

## Supplemental Information

### **Controlled Branching of Polyglycidol and Formation of Protein-Glycidol Bioconjugates via a Grafting-from Approach with “PEG-like” Arms**

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**Materials.** Glycidol (Sigma Aldrich, 96%) was vacuum distilled prior to use. 3-Buten-1-ol (Sigma Aldrich Inc.), Tin(II) trifluoromethanesulfonate (Strem Chemicals Inc., 99%), N,N-Dimethylformamide (DMF, Sigma Aldrich 99.8%), N,N,N',N'-ethylenediaminetetraacetic acid (EDTA, Sigma Aldrich, 99.4-100.6%), triethylamine (Sigma Aldrich, ≥99%), bovine serum albumin (BSA, Aldrich, Approx. 99%), 5,5'- dithio-bis-(2-nitrobenzenzoic acid) (Ellman's reagent, Aldrich, 99%), tris(2-carboxyethyl)phosphine hydrochloride (Sigma Aldrich, TCEP, ≥98 %), and 4-nitrophenyl acetate (Fluka, ≥99%) were used as received. Dialysis membranes (Spectra/Por® 7, molecular weight cut-off (MWCO): 25,000 and 50,000 Da) were obtained from Spectrum Laboratories, Inc. Precise Tris-glycine 12% pre-cast PAGE gels were obtained from Fisher Scientific Company.

**Synthesis of GLY Homopolymer at a Range of Temperatures.** In order to maintain a consistent temperature throughout the entirety of the polymerization, a specialized round bottom flask and cooling system were employed. The round bottom flask was formed with a 25mL reaction vessel enveloped by a glass "jacket" with an inlet at the bottom and an outlet at the top. This allowed for a temperature controlled liquid to constantly surround the reaction vessel, thus allowing for sustained reaction temperature. The pump system used was a 7 Liter Refrigerated Circulator obtained from PolyScience with a temperature range of -40°C to 200°C and a temperature stability of ±0.01°C.

Two 1-dram vials were fitted with septa and flame dried under N<sub>2</sub>(g). The first 1-dram vial was used to make a 1.7M Iso-Amyl alcohol (IAOH) stock solution was formed using dry THF, while the second was used to create a 3.7x10<sup>-2</sup>M tin triflate stock solution, also using dry THF. Stock IAOH (0.20mL, 3.33x10<sup>-4</sup> mol, 0.066eq) and Sn(OTf)<sub>2</sub> (0.26mL, 9.45x10<sup>-6</sup> mol, 0.00035 eq) were added to the jacketed reaction flask, equipped with a magnetic stir bar, at the chosen reaction temperature, and the reaction solution was allowed to stir for 30 minutes to allow for the alcohol/catalyst conjugation. After the activation step, the glycidol monomer (2.00g, 27mmol, 1.0eq) was added drop wise. After stirring was completely impeded (Reaction time varied with temperature), the crude viscous polymer product was dissolved in a minimal amount of methanol and precipitated into vigorously stirring

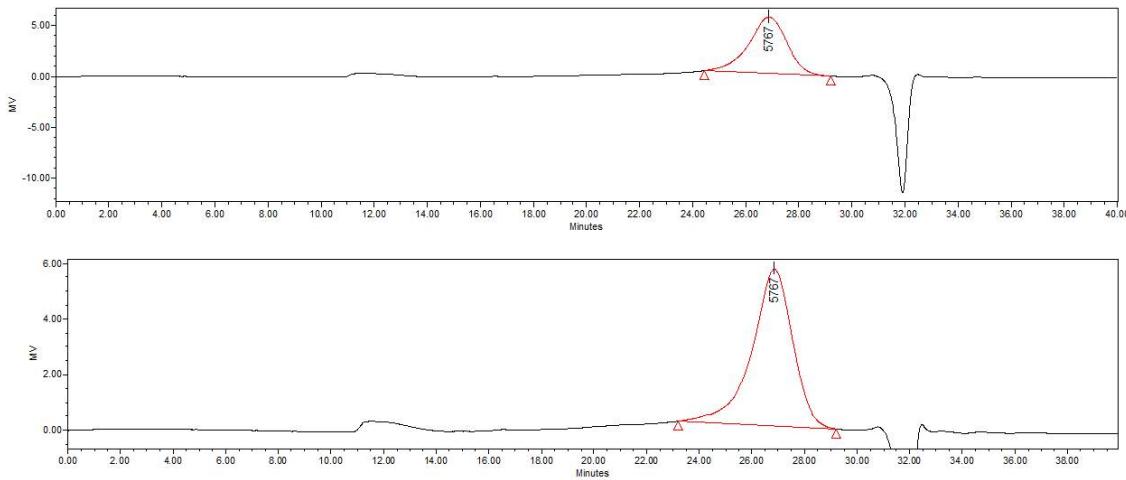
hexanes, which were then decanted to afford the pure glycidol homopolymer product as translucent viscous material.  $^1\text{H-NMR}$  (600MHz,  $\text{CDCl}_3$ )  $\delta$ : 3.31-3.94 (6H).  $^{13}\text{C-NMR}$  (150MHz,  $\text{CDCl}_3$ )  $\delta$ : 81.37, 79.81, 75.12, 73.88, 72.01-72.94, 70.42-71.17, 64.41, 62.53, 62.06.

