Supporting Information

For

Capture and Metathesis-based Release of

Potassium Salts by a Multitopic Ion Receptor

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Figure S1. Partial ¹H NMR spectra of (a) **2** only, (b) **2** with 5.0 equiv of $CsClO_4$, (c) **2** with 5.0 equiv of CsF, (d) **2** with 5.0 equiv of CsCl, (e) **2** with 5.0 equiv of CsBr, and (f) **2** with 5.0 equiv of $CsNO_3$ in $CD_3OD/CDCl_3$ (1:9, v/v).



Figure S2. Partial ¹H NMR spectra of (a) **2** only, (b) **2** with 5.0 equiv of $CsClO_4$, (c) **2** with 5.0 equiv of CsF, (d) **2** with 5.0 equiv of CsCl, (e) **2** with 5.0 equiv of CsBr, and (f) **2** with 5.0 equiv of CsNO₃ in CD₃OD/CDCl₃ (1:9, v/v).



Figure S3. Proposed binding interactions involving receptor **2** and various Cs^+ ion pairs in 10% methanol in chloroform (10% CD₃OD in CDCl₃ for ¹H NMR spectral studies).



Figure S4. Partial ¹H NMR spectra of nitrobenzene- d_5 solutions of **2** after contacting with (a) an ion-free aqueous D₂O solution, (b) an aqueous D₂O solution of CsF (5 equiv), (c) an aqueous D₂O solution of CsCl (5 equiv), (d) an aqueous D₂O solution of CsBr (5 equiv), (e) an aqueous D₂O solution of CsNO₃ (5 equiv), and (f) an aqueous D₂O solution of CsClO₄ (5 equiv).



Figure S5. Partial ¹H NMR spectra of nitrobenzene- d_5 solutions of **2** after exposure to (a) an ionfree aqueous D₂O solution, (b) an aqueous D₂O solution of KNO₃ (5 equiv), (c) an aqueous D₂O solution of CsCl (5 equiv), (d) an aqueous D₂O solution of CsBr (5 equiv), (e) an aqueous D₂O solution of CsNO₃ (5 equiv), and (f) an aqueous D₂O solution of CsClO₄ (5 equiv).



Figure S6. Partial ¹H NMR spectra of nitrobenzene- d_5 solutions of **2** (a) after being washed with D₂O, (b) after contacting with an aqueous KF solution (5 equiv), (c) after contacting with an aqueous KCl solution (5 equiv), (d) after contacting with an aqueous KNO₃ solution (5 equiv), and (e) after the nitrobenzene phase obtained from (d) was washed with first D₂O and then an aqueous D₂O CsClO₄ solution (5 equiv).



Figure S7. Partial ¹H NMR spectra of nitrobenzene- d_5 solutions of **3** (a) after being washed with D₂O and (b) after contacting with an aqueous KNO₃ solution (5 equiv).



Figure S8. Partial ¹H NMR spectra of nitrobenzene- d_5 solutions of **4** (a) after being washed with D₂O and (b) after contacting with an aqueous KNO₃ solution (5 equiv).

Ion pair	crown/crown	crown/pyrrole	glycol/pyrrole	pyrrole/pyrrole
CsF	-148.3	-125.2	-155.1	-148.2
CsCl	-140	-111.2	-138.9	-125.7
CsBr	-136.4	-104.8	-131.7	-114.5
CsNO ₃	-130.3	-110.6	-128.2	-104.3
CsClO ₄	-133.2	-110.1	-123.8	-99.9
KF	-162.5	-140.7	-163.3	-162.2
KCl	-150.1	-128	-150.1	-138.4
KBr	-146.5	-121.9	-146.1	-126.5
KNO3	-140.3	-119.1	-143.5	-115.9
KClO ₄	-142.6	-123.1	-141.8	-112
Anion	pyrrole	cation	crown	glycol
F^{-}	-41.6	\mathbf{K}^+	-75.7	-63.6
Cl	-26.1	Cs^+	-66.5	-40.9
Br	-16.4			
NO ₃ ⁻	-13.2			
ClO ₄ ⁻	-10.1			

Table S1. Calculated Gas-Phase Binding Energy for Ion Pairs and Individual Ions in Different Binding Modes of Receptor 2, ΔE^a (kcal/mol).

 ${}^{a} \Delta E = E(\text{complex}) - E(\text{ligand}) - E(\text{cation}) - E(\text{anion}).$

Supporting info for modeling:

Molecular mechanics calculations were performed using the MMFF94 force field¹ as implemented in PCModel.² Non-default van der Waals parameters used for the cesium cation are described elsewhere.³ Optimized Cartesian coordinates for the four CsF binding modes are provided below.

