

Supporting Information

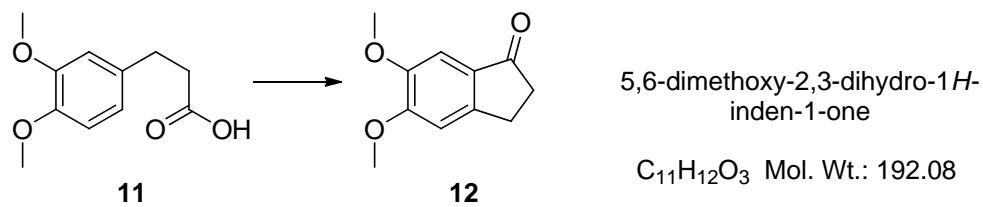
Enantioselective Total Synthesis of (+)-Brazilin, (-)-Brazilein and (+)-Brazilide A

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General Experimental: Proton nuclear magnetic resonance (¹H-NMR) spectra were recorded on Bruker Avance 300 and 400 spectrometer at 300 and 400 MHz. Carbon-13 nuclear magnetic resonance (¹³C-NMR) was recorded on Bruker Avance 300 and 400 spectrometer at 75 and 100 MHz. Chemical shifts are reported as δ values in parts per million (ppm) relative to tetramethylsilane (TMS) for all recorded NMR spectra. Low-resolution Mass spectra were recorded on a VG Auto Spec-3000 magnetic sector MS spectrometer. High Resolution Mass spectra were taken on AB QSTAR Pulsar mass spectrometer. The infrared (IR) spectra were acquired as thin films on a FT-IR spectrometer and the absorption frequencies are reported in cm⁻¹. Chiral HPLC analyses were performed on Agilent 1100 series with a tunable UV detector at wavelength $\lambda = 254$ nm. Melting points were determined on a capillary melting point apparatus and are uncorrected. Optical rotations were obtained on a UV-210A spectrometer. Starting materials and reagents used in reactions were obtained commercially from Acros, Aldrich, Fluka and were used without purification, unless otherwise indicated. THF and diethyl ether used in the reactions were dried by distillation over metallic sodium and benzophenone; dichloromethane were distilled over P₂O₅. Unless otherwise stated, all reactions were conducted in dried glassware under a positive pressure of dry nitrogen or argon. Silica gel (Qingdao, 300-400 mesh) was used for column chromatography.

Synthesis of compound 12



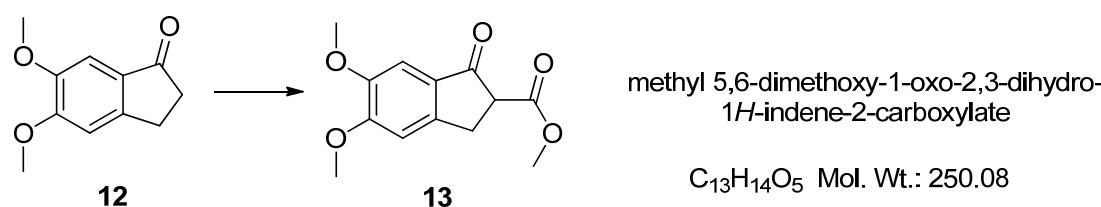
To a solution of 3-(3,4-dimethoxyphenyl)propanoic acid **11** (21 g, 0.1 mol) in anhydrous dichloromethane (300 mL) and DMF (0.5 mL) was added oxalyl chloride (24 mL, 0.25 mol) dropwise over a period of 1 hour. The mixture was then stirred at room temperature overnight. The resulting mixture was concentrated under reduced pressure and the crude aryl propionyl chloride was redissolved in anhydrous dichloromethane (300 mL) at 0 °C. To the solution of aryl propionyl

chloride was added portionwise a powder of AlCl₃ (16 g, 0.12 mol). The reaction mixture was then stirred at room temperature for 2 h. Ice (20 g) was added slowly at 0 °C (to quench the excess AlCl₃) followed by ice-water (100 mL). The organic phase was separated, and the aqueous phase was extracted with dichloromethane (3 × 100 mL). The combined organic phases were washed with brine, dried over Na₂SO₄, decolorized with activated carbon, filtered, and concentrated to afford the product (**12**, 18.6 g, 97%) as pale yellow solid.¹

Ref. 1: S. R. Haadsma-Svensson, K. A. Cleek, D. M. Dinh, J. N. Duncan, C. L. Haber, R. M. Huff, M. E. Lajiness, N. F. Nichols, M. W. Smith, K. A. Svensson, M. J. Zaya, A. Carlsson, C. H. Lin, *J. Med. Chem.* **2001**, *44*, 4716.

m.p.: 116-118 °C. IR: ν_{max} (KBr) cm⁻¹: 3055, 2928, 2849, 1699, 1592, 1497, 1450, 1313, 1258, 1116, 1037, 896, 851, 813, 708. ¹H-NMR (400 MHz, CDCl₃) δ: 7.17 (1H, s), 6.89 (1H, s), 3.96 (3H, s), 3.90 (3H, s), 3.05 (2H, t, *J* = 5.4 Hz), 2.68-2.65 (2H, m). ¹³C-NMR (100 MHz, CDCl₃) δ: 205.77, 155.39, 150.43, 149.37, 129.91, 107.47, 104.16, 56.23, 56.09, 36.53, 25.58. ES+ *m/z* (%): 192 (M⁺, 100), 177 (27), 164 (12), 149 (32), 135 (13), 121 (67), 107 (34), 91 (47), 77 (81), 63 (59), 51 (69). HRMS *m/z* Found: 192.0793, Calcd. for C₁₁H₁₂O₃ (M)⁺: 192.0786.

Synthesis of compound **13**

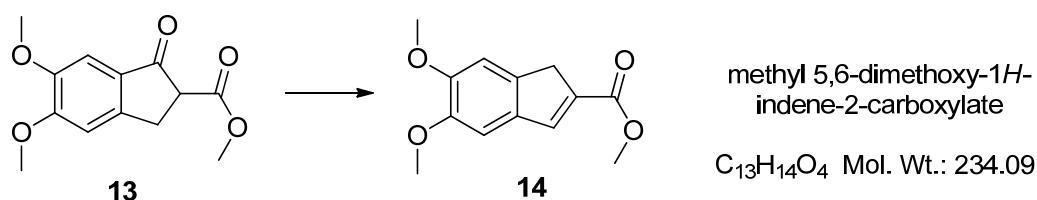


To a solution of 1-indanone **12** (9.6 g, 50 mmol) in dimethylcarbonate (80 mL) was added NaH (3.42 g, 100 mmol, 70% in paraffin), and the mixture was stirred at 90 °C for 3-4 h. The resulting solid was cooled with an ice-water bath at 0 °C. Dichloromethane (200 mL) was added, followed by aqueous solution of HCl (6M, 18 mL) and H₂O (100 mL). The organic phase was separated, and the aqueous layer was extracted with dichloromethane (3 × 100 mL). The combined organic phases were washed with brine, dried over Na₂SO₄ and concentrated. The desired product (**13**, 11.5 g, 92%) was obtained as pale yellow solid by flash column chromatography (Hexane : EtOAc = 8 : 1 → 4 : 1) on silica gel.²

Ref. 2: C. X. Pan, X. H. Zeng, Y. F. Guan, X. L. Jiang, L. Li, H. Zhang, *Synlett*, **2011**, 425.

m.p.: 161-162 °C. IR: ν_{max} (KBr) cm⁻¹: 3068, 2929, 2848, 1704, 1591, 1506, 1452, 1314, 1264, 1207, 1115, 1025, 956, 876, 723. ¹H-NMR (400 MHz, CDCl₃) δ: 7.16 (1H, s), 6.90 (1H, s), 3.96 (3H, s), 3.89 (3H, s), 3.77 (3H, s), 3.71 (1H, dd, *J* = 3.4, 7.9 Hz), 3.44 (1H, dd, *J* = 3.3, 17.0 Hz), 3.26 (1H, dd, *J* = 7.9, 17.0 Hz). ¹³C-NMR (100 MHz, CDCl₃) δ: 197.94, 169.90, 156.11, 149.78, 149.27, 127.92, 107.26, 104.83, 56.34, 56.15, 53.41, 52.76, 30.02. ES+ *m/z* (%): 250 (M⁺, 16), 218 (18), 190 (34), 175 (11), 125 (15), 111 (28), 97 (38), 85 (40), 83 (40), 77 (18), 71 (52), 57 (100), 55 (72). HRMS *m/z* Found: 250.0852, Calcd. for C₁₃H₁₄O₅ (M)⁺: 250.0841.

Synthesis of compound **14**

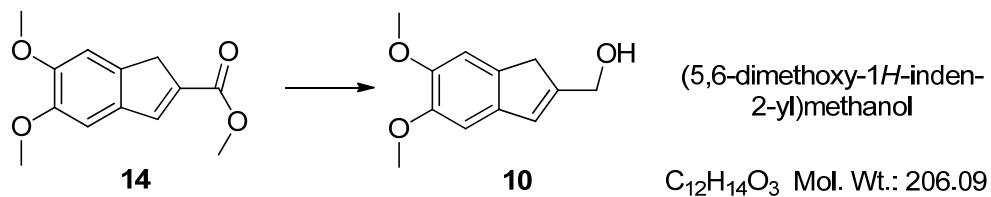


1-Indanone-2-carboxylic acid methyl ester **13** (5.0 g, 20 mmol) was dissolved in dichloromethane (100 mL) and methanol (10 mL). NaBH₄ (912 mg, 24 mmol) was added portionwise at 0 °C and the mixture was stirred at room temperature for 1 h. A solution of aqueous HCl (1 M, 30 mL) was added. The organic phase was separated and the aqueous phase was extracted with dichloromethane (3 × 50 mL). The combined organic phases were washed with brine, dried over Na₂SO₄ and concentrated. The crude product was re-dissolved in toluene (100 mL) and PPTS (Pyridinium *p*-toluenesulfonate, 0.5 g, 2 mmol) was added. The resulting mixture was stirred at reflux with a Dean-Stark trap for 3-6 h. The reaction progress was monitored by TLC. The solvent was removed under reduced pressure and the residue was dissolved in dichloromethane (250 mL) and washed with brine, dried over Na₂SO₄ and concentrated. Flash column chromatography on silica gel (Hexane : DCM = 1:1) afforded the product (**14**, 3.51 g, 75%) as white solid.²

Ref. 2: C. X. Pan, X. H. Zeng, Y. F. Guan, X. L. Jiang, L. Li, H. Zhang, *Synlett*, **2011**, 425.

m.p.: 122-123 °C. IR: ν_{max} (KBr) cm⁻¹: 3065, 2949, 2838, 1696, 1604, 1558, 1485, 1438, 1320, 1229, 1092, 989, 849, 737. ¹H-NMR (400 MHz, CDCl₃) δ: 7.64 (1H, s), 7.06 (1H, s), 7.02 (1H, s), 3.92 (3H, s), 3.91 (3H, s), 3.82 (3H, s), 3.61 (2H, s). ¹³C-NMR (100 MHz, CDCl₃) δ: 165.37, 149.77, 148.69, 141.48, 138.32, 135.35, 135.25, 107.54, 105.95, 56.12, 51.50, 38.37. ES+ *m/z* (%): 234 (M⁺, 76), 219 (24), 203 (24), 191 (16), 175 (52), 160 (26), 145 (20), 131 (93), 117 (58), 89 (100), 83 (81), 75 (27), 63 (72), 59 (67). HRMS *m/z* Found: 234.0893, Calcd. for C₁₃H₁₄O₄ (M)⁺ : 234.0892.

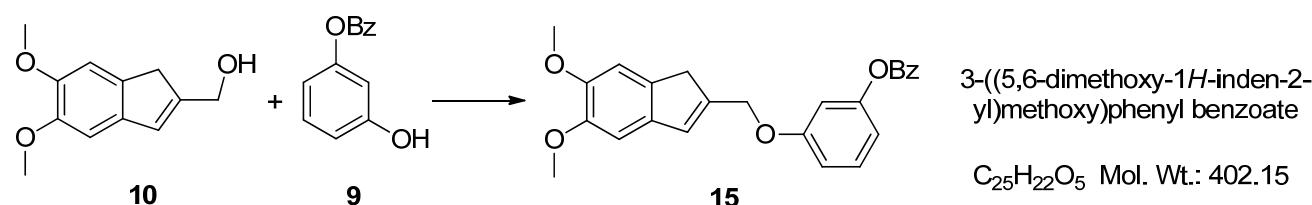
Synthesis of compound **10**



To a solution of methyl ester **14** (5.0 g, 21.4 mmol) in dichloromethane (100 mL) at -78 °C under argon was added dropwise a solution of DIBAL-H (31.4 mL, 1.5 M in toluene, 47.1 mmol). After 3 h, the reaction was quenched slowly with MeOH (5.0 mL) at -78 °C followed by addition of a saturated aqueous solution of NaHCO₃ (27 mL). The reaction mixture was allowed to reach room temperature and filtered through celite. The organic phase was separated and the aqueous phase was extracted with dichloromethane (3 × 50 mL). The combined organic phases were washed with brine (80 mL), dried over Na₂SO₄ and concentrated. Flash column chromatography on silica gel (Hexane : DCM = 2:1) afforded product **10** (4.01 g, 91%) as white solid.

m.p.: 105–106 °C. IR: ν_{max} (KBr) cm^{-1} : 3356, 3064, 2927, 2837, 1610, 1485, 1320, 1267, 1215, 1101, 1026, 992, 856, 759. $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ : 6.99 (1H, s), 6.87 (1H, s), 6.61 (1H, s), 4.51 (2H, s), 3.87 (6H, s), 3.33 (2H, s), 2.05 (1H, br, s). $^{13}\text{C-NMR}$ (100 MHz, CDCl_3) δ : 148.19, 147.36, 147.01, 137.22, 135.84, 127.54, 108.09, 104.54, 61.75, 56.24, 56.11, 38.93. ES+ m/z (%): 206 (M^+ , 67), 191 (22), 175 (37), 161 (27), 145 (36), 131 (52), 115 (80), 103 (78), 91 (100), 83 (44), 77 (74), 63 (87), 51 (70). HRMS m/z Found: 206.0945, Calcd. for $\text{C}_{12}\text{H}_{14}\text{O}_3$ (M^+): 206.0943.

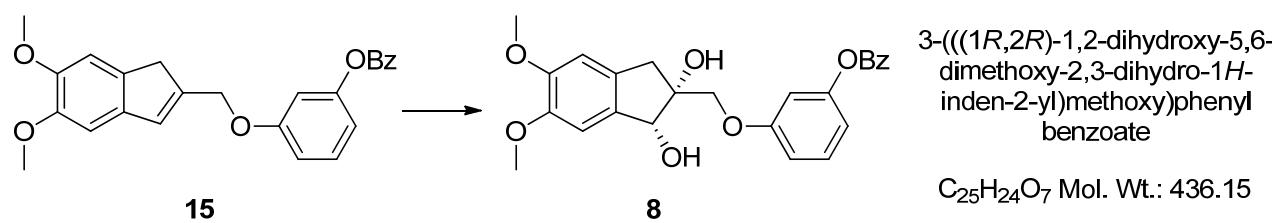
Synthesis of compound 15



1-Indene-2-methanol **10** (1.03 g, 5.0 mmol), triphenylphosphine (1.58 g, 6.0 mmol) and 3-hydroxyphenyl benzoate **9** (1.18 g, 5.5 mmol) were dissolved in dry THF (80 mL) then cooled to –78 °C. To this mixture was added dropwise a solution of diethyl azodicarboxylate (1.0 mL, 6.0 mmol) in dry THF (5 mL) over 30 min. The resulting yellow solution was then allowed to warm up and stirred at room temperature for 1 h. The reaction progress was monitored by TLC. The solvent was removed under reduced pressure and the residue was chromatographed on silica gel (Hexane : DCM = 3:1→1:1) to give the product (**15**, 1.21 g, 60%) as white solid.

m.p.: 121–122 °C. IR: ν_{max} (KBr) cm^{-1} : 3064, 2991, 2932, 2831, 1736, 1603, 1487, 1313, 1261, 1144, 1100, 1012, 878, 778, 707. $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ : 8.21 (2H, *d*, J = 7.2 Hz), 7.64 (1H, *t*, J = 7.2 Hz), 7.52 (2H, *t*, J = 7.8 Hz), 7.33 (1H, *t*, J = 8.2 Hz), 7.05 (1H, *s*), 6.93 (1H, *s*), 6.89 (1H, *dd*, J = 2.0, 8.4 Hz), 6.86 (1H, *t*, J = 2.0 Hz), 6.83 (1H, *dd*, J = 2.0, 8.4 Hz), 6.78 (1H, *s*), 4.95 (2H, *s*), 3.90 (6H, *s*), 3.46 (2H, *s*). $^{13}\text{C-NMR}$ (100 MHz, CDCl_3) δ : 165.12, 159.71, 151.90, 148.27, 147.29, 142.46, 136.96, 136.00, 133.64, 130.20, 129.93, 129.77, 129.53, 128.60, 114.14, 112.54, 108.60, 108.04, 104.71, 67.01, 56.26, 56.13, 39.37. ES+ m/z (%): 402 (M^+ , 5), 250 (10), 234 (23), 206 (100), 191 (50), 175 (60), 161 (33), 145 (41), 131 (59), 117 (36), 105 (49), 91 (66), 77 (68), 63 (46). HRMS m/z Found: 402.1478, Calcd. for $\text{C}_{25}\text{H}_{22}\text{O}_5$ (M^+): 402.1467.

Synthesis of compound 8



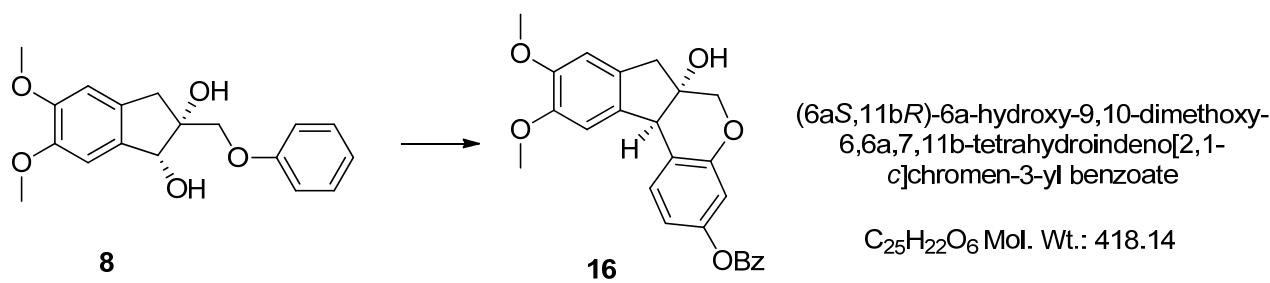
A 50 mL round-bottomed flask equipped with a magnetic stirring bar was charged with 1.4 g AD-mix- β ³ and methanesulfonyl amide (98 mg, 1.0 mmol). *tert*-Butanol (10 mL) and water (10 mL) were added and the slurry was

stirred at room temperature until all solids dissolved. The orange solution was then cooled to 0 °C. The 1-indene-2-benzoate (**15**, 402 mg, 1 mmol) and 3-hydroxyphenyl benzoate (**9**, 214 mg, 1.0 mmol) in dichloromethane (1 mL) were added. A solution of OsO₄ (0.13 mL, 20 mg/mL, 1% mmol) was added and the resulting mixture was vigorously stirred at room temperature for 3 days. The reaction progress was monitored by TLC. Saturated aqueous solution of Na₂S₂O₃ (13 mL) was added and the reaction was stirred for 15 min. The mixture was then extracted with ethyl acetate (3 × 30 mL). The combined organic phases were washed with brine (15 mL), dried over Na₂SO₄ and concentrated. Flash column chromatography on silica gel (Hexane : EtOAc = 3:1→2:1) afforded the product (**8**, 370 mg, 85%) as a yellow syrup. 80.5% ee, determined by HPLC on a Chiracel AS column using 2-propanol in hexane (30%) as the eluent.

Ref. 3: K. B. Sharpless, W. Amberg, Y. L. Bennani, G. A. Crispino, J. Hartung, K. S. Jeong, H. L. Kwong, K. Morikawa, Z. M. Wang, D. Xu, X. L. Zhang, *J. Org. Chem.*, **2011**, *76*, 2768.

IR: ν_{max} (KBr) cm⁻¹: 3443, 3072, 2924, 2848, 1732, 1603, 1497, 1455, 1320, 1264, 1143, 990, 872, 766, 708. ¹H-NMR (400 MHz, CDCl₃) δ: 8.19 (2H, *d*, *J* = 7.6 Hz), 7.64 (1H, *t*, *J* = 7.2 Hz), 7.51 (2H, *t*, *J* = 7.6 Hz), 7.34 (1H, *t*, *J* = 8.0 Hz), 6.96 (1H, *s*), 6.87-6.84 (3H, *m*), 6.75 (1H, *s*), 5.09 (1H, *d*, *J* = 6.9 Hz), 4.15 (1H, *d*, *J* = 9.3 Hz), 4.10 (1H, *d*, *J* = 9.2 Hz), 3.88 (3H, *s*), 3.87 (3H, *s*), 3.20 (1H, *s*), 3.08 (1H, *d*, *J* = 16.2 Hz), 3.05 (1H, *d*, *J* = 16.2 Hz), 2.80 (1H, *d*, *J* = 6.8 Hz). ¹³C-NMR (100 MHz, CDCl₃) δ: 165.17, 159.37, 151.90, 149.96, 148.90, 133.73, 133.39, 131.16, 130.19, 130.09, 129.37, 128.62, 114.61, 112.36, 108.53, 108.01, 107.76, 80.67, 77.25, 72.28, 56.07, 40.87. ES+ *m/z* (%): 436 (M⁺, 92), 418 (14), 400 (5), 277 (18), 222 (11), 205 (22), 189 (33), 151 (9), 105 (100), 91 (6), 77 (58). HRMS *m/z* Found: 436.1529, Calcd. for C₂₅H₂₄O₇ (M)⁺: 436.1522.

Synthesis of compound **16**

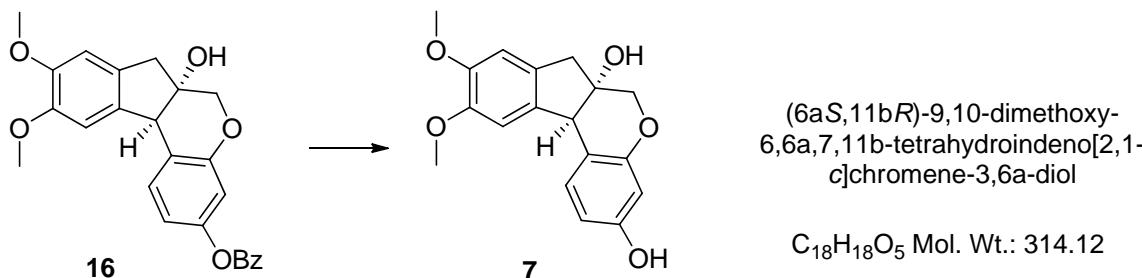


To a solution of diol **8** (930 mg, 2.13 mmol) in toluene (25 mL) was added pyridinium *p*-toluenesulfonate (PPTS, 643 mg, 2.56 mmol). The resulting mixture was stirred at 90 °C for 2 h. The solvent was removed under reduced pressure and the residue was dissolved in dichloromethane (100 mL) and washed with brine (20 mL), dried over Na₂SO₄ and concentrated. Flash column chromatography on silica gel (Hexane : EtOAc = 3:1→2:1) afforded the product (**16**, 713 mg, 80%) as yellow plates.

m.p.: 130-131 °C. IR: ν_{max} (KBr) cm⁻¹: 3509, 3052, 2926, 2836, 1716, 1599, 1499, 1454, 1319, 1274, 1130, 1032, 960, 889, 710. ¹H-NMR (400 MHz, CDCl₃) δ: 8.20 (2H, *d*, *J* = 7.8 Hz), 7.64 (1H, *t*, *J* = 7.4 Hz), 7.52 (2H, *t*, *J* = 7.6 Hz), 7.45 (1H, *d*, *J* = 8.4 Hz), 6.94 (1H, *dd*, *J* = 2.0, 8.4 Hz), 6.82 (1H, *dd*, *J* = 2.0, 7.2 Hz), 6.81 (1H, *s*), 6.75 (1H, *s*), 4.21 (1H, *s*), 4.08 (1H, *d*, *J* = 11.2 Hz), 3.87 (1H, *d*, *J* = 11.2 Hz), 3.86 (3H, *s*), 3.84 (3H, *s*), 3.28 (1H, *d*, *J* = 15.6 Hz), 2.91 (1H, *d*, *J* =

15.6 Hz), 2.51 (1H, s). ^{13}C -NMR (100 MHz, CDCl_3) δ : 165.15, 154.31, 150.35, 148.86, 148.46, 135.42, 133.69, 131.15, 130.58, 130.22, 129.41, 128.60, 119.99, 115.28, 110.94, 108.46, 107.64, 70.33, 56.15, 56.08, 50.78, 41.33. ES+ m/z (%): 418 (M^+ , 22), 400 (9), 295 (4), 267 (8), 239 (5), 211 (6), 194 (86), 165 (61), 118 (100), 105 (27). HRMS m/z Found: 418.1408, Calcd. for $\text{C}_{25}\text{H}_{22}\text{O}_6$ (M^+) 418.1416.

Synthesis of compound 7



To a solution of benzoate **16** (667 mg, 1.6 mmol) in THF (25 mL) was added a solution of LiOH (12.5 mL, 1.0 M in methanol, 20 mmol). The reaction mixture was then stirred at room temperature for 30 min until TLC showed complete conversion. After removal of the solvents, the residue was then diluted with EtOAc (50 mL), aq. HCl (19 mL, 1.0 M) and brine (30 mL) and the organic phase was then separated. The aqueous layer was extracted with EtOAc (3×30 mL). The combined organic phases were washed with brine (30 mL), dried over Na_2SO_4 and concentrated. Flash column chromatography on silica gel (Hexane : EtOAc = 1:1) afforded the product (**7**, 462 mg, 92%) as yellow crystals. After recrystallization (311 mg, 67%), the isomeric purity of phenol **7** was 99.8% ee, as determined by HPLC on a Chiracel AS column using 2-propanol in hexane (35%) as the eluent.

m.p.: 186–188 °C. $[\alpha]_D^{20} +125.5$ (*c* 1.00, MeOH). IR: ν_{max} (KBr) cm^{-1} : 3359, 3259, 3011, 2925, 2840, 1608, 1500, 1466, 1287, 1224, 1148, 1037, 960, 841, 764. ^1H -NMR (400 MHz, MeOD) δ : 7.25 (1H, *d*, *J* = 8.3 Hz), 6.86 (1H, *s*), 6.80 (1H, *s*), 6.49 (1H, *dd*, *J* = 1.4, 8.3 Hz), 6.30 (1H, *s*), 4.03 (1H, *s*), 3.94 (1H, *d*, *J* = 11.3 Hz), 3.77 (3H, *s*), 3.76 (3H, *s*), 3.70 (1H, *d*, *J* = 11.3 Hz), 3.08 (1H, *d*, *J* = 15.8 Hz), 2.89 (1H, *d*, *J* = 15.8 Hz). ^{13}C -NMR (100 MHz, MeOD) δ : 156.62, 154.40, 148.72, 148.35, 137.03, 131.37, 130.78, 113.86, 108.76, 108.69, 108.22, 102.94, 76.79, 69.41, 55.29, 55.21, 50.03, 41.70. ES+ m/z (%): 314 (M^+ , 7), 279 (3), 237 (7), 191 (4), 177 (6), 153 (17), 149 (31), 137 (29), 125 (12), 83 (17), 71 (46), 57 (100). HRMS m/z Found: 314.1148, Calcd. for $\text{C}_{18}\text{H}_{18}\text{O}_5$ (M^+) : 314.1154.

Synthesis of (+)-Brazilin (1)

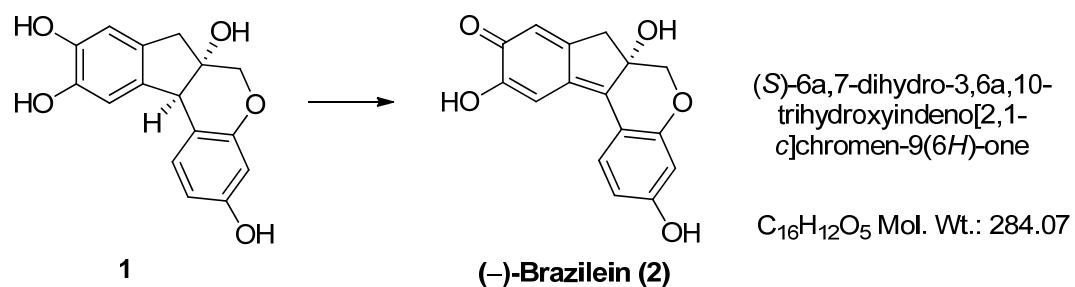


To a solution of compound **7** (30 mg, 0.1 mmol) in dry dichloromethane (10 mL) at -78 °C was added dropwise a solution of boron tribromide (0.09 mL, 4 N solution in dichloromethane, 0.35 mmol). The reaction mixture was then stirred at -78 °C for 2 h and stirred at room temperature overnight. The reaction was quenched by the addition of methanol (0.6 mL), then diluted with water (5 mL). The aqueous phase was extracted with EtOAc (3 × 10 mL). The combined organic phases were washed with brine (5 mL), dried over Na₂SO₄ and concentrated. Flash column chromatography on silica gel (Hexane : EtOAc = 1:1) afforded the natural product (**1**, 22 mg, 81%) as yellow solid.⁴

m.p.: 68-69 °C. [α]_D²⁰ +76.1 (c 1.00, MeOH), IR: ν_{max} (KBr) cm⁻¹: 3364, 2966, 2913, 1614, 1505, 1462, 1306, 1236, 1163, 1119, 1034, 979, 844, 771. ¹H-NMR (400 MHz, MeOD) δ: 7.19 (1H, d, J = 8.3 Hz), 6.71 (1H, s), 6.60 (1H, s), 6.47 (1H, dd, J = 2.4, 8.3 Hz), 6.30 (1H, d, J = 2.4 Hz), 3.96 (1H, s), 3.93 (1H, d, J = 11.3 Hz), 3.69 (1H, d, J = 11.3 Hz), 3.02 (1H, d, J = 15.6 Hz), 2.78 (1H, d, J = 15.6 Hz). ¹³C-NMR (100 MHz, MeOD) δ: 157.84, 155.71, 145.64, 145.32, 137.45, 132.23, 131.37, 115.59, 112.89, 112.47, 109.97, 104.27, 78.08, 70.86, 51.07, 42.90. ES+ m/z (%): 286 (M⁺, 39), 268 (16), 229 (6), 163 (3), 147 (3), 105 (6), 84 (82), 77 (5), 66 (100), 57 (18). HRMS m/z Found: 286.0844, Calcd. for C₁₆H₁₄O₅ (M)⁺ : 286.0841.

Ref. 4: a) C. F. J. Yamahara, T. Shimokawa, J. Kinjo, T. Tomimatsu, T. Nohara, *Phytochemistry* **1985**, 24, 2403; b) D. S. Kim, N. Baek, S. P. Oh, K. Y. Jung, I. S. Lee, H.-K. Lee, *Phytochemistry* **1997**, 46, 177.