CAUTION: When the scale of the reaction is above 1g, the reaction can become very exothermic during monomer addition and a cooling of the reaction vessel with an ice bath is recommended to avoid overheating.

Polymer	Temperature	Reaction Time	Yield
Glycidol Homopolymer	40°C	1h	92.3%
Glycidol Homopolymer	20°C	2.5h	90.7%
Glycidol Homopolymer	0°C	42h	93.5%
Glycidol Homopolymer	-20°C	4d	94.8%

#### GPC – Data:

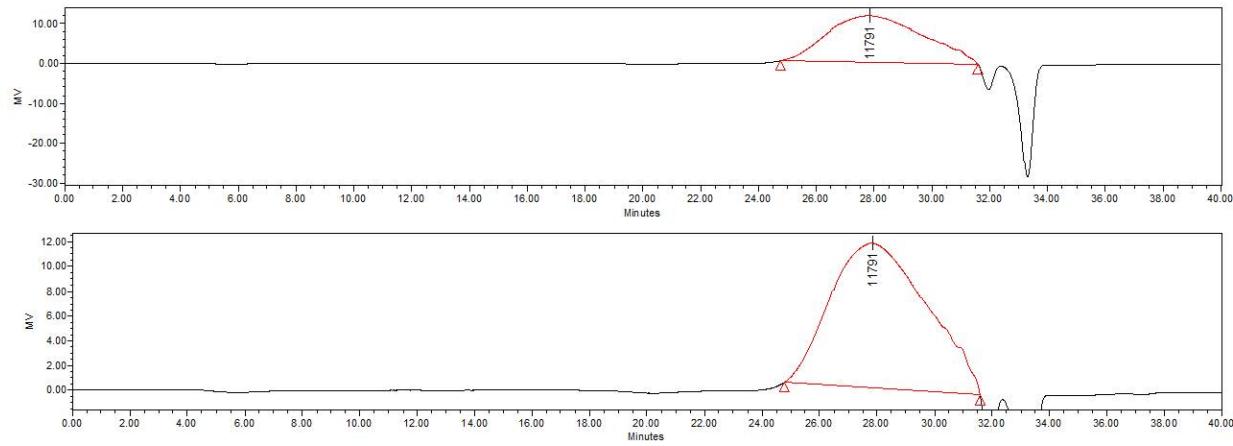
##### -40°C: Homoglycidol polymer



	Name	Retention Time	Peak	Dist	Name	Mn	Mw	Mz	Mz+1	Polydispersity	R/P	K	alpha	Mz/Mw	Id	Peak Sigma	Systm Sigma	Aea	% Aea	Height	Int Type	MW <sub>n</sub> × 10 <sup>3</sup> pcp	MW <sub>w</sub> × 10 <sup>3</sup> pcp	Mn <sub>above</sub> 90pcp st	Mn <sub>below</sub> 90pcp st	Mw <sub>above</sub> 90pcp st	Mw <sub>below</sub> 90pcp st	Mn <sub>above</sub> 10pcp D st	Mn <sub>below</sub> 10pcp D st
1	Broad	26.83	5762	784	5761	998	1479	126807						1.3720	2.0435	0	60591	10.00	5624.36		1813	3712	3079	16291	3151	1792	3113	19895	

