

1 **Ring-Opening Polymerization Reaction of Polyhedral**
2 **Oligomeric Silsesquioxanes (POSSs) for Preparation of**
3 **Well-controlled 3D Skeletal Hybrid Monoliths**

4
5 **Supporting Information**

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25 **Experimental section**

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27 **Materials**

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29 Octaglycidyl dimethylsilyl POSS (PSS-octa[(3-glycidyloxypropyl)dimethylsiloxy] substituted)
30 (POSS-epoxy), hexamethylene diamine (HDA), 1,8-diaminooctane (DAO), 1,10-diaminodecane
31 (DAD), 1,12-diaminododecane (DADD), poly(ethylene glycol) (PEG, Mn=10,000) and
32 (3-aminopropyl)triethoxysilane (APTES) were purchased from Aldrich (Milwaukee, WI, USA).
33 The fused-silica capillaries with dimension of 50 and 75 μm i.d. and 365 μm o.d. were obtained
34 from the Refine Chromatography Ltd. (Yongnian, Hebei, China). Trypsin was purchased from
35 Promega (Madison, WI, USA). Bovine serum albumin (BSA), thiourea, benzene, toluene, phenol
36 and other standard compounds were all obtained from Sigma (St Louis, MO, USA). Dithiothreitol
37 (DTT) and iodoacetamide (IAA) were purchased from Sino-American Biotechnology Corporation
38 (Beijing, China). HPLC-grade acetonitrile (ACN) was used for the preparation of mobile phases.
39 The water used in all experiments was doubly distilled and purified by a Milli-Q system
40 (Millipore Inc., Milford, MA, USA). C18-particles (5 μm , 120 \AA pore) were purchased from
41 DAISO (Osaka, Japan). Other chemical reagents were all of analytical grade.

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43 **Preparation and modification of the POSS-based hybrid monoliths**

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45 Before the preparation of the monolithic columns, the fused-silica capillary was pretreated and
46 rinsed by 1.0 M NaOH for 4 h, water for 30 min, 1.0 M HCl for 14 h, and water for another 30
47 min, successively, and then dried by a nitrogen stream at room temperature. The mixture of
48 (3-aminopropyl)triethoxysilane (APTES)/methanol (50%, v/v) was used to introduce the $-\text{NH}_2$
49 group onto the inner surface of capillary for anchoring monolith matrices to the capillary inner
50 wall, as described in the previous report,¹ and then the prepolymerization mixture with the feed
51 recipes listed in Table 1 was injected into the modified capillaries with a syringe. The fulfilled
52 capillaries were then sealed with rubber stoppers and immersed in a water bath at 50 $^\circ\text{C}$ for 24 h.
53 After that, the capillaries were flush with methanol. For bulk hybrid monoliths, the

54 prepolymerization mixture was placed in a centrifuge tube and reacted under the same conditions.
55 After polymerization, the bulk hybrid monoliths were cut into smaller pieces, extracted with
56 ethanol overnight in a Soxhlet apparatus and dried in a vacuum. For the modification of the
57 resulting hybrid capillary monolithic column, a H₂O/ACN (50%, v/v) mixture contain 2.5% (v/v)
58 ammonia was pumped through the column and reacted at 50 °C overnight, then the columns were
59 washed with methanol.

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61 **Instrumentation**

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63 The chromatographic evaluation of hybrid monolithic columns was performed on an LC system
64 equipped with an Agilent 1100 (Hewlett-Packard) micropump and a UV detector (K-2501, Knauer,
65 Germany). Data was collected at 214 nm, and processed by a chromatography workstation
66 (Beijing Cailu Scientific Instrument Ltd., Beijing, China). A 7725i injector with a 20 µL sample
67 loop was used. A T-union connector served as a splitter with one end connected to the capillary
68 monolithic column and the other end to a blank capillary (95-cm long, 50 µm i.d. and 365 µm o.d.).
69 The split ratio was controlled at about 1/240. The outlet of the hybrid monolithic column was
70 connected with a Teflon tube to a empty fused-silica capillary (75 µm i.d. and 365 µm o.d.), where
71 a detection window was made by removing a 2 mm length of the polyimide coating in a position
72 of 5.5 cm from the separation monolithic column outlet.

73 SEM images were obtained by using a JEOL JSM-5600 scanning electron microscope (JEOL,
74 Tokyo, Japan). FT-IR spectra were measured with a Bruker Tensor 27 FT-IR spectrometer (Bruker
75 Daltonics, Ettlingen, Germany). Pore size measurement was performed on an Autopore IV 9500
76 (Micromeritics, Norcross, USA). Thermal gravimetric analysis was carried out on a Setsys 16/18
77 (Setaram, Caluire, France). Nitrogen adsorption/desorption measurements of dry bulk monoliths
78 were performed on a Quadrasorb SI surface area analyzer and pore size analyzer (Quantachrome,
79 Boynton Beach, USA).

