

Supporting Information

HBC-Porphyrin – Close contact chromophores

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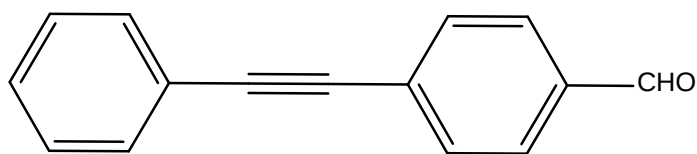
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Experimental Section

All chemicals were purchased from chemical suppliers and used without further purification. All analytical reagent-grade solvents were purified by distillation. Hexakis-*tert*.butyl-hexaperibenzocoronene was synthesized following literature procedures.¹ Thin layer chromatography (TLC): Merck silica gel 60 F254. Detection: UV lamp ($\lambda = 254$ & 366 nm). Flash- chromatography (FC): Macherey–Nagel silica gel 60 M (230–400 mesh, 0.04 – 0.063 mm). Solvents were purified by distillation prior to use. Mass spectrometry: AXIMA Confidence MALDI TOF mass spectrometer, nitrogen UV laser, 50 Hz, $\lambda=337$ nm wavelength, Shimadzu. 3,5-dimethoxy-4-hydroxycinnamic acid (SIN), 2,5-dihydroxybenzoicacid (DHB) and 2-[(2*E*)-3-(4-*tert*-Butylphenyl)-2-methylprop-2-enylidene]malononitrile (DCTB) were used as matrix or without matrix (OM). High-resolution electrospray ionization (ESI) mass spectrometry: Bruker maXis 4G and Bruker microTOF II. NMR spectroscopy: JEOL JNM EX 400 and JEOL JNM GX 400 and Bruker Avance 300. Chemical shifts are referenced to residual protic impurities in the solvents (^1H) or the deuterated solvent itself (^{13}C) and reported relative to external SiMe_4 . The resonance multiplicities are indicated as br (broad), s (singlet), d (doublet), t (triplet), q (quartet) and m (multiplet).

1. L. Zhai, R. Shukla, S. H. Wadumethrige and R. Rathore, *J. Org. Chem.*, 2010, **75**, 4748-4760.

4-(phenylethynyl)benzaldehyde (1)



Chemical Formula: C₁₅H₁₀O

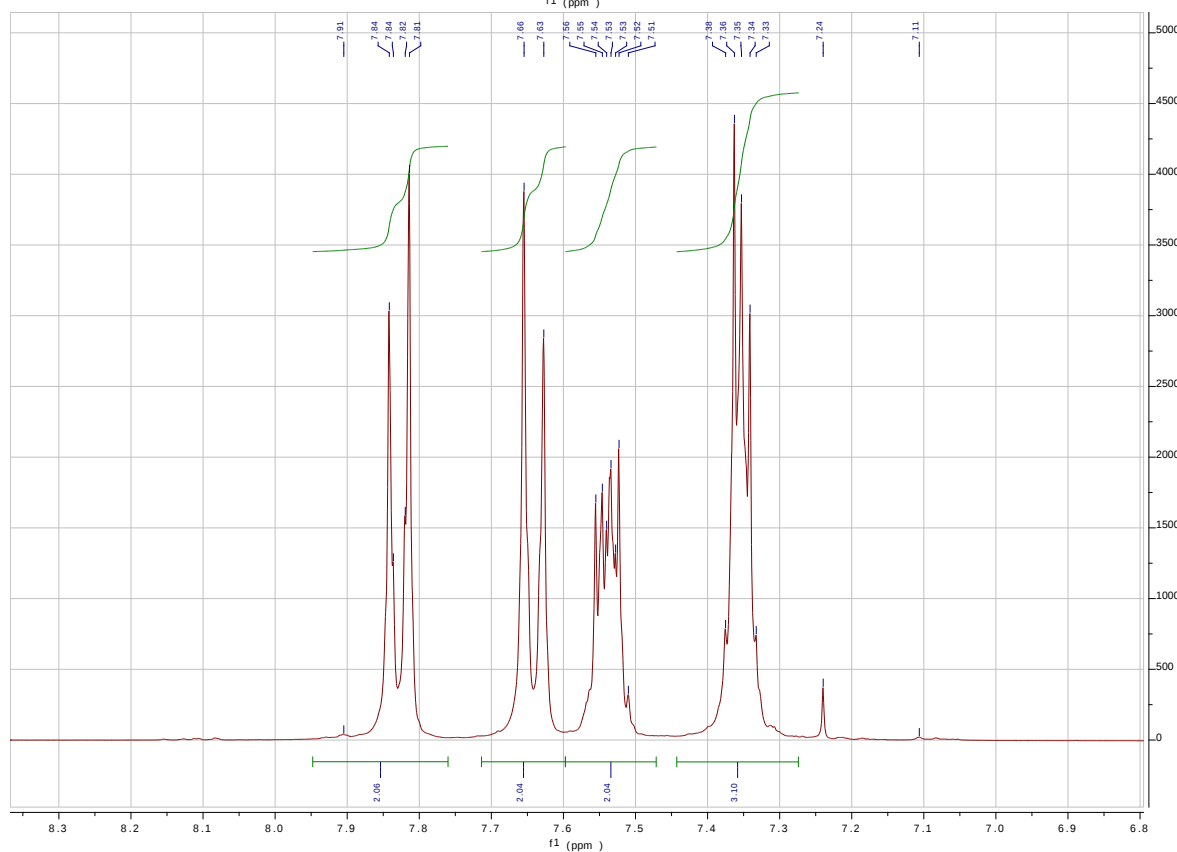
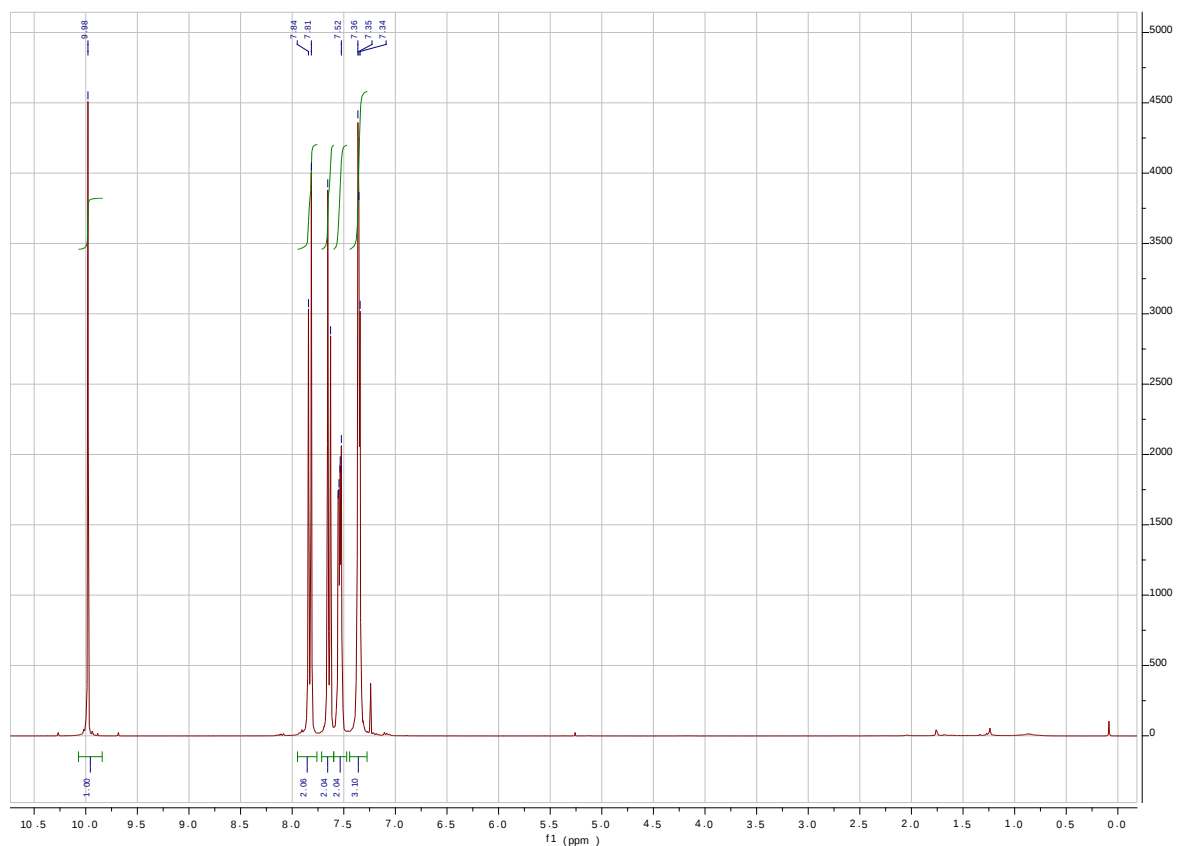
Exact Mass: 206,07

Molecular Weight: 206,24

m/z: 206.07 (100.0%), 207.08 (16.4%), 208.08 (1.5%)

To a solution of phenylacetylene (4.85 g, 47.6 mmol) and 4-bromobenzaldehyde (8.00 g, 43.2 mmol) in 150 ml dry MeCN with 15 ml NEt₃, copper(I)iodide (0.16 g, 0.9 mmol) and Pd(Cl₂(PPh₃)₂) (0.30 g, 0.04 mmol) were added under argon atmosphere. The reaction mixture was stirred under reflux overnight. After cooling to room temperature dichloromethane was added and the crude mixture was washed with water and diluted hydrochloric acid. The organic layer was dried over magnesium sulfate and after evaporation of the solvent the product was obtained as beige shining pellets in 60% yield after recrystallization from small amounts of *n*-hexane.

