Electronic Supplementary Information

In situ amplified electronic signal for determination of low-abundance proteins coupling with nanocatalyst-based redox cycling

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1 EXPERIMENTAL SECTION

Materials. Polyclonal anti-CEA antibody produced in rabbit (anti-CEA, 1.0 mg/mL) was 2 3 purchased from Beijing Biosynthesis Biotechnol. Co., Ltd. (China). CEA ELISA kits with 0, 5.0, 10, 20, 40 and 80 ng/mL CEA standards were obtained from Biocell Biotechnol. Co., Ltd. 4 (China). *N*-(3-Dimethylaminopropyl)-*N*'-ethylcarbodiimide 5 hydrochloride (EDC), N-Hydroxysuccinimide (NHS), and ferrocenecarboxylic acid were supplied by Sigma-Aldrich 6 (USA). Cerium nitrate hexahydrate, chitosan, NaBH₄, p-nitrophenol (PNP), p-aminophenol 7 (PAP), and bovine serum albumin (BSA, 96–99 wt %) were obtained from Sinopharm Chem. Re. 8 Co. Ltd. (Shanghai, China). All other reagents were of analytical grade and were used without 9 further purification. Ultrapure water obtained from a Millipore water purification system (≥ 18 10 11 M Ω , Milli-Q, Millipore) was used in all runs. 0.1 mol/L phosphate buffered saline (PBS) solutions with various pH values was prepared by mixing 0.1 mol/L K₂HPO₄ and 0.1 mol/L 12 KH₂PO₄, and 0.1 mol/L KCl was added as the supporting electrolyte. 13

Preparation and Bioconjugation of Platinum/Cerium Oxide Hybrid Nanostructures (CeO₂/Pt). Cerium oxide (CeO₂) nanoparticles were synthesized according to the literature with a little modification.^{S1} Briefly, cerium nitrate solution (5.82 mL, 0.05 mol/L) was initially dropped into ammonia (3.4 mL, 25 wt %) under vigorous stirring, and then further reacted at room temperature (RT) until the color of the suspension changed from purple to yellow. Following that, the suspension was centrifuged for 10 min at 10,000 rpm, and the obtained precipitate was re-dispersed into 3 mL of distilled water.

Next, the platinum/cerium oxide hybrid nanostructures were synthesized consulting to the literature.^{S2} 20 mg of poly(*N*-vinyl-2-pyrrolidone) (PVP) was added into the as-prepared solution, and stirred for 12 h at RT. The PVP-functionalized CeO₂ nanoparticles were centrifuged, and re-dispersed in 2.5 mL of distilled water. Following that, 1.0 mL of K₂PtCl₄ (21 mg/mL) was added in the suspension, and heated to 95 °C for 1 h under gentle stirring. Subsequently, 1.0 mL of NaBH₄ (4.0 mg/mL) was added into the mixture. The reaction was maintained for another 30

min at 95 °C. Finally, platinum/cerium oxide hybrid nanostructures (designated as CeO₂/Pt) were 27 collected by centrifugation, and dispersed in 2.0 mL of distilled water for the conjugation of 28 anti-CEA antibody as follows: 200 µL of anti-CEA (1.0 mg/mL) was initially added into 1.0 mL 29 of CeO₂/Pt colloids ($C_{[CeO2]} \approx 0.14$ mM), and then the mixture was incubated for 12 h at 4 °C with 30 gentle stirring. During this process, anti-CEA antibodies were covalently conjugated onto the 31 CeO₂/Pt nanostructures via the Pt-NH bonds. Finally, anti-CEA-functionalized CeO₂/Pt 32 (designated as CeO₂/Pt-anti-CEA) were obtained by centrifugation, and stored in 2.0 mL of pH 33 7.4 PBS at 4 °C until use. For comparison, pure Pt and CeO₂ nanoparticles were utilized for the 34 label of anti-CEA antibodies using the same method, respectively (Note: Pure platinum and CeO₂ 35 nanoparticles were prepared according to Ref. [S3] and Ref. [S4], respectively. The antibodies 36 37 were labeled onto the CeO_2 nanoparticles based on the encapsulation method by using chitosan 38 acidic suspension).