Synthesis of (-)-Brazilein (**2**)

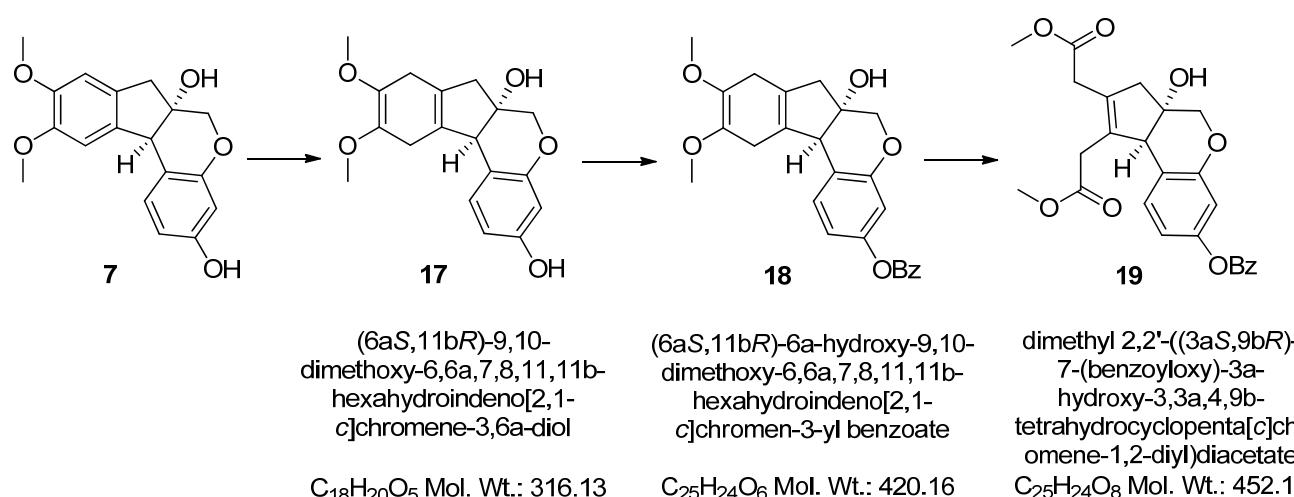


To a solution of (+)-Brazilin (**1**, 14.3 mg, 0.05 mmol) in anhydrous THF (3 mL) at 0 °C was added IBD (iodobenzene diacetate, 16.1 mg, 0.05 mmol). The reaction mixture was stirred for 10 min. The reaction progress was monitored by TLC. After removal of the solvent, the residue was chromatographed on silica gel (CHCl₃ : MeOH = 5:1) to afford the product (**2**, 10.8 mg, 76%) as brown solid.

m.p.: 260-262 °C. [α]_D²⁰ -871.0 (c 1.20, MeOH), ¹H-NMR (300 MHz, DMSO) δ: 7.81 (1H, d, J = 8.7 Hz), 7.10 (1H, s), 6.56 (1H, dd, J = 2.1, 8.7 Hz), 6.35 (1H, d, J = 2.1 Hz), 6.32 (1H, s), 5.66 (1H, br, s), 4.45 (1H, d, J = 11.7 Hz), 4.00 (1H, d, J = 11.7 Hz), 2.85 (2H, s). ¹³C-NMR (75 MHz, DMSO) δ: 179.29, 162.33, 158.92, 157.73, 152.34, 151.60, 130.53, 126.03, 117.48, 110.11, 110.81, 104.12, 102.88, 74.21, 72.94, 39.65.

Ref. 4: D. S. Kim, N. Baek, S. P. Oh, K. Y. Jung, I. S. Lee, H.-K. Lee, *Phytochemistry* **1997**, 46, 177.

Synthesis of compounds 17, 18 and 19



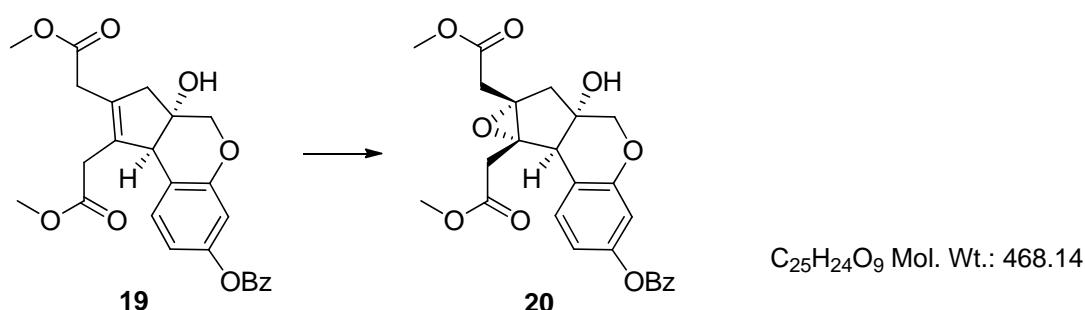
To a mixture of compound **7** (500 mg, 1.59 mmol), anhydrous THF (5 mL), anhydrous EtOH (5 mL), and liquid ammonia (150 mL) at -78°C under nitrogen was added lithium metal (445 mg, 63.6 mmol) in small pieces. The reaction mixture was stirred for 12 h at -78°C and then warmed to room temperature with evaporation of ammonia. Ice water (20 mL) was added slowly. The mixture was then extracted with EtOAc (4×10 mL), and the combined organic phases were washed with brine (20 mL), and concentrated to afford the product **17** as a yellow syrup. This diene product (**17**) was then dissolved in dry dichloromethane (20 mL) and Et₃N (0.44 mL, 3.18 mmol), DMAP (4-dimethylaminoypyridine, 19 mg, 0.159 mmol) and BzCl (0.22 mL, 1.91 mmol) were introduced at 0 °C. The reaction mixture was stirred for 20 min and the solvent was then removed under reduced pressure. After flash column chromatography on silica gel (Hexane : EtOAc = 3:1), the benzoate (**18**) was obtained as a yellow oil. The benzoate **18** was quickly dissolved in EtOAc (90 mL) and divided into three portions (30 mL/part). Each portion was diluted with EtOAc (10 mL) and MeOH (4 mL) then treated with pyridinium *p*-toluenesulfonate (PPTS, 133 mg, 0.53 mmol) at 0 °C. Ozone (O₃) was then bubbled through the mixture at 0 °C. The reaction progress was monitored by TLC. After ozonolysis, the three reaction mixtures were combined and treated with dimethyl sulfide (4 mL, 54.7 mmol) overnight at room temperature. The solvent was removed under reduced pressure and the residue was chromatographed on silica gel (Hexane : EtOAc = 8:1 → 5:1 → 3:1) to afford product **19** (359 mg, 50% for 3 steps) as a yellow oil.

Compound 17: ¹H-NMR (300 MHz, MeOD) δ: 6.98 (1H, *d*, *J* = 8.1 Hz), 6.42 (1H, *dd*, *J* = 2.4, 8.4 Hz), 6.32 (1H, *d*, *J* = 2.4 Hz), 3.84 (1H, *d*, *J* = 11.1 Hz), 3.77 (1H, *d*, *J* = 11.1 Hz), 3.56 (7H, overlap, *s*), 2.95-2.86 (1H, *m*), 2.79-2.66 (3H, *m*), 2.53 (1H, *d*, *J* = 16.2 Hz), 2.31 (1H, *d*, *J* = 16.2 Hz). ¹³C-NMR (75 MHz, MeOD) δ: 160.36, 159.32, 140.39, 140.17, 137.20, 134.31, 132.85, 117.09, 112.52, 107.13, 80.15, 74.70, 60.39, 60.19, 56.68, 48.49, 31.81, 30.95.

Compound 18: ¹H-NMR (300 MHz, CDCl₃) δ: 8.18 (2H, *d*, *J* = 7.2 Hz), 7.63 (1H, *t*, *J* = 7.4 Hz), 7.50 (2H, *t*, *J* = 7.5 Hz), 7.17 (1H, *d*, *J* = 7.8 Hz), 6.83 (1H, *d*, *J* = 7.8 Hz), 6.81 (1H, *s*), 3.97 (1H, *d*, *J* = 11.1 Hz), 3.90 (1H, *d*, *J* = 11.1 Hz), 3.69 (1H, *s*), 3.62 (3H, *s*), 3.61 (3H, *s*), 2.93-2.78 (4H, *m*), 2.66 (1H, *d*, *J* = 16.2 Hz), 2.64 (1H, *s*), 2.33 (1H, *d*, *J* = 16.2 Hz). ¹³C-NMR (75 MHz, CDCl₃) δ: 165.13, 155.16, 150.10, 136.36, 136.12, 133.63, 132.79, 130.64, 130.19, 129.74, 129.46, 128.57, 119.49, 114.93, 110.95, 76.81, 71.46, 57.57, 57.25, 53.45, 44.25, 28.60, 27.66.

Compound 19: $[\alpha]_D^{20} +28.0$ (c 1.13, Acetone), IR: ν_{max} (KBr) cm^{-1} : 3458, 3060, 2949, 2844, 1733, 1601, 1497, 1441, 1311, 1257, 1149, 1047, 887, 800, 709. $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ : 8.18 (2H, *d*, J = 7.8 Hz), 7.63 (1H, *t*, J = 7.4 Hz), 7.51 (2H, *t*, J = 7.7 Hz), 7.29 (1H, *d*, J = 8.2 Hz), 6.82 (1H, *dd*, J = 1.8, 8.4 Hz), 6.80 (1H, *s*), 4.02 (1H, *d*, J = 11.2 Hz), 3.96 (1H, *d*, J = 11.6 Hz), 3.93 (1H, *s*), 3.68 (3H, *s*), 3.66 (3H, *s*), 3.21-3.07 (4H, *m*), 2.85 (1H, *d*, J = 16.2 Hz), 2.56 (1H, *d*, J = 4.1 Hz), 2.42 (1H, *d*, J = 16.3 Hz). $^{13}\text{C-NMR}$ (100 MHz, CDCl_3) δ : 170.94, 170.64, 165.09, 155.16, 150.38, 134.79, 133.65, 131.48, 131.35, 130.20, 129.43, 128.58, 118.46, 114.88, 110.84, 75.29, 70.57, 53.14, 52.13, 52.07, 44.68, 34.17, 31.98. ES+ m/z (%): 452 (M^+ , 6), 434 (15), 374 (5), 118 (11), 105 (24), 83 (100), 77 (13). HRMS m/z Found: 452.1482, Calcd. for $\text{C}_{25}\text{H}_{24}\text{O}_8$ (M^+): 452.1471.

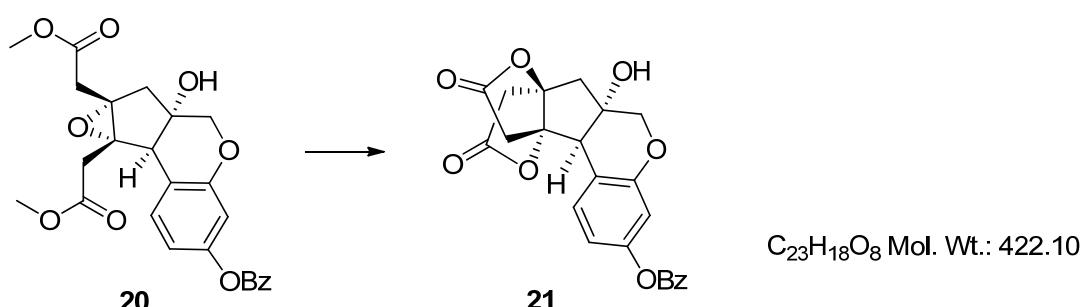
Synthesis of compound 20



To a solution of compound **19** (42 mg, 0.093 mmol) in dry dichloromethane (3 mL) was added *m*-CPBA (3-chloroperoxybenzoic acid, 25 mg, 77%, 0.112 mmol) at 0 °C. The reaction mixture was stirred for 30 min at 0 °C then at room temperature overnight. Saturated aqueous solution of NaHSO_3 (8 mL) was added. The mixture was extracted with dichloromethane (3 × 10 mL), and the combined organic phases were washed with saturated aqueous solution of NaHCO_3 (8 mL) and brine (8 mL), and concentrated. Flash column chromatography on silica gel (Hexane : EtOAc = 3:1) afforded the product (**20**, 40 mg, 92%) as a colorless oil.

$[\alpha]_D^{20} +46.1$ (c 2.88, Acetone). IR: ν_{max} (KBr) cm^{-1} : 3514, 3068, 2952, 1737, 1602, 1496, 1437, 1348, 1254, 1152, 1039, 939, 887, 708. $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ : 8.17 (2H, *d*, J = 7.5 Hz), 7.64 (1H, *t*, J = 7.4 Hz), 7.51 (2H, *t*, J = 7.7 Hz), 7.30 (1H, *d*, J = 8.4 Hz), 6.85 (1H, *dd*, J = 2.3, 8.4 Hz), 6.82 (1H, *d*, J = 2.3 Hz), 4.02 (1H, *d*, J = 10.7 Hz), 3.74 (3H, *s*), 3.68 (1H, *d*, J = 10.7 Hz), 3.61 (3H, *s*), 2.71 (1H, *s*), 3.05 (1H, *d*, J = 17.1 Hz), 2.71 (2H, *s*), 2.55 (1H, *d*, J = 17.1 Hz), 2.54 (1H, *d*, J = 14.9 Hz), 2.30 (1H, *d*, J = 14.9 Hz). $^{13}\text{C-NMR}$ (100 MHz, CDCl_3) δ : 170.10, 169.33, 164.96, 156.10, 150.62, 133.73, 130.47, 130.18, 129.31, 128.62, 118.01, 115.42, 111.43, 75.25, 70.95, 69.05, 52.33, 52.13, 51.67, 41.00, 35.99, 34.31. ES+ m/z (%): 468 (M^+ , 6), 268 (7), 256 (4), 239 (5), 156 (9), 139 (11), 125 (17), 118 (100), 111 (33), 105 (76), 97 (59). HRMS m/z Found: 468.1425, Calcd. for $\text{C}_{25}\text{H}_{24}\text{O}_9$ (M^+): 468.1420.

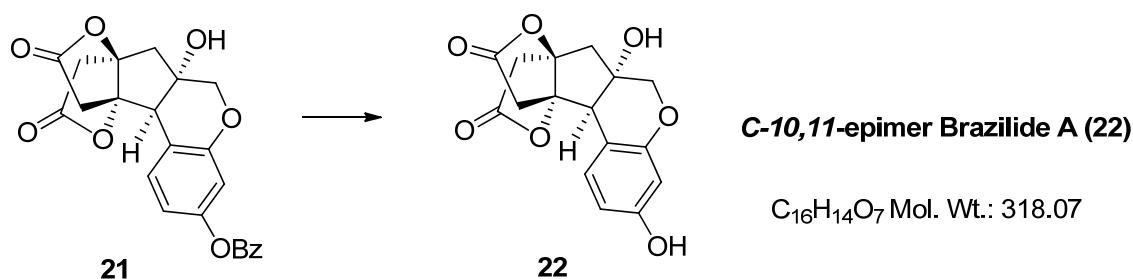
Synthesis of compound 21



To a solution of epoxide **20** (22 mg, 0.047 mmol) in dry dichloromethane (1 mL) was added a solution of $BF_3 \cdot Et_2O$ (boron trifluoride-diethyl etherate, 0.24 mL, 0.1 M in dichloromethane, 0.024 mmol) at 0 °C under nitrogen. The reaction mixture was then stirred for 30 min at 0 °C before stirring at room temperature overnight. A saturated aqueous solution of $NaHCO_3$ (5 mL) was added. The mixture was then extracted with dichloromethane (3×5 mL). The combined organic phases were washed with brine (5 mL) and concentrated. Flash column chromatography on silica gel (Hexane : EtOAc = 3:1→2:1) afforded the product (**21**, 16.7 mg, 84%) as white solid.

m.p.: 133–135 °C. $[\alpha]_D^{20} -16.8$ (*c* 1.14, Acetone), IR: ν_{max} (KBr) cm^{-1} : 3431, 3056, 2933, 1793, 1736, 1601, 1497, 1442, 1397, 1263, 1151, 1053, 992, 888, 814, 709. 1H -NMR (400 MHz, $CDCl_3$) δ : 8.17 (2H, *d*, *J* = 7.9 Hz), 7.66 (1H, *t*, *J* = 7.3 Hz), 7.53 (2H, *t*, *J* = 7.7 Hz), 7.22 (1H, *d*, *J* = 8.4 Hz), 6.95 (1H, *dd*, *J* = 1.6, 8.4 Hz), 6.89 (1H, *s*), 3.94 (1H, *d*, *J* = 11.8 Hz), 3.73 (1H, *d*, *J* = 11.8 Hz), 3.58 (1H, *s*), 3.34 (1H, *d*, *J* = 19.1 Hz), 3.16 (1H, *d*, *J* = 19.1 Hz), 2.88 (1H, *br*, *s*), 2.77 (1H, *d*, *J* = 19.6 Hz), 2.61 (1H, *d*, *J* = 14.4 Hz), 2.56 (1H, *d*, *J* = 19.2 Hz), 2.50 (1H, *d*, *J* = 15.3 Hz). ^{13}C -NMR (100 MHz, $CDCl_3$) δ : 173.01, 172.15, 165.19, 154.50, 151.32, 134.03, 130.63, 130.27, 128.92, 128.73, 117.26, 116.95, 111.80, 97.58, 92.36, 76.16, 70.61, 55.44, 45.21, 42.04, 38.60. ES+ *m/z* (%): 422 (M^+ , 3), 219 (2), 178 (4), 149 (4), 105 (84), 83 (100), 77 (31), 66 (30), 57 (32). HRMS *m/z* Found: 422.0996, Calcd. for $C_{23}H_{18}O_8$ (M^+) : 422.1022.

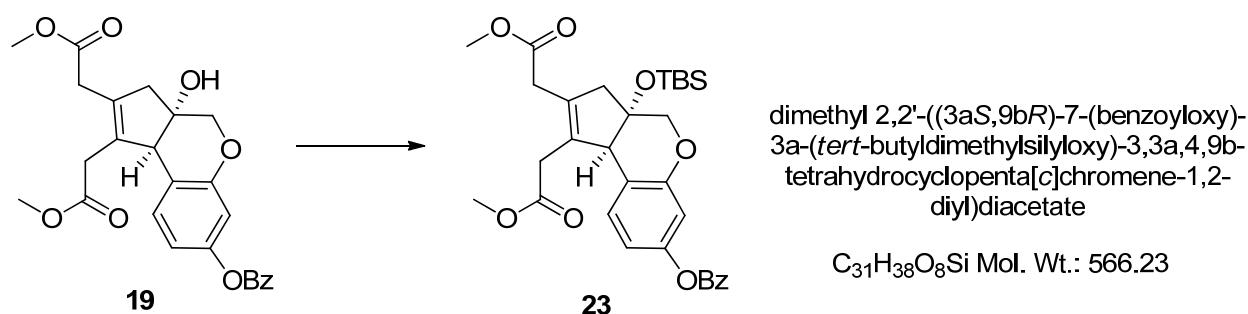
Synthesis of *C*-10,11-epimer of Brazilide A (**22**)



To a solution of bis-lactone **21** (19 mg, 0.045 mmol) in acetic acid (2 mL) was added 65% H_2SO_4 solution (0.14 mL). The reaction mixture was then stirred at 110 °C for 5 h, by which time TLC analysis showed no remaining starting material. After cooling to room temperature, EtOAc (5 mL) and water (5 mL) was added. After separation of the organic phase, the aqueous phase was extracted with EtOAc (3×10 mL). The combined organic layer was washed with brine (8 mL), dried over Na_2SO_4 and concentrated. Flash column chromatography on silica gel (Hexane : EtOAc = 3:1→2:1) afforded the product (**22**, 12.7 mg, 90%) as white solid.

m.p.: 127–128 °C. $[\alpha]_D^{20} +4.4$ (*c* 1.30, Acetone), IR: ν_{max} (KBr) cm^{-1} : 3407, 3077, 2987, 2931, 1782, 1621, 1507, 1462, 1401, 1324, 1250, 1154, 1041, 994, 845, 639. $^1\text{H-NMR}$ (400 MHz, Acetone-*d*₆) δ : 7.00 (1H, *d*, *J* = 8.4 Hz), 6.57 (1H, *dd*, *J* = 1.4, 8.3 Hz), 6.40 (1H, *s*), 3.89 (1H, *d*, *J* = 11.6 Hz), 3.76 (1H, *d*, *J* = 11.6 Hz), 3.55 (1H, *s*), 3.44 (1H, *d*, *J* = 18.8 Hz), 3.16 (1H, *d*, *J* = 18.8 Hz), 2.80 (1H, *d*, *J* = 19.3 Hz), 2.64 (1H, *d*, *J* = 14.8 Hz), 2.56 (1H, *d*, *J* = 19.3 Hz), 2.47 (1H, *d*, *J* = 14.8 Hz). $^{13}\text{C-NMR}$ (100 MHz, Acetone-*d*₆) δ : 174.08, 173.37, 158.78, 156.15, 131.69, 112.17, 111.21, 104.56, 98.65, 93.59, 76.84, 71.15, 55.97, 46.02, 42.44, 39.10. ES+ *m/z* (%): 318 (M⁺, 6), 192 (5), 161 (18), 147 (4), 131 (3), 118 (100), 105 (12), 97 (9). HRMS *m/z* Found: 318.0743, Calcd. for C₁₆H₁₄O₇ (M)⁺ : 318.0740.

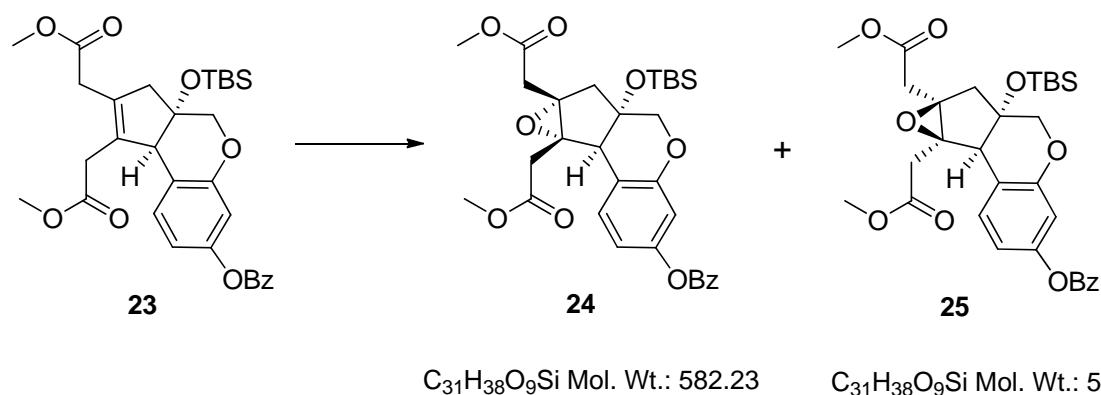
Synthesis of compound 23



To a solution of compound **19** (240 mg, 0.53 mmol) in anhydrous dichloromethane (20 mL) at 0 °C under nitrogen was added a solution of 2,6-Lutidine (0.31 mL, 2.65 mmol) followed by addition of *tert*-butyldimethylsilyl trifluoromethanesulfonate (TBSOTf, 0.52 mL, 2.12 mmol). The reaction mixture was then warmed up to room temperature and stirred at room temperature overnight. The reaction mixture was then cooled to 0 °C and quenched with brine (10 mL). The aqueous phase was extracted with dichloromethane (3 × 15 mL). The combined organic phases were washed with brine (15 mL), dried over Na₂SO₄ and concentrated. Flash column chromatography on silica gel (Hexane : EtOAc = 8:1) afforded the product (**23**, 255 mg, 85%) as a yellow oil.

$[\alpha]_D^{20} +46.2$ (*c* 1.24, Acetone), IR: ν_{max} (KBr) cm^{-1} : 3064, 2944, 2856, 1738, 1606, 1496, 1442, 1317, 1256, 1150, 1010, 939, 832, 777, 709. $^1\text{H-NMR}$ (300 MHz, CDCl₃) δ : 8.16 (2H, *d*, *J* = 7.8 Hz), 7.60 (1H, *t*, *J* = 7.2 Hz), 7.47 (2H, *t*, *J* = 7.5 Hz), 7.22 (1H, *d*, *J* = 7.8 Hz), 6.77 (1H, *d*, *J* = 8.7 Hz), 6.74 (1H, *s*), 3.96–3.89 (3H, *m*), 3.65 (3H, *s*), 3.63 (3H, *s*), 3.18–3.02 (4H, *m*), 2.78 (1H, *d*, *J* = 16.2 Hz), 2.37 (1H, *d*, *J* = 16.2 Hz), 0.79 (9H, *s*), 0.02 (3H, *s*), –0.03 (3H, *s*). $^{13}\text{C-NMR}$ (75 MHz, CDCl₃) δ : 170.92, 170.57, 165.01, 155.54, 150.29, 135.09, 133.49, 131.11, 130.14, 129.65, 128.53, 118.80, 114.11, 110.35, 70.69, 53.67, 51.95, 46.17, 34.29, 31.95, 25.63, 17.99, –2.85, –2.99. ES+ *m/z* (%): 565 (M⁺–1, 10), 551 (40), 535 (52), 509 (27), 477 (8), 449 (8), 375 (15), 285 (3), 271 (5), 211 (4), 105 (100), 77 (35). HRMS *m/z* Found: 566.2330, Calcd. for C₃₁H₃₈O₈Si (M)⁺ : 566.2336.

Synthesis of compounds 24 and 25

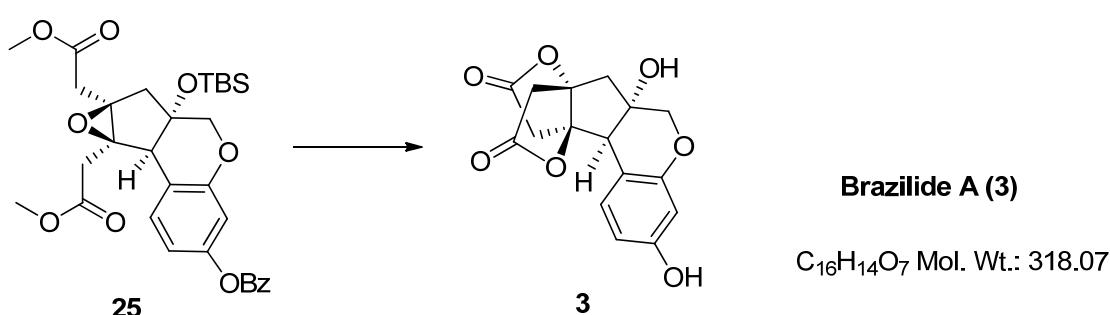


To a solution of compound **23** (246 mg, 0.435 mmol) in anhydrous dichloromethane (30 mL) was added *m*-CPBA (3-chloroperoxybenzoic acid, 97 mg, 77%, 0.435 mmol) at 0 °C. The reaction mixture was stirred for 30 min at 0 °C then warmed up to 32 °C. The reaction mixture was then stirred at 32 °C for 3 days and during this period further *m*-CPBA was added every 12 h (6 × 49 mg, 77%, 6 × 0.22 mmol). Saturated aqueous solution of NaHSO₃ (20 mL) was added. The mixture was extracted with dichloromethane (3 × 25 mL), and the combined organic phases were washed with saturated aqueous solution of NaHCO₃ (15 mL), brine (15 mL), dried over Na₂SO₄ and concentrated. Flash column chromatography on silica gel (Hexane : Acetone = 30:1→20:1) afforded the epoxides (**24**, 160 mg, 63%; **25**, 54 mg, 22%) as colorless oil.

Compound 24: $[\alpha]_D^{20} +11.9$ (*c* 1.51, Acetone), IR: ν_{\max} (KBr) cm⁻¹: 3068, 2946, 2844, 1738, 1606, 1497, 1440, 1321, 1256, 1150, 1049, 939, 832, 777, 708. ¹H-NMR (400 MHz, CDCl₃) δ: 8.17 (2H, *d*, *J* = 7.7 Hz), 7.63 (1H, *t*, *J* = 7.4 Hz), 7.50 (2H, *t*, *J* = 7.7 Hz), 7.27 (1H, *d*, *J* = 8.3 Hz), 6.82 (1H, *dd*, *J* = 2.2, 8.3 Hz), 6.76 (1H, *s*), 4.08 (1H, *d*, *J* = 11.2 Hz), 3.76 (1H, *d*, *J* = 11.2 Hz), 3.69 (3H, *s*), 3.52 (3H, *s*), 3.44 (1H, *s*), 2.81 (1H, *d*, *J* = 17.1 Hz), 2.75 (1H, *d*, *J* = 16.3 Hz), 2.61 (1H, *d*, *J* = 16.3 Hz), 2.55 (1H, *d*, *J* = 17.1 Hz), 2.39 (1H, *d*, *J* = 14.6 Hz), 2.28 (1H, *d*, *J* = 14.6 Hz), 0.79 (9H, *s*), 0.07 (3H, *s*), 0.06 (3H, *s*). ¹³C-NMR (100 MHz, CDCl₃) δ: 169.98, 169.90, 164.92, 155.76, 150.77, 133.63, 131.66, 130.15, 129.46, 128.59, 117.35, 114.72, 110.49, 78.90, 71.10, 70.84, 68.22, 52.05, 51.59, 42.57, 36.95, 34.76, 25.61, 18.01, -2.65, -2.74. ES+ *m/z* (%): 582 (M⁺, 12), 581 (M⁺-1, 23), 567 (82), 525 (100), 509 (18), 493 (25), 465 (21), 433 (32), 373 (25), 325 (25), 173 (18), 105 (53), 89 (12), 77 (34). HRMS *m/z* Found: 582.2274, Calcd. for C₃₁H₃₈O₉Si (M)⁺: 582.2285.

Compound 25: $[\alpha]_D^{20} -8.1$ (*c* 1.51, Acetone), IR: ν_{\max} (KBr) cm⁻¹: 3068, 2948, 2856, 1739, 1608, 1498, 1442, 1317, 1256, 1148, 1049, 1017, 833, 778, 708. ¹H-NMR (400 MHz, CDCl₃) δ: 8.20 (2H, *d*, *J* = 7.6 Hz), 7.63 (1H, *t*, *J* = 7.4 Hz), 7.51 (2H, *t*, *J* = 7.7 Hz), 7.28 (1H, *d*, *J* = 8.4 Hz), 6.84 (1H, *dd*, *J* = 2.2, 8.4 Hz), 6.79 (1H, *d*, *J* = 2.2 Hz), 4.00 (1H, *d*, *J* = 11.4 Hz), 3.93 (1H, *d*, *J* = 11.4 Hz), 3.81 (3H, *s*), 3.74 (3H, *s*), 3.43 (1H, *s*), 3.39 (1H, *d*, *J* = 16.7 Hz), 2.84 (1H, *d*, *J* = 16.3 Hz), 2.70 (1H, *d*, *J* = 16.3 Hz), 2.39 (1H, *d*, *J* = 16.7 Hz), 2.32 (1H, *d*, *J* = 14.7 Hz), 2.27 (1H, *d*, *J* = 14.7 Hz), 0.79 (9H, *s*), 0.00 (3H, *s*), -0.01 (3H, *s*). ¹³C-NMR (100 MHz, CDCl₃) δ: 169.95, 169.88, 165.03, 155.64, 150.47, 133.57, 130.87, 130.19, 129.57, 128.56, 115.48, 114.25, 110.49, 73.32, 73.06, 67.10, 64.34, 52.30, 52.05, 46.47, 41.30, 36.19, 34.63, 25.59, 17.93, -2.95, -3.15. ES+ *m/z* (%): 582 (M⁺, 9), 581 (M⁺-1, 16), 567 (17), 559 (22), 525 (43), 505 (21), 495 (59), 463 (100), 435 (38), 421 (18), 391 (9), 375 (14), 330 (13), 313 (21), 271 (13), 192 (16), 105 (67), 77 (24). HRMS *m/z* Found: 582.2285, Calcd. for C₃₁H₃₈O₉Si (M)⁺: 582.2285.

