

# Facile carbon-sulfur bond cleavage in diarylsulfonium ylides: a catalytic sulfur-to-silicon group transfer

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## Part I Experimental part

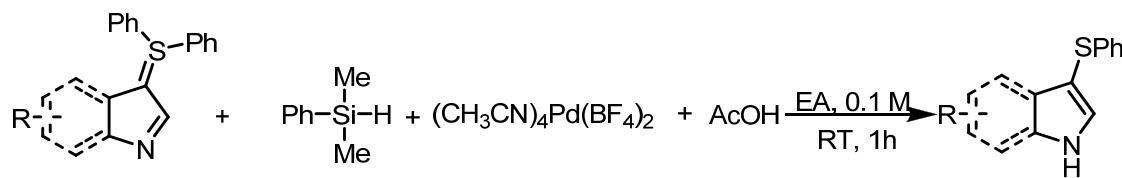
### General information

Unless otherwise indicated, all glassware was oven dried by a heat gun before use and all reactions were performed under an atmosphere of Argon. All solvents were distilled from appropriate drying agents prior to use. All reagents were used as received from commercial suppliers unless otherwise stated. All the ylides were prepared according to the procedures reported by our group.<sup>1</sup> Reaction progress was monitored by thin layer chromatography (TLC) performed on plastic plates coated with kieselgel F<sub>254</sub> with 0.2 mm thickness or GC-MS. Visualization was achieved by ultraviolet light (254 nm). Flash column chromatography was performed using silica gel 60 (230-400 mesh, Merck and co.). Mass spectra were obtained using a Finnigan MAT 8200 or (70 eV) or an Agilent 5973 (70 eV) spectrometer, using electrospray ionization (ESI). All <sup>1</sup>H NMR, <sup>13</sup>C NMR spectra were recorded on Bruker AV-500, AV-400 or AV-300 in CDCl<sub>3</sub>. Chemical shifts were given in parts per million (ppm,  $\delta$ ), referenced to the peak of tetramethylsilane, defined at  $\delta$  = 0.00 (<sup>1</sup>H NMR), or the solvent peak of CDCl<sub>3</sub>, defined at  $\delta$  = 77.0 (<sup>13</sup>C NMR). Coupling constants were quoted in Hz ( $J$ ). <sup>1</sup>H NMR Spectroscopy splitting patterns were designated as singlet (s), doublet (d), triplet (t), quartet (q), pentet (p), sextet (se), septet (sep), octet (o). Splitting patterns that could not be interpreted or easily visualized were designated as multiplet (m) or broad (br).

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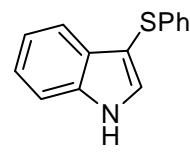
<sup>1</sup> X. Huang, R. Goddard, N. Maulide, *Angew. Chem. Int. Ed.* **2010**, *49*, 8979.

## 1.1 Palladium-catalysed dearylation of diphenyl sulfonium ylides



**Typical procedure** for the dephenylation of sulfonium ylides: Under an argon atmosphere,  $(\text{CH}_3\text{CN})_4\text{Pd}(\text{BF}_4)_2$  (0.005 mmol, 2.2 mg) and sulfonium ylide (0.1 mmol, 37.3 mg) were added into a dried Schlenk tube. The tube was stopped, evacuated and back filled with argon three times. Ethyl acetate (1 mL, 0.1 M), AcOH (57 $\mu$ L, 1.0 mmol) were added by syringe, followed by dimethylphenylsilane (30  $\mu$ L, 0.2 mmol). The mixture was stirred under argon atmosphere at room temperature for 1h. The resultant mixture was filtered through a short plug of silica gel, and eluted with ethyl acetate. The filtrate was concentrated on a rotatory evaporator to afford a crude mixture, which was purified by flash column chromatography on silica gel to afford the title compound.

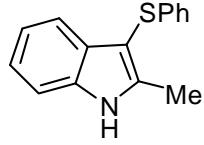
### 3-(Phenylthio)-1H-indole (3b)<sup>2</sup>



Compound **3b** was obtained in 67% yield as white solid.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  8.39 (br, 1H), 7.61 (d,  $J = 7.9$  Hz, 1H), 7.48-7.42 (m, 2H), 7.28-7.24 (m, 1H), 7.18-7.14 (m, 3H), 7.11-7.09 (m, 2H) 7.06-7.03 (m, 1H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  139.2, 136.4, 130.7, 129.1, 128.7, 125.8, 124.7, 123.0, 123.9, 119.6, 111.5, 102.8.

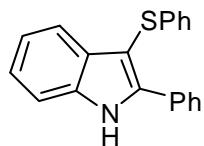
<sup>2</sup> Fang, X.-L.; Tang, R.-Y.; Zhong, P.; Li, J.-H. *Synthesis* **2009**, 4183.

**2-Methyl-3-(phenylthio)-1H-indole (3c)<sup>1</sup>**



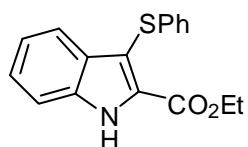
Compound **3c** was obtained in 92% yield as pale yellow solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.15 (br, 1H), 7.47 (d, *J* = 7.8 Hz, 1H), 7.28-7.26 (m, 1H), 7.14-7.03 (m, 4H), 6.98-6.94 (m, 3H), 2.45 (s, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 141.1, 139.3, 135.4, 130.3, 128.4, 124.5, 122.2, 119.0, 110.6, 99.4, 12.1.

**2-Phenyl-3-(phenylthio)-1H-indole (3d)<sup>3</sup>**



Compound **3d** was obtained in 89% yield as colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.47 (br, 1H), 7.67-7.66 (m, 2H), 7.55 (d, *J* = 7.6 Hz, 1H), 7.37-7.33 (m, 3H), 7.31-7.28 (m, 1H), 7.19 (t, *J* = 7.3 Hz, 1H), 7.10-7.06 (m, 3H), 7.02-7.01 (m, 2H), 6.98-6.95 (m, 1H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 142.0, 139.2, 135.8, 131.4, 131.1, 128.8, 128.74, 128.67, 128.1, 125.5, 124.6, 123.3, 121.1, 119.9, 111.1.

**Ethyl 3-(phenylthio)-1H-indole-2-carboxylate (3a)<sup>4</sup>**

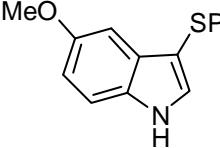


Compound **3a** was obtained in 81% yield as white solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 9.26 (br, 1H), 7.54 (dd, *J* = 8.1 Hz, *J* = 0.9 Hz, 1H), 7.39-7.36 (m, 1H), 7.31-7.25 (m, 1H), 7.10-6.98 (m, 6H), 4.32 (q, *J* = 7.1 Hz, 2H), 1.23 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 161.3, 137.9, 135.7, 130.1, 128.8, 128.7, 127.4, 126.1, 125.3, 121.7, 121.5, 112.1, 110.5, 61.5, 14.2.

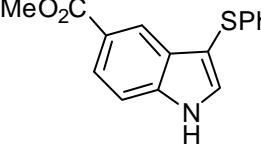
<sup>3</sup> Gou, Y.-J.; Tang, R.-Y.; Li, J.-H.; Zhong, P.; Zhang, X.-G. *Adv. Synth. Catal.* **2009**, *351*, 2615.

<sup>4</sup> Tudge, M.; Tamiya, M.; Savarin, C.; Humphrey, G. R. *Org. Lett.* **2006**, *8*, 565.