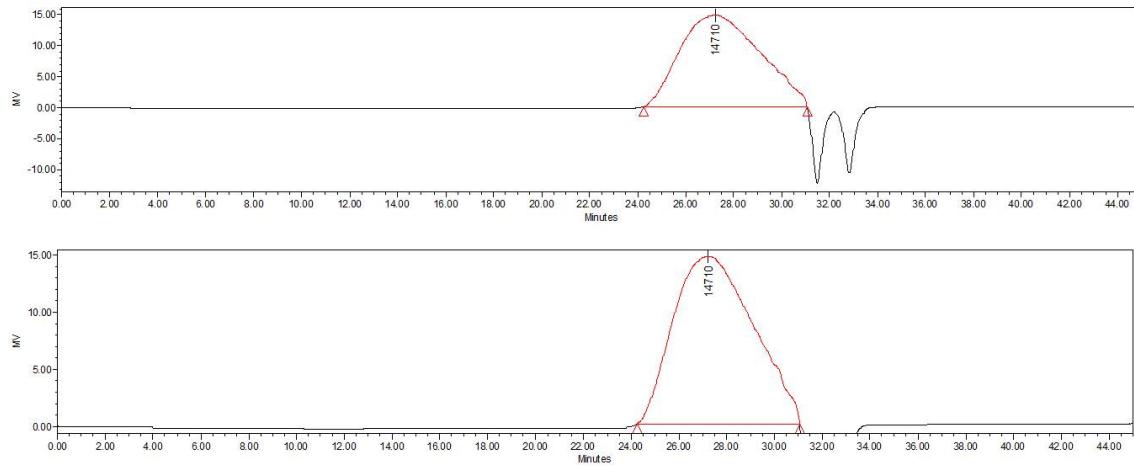
Mn: 5762; Mw: 7242, PDI 1.25

### 0°C: Homoglycidol polymer



Mn: 8566, Mw: 11645, PDI: 1.35

### 20°C: Homoglycidol polymer



	Name	Retention Time	Peak Codes	Dist Name	Mn	Mw	Mf	Mr	Mz	Mz+1	Polydispersity	[η] <sub>P</sub>	K	alpha	Mz/Mw	Mz+1/Mw	Id	Peak Sigma	System Sigma	Aea	% Aea	Int Type	MW_d1_10 <sub>pd1</sub>	MW_d1_50 <sub>pd1</sub>	Mn_above_90pc_d1 <sub>pd1</sub>	Mn_below_90pc_d1 <sub>pd1</sub>	Mn_above_90pc_d1 <sub>pd2</sub>	Mn_below_90pc_d1 <sub>pd2</sub>	Mw_above_90pc_d1 <sub>pd1</sub>	Mw_below_90pc_d1 <sub>pd1</sub>	Mw_above_90pc_d1 <sub>pd2</sub>	Mw_below_90pc_d1 <sub>pd2</sub>	Mw_above_90pc_d2 <sub>pd1</sub>	Mw_below_90pc_d2 <sub>pd1</sub>	Mw_above_90pc_d2 <sub>pd2</sub>	Mw_below_90pc_d2 <sub>pd2</sub>
1	Bead	27.24	C08		10976	14589	14710		18245	21472	1329218				1.250540	1.471737	0	3421576		100.00		14637	Bs			5947	4818	28869	4755	29354	4882	29777				