Synthesis of (+)-Brazilide A (3)



To a stirred solution of epoxide **25** (37 mg, 0.064 mmol) in anhydrous dichloromethane (5 mL) was added a solution of $BF_3 \cdot Et_2O$ (boron trifluoride-diethyl etherate, 0.32 mL, 0.1 M in dichloromethane, 0.032 mmol) at 0 °C under nitrogen. The reaction mixture was stirred for 30 min at 0 °C then stirred at room temperature overnight. A saturated aqueous solution of $NaHCO_3$ (8 mL) was added. The resulting mixture was extracted with dichloromethane (3×15 mL), and the combined organic phases were washed with brine (8 mL) and concentrated. The residue was re-dissolved in acetic acid (3 mL), and a solution of 65% H_2SO_4 (0.2 mL) was added. The reaction mixture was stirred at 110 °C for 5 h, by which time TLC analysis showed no remaining starting material. After cooling to room temperature, $EtOAc$ (5 mL) and water (5 mL) was added. The organic phase was separated. The aqueous phase was extracted with $EtOAc$ (3×10 mL). The combined organic phases were washed with brine (8 mL), dried over Na_2SO_4 and concentrated. Flash column chromatography on silica gel (Hexane : $EtOAc$ = 3:1→2:1) afforded the product (**2**, 16.6 mg, 82%) as white solid.⁵

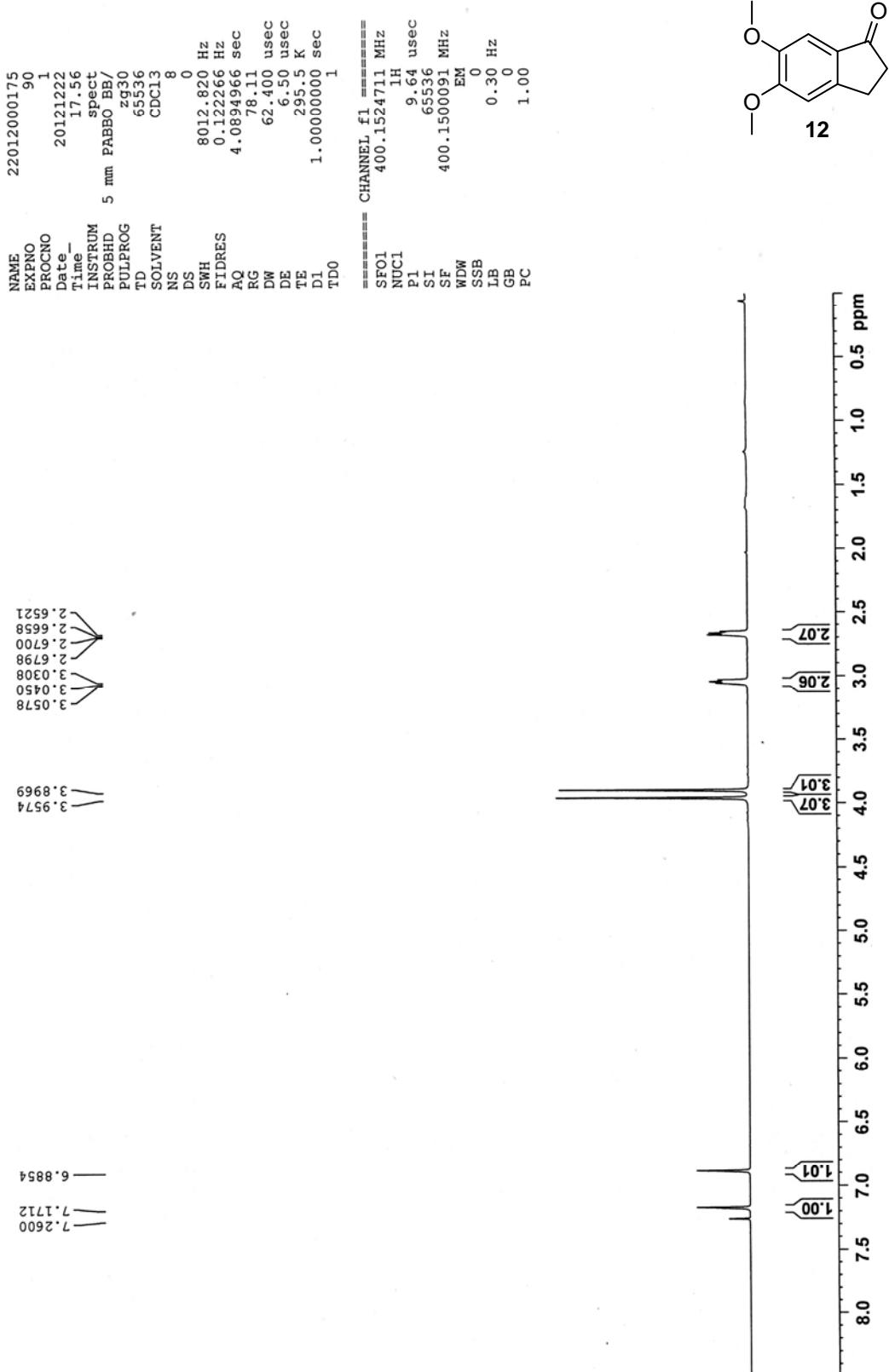
m.p.: 252–253 °C. $[\alpha]_D^{20} +3.4$ (c 1.36, Acetone), IR: ν_{max} (KBr) cm^{-1} : 3389, 2921, 2854, 1792, 1622, 1507, 1457, 1328, 1283, 1219, 1161, 1096, 1037, 996, 842, 713. 1H -NMR (400 MHz, Acetone- d_6) δ : 7.12 (1H, d, J = 8.6 Hz), 6.46 (1H, dd, J = 2.4, 8.5 Hz), 6.31 (1H, d, J = 2.4 Hz), 3.90 (1H, d, J = 10.9 Hz), 3.82 (1H, d, J = 10.9 Hz), 3.67 (1H, s), 3.53 (1H, d, J = 18.7 Hz), 3.23 (1H, d, J = 18.7 Hz), 3.00 (1H, d, J = 19.2 Hz), 2.65 (1H, d, J = 19.2 Hz), 2.62 (1H, d, J = 15.4 Hz), 2.43 (1H, d, J = 15.6 Hz). ^{13}C -NMR (100 MHz, Acetone- d_6) δ : 174.12, 173.73, 158.47, 155.77, 134.52, 110.13, 109.51, 104.00, 96.52, 94.63, 78.52, 69.80, 54.30, 46.02, 44.45, 42.94. ES+ m/z (%): 318 (100), 300 (10), 232 (5), 174 (5), 163 (26), 147 (15), 105 (20), 91 (16), 77 (16), 57 (16). HRMS m/z Found: 318.0740, Calcd. for $C_{16}H_{14}O_7$ (M^+): 318.0740.

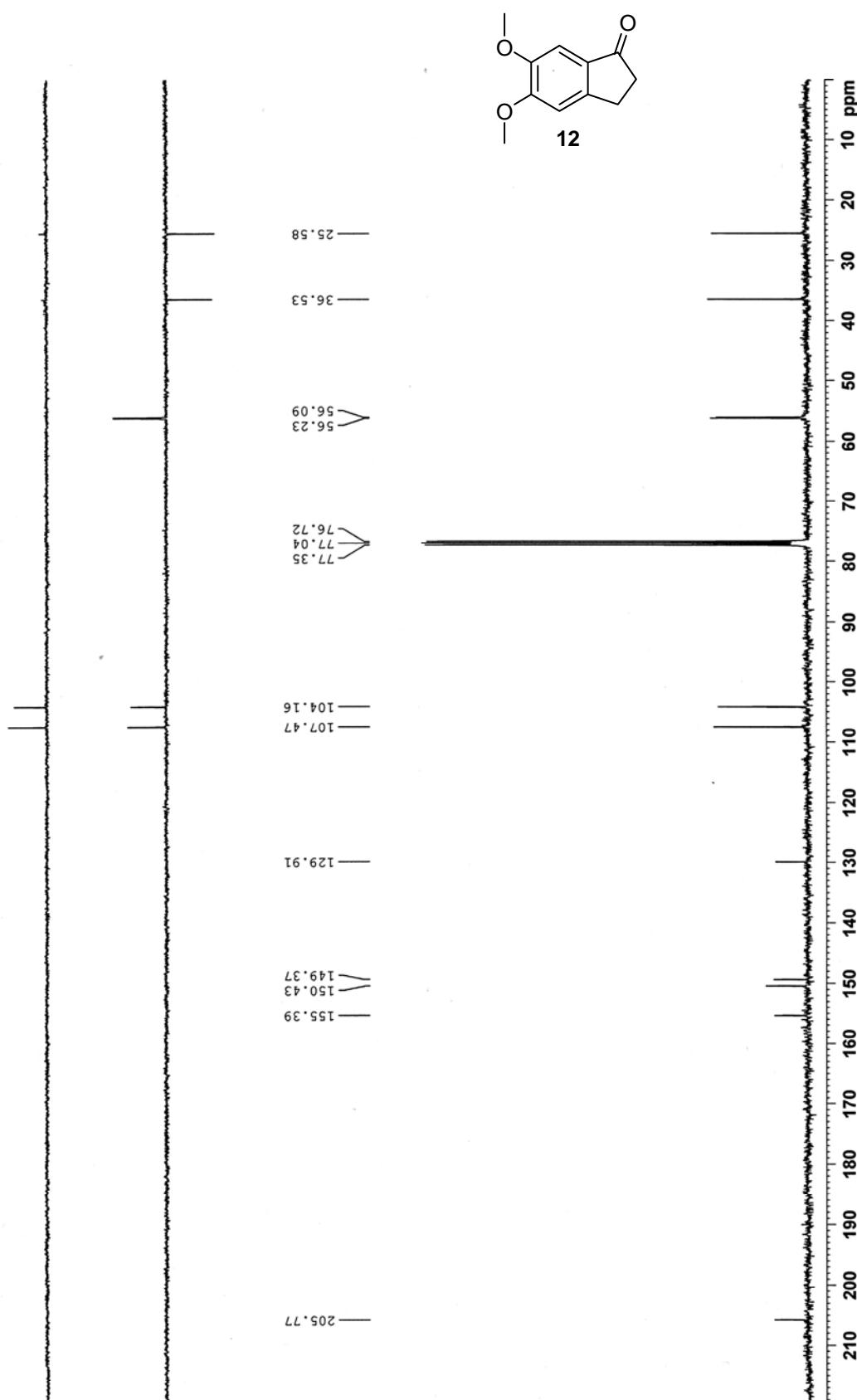
Ref. 5: B. O. Yang, C.-Q. Ke, Z.-S. He, Y.-P. Yang, Y. Ye, *Tetrahedron Lett.* **2002**, *43*, 1731.

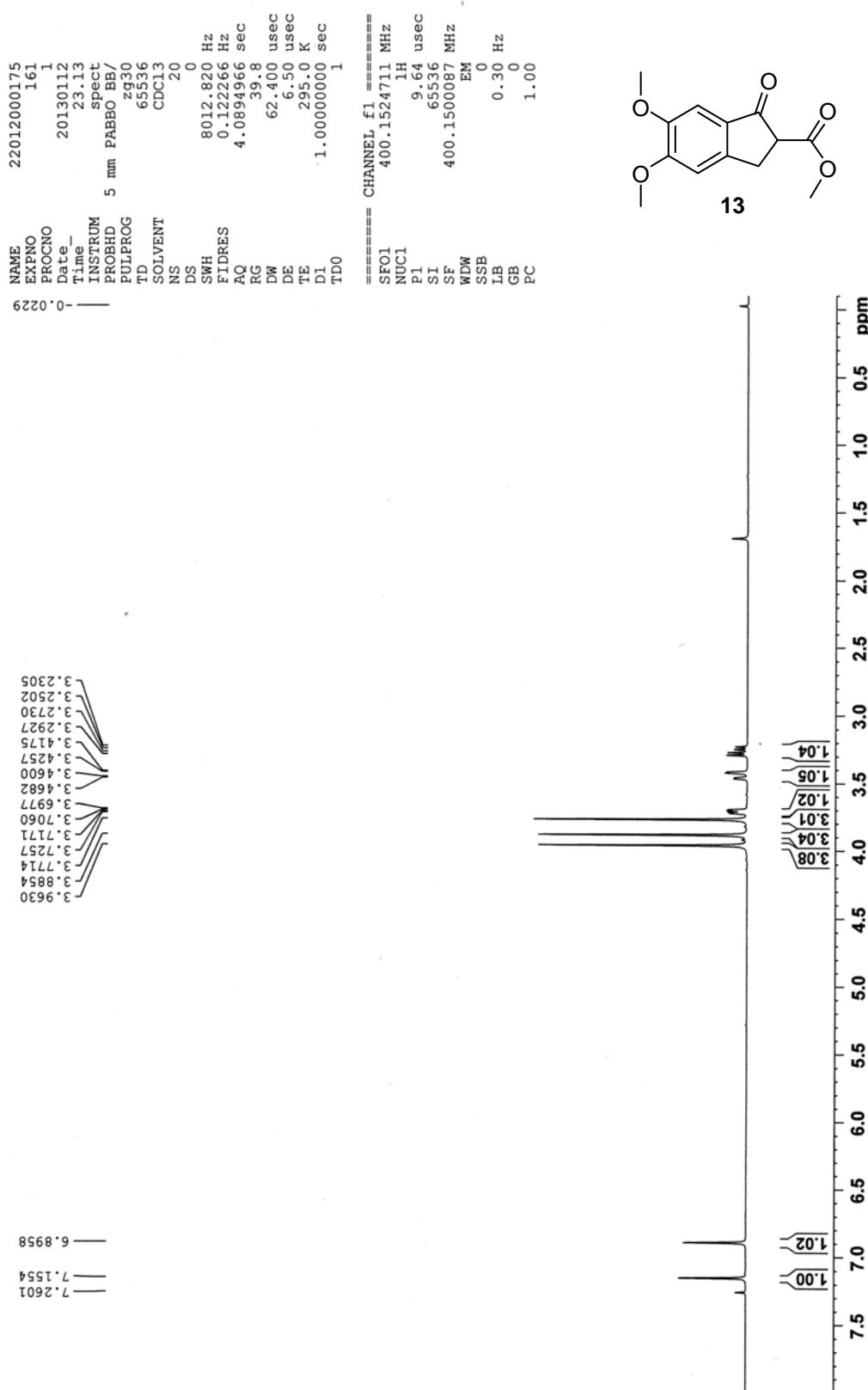
**Spectrum for
Enantioselective Total Synthesis of (+)-Brazilin, (-)-Brazilein and (+)-Brazilide A**

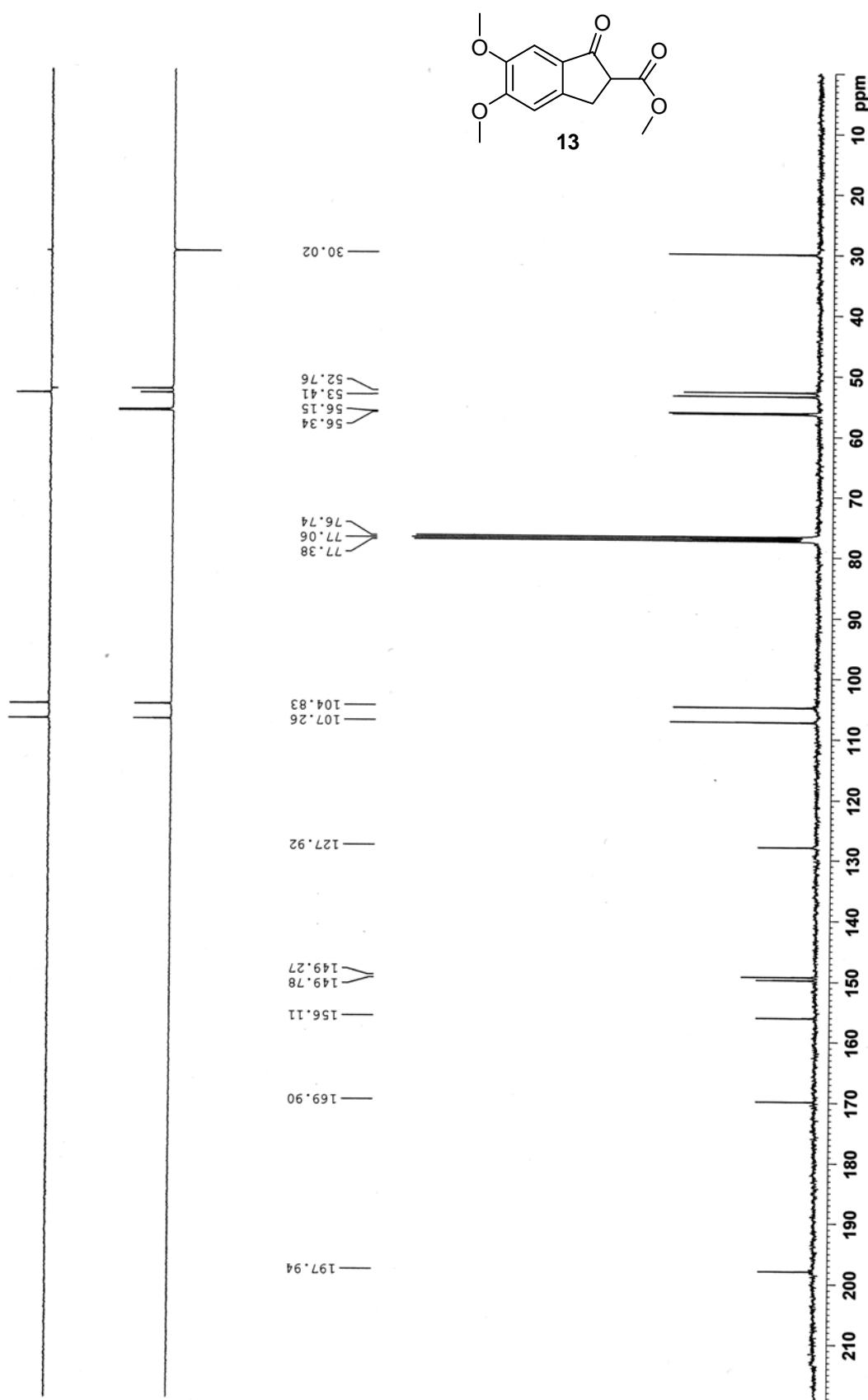
Xuequan Wang, Hongbin Zhang, Xiaodong Yang*, Jingfeng Zhao, and Chengxue Pan*

Key Laboratory of Medicinal Chemistry for Natural Resource, Ministry of Education, School of Chemical Science and Technology, Yunnan University, Kunming, Yunnan 650091, P. R. China Fax: 86-871-5035538. E-mail:
zhanghb@ynu.edu.cn; xdyang@ynu.edu.cn



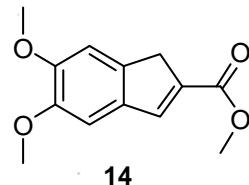




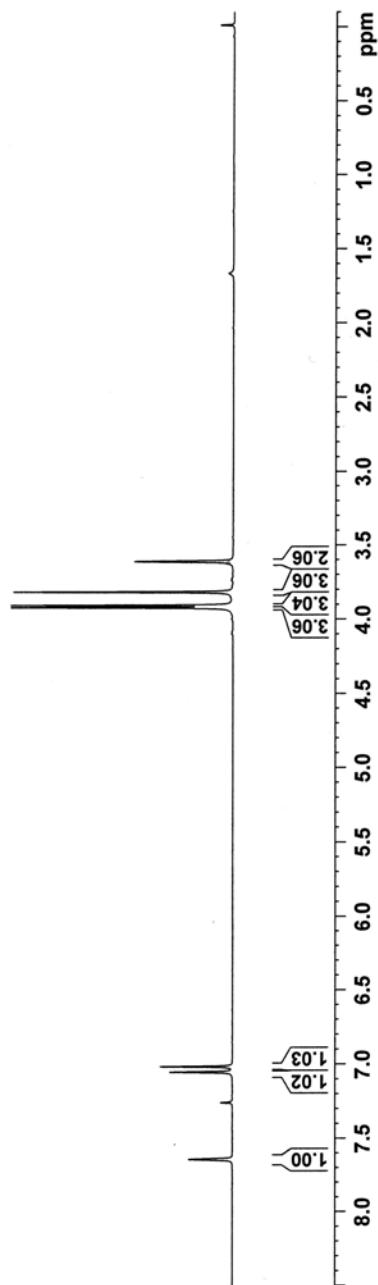


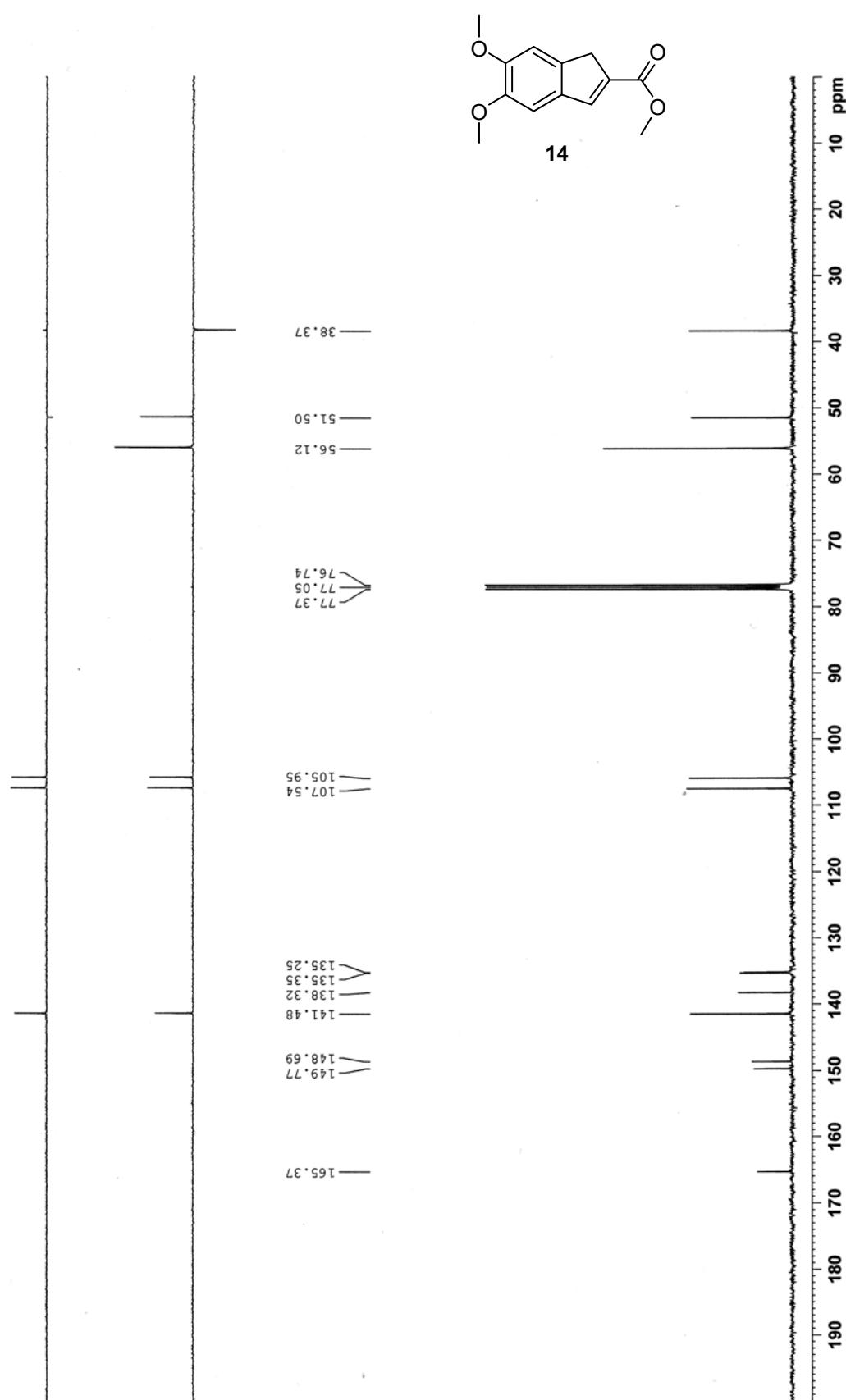
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PULPROG zg30
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SOLVENT CDC13
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DS 0
SWH 8012.820 Hz
FIDRES 0.0122266 Hz
AQ 4.0894966 sec
RG 63.8
DW 62.400 usec
DE 6.50 usec
TE 295.0 K
D1 1.0000000 sec
TD0 1
===== CHANNEL f1 =====
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P1 9.64 usec
SI 65536
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SSB 0
LB 0.30 Hz
GB 0
PC 1.00

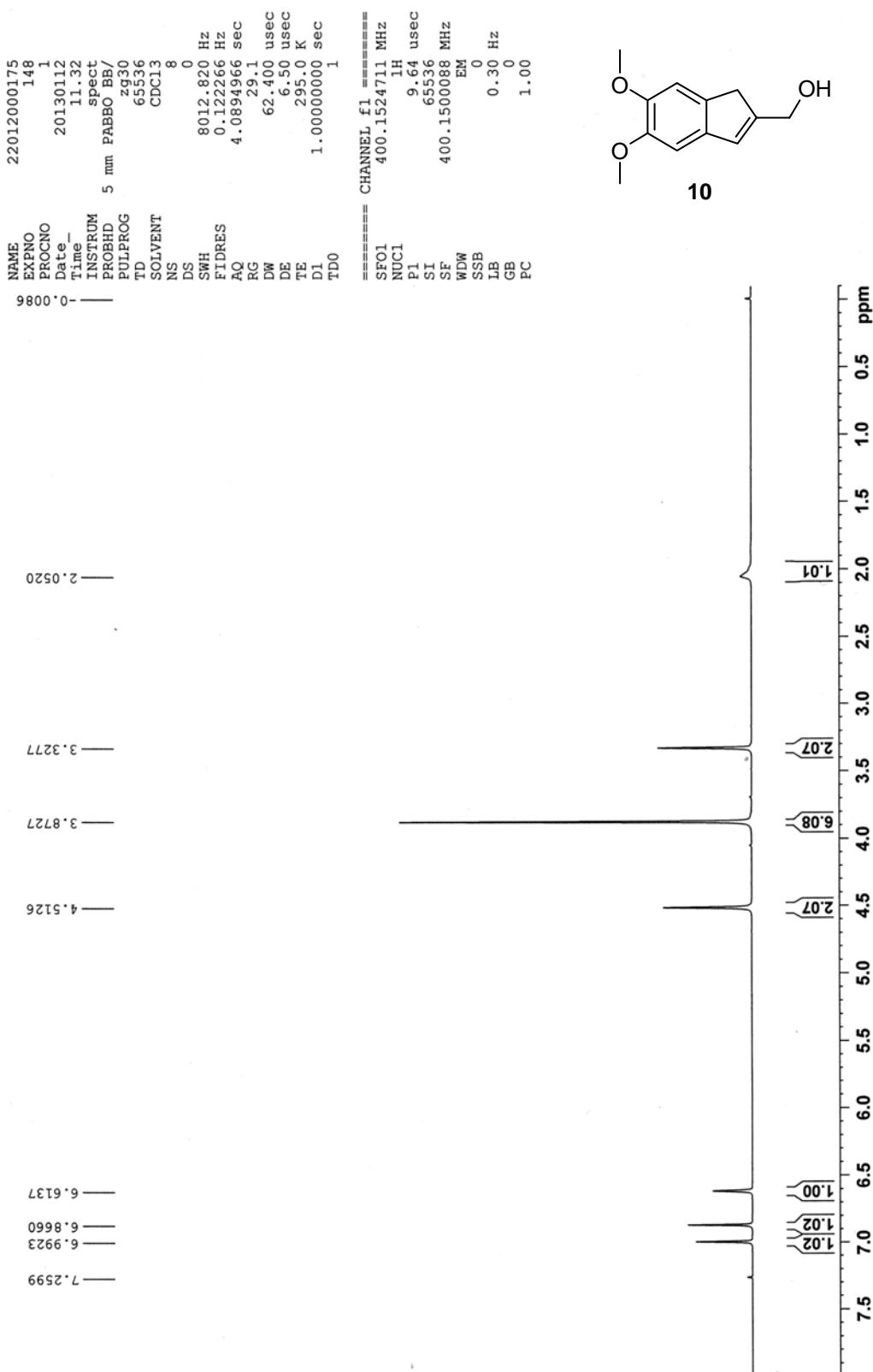
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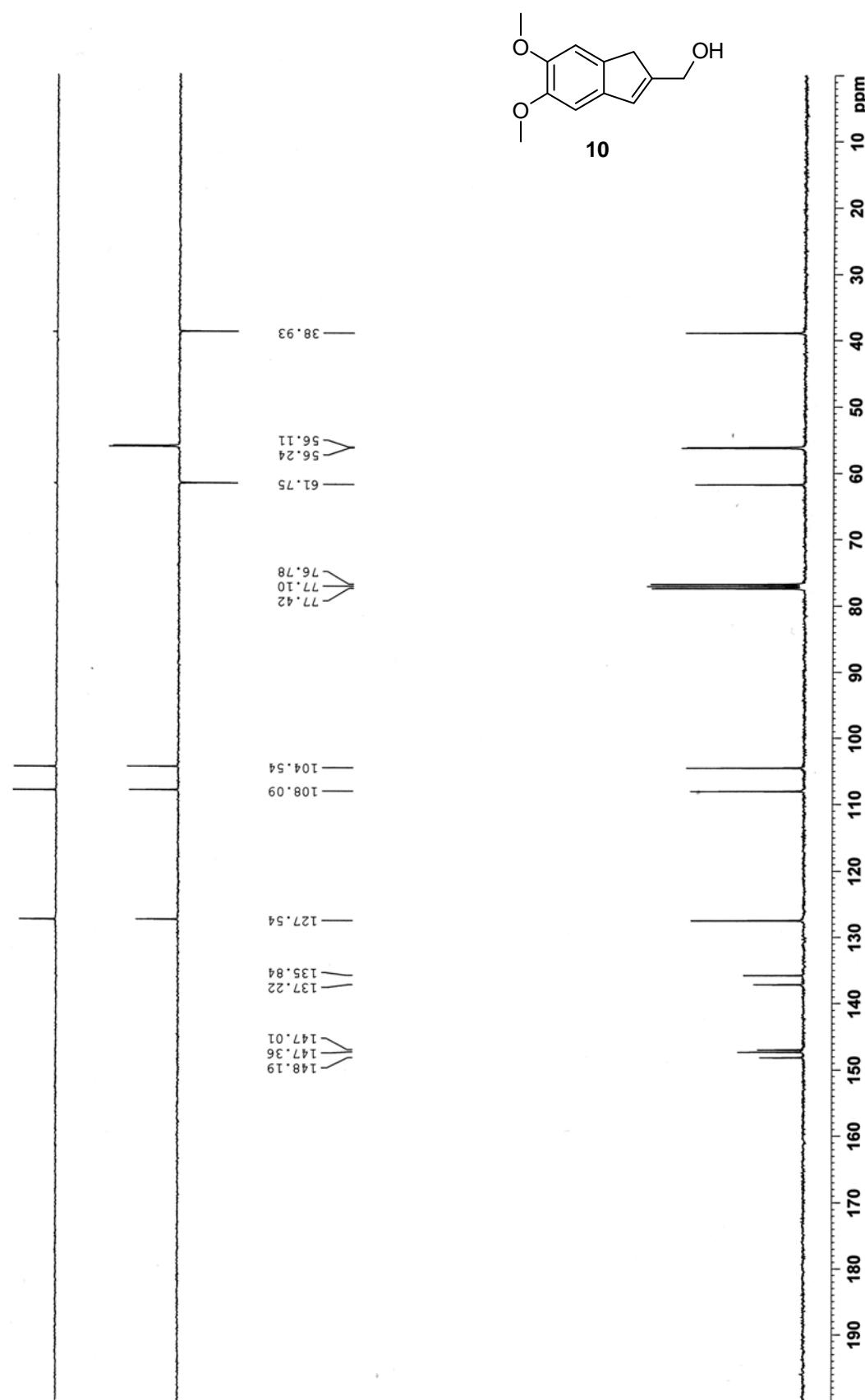