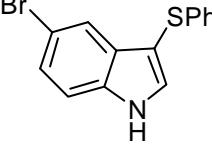
**5-Methoxy-3-(phenylthio)-1H-indole (3e)<sup>3</sup>**

Compound **3e** obtained in 86% yield as colorless oil. <sup>1</sup>H NMR  
 (500 MHz, CDCl<sub>3</sub>) δ 8.34 (br, 1H), 7.42 (d, *J* = 2.7 Hz, 1H), 7.31 (d, *J* = 8.8 Hz, 1H), 7.18-7.15 (m, 2H), 7.10-7.04 (m, 4H), 6.91 (dd, *J* = 8.8 Hz, *J* = 2.4 Hz, 1H), 3.78 (s, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 155.1, 139.3, 131.3, 129.9, 128.7, 125.6, 124.7, 113.6, 112.4, 102.1, 100.7, 55.7.

**Methyl 3-(phenylthio)-1H-indole-5-carboxylate (3f)<sup>3</sup>**

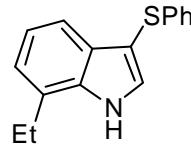
Compound **3f** was obtained in 88% yield as colorless oil. <sup>1</sup>H NMR  
 (500 MHz, CDCl<sub>3</sub>) δ 8.81 (br, 1H), 8.38 (s, 1H), 7.97 (dd, *J* = 8.6 Hz, *J* = 1.5 Hz, 1H), 7.54 (d, *J* = 2.5 Hz, 1H), 7.45 (d, *J* = 8.6 Hz, 1H), 7.16 (t, *J* = 7.6 Hz, 2H), 7.09-7.05 (m, 3H), 3.89 (s, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 167.9, 139.1, 138.8, 132.2, 128.9, 128.8, 125.9, 125.0, 124.4, 123.1, 122.4, 111.5, 104.6, 52.0.

**5-Bromo-3-(phenylthio)-1H-indole (3g)<sup>1,5</sup>**

Compound **3g** was obtained in 66% yield as colorless oil. <sup>1</sup>H NMR  
 (500 MHz, CDCl<sub>3</sub>) δ 8.38 (br, 1H), 7.68 (d, *J* = 1.6 Hz, 1H), 7.40 (d, *J* = 2.5 Hz, 1H), 7.28-7.22 (m, 2H), 7.12-7.09 (m, 2H), 7.01-6.99 (m, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 138.7, 135.1, 131.8, 130.9, 128.8, 126.1, 125.8, 125.0, 122.2, 114.5, 113.0, 102.7.

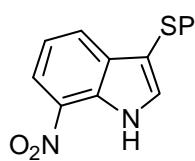
<sup>5</sup> Yadav, J. S.; Reddy, B. V. S.; Reddy, Y. J.; Praneeth, K. *Synthesis* **2009**, 1520.

**7-Ethyl-3-(phenylthio)-1H-indole (3h)<sup>1,4</sup>**



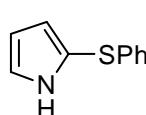
Compound **3h** was obtained in 95% yield as colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.27 (br, 1H), 7.40-7.37 (m, 2H), 7.09-7.02 (m, 6H), 6.98-6.95 (m, 1H), 2.81 (q, *J* = 7.6 Hz, 2H), 1.31 (t, *J* = 7.6 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 139.3, 135.3, 130.3, 128.8, 128.6, 126.9, 125.8, 124.7, 121.5, 121.1, 117.3, 103.1, 23.8, 13.8.

**7-Nitro-3-(phenylthio)-1H-indole (3i)**



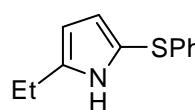
Compound **3i** was obtained in 74% yield as yellow solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 10.1 (br, 1H), 8.15-8.13 (m, 1H), 7.87 (d, *J* = 7.8 Hz, 1H), 7.62 (d, *J* = 2.5 Hz, 1H), 7.19-7.17 (m, 1H), 7.13-7.10 (m, 2H), 7.03-7.01 (m, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 137.9, 133.4, 132.7, 132.5, 129.9, 128.9, 127.8, 126.3, 125.4, 120.3, 105.6; EIMS *m/z* (%): 270 (100), 253 (19), 223 (41); HRMS-(EI) (*m/z*): [M]<sup>+</sup> calcd for C<sub>14</sub>H<sub>10</sub>N<sub>2</sub>O<sub>2</sub>S, 270.0463; found 270.0461.

**2-(Phenylthio)-1H-pyrrole (3j)<sup>6</sup>**



Compound **3j** was obtained in 88% yield as colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.22 (br, 1H), 7.15-7.12 (m, 2H), 7.04-7.00 (m, 1H), 6.95-6.94 (m, 2H), 6.87 (se, *J* = 1.4 Hz, 1H), 6.50 (Hep, *J* = 1.2 Hz, 1H), 6.25 (q, *J* = 3.0 Hz, 1H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 139.1, 128.9, 125.6, 125.3, 121.8, 118.7, 115.5, 110.4.

**2-Ethyl-5-(phenylthio)-1H-pyrrole (3k)**

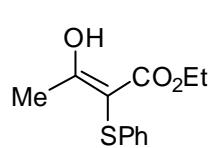


Compound **3k** was obtained in 88% yield as colorless oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.91 (br, 1H), 7.16-7.13 (m, 2H), 7.04-7.00 (m,

<sup>6</sup> (a) Gillis, H. M.; Greene, L.; Thompson, A. *Synlett* **2009**, 112. (b) Thamyongkit, P.; Bhise, A. D.; Taniguchi, M.; Lindsey, J. S.; *J. Org. Chem.* **2006**, 71, 903.

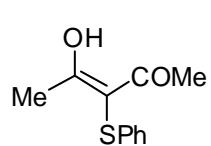
1H), 6.97-6.96 (m, 2H), 6.41 (t,  $J$  = 3.0 Hz, 1H), 5.95-5.94 (m, 1H), 2.56 (q,  $J$  = 7.6 Hz, 2H), 1.18 (t,  $J$  = 7.6 Hz, 3H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  139.7, 138.6, 128.9, 125.5, 125.2, 119.3, 112.9, 106.7, 21.1, 13.3; EIMS  $m/z$  (%): 203 (78), 188 (100); HRMS-(EI) ( $m/z$ ): [M]<sup>+</sup> calcd for  $\text{C}_{12}\text{H}_{13}\text{NS}$ , 203.0769; found 203.0770.

**(E)-Ethyl 3-hydroxy-2-(phenylthio)but-2-enoate (6a)<sup>7</sup>**



$\text{Pd}(\text{OAc})_2$  (10 mol%) was employed as the catalyst. The mixture was stirred in 1,2-dichloroethane at room temperature for 2h. Compound **6a** was obtained in 76% yield as colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  13.78 (d,  $J$  = 0.5 Hz, 1H), 7.19-7.16 (m, 2H), 7.07-7.03 (m, 3H), 4.14 (q,  $J$  = 7.1 Hz, 2H), 2.26 (d,  $J$  = 0.5 Hz, 3H), 1.11 (t,  $J$  = 7.1 Hz, 3H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  184.3, 172.6, 137.8, 128.5, 125.3, 124.8, 61.3, 20.6, 13.7.