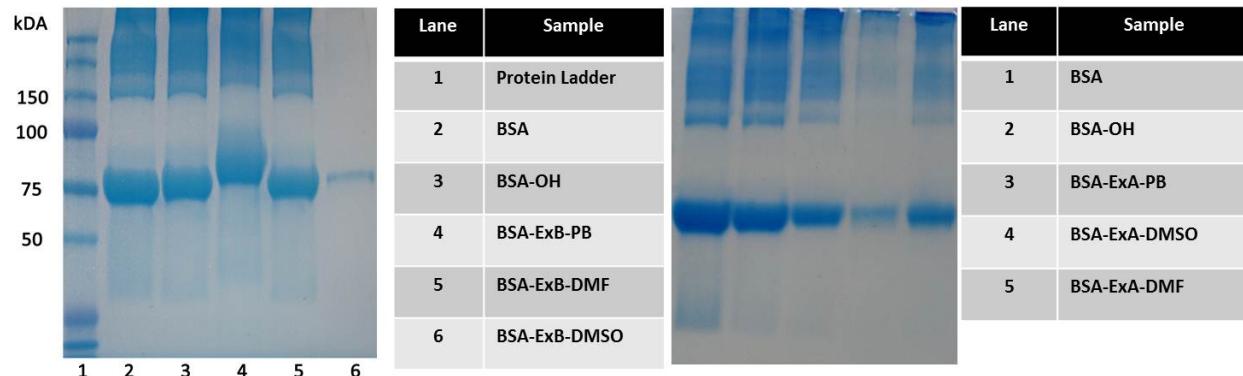
Mn: 10976, Mw: 14589, PDI: 1.32

**Conjugation of BSA with Maleimide Alcohol to Obtain BSA Alcohol Conjugate.** BSA (0.5g, 7.5  $\mu$ mol, 1 eq) was dissolved in phosphate buffer (PB) solution (37.5mL, pH 7.2) in a 100mL round bottom flask equipped with a magnetic stir bar and was then purged under N<sub>2</sub>(g) for 1h. 1-(2-hydroxyethyl)-1H-pyrrole-2,5-dione (21.23mg, 6x10<sup>-5</sup> mol, 20 eq) was solubilized in nitrogen-purged DMF (2.5mL). The maleimide solution was then added drop wise to the stirring BSA solution and the reaction was allowed to run for 20 @ 25°C. The resulting reaction mixture was dialyzed against deionized (DI) water for 36h using a MWCO of 25,000 Da, and then lyophilized to isolate the BSA Alcohol Conjugate (BSA-OH).

**General Procedure for the Polymerization of GLY using BSA-OH in Different Solvent Systems.** Sn(OTf)<sub>2</sub> (0.87mg, 2.087 $\mu$ mol), BSA-OH (80mg, 1.198 $\mu$ mol), and PB pH 6.0 (3.6mL) were added to a 25mL round bottom flask equipped with a magnetic stir bar and purged with N<sub>2</sub>(g) for 30 minutes in an ice bath. The glycidol monomer (1.194mmol, 79.4 $\mu$ L) was then added by syringe and the reaction was allowed to run for 24h at 25°C. The resulting reaction mixture was dialyzed against DI water for 36h using a MWCO of 25,000 Da, and then lyophilized to isolate the BSA Glycidol Conjugate (BSA-GLY).

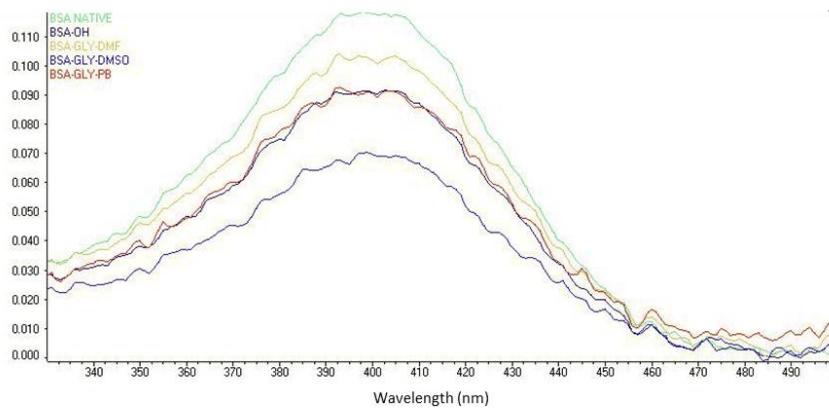
This reaction was repeated with the PB replaced with either DMF or DMSO in order to ascertain the optimal reaction conditions. The products obtained from these three solvent systems were visibly different with a seeming decrease in solubility in the trend: PB>DMF>DMSO. However, apart from this caveat, the reaction procedure remained unchanged.