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81 **Preparation of BSA Tryptic Digest and cHPLC-MS Analysis.**

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83 The 2 mg BSA was dissolved in 1 mL of denaturing buffer containing 8 M urea and 100 mM

84 ammonium bicarbonate. After the addition of 20 μ L of DTT (20 mM) solution, the mixture was
85 incubated at 60 $^{\circ}$ C for 1 h to reduce the disulfide bonds of the protein. Subsequently, 7.48 mg IAA
86 was added, and then the mixture was incubated at room temperature in the dark for 40 min. After
87 that, the mixture was diluted 8-fold with 100 mM ammonium bicarbonate buffer (pH 8.2) and
88 digested at 37 $^{\circ}$ C for 16 h with trypsin at enzyme-to-substrate ratio of 1:25 (w/w). After digesting,
89 the pH value of the tryptic digestion solution was adjusted to 2-3 by 10% trifluoroacetic acid
90 aqueous solution. Solid-phase extraction (SPE) was performed with a homemade C18 cartridge.
91 The collected peptides were dried under vacuum and dissolved in a 0.1% formic acid aqueous
92 solution (2 mL), and then stored in a -20 $^{\circ}$ C freezer before cLC-MS/MS analysis.

93 The cLC-MS/MS analysis was carried out on a Finnigan LTQ ion trap mass spectrometer
94 (Finnigan, San Jose, USA) which interfaced with a surveyor MS pump. Buffer A was water
95 (containing 0.1% formic acid), and buffer B was 100% ACN (containing 0.1% formic acid).
96 Tryptic digests were automatically injected onto the column with buffer A for 3 min at the flow
97 rate of 5 μ L/min, the trapped peptides were then separated at a flow rate of ca. 200 nL/min on the
98 POSS monolith (40 cm in length \times 75 μ m i.d.) or the C18-particles packed column (12 cm in
99 length \times 75 μ m i.d.) with an integrated emitter, which was prepared by directly tapering the tip
100 from the outlet of the capillary. The separation was performed with gradient elution from 5 to 80%
101 ACN (containing 0.1% formic acid) within 65 min. The LTQ linear ion trap mass spectrometer
102 was equipped with a nanospray ion source, and the temperature of the ion transfer capillary was
103 set at 200 $^{\circ}$ C. The spray voltage was set at 1.8 kV, and the normalized collision energy was set at
104 35.0%. One microscan was set for each MS and MS/MS scan. All MS and MS/MS spectra were
105 acquired in the data dependent mode. The mass spectrometer was set such that one full MS scan
106 was followed by six MS/MS scans on the six most intense ions. The dynamic exclusion function
107 was set as follows: repeat count 2, repeat duration 30 s, and exclusion duration 90 s. System
108 control and data collection were done by Xcalibur software version 1.4 (Thermo, USA). The scan
109 range was set from m/z 400 to m/z 1600.

110 The acquired MS/MS spectra were searched on a database using the SEQUEST (version 0.28)
111 against an IPI_bovine_BOVIN_3.32 (32946 sequences; 16109453 residues). Cysteine residues
112 were searched as fixed modification of 57.0215 Da, and methionine residues as variable
113 modification of 15.9949 Da. Peptides were searched using fully tryptic cleavage constraints and

114 up to two internal cleavage sites were allowed for tryptic digestion. The mass tolerances were 2
115 Da for parent mass and 1 Da for fragment masses.

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117 **References**

118 1. L. Ren, Z. Liu, Y. Liu, P. Dou and H.-Y. Chen, *Angew. Chem. Int. Ed.* 2009, **48**, 6704-6707.

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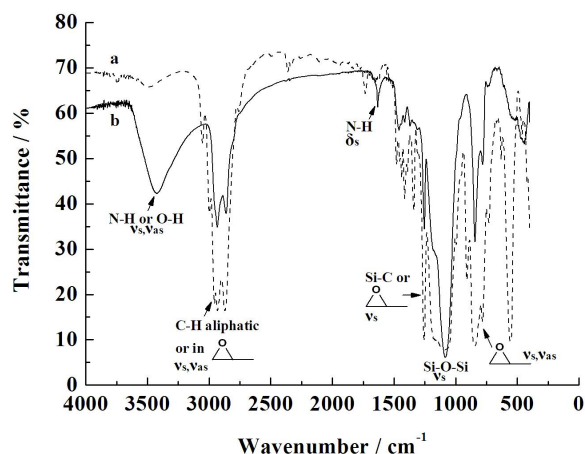
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144 **Supplementary Figures**



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146 **Fig. S1** FT-IR spectra of a) octaglycidyl dimethylsilyl POSS (POSS-epoxy), b) POSS-HDA
147 monolith. v_a : symmetric vibration, v_{as} : asymmetric vibration, δ_s : in-plane bending (scissoring).²

148 As shown in Fig. S1, a strong band at about 1085 cm⁻¹ is almost unchanged before and after
149 reaction, whilst the peak at 1255 cm⁻¹ and the peaks range from 725 cm⁻¹ to 910 cm⁻¹ (assigned to
150 the epoxy groups) are decreased. The broad peak at 3428 cm⁻¹ (assigned to the N-H and O-H
151 groups) is increased remarkably, which was the product were production of the
152 ring-opening reaction. In addition, a N-H peak at 1630 cm⁻¹ is observed only on the spectra of the
153 hybrid materials. All these results demonstrated the occurrence of the ring-opening reaction
154 forcefully.