**(E)-4-hydroxy-3-(phenylthio)pent-3-en-2-one (6b)<sup>8</sup>**

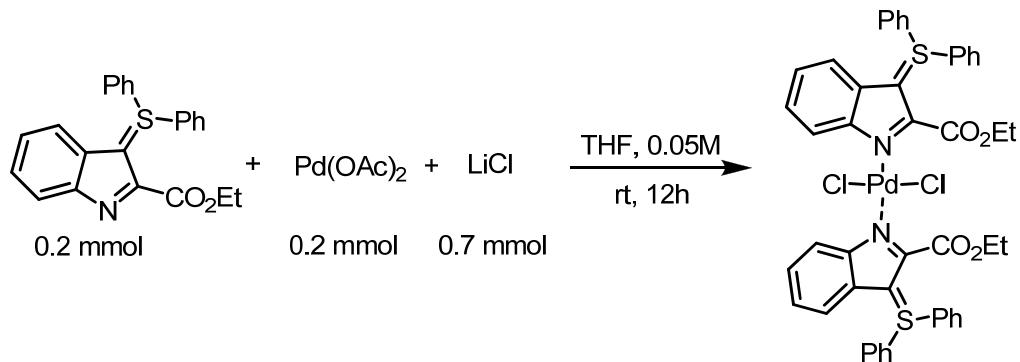


$\text{Pd}(\text{OAc})_2$  (10 mol%) was employed as the catalyst. The mixture was stirred in 1,2-dichloroethane at room temperature for 2h. The reaction was carried out in 0.2 mmol scale. Compound **6b** was obtained in 83% yield as colorless oil.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.33-7.30 (m, 2H), 7.18-7.12 (m, 3H), 2.37 (s, 6H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  198.3, 137.7, 129.2, 125.2, 124.6, 101.5, 24.4.

<sup>7</sup> Brownbridge, P.; Hunt, P. G.; Warren, S. *J. Chem. Soc. Perkin Trans. I* **1986**, 1695.

<sup>8</sup> Rashid, M. A.; Rasool, N.; Adeel, M.; Reinke, H.; Fischer, C.; Langer, P. *Tetrahedron* **2008**, 64, 3782.

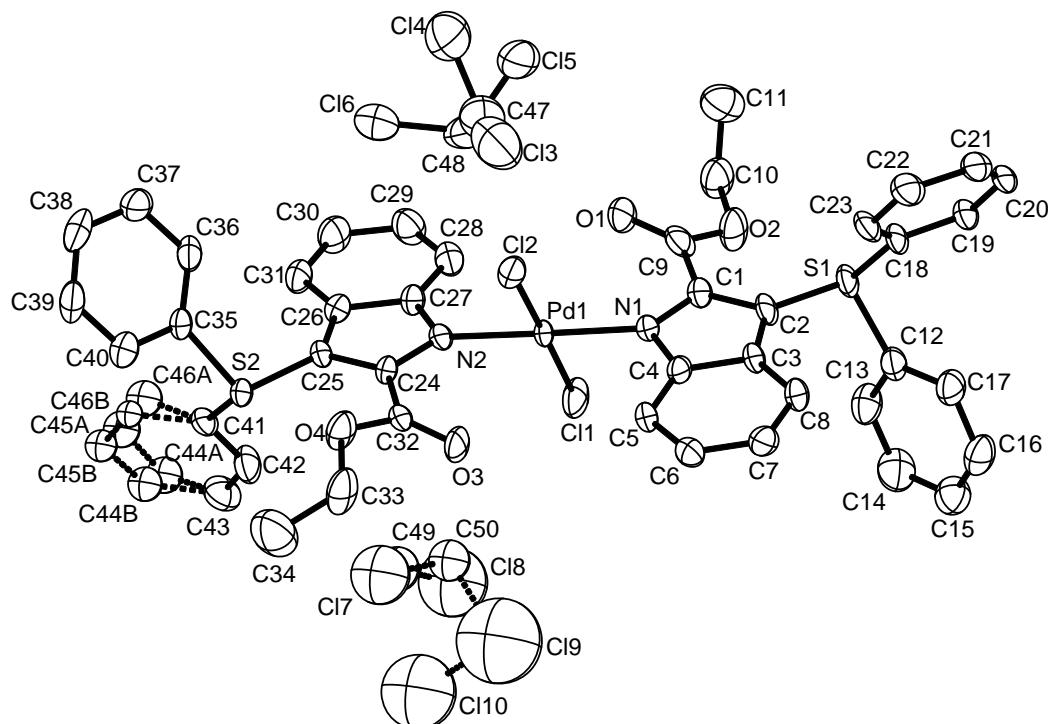
## 1.2 Preparation of palladium complex 7



To a solution of ylide (74.6 mg, 0.2 mmol) and Pd(OAc)<sub>2</sub> (44.8 mg, 0.2 mmol) in dry THF (2 mL, 0.1M), was added LiCl (30 mg, 0.7 mmol). The mixture was stirred at room temperature for 36h and the solvent was removed under vacuum. The resulting solid was washed carefully with THF, then diethyl ether. The crude yellow solid (89 mg) was redissolved by large amount of dichloromethane, filtered, and a portion of the filtrate was used for the preparation of the crystal. NMR of the crude solid: <sup>1</sup>H NMR (500 MHz, CD<sub>2</sub>Cl<sub>2</sub>) δ 9.30 (d, *J* = 8.4 Hz, 2H), 7.72-7.69 (m, 4H), 7.65-7.59 (m, 16H), 7.39 (t, *J* = 7.6 Hz, 2H), 6.98 (t, *J* = 7.6 Hz, 2H), 6.57 (d, *J* = 8.2 Hz, 2H), 4.79 (q, *J* = 7.0 Hz, 4H), 1.63 (t, *J* = 7.10 Hz, 6H); <sup>13</sup>C NMR (125 MHz, CD<sub>2</sub>Cl<sub>2</sub>) δ 163.4, 147.6, 144.9, 133.2, 131.3, 131.1, 129.8, 129.7, 128.8, 128.0, 123.3, 122.8, 122.6, 117.4, 84.6, 62.6, 15.0. EIMS *m/z* (%): 889 (100), 701 (15).

### 1.3 Crystal structure of the palladium complex 7

Crystals were prepared by slow diffusion of pentane into a solution of **7** in dichloromethane. Several crystals were investigated, all of which contained solute dichloromethane. The results described here stem from the crystal of best quality.



**Figure S-1.** The molecular structure of **7**. Atomic displacement ellipsoids are shown at the 50% probability level. H atoms omitted for clarity.

**Crystal Data for **7**:**  $[C_{46}H_{38}Cl_2N_2O_4PdS_2] \cdot 3(C_2H_2Cl_2)$ ,  $M_r = 1178.98 \text{ g} \cdot \text{mol}^{-1}$ , yellow-orange plate, crystal size  $0.06 \times 0.14 \times 0.23 \text{ mm}^3$ , monoclinic, space group  $C2/c$ ,  $a = 28.301(2) \text{ \AA}$ ,  $b = 20.495(2) \text{ \AA}$ ,  $c = 18.092(1) \text{ \AA}$ ,  $\beta = 105.211(7)^\circ$ ,  $V = 10126.3(15) \text{ \AA}^3$ ,  $T = 100 \text{ K}$ ,  $Z = 8$ ,  $D_{calc} = 1.547 \text{ g} \cdot \text{cm}^{-3}$ , Mo- $K_\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$ ,  $\mu = 0.917 \text{ mm}^{-1}$ ,

Gaussian absorption correction ( $T_{\min} = 0.88566$ ,  $T_{\max} = 0.95945$ ), minimum and maximum estimated transmissions from the multi-scan scaling: 0.5513 and 1.0 (SADABS), Bruker AXS Enraf-Nonius KappaCCD diffractometer,  $2.81 < \theta < 27.10^\circ$ , 100126 measured reflections, 11175 independent reflections ( $R_{\text{int}} = 0.104$ ), 7729 reflections with  $I > 2\sigma(I)$ . Structure solved by direct methods and refined by full-matrix least-squares against  $F^2$  to  $R_I = 0.0799$  [ $I > 2\sigma(I)$ ],  $wR_2 = 0.2300$ , 590 parameters (G. M. Sheldrick, *Acta Cryst.* **2008**, A64, 112-122).  $S = 1.530$ , indicating that on average the standard uncertainties of the intensities are systematically underestimated, residual electron density  $+1.78 / -1.52 \text{ e } \text{\AA}^{-3}$ . CCDC 894535.