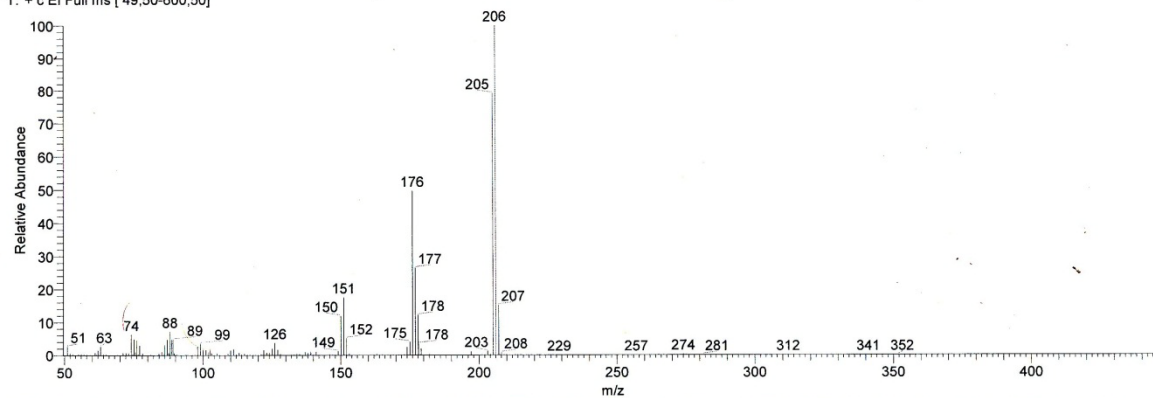
^1H NMR (400 MHz, CDCl_3 , rt): δ = 9.98 (s, 1H, CHO), 7.83 (d, 2H, 3J = 8 Hz, $H\text{-C}_{\text{arom}}$), 7.65 (d, 2H, 3J = 8 Hz, $H\text{-C}_{\text{arom}}$), 7.54 (m, aromatic), 7.35 (m, aromatic) ppm.



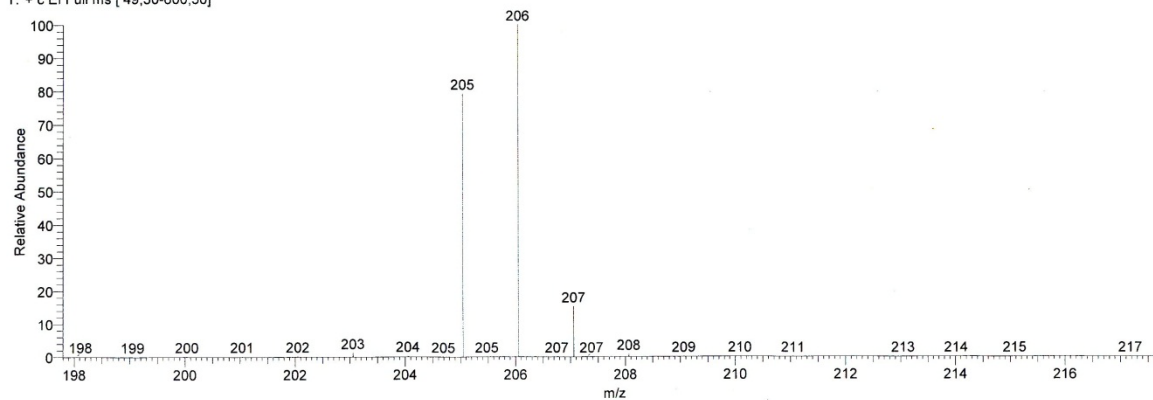
^{13}C NMR (100 MHz, CDCl_3 , rt): δ = 191.5, 135.4, 132.2, 131.9, 129.6, 129.6, 129.1, 128.6, 122.5 ppm.

MS-EI(+): m/z [M^+] $\text{C}_{15}\text{H}_{10}\text{O}$: calc.: 206.07, found: 206.

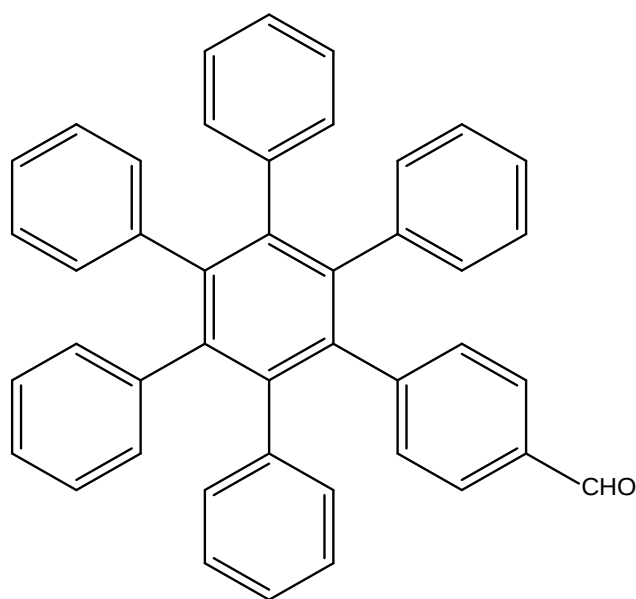
T: + c EI Full ms [49,50-600,50]



T: + c EI Full ms [49,50-600,50]



3',4',5',6'-tetraphenyl-[1,1':2',1''-terphenyl]-4-carbaldehyde – HPB-CHO (2)



Chemical Formula: $C_{43}H_{30}O$

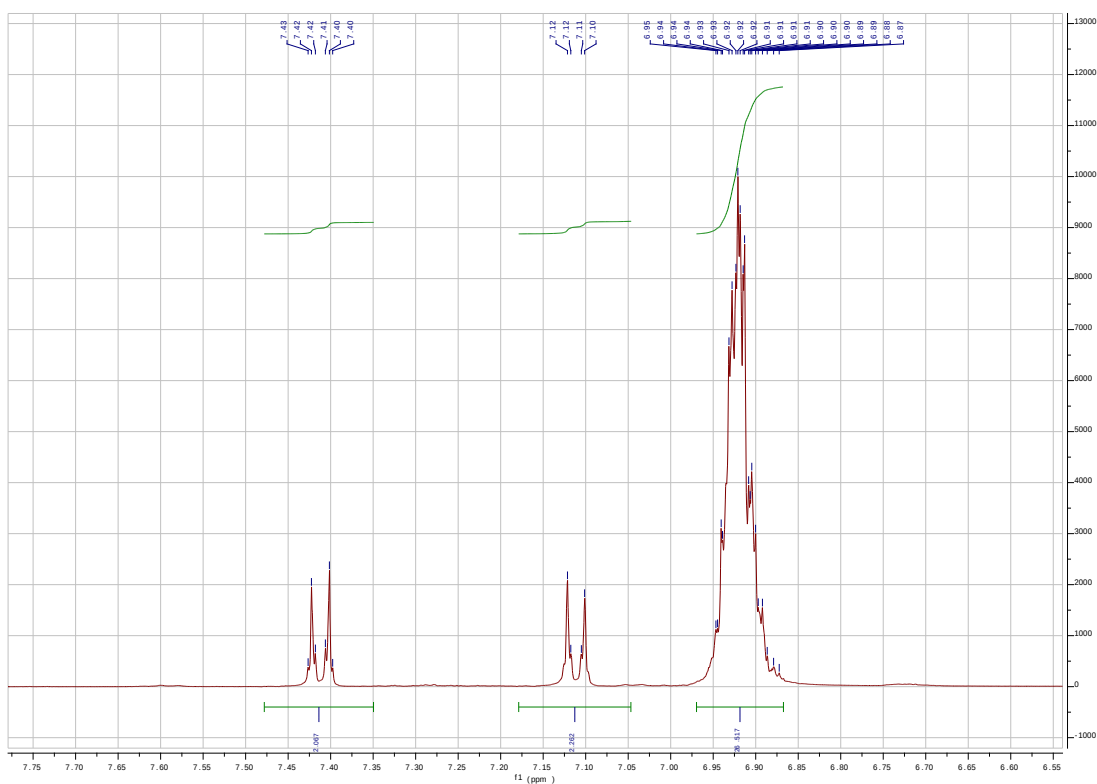
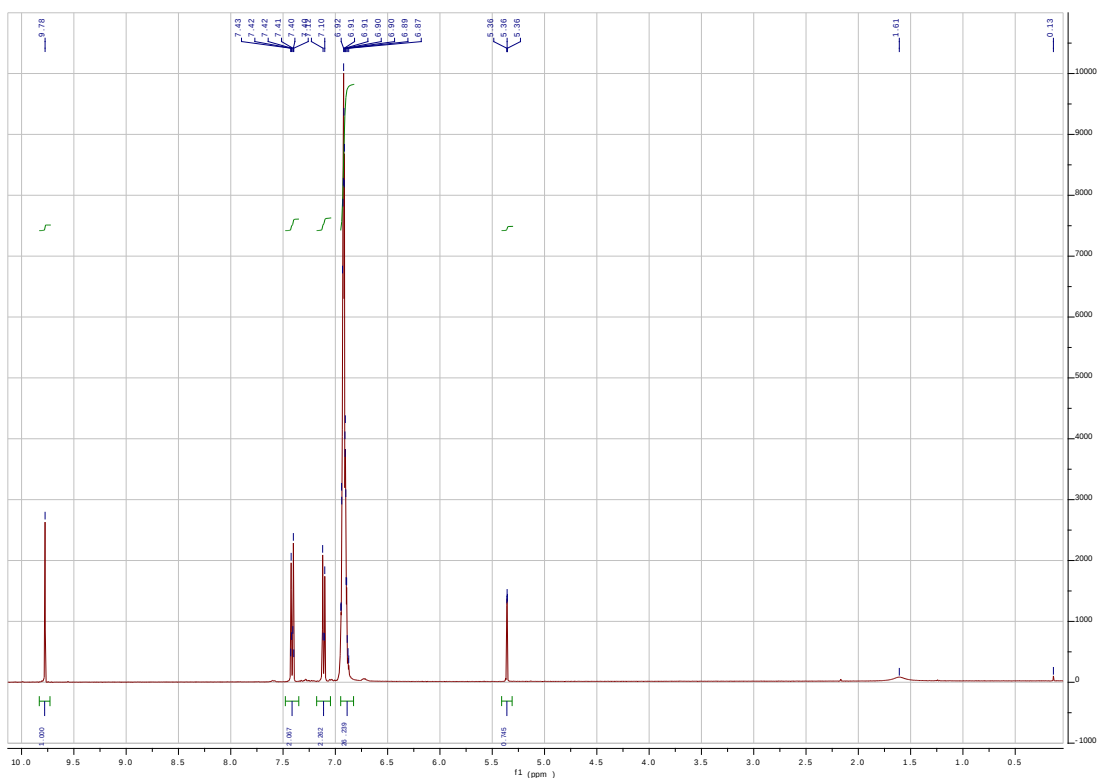
Exact Mass: 562,23