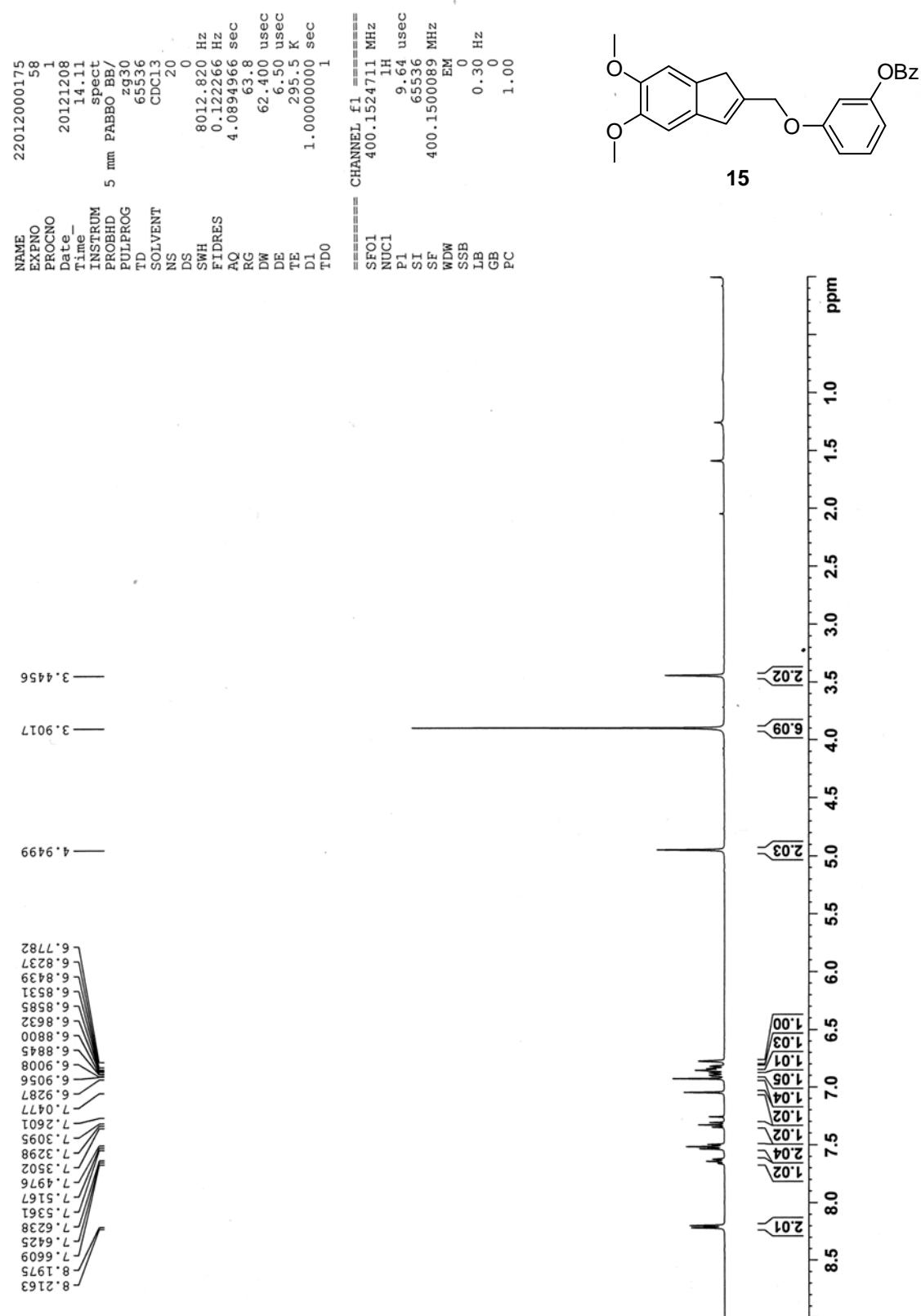
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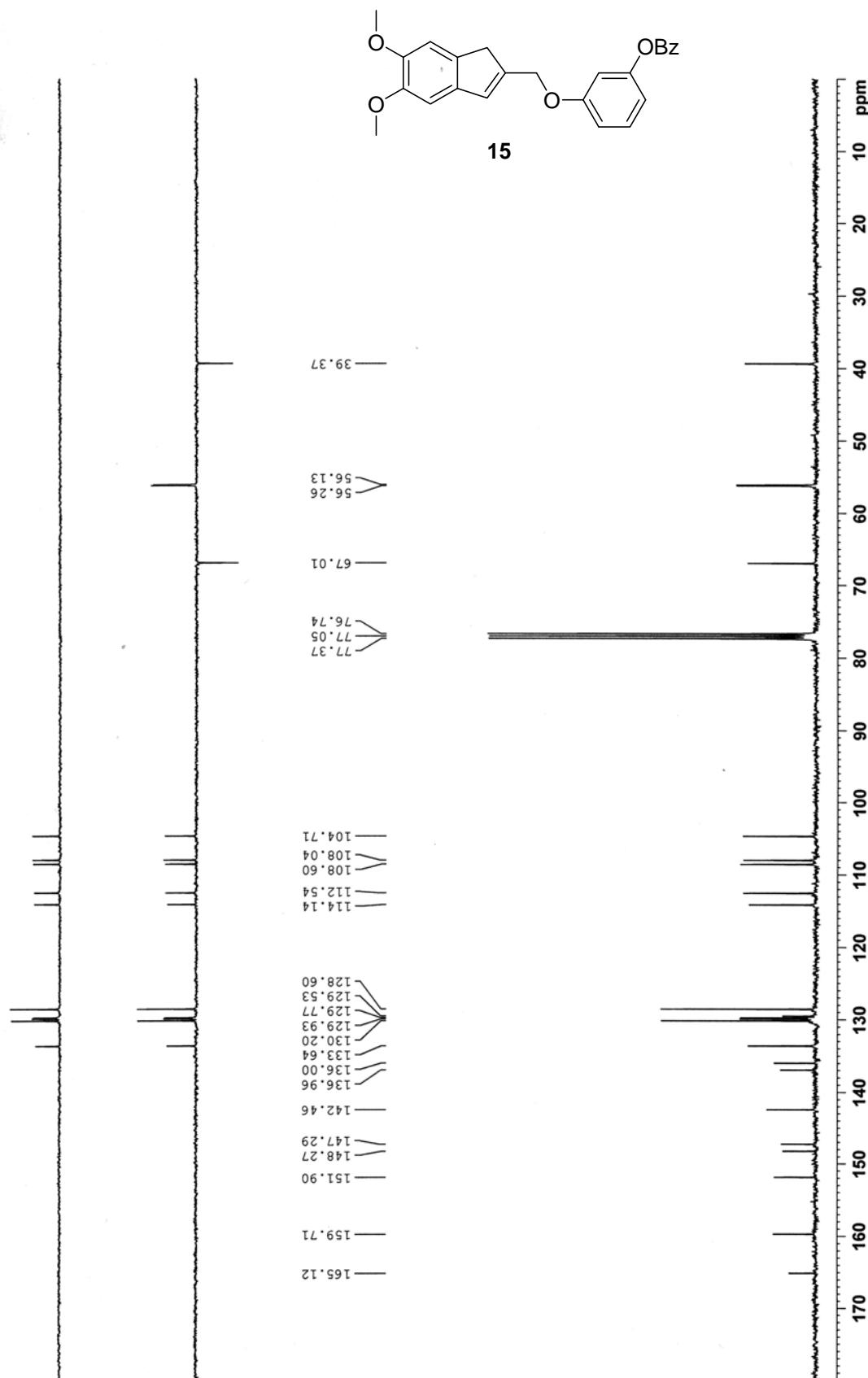