Data were collected to a resolution of  $0.78 \text{ \AA}$  with an average redundancy of almost 9 (see statistics, below). A minimum theta value of  $2.81^\circ$  was chosen in order to avoid possible loss of intensity resulting from the shadow cast by the beamstop. 13 reflections diffract at angles less than this minimum theta value, owing to the large unit cell parameters, and were not included in the data set. Although the face-indexed absorption correction was undertaken with as much care as possible, the observed residual electron density of  $1.78 \text{ e\AA}^{-3}$ ,  $0.92 \text{ \AA}$  from Pd1, may be due to the difficulty of accurately correcting for the effects of X-ray absorption for a thin plate composed of a substance with a relatively high linear absorption coefficient ( $\mu = 0.917 \text{ mm}^{-1}$ ).

#### INTENSITY STATISTICS FOR DATASET

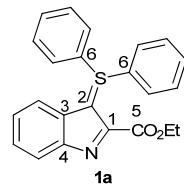
Resolution	#Data	#Theory	%Complete	Redundancy	Mean I	Mean I/s	R(int)	Rsigma
Inf - 2.14	576	593	97.1	11.08	22.8	30.75	0.0664	0.0255
2.14 - 1.70	585	585	100.0	11.77	12.1	28.33	0.0628	0.0256
1.70 - 1.48	576	576	100.0	11.69	7.9	24.70	0.0686	0.0278
1.48 - 1.34	615	615	100.0	11.34	6.1	22.32	0.0729	0.0308
1.34 - 1.24	613	613	100.0	10.75	4.8	18.91	0.0843	0.0362
1.24 - 1.16	659	659	100.0	10.24	3.9	15.86	0.0983	0.0433

1.16 - 1.10	612	612	100.0	9.78	3.6	14.23	0.1072	0.0481
1.10 - 1.05	650	650	100.0	9.52	2.8	12.18	0.1254	0.0580
1.05 - 1.01	606	606	100.0	9.10	2.4	10.67	0.1428	0.0681
1.01 - 0.97	709	709	100.0	8.90	2.2	9.21	0.1598	0.0782
0.97 - 0.94	604	604	100.0	8.38	2.0	8.39	0.1749	0.0893
0.94 - 0.91	705	705	100.0	8.26	1.7	7.01	0.1940	0.1066
0.91 - 0.88	781	781	100.0	7.99	1.4	5.61	0.2343	0.1417
0.88 - 0.86	599	599	100.0	7.60	1.2	4.84	0.2737	0.1691
0.86 - 0.84	632	632	100.0	6.71	1.0	3.54	0.3388	0.2363
0.84 - 0.82	759	759	100.0	6.05	1.0	3.35	0.3233	0.2489
0.82 - 0.80	767	767	100.0	5.52	0.9	2.84	0.3444	0.2992
0.80 - 0.78	444	444	100.0	5.20	0.7	2.09	0.4334	0.4348
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0.88 - 0.78	3454	3454	100.0	6.35	1.0	3.53	0.3159	0.2446
Inf - 0.78	11492	11509	99.9	8.82	4.1	12.10	0.0981	0.0544

Merged [A], lowest resolution = 7.25 Angstroms, 3194 outliers downweighted

The crystal contains three solute dichloromethane molecules per Pd complex. One of the dichloromethane molecules is badly disordered about a two-fold axis of symmetry. Allocation of atoms was based on chemical intuition. The deepest hole in the residual electron density map is -1.51 eÅ<sup>-3</sup> at x = 0.0789, y = 0.3393 and z = 0.3834, 0.62 Å from Cl7 in the disordered dichloromethane molecule. One of the phenyl groups on one of the diphenylsulphonium ylide groups in close vicinity to the disordered dichloromethane is also slightly disordered. Disordered atoms were refined with isotropic atomic displacement parameters. H atom positions were calculated and refined using a riding model. Intensities at high 2θ angles are disproportionately weak, presumably owing to loss of solute from the crystal during transfer from solution to the diffractometer at 100 K. The suggested second term for weighting scheme proposed by SHELXL was not adopted, since it would have given too little weight to these weak reflections, which also contain important structural information.

**Table S-1.** Comparison of selected distances ( $\text{\AA}$ ) and angles ( $^\circ$ ) for the free ligand **1a** and the Pd complex **7**.



bond	isolated ligand (CSD refcode LAFTAP) <sup>1</sup>	Pd complex <b>7</b>
N=C1	1.336(1)	1.342(6) <sup>a</sup>
C1-C2	1.424(1)	1.414(4) <sup>a</sup>
C2=S	1.710(1)	1.721(11) <sup>a</sup>
S-C6 (Ph)	1.790(4) <sup>a</sup>	1.799(5) <sup>a</sup>
C1-C5 (O <sub>2</sub> Et)	1.482(2)	1.481(23) <sup>a</sup>
C2-C3	1.434(1)	1.419(24) <sup>a</sup>
C3-C4	1.426(1)	1.415(4) <sup>a</sup>
N-C4	1.378(1)	1.395(8) <sup>a</sup>
C4-N-C1	105.43(8)	107.8(2)
N-C1-C2	112.62(9)	110.0(2)
C1-C2-C3	105.68(8)	107.3(2) <sup>a</sup>

<sup>a</sup> Average value. <sup>1</sup> Reference 1.