Molecular Weight: 562,70

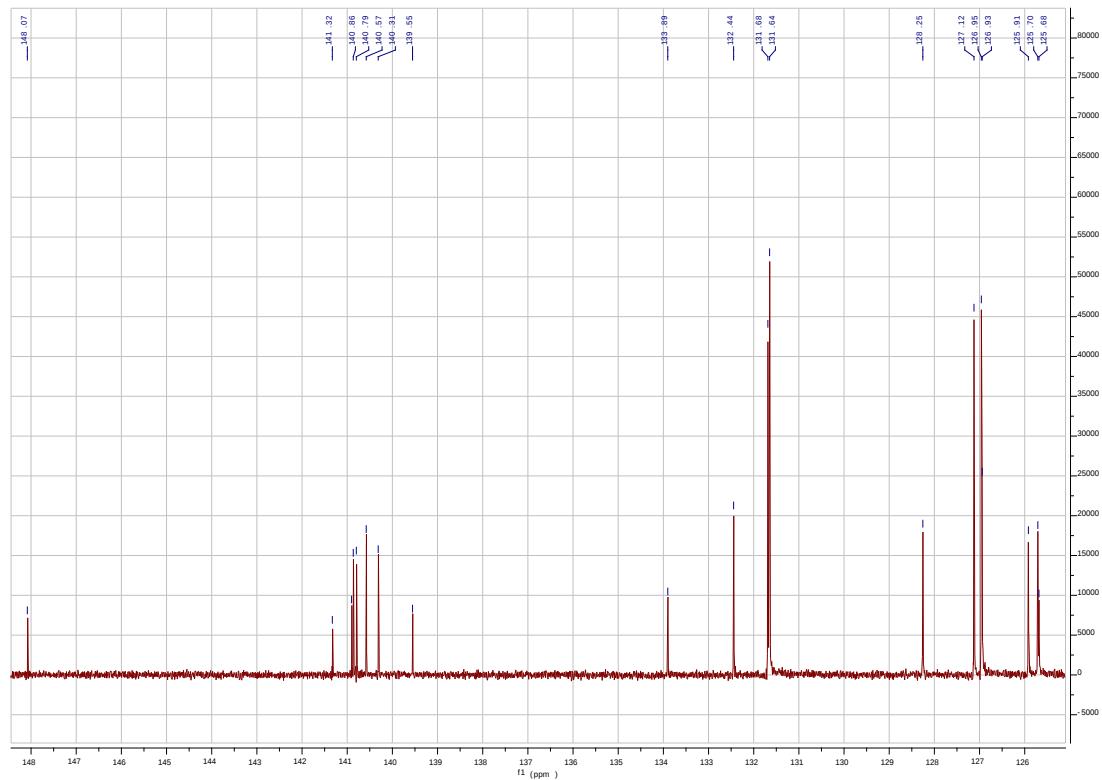
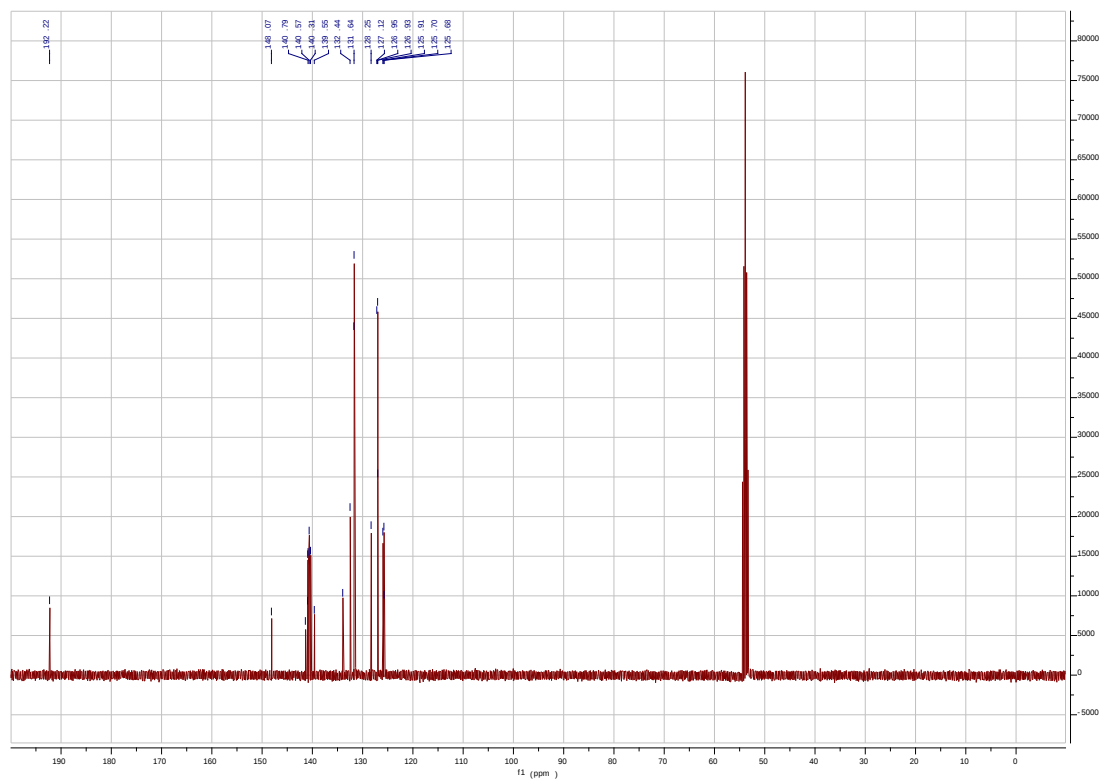
m/z: 562.23 (100.0%), 563.23 (46.5%), 564.24 (10.7%), 565.24 (1.7%)

4.85 g (12.6 mmol) neat tetraphenylcyclopentadienone were mixed with 5.2 g (25.3 mmol) **1** in a 30 cm long Schlenck tube equipped with a bubble counter under low, but, constant argon stream. The tube was heated to 240°C in a mono mode laboratory microwave oven (CEM discover – open vessel dynamic) under dynamic irradiation with peakpower of 300W. After 30 minutes the tube was removed from the oven and cooled down to room temperature. The crude product was dissolved in as little dichloromethane as possible and transferred to a 300 mL Erlenmeyer flask. After evaporation to dryness the raw product was repeatedly washed and filtered from boiling ethanol to yield HPB-CHO (**2**) as white powder in typically >60 % yield.

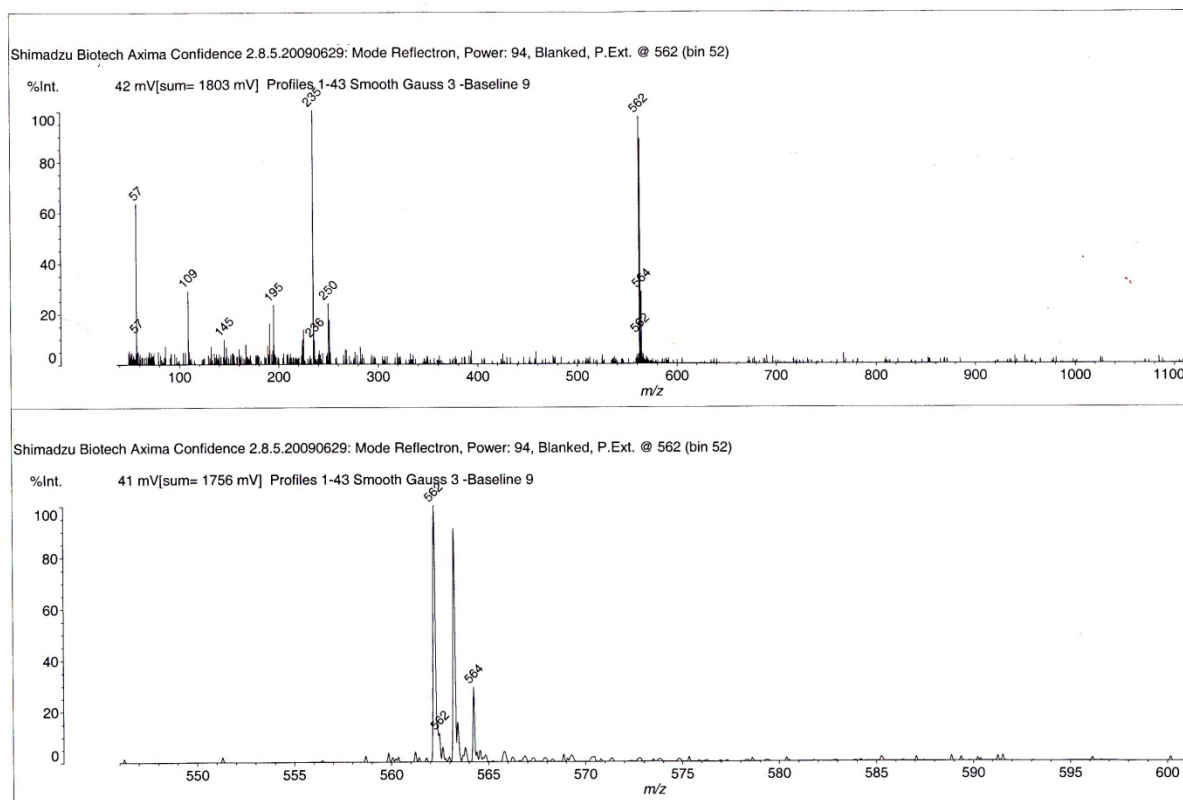
^1H NMR (400 MHz, CD_2Cl_2 , rt): δ = 9.78 (s, 1H, CHO), 7.42 (d, 2H, 3J = 8 Hz, $H\text{-C}_{\text{arom,CHO}}$), 7.11 (d, 2H, 3J = 8 Hz, $H\text{-C}_{\text{arom,CHO}}$), 6.91 (m, 25H, $H\text{-C}_{\text{arom}}$) ppm.