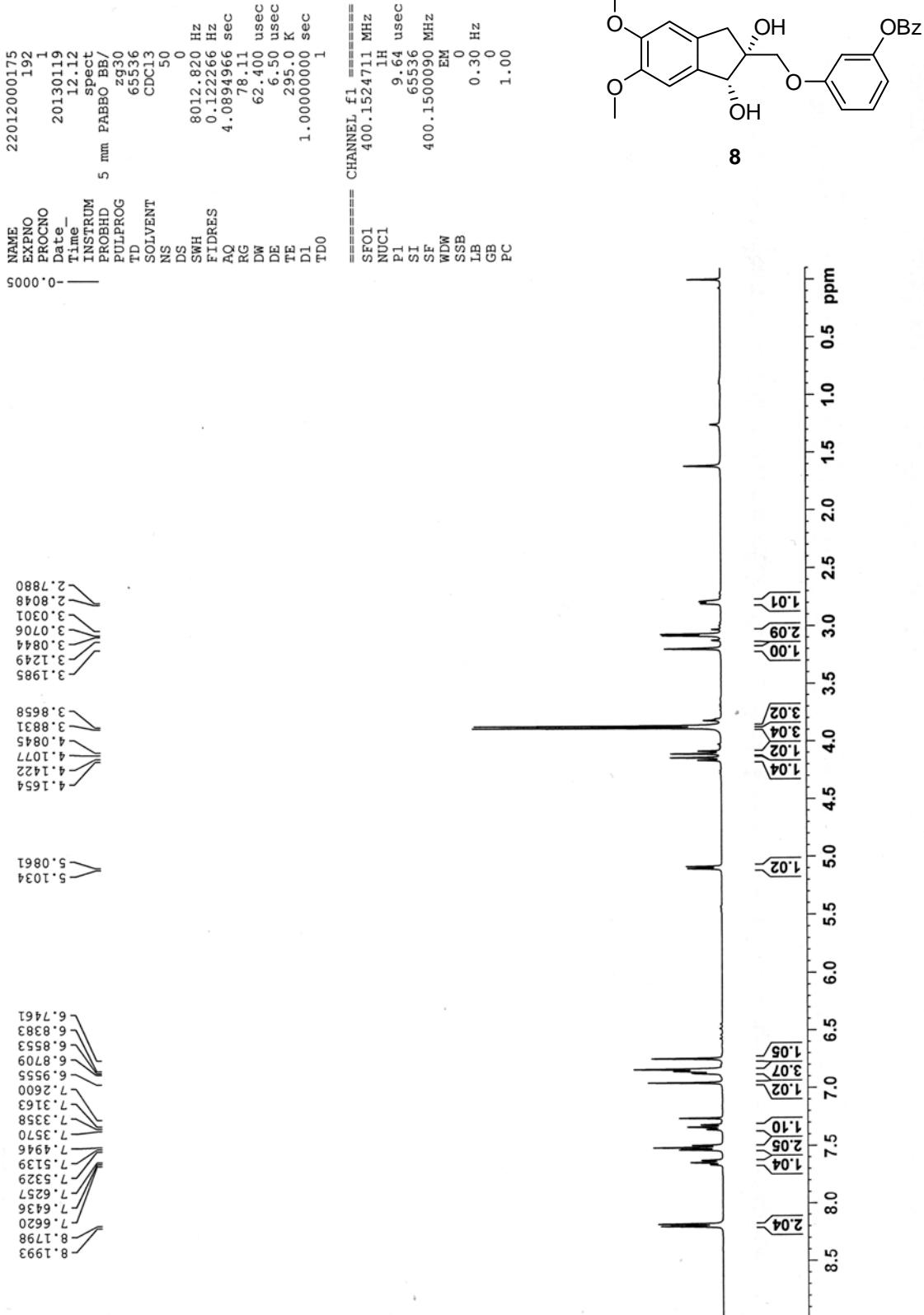


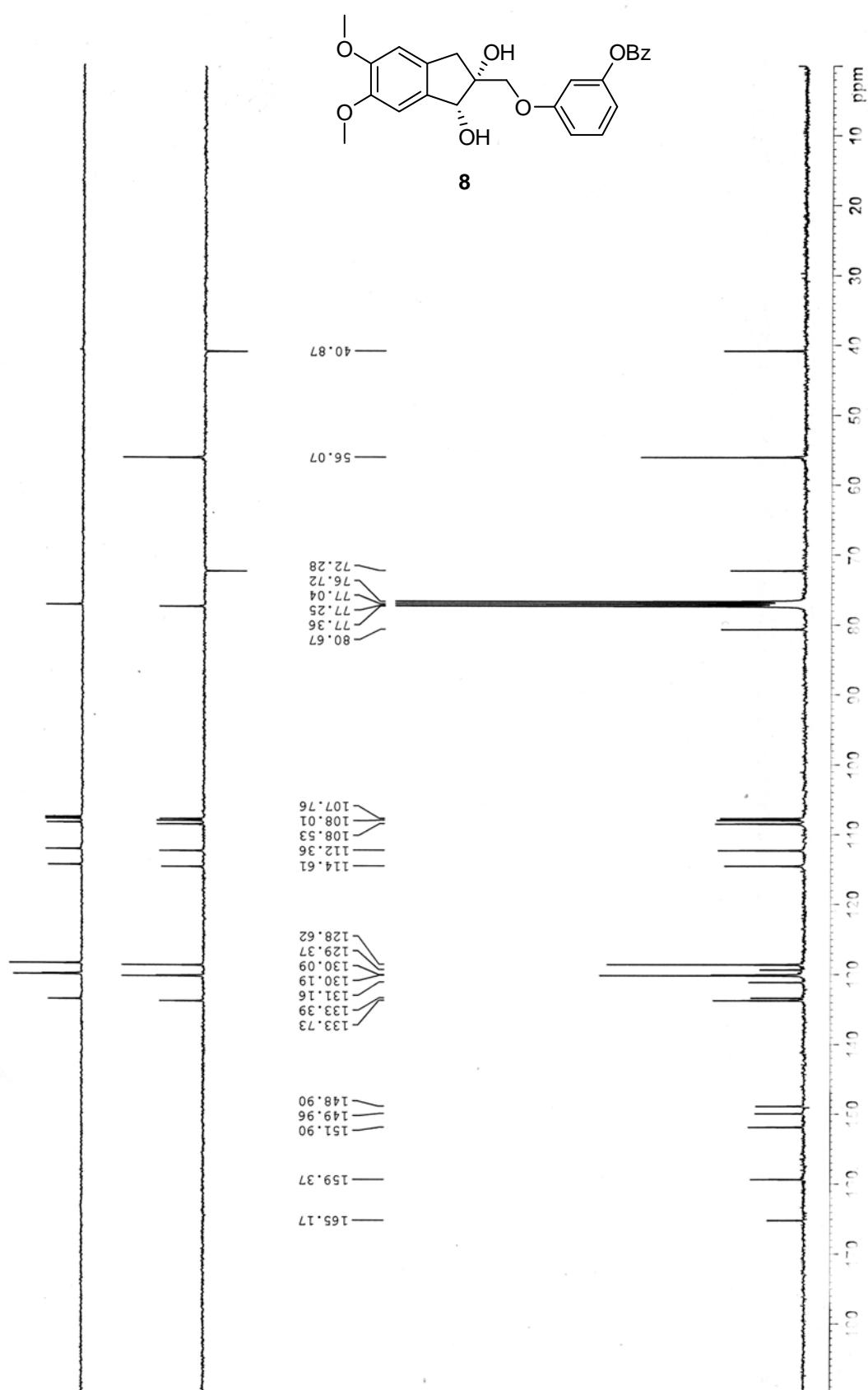


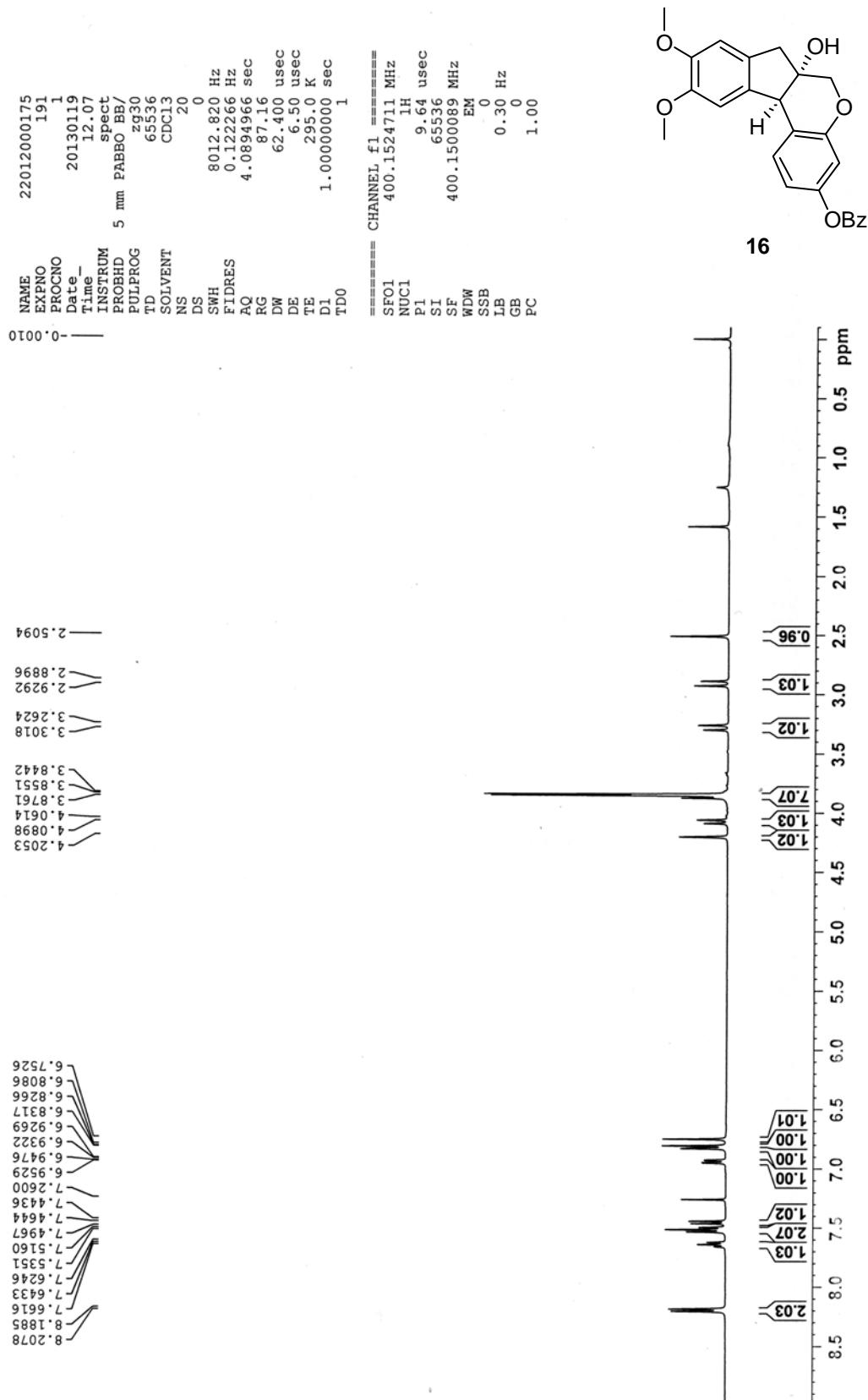


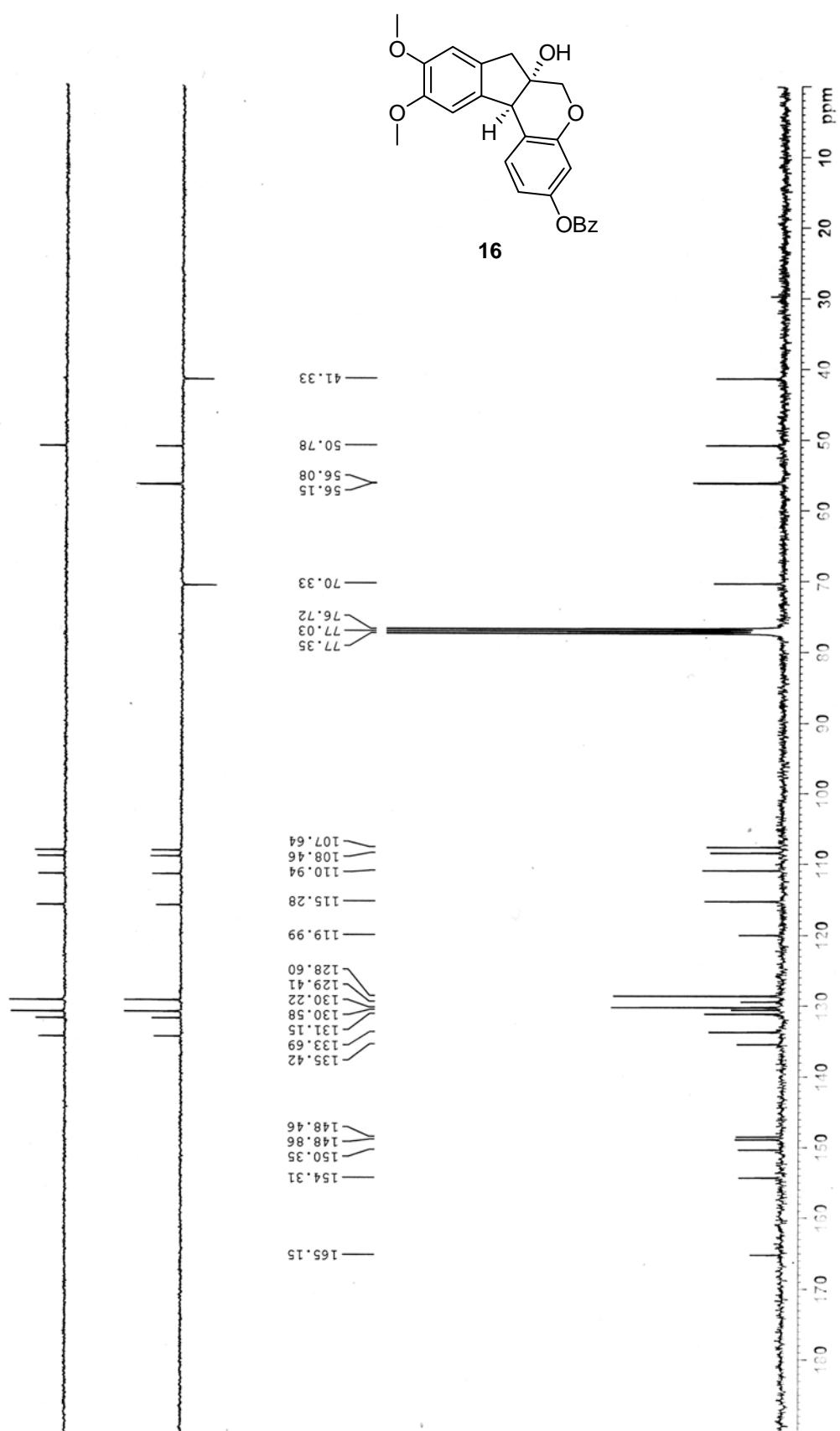


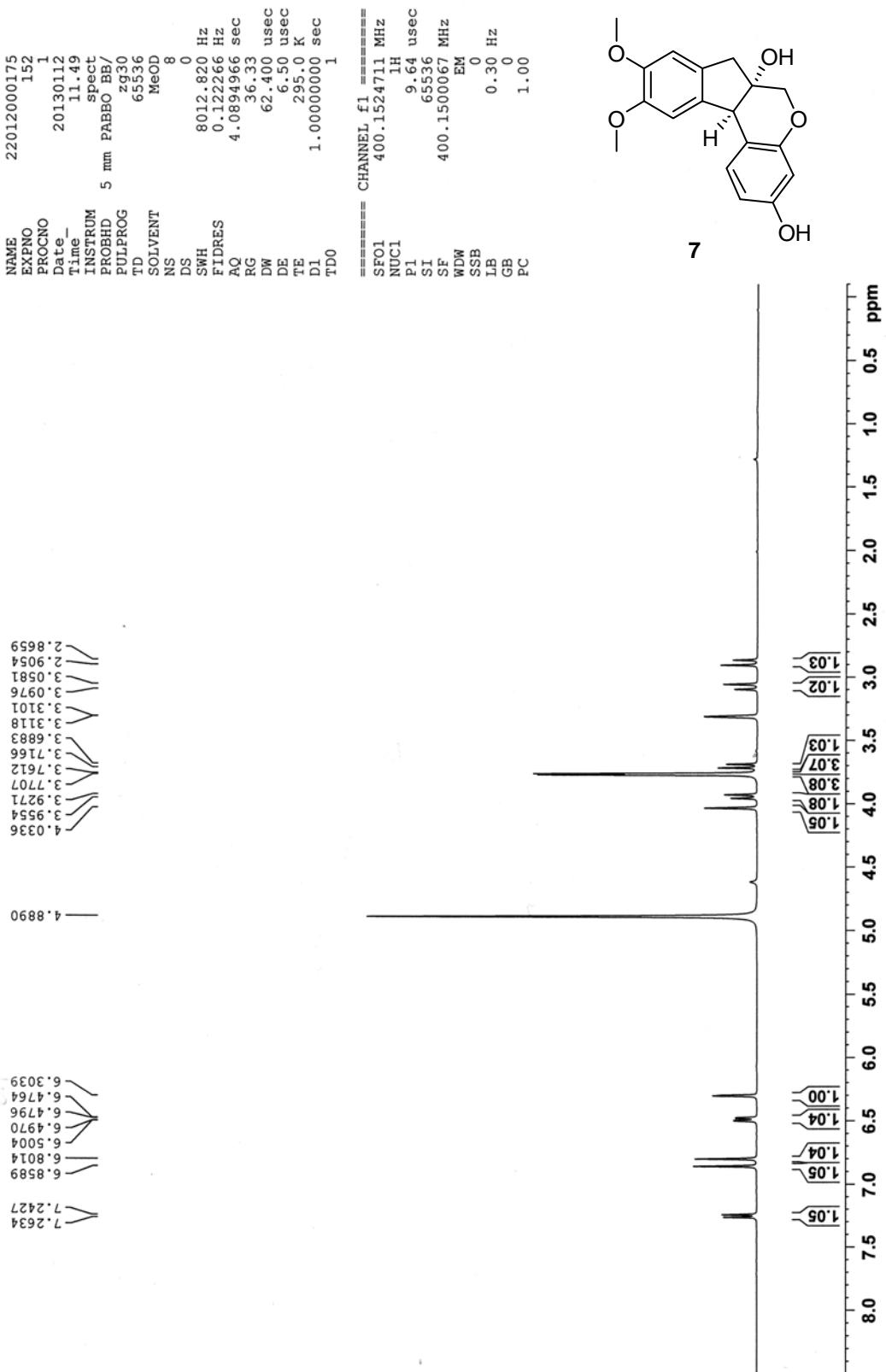


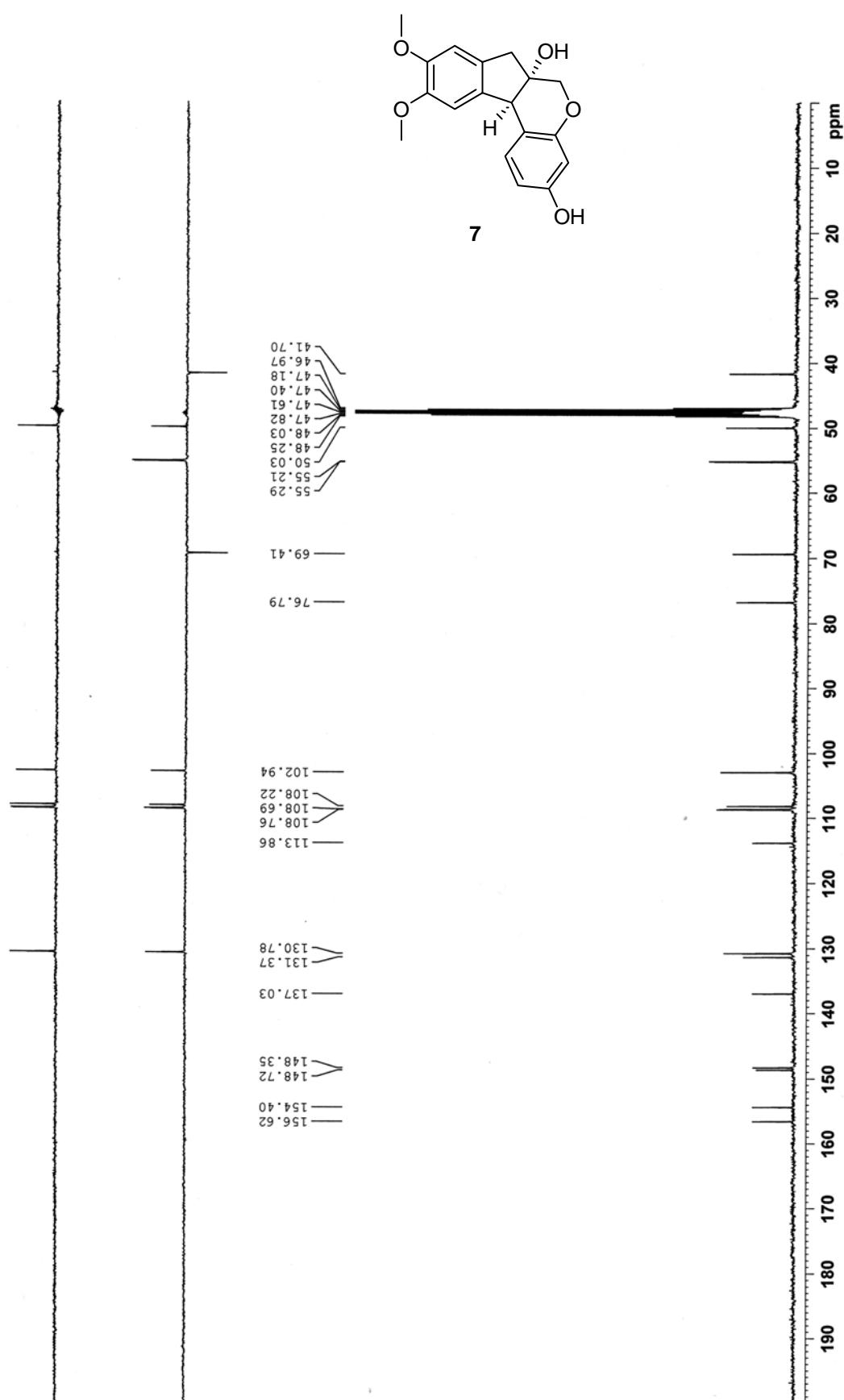


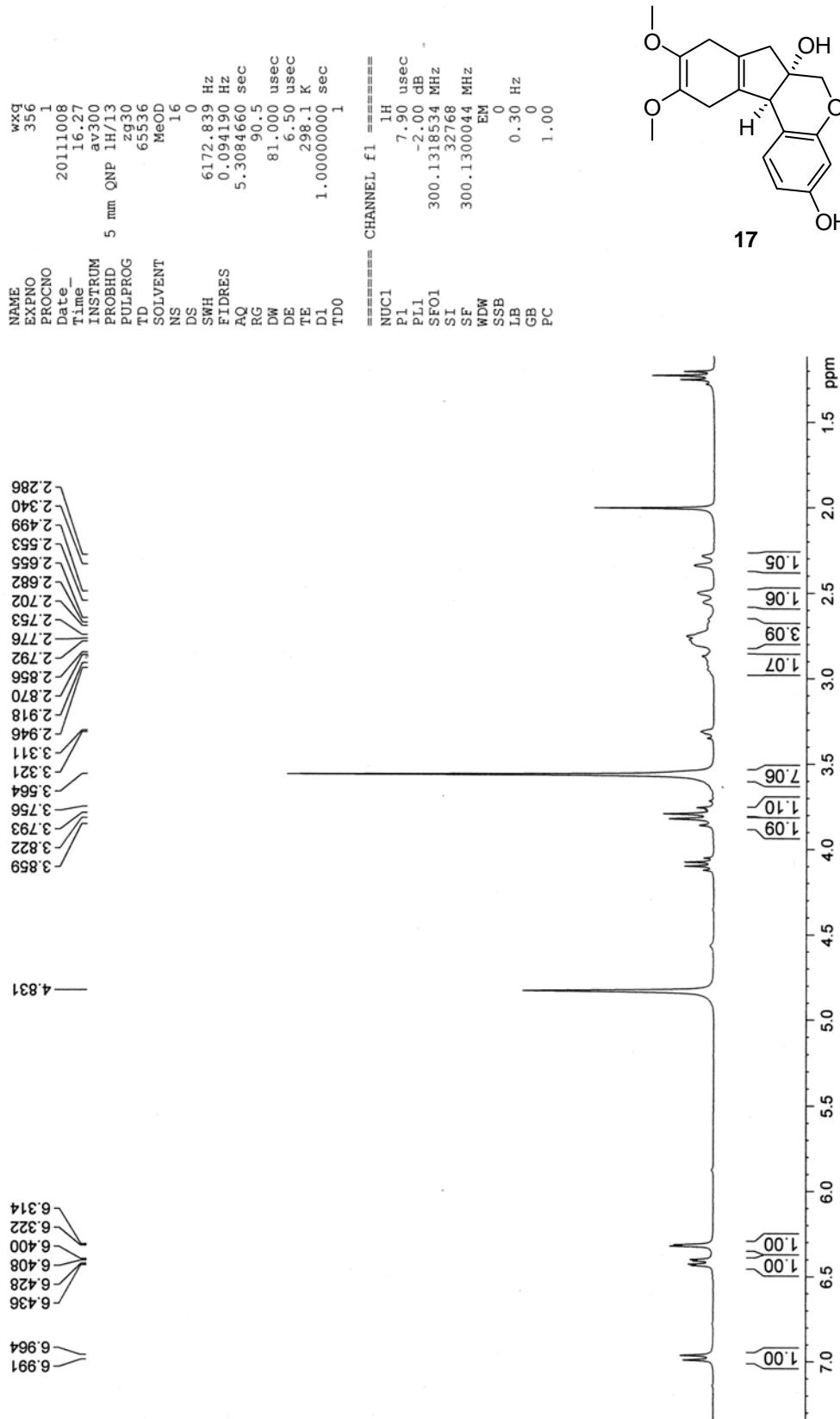


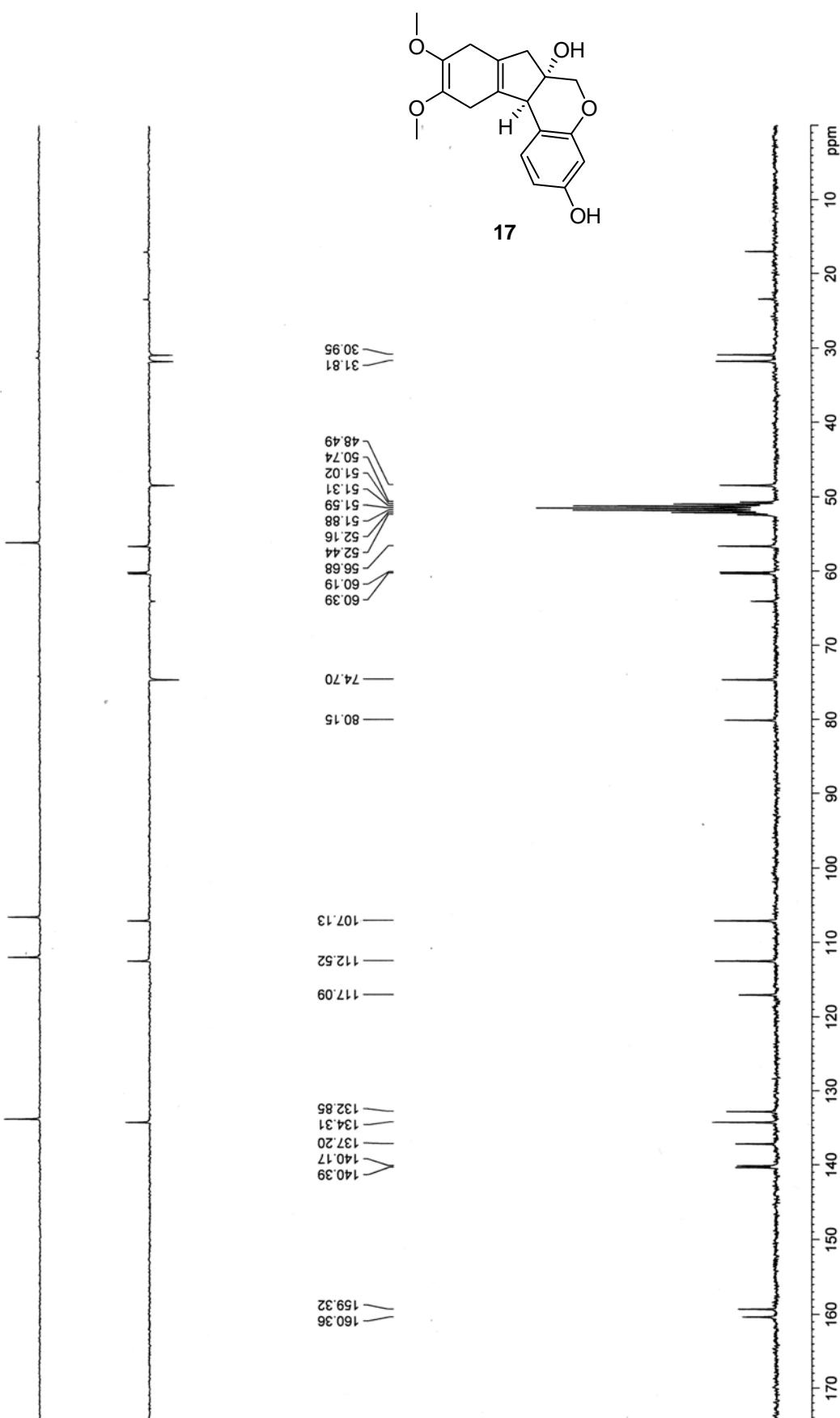










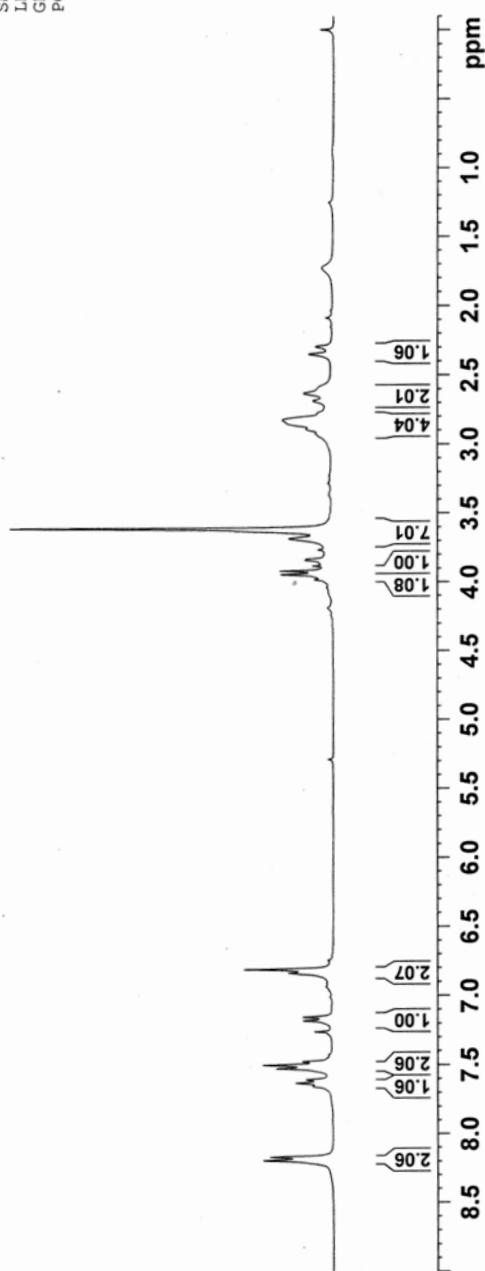
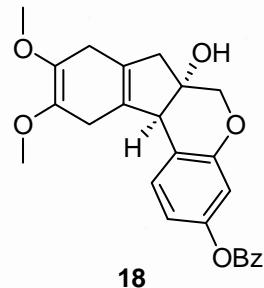


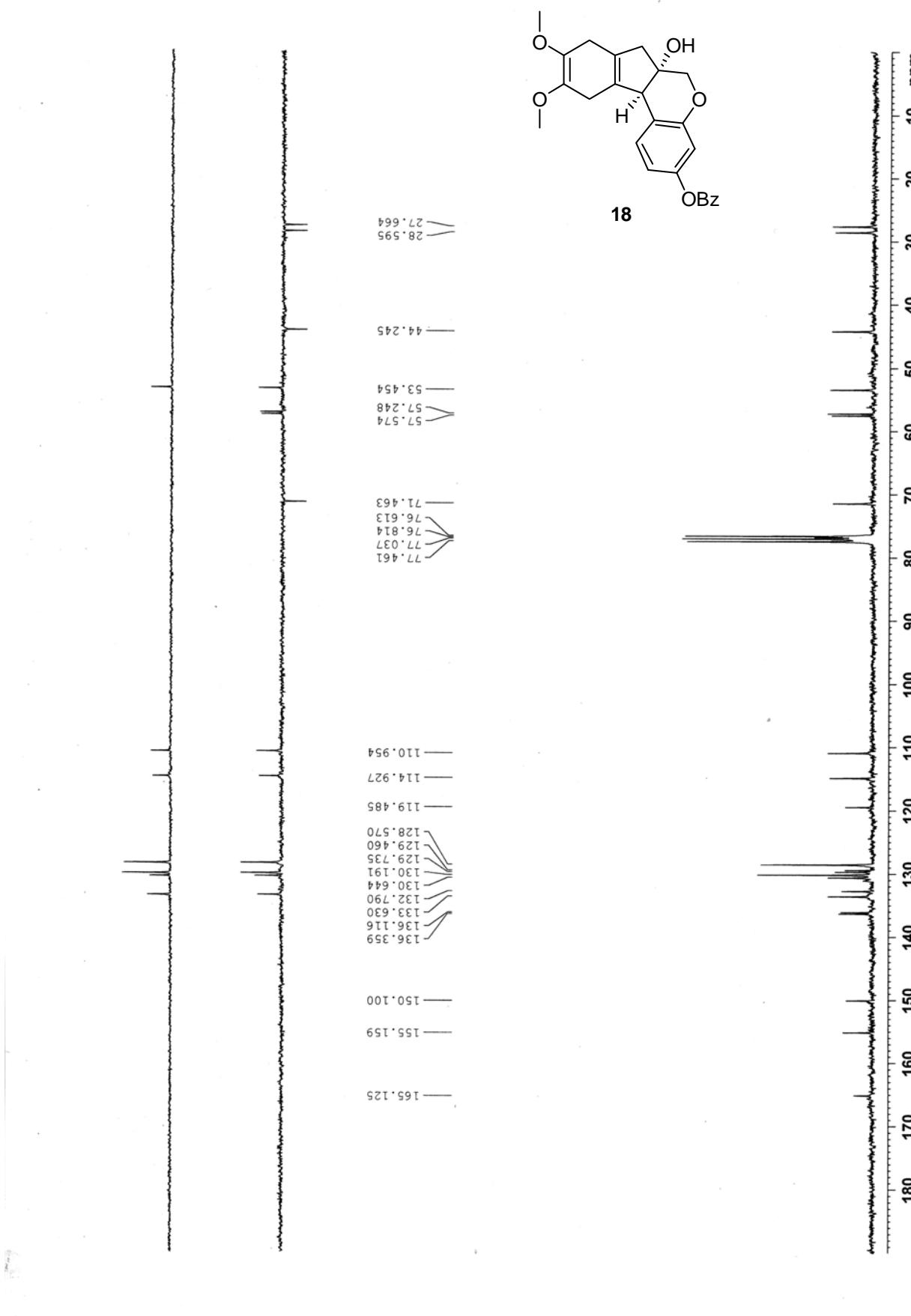
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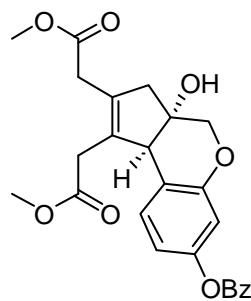
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Time_      9.28
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TD       65536
SOLVENT  CDC13
NS        16
DS        2
SWH      6172.839 Hz
FIDRES   0.00190 Hz
AQ       5.3004660 s
RG       181
DW       81.000 u
DE       6.50 u
TE       298.8 K
D1      1.0000000 s
TD      1

===== CHANNEL f1 =====
NUC1      1H
P1        7.90 u
PPL1     -2.00 d
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SI        32768
SF        300.1300057 MHz
WDW      EM
SSB      0
LB      0.30 Hz
GB      1.00
PC

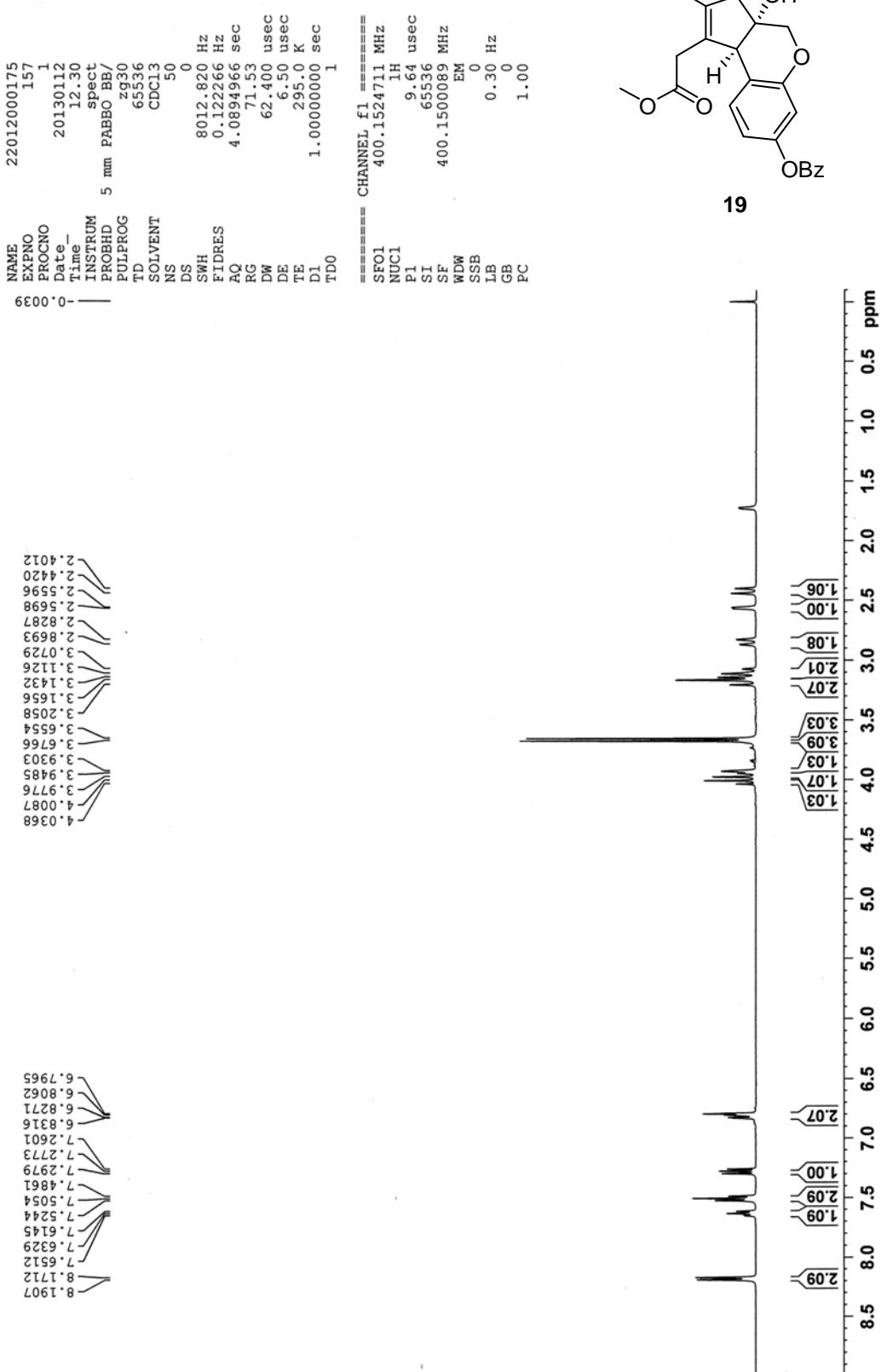
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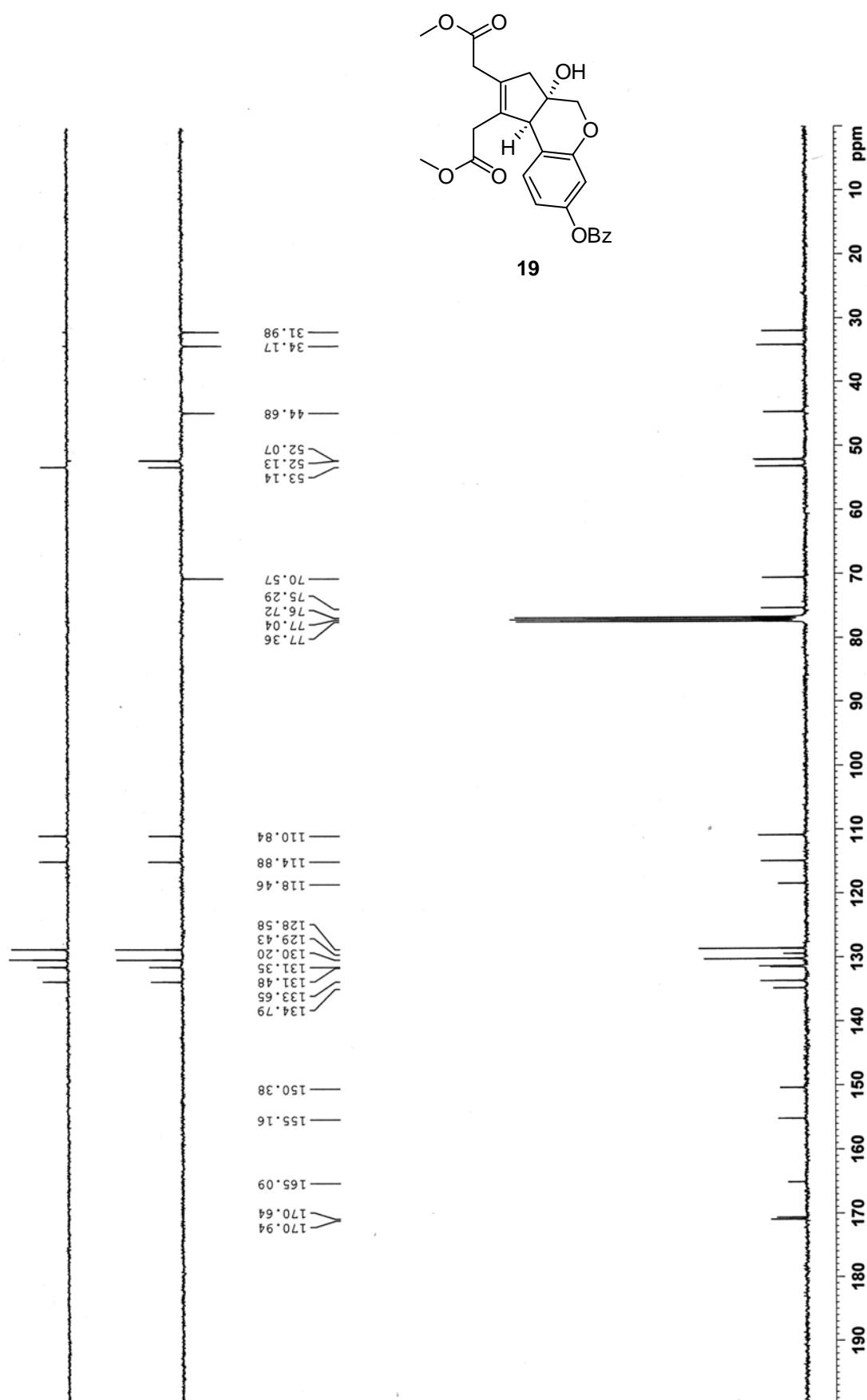


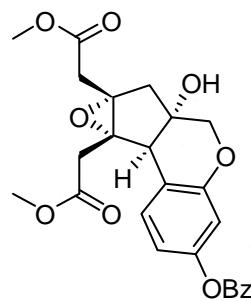




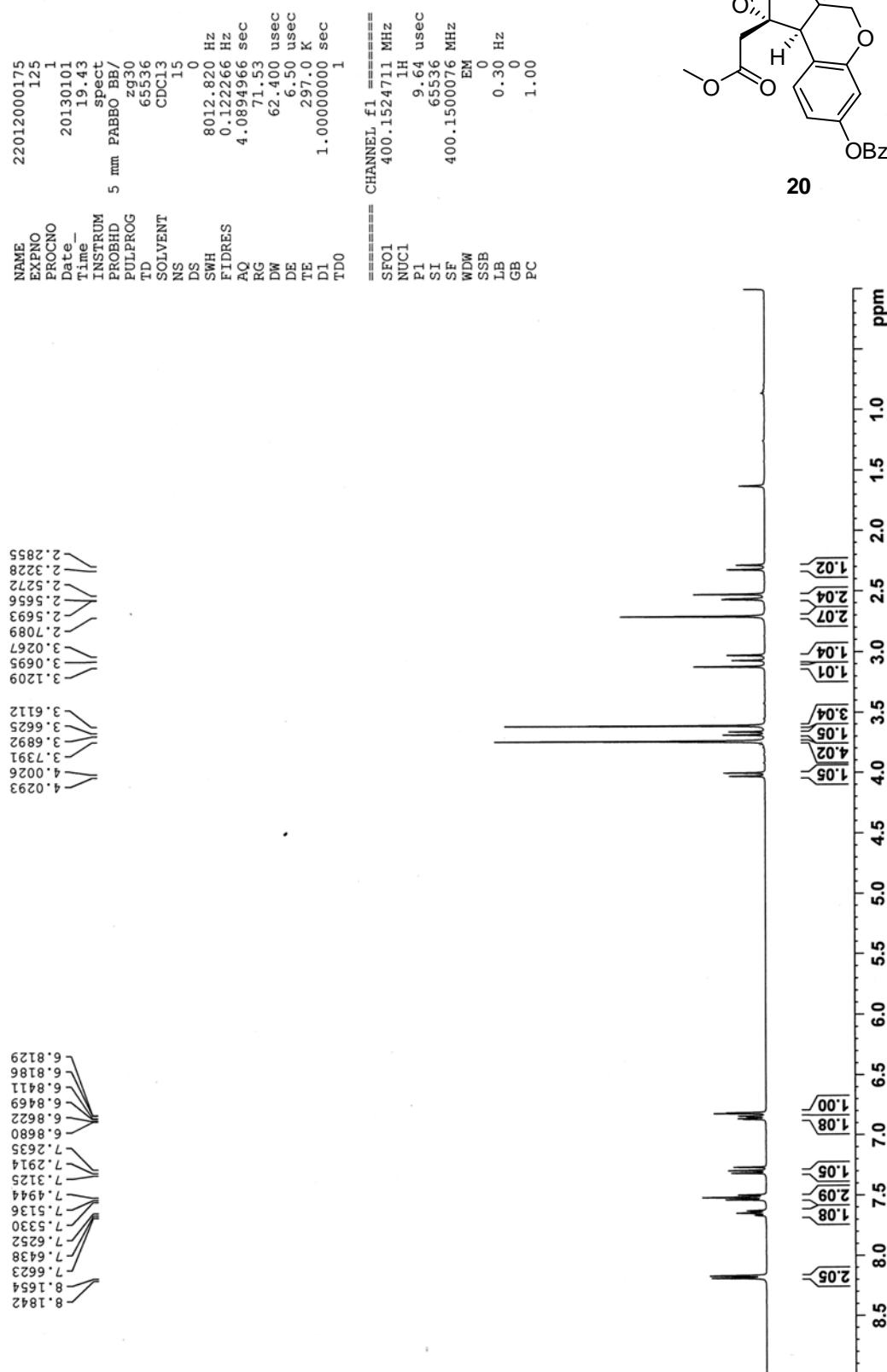
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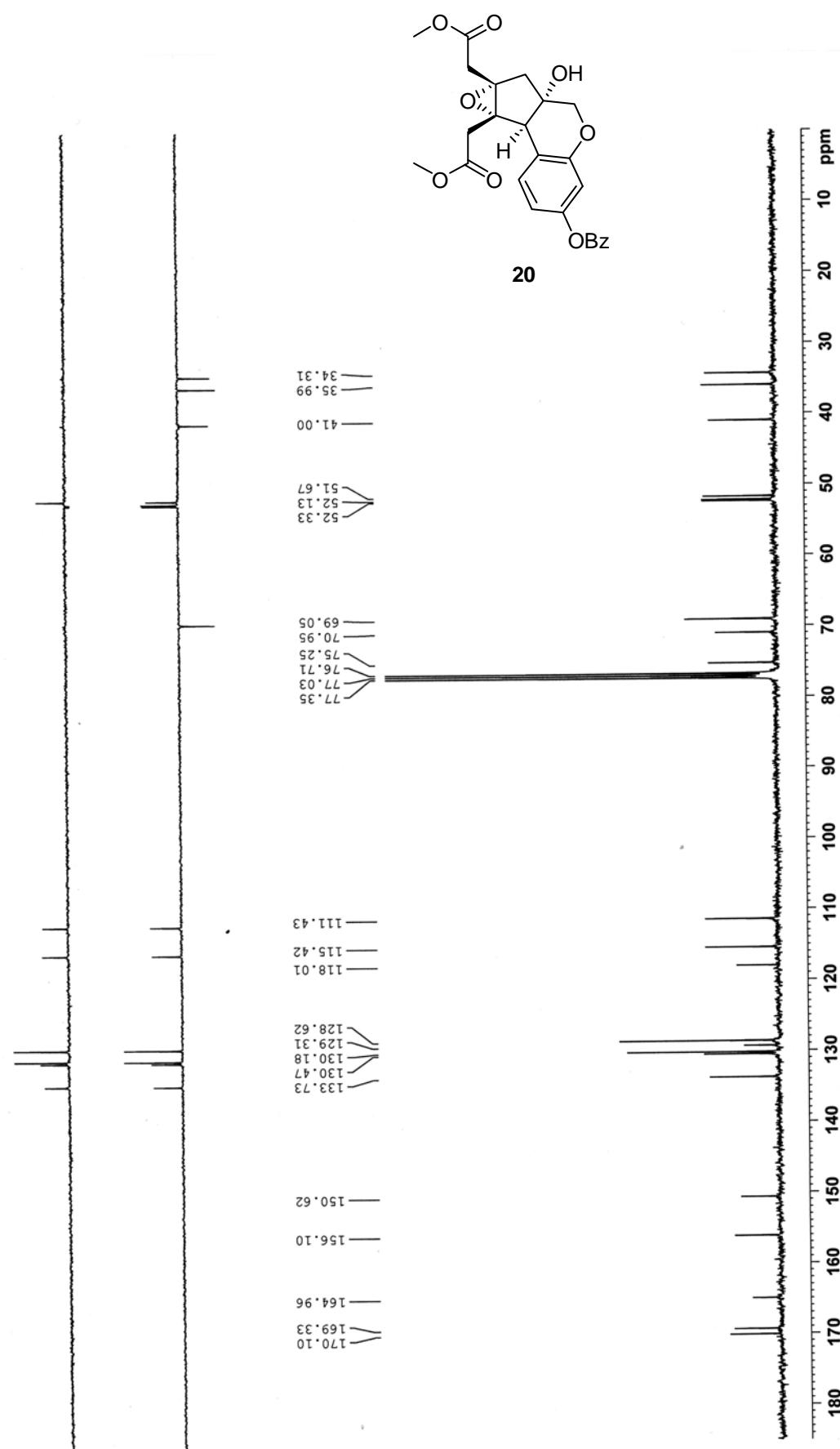


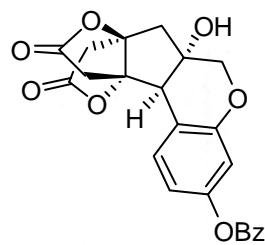




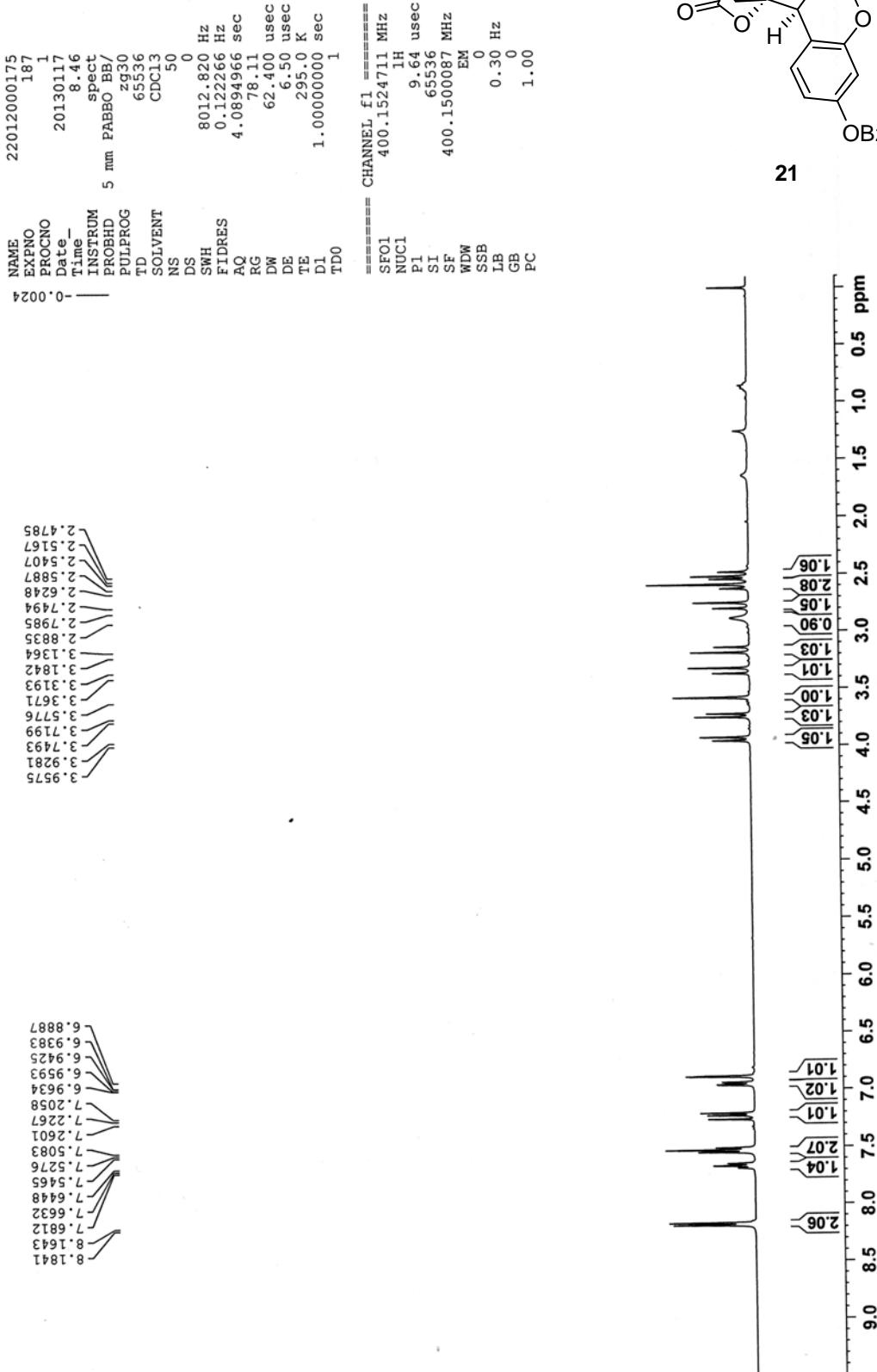
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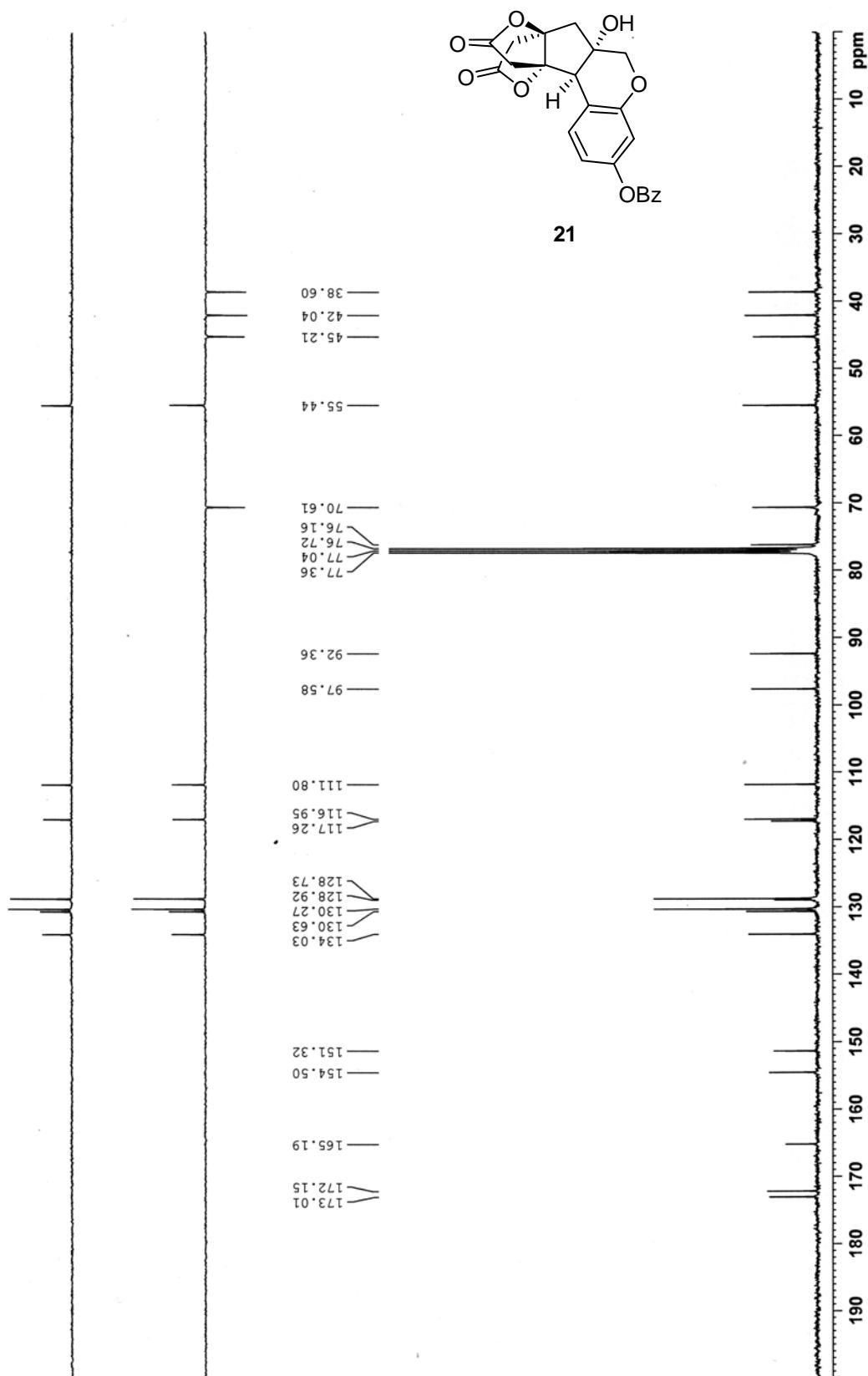