39 Preparation of Electrochemical Immunosensor. Initially, 1.0 mg of chitosan (CS) flakes was 40 added into 1.0 mL of 1.0 wt % acetic acid solution, and then the mixture was adequately stirred at 41 RT until chitosan flake were completely dissolved. Following that, 1.0 mg of ferrocenecarboxylic 42 acid was added into the chitosan solution with adequately stirring until a homogeneous mixture 43 was obtained (designated as Fc-CS).

44 Next, a glassy carbon electrode (GCE, 4 mm in diameter) was polished with 0.3 µm and 0.05 um alumina, followed by successive sonication in acetone, ethanol and deionized water for 5 min 45 46 and dried in air. Afterwards, 10 µL of the as-prepared Fc-CS suspension was cast on the surface of the cleaned GCE. After dry, 5 µL of 0.025% (w/v) of glutaraldehyde was added on the 47 electrode and incubated for 1 h at RT. Excess glutaraldehyde was removed by washing with pH 48 7.4 PBS. Subsequently, the modified electrode was immersed into the anti-CEA solution, and 49 incubated for 12 h at 4 °C. During this process, anti-CEA antibodies were conjugated onto the 50 GCE via the glutaraldehyde. Finally, anti-CEA-modified GCE was incubated with 2.5 wt % BSA 51 52 for 1.0 h at RT to block the unreacted and nonspecific sites. The as-prepared immunosensor was used for detection of CEA. 53

54 Electrochemical Measurement. All electrochemical measurements were carried out with a

CHI 620D Electrochemical Workstation (Shanghai, China) with a conventional three-electrode 55 system using a modified GCE as working electrode, a platinum wire as auxiliary electrode, and a 56 saturated calomel electrode (SCE) as reference electrode. Initially, 5 µL of CEA samples with 57 various concentrations was thrown onto the modified electrode and incubated for 30 min at RT. 58 After washing with pH 7.4 PBS, 10 µL of as-prepared CeO₂/Pt-anti-CEA suspension were 59 dropped onto the modified electrode, and incubated for another 30 min under the same conditions. 60 After rinsing thoroughly with pH 7.4 PBS, the electrochemical response of the immunosensor 61 were performed in pH 8.0 PBS containing 6 mM p-nitrophenol and 6 mM NaBH₄ by using 62 63 differential pulse voltammetry (DPV) from -100 to 600 mV (vs. SCE) with a pulse amplitude of 50 mV and a pulse width of 50 ms. All electrochemical measurements were done in an unstirred 64 65 electrochemical cell at room temperature (RT, 25 ± 1.0 °C). Analyses are always made in 66 triplicate. Scheme 1 represents the assay protocol of the electrochemical immunosensor.

Elucidation of Signal Amplification by Nanocatalyst-Based Redox Cycling. To verify the 67 advantages of the as-synthesized nanolabels in the redox cycling system, we also prepared 68 69 another two types of nanomaterials, i.e., Pt and CeO₂ nanoparticles, which were used as the 70 nanolabels for conjugation of anti-CEA antibodies. The comparative study was carried out by assaying 0.1 ng/mL CEA (as an example) on the same-batch immunosensors. The judgement was 71 based on the change in the current in the absence and presence of PNP and NaBH₄ after the 72 formation of sandwiched immunocomplexes. As seen from Fig. S2-A, upon addition the PNP and 73 NaBH₄ in pH 8.0 PBS, redox peak current of the CeO₂ based immunoassay was not almost 74 changed in comparison with the background current. The results revealed that CeO₂ could not 75 reduce the PNP to PAP. However, an obvious increase of the oxidation peak was obtained for the 76 Pt-based immunoassay (Fig. S2-B), indicating Pt nanoparticles possessed excellent catalytic 77 ability for the reduction of PNP. More inspiringly, the peak currents would be greatly improved 78 by using CeO_2/Pt as the nanolabels (Fig. S2-C). The reason might be the fact that the CeO_2 79 nanoparticles with a large surface area can load a lot of Pt nanoparticles. When one secondary 80 antibody was conjugated with one CEA antigen on the electrode, more Pt nanoparticles will 81

