File:
             S:\4\DATA\RC\EP_283_284_R040.D
Sample name: EP_283_284_R
Sample Info: AD-H, 1 ml/min, 8:2 Hex/i-PrOH, 230 nm, 25°
               x rac rac seconda prova
                                                                              ->
 Acquired by: EP
                                                      18/07/2012 12.02.23
                                            on:
      Method: S:\4\METHODS\RGM001-C2-A90-B10-0.75-20C-S20.M
    Location: Vial 41
                                   Volume:
                                                     5uL
      Column: Lux 5u Amylose-2
                                                  250 mm x 4.6 mm
        Flow: 1.000 mL/min
                                    Temp.: 24.99°C
```



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```
Data File:
             S:\4\DATA\EP\EP_283_000005.D
Sample name: EP_283_
Sample Info: AD-H 1ml-min, Hex/i-PrOH 80:20, 230 nm, 25°C,
             4-(boc-amino)-1-naphthol + ciclohexenone cat Q-NH2
                                                                        ->
Acquired by: EP
                                              29/06/2012 12.55.50
                                        on:
     Method: S:\4\METHODS\RGM001-C2-A90-B10-0.75-20C-S20.M
   Location: Vial 41
                                Volume:
                                                 5uL
     Column: Lux 5u Amylose-2
                                              250 mm x 4.6 mm
       Flow: 1.000 mL/min
                                 Temp.: 24.99°C
```







0.000

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```
Data File:
             S:\4\DATA\EP\EP_268_269_0005.D
Sample name: EP_268_269_
Sample Info: OJ-H, 0.7 ml/min, 9:1 Hex/i-PrOH, 230 nm, 23°C.
              1-naphthol + 4-methyl-cyclohexenone rac.
                                                                        ->
Acquired by: EP
                                                 12/06/2012 15.35.14
                                        on:
     Method: S:\4\METHODS\RGM001-C2-A90-B10-0.75-20C-S20.M
   Location: Vial 26
                                                 5uL
                                Volume:
     Column: Lux 5u Amylose-2
                                              250 mm x 4.6 mm
       Flow: 0.700 mL/min
                                            23°C
                                 Temp.:
```



```
Data File:
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Sample name: EP_268_
Sample Info: OJ-H, 0.7 ml/min, 9:1 Hex/i-PrOH, 230 nm, 23°C.
              1-naphthol + 4-methyl-cyclohexenone Q-NH2.
                                                                           ->
 Acquired by: EP
                                                   12/06/2012 16.30.41
                                          on:
      Method: S:\4\METHODS\RGM001-C2-A90-B10-0.75-20C-S20.M
    Location: Vial 27
                                 Volume:
                                                   5uL
                                                250 mm x 4.6 mm
      Column: Lux 5u Amylose-2
        Flow: 0.700 mL/min
                                   Temp.: 22.99°C
```



0.000













```
Data File: S:\4\DATA\EP\EP_247_248_0044.D
Sample name: EP_247_248_
Sample Info: AD-H, 0.7 ml/min, 96:4 Hex/i-PrOH, 230 nm, 25°
              rac per vedere se i picchi si separano.
                                                                            ->
 Acquired by: EP
                                          on:
                                                 18/07/2012 15.51.08
      Method: S:\4\METHODS\RGM001-C2-A90-B10-0.75-20C-S20.M
    Location: Vial 41
                                  Volume:
                                                  5uL
      Column: Lux 5u Amylose-2
                                                 250 mm x 4.6 mm
                                               25°C
        Flow: 0.700 mL/min
                                   Temp.:
```






















*** End of Report ***





Data File D:\HPCHEM\2\DATA\GIORGIO\6236244.D



 1
 18.943 VV
 0.4136 1840.32117
 67.60520 48.8476

 2
 20.582 VB
 0.4407 1927.15125
 66.77039 51.1524

 Totals :
 3767.47241
 134.37559

Results obtained with enhanced integrator!

*** End of Report ***

Sample Name: GB1624+GB1623





Sample Name: GB1631+GB1630

5Cl-indenone, 4Meo-Alfanaftolo, rac OD-H 95:5 Hexane:IPA 0.750 ml/min

Injection Date	:	27/05/02 0.45.28					
Sample Name	:	GB1631+GB1630	Location	:	Vial	1	
Acq. Operator	:	giorgio					
Acq. Method	:	D:\HPCHEM\2\METHODS\PAOLO.M					
Last changed	:	26/05/02 21.15.03 by giorgio					
		(modified after loading)					
Analysis Method	:	D:\HPCHEM\2\METHODS\PAOLO.M					
Last changed	:	03/07/02 3.44.28 by Riccardo					
		(modified after loading)					
DAD1 C, Sig	=21	4,8 Ref=360,100 (GIORGIO\6306317.Ď)					
mALL ±						10	



_____ Area Percent Report -----

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 sense same more when were same and same same same same were wort from their base whet inter which this half to	and some shark more many many many state state many sales when state and some state along state and state some

Sorted By	:	Signal
Multiplier	:	1.0000
Dilution	:	1.0000

Signal 1: DAD1 C, Sig=214,8 Ref=360,100

Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	%
1	32.115	BB	1.2359	3444.74634	42.94129	48.4821
2	38.047	VB	1.3987	3660.44580	39.58815	51.5179
Total	_S :			7105.19214	82.52944	
Resu	ults obta	ained	with enh	nanced integ	grator!	

*** End of Report ***



*** End of Report ***