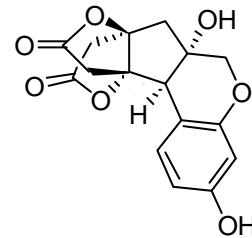
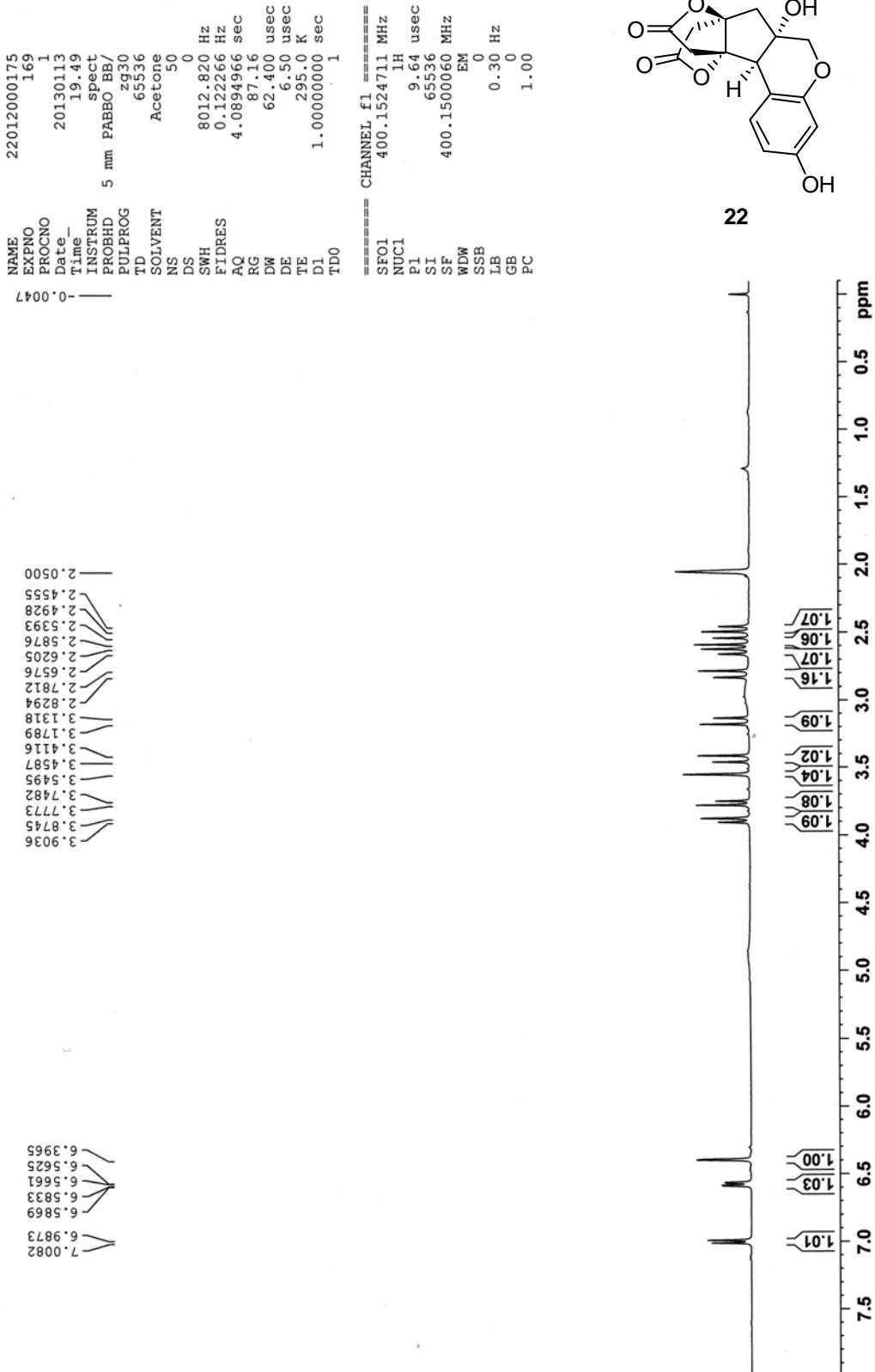




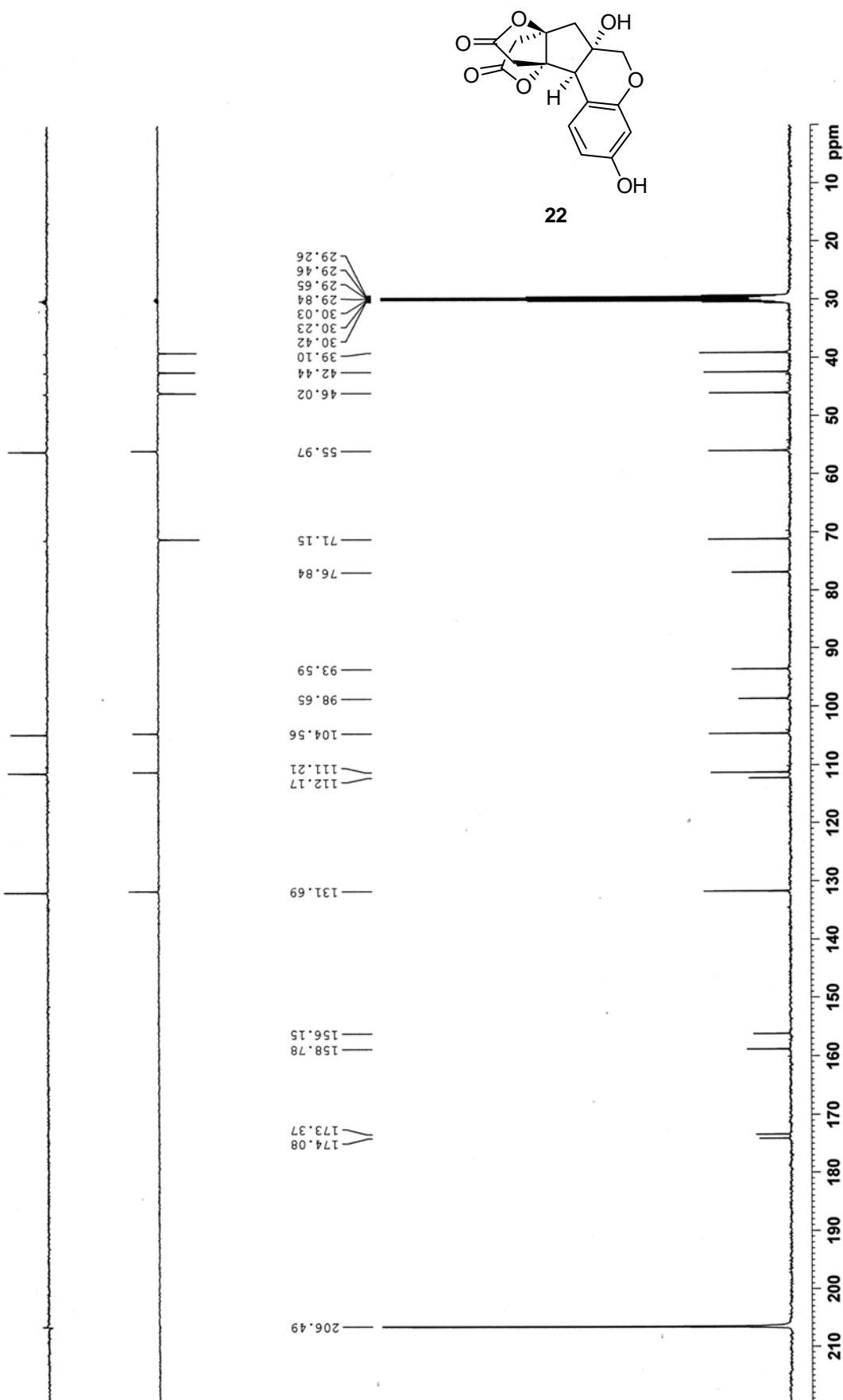
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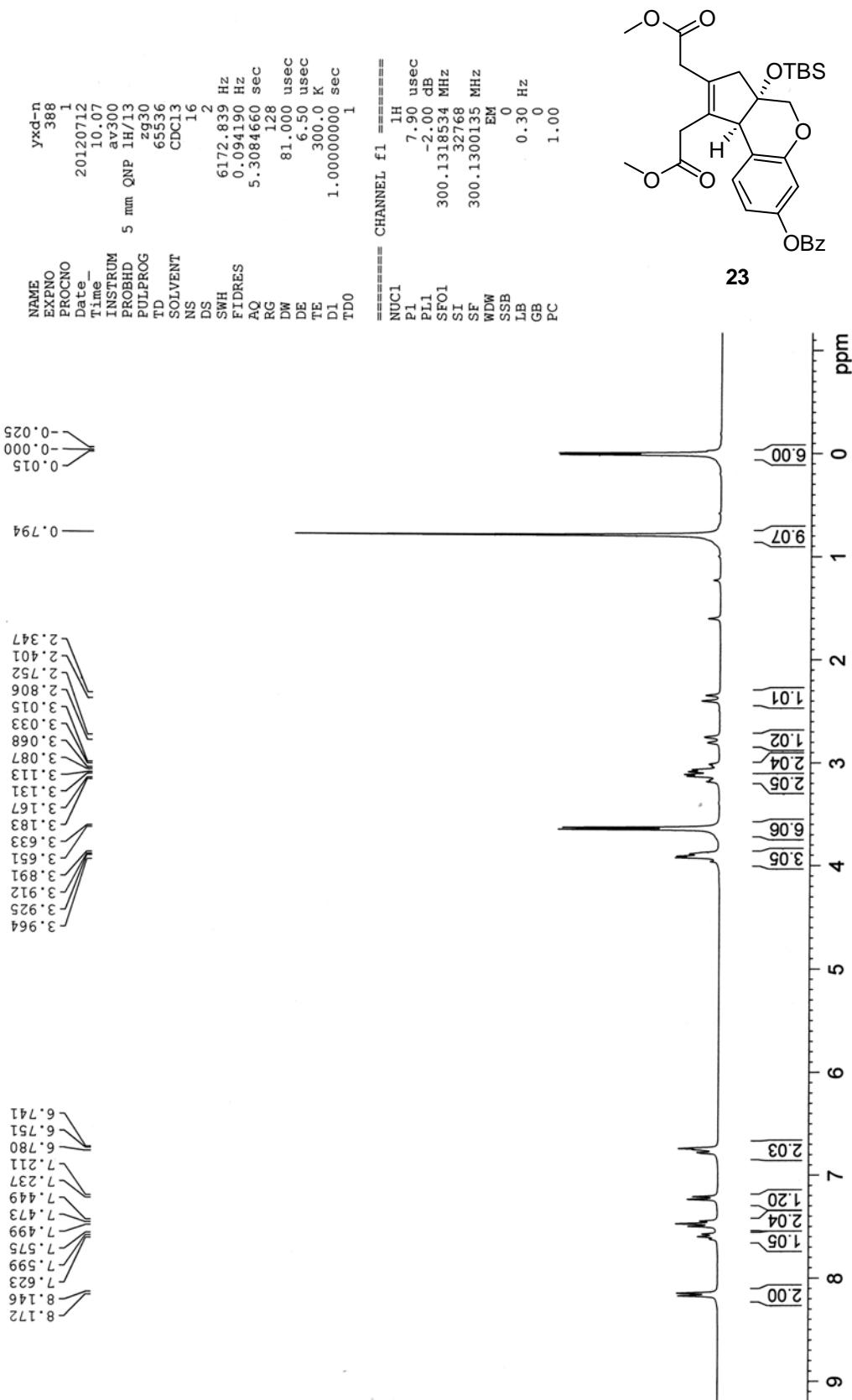


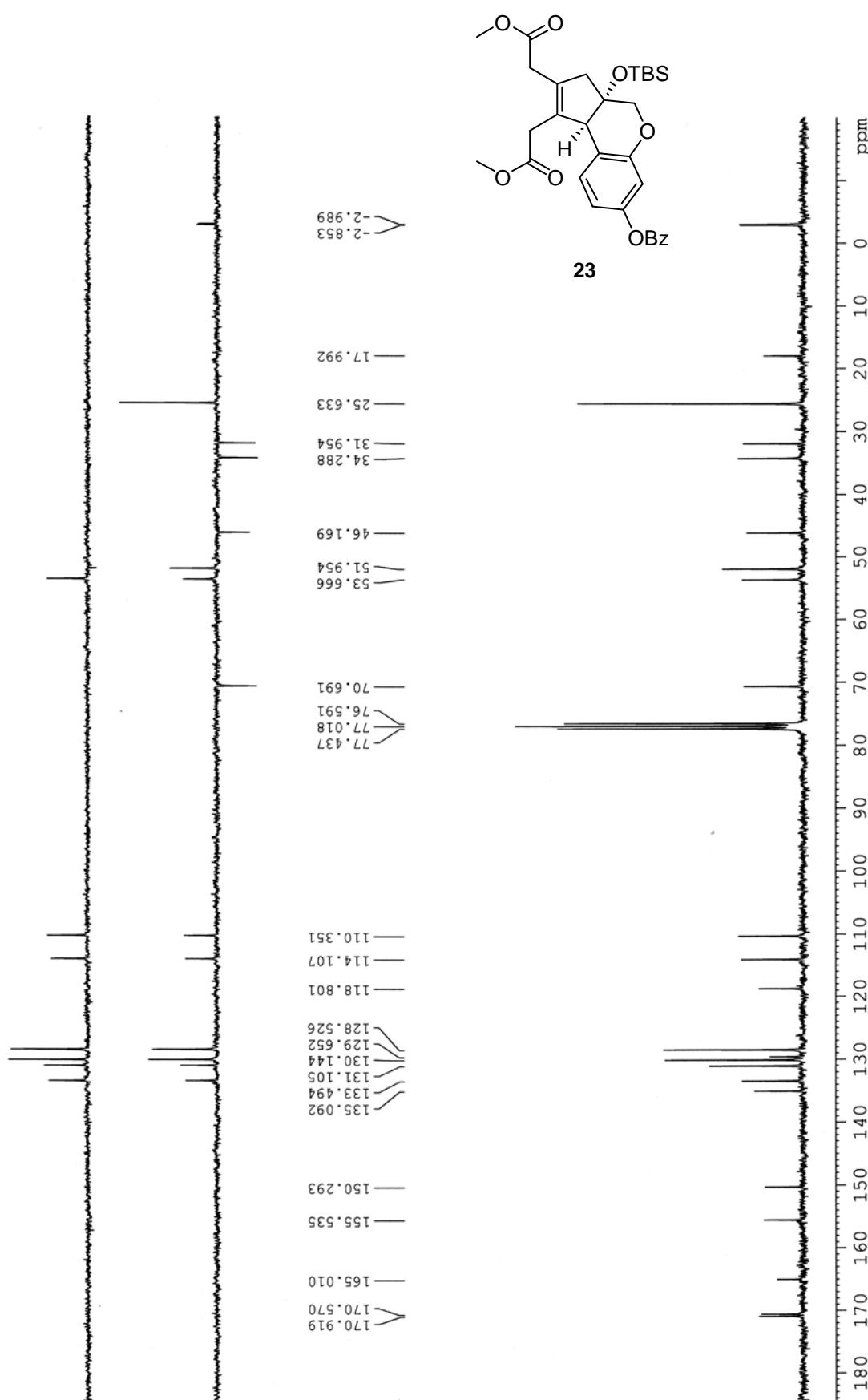


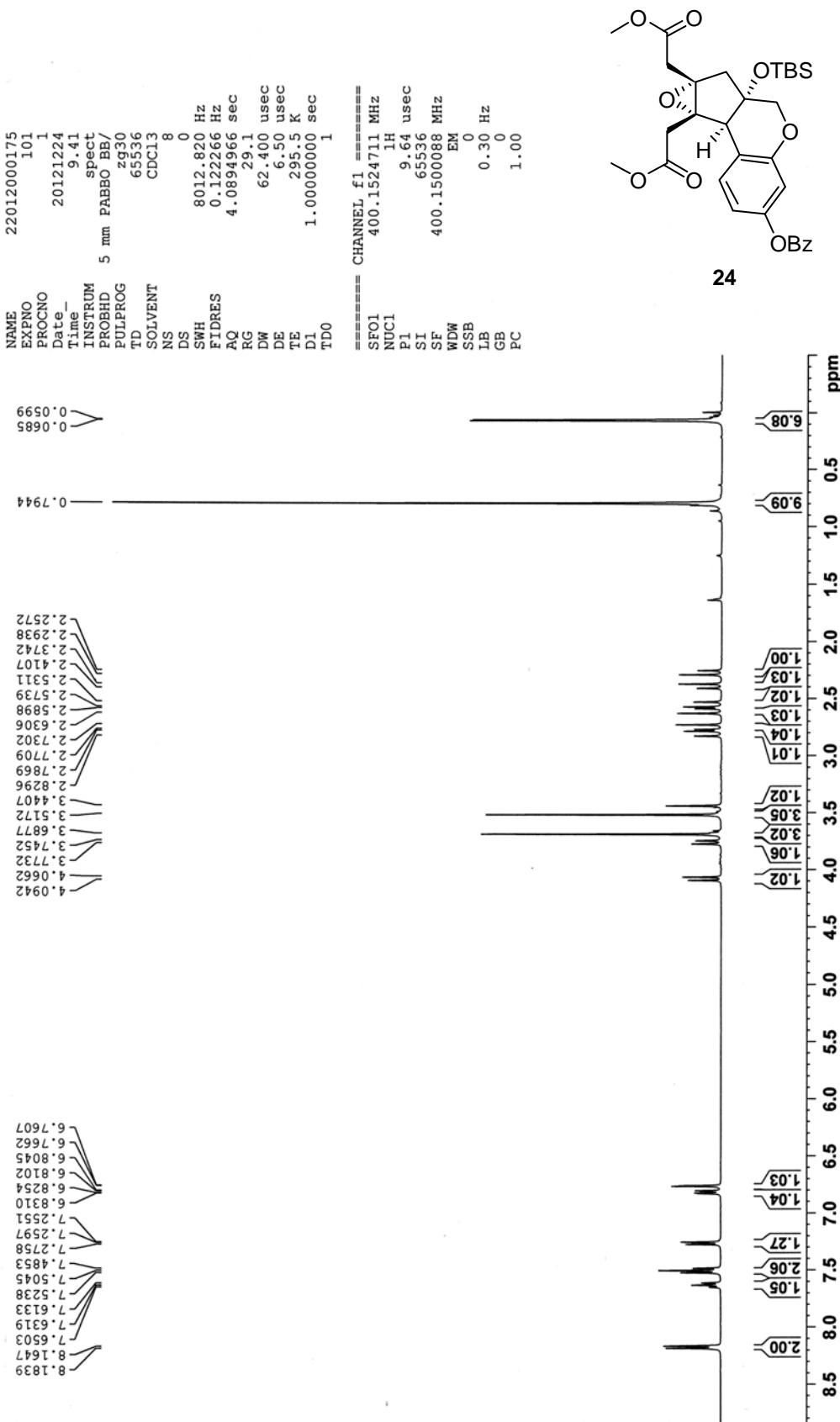


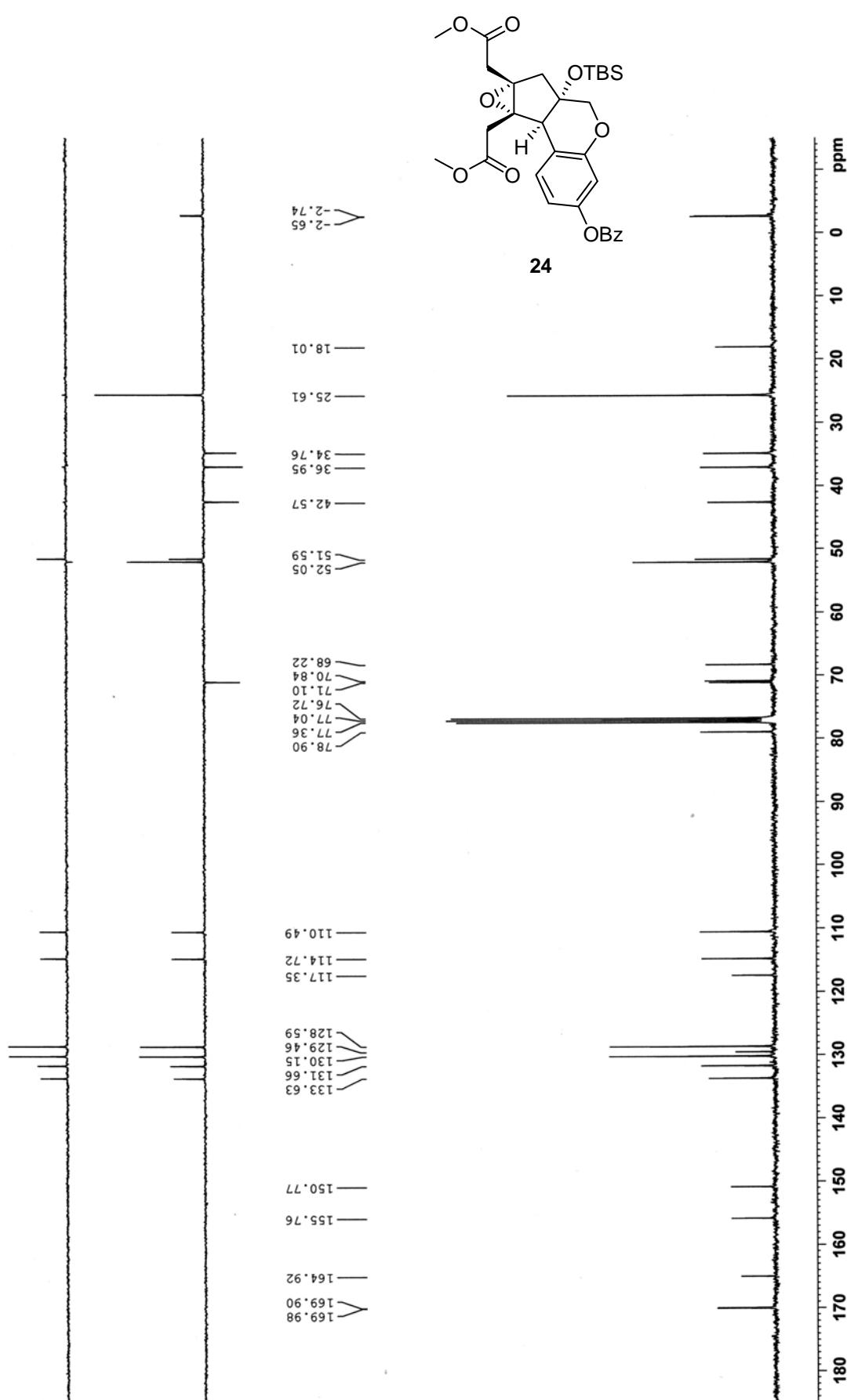
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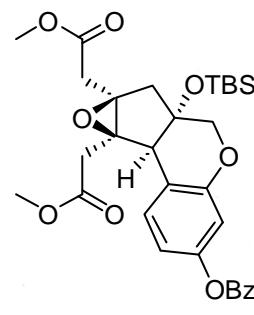




