

Coordinative trapping of the boron β -diketiminato system [B(NMesCMe)₂CH] via metal-templated synthesis

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1. General methods and instrumentation

All manipulations were carried out using standard Schlenk line or dry-box techniques under an atmosphere of argon or dinitrogen. Solvents were degassed by sparging with dinitrogen and dried by passing through a column of the appropriate drying agent. Fluorobenzene was dried by refluxing over calcium hydride, distilled, sparged and stored over activated molecular sieve. NMR spectra were recorded in C₆D₆ or CD₂Cl₂ which was dried over potassium (C₆D₆) or molecular sieve (CD₂Cl₂), distilled under reduced pressure and stored under dinitrogen in Teflon valve ampoules. NMR samples were prepared under dinitrogen in 5 mm Wilmad 507-PP tubes fitted with J. Young Teflon valves. ¹H, ¹³C, ¹¹B and ¹⁹F NMR spectra were recorded on a Varian Mercury-VX-300 spectrometer and referenced internally to residual protio-solvent (¹H) or solvent (¹³C) resonances and are reported relative to tetramethylsilane (δ = 0 ppm). ¹¹B and ¹⁹F NMR spectra were referenced to Et₂O·BF₃ and CFCl₃, respectively. Chemical shifts are quoted in δ (ppm) and coupling constants in Hz. Infrared spectra were measured on a Nicolet 500 FT-IR spectrometer. Mass spectra of compound **4** were recorded on a Bruker Microtof mass spectrometer; all other mass spectra were measured by the EPSRC National Mass Spectrometry Service Centre, Swansea University. Elemental analyses were carried out by Stephen Boyer at London Metropolitan University.

Starting materials **1**,^{s1} MesN(H)C(Me)CHC(Me)NMes,^{s2} Li[(NMesCMe)₂CH],^{s3} Li[(NⁱPr)₂CPh],^{s4} and Na[BAr^f₄] (Ar^f = C₆H₃(CF₃)₂-3,5)^{s5} were prepared by literature procedures.

2. Syntheses of new compounds

Cp*Fe(CO)₂B(Cl){(N*i*Pr)₂CPh}, 2: A solution **1** prepared in situ from Na[Cp*Fe(CO)₂] (0.27 g, 1.00 mmol) and BCl₃ (1.0 mL of a 1.0 M solution in heptane, 1.00 mmol) in toluene (10 mL) was transferred onto a suspension of Li[(N*i*Pr)₂CPh] (1.0 equiv.) also in toluene (10 mL) at -78 °C. After warming to -30 °C and removal of the solvent in vacuo, the resulting brown residue was extracted into hexanes (30 mL). The dark red solution was concentrated to a volume of about 10 mL and stored at -30 °C, affording pale yellow crystals of **2** (isolated yield 0.13 g, 26%). ¹H NMR (300 MHz, C₆D₆, 298 K): δ_H 1.17 (d, ³J_{HH}= 6 Hz, 6H, *i*Pr CH₃), 1.58 (d, ³J_{HH}= 6 Hz, 6H, *i*Pr CH₃), 1.83 (s, 15H, Cp* CH₃), 3.89 (sept, ³J_{HH}= 6 Hz, 2H, *i*Pr CH), 6.95-7.36 (m, Ph H, 5H). ¹³C NMR (75 MHz, C₆D₆, 298 K): δ_C 10.4 (Cp* CH₃), 22.5 (*i*Pr CH₃), 24.0 (*i*Pr CH₃), 46.3 (*i*Pr CH), 95.2 (Cp*), 127.8, 128.2, 128.9, 130.2 (Ph C), 163.0 (NCN), 221.8 (CO). ¹¹B NMR (96 MHz, C₆D₆, 298 K): δ_B 21. IR (CH₂Cl₂, ν_{CO}/cm⁻¹): 1962, 1900. HR-MS (EI): *m/z*: 466.1843, calcd. for (C₂₄H₃₄BClFeN₂O)⁺= 466.1846 [(M-CO)⁺]