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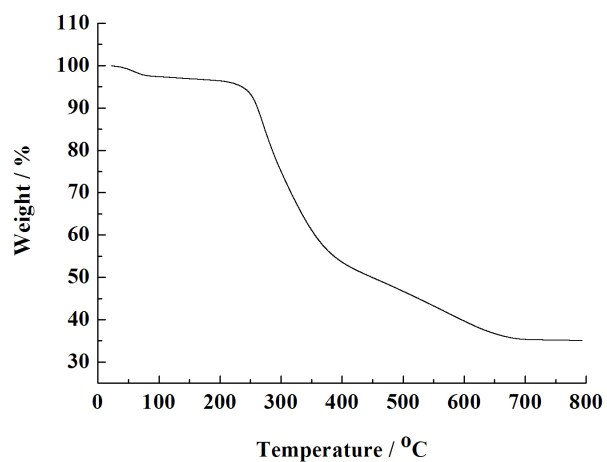
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168 **Fig. S2** Thermal gravimetric analysis of the monolith POSS-HDA at a heating rate of 10 °C/min in
169 air. The significant weight lost indicates the occurrence of the ring-opening polymerization.

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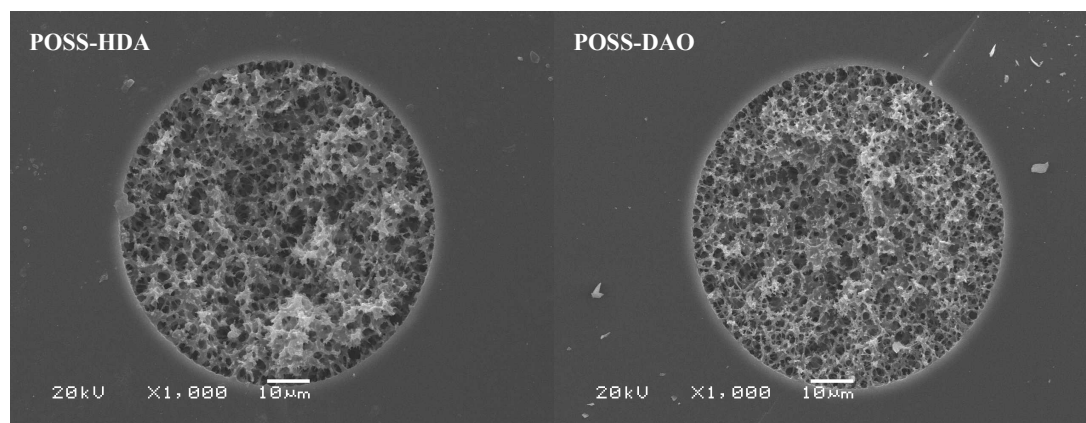
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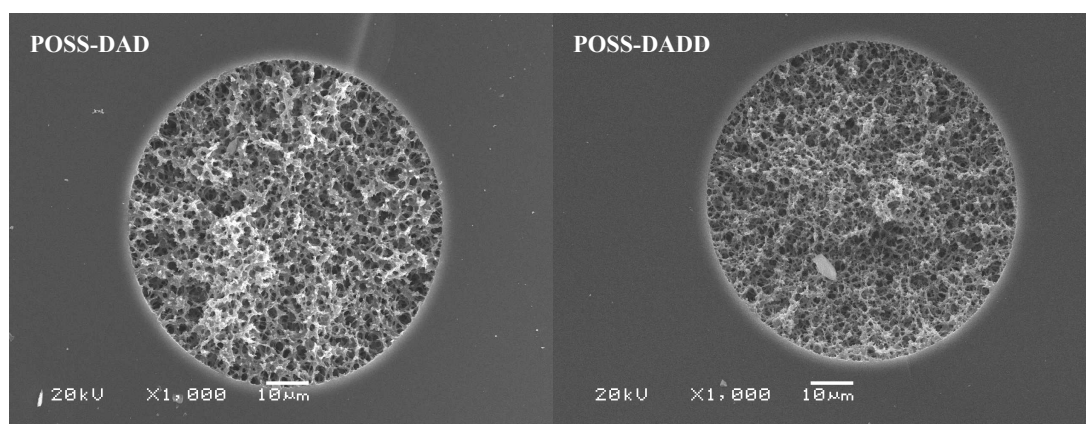
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191 **Fig. S3** The cross-section of the monolithic columns prepared with different diamines.

192 Magnification: 1000×.

193 The monolithic matrices anchored to the inner wall of the capillaries tightly, and well-controlled

194 3D skeleton was all achieved on these four monolithic columns.

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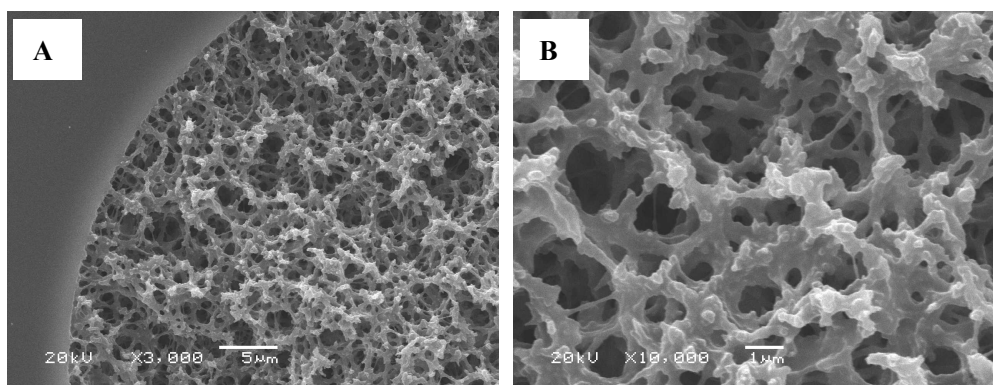
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207 **Fig. S4** SEM images of the cross-section of the hybrid POSS-DADD monolithic column.