**Sodium Dodecyl Sulfate-Polyacrylamide Gel Electrophoresis (PAGE).** PAGE was accomplished using an Invitrogen Novex Mini-Cell system. Precast 12% Precise™ Protein Gels (Thermo Scientific) were run at 100V and 120mA for 1.75 hours. Solutions were prepared with a BSA concentration of 6g/L in DI water. PAGE samples were formed by mixing 7.5 $\mu$ L of BSA solution with 2.5 $\mu$ L 4x LDS sample buffer (Invitrogen) before loading the 10 $\mu$ L solutions onto the gel plate. Staining of the gels was accomplished using coomassie blue.

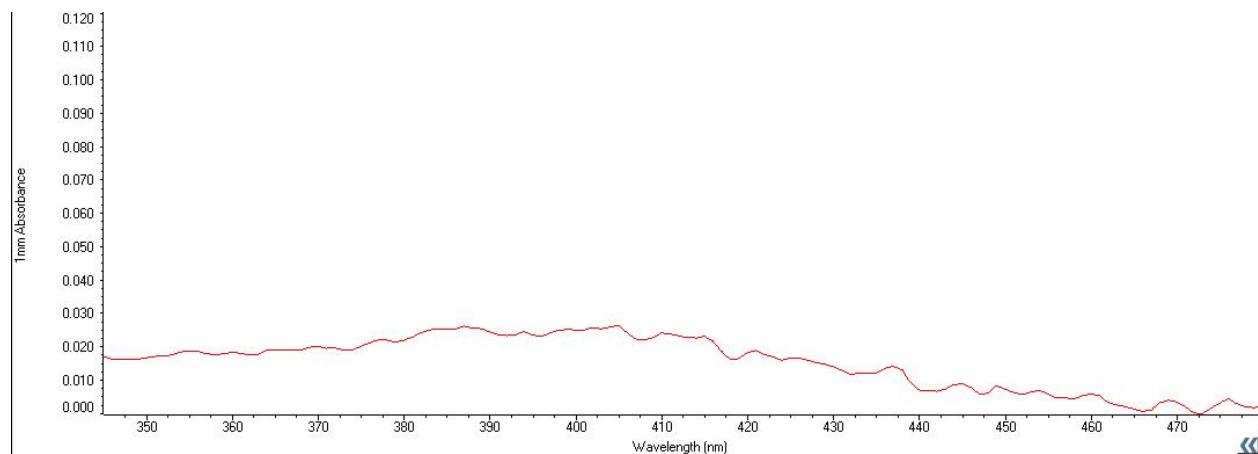


**Figure 1.** (left), Experiment B (ExB) higher concentration of glycidol monomer 1.33M (4 fold): PAGE of BSA and the synthesized BSA-polymer conjugates. Lane1: Protein Ladder. Lane 2: BSA. Lane 3: BSA-OH initiator. Lane 4: BSA-polymer reaction in PB. Lane 5: BSA-polymer reaction in DMF. Lane 6: BSA-polymer in DMSO. (right), Experiment A (ExA) lower concentration of glycidol monomer of 0.33M. Lane1: Native BSA protein. Lane 2: BSA-OH. Lane 3: BSA-polymer reaction in PB. Lane 5: BSA-polymer reaction in DMSO. Lane 6: BSA-polymer in DMF.

**Bioactivity assay.** The bioactivity of the synthesized BSA-OH conjugates and subsequent polymer coated species was determined and compared to the native BSA protein by measuring the absorbance related to the hydrolysis product with 4-nitrophenyl acetate as previously published by Sumerlin.<sup>1</sup> Sample preparation was performed by combining a BSA or BSA conjugate solution (50µL, [BSA]=0.27mM) in PB (pH=8.0) with a 10mM solution of 4-nitrophenyl acetate dissolved in acetonitrile (10µL) and PB (0.94mL, pH=8.0) and centrifuging the samples for 5 minutes at 6,000 rpm. The samples were allowed to incubate for 30 minutes at room temperature before measuring the absorbance at  $\lambda= 405\text{nm}$  to determine the bioactivity of each sample.



Sample	Absorbance
BSA	0.117
BSA-OH	0.93
BSA-Gly-DMF	0.103
BSA-Gly-DMSO	0.069
BSA-Gly-PB	0.091



**Figure 2a.** Bioactivity assay performed using UV-vis spectroscopy. Data presented includes BSA-polyglycidol conjugate run in PB at lower (top) and 4-fold higher glycidol concentration (bottom). Sample in the bottom trace represented the PB only with 10 K molecular weight increase (MALDI) since the molecular weight increase is not as significant in the other samples run in organic solvents.