Cp*Fe(CO)₂B(Cl)[κ¹-{(NMesCMe)₂CH}], 3: A solution of **1** prepared in situ from Na[Cp*Fe(CO)₂] (0.27 g, 1.00 mmol) and BCl₃ (1.0 mL of a 1.0 M solution in heptane, 1.00 mmol) in toluene (10 mL) was transferred onto a solution of Li[(NMesCMe)₂CH] (1.0 equiv.) also in toluene (10 mL) at -78 °C, and the reaction mixture warmed to room temperature. The resulting mixture was slowly warmed to room temperature and stirred for 12 h. Removal of volatiles in vacuo gave a brown residue, which was extracted with pentane (50 mL), concentrated (to ca. 20 mL) and cooled to -30 °C, affording colourless crystals of **3** (isolated yield 0.15 g, 24%). ¹H NMR (300 MHz, C₆D₆, 298 K): δ_H 1.67 (s, 15H, Cp* CH₃), 1.69 (s, 3H, Mes *p*-CH₃), 2.09 (s, 6H, Mes *o*-CH₃), 2.12 (s, 3H, CCH₃), 2.24 (s, 3H, CCH₃), 2.34 (s, 3H, Mes *p*-CH₃), 2.44 (s, 6H, Mes *o*-CH₃), 6.19 (s, 1H, CCHC), 6.82 (s, 2H, Ar-H), 6.87 (s, 2H, Ar-H). ¹³C NMR (75 MHz, C₆D₆, 298 K): δ_C 10.2 (Cp* CH₃), 18.5 (CH₃), 19.6 (CH₃), 19.8 (CH₃), 20.8 (CH₃), 21.0 (CH₃), 21.6 (CH₃), 96.6 (Cp*), 125.3 (CCHC), 129.1, 129.9, 130.1, 131.5, 135.8, 136.1, 143.1, 147.8 (ArC), 151.4 (CCN), 164.4 (CC=N), 216.9 (CO). ¹¹B NMR (96 MHz, C₆D₆, 298 K): δ_B 73 (br). IR (CH₂Cl₂, ν_{CO}/cm⁻¹): 1986, 1930. HR-MS (EI): *m/z*: 596.2626, calcd. for (C₃₄H₄₄BClFeN₂O)⁺= 596.2627 [(M-CO)⁺]

3. Crystallographic details

2: $C_{25}H_{34}N_2BClFeO_2$, $M_r = 496.67$, monoclinic, $P2_1/n$, $a = 12.3060(1)$, $b = 15.6594(2)$, $c = 13.3532(2)$ Å, $\beta = 92.941(1)^\circ$, $V = 2569.8(1)$ Å³, $Z = 4$, $\rho_c = 1.284$ Mg m⁻³, $T = 150(2)$ K, $\lambda = 0.71073$ Å. 5857 independent reflections [$R(int) = 0.027$], used in all calculations. $R_1 = 0.0358$, $wR_2 = 0.0810$ for $F^2 > 2\sigma(F^2)$, and $R_1 = 0.0577$, $wR_2 = 0.1039$ for all unique reflections. Max./min. residual electron densities 0.56 and -0.48 e Å⁻³. CSD ref.: 908735.

3: $C_{35}H_{44}N_2BClFeO_2$, $M_r = 626.86$, monoclinic, $P2_1/n$, $a = 13.9947(2)$, $b = 16.4994(2)$, $c = 14.7643(2)$ Å, $\beta = 95.885(1)^\circ$, $V = 3391.2(1)$ Å³, $Z = 4$, $\rho_c = 1.228$ Mg m⁻³, $T = 150(2)$ K, $\lambda = 0.71073$ Å. 7705 independent reflections [$R(int) = 0.044$], used in all calculations. $R_1 = 0.0470$, $wR_2 = 0.0955$ for $F^2 > 2\sigma(F^2)$, and $R_1 = 0.0860$, $wR_2 = 0.1316$ for all unique reflections. Max./min. residual electron densities 0.63 and -0.66 e Å⁻³. CSD ref.: 908736.

4. Details of DFT calculations

The DFT calculations were performed using the Amsterdam Density Functional (ADF) Package Software 2012.^{s6} Calculations were performed using the Vosko-Wilk-Nusair local density approximation with exchange from Becke^{s7} and correlation corrections from Perdew^{s8} (BP). Slater-type orbitals (STOs)^{s9} were used for the triple zeta basis set with an additional set of polarization functions (TZP). The large frozen core basis set approximation was applied with no molecular symmetry. The general numerical integration was 6. Frequency calculations were performed for the cationic metal complexes and no significant imaginary frequencies were observed. Estimates of binding energies were obtained following the strategy outlined by Baerends^{s10} using the counterpoise method.^{s11} Calculations of ¹¹B NMR chemical shifts were performed using the NMR program contained in the ADF Package.^{s12} Chemical shifts are referenced to Et₂O·BF₃ (δ = 0 ppm) as the experimental standard. For optimized coordinate of the calculated complexes, see the frequency calculation run files (below).