82 participate in the catalytic cycling amplification reaction.



Fig. S1 (a) Typical TEM image of platinum/cerium oxide hybrid nanostructures. Dynamic light scattering (DLS) data of platinum/cerium oxide hybrid nanostructures before (b) and after (c) conjugation with anti-CEA antibodies. Nanostructures were characterized using Hitachi H-7650 transmission electron microscope (TEM, Japan). DLS measurements were used to determine the hydrodynamic sizes of the nanostructures and the aggregates using a Malvern Zetasizer 3000 HS particle size analyzer (Malvern Instruments, UK).



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Fig. S2 Cyclic voltammograms of the immunosensor after reaction with 0.1 ng/mL CEA using various
nanolabels: (A) anti-CEA-CeO₂, (B) anti-CEA-PtNP, and (C) CeO₂/Pt-anti-CEA, in pH 8.0 PBS in the absence
(a) and presence (b) of 6.0 mM PNP + 6.0 mM NaBH₄.

| Method | Linear range (ng/mL) | LOD (ng/mL) | Ref. |
|---|----------------------|--------------------------|-----------|
| Nanocatalyst-based electrochemical immunoassay | 0.0005 - 20 | 0.0005 | This work |
| CdS/DNA nanochain-based electrochemical immunoassay | 0.1 - 100 | 0.0033 | [85] |
| Chemiluminescent multiplex immunoassay | 1.0-70 ng/mL | 0.65 ng/mL | [S6] |
| Multianalyte electrochemical immunoassay | \leq 188 ng/mL | 1.1 ng/mL | [S7] |
| Multiplexed electrical detection | - | $\leq 0.9 \text{ pg/mL}$ | [S8] |
| Time-resolved fluoroimmunoassay | - | ≤70 pg/mL | [89] |
| Immunofluorescence assay | 0.5-1000 pmol/L | 1.31 pmol/L | [S10] |
| Amperometric immunoassay | 0.01-160 ng/mL | 5.0 pg/mL | [S11] |
| Piezoelectric immunoassay | 1.5-30 μg/mL | 1.5 μg/mL | [S12] |
| Chemiluminescence enzyme immunoassay | 2-162 ng/mL | 0.69 ng/mL | [813] |
| Potentiometric immunoassay | 1.5-200 ng/mL | 0.5 ng/mL | [S14] |

Table S1 Comparison of analytical properties of various CEA immunosensors or immunoassays