**Gel Permeation Chromatography (GPC).** Size exclusion chromatography of the glycidol homopolymers was performed in DMF at 70°C with a flow rate of 1.0mL/min (Waters 1525 binary HPLC pump; columns: 7.8x300mm; Styragel HR 5 DMF, Styragel HR 4E, and Styragel HR 3: molecular weight range 50,000 to 4x10<sup>6</sup>, 50 to 100,000, and 500 to 30,000 g/mol,

respectively). Detection was accomplished using a Waters 2414 refractive index detector. Molecular weights were determined using polystyrene standards.

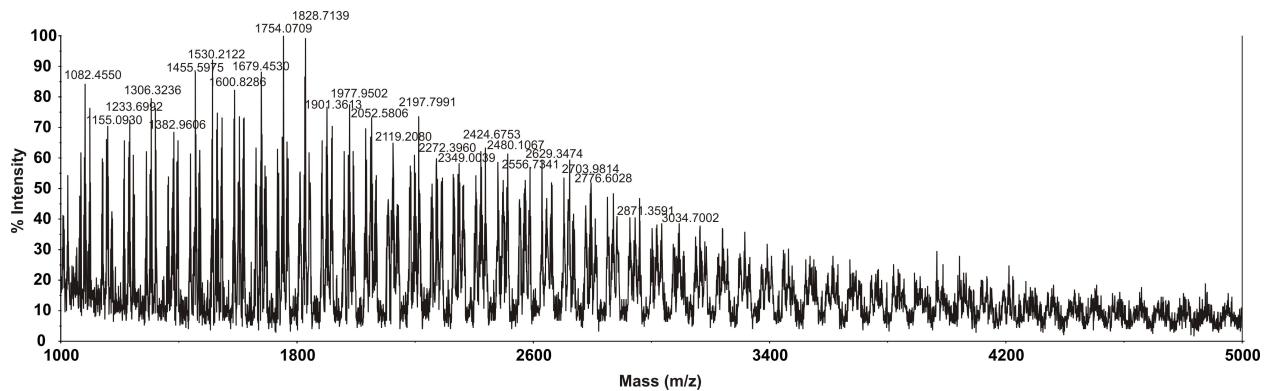
**UV-vis Spectroscopy.** UV-vis spectra were obtained using a Thermo Scientific Nanodrop 2000

**Matrix-Assisted Laser Desorption/Ionization Time-of-Flight Mass Spectroscopy (MALDI-TOF).** MALDI-TOF MS of glycidol homopolymers was performed using product samples of 3mg/mL and a  $\alpha$ -cyano-4-hydroxycinnamic acid matrix solution of 20mg/mL, both in methanol. To confirm the addition of the maleimide alcohol initiator to the BSA protein and the subsequent glycidol polymer growth from the BSA-OH conjugate similar parameters were used. The MALDI samples were formed using 6mg/mL product samples in DI water and a 5mg/mL matrix solution of sinapic acid dissolved in 2 parts 0.1% formic acid, 1 part acetonitrile.

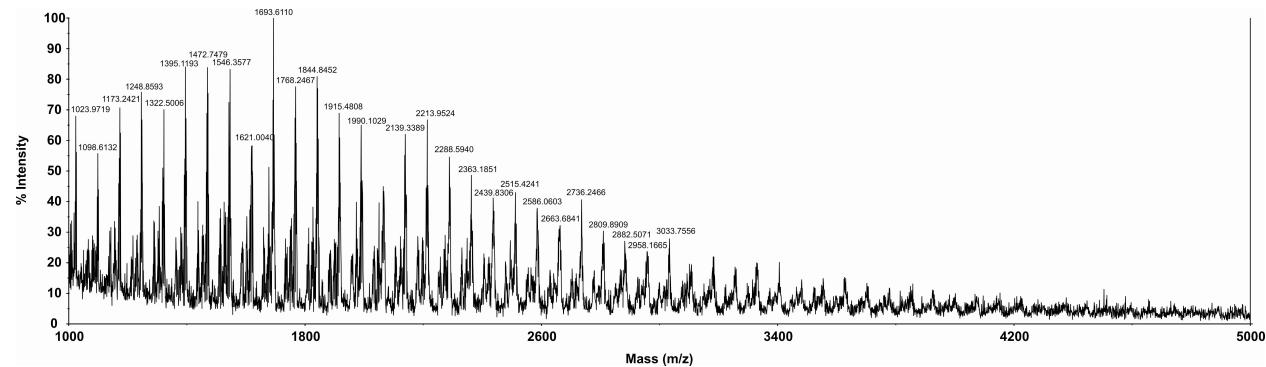
The sample and matrix were mixed in a 1:1 ratio and 1.4 $\mu$ L was spotted onto the MALDI plate for characterization. The instrument used was a Voyager DE-STR MS in linear TOF mode equipped with a nitrogen gas laser at  $\lambda = 337\text{nm}$  with external calibration. The instrument parameters were set to: 25,000V; 90% grid; 600ns delay; 1000 shots per spectrum.