- 1. Halgren, T. A. J. Comp. Chem. 1996, 17, 490-519.
- 2. PC Model, v9.0, Serena Software, Serena Software, Box 3076, Bloomington, IN 47402-3076.
- Kim, S. K.; Sessler, J. L.; Gross, D. E.; Lee, C.-H.; Kim, J. S.; Lynch, V. M.; Delmau, L. H.; Hay, B. P. J. Am. Chem. Soc. 2010, 132, 5827-5836.

CsCl crown/pyrrole mode 182 -0.019 4.928 Cs 1.313 -5.078 Cl -0.004 0.268 3.574 -1.019 -1.693 0 2.210 С 2.271 -0.726 2.799 С 3.011 0.312 С 3.230 4.183 0.255 С 4.955 -0.750 2.641 С 1.756 4.367 -1.658 С 1.528 2.985 -1.632 С 3.570 1.976 1.462 С 1.597 1.769 3.116 С 0.582 1.009 3.733 С -0.378 0.572 3.513 С 1.515 -0.986 2.690 С 2.489 -0.214 2.063 С 2.538 1.177 2.244 0 -1.248 -2.205 6.336 С 1.606 -0.538 4.573 С -2.300 1.841 2.746 С -3.043 2.630 1.851 С -3.265 3.971 2.193 С -2.678 4.521 3.336 С -1.791 3.761 4.104 С -1.559 2.414 3.792 С -3.599 2.061 0.556 С -1.626 2.205 -1.104 С -2.565 1.443 -0.374 С -2.512 0.044 -0.484 -0.578 С -1.536 -1.256 С -0.595 0.189 -1.936 С -0.609 1.592 -1.865 0 -1.186 8.324 0.707 С 0.509 2.352 -2.563 0 1.140 8.129 2.607

0	-2.323	0.474	2.619
0	-2.590	-2.200	4.097
0	2.164	5.783	4.110
0	1.660	3.126	3.315
Ν	0.745	-7.637	2.326
C	-2 424	4 103	-2 140
C	-3 159	5 359	-1 686
c	2 956	J.JJJ	0 970
d	-2.050	7.550	-0.879
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C	-2 844	-9 479	1 534
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и ц	-2 1/5	6 173	0 140
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п บ	2 055	-0 541	-3 053
п u	_0 139	-0.541	-3.033
п u	-0.130	-1.040	-3.320
п u	-2.220	1 215	-2.909
п u	2.200	2 720	-2.302
п	2.021	4 920	-3.375
п 11	2.600	4.920	2.000
H	2.509	6.058	-0.096
H	2.505	4.771	-2.188
H T	3.504	1.548	2.688
H	2.861	2.996	3.397
H	2.269	-0.374	3.282
H	0.121	-1.531	3.686
H TT	-2.017	-0.335	3.356
н	-3.233	1.540	2.717
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H TT	0.475	3.503	4.997
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и ц	0 185	9 5 8 5	2.310
н	-1 289	8 748	-0.030
н	0 092	9 852	0.020
и ц	1 396	8 206	-1 401
н	-0 073	7 220	-1 533
и ц	0.075	9 641	-4 351
и ц	1 431	9.041 8.578	-3 667
и ц	_1 ///	7 581	-4 069
п п	_0 215	7.501	-5 457
п u	-0.313	5 190	-4 210
п п	-1.400	5.190	-4.210
п u	-0.470	2.574	-5.000
п п	1 251	3.137	-4.900
п u	1.351	-9 964	-4.422
п u	4.337	-0.904	-1.920
п u	3.205	-7.708	-1.111
п u	4.732	- / . 4 / L E 1 E 1	-2.119
п 11	-1.302	-5.151	-2.403
H	-2.939	-5.494	-3.212
н 	-1.464	-5.427	-4.192
н т	-2.233	-9.040	-4.221
н т	-1.958	-/.586	-5.182
п 11	-3.440	-/./49	-4.230
н т	-4./89	-5.593	-0.180
H	-5.107	-3.209	-0.425
H	-3.027	-2.543	3.292
H	-2.694	-4.945	3.541
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C	0 245	4 659	3 964
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C	4,026	0.410	1.350 N 369
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C	-3 35/	-0 902	0.200
C	-3 548	0 601	-0 049
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Ν	2.451	-7.314	0.059
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С	2.410	-9.079	-1.273
C	2.754	-9.506	0.031
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C	-0.117	-6.513	3.347
Ν	-2.395	-7.145	0.882
С	-2.161	-7.449	2.206
С	-2.454	-8.795	2.354
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С	-2.224	-6.728	4.614
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182			
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н	-3 428	1 277	-2 781
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л U	-2 000	-2.232	4.390
п u	-2.099	-9./14	3.554
п	-4.010	- 2 . / 3 /	0.029

Η	-3.235	-9.656	-2.531
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Н	1.079	8.120	0.902
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Η	2.677	-8.745	4.581
Н	2.047	-7.374	5.486
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Η	4.813	-7.519	-3.301
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Η	0.097	-3.281	-5.087
Η	-3.554	-3.094	-2.816
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