**Table S-2. Atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>).**

$U_{eq}$  is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

	x	y	z	$U_{eq}$
C(1)	0.1068(2)	0.8882(2)	0.2069(3)	0.032(1)
C(2)	0.1079(2)	0.9564(2)	0.1968(3)	0.031(1)
C(3)	0.1295(2)	0.9679(2)	0.1367(3)	0.030(1)
C(4)	0.1402(2)	0.9056(2)	0.1119(3)	0.029(1)
C(5)	0.1606(2)	0.8969(3)	0.0507(3)	0.034(1)
C(6)	0.1713(2)	0.9524(3)	0.0151(3)	0.038(1)
C(7)	0.1627(2)	1.0151(3)	0.0388(3)	0.037(1)
C(8)	0.1412(2)	1.0238(3)	0.0991(3)	0.035(1)
C(9)	0.0904(2)	0.8480(3)	0.2647(3)	0.038(1)
C(10)	0.0410(3)	0.8415(3)	0.3503(4)	0.053(2)
C(11)	0.0765(3)	0.8467(4)	0.4264(4)	0.061(2)
C(12)	0.0392(2)	1.0544(3)	0.1671(3)	0.035(1)
C(13)	-0.0004(2)	1.0179(3)	0.1273(4)	0.046(2)
C(14)	-0.0332(3)	1.0451(4)	0.0647(4)	0.059(2)
C(15)	-0.0262(2)	1.1086(3)	0.0421(4)	0.051(2)
C(16)	0.0117(2)	1.1440(3)	0.0827(4)	0.049(2)
C(17)	0.0460(2)	1.1181(3)	0.1468(4)	0.041(1)
C(18)	0.1249(2)	1.0716(2)	0.2883(3)	0.034(1)
C(19)	0.1077(3)	1.1218(3)	0.3277(3)	0.040(1)
C(20)	0.1413(3)	1.1675(3)	0.3650(3)	0.045(2)
C(21)	0.1906(3)	1.1637(3)	0.3648(3)	0.045(2)
C(22)	0.2069(3)	1.1128(3)	0.3273(4)	0.045(2)
C(23)	0.1732(2)	1.0661(3)	0.2881(3)	0.041(2)
C(24)	0.1484(2)	0.6318(2)	0.0722(3)	0.029(1)
C(25)	0.1444(2)	0.5637(2)	0.0823(3)	0.027(1)
C(26)	0.1187(2)	0.5544(3)	0.1402(3)	0.033(1)
C(27)	0.1088(2)	0.6185(2)	0.1616(3)	0.032(1)
C(28)	0.0840(2)	0.6298(3)	0.2169(3)	0.041(1)
C(29)	0.0688(2)	0.5754(3)	0.2506(4)	0.046(2)

C(30)	0.0783(2)	0.5116(3)	0.2302(4)	0.046(2)
C(31)	0.1021(2)	0.5010(3)	0.1743(3)	0.038(1)
C(32)	0.1680(2)	0.6674(2)	0.0166(3)	0.033(1)
C(33)	0.2097(3)	0.6520(3)	-0.0811(4)	0.057(2)
C(34)	0.2043(4)	0.5951(5)	-0.1352(5)	0.095(3)
C(35)	0.2104(2)	0.4578(2)	0.0837(3)	0.032(1)
C(36)	0.2196(2)	0.4533(3)	0.1610(4)	0.040(1)
C(37)	0.2575(2)	0.4109(3)	0.2003(4)	0.047(2)
C(38)	0.2826(2)	0.3750(3)	0.1592(4)	0.046(2)
C(39)	0.2743(3)	0.3827(3)	0.0811(4)	0.049(2)
C(40)	0.2375(2)	0.4247(3)	0.0421(4)	0.045(2)
C(41)	0.1106(2)	0.4520(3)	-0.0010(3)	0.035(1)
C(42)	0.0658(2)	0.4798(3)	-0.0334(4)	0.043(2)
C(43)	0.0253(2)	0.4415(3)	-0.0603(4)	0.050(2)
C(44B)	0.0349(6)	0.3710(8)	-0.0620(10)	0.043(4)
C(44A)	0.0235(7)	0.3811(8)	-0.0471(10)	0.048(4)
C(45A)	0.0661(6)	0.3518(7)	-0.0106(10)	0.049(4)
C(45B)	0.0811(5)	0.3442(6)	-0.0347(8)	0.037(4)
C(46A)	0.1097(6)	0.3885(7)	0.0172(11)	0.047(4)
C(46B)	0.1214(5)	0.3845(5)	-0.0042(8)	0.025(3)
C(47)	0.3203(4)	0.7505(5)	0.3286(6)	0.089(3)
C(48)	0.1871(4)	0.6711(4)	0.3397(5)	0.082(3)
C(49)	0.0354(6)	0.6315(9)	-0.1239(11)	0.073(5)
C(50)	0.0390(5)	0.6811(8)	-0.1084(9)	0.062(4)
Cl(1)	0.0445(1)	0.7635(1)	0.0872(1)	0.047(1)
Cl(2)	0.2124(1)	0.7602(1)	0.1834(1)	0.038(1)
Cl(3)	0.3643(2)	0.7937(2)	0.2929(2)	0.141(1)
Cl(4)	0.3465(1)	0.7117(1)	0.4091(2)	0.096(1)
Cl(5)	0.2099(1)	0.7125(1)	0.4209(1)	0.089(1)
Cl(6)	0.2215(1)	0.5996(1)	0.3305(1)	0.089(1)
Cl(7)	0.0784(2)	0.6399(3)	-0.1416(3)	0.128(2)
Cl(8)	-0.0142(3)	0.6619(5)	-0.1211(6)	0.176(3)
Cl(9)	0.0263(5)	0.7416(6)	-0.1886(8)	0.271(6)
Cl(10)	0.0000	0.6639(10)	-0.2500	0.206(7)
N(1)	0.1266(2)	0.8579(2)	0.1568(2)	0.028(1)
N(2)	0.1275(2)	0.6642(2)	0.1201(2)	0.030(1)

O(1)	0.1054(2)	0.7945(2)	0.2834(2)	0.046(1)
O(2)	0.0581(2)	0.8788(2)	0.2919(3)	0.049(1)
O(3)	0.1580(2)	0.7222(2)	-0.0035(3)	0.055(1)
O(4)	0.1975(2)	0.6296(2)	-0.0111(3)	0.045(1)
Pd(1)	0.1281(1)	0.7614(1)	0.1372(1)	0.028(1)
S(1)	0.0803(1)	1.0115(1)	0.2445(1)	0.033(1)
S(2)	0.1621(1)	0.5070(1)	0.0254(1)	0.030(1)

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**Table S-3.** Selected bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ].

C(1)-N(1)	1.337(7)	C(1)-C(2)	1.410(7)
C(1)-C(9)	1.497(8)	C(2)-C(3)	1.402(7)
C(2)-S(1)	1.728(5)	C(3)-C(4)	1.411(7)
C(3)-C(8)	1.415(7)	C(4)-C(5)	1.388(8)
C(4)-N(1)	1.388(7)	C(5)-C(6)	1.381(8)
C(5)-H(5)	0.9500	C(6)-C(7)	1.395(8)
C(6)-H(6)	0.9500	C(7)-C(8)	1.394(8)
C(7)-H(7)	0.9500	C(8)-H(8)	0.9500
C(9)-O(1)	1.193(7)	C(9)-O(2)	1.308(7)
C(10)-C(11)	1.480(10)	C(10)-O(2)	1.484(7)
C(10)-H(10A)	0.9900	C(10)-H(10B)	0.9900
C(11)-H(11B)	0.9800	C(11)-H(11C)	0.9800
C(11)-H(11A)	0.9800	C(12)-C(13)	1.381(8)
C(12)-C(17)	1.382(8)	C(12)-S(1)	1.798(6)
C(13)-C(14)	1.380(9)	C(13)-H(13)	0.9500
C(14)-C(15)	1.395(10)	C(14)-H(14)	0.9500
C(15)-C(16)	1.341(9)	C(15)-H(15)	0.9500
C(16)-C(17)	1.408(9)	C(16)-H(16)	0.9500
C(17)-H(17)	0.9500	C(18)-C(23)	1.375(9)
C(18)-C(19)	1.407(7)	C(18)-S(1)	1.792(5)
C(19)-C(20)	1.379(8)	C(19)-H(19)	0.9500
C(20)-C(21)	1.399(10)	C(20)-H(20)	0.9500
C(21)-C(22)	1.389(9)	C(21)-H(21)	0.9500
C(22)-C(23)	1.404(8)	C(22)-H(22)	0.9500
C(23)-H(23)	0.9500	C(24)-N(2)	1.346(7)
C(24)-C(25)	1.416(7)	C(24)-C(32)	1.465(7)
C(25)-C(26)	1.438(7)	C(25)-S(2)	1.713(5)
C(26)-C(31)	1.397(8)	C(26)-C(27)	1.418(7)
C(27)-C(28)	1.383(8)	C(27)-N(2)	1.389(7)
C(28)-C(29)	1.392(8)	C(28)-H(28)	0.9500
C(29)-C(30)	1.404(9)	C(29)-H(29)	0.9500
C(30)-C(31)	1.371(8)	C(30)-H(30)	0.9500
C(31)-H(31)	0.9500	C(32)-O(3)	1.191(6)
C(32)-O(4)	1.329(7)	C(33)-O(4)	1.471(7)