Homopolymers:

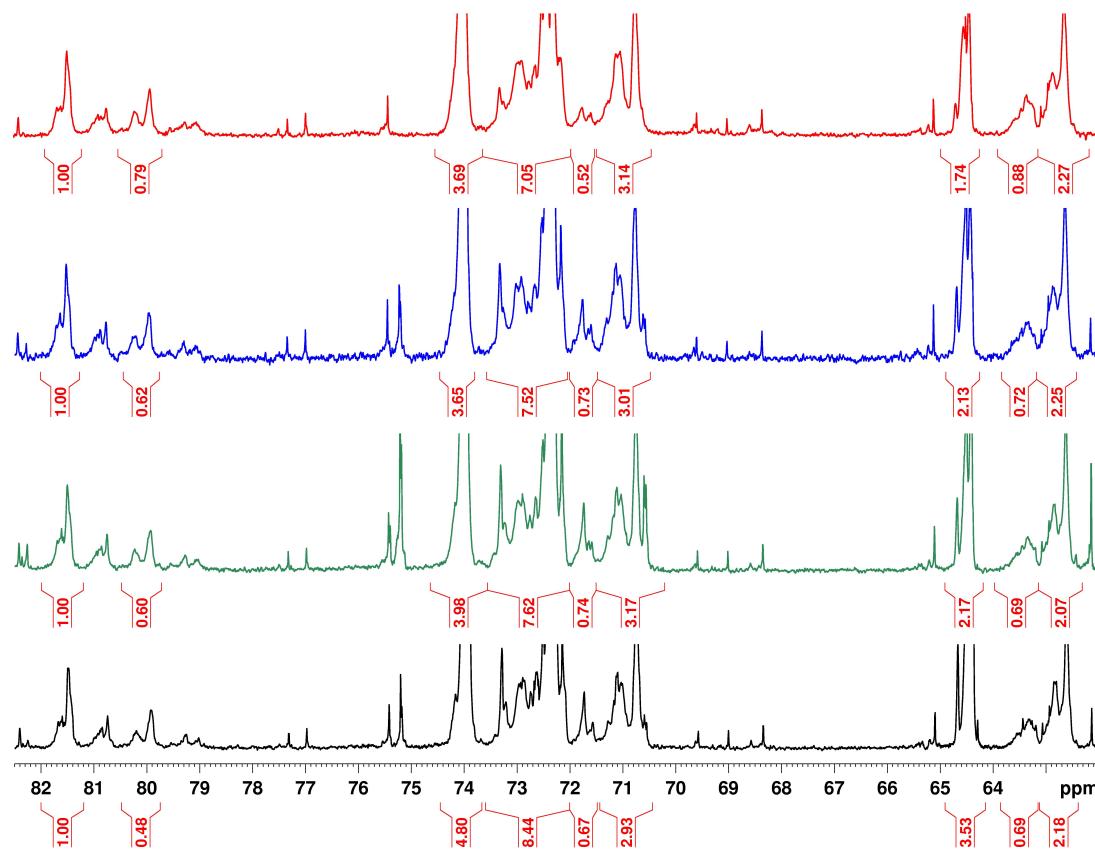
40°C experiment



0°C experiment



**NMR Spectroscopy.**  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopy was performed using a Bruker 600 spectrometer operating at 600 and 150 MHz, respectively. The instrument is equipped with a 14.1 Tesla Bruker magnet, which is controlled by a Bruker AV-II console, and a 5mm Z-gradient TCI Cryo-probe. A 20 second recycle delay was used to insure full relaxation for quantitative measurements, which allowed for the elucidation of specific monomer unit peaks and subsequent calculation of the degree of branching present in each glycidol homopolymer.