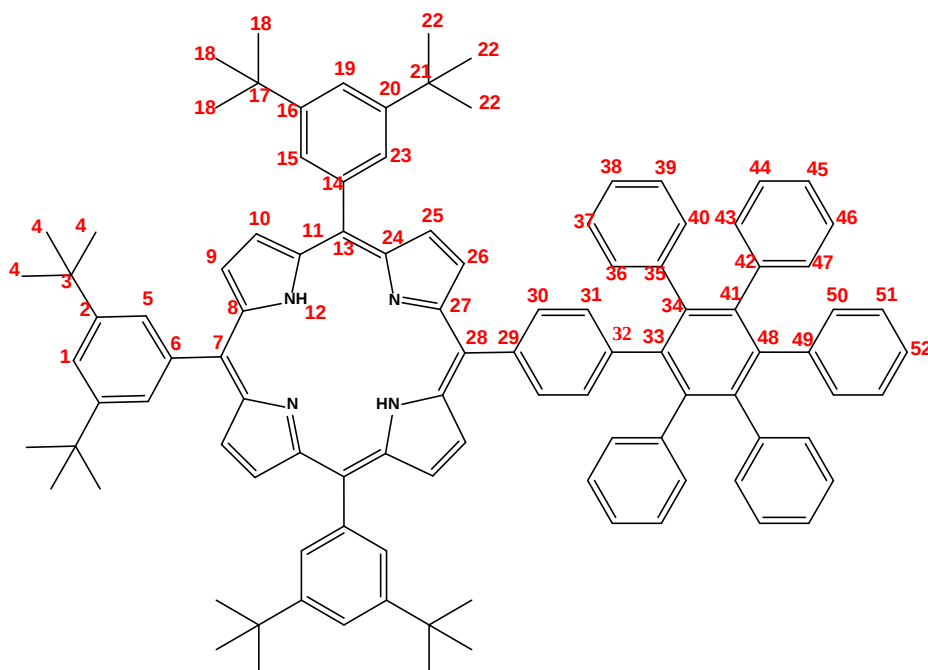
¹³C NMR (100.5 MHz, CD₂Cl₂, rt): δ = 192.22, 148.07, 140.79, 140.57, 140.31, 139.55, 132.44, 131.64, 128.25, 127.12, 126.95, 126.93, 125.91, 125.70, 125.68



MALDI-TOF-MS (DCTB): m/z [M^+] $C_{43}H_{30}O$: calc.: 562.23, found: 562.



5-(4'-2'',3'',4'',5'',6''-pentaphenyl-phenyl)-10,15,20-tris(3',5'-*tert*-butyl-phenyl)-porphyrin – mono-HPB-porphyrin (**3**)



Chemical Formula: C₁₀₄H₁₀₂N₄

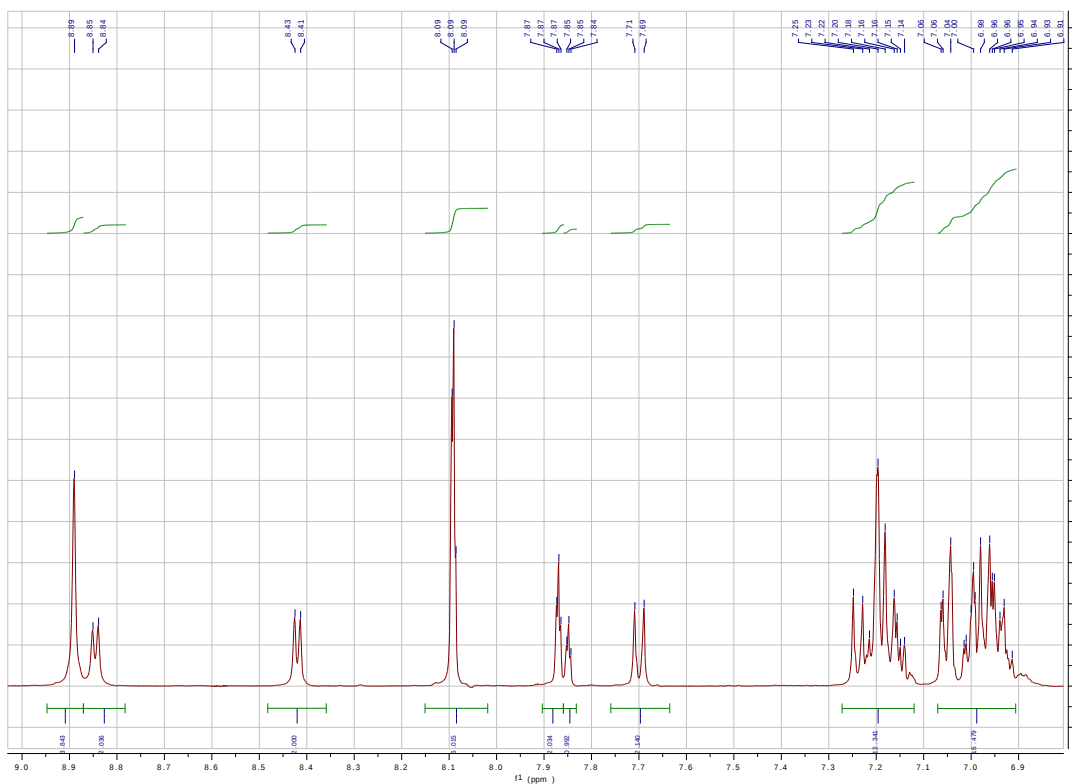
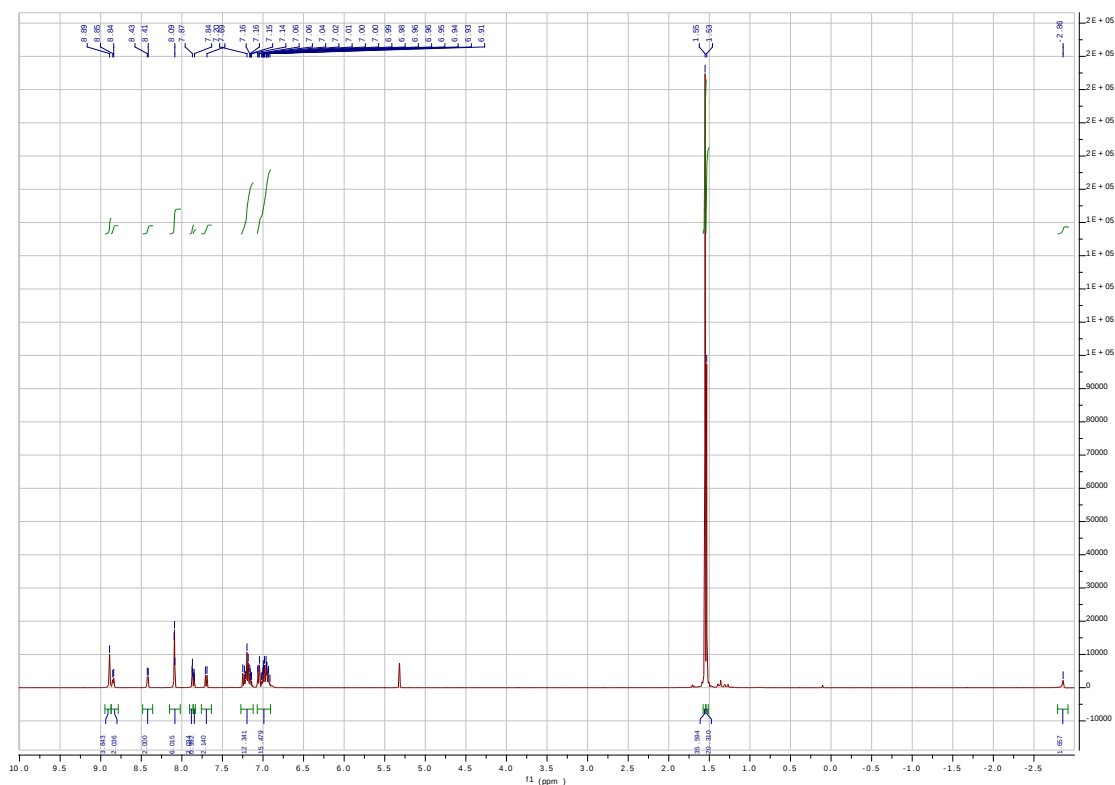
Exact Mass: 1406,81