| | Method; ^{<i>a</i>} Concentration [mean \pm SD (RSD, %), <i>n</i> = 3, ng mL ⁻¹] | | _ |
|-------------------------|--|---------------------|------------------|
| Sample no. ^b | Found by immunosensor | Found by the ELISA | t _{exp} |
| 1 | 8.4 ± 1.2 (7.6%) | 7.9 ± 0.8 (10.1%) | 0.48 |
| 2 | 15.3 ± 0.6 (3.9%) | 15.4 ± 2.5 (16.2%) | 0.67 |
| 3 | 2.6 ± 0.3 (11.5%) | 2.4 ± 0.4 (16.7%) | 0.69 |
| 4 | 5.7 ± 0.5 (8.8%) | 6.5 ± 0.7 (10.8%) | 1.61 |
| 5 | 0.8 ± 0.15 (18.8%) | 0.7 ± 0.1 (14.3%) | 0.96 |
| 6 | 18.5 ± 2.1 (11.4%) | 17.5± 1.3 (7.4%) | 0.70 |
| 7 | 6.1 ± 0.5 (8.2%) | 5.8 ± 0.4 (6.9%) | 0.81 |
| 8 | 4.1 ± 0.3 (7.3%) | 3.7 ± 0.3 (8.1%) | 1.63 |
| 9 | 17.5±2.0 (11.4%) | 17.7±1.9 (11.1%) | 0.25 |
| 10 | 10.9 ± 2.3 (17.8%) | 10.4± 1.6 (15.4%) | 1.54 |
| 11 | 9.3 ± 1.1 (11.8%) | 9.8 ± 2.7 (27.0%) | 0.29 |
| 12 | 13.9 ± 1.5 (10.8%) | 13.2 ± 1.8 (13.6%) | 0.52 |
| 13 | 19.3 ± 1.1 (5.7%) | 19.7 ± 0.9 (4.6%) | 0.49 |
| 14 | 10.2 ± 0.6 (5.9%) | 9.8 ± 0.5 (5.1%) | 0.89 |
| 15 | 4.9 ± 0.3 (6.1%) | 5.1 ± 0.2 (3.9%) | 0.96 |
| 16 | 0.51 ± 0.08 (15.6%) | 0.54 ± 0.09 (16.7%) | 0.43 |

 Table S2 Comparison of the assayed results for clinical serum specimens or standard samples with various

 dilution ratios using the developed protocol and the referenced ELISA method

^{*a*} The regression equation (linear) for these data is as follows: y = 0.99x - 0.0243 ($R^2 = 0.9942$) (x-axis: by magnetic electrochemical immunoassay; y-axis: by the ELISA).

^b Clinical serum samples (no.: 1-12) and CEA standard samples with various dilution ratios (no.: 13-16) were appropriately diluted if CEA levels were over the calibration ranges.

Notes and references

- S1. S. Ghosh, D. Divya, K. Remani and T. Sreeremya, J. Nanoparticle Res., 2010, 12, 1905.
- S2. T. Yu, J. Zeng, B. Lim and Y. Xia, Adv. Mater., 2010, 22, 5188.
- S3. R. Polsky, R. Gill, I. Kaganovasky and I. Willner, Anal. Chem., 2006, 78, 2268.
- S4. Y. Wu, C. Chen and S. Liu, Anal. Chem., 2009, 81, 1600.
- S5. F. Kong, B. Xu, J. Xu and H. Chen, Biosens. Bioelectron., 2013, 39, 177.
- S6. Z. Fu, Z. Yang, J. Tang, Y. Liu, F. Yan and H. Ju, Anal. Chem., 2007, 79, 7376.
- S7. J. Wu, F. Yan, J. Tang, C. Zhai and H. Ju, Clin. Chem., 2007, 53, 1495.
- S8. G. Zheng, F. Patolsky, Y. Cui, W. Wang and C. Lieber, Nat. Biotechnol., 2005, 23, 1294.
- S9. J. Yuan, G. Wang, K. Majima and K. Matsumoto, Anal. Chem., 2001, 73, 1869.
- S10. N. Honda, U. Lindberg, P. Andersson, S. Hoffmann and H. Takei, Clin. Chem., 2005, 51, 1955.
- S11. D. Tang, R. Yuan and Y. Chai, Anal. Chem., 2008, 80, 1582.
- S12. B. Zhang, X. Zhang, H. Yan, S. Xu, D. Tang and W. Fu, Biosens. Bioelectron., 2007, 23, 19.
- S13. W. Dungchai, W. Siangproh, J. Lin, O. Chailapakul, S. Lin and X. Ying, Anal. Bioanal. Chem., 2007, 387, 1965.
- S14. D. Tang, R. Yuan and Y. Chai, J. Phys. Chem. B, 2006, 110, 11640.