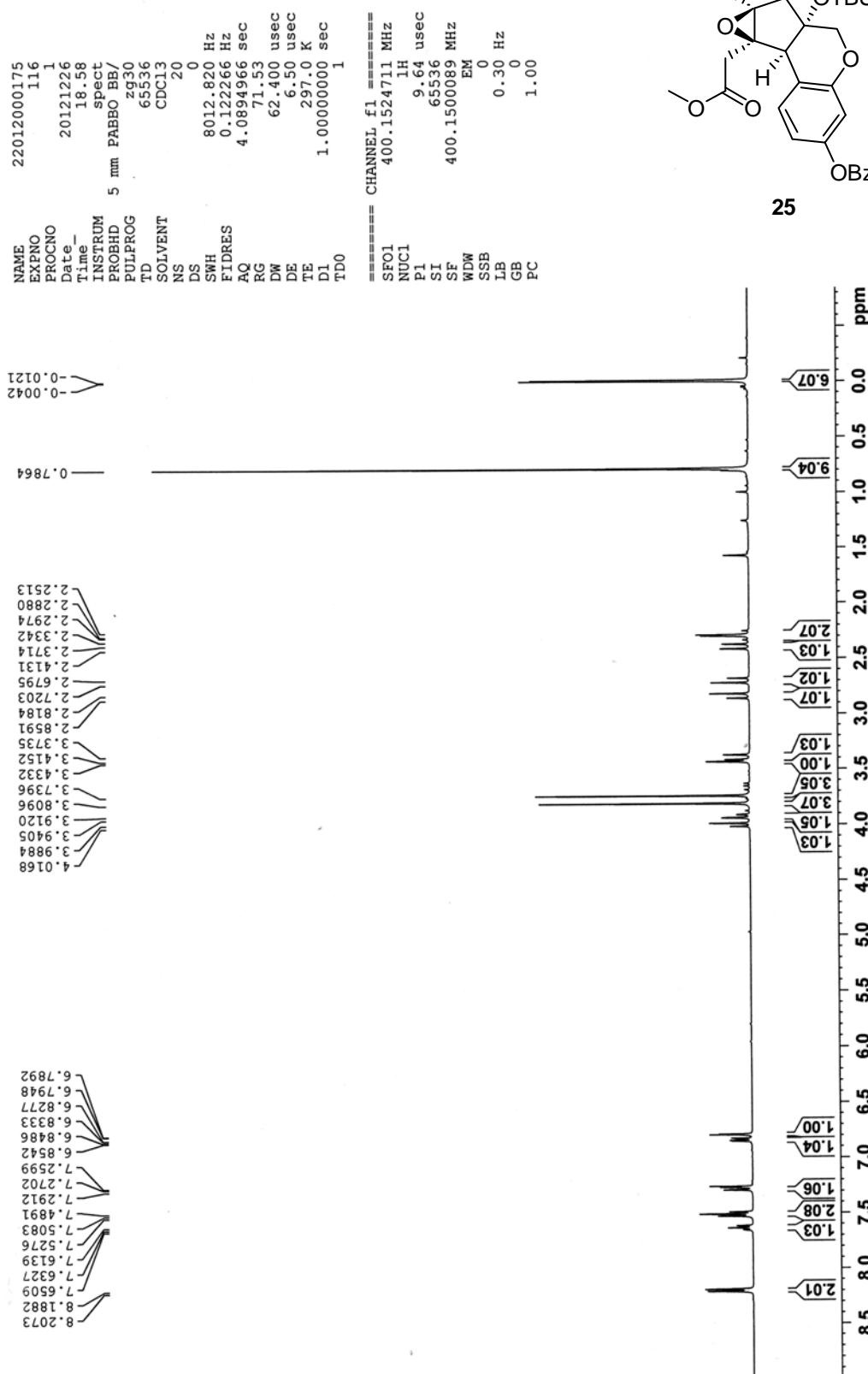


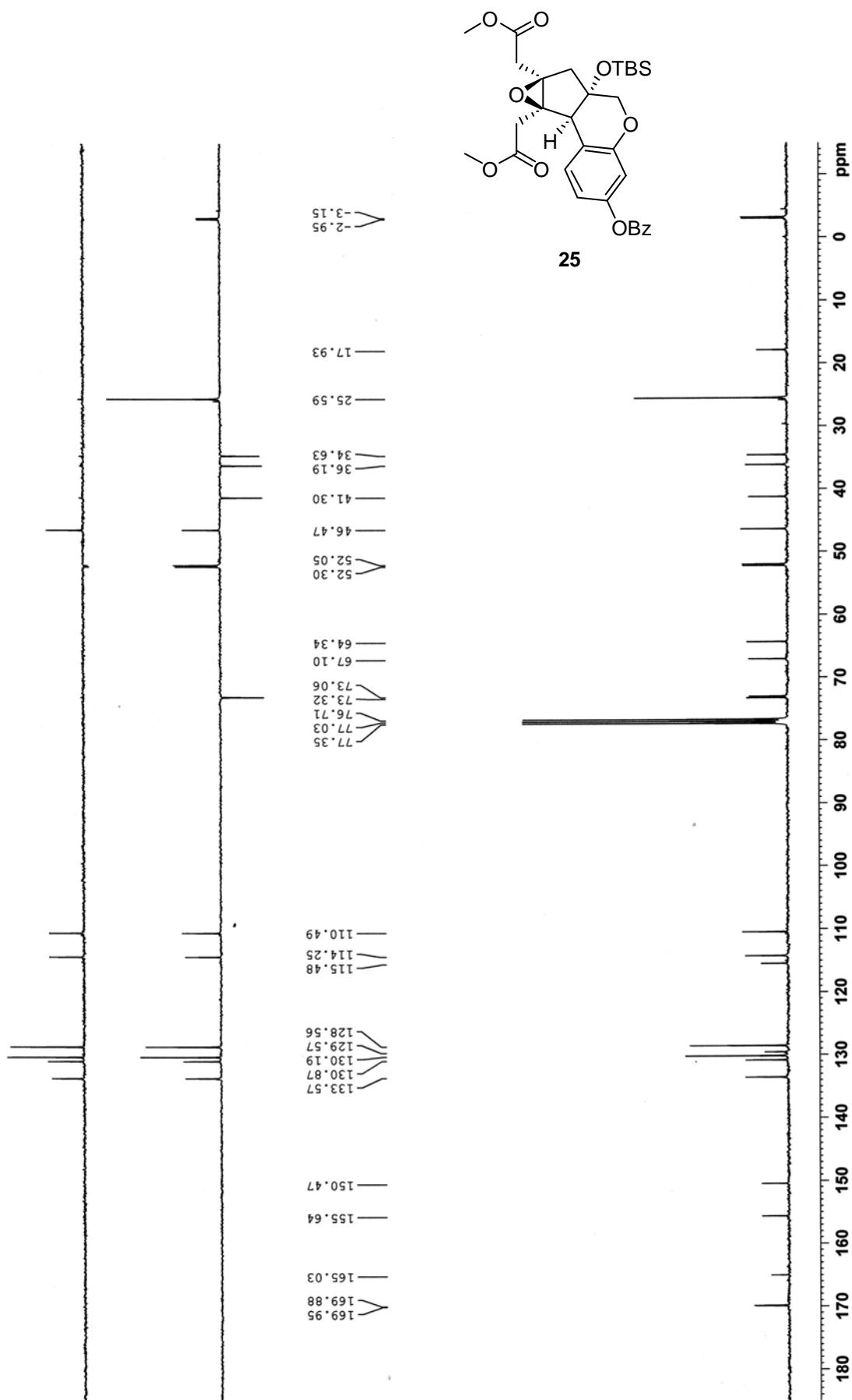


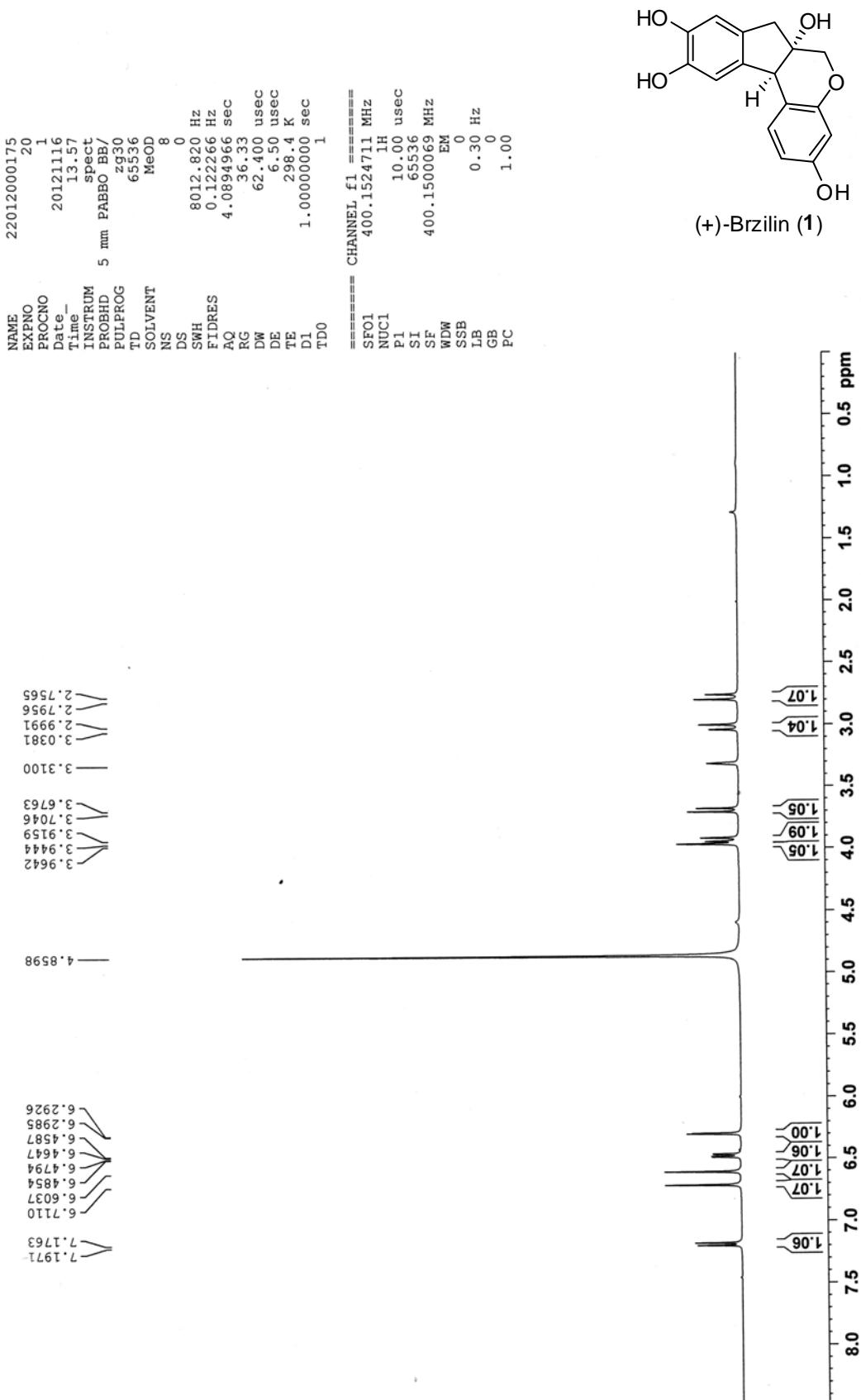


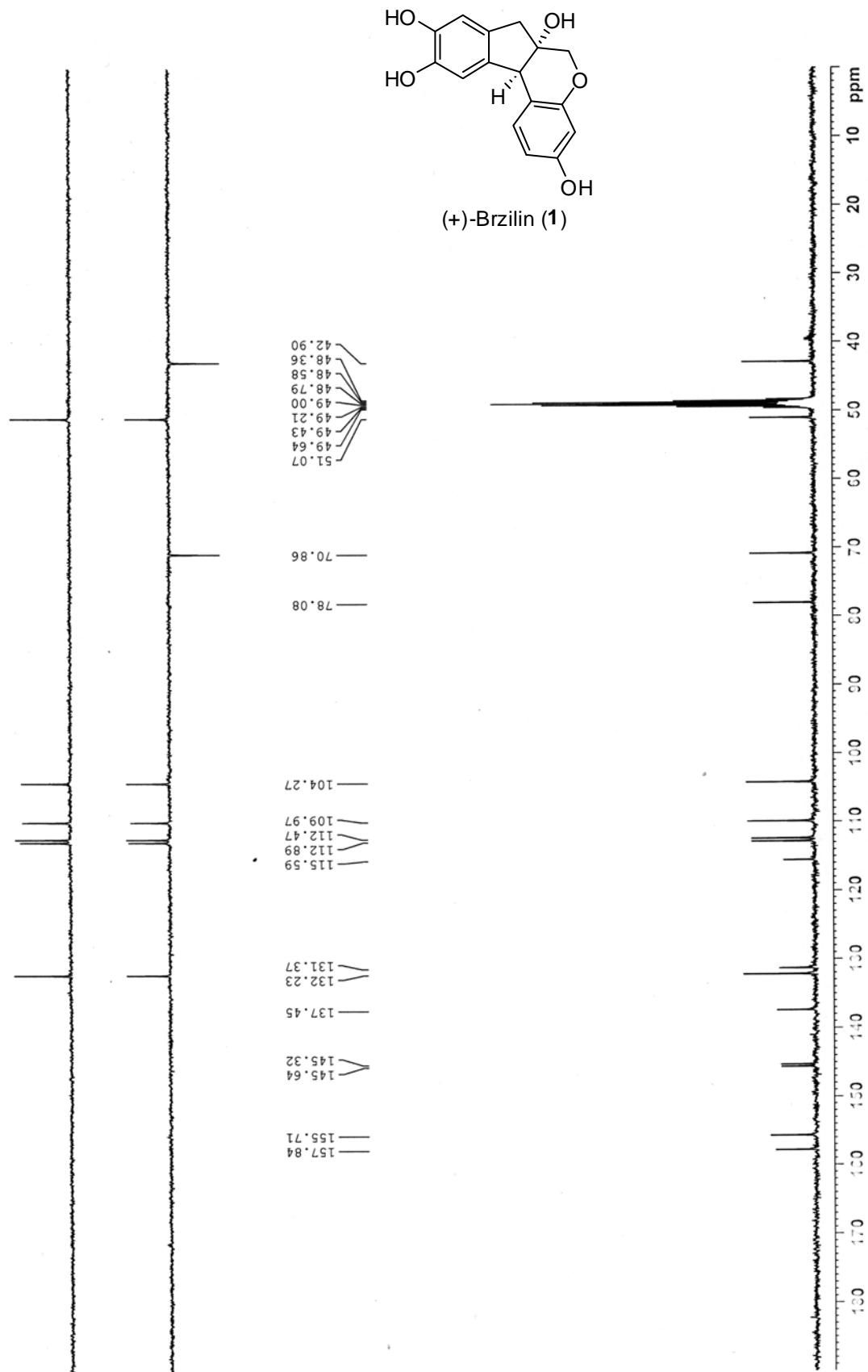


25



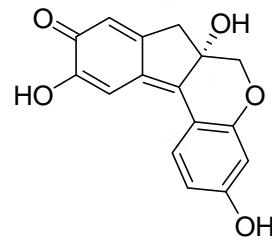




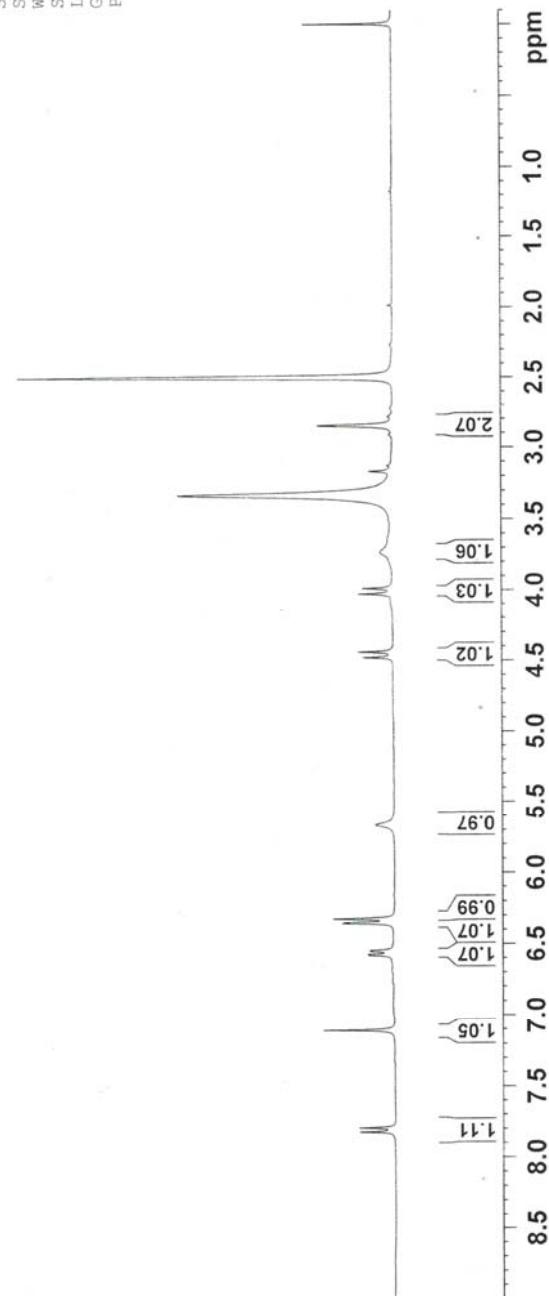


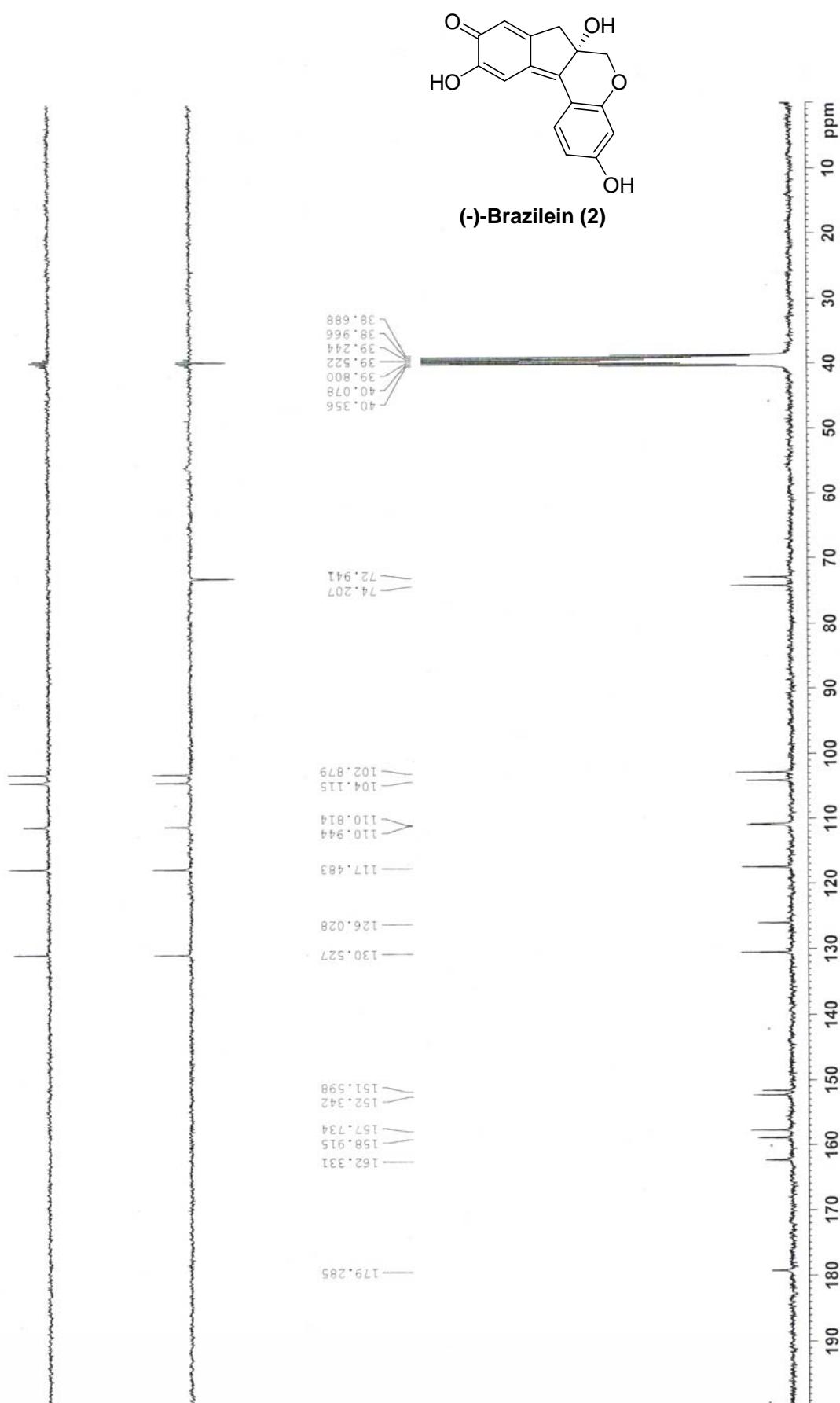
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PROCNO	1	P1	P1
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Time	17.08	SFO1	SFO1
INSTRUM	5 mm QNP av300	SI	SI
PROBHD	1H/13	SF	SF
PULPROG	zg30	WDW	WDW
TD	65536	SSB	SSB
SOLVENT	DMSO	LB	LB
NS	36	GB	GB
DS	0	PC	PC
SWH	6172.839 Hz		
FIDRES	0.094190 Hz		
AQ	5.3984660 sec		
RG	456.1		
DW	81.000 usec		
DE	6.50 usec		
TE	297.3 K		
D1	2.0000000 sec		
TDO	1		

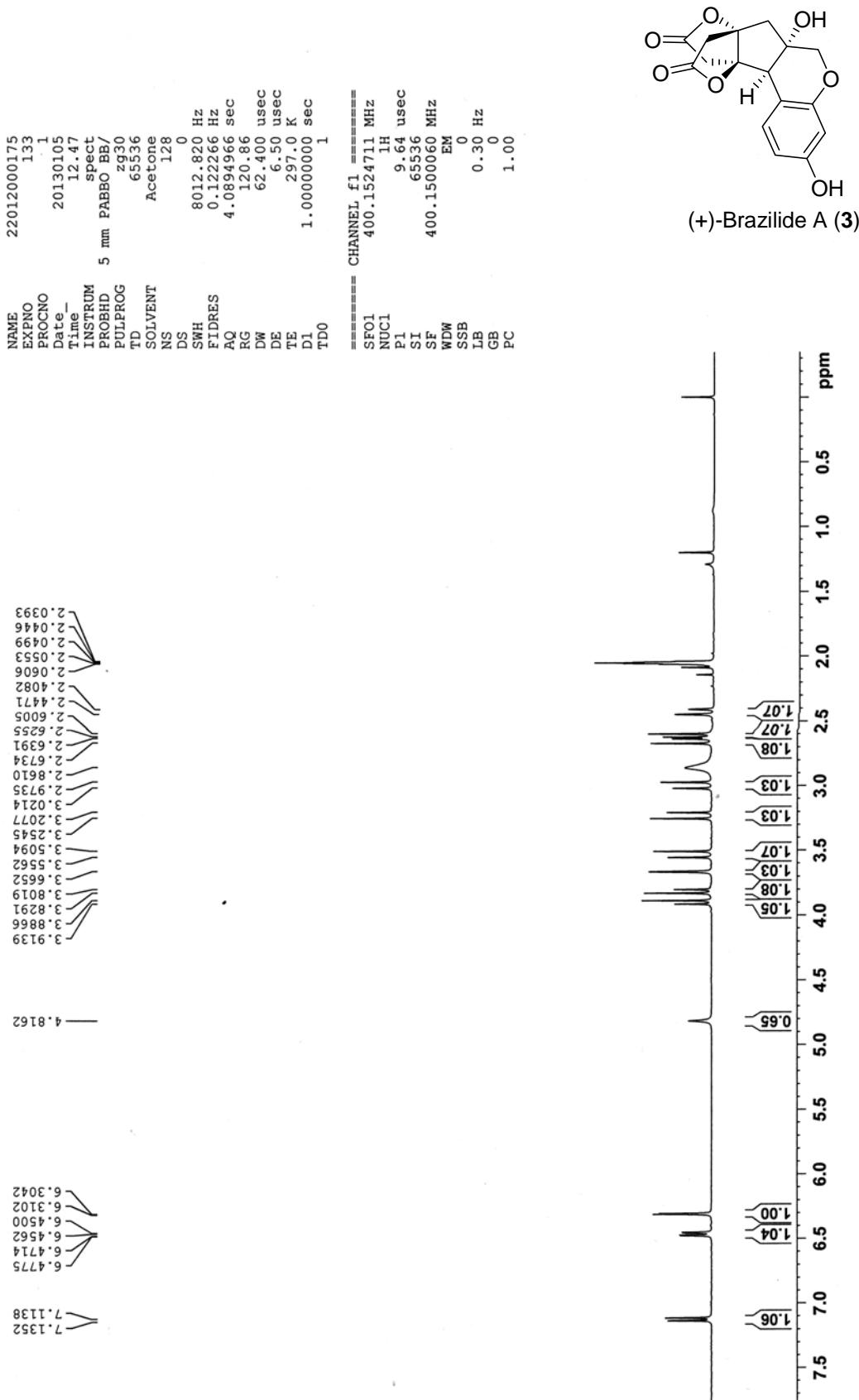
2.500			
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3.331			
3.730			
3.987			
4.026			
4.438			
4.477			
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6.543			
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7.793			
7.822			

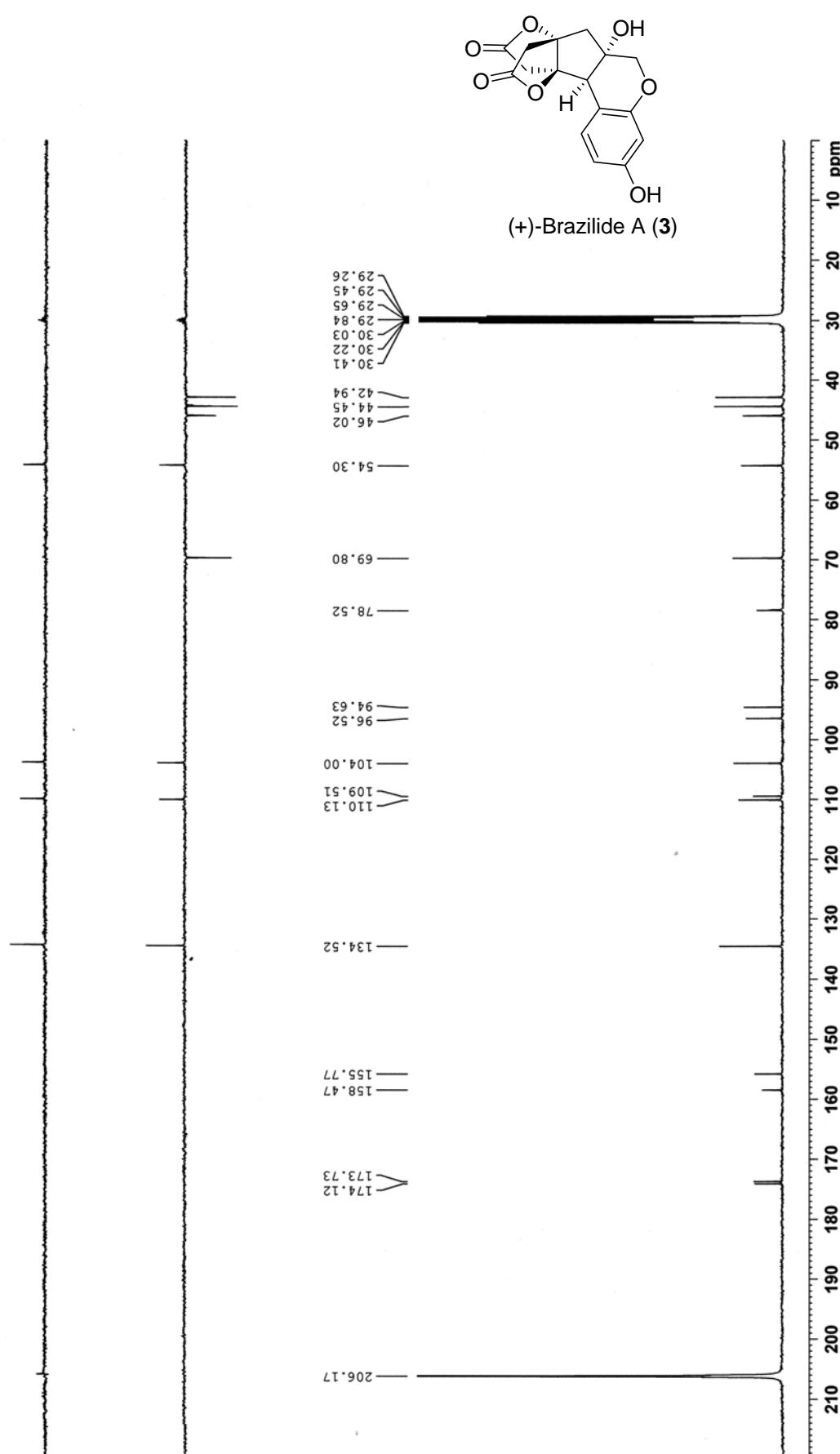


(-)-Brazilein (2)







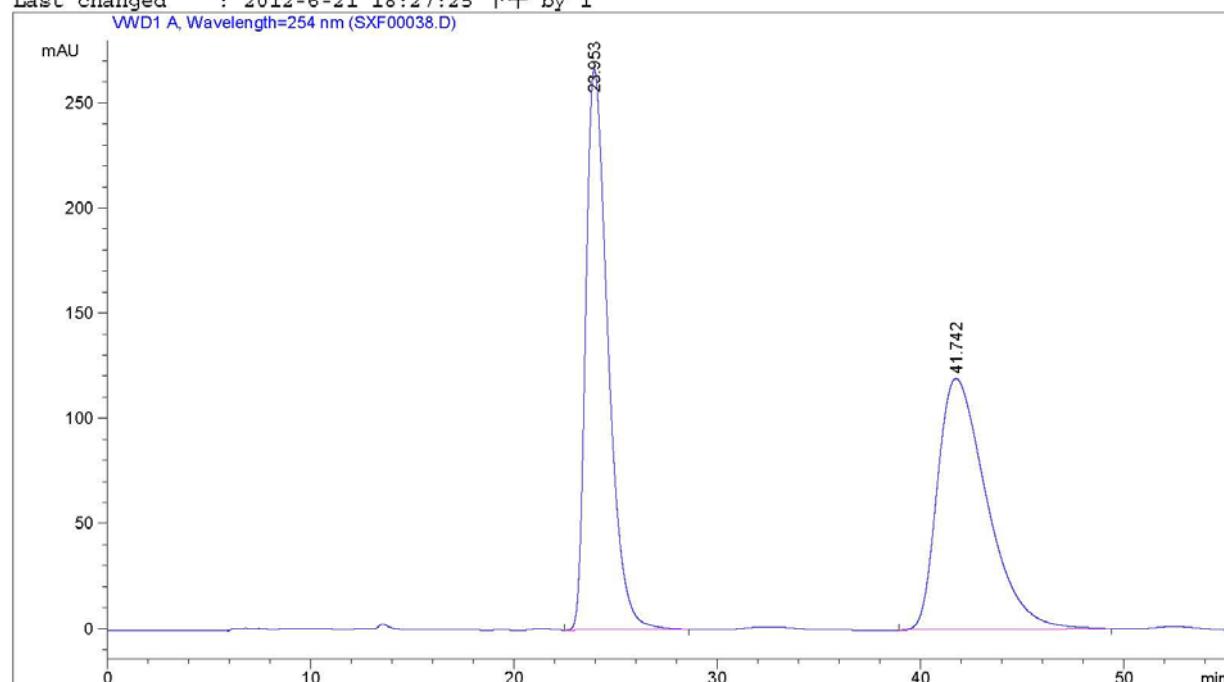


Data File C:\HPCHEM\1\VERIFY\DEFAULT.VAL\SXF00038.D

Sample Name: zero

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Last changed : 2012-7-5 20:20:29 下午 by 1
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Last changed : 2012-6-21 18:27:25 下午 by 1



=====

Area Percent Report

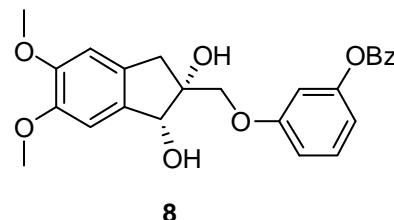
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Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000

Signal 1: VWD1 A, Wavelength=254 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU	*s	Height [mAU]	Area %
1	23.953	VB	1.1658	2.02632e4		266.74408	50.1289
2	41.742	PB	2.6088	2.01590e4		119.49933	49.8711

Totals : 4.04223e4 386.24341



Results obtained with enhanced integrator!

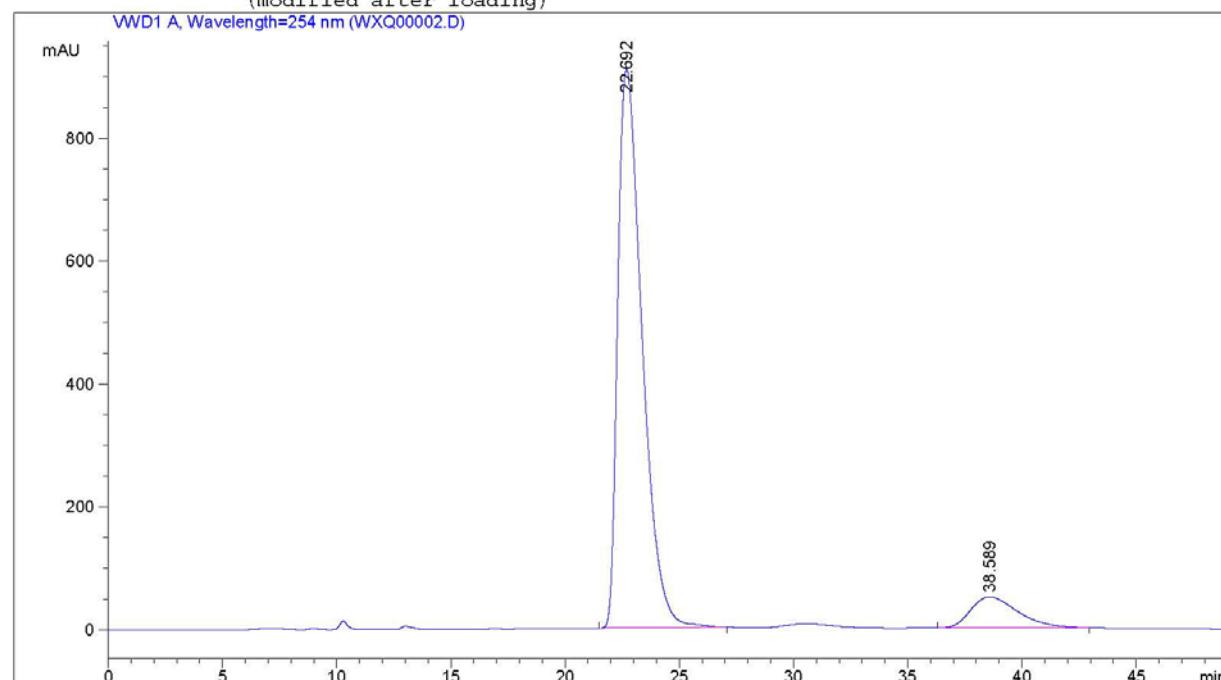
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*** End of Report ***

Data File C:\HPCHEM\1\VERIFY\DEFAULT.VAL\WXQ00002.D

Sample Name: zero

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Injection Date : 2012-9-5 16:11:36 下午
Sample Name : zero Location : -
Acq. Operator :
Method : C:\HPCHEM\1\METHODS\CQ.M
Last changed : 2012-9-5 14:00:57 下午 by 1
(modified after loading)



=====
Area Percent Report
=====

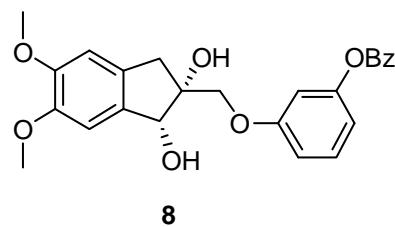
Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000

Signal 1: VWD1 A, Wavelength=254 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU	Height *s	[mAU]	Area %
1	22.692	BB	1.1317	6.72658e4	910.06598	90.2326	
2	38.589	VB	2.0693	7281.29785	50.85722	9.7674	

Totals : 7.45471e4 960.92320

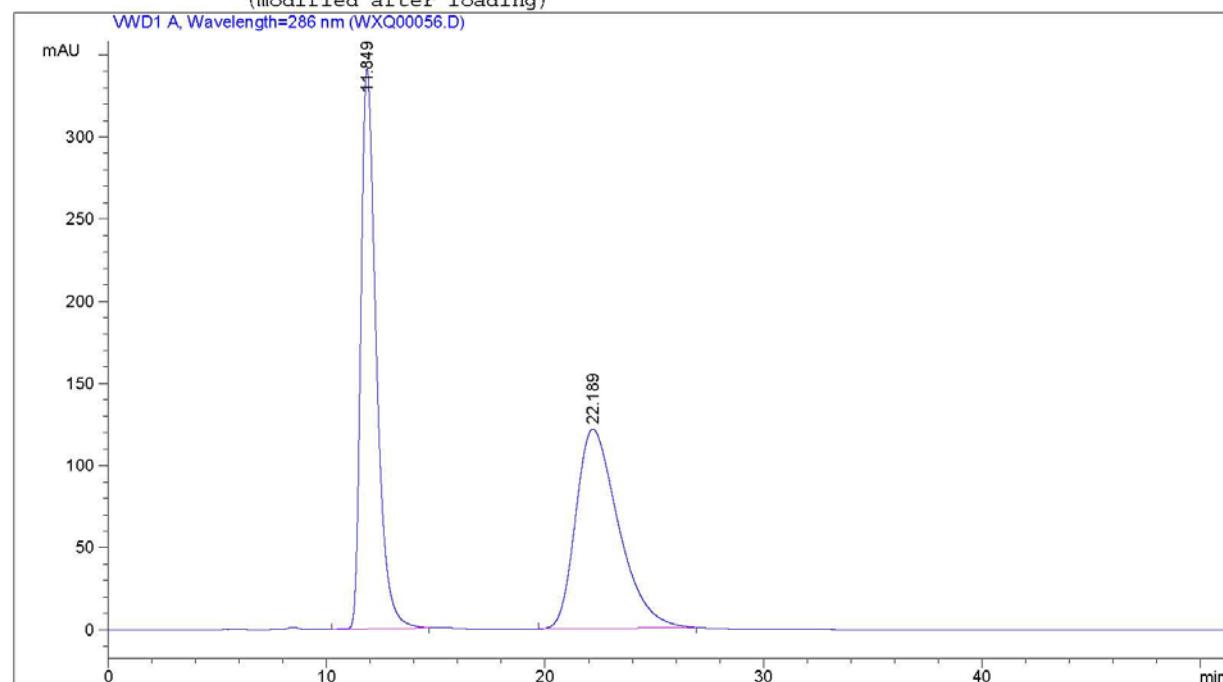
Results obtained with enhanced integrator!