Molecular Weight: 1407,95

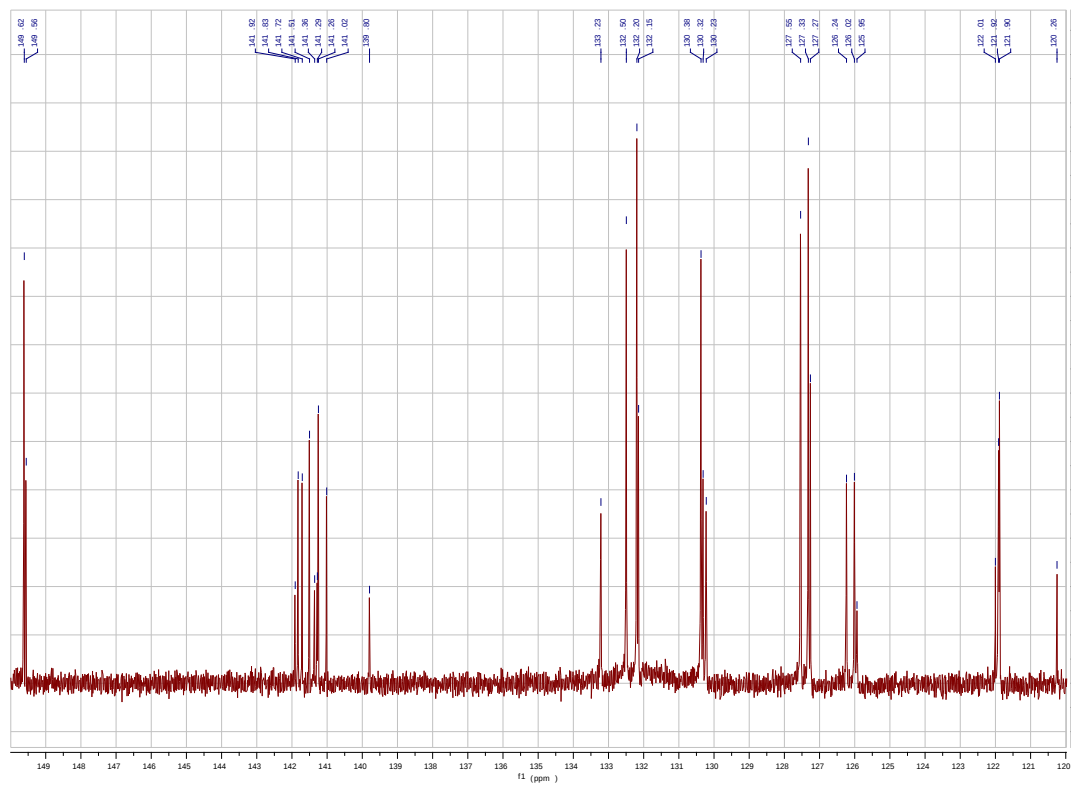
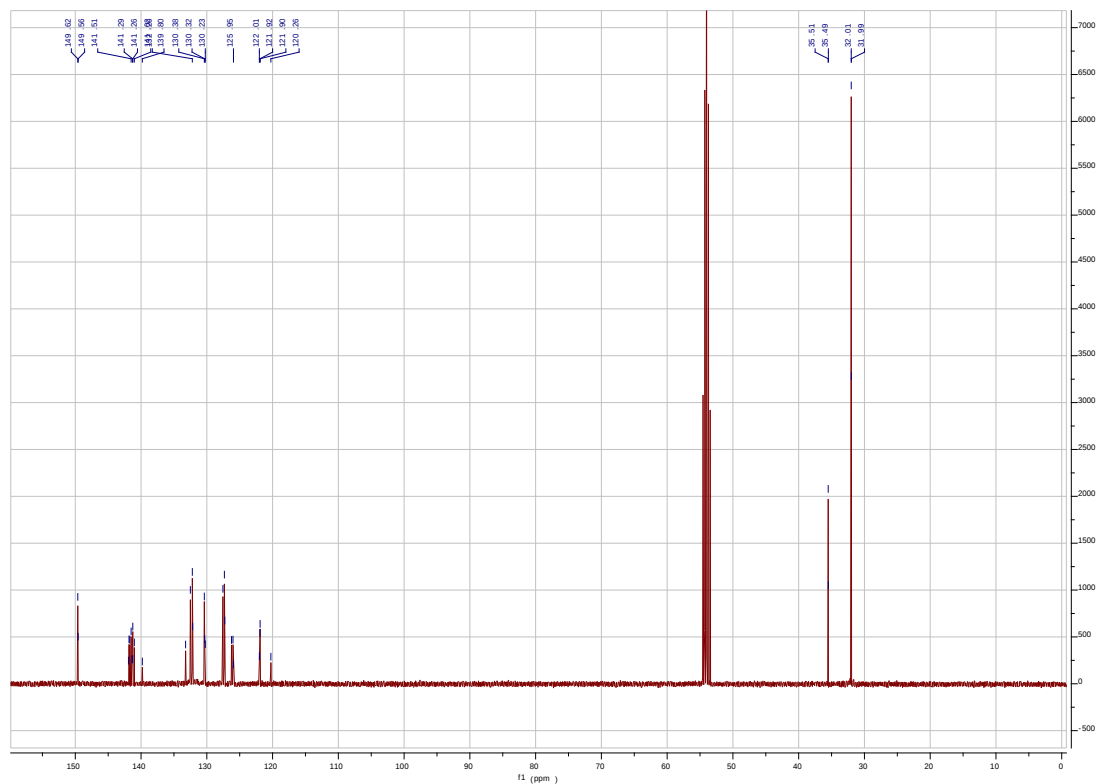
m/z: 1407.81 (100.0%), 1406.81 (87.7%), 1408.82 (56.1%), 1409.82 (20.9%), 1410.82 (5.8%), 1408.81 (1.5%), 1411.83 (1.2%), 1407.82 (1.0%)

In 1.4 L dichloromethane and 10 mL EtOH compound **2** (2.00 g, 3.55 mmol) together with 3,5-*tert*-butylbenzaldehyde (2.33 g, 10.66 mmol) and pyrrole (2.15 g, 31.98 mmol) was dissolved and purged with nitrogen. After addition of BF₃·OEt₂ (0.05 g, 0.36 mmol) the reaction mixture was stirred for one hour at room temperature. DDQ (1.21 g, 5.33 mmol) was added and the reaction mixture stirred for two hours at room temperature. The solvent was evaporated and the crude reaction mixture first purified by a plug followed by column chromatography with hexane:dichloromethane (5:1) as eluent. The purple crystals of porphyrin **3** were obtained as second fraction and recrystallized from MeOH/hexane in up to 19 % yield.

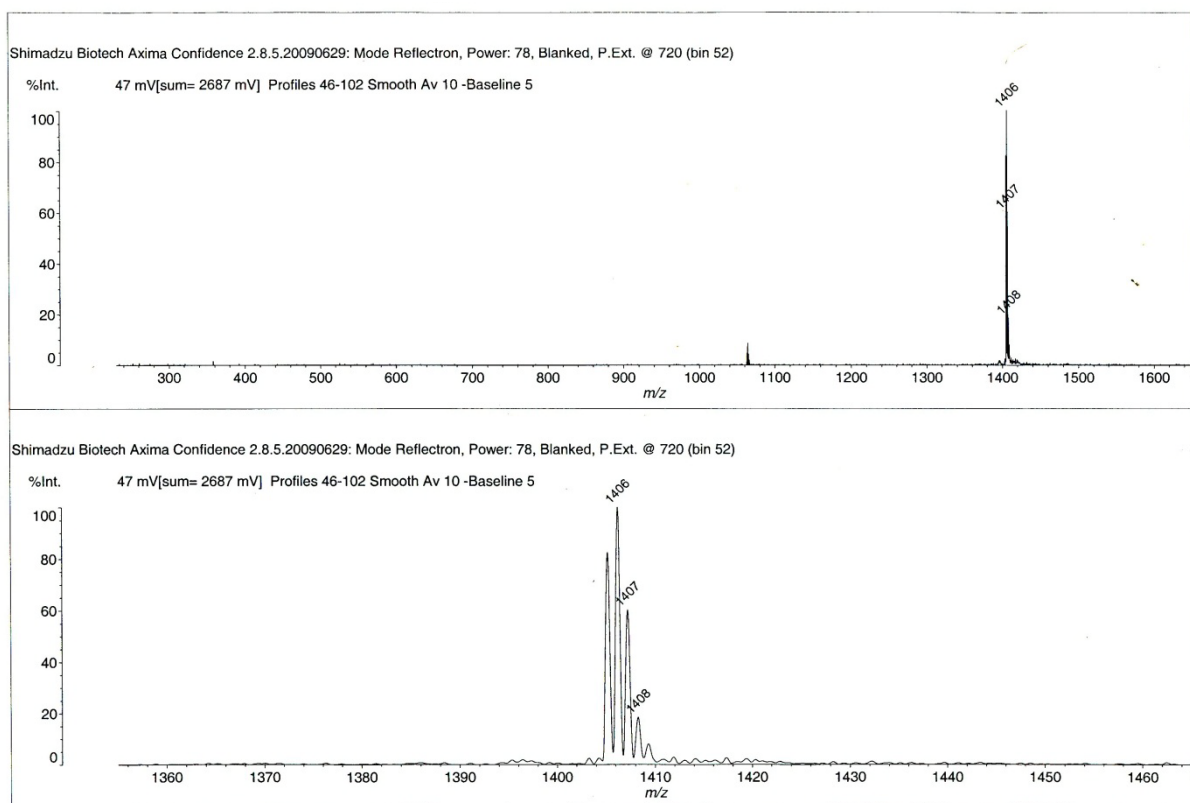
^1H NMR (400 MHz, CD_2Cl_2 , rt): δ [ppm] = 8.89 (s, 4H, 9/10), 8.85 (d, 2H, 3J = 4.7 Hz, 25), 8.42 (d, 2H, 3J = 4.7 Hz, 26), 8.09 (m, 6H, 15/19/23), 7.87 (t, 2H, 4J = 1.7 Hz, 5), 7.85 (t, 1H, 4J = 1.7 Hz, 1), 7.70 (d, 2H, 3J = 7.8 Hz, 30), 7.25-7.14 (m, 12H, 31/36/40/43/47/50), 7.06-6.91 (m, 15H, 37-39/44-46/51/52), 1.55 (s, 36H, 18/22), 1.53 (s, 18H, 4), -2.86 (s, 2H, 12)