208 Magnification: (A) 3000× and (B) 10000×.

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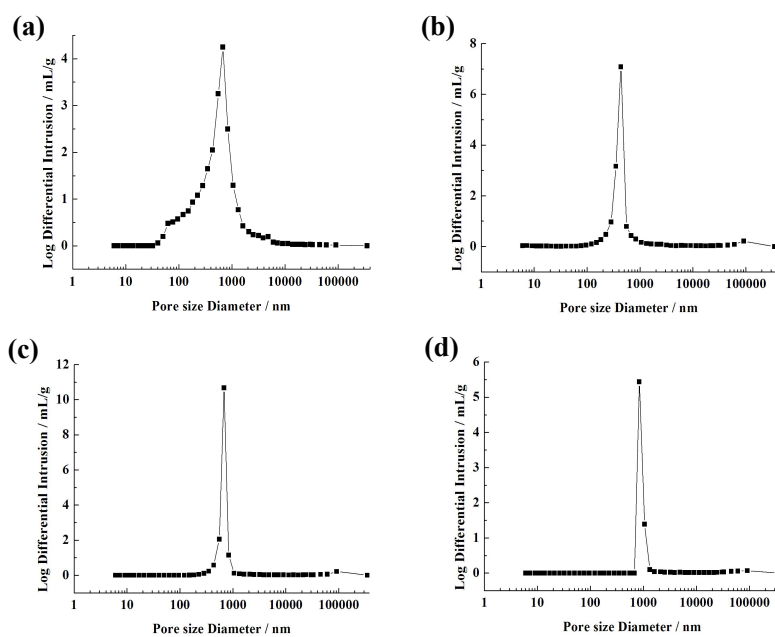
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240 **Fig. S5** Pore size distribution of the monolith POSS-DADD (a) and other monoliths 1 (b), 4 (c)

241 and 5 (d) in Table S2 in the Supporting Information by the mercury intrusion method.

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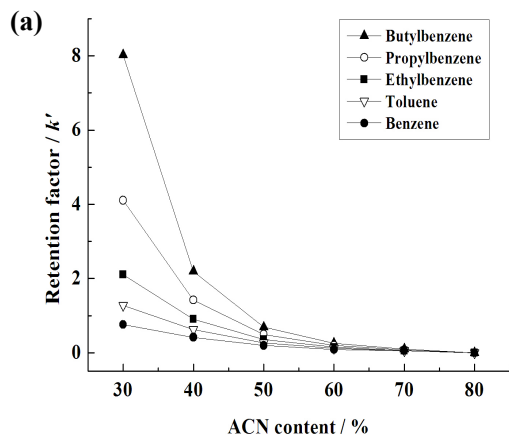
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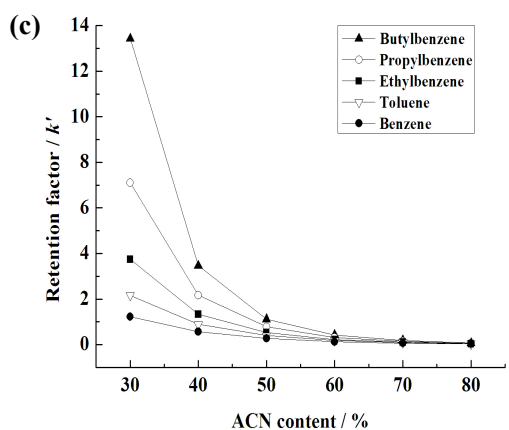
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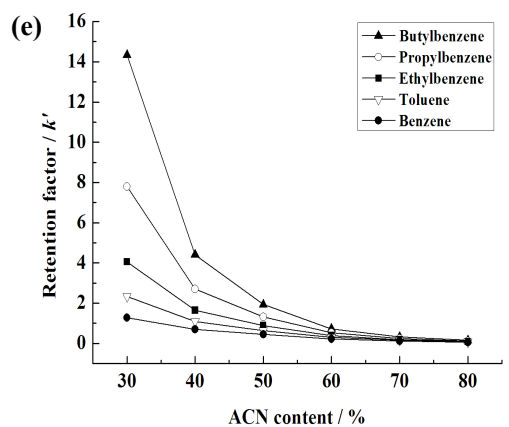
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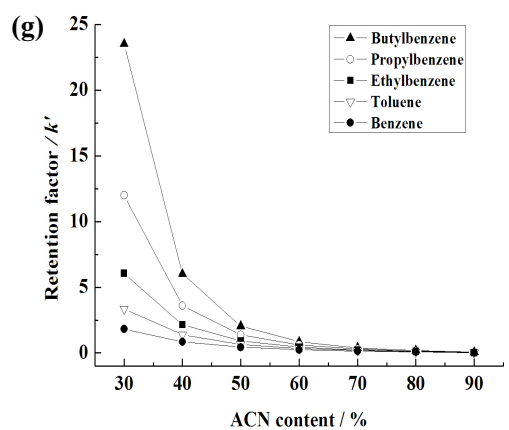
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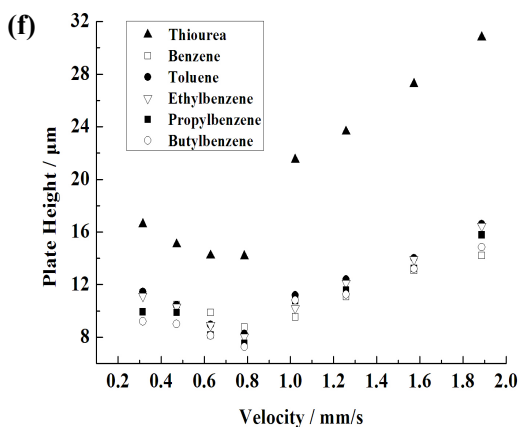
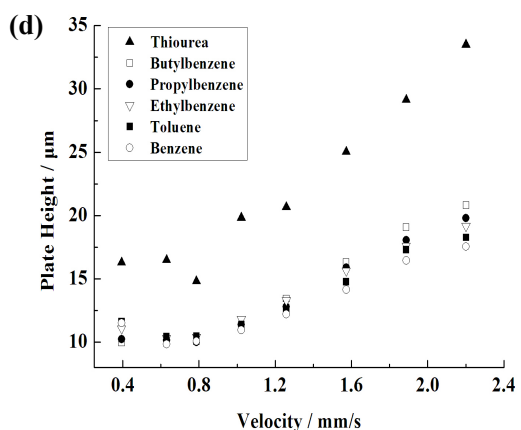
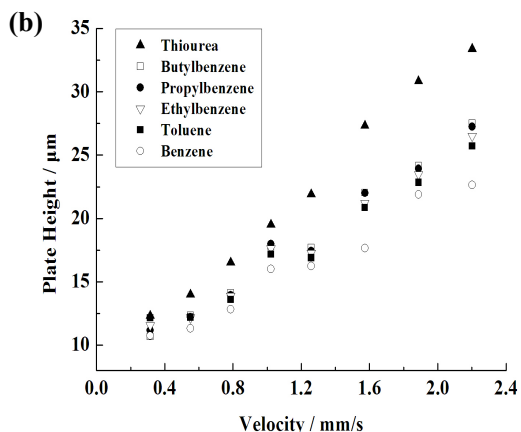
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289 **Fig. S6** The influence of ACN concentration on the retention factors of alkylbenzenes (a, c, e, g)
290 and the dependence of their plate heights on the linear velocity of the mobile phase (b, d, f) by the
291 monoliths POSS-HDA (a, b), POSS-DAO (c, d), POSS-DAD (e, f) and POSS-DADD (g) in cLC.
292 Experimental conditions: column dimension for (a, b), 41 cm × 75 μm i.d., for (c, d), 49 cm × 75
293 μm i.d., for (e, f), 49 cm × 75 μm i.d., for (g), 48 cm × 75 μm i.d.; flow rate for (a, e), 50 μL/min
294 (before split), for (c), 80 μL/min (before split), for (g), 100 μL/min (before split); mobile phase for
295 (b, f), ACN/water (50/50, v/v), for (d), ACN/water (45/55, v/v); injection volume, 2.5 μL in split
296 mode; detection wavelength, 214 nm.