C(33)-C(34)	1.504(11)	C(33)-H(33B)	0.9900
C(33)-H(33A)	0.9900	C(34)-H(34B)	0.9800
C(34)-H(34C)	0.9800	C(34)-H(34A)	0.9800
C(35)-C(36)	1.356(8)	C(35)-C(40)	1.386(8)
C(35)-S(2)	1.799(6)	C(36)-C(37)	1.418(8)
C(36)-H(36)	0.9500	C(37)-C(38)	1.371(9)
C(37)-H(37)	0.9500	C(38)-C(39)	1.379(10)
C(38)-H(38)	0.9500	C(39)-C(40)	1.392(9)
C(39)-H(39)	0.9500	C(40)-H(40)	0.9500
C(41)-C(46A)	1.343(15)	C(41)-C(42)	1.373(8)
C(41)-C(46B)	1.421(12)	C(41)-S(2)	1.806(6)
C(42)-C(43)	1.368(8)	C(42)-H(42)	0.9500
C(43)-C(44A)	1.265(17)	C(43)-C(44B)	1.474(17)
C(43)-H(43)	0.9500	C(44B)-C(45B)	1.38(2)
C(44B)-H(44)	0.9500	C(44A)-C(45A)	1.35(2)
C(44A)-H(44A)	0.9500	C(45A)-C(46A)	1.42(2)
C(45A)-H(45)	0.9500	C(45B)-C(46B)	1.399(16)
C(45B)-H(45A)	0.9500	C(46A)-H(46)	0.9500
C(46B)-H(46A)	0.9500	C(47)-Cl(4)	1.655(10)
C(47)-Cl(3)	1.779(11)	C(47)-H(47B)	0.9900
C(47)-H(47A)	0.9900	C(48)-Cl(5)	1.672(9)
C(48)-Cl(6)	1.793(10)	C(48)-H(48B)	0.9900
C(48)-H(48A)	0.9900	C(49)-Cl(7)	1.348(17)
C(49)-Cl(8)	1.549(18)	C(49)-H(49A)	0.9900
C(49)-H(49B)	0.9900	C(50)-Cl(8)	1.516(17)
C(50)-Cl(7)	1.633(16)	C(50)-H(50A)	0.9900
C(50)-H(50B)	0.9900	Cl(1)-Pd(1)	2.3010(16)
Cl(2)-Pd(1)	2.3131(15)	Cl(8)-Cl(10)	2.470(10)
Cl(9)-Cl(9)#1	2.33(3)	Cl(10)-Cl(8)#1	2.470(10)
N(1)-Pd(1)	2.013(4)	N(2)-Pd(1)	2.015(4)
N(1)-C(1)-C(2)	110.2(5)	N(1)-C(1)-C(9)	118.7(5)
C(2)-C(1)-C(9)	131.0(5)	C(3)-C(2)-C(1)	107.2(5)
C(3)-C(2)-S(1)	128.7(4)	C(1)-C(2)-S(1)	123.7(4)
C(2)-C(3)-C(4)	105.5(5)	C(2)-C(3)-C(8)	135.5(5)
C(4)-C(3)-C(8)	119.0(5)	C(5)-C(4)-N(1)	127.8(5)

C(5)-C(4)-C(3)	122.6(5)	N(1)-C(4)-C(3)	109.7(5)
C(6)-C(5)-C(4)	117.0(5)	C(6)-C(5)-H(5)	121.5
C(4)-C(5)-H(5)	121.5	C(5)-C(6)-C(7)	122.6(5)
C(5)-C(6)-H(6)	118.7	C(7)-C(6)-H(6)	118.7
C(8)-C(7)-C(6)	120.4(5)	C(8)-C(7)-H(7)	119.8
C(6)-C(7)-H(7)	119.8	C(7)-C(8)-C(3)	118.5(5)
C(7)-C(8)-H(8)	120.8	C(3)-C(8)-H(8)	120.8
O(1)-C(9)-O(2)	124.6(6)	O(1)-C(9)-C(1)	123.5(5)
O(2)-C(9)-C(1)	111.9(5)	C(11)-C(10)-O(2)	110.9(6)
C(11)-C(10)-H(10A)	109.5	O(2)-C(10)-H(10A)	109.5
C(11)-C(10)-H(10B)	109.5	O(2)-C(10)-H(10B)	109.5
H(10A)-C(10)-H(10B)	108.1	C(10)-C(11)-H(11B)	109.5
C(10)-C(11)-H(11C)	109.5	H(11B)-C(11)-H(11C)	109.5
C(10)-C(11)-H(11A)	109.5	H(11B)-C(11)-H(11A)	109.5
H(11C)-C(11)-H(11A)	109.5	C(13)-C(12)-C(17)	121.5(6)
C(13)-C(12)-S(1)	114.3(4)	C(17)-C(12)-S(1)	124.1(5)
C(14)-C(13)-C(12)	119.3(6)	C(14)-C(13)-H(13)	120.3
C(12)-C(13)-H(13)	120.3	C(13)-C(14)-C(15)	120.0(6)
C(13)-C(14)-H(14)	120.0	C(15)-C(14)-H(14)	120.0
C(16)-C(15)-C(14)	119.9(7)	C(16)-C(15)-H(15)	120.0
C(14)-C(15)-H(15)	120.0	C(15)-C(16)-C(17)	121.8(6)
C(15)-C(16)-H(16)	119.1	C(17)-C(16)-H(16)	119.1
C(12)-C(17)-C(16)	117.4(6)	C(12)-C(17)-H(17)	121.3
C(16)-C(17)-H(17)	121.3	C(23)-C(18)-C(19)	122.5(5)
C(23)-C(18)-S(1)	121.9(4)	C(19)-C(18)-S(1)	115.4(5)
C(20)-C(19)-C(18)	117.4(6)	C(20)-C(19)-H(19)	121.3
C(18)-C(19)-H(19)	121.3	C(19)-C(20)-C(21)	121.4(6)
C(19)-C(20)-H(20)	119.3	C(21)-C(20)-H(20)	119.3
C(22)-C(21)-C(20)	120.2(6)	C(22)-C(21)-H(21)	119.9
C(20)-C(21)-H(21)	119.9	C(21)-C(22)-C(23)	119.4(6)
C(21)-C(22)-H(22)	120.3	C(23)-C(22)-H(22)	120.3
C(18)-C(23)-C(22)	119.1(5)	C(18)-C(23)-H(23)	120.4
C(22)-C(23)-H(23)	120.4	N(2)-C(24)-C(25)	109.9(5)
N(2)-C(24)-C(32)	120.5(4)	C(25)-C(24)-C(32)	129.5(5)
C(24)-C(25)-C(26)	107.3(4)	C(24)-C(25)-S(2)	123.1(4)
C(26)-C(25)-S(2)	129.2(4)	C(31)-C(26)-C(27)	119.5(5)