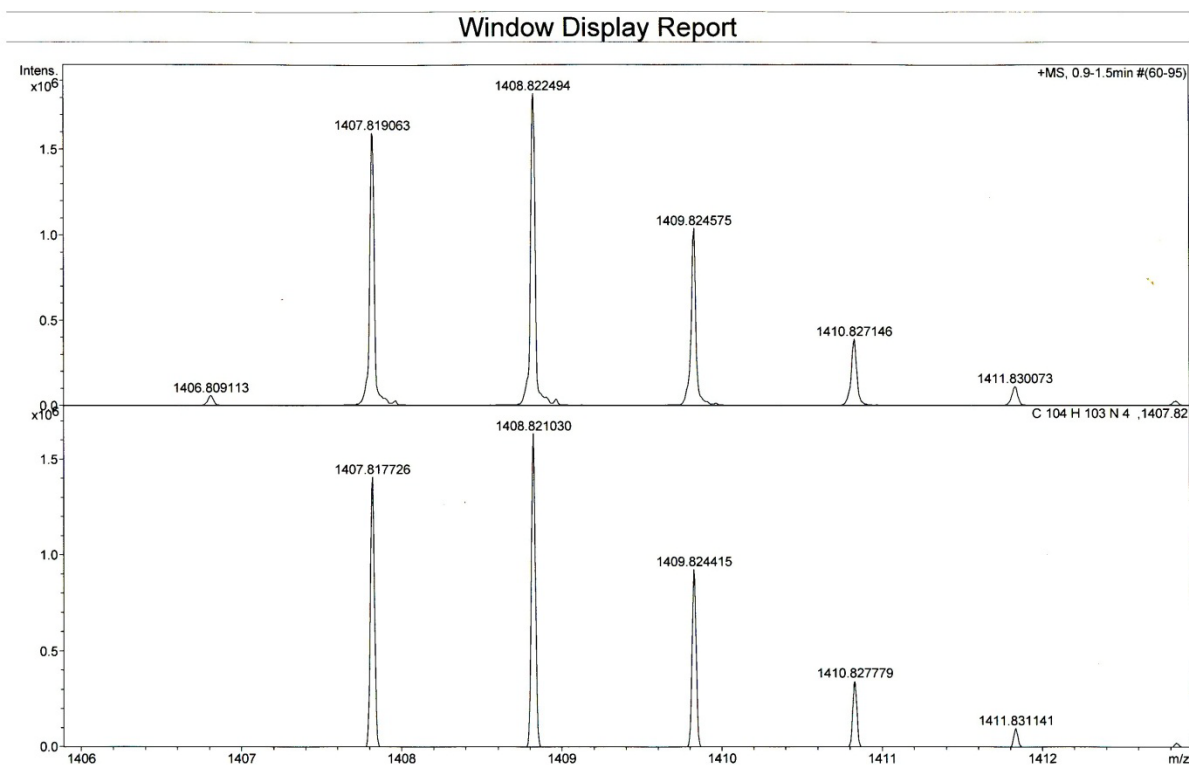
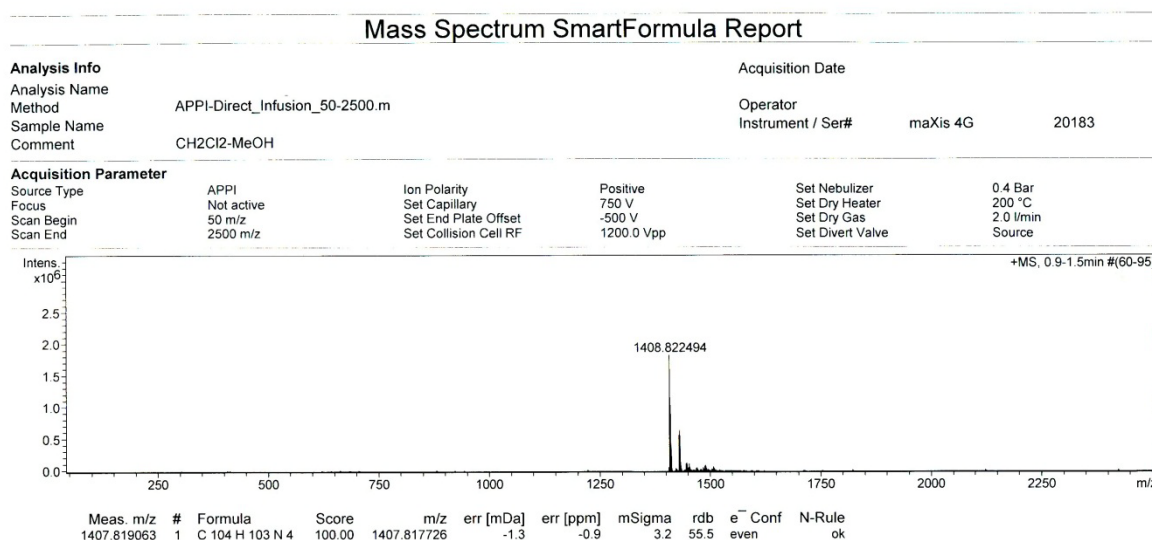
¹³C NMR (100 MHz, CD₂Cl₂, rt): δ [ppm] = 149.62, 149.56, 141.92, 141.83, 141.72, 141.51, 141.36, 141.29, 141.26, 141.02, 139.80, 131.23, 132.50, 132.20, 132.15, 130.38, 130.32, 130.23, 127.55, 127.33, 127.27, 126.24, 126.02, 125.95, 122.01, 121.92, 121.90, 120.26, 35.51, 35.49, 32.01, 31.99



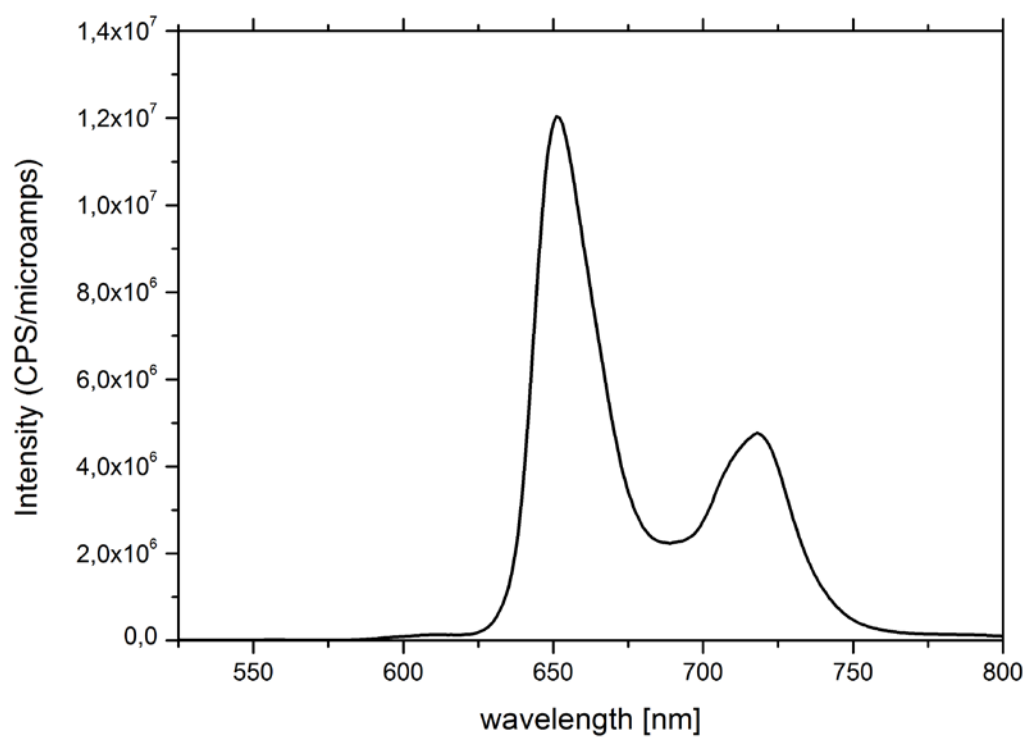
MALDI-TOF-MS (OM): m/z [M^+] $C_{104}H_{102}N_4$: calc.: 1407.95, found: 1405.



HRMS (ESI⁺) for [M-H⁺] C₁₀₄H₁₀₃N₄: calc.: 1407.817726, found: 1407.819063.

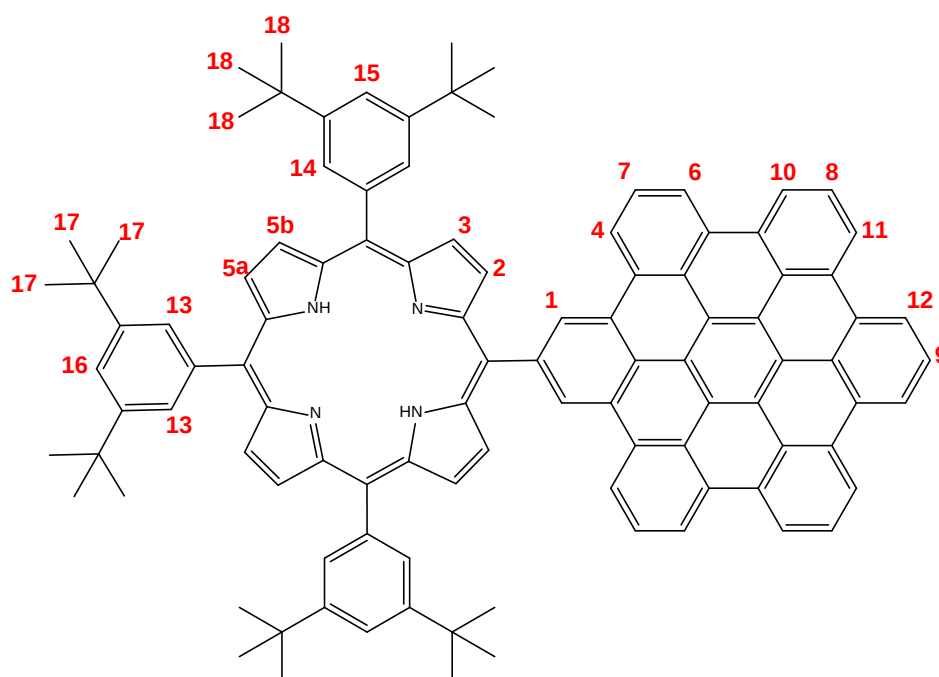


UV/Vis (THF): $\lambda_{max}(\epsilon[M^{-1}cm^{-1}]) = 419 (472200), 515 (14500), 549 (6000), 592 (1700), 647 (1300) \text{ nm.}$



Emission spectra of the HPB-Porphyrin **3** after excitation at 515 nm.