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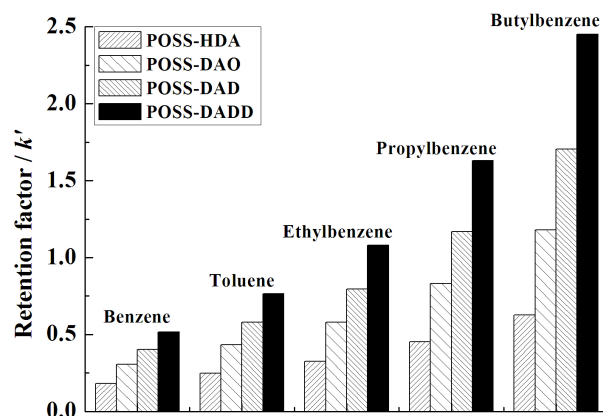
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321 **Fig. S7** Comparison of the hydrophobicity of the hybrid monolithic columns prepared with
322 different diamines. Experimental conditions: flow rate, 50 $\mu\text{L}/\text{min}$ (before split); mobile phase,
323 ACN/water (50/50, v/v); injection volume, 2.5 μL in split mode; detection wavelength, 214 nm.

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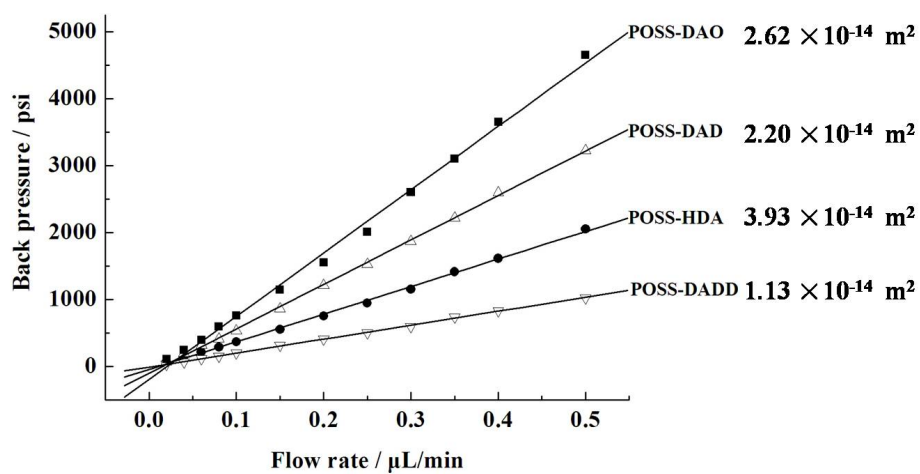
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343 **Fig. S8** The relationships between flow rate and the back pressure drop of the monolithic columns.