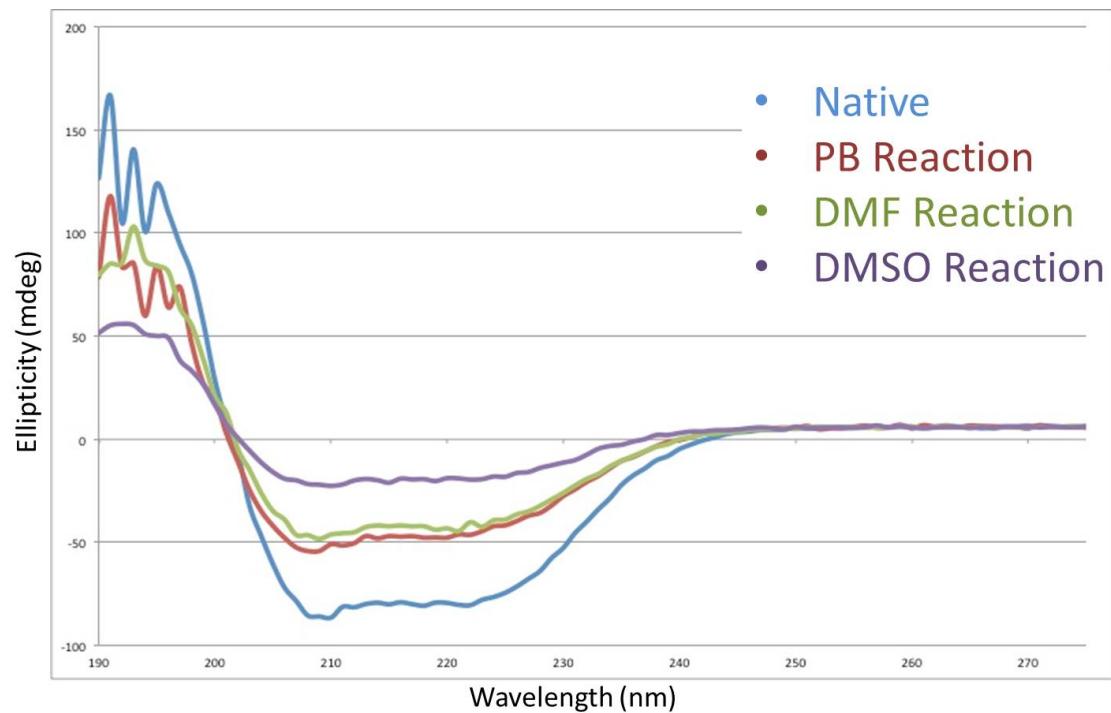


From 40°C (top, see also Figure 1 in article), 20°C, 0°C, -20°C (bottom), basis for data of table in Figure 1.

Calculation of the Degree of Branching (DB) for data of table in Figure 1 reported by Frey<sup>2</sup>.

$$DB = \frac{2D}{2D+L13+L14}$$

**Circular Dichroism (CD):** The use of CD spectrophotometry was employed to verify the retention of the secondary structure of the BSA protein. Spectra were obtained with a Jasco J-720 at 25 ± 1 °C in DI water (1.5x10<sup>-5</sup> M) in a cell with a path length of 1cm (bandwidth=1 nm; step resolution = 0.5 nm; scan speed = 50nm/min; response time = 0.5s). Spectral data was collected from 190-275 nm to ascertain information about the α-helix structure of the BSA protein as well as the structure of the BSA conjugates formed.



1. P. De, M. Li, S. R. Gondi and B. S. Sumerlin, *J Am Chem Soc*, 2008, **130**, 11288-+.
2. A. Sunder, R. Hanselmann, H. Frey and R. Mulhaupt, *Macromolecules*, 1999, **32**, 4240-4246.