5-(HBC)-10,15,20-tris(3',5'-*tert*-butyl-phenyl)-porphyrin – mono-HBC-porphyrin (4)



Chemical Formula: $C_{104}H_{90}N_4$

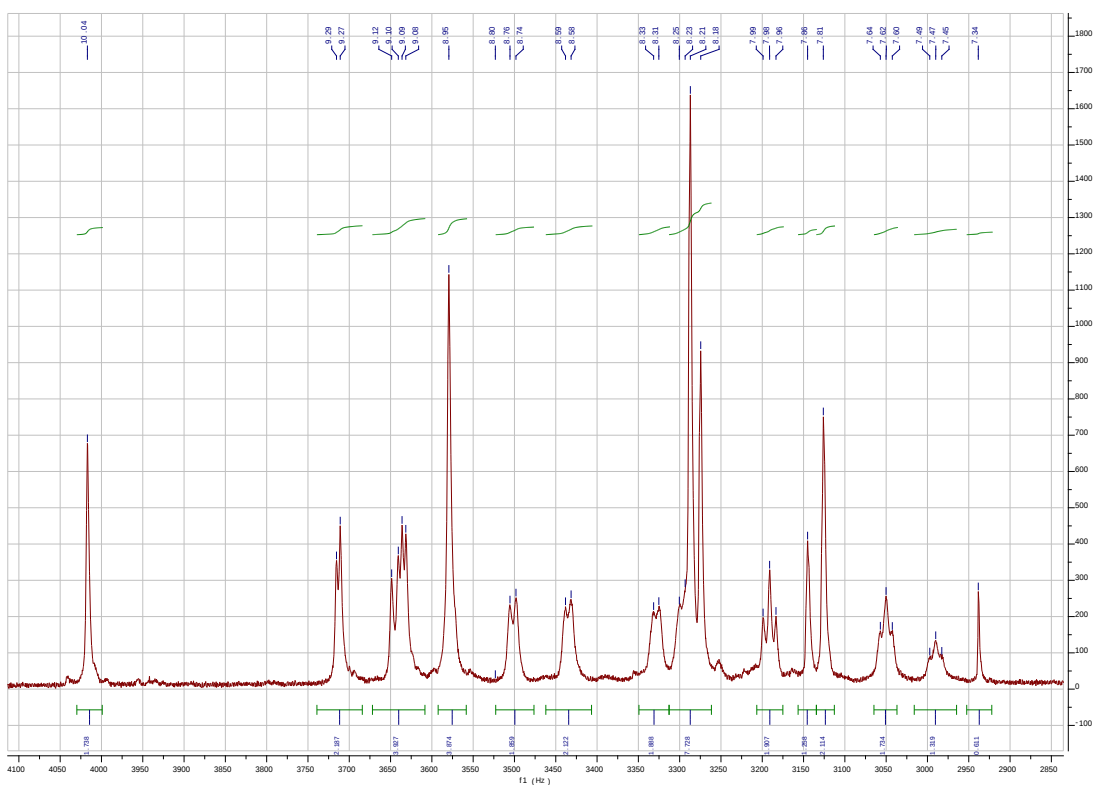
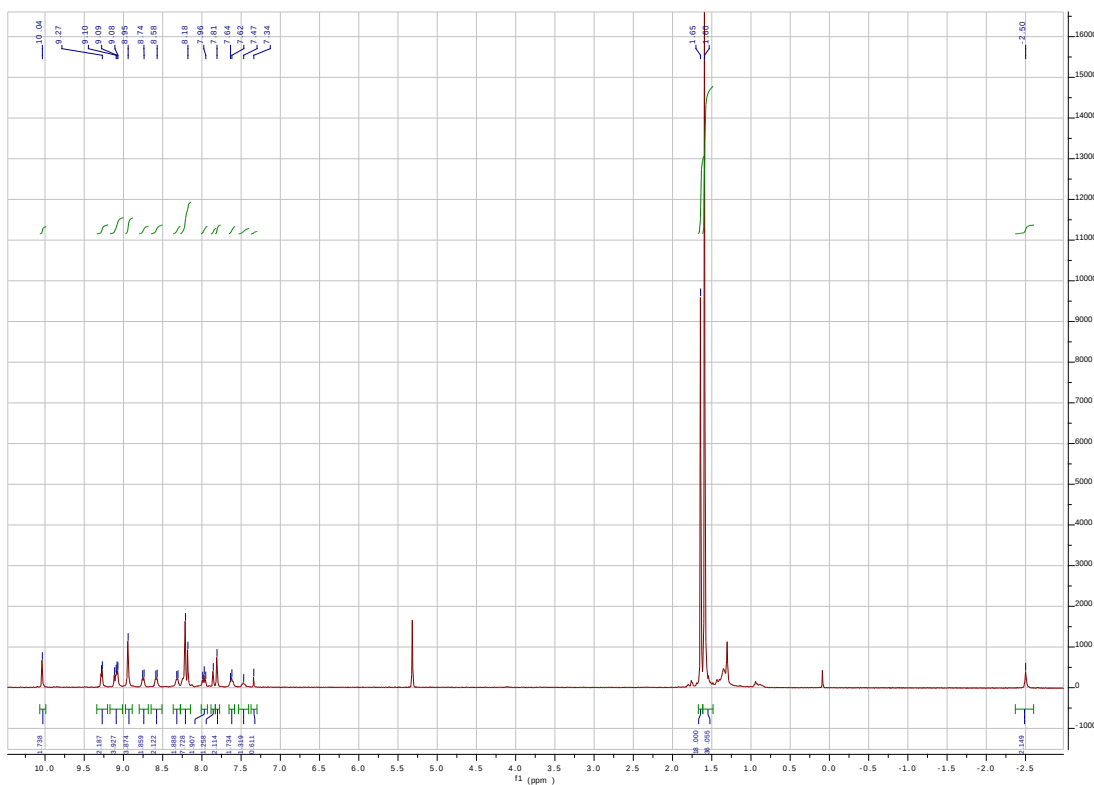
Exact Mass: 1394,72

Molecular Weight: 1395,85

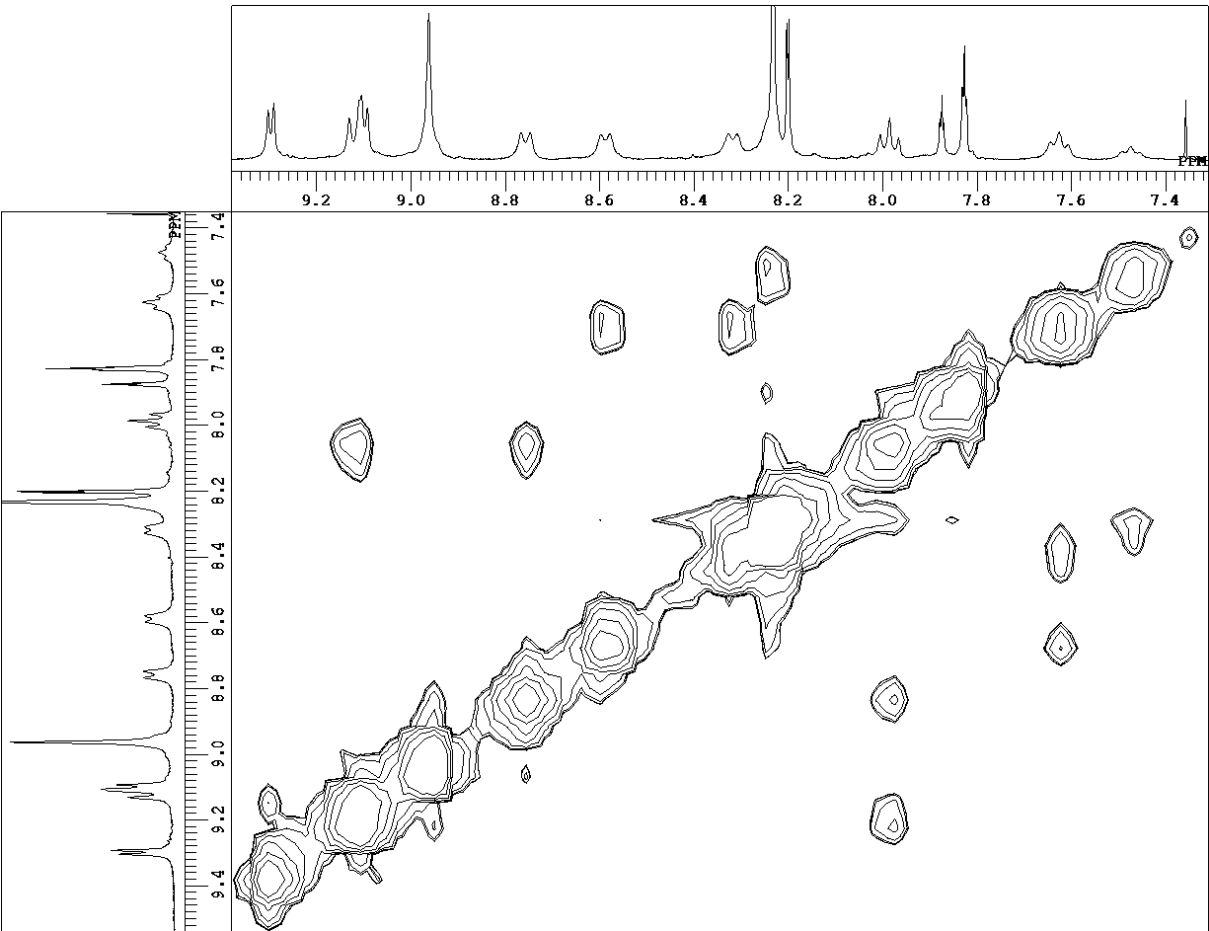
m/z: 1395.72 (100.0%), 1394.72 (88.1%), 1396.72 (56.7%), 1397.73 (20.9%), 1398.73 (5.7%), 1395.71 (1.3%), 1399.73 (1.3%), 1396.73 (1.0%)

To an ice cooled, argon saturated solution of 15 mg (0.011 mmol) of **4** in 50 mL dry dichloromethane 0.57 ml of a freshly prepared solution of 300 mg/mL water free iron(III)chloride in dry nitromethane are added dropwise by syringe through a pressure equilibrated septum while maintaining a low but constant stream of argon bubbling through the solution. Cooling is maintained for typically one hour and evaporated solvent is replaced if necessary. Subsequently, the flask is sealed, stirred for 8 hours then quenched by addition of 50 mL cold methanol and neutralized with sat. NaHCO_3 . The mixture is poured into 200 mL water/cyclohexane (1:1 v/v), the organic layer is washed two times with distilled water and the solvent is removed under reduced pressure. Further purification can be performed by column chromatography (silica gel, DCM/Hexane 1:2), if necessary. The final product is obtained as bronze solid in good to quantitative yields.