344 Experimental condition: mobile phase, ACN/water (40/60, v/v).

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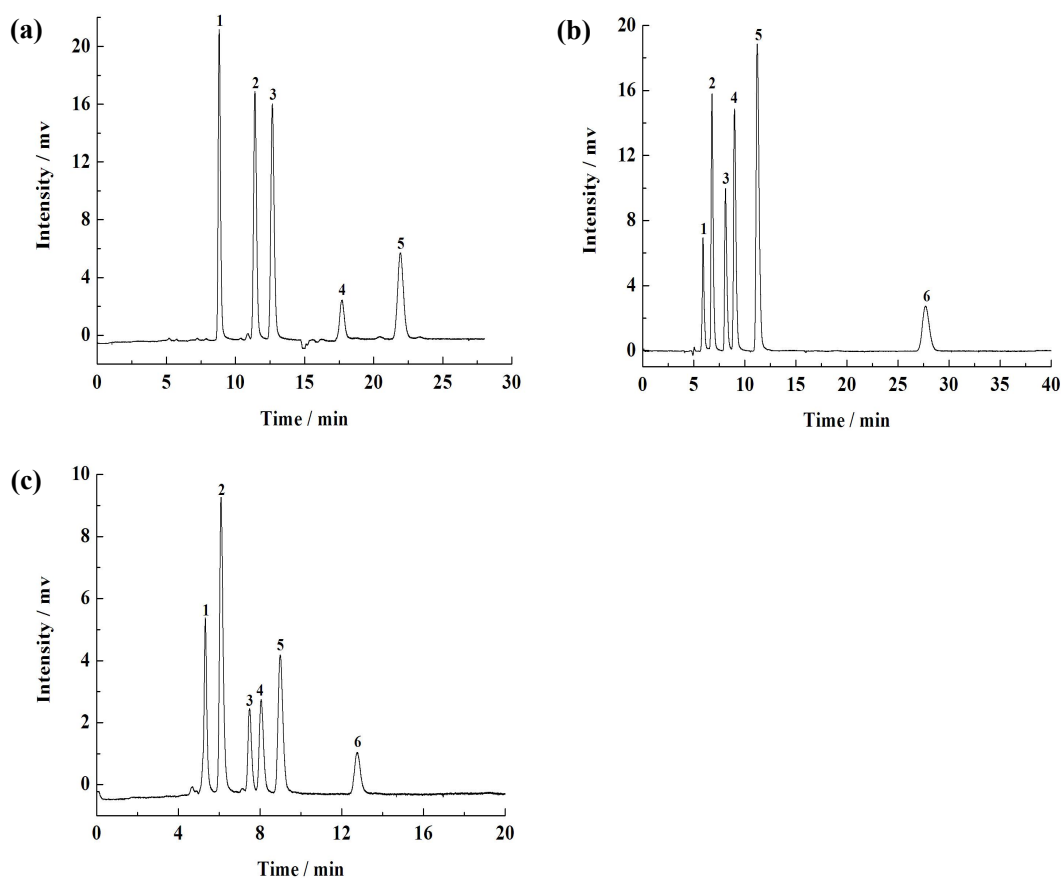
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377 **Fig. S9** Separations of polycyclic aromatic hydrocarbons (PAHs) (a), phenols (b) and anilines (c)

378 on the POSS-DADD monolithic column in cLC. Analytes: (a), (1) naphthalene, (2) acenaphthene,

379 (3) 4,4'-dimethylbiphenyl, (4) p-terphenyl, (5) pyrene; (b), (1) hydroquinone, (2) resorcinol, (3)

380 pyrocatechol, (4) phenol, (5) 4-cresol, (6) 4-tert-butylphenol; (c), (1) 2,4-diaminotoluene, (2)

381 benzidine, (3) 2-nitroaniline, (4) 2,4-dinitroaniline, (5) 4-aminobiphenyl, (6)

382 2,6-dichloro-4-nitroaniline; Experimental conditions: column dimensions, 47 cm × 75 μm i.d.;

383 mobile phase for (a), ACN/water (60/40, v/v), for (b), ACN/water (35/65, v/v), for (c), ACN/water

384 (50/50, v/v); flow rate, 60 μL/min (before split); injection volume, 2.5 μL in split mode; detection

385 wavelength, 214 nm.

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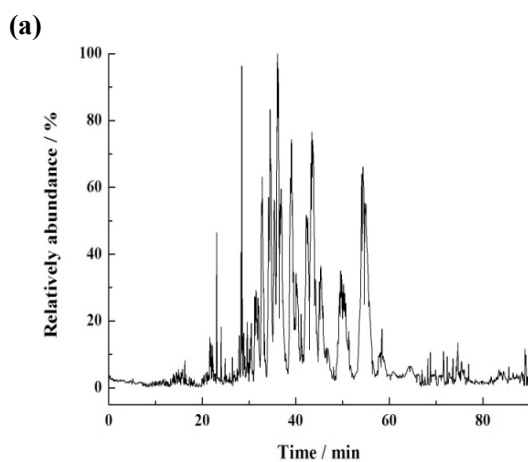
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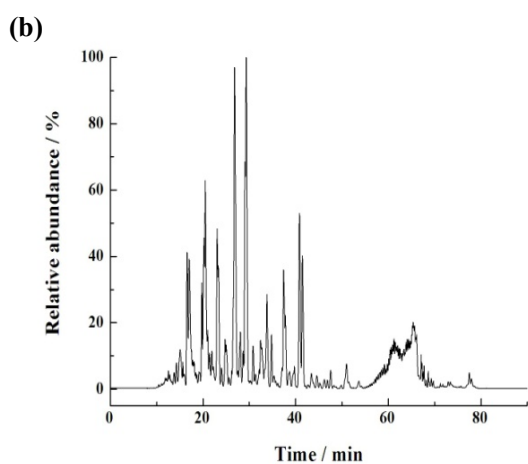
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408 **Fig. S10** Base-peak chromatograms of cLC-MS/MS analysis of a BSA tryptic digest on monolith