C(31)-C(26)-C(25)	135.9(5)	C(27)-C(26)-C(25)	104.5(5)
C(28)-C(27)-N(2)	128.0(5)	C(28)-C(27)-C(26)	121.7(5)
N(2)-C(27)-C(26)	110.3(5)	C(27)-C(28)-C(29)	117.2(5)
C(27)-C(28)-H(28)	121.4	C(29)-C(28)-H(28)	121.4
C(28)-C(29)-C(30)	121.8(6)	C(28)-C(29)-H(29)	119.1
C(30)-C(29)-H(29)	119.1	C(31)-C(30)-C(29)	120.5(6)
C(31)-C(30)-H(30)	119.8	C(29)-C(30)-H(30)	119.8
C(30)-C(31)-C(26)	119.2(5)	C(30)-C(31)-H(31)	120.4
C(26)-C(31)-H(31)	120.4	O(3)-C(32)-O(4)	123.9(5)
O(3)-C(32)-C(24)	125.1(5)	O(4)-C(32)-C(24)	111.0(4)
O(4)-C(33)-C(34)	108.2(5)	O(4)-C(33)-H(33B)	110.1
C(34)-C(33)-H(33B)	110.1	O(4)-C(33)-H(33A)	110.1
C(34)-C(33)-H(33A)	110.1	H(33B)-C(33)-H(33A)	108.4
C(33)-C(34)-H(34B)	109.5	C(33)-C(34)-H(34C)	109.5
H(34B)-C(34)-H(34C)	109.5	C(33)-C(34)-H(34A)	109.5
H(34B)-C(34)-H(34A)	109.5	H(34C)-C(34)-H(34A)	109.5
C(36)-C(35)-C(40)	122.8(5)	C(36)-C(35)-S(2)	123.7(4)
C(40)-C(35)-S(2)	113.5(4)	C(35)-C(36)-C(37)	118.4(5)
C(35)-C(36)-H(36)	120.8	C(37)-C(36)-H(36)	120.8
C(38)-C(37)-C(36)	119.2(6)	C(38)-C(37)-H(37)	120.4
C(36)-C(37)-H(37)	120.4	C(37)-C(38)-C(39)	121.4(6)
C(37)-C(38)-H(38)	119.3	C(39)-C(38)-H(38)	119.3
C(38)-C(39)-C(40)	119.6(6)	C(38)-C(39)-H(39)	120.2
C(40)-C(39)-H(39)	120.2	C(35)-C(40)-C(39)	118.4(6)
C(35)-C(40)-H(40)	120.8	C(39)-C(40)-H(40)	120.8
C(46A)-C(41)-C(42)	115.7(9)	C(42)-C(41)-C(46B)	124.9(7)
C(46A)-C(41)-S(2)	127.2(8)	C(42)-C(41)-S(2)	116.4(4)
C(46B)-C(41)-S(2)	116.7(6)	C(43)-C(42)-C(41)	120.4(6)
C(43)-C(42)-H(42)	119.8	C(41)-C(42)-H(42)	119.8
C(44A)-C(43)-C(42)	124.5(10)	C(42)-C(43)-C(44B)	115.4(8)
C(44A)-C(43)-H(43)	117.8	C(42)-C(43)-H(43)	117.8
C(44B)-C(43)-H(43)	123.3	C(45B)-C(44B)-C(43)	122.9(12)
C(45B)-C(44B)-H(44)	118.5	C(43)-C(44B)-H(44)	118.5
C(43)-C(44A)-C(45A)	116.7(14)	C(43)-C(44A)-H(44A)	121.6
C(45A)-C(44A)-H(44A)	121.6	C(44A)-C(45A)-C(46A)	121.2(14)
C(44A)-C(45A)-H(45)	119.4	C(46A)-C(45A)-H(45)	119.4

C(44B)-C(45B)-C(46B)	120.3(12)	C(44B)-C(45B)-H(45A)	119.9
C(46B)-C(45B)-H(45A)	119.9	C(41)-C(46A)-C(45A)	119.7(13)
C(41)-C(46A)-H(46)	120.1	C(45A)-C(46A)-H(46)	120.1
C(45B)-C(46B)-C(41)	115.3(10)	C(45B)-C(46B)-H(46A)	122.4
C(41)-C(46B)-H(46A)	122.4	Cl(4)-C(47)-Cl(3)	111.4(6)
Cl(4)-C(47)-H(47B)	109.4	Cl(3)-C(47)-H(47B)	109.4
Cl(4)-C(47)-H(47A)	109.4	Cl(3)-C(47)-H(47A)	109.4
H(47B)-C(47)-H(47A)	108.0	Cl(5)-C(48)-Cl(6)	113.7(5)
Cl(5)-C(48)-H(48B)	108.8	Cl(6)-C(48)-H(48B)	108.8
Cl(5)-C(48)-H(48A)	108.8	Cl(6)-C(48)-H(48A)	108.8
H(48B)-C(48)-H(48A)	107.7	Cl(7)-C(49)-Cl(8)	147.1(15)
Cl(7)-C(49)-H(49A)	100.0	Cl(8)-C(49)-H(49A)	100.0
Cl(7)-C(49)-H(49B)	100.0	Cl(8)-C(49)-H(49B)	100.0
H(49A)-C(49)-H(49B)	104.2	Cl(8)-C(50)-Cl(7)	123.9(12)
Cl(8)-C(50)-H(50A)	106.4	Cl(7)-C(50)-H(50A)	106.4
Cl(8)-C(50)-H(50B)	106.4	Cl(7)-C(50)-H(50B)	106.4
H(50A)-C(50)-H(50B)	106.4	C(50)-Cl(8)-Cl(10)	74.7(7)
C(49)-Cl(8)-Cl(10)	66.5(8)	Cl(8)-Cl(10)-Cl(8)#1	178.1(11)
C(1)-N(1)-C(4)	107.4(4)	C(1)-N(1)-Pd(1)	127.4(4)
C(4)-N(1)-Pd(1)	124.6(3)	C(24)-N(2)-C(27)	108.1(4)
C(24)-N(2)-Pd(1)	126.8(3)	C(27)-N(2)-Pd(1)	124.9(3)
C(9)-O(2)-C(10)	114.2(5)	C(32)-O(4)-C(33)	117.0(4)
N(1)-Pd(1)-N(2)	177.66(18)	N(1)-Pd(1)-Cl(1)	89.07(13)
N(2)-Pd(1)-Cl(1)	89.58(13)	N(1)-Pd(1)-Cl(2)	90.75(13)
N(2)-Pd(1)-Cl(2)	90.66(13)	Cl(1)-Pd(1)-Cl(2)	178.05(6)
C(2)-S(1)-C(18)	108.2(3)	C(2)-S(1)-C(12)	102.4(3)
C(18)-S(1)-C(12)	104.2(3)	C(25)-S(2)-C(35)	108.9(3)
C(25)-S(2)-C(41)	103.9(3)	C(35)-S(2)-C(41)	104.0(3)

Symmetry transformations used to generate equivalent atoms:

#1 -x,y,-z-1/2

**Table S-4. Anisotropic displacement parameters ( $\text{\AA}^2$ ).**

The anisotropic displacement factor exponent takes the form:

$$-2\pi^2 [ h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12} ].$$

	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>23</sub>	U <sub>13</sub>	U <sub>12</sub>
C(1)	0.044(3)	0.020(3)	0.032(3)	-0.003(2)	0.007(3)	-0.001(2)
C(2)	0.044(3)	0.016(2)	0.034(3)	-0.010(2)	0.009(3)	-0.006(2)
C(3)	0.040(3)	0.017(2)	0.031(3)	-0.007(2)	0.007(2)	-0.002(2)
C(4)	0.034(3)	0.018(2)	0.030(3)	-0.001(2)	0.001(2)	-0.002(2)
C(5)	0.048(4)	0.020(3)	0.031(3)	-0.006(2)	0.006(3)	0.004(2)
C(6)	0.052(4)	0.033(3)	0.029(3)	-0.002(2)	0.014(3)	0.003(3)
C(7)	0.051(4)	0.026(3)	0.033(3)	0.004(2)	0.008(3)	-0.004(2)
C(8)	0.047(3)	0.018(3)	0.039(3)	-0.006(2)	0.007(3)	-0.003(2)
C(9)	0.049(4)	0.032(3)	0.032(3)	-0.012(2)	0.009(3)	-0.008(3)
C(10)	0.072(5)	0.036(3)	0.058(5)	0.006(3)	0.029(4)	0.002(3)
C(11)	0.077(5)	0.059(5)	0.049(4)	0.002(3)	0.019(4)	-0.017(4)
C(12)	0.042(3)	0.025(3)	0.039(3)	-0.005(2)	0.013(3)	0.001(2)
C(13)	0.042(4)	0.031(3)	0.064(4)	-0.002(3)	0.012(3)	-0.002(3)
C(14)	0.043(4)	0.053(4)	0.072(5)	-0.012(4)	0.001(4)	-0.003(3)
C(15)	0.046(4)	0.049(4)	0.056(4)	-0.001(3)	0.011(3)	0.004(3)
C(16)	0.059(4)	0.031(3)	0.061(4)	0.010(3)	0.023(4)	0.004(3)
C(17)	0.046(4)	0.027(3)	0.048(4)	-0.001(3)	0.009(3)	-0.002(3)
C(18)	0.056(4)	0.014(2)	0.031(3)	-0.006(2)	0.011(3)	-0.005(2)
C(19)	0.064(4)	0.021(3)	0.031(3)	-0.005(2)	0.009(3)	0.000(3)
C(20)	0.085(5)	0.018(3)	0.026(3)	-0.006(2)	0.002(3)	0.001(3)
C(21)	0.073(5)	0.021(3)	0.032(3)	-0.003(2)	-0.003(3)	-0.012(3)
C(22)	0.057(4)	0.036(3)	0.038(4)	-0.003(3)	0.006(3)	-0.008(3)
C(23)	0.069(5)	0.024(3)	0.029(3)	-0.010(2)	0.014(3)	-0.006(3)
C(24)	0.042(3)	0.015(2)	0.029(3)	0.001(2)	0.005(2)	0.003(2)
C(25)	0.039(3)	0.017(2)	0.025(3)	-0.004(2)	0.007(2)	0.001(2)
C(26)	0.044(3)	0.023(3)	0.031(3)	0.003(2)	0.007(3)	0.004(2)
C(27)	0.041(3)	0.018(2)	0.034(3)	-0.001(2)	0.007(3)	-0.001(2)
C(28)	0.053(4)	0.032(3)	0.040(4)	-0.004(3)	0.015(3)	0.004(3)
C(29)	0.059(4)	0.043(4)	0.043(4)	-0.005(3)	0.023(3)	-0.002(3)
C(30)	0.061(4)	0.033(3)	0.049(4)	0.007(3)	0.023(3)	-0.002(3)

C(31)	0.051(4)	0.023(3)	0.041(3)	0.003(2)	0.014(3)	0.003(2)
C(32)	0.052(4)	0.019(3)	0.024(3)	-0.005(2)	0.003(3)	0.001(2)
C(33)	0.081(5)	0.031(3)	0.075(5)	0.019(3)	0.051(4)	0.008(3)
C(34)	0.153(10)	0.074(6)	0.068(6)	-0.005(5)	0.047(6)	-0.009(6)
C(35)	0.042(3)	0.016(2)	0.036(3)	-0.002(2)	0.007(3)	0.001(2)
C(36)	0.050(4)	0.024(3)	0.046(4)	-0.003(3)	0.015(3)	0.007(3)
C(37)	0.061(4)	0.031(3)	0.041(4)	0.002(3)	0.001(3)	-0.002(3)
C(38)	0.032(3)	0.025(3)	0.076(5)	0.007(3)	0.005(3)	0.004(2)
C(39)	0.060(4)	0.026(3)	0.065(5)	-0.001(3)	0.023(4)	0.011(3)
C(40)	0.057(4)	0.028(3)	0.048(4)	0.003(3)	0.012(3)	0.004(3)
C(41)	0.053(4)	0.021(3)	0.027(3)	-0.005(2)	0.005(3)	-0.006(2)
C(42)	0.041(4)	0.029(3)	0.055(4)	-0.014(3)	0.004(3)	0.001(3)
C(43)	0.047(4)	0.052(4)	0.045(4)	-0.012(3)	0.002(3)	-0.006(3)
C(47)	0.085(7)	0.086(7)	0.077(7)	-0.016(5)	-0.012(5)	0.014(5)
C(48)	0.101(7)	0.073(6)	0.053(5)	0.015(4)	-0.014(5)	-0.008(5)
Cl(1)	0.043(1)	0.021(1)	0.067(1)	-0.007(1)	-0.002(1)	0.000(1)
Cl(2)	0.043(1)	0.027(1)	0.040(1)	0.002(1)	0.005(1)	0.005(1)
Cl(3)	0.246(4)	0.095(2)	0.086(2)	-0.014(2)	0.049(3)	0.010(2)
Cl(4)	0.106(2)	0.072(2)	0.094(2)	-0.012(1)	-0.001(2)	0.000(1)
Cl(5)	0.150(2)	0.043(1)	0.057(1)	0.007(1)	-0.001(1)	-0.007(1)
Cl(6)	0.138(2)	0.079(2)	0.045(1)	-0.002(1)	0.014(1)	0.017(1)
N(1)	0.037(3)	0.019(2)	0.025(2)	-0.003(2)	0.002(2)	0.002(2)
N(2)	0.046(3)	0.017(2)	0.024(2)	-0.004(2)	0.003(2)	-0.002(2)
O(1)	0.074(3)	0.021(2)	0.043(3)	0.001(2)	0.016(2)	-0.004(2)
O(2)	0.071(3)	0.024(2)	0.064(3)	0.005(2)	0.037(3)	0.000(2)
O(3)	0.109(4)	0.021(2)	0.042(3)	0.011(2)	0.032(3)	0.017(2)
O(4)	0.060(3)	0.029(2)	0.054(3)	0.013(2)	0.028(2)	0.005(2)
Pd(1)	0.043(1)	0.012(1)	0.025(1)	-0.003(1)	0.004(1)	0.002(1)
S(1)	0.049(1)	0.016(1)	0.036(1)	-0.008(1)	0.015(1)	-0.003(1)
S(2)	0.045(1)	0.015(1)	0.028(1)	-0.001(1)	0.007(1)	0.000(1)

## Part II NMR spectra