Table s1: Binding Energy Summary { $\Delta E_{(\text{Binding})} = \Delta E_{(\text{Complex})} - \Delta E_{(\text{Free Fragments})}$ }^a

	From Closed Shell Fragments (kcal mol ⁻¹)	From Open Shell Fragments (kcal mol ⁻¹)
[Cp*Fe(CO) ₂ {B(N ⁱ Pr) ₂ CPh}] ⁺	-103.65	-84.57
[Cp*Fe(CO) ₂ {B(NMesCMe) ₂ CH}] ⁺ (4)	-82.76	-44.22
[CpFe(CO) ₂ {IMes}] ⁺	-81.05	-73.89

^a For a given complex, a more negative binding energy reflects less stable fragments

Table s2: Calculated spectroscopic properties

	Calculated ¹¹ B NMR Shift (ppm)	Calculated CO Stretching Frequencies (cm ⁻¹)
[Cp*Fe(CO) ₂ {B(N ⁱ Pr) ₂ CPh}] ⁺	91.3	1981, 1940
[Cp*Fe(CO) ₂ {B(NMesCMe) ₂ CH}] ⁺ (4)	66.3	1959, 1912

Run Files

```
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FULLSCF
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2 C 1.743267875000 14.829888240000 6.160393845000
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7 H -0.083961876780 17.130674200000 4.531701510000
8 H 2.426036151000 16.877385520000 5.529022463000
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10 H 0.251032030900 13.154591900000 6.335518228000
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12 O 1.498861513000 16.407691190000 1.703924756000
13 C 0.274647150300 13.510401060000 3.087733306000
14 O -0.499766795600 12.909911570000 2.471603925000
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17 H 3.811720288000 11.388075090000 6.376110023000
18 N 4.274841488000 14.117112880000 3.547607168000
19 C 4.565888205000 11.931762160000 3.672327134000
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21 C 5.223503649000 13.100883590000 3.475767047000
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23 H 4.917214084000 10.907157630000 3.693062831000
24 H 6.269941574000 13.311667650000 3.291376126000
25 C 4.658164498000 15.501226550000 3.328013078000
26 C 4.707940813000 15.978875090000 2.001435636000
27 C 5.129037882000 17.296302850000 1.799655793000
28 C 5.521624817000 18.126718020000 2.859523340000
29 C 5.505572280000 17.594790020000 4.152480043000
30 C 5.096263373000 16.278507010000 4.414080157000
31 H 5.258663194000 14.498069900000 0.529564032300
32 H 5.161787169000 17.681601290000 0.777681810700
33 H 6.762887463000 19.575206190000 1.842370844000
34 H 5.845073804000 18.212338350000 4.987913881000
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36 C 1.668015143000 10.591177560000 2.921929643000
37 C 0.778990080800 9.526784122000 3.127487244000
38 C 0.487152407900 9.034350857000 4.402718565000
39 C 1.129004579000 9.626069379000 5.501215379000
40 C 2.019518409000 10.694760390000 5.357078195000
41 H 1.200970505000 10.763567480000 0.824787791800
42 H 0.313828898600 9.061381646000 2.255584996000
43 H -0.024768045820 7.098552984000 5.220760326000
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46 H 2.333082239000 12.206183050000 6.892138455000
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50 H 2.917415110000 10.490590670000 1.164327127000
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15 9 2 1.0 | 64 58 57 1.0
16 10 1 1.0 | 65 59 57 1.0
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20 2 3 1.0 | 69 60 28 1.0
21 20 2 1.0 | END
22 3 4 1.0 |
23 13 14 2.0 | CHARGE 1.0
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27 8 3 1.0 | core Large
28 20 3 1.0 | createoutput None
29 21 18 1.0 | END
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32 27 28 1.5 | END
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36 31 57 1.0 | AnalyticalFreq
37 32 27 1.0 | END
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40 25 18 1.0 | INTEGRATION 6.0
41 35 36 1.5 | NOPRINT LOGFILE
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48 42 37 1.0 |

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