409 POSS-DADD (a) and C18-particles packed column (b). Experimental conditions: column

410 dimensions for (a), 40 cm \times 75 μ m i.d., for (b), 12 cm \times 75 μ m i.d.; mobile phase: buffer A, 100%

411 water (containing 0.1% formic acid), buffer B, 100% ACN (containing 0.1% formic acid);

412 separation gradient, buffer B from 5% to 80% in 65 min, flow rate, 200 nL/min (after split).

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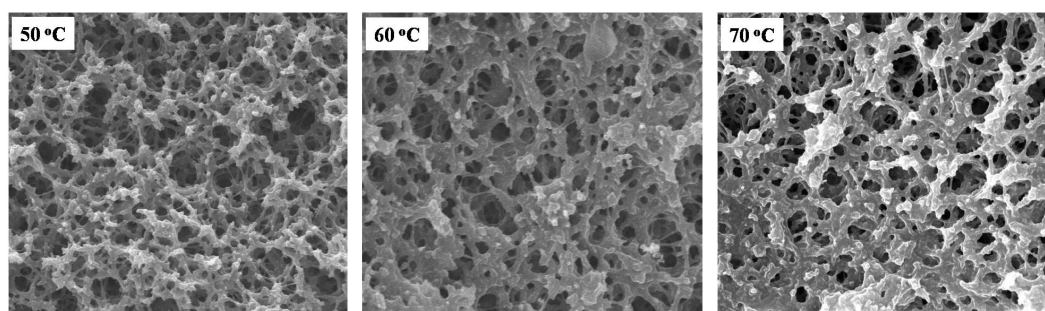
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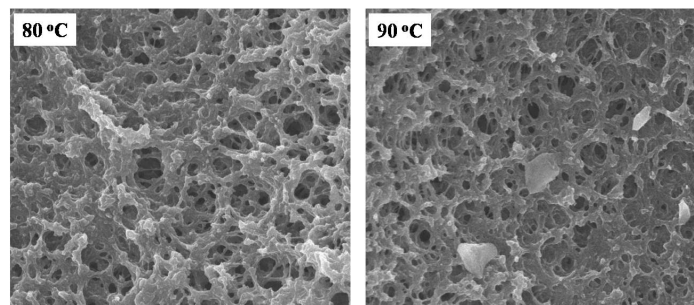
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425 **Fig. S11** The SEM images of the monolith POSS-DADD prepared at different temperatures.

426 Magnification: 3000×.

427 By comparing the monoliths prepared at different temperatures, it could be found that the pore
428 size decreased as the polymerization temperatures increased.