^1H NMR (400 MHz, CS_2 , CD_2Cl_2 , rt): δ [ppm] = 10.04 (s, 2H, 1), 9.28 (d, 2H, $^3J = 4.7$ Hz, 2), 9.11 (d, 2H, $^3J = 7.9$ Hz, 4), 9.09 (d, 2H, $^3J = 4.7$ Hz, 3), 8.95 (s, 4H, 5a/5b), 8.75 (d, 2H, $^3J = 7.2$ Hz, 6), 8.59 (d, 2H, $^3J = 6.7$ Hz, 10), 8.32 (d, 2H, $^3J = 6.4$ Hz, 11), 8.24 (d, 2H, 12), 8.21 (s, 4H, 14), 8.18 (s, 2H, 13), 7.98 (t, 2H, $^3J = 7.6$ Hz, 7), 7.86 (s, 1H, 16), 7.81 (s, 2H, 15), 7.62 (t, 2H, $^3J = 7.3$, 8), 7.47 (t, 1H, $^3J = 7.7$ Hz, 9), 1.65 (s, 18H, 17), 1.60 (s, 36H, 18)



COSY-2D NMR (400 MHz, CS₂/CD₂Cl₂) – aromatic region



HRMS (ESI⁺) for C₁₀₄H₉₀N₄: calc.: 1395.72383, found: 1395.72197

Mass Spectrum SmartFormula Report

Analysis Info

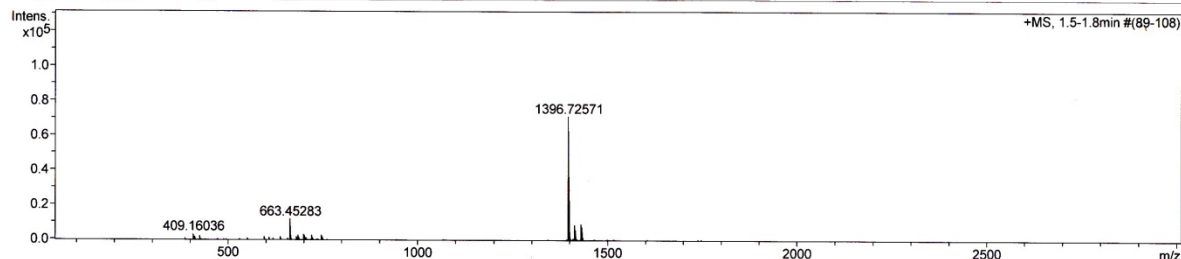
Analysis Name
Method tune_pos_wide.m
Sample Name
Comment ACN-Tol

Acquisition Date

Operator
Instrument / Ser# micrOTOF 10364

Acquisition Parameter

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Scan Begin	50 m/z	Set Capillary	4500 V	Set Dry Gas	4.0 l/min
Scan End	3000 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste



Meas. m/z	#	Formula	Score	m/z	err [mDa]	err [ppm]	mSigma	rdb	e ⁻ Conf	N-Rule
1395.72197	1	C 104 H 91 N 4	100.00	1395.72383	1.9	1.3	70.8	61.5	even	ok
	2	C 99 H 91 N 6 O 2	50.79	1395.71980	-2.2	-1.6	81.9	57.5	even	ok
	3	C 92 H 95 N 6 O 7	3.75	1395.72568	3.7	2.7	106.9	48.5	even	ok

Display Report

Analysis Info

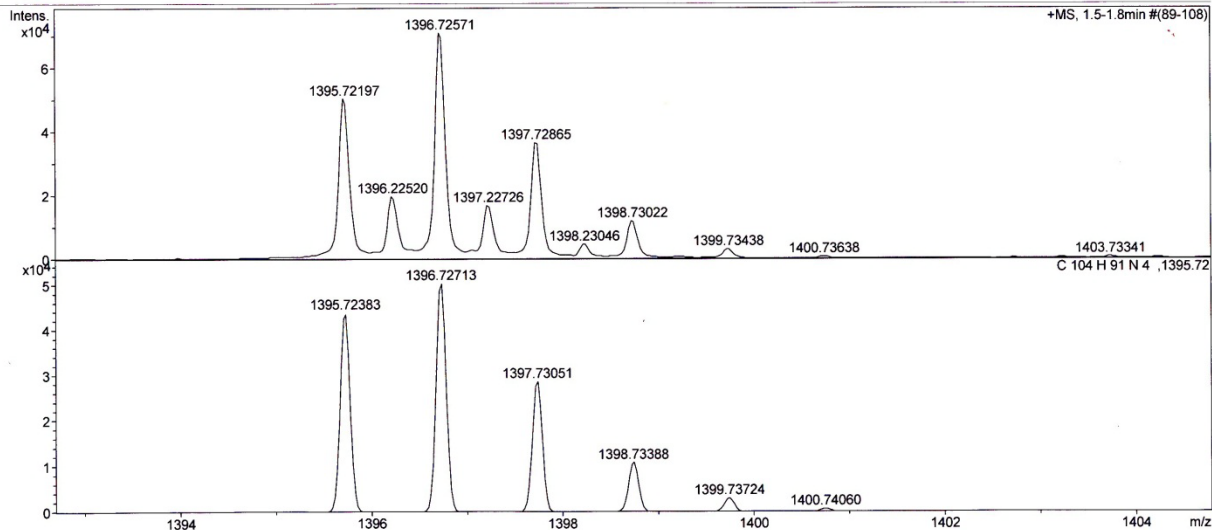
Analysis Name
Method tune_pos_wide.m
Sample Name
Comment ACN-Tol

Acquisition Date

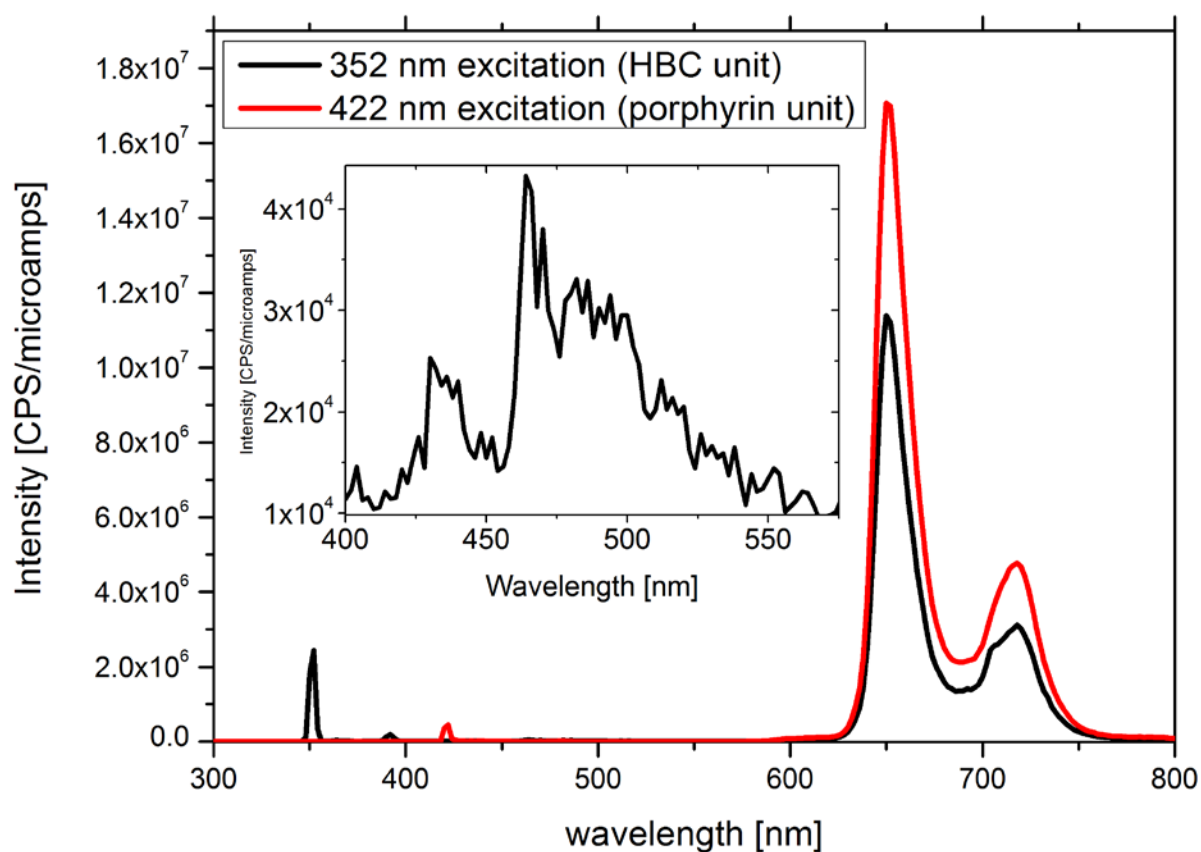
Operator
Instrument / Ser# micrOTOF 10364

Acquisition Parameter

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Scan End	3000 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste



UV/Vis (THF): $\lambda_{\text{max}}(\epsilon[M^{-1}cm^{-1}]) = 353 (64400), 426 (208400), 516 (11000), 552 (6000), 591 (2900), 648 (2600) \text{ nm.}$



Emission spectra of HBC-porphyrin **4** for different excitation wavelengths are shown, which excite either the HBC (352 nm) or the porphyrin (422 nm) unit of the conjugate. In both cases strong emission of the porphyrin is observed. When the HBC unit is excited emission from the HBC is completely suppressed as deduced from the zoomed region of the expected emission in the inset.