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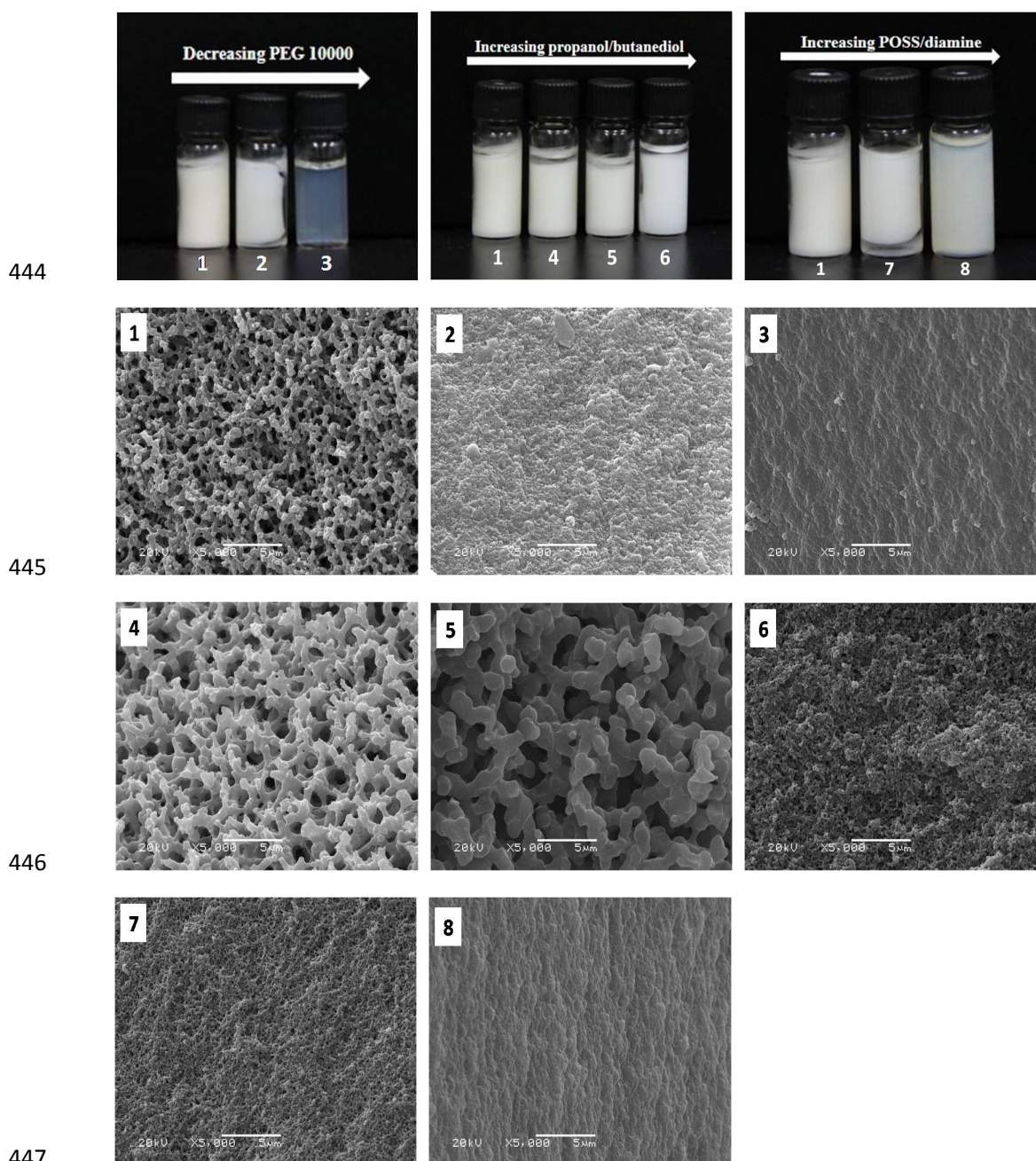
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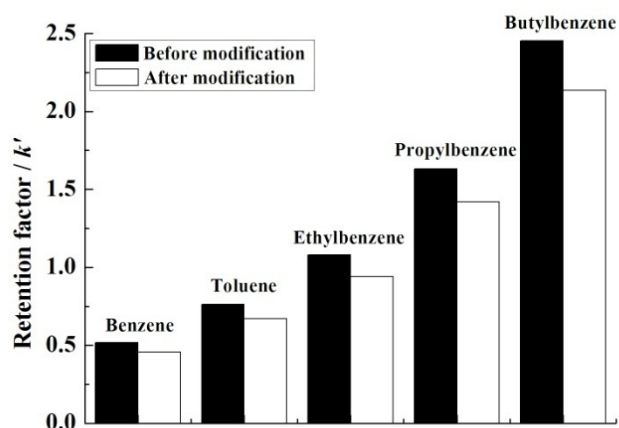


448 **Fig. S12** The bulk hybrid materials prepared with different feed recipes and their SEM images.

449 Magnification: 5000×.

450 As shown above, the pore size of the hybrid monoliths decreased with either a decrease of the
451 fraction of PEG 10,000 or an increase of the fraction of propanol. It is promising that hybrid
452 materials with different high surface areas and high ordered 3D skeleton could be achieved
453 conveniently by slightly adjusting the component of the porogenic solvents (see Table S2 for the
454 detail preparation information).

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457 **Fig. S13** The retention factors of alkylbenzenes on the monolith POSS-DADD before and after
458 $\text{NH}_3 \cdot \text{H}_2\text{O}$ modification. Experimental conditions: column dimension, 38 cm \times 75 μm i.d.; flow
459 rate, 50 $\mu\text{L}/\text{min}$ (before split); mobile phase, ACN/water (50/50, v/v); injection volume, 2.5 μL in
460 split mode; detection wavelength, 214 nm.

461 The decrease in retention indicated the existence of the residual epoxy groups, and demonstrated
462 the good tailorability of the prepared hybrid materials.

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477 **Supplementary Tables**

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479 **Table S1.** Comparison of the performance of the monolithic column and C18-particles packed

480 column in cLC-MS/MS.

| Column | Unique peptides | Sequence coverage / % |
|------------------------------------|-----------------|-----------------------|
| POSS-DADD monolith | 51 | 57.0 |
| C18-particles packed column | 44 | 59.3 |

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504 **Table S2.** Components of the prepolymerization solution of the bulk hybrid materials.

| Monolith | POSS (mg) | HDA (mg) | Propanol (μL) | 1,4-Butanediol (μL) | PEG10,000 (mg) | Temp ($^{\circ}$C) | BET Surface area (m^2/g) | Total pore volume (mL/g) | Average pore size (nm) |
|-----------------|----------------------|---------------------|---|---|---------------------------|--|--|-------------------------------------|-----------------------------------|
| 1 | 50 | 15 | 280 | 40 | 30 | 70 | 11.46 | 1.63 | 336 |
| 2 | 50 | 15 | 280 | 40 | 15 | 70 | 0.57 | / | / |
| 3 | 50 | 15 | 280 | 40 | 0 | 70 | 0.47 | / | / |
| 4 | 50 | 15 | 280 | 80 | 30 | 70 | 6.78 | 1.57 | 802 |
| 5 | 50 | 15 | 280 | 0 | 30 | 70 | 3.75 | 0.79 | 1109 |
| 6 | 50 | 15 | 560 | 0 | 30 | 70 | 46.53 | 0.64 | 64 |
| 7 | 50 | 5 | 280 | 40 | 30 | 70 | 79.09 | 0.25 | 47 |
| 8 | 100 | 5 | 280 | 40 | 30 | 70 | 43.82 | 0.07 | 94 |

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