Data File C:\HPCHEM\1\VERIFY\DEFAULT.VAL\WXQ00056.D

Sample Name: zero

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Injection Date : 2013-1-19 15:22:48 下午
Sample Name : zero Location : -
Acq. Operator : 1
Method : C:\HPCHEM\1\METHODS\1.M
Last changed : 2013-1-19 9:50:30 下午 by 1
(modified after loading)



=====
Area Percent Report
=====

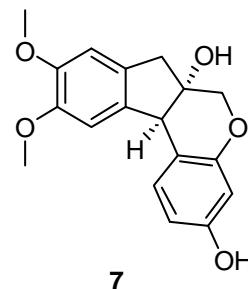
Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000

Signal 1: VWD1 A, Wavelength=286 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU *s	Height [mAU]	Area %
1	11.849	BB	0.7432	1.66939e4	340.88220	50.3205
2	22.189	BB	2.0087	1.64812e4	121.19862	49.6795

Totals : 3.31751e4 462.08083

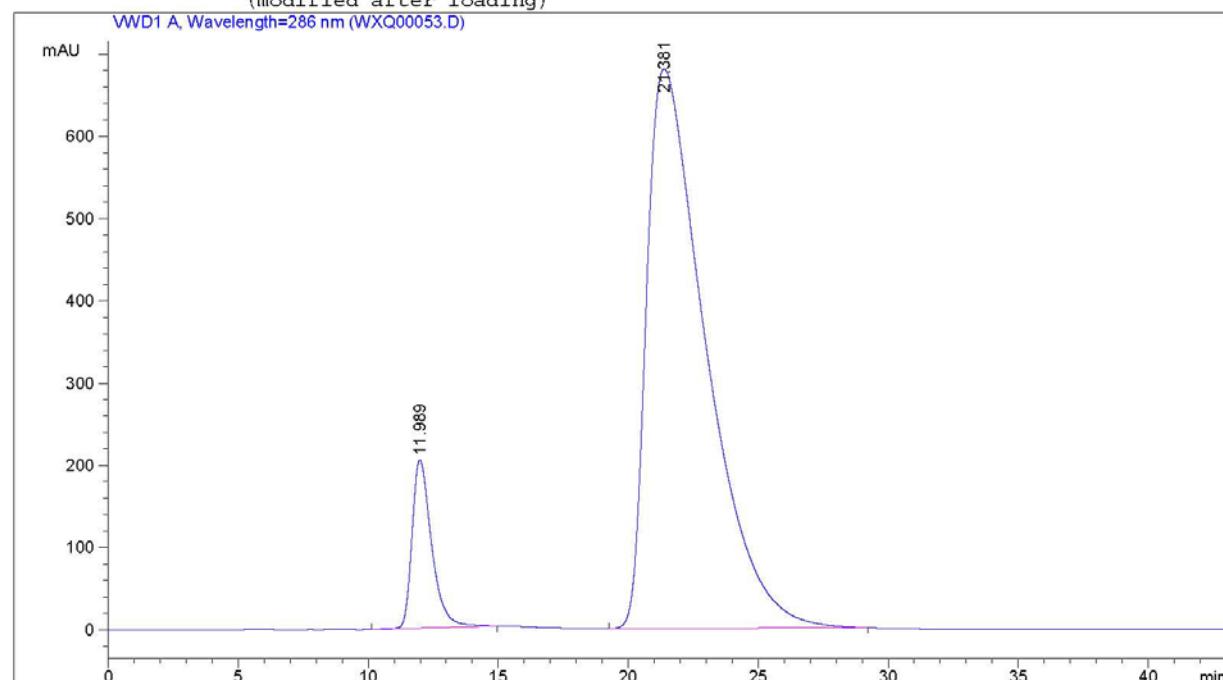
Results obtained with enhanced integrator!



Data File C:\HPCHEM\1\VERIFY\DEFAULT.VAL\WXQ00053.D

Sample Name: zero

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Injection Date : 2013-1-19 13:18:28 下午
Sample Name : zero Location : -
Acq. Operator : 1
Method : C:\HPCHEM\1\METHODS\1.M
Last changed : 2013-1-19 9:50:30 下午 by 1
(modified after loading)



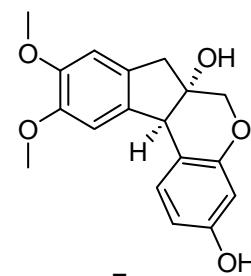
=====
Area Percent Report
=====

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000

Signal 1: VWD1 A, Wavelength=286 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU *s	Height [mAU]	Area %
1	11.989	PB	0.8016	1.07707e4	204.72748	9.1332
2	21.381	BB	2.3459	1.07158e5	680.28156	90.8668

Totals : 1.17929e5 885.00903



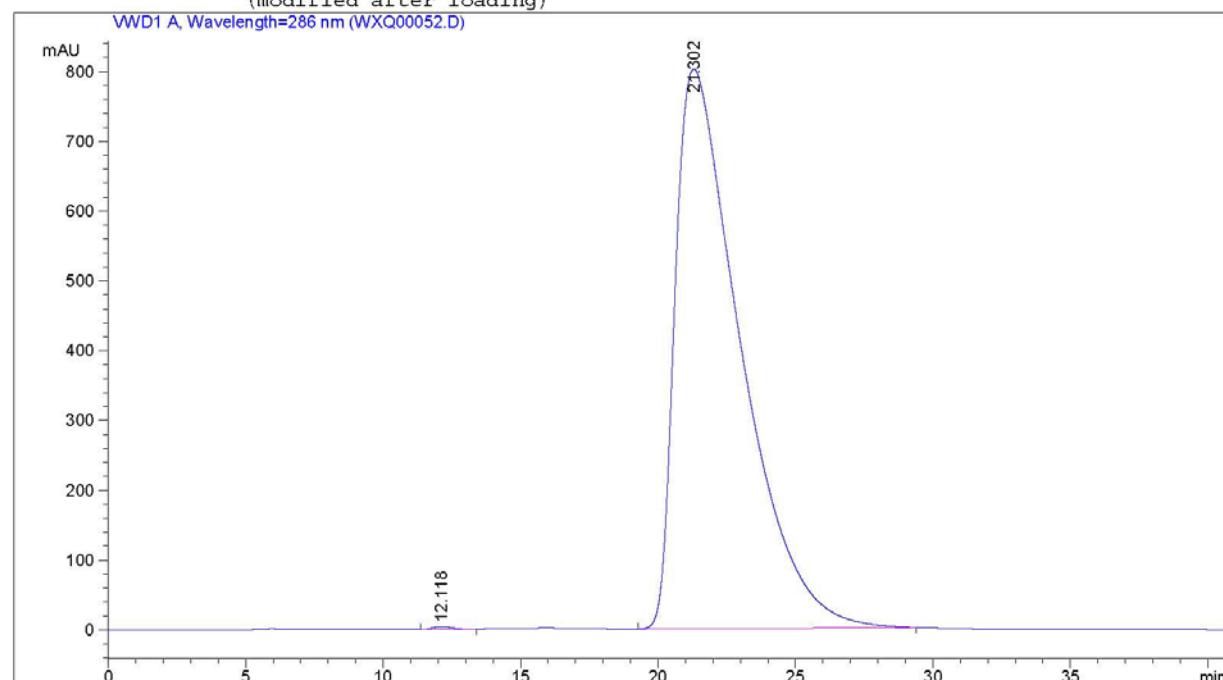
Results obtained with enhanced integrator!

=====*** End of Report ***

Data File C:\HPCHEM\1\VERIFY\DEFAULT.VAL\WXQ00052.D

Sample Name: zero

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Sample Name : zero Location : -
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(modified after loading)

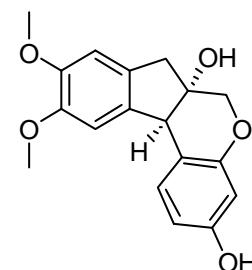


=====
Area Percent Report
=====

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000

Signal 1: VWD1 A, Wavelength=286 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU	Height *s	Area [mAU]	Area %
1	12.118	PP	0.6813	165.04034	3.60430	0.1247	
2	21.302	BB	2.4032	1.32183e5	802.06293	99.8753	
Totals :				1.32348e5		805.66723	



Results obtained with enhanced integrator!

=====

*** End of Report ***

Compound 21 crystal structure analysis

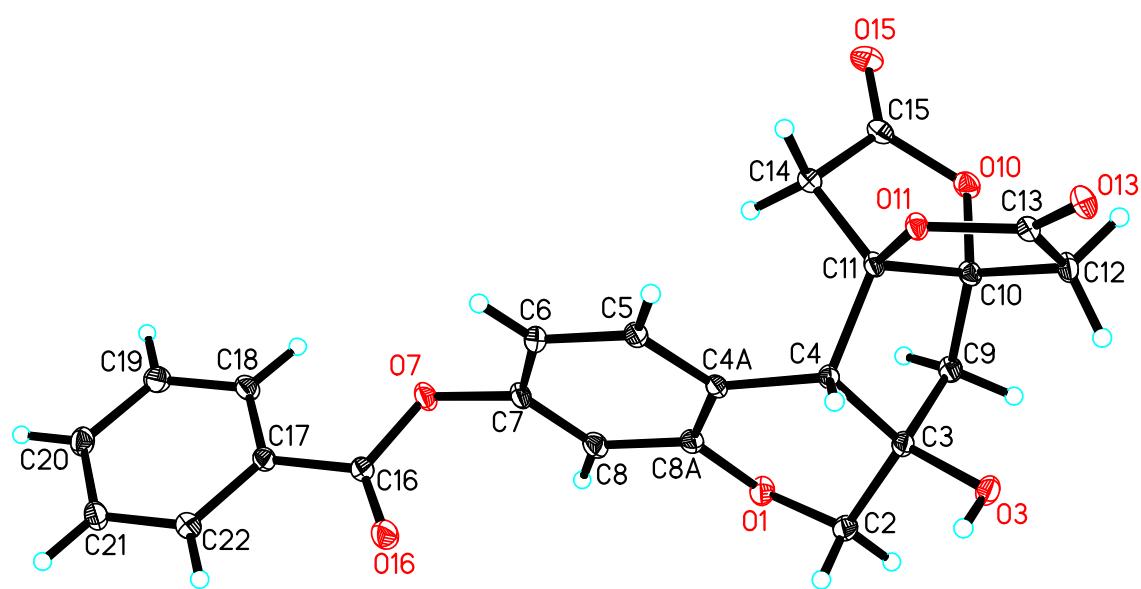


Figure 1 X-crystal structure of compound 21 (zhb_yxd_1)

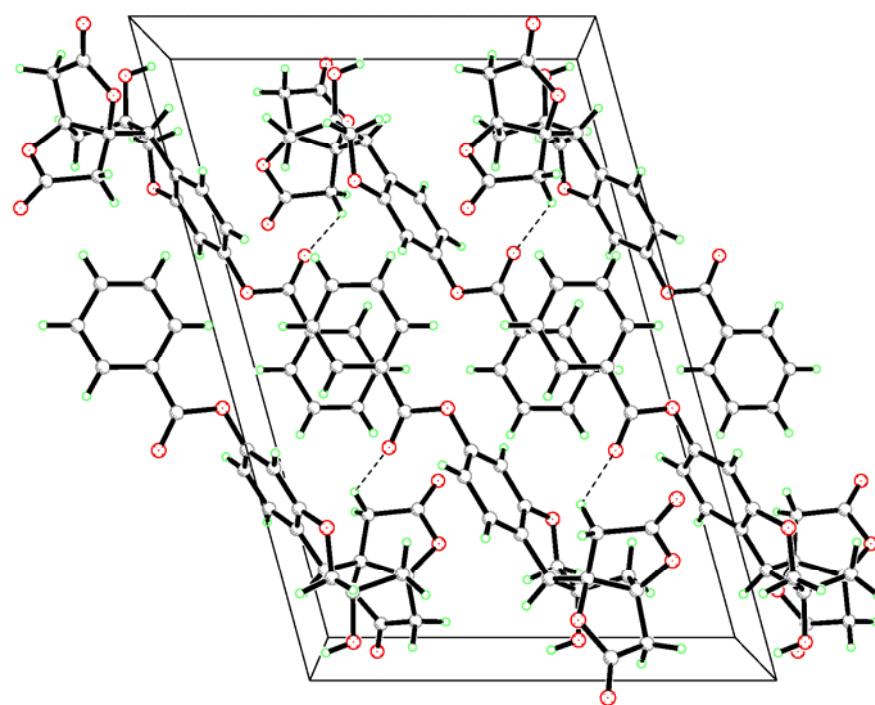


Table 1. Crystal data and structure refinement for zhb_yxd_1.

Identification code	mo_zhb_yxd_1_0m
Empirical formula	C23 H18 O8
Formula weight	422.37
Temperature	100(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, P21/c
Unit cell dimensions	a = 19.082(4) Å alpha = 90 deg. b = 8.1761(16) Å beta = 105.231(3) deg. c = 12.919(3) Å gamma = 90 deg.
Volume	1944.8(6) Å ³
Z, Calculated density	4, 1.443 Mg/m ³
Absorption coefficient	0.110 mm ⁻¹
F(000)	880
Crystal size	0.67 x 0.37 x 0.13 mm
Theta range for data collection	1.11 to 30.45 deg.
Limiting indices	-26<=h<=26, -11<=k<=10, -18<=l<=17
Reflections collected / unique	19227 / 5453 [R(int) = 0.0482]
Completeness to theta = 30.45	92.1 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.9858 and 0.9298
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	5453 / 0 / 281
Goodness-of-fit on F ²	1.079
Final R indices [I>2sigma(I)]	R1 = 0.0469, wR2 = 0.1263
R indices (all data)	R1 = 0.0675, wR2 = 0.1455
Largest diff. peak and hole	0.473 and -0.329 e.Å ⁻³

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for zhb_yxd_1.
U(eq) is defined as one third of the trace of the orthogonalized Uij tensor.

	x	y	z	U(eq)
O(15)	7204(1)	7023(2)	6197(1)	29(1)
O(7)	5938(1)	2554(2)	10477(1)	24(1)
O(13)	10069(1)	7429(1)	8758(1)	22(1)
O(11)	9007(1)	6537(1)	8924(1)	17(1)
O(10)	8183(1)	5421(1)	6553(1)	22(1)
O(1)	7734(1)	774(1)	8784(1)	23(1)
O(16)	6574(1)	1470(2)	12043(1)	28(1)
O(3)	9577(1)	2268(1)	8969(1)	22(1)
C(15)	7654(1)	6277(2)	6844(1)	21(1)
C(14)	7736(1)	6115(2)	8036(1)	18(1)
C(11)	8464(1)	5294(2)	8473(1)	15(1)
C(4)	8541(1)	3832(2)	9266(1)	15(1)
C(4A)	7863(1)	3446(2)	9615(1)	16(1)
C(5)	7584(1)	4591(2)	10201(1)	19(1)
C(6)	6949(1)	4300(2)	10503(1)	21(1)
C(7)	6598(1)	2830(2)	10210(1)	21(1)
C(16)	6000(1)	1873(2)	11457(1)	20(1)
C(17)	5288(1)	1746(2)	11715(1)	18(1)
C(18)	4640(1)	2205(2)	10992(1)	22(1)
C(19)	3997(1)	2152(2)	11300(1)	24(1)
C(20)	4002(1)	1618(2)	12323(1)	24(1)
C(13)	9558(1)	6521(2)	8449(1)	18(1)
C(10)	8670(1)	4633(2)	7465(1)	18(1)
C(12)	9435(1)	5259(2)	7575(1)	20(1)
C(9)	8552(1)	2799(2)	7496(1)	19(1)
C(3)	8801(1)	2377(2)	8689(1)	18(1)
C(8A)	7491(1)	1972(2)	9350(1)	19(1)
C(2)	8506(1)	760(2)	8974(1)	22(1)
C(8)	6853(1)	1657(2)	9643(1)	21(1)
C(21)	4644(1)	1118(2)	13031(1)	24(1)
C(22)	5291(1)	1185(2)	12733(1)	22(1)

Table 3. Bond lengths [Å] and angles [deg] for zhb_yxd_1.

O(15)-C(15)	1.1949(19)
O(7)-C(16)	1.3592(19)
O(7)-C(7)	1.4102(18)
O(13)-C(13)	1.2077(18)
O(11)-C(13)	1.3502(17)
O(11)-C(11)	1.4591(16)
O(10)-C(15)	1.3600(19)
O(10)-C(10)	1.4461(17)
O(1)-C(8A)	1.3741(18)
O(1)-C(2)	1.4293(19)
O(16)-C(16)	1.2032(18)
O(3)-C(3)	1.4309(17)
O(3)-H(3)	0.8400
C(15)-C(14)	1.512(2)
C(14)-C(11)	1.512(2)
C(14)-H(14A)	0.9900
C(14)-H(14B)	0.9900
C(11)-C(10)	1.5539(19)
C(11)-C(4)	1.555(2)
C(4)-C(4A)	1.5113(19)
C(4)-C(3)	1.553(2)
C(4)-H(4)	1.0000
C(4A)-C(8A)	1.394(2)
C(4A)-C(5)	1.395(2)
C(5)-C(6)	1.387(2)
C(5)-H(5)	0.9500
C(6)-C(7)	1.380(2)
C(6)-H(6)	0.9500
C(7)-C(8)	1.371(2)
C(16)-C(17)	1.484(2)
C(17)-C(18)	1.391(2)
C(17)-C(22)	1.392(2)
C(18)-C(19)	1.387(2)
C(18)-H(18)	0.9500
C(19)-C(20)	1.390(2)
C(19)-H(19)	0.9500
C(20)-C(21)	1.383(2)
C(20)-H(20)	0.9500
C(13)-C(12)	1.502(2)
C(10)-C(12)	1.517(2)
C(10)-C(9)	1.518(2)
C(12)-H(12A)	0.9900
C(12)-H(12B)	0.9900
C(9)-C(3)	1.528(2)
C(9)-H(9A)	0.9900
C(9)-H(9B)	0.9900
C(3)-C(2)	1.519(2)
C(8A)-C(8)	1.391(2)
C(2)-H(2A)	0.9900
C(2)-H(2B)	0.9900
C(8)-H(8)	0.9500
C(21)-C(22)	1.387(2)
C(21)-H(21)	0.9500
C(22)-H(22)	0.9500

C(16)-O(7)-C(7)	115.41(12)
C(13)-O(11)-C(11)	111.57(11)
C(15)-O(10)-C(10)	111.72(11)
C(8A)-O(1)-C(2)	112.99(12)
C(3)-O(3)-H(3)	109.5
O(15)-C(15)-O(10)	121.17(14)
O(15)-C(15)-C(14)	128.37(15)
O(10)-C(15)-C(14)	110.46(12)
C(11)-C(14)-C(15)	105.04(12)
C(11)-C(14)-H(14A)	110.7
C(15)-C(14)-H(14A)	110.7
C(11)-C(14)-H(14B)	110.7
C(15)-C(14)-H(14B)	110.7
H(14A)-C(14)-H(14B)	108.8
O(11)-C(11)-C(14)	108.98(11)
O(11)-C(11)-C(10)	105.52(10)
C(14)-C(11)-C(10)	104.54(11)
O(11)-C(11)-C(4)	109.76(11)
C(14)-C(11)-C(4)	120.40(11)
C(10)-C(11)-C(4)	106.53(11)
C(4A)-C(4)-C(3)	112.94(11)
C(4A)-C(4)-C(11)	114.64(11)
C(3)-C(4)-C(11)	105.03(11)
C(4A)-C(4)-H(4)	108.0
C(3)-C(4)-H(4)	108.0
C(11)-C(4)-H(4)	108.0
C(8A)-C(4A)-C(5)	118.14(13)
C(8A)-C(4A)-C(4)	121.83(12)
C(5)-C(4A)-C(4)	120.02(13)
C(6)-C(5)-C(4A)	121.53(14)
C(6)-C(5)-H(5)	119.2
C(4A)-C(5)-H(5)	119.2
C(7)-C(6)-C(5)	117.97(14)
C(7)-C(6)-H(6)	121.0
C(5)-C(6)-H(6)	121.0
C(8)-C(7)-C(6)	122.81(14)
C(8)-C(7)-O(7)	118.96(14)
C(6)-C(7)-O(7)	118.20(14)
O(16)-C(16)-O(7)	122.69(14)
O(16)-C(16)-C(17)	125.17(14)
O(7)-C(16)-C(17)	112.11(12)
C(18)-C(17)-C(22)	120.36(14)
C(18)-C(17)-C(16)	122.32(14)
C(22)-C(17)-C(16)	117.30(13)
C(19)-C(18)-C(17)	119.74(14)
C(19)-C(18)-H(18)	120.1
C(17)-C(18)-H(18)	120.1
C(18)-C(19)-C(20)	119.81(14)
C(18)-C(19)-H(19)	120.1
C(20)-C(19)-H(19)	120.1
C(21)-C(20)-C(19)	120.34(14)
C(21)-C(20)-H(20)	119.8
C(19)-C(20)-H(20)	119.8
O(13)-C(13)-O(11)	120.14(13)
O(13)-C(13)-C(12)	128.62(13)
O(11)-C(13)-C(12)	111.23(12)
O(10)-C(10)-C(12)	108.89(12)
O(10)-C(10)-C(9)	113.16(12)
C(12)-C(10)-C(9)	118.46(13)

O(10)-C(10)-C(11)	106.11(11)
C(12)-C(10)-C(11)	104.65(11)
C(9)-C(10)-C(11)	104.40(11)
C(13)-C(12)-C(10)	105.02(12)
C(13)-C(12)-H(12A)	110.7
C(10)-C(12)-H(12A)	110.7
C(13)-C(12)-H(12B)	110.7
C(10)-C(12)-H(12B)	110.7
H(12A)-C(12)-H(12B)	108.8
C(10)-C(9)-C(3)	103.94(12)
C(10)-C(9)-H(9A)	111.0
C(3)-C(9)-H(9A)	111.0
C(10)-C(9)-H(9B)	111.0
C(3)-C(9)-H(9B)	111.0
H(9A)-C(9)-H(9B)	109.0
O(3)-C(3)-C(2)	108.01(12)
O(3)-C(3)-C(9)	107.20(12)
C(2)-C(3)-C(9)	113.82(13)
O(3)-C(3)-C(4)	111.49(12)
C(2)-C(3)-C(4)	111.40(12)
C(9)-C(3)-C(4)	104.88(12)
O(1)-C(8A)-C(8)	116.99(13)
O(1)-C(8A)-C(4A)	121.77(13)
C(8)-C(8A)-C(4A)	121.24(14)
O(1)-C(2)-C(3)	112.51(12)
O(1)-C(2)-H(2A)	109.1
C(3)-C(2)-H(2A)	109.1
O(1)-C(2)-H(2B)	109.1
C(3)-C(2)-H(2B)	109.1
H(2A)-C(2)-H(2B)	107.8
C(7)-C(8)-C(8A)	118.30(14)
C(7)-C(8)-H(8)	120.9
C(8A)-C(8)-H(8)	120.9
C(20)-C(21)-C(22)	120.23(15)
C(20)-C(21)-H(21)	119.9
C(22)-C(21)-H(21)	119.9
C(21)-C(22)-C(17)	119.48(14)
C(21)-C(22)-H(22)	120.3
C(17)-C(22)-H(22)	120.3

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{Å}^2 \times 10^3$) for zhb_yxd_1.

The anisotropic displacement factor exponent takes the form:

$$-2 \pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$$

	U11	U22	U33	U23	U13	U12
O(15)	33(1)	27(1)	25(1)	9(1)	5(1)	4(1)
O(7)	19(1)	33(1)	23(1)	9(1)	10(1)	0(1)
O(13)	21(1)	23(1)	23(1)	0(1)	9(1)	-7(1)
O(11)	18(1)	15(1)	20(1)	-3(1)	9(1)	-3(1)
O(10)	27(1)	23(1)	16(1)	2(1)	7(1)	0(1)
O(1)	27(1)	15(1)	32(1)	-6(1)	15(1)	-4(1)
O(16)	21(1)	39(1)	27(1)	11(1)	9(1)	5(1)
O(3)	20(1)	26(1)	23(1)	-1(1)	9(1)	4(1)
C(15)	23(1)	17(1)	22(1)	3(1)	7(1)	-3(1)
C(14)	19(1)	17(1)	20(1)	2(1)	7(1)	0(1)
C(11)	17(1)	14(1)	17(1)	-1(1)	8(1)	-3(1)
C(4)	17(1)	13(1)	17(1)	-1(1)	8(1)	-1(1)
C(4A)	20(1)	15(1)	17(1)	2(1)	9(1)	0(1)
C(5)	23(1)	16(1)	20(1)	0(1)	10(1)	0(1)
C(6)	24(1)	21(1)	21(1)	2(1)	12(1)	4(1)
C(7)	18(1)	25(1)	21(1)	6(1)	10(1)	0(1)
C(16)	21(1)	18(1)	21(1)	2(1)	9(1)	1(1)
C(17)	20(1)	16(1)	21(1)	1(1)	9(1)	-1(1)
C(18)	22(1)	23(1)	22(1)	4(1)	8(1)	-2(1)
C(19)	20(1)	25(1)	29(1)	4(1)	6(1)	0(1)
C(20)	22(1)	26(1)	29(1)	0(1)	12(1)	-1(1)
C(13)	18(1)	17(1)	19(1)	3(1)	8(1)	0(1)
C(10)	22(1)	17(1)	15(1)	0(1)	8(1)	-1(1)
C(12)	22(1)	20(1)	23(1)	-2(1)	14(1)	-3(1)
C(9)	22(1)	16(1)	19(1)	-3(1)	9(1)	-2(1)
C(3)	19(1)	16(1)	21(1)	-1(1)	10(1)	1(1)
C(8A)	23(1)	15(1)	20(1)	0(1)	10(1)	0(1)
C(2)	26(1)	14(1)	30(1)	0(1)	14(1)	2(1)
C(8)	22(1)	19(1)	24(1)	2(1)	9(1)	-4(1)
C(21)	28(1)	27(1)	21(1)	1(1)	12(1)	0(1)
C(22)	23(1)	23(1)	21(1)	2(1)	7(1)	0(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{Å}^2 \times 10^3$) for zhb_yxd_1.

	x	y	z	U(eq)
H(3)	9735	2189	9638	33
H(14A)	7729	7204	8369	22
H(14B)	7340	5440	8176	22
H(4)	8937	4101	9922	18
H(5)	7835	5594	10400	23
H(6)	6761	5089	10898	25
H(18)	4638	2553	10289	26
H(19)	3554	2480	10813	29
H(20)	3563	1596	12538	29
H(12A)	9793	4360	7774	24
H(12B)	9477	5753	6895	24
H(9A)	8846	2209	7088	22
H(9B)	8034	2519	7195	22
H(2A)	8646	-123	8544	27
H(2B)	8730	517	9740	27
H(8)	6600	654	9455	26
H(21)	4642	729	13722	29
H(22)	5732	849	13221	26

Table 6. Torsion angles [deg] for zhb_yxd_1.

C(10)-O(10)-C(15)-O(15)	179.27(14)
C(10)-O(10)-C(15)-C(14)	-0.80(17)
O(15)-C(15)-C(14)-C(11)	-170.30(16)
O(10)-C(15)-C(14)-C(11)	9.77(16)
C(13)-O(11)-C(11)-C(14)	-121.79(12)
C(13)-O(11)-C(11)-C(10)	-10.01(14)
C(13)-O(11)-C(11)-C(4)	104.41(13)
C(15)-C(14)-C(11)-O(11)	98.48(13)
C(15)-C(14)-C(11)-C(10)	-13.95(14)
C(15)-C(14)-C(11)-C(4)	-133.48(13)
O(11)-C(11)-C(4)-C(4A)	122.46(12)
C(14)-C(11)-C(4)-C(4A)	-5.23(18)
C(10)-C(11)-C(4)-C(4A)	-123.77(12)
O(11)-C(11)-C(4)-C(3)	-112.99(12)
C(14)-C(11)-C(4)-C(3)	119.32(13)
C(10)-C(11)-C(4)-C(3)	0.79(14)
C(3)-C(4)-C(4A)-C(8A)	-5.07(19)
C(11)-C(4)-C(4A)-C(8A)	115.19(15)
C(3)-C(4)-C(4A)-C(5)	176.14(13)
C(11)-C(4)-C(4A)-C(5)	-63.60(17)
C(8A)-C(4A)-C(5)-C(6)	-1.1(2)
C(4)-C(4A)-C(5)-C(6)	177.72(13)
C(4A)-C(5)-C(6)-C(7)	0.4(2)
C(5)-C(6)-C(7)-C(8)	0.3(2)
C(5)-C(6)-C(7)-O(7)	-177.59(13)
C(16)-O(7)-C(7)-C(8)	93.40(17)
C(16)-O(7)-C(7)-C(6)	-88.64(17)
C(7)-O(7)-C(16)-O(16)	-2.9(2)
C(7)-O(7)-C(16)-C(17)	175.36(13)
O(16)-C(16)-C(17)-C(18)	-178.66(17)
O(7)-C(16)-C(17)-C(18)	3.1(2)
O(16)-C(16)-C(17)-C(22)	3.0(2)
O(7)-C(16)-C(17)-C(22)	-175.15(14)
C(22)-C(17)-C(18)-C(19)	2.1(2)
C(16)-C(17)-C(18)-C(19)	-176.13(14)
C(17)-C(18)-C(19)-C(20)	-1.0(2)
C(18)-C(19)-C(20)-C(21)	-0.9(3)
C(11)-O(11)-C(13)-O(13)	-176.90(13)
C(11)-O(11)-C(13)-C(12)	1.69(16)
C(15)-O(10)-C(10)-C(12)	-120.47(13)
C(15)-O(10)-C(10)-C(9)	105.58(14)
C(15)-O(10)-C(10)-C(11)	-8.31(15)
O(11)-C(11)-C(10)-O(10)	-101.08(12)
C(14)-C(11)-C(10)-O(10)	13.80(14)
C(4)-C(11)-C(10)-O(10)	142.28(11)
O(11)-C(11)-C(10)-C(12)	13.99(14)
C(14)-C(11)-C(10)-C(12)	128.87(12)
C(4)-C(11)-C(10)-C(12)	-102.65(13)
O(11)-C(11)-C(10)-C(9)	139.13(11)
C(14)-C(11)-C(10)-C(9)	-105.98(13)
C(4)-C(11)-C(10)-C(9)	22.50(14)
O(13)-C(13)-C(12)-C(10)	-174.01(15)
O(11)-C(13)-C(12)-C(10)	7.56(16)
O(10)-C(10)-C(12)-C(13)	100.22(13)
C(9)-C(10)-C(12)-C(13)	-128.62(14)

C(11)-C(10)-C(12)-C(13)	-12.90(15)
O(10)-C(10)-C(9)-C(3)	-152.30(11)
C(12)-C(10)-C(9)-C(3)	78.48(15)
C(11)-C(10)-C(9)-C(3)	-37.38(14)
C(10)-C(9)-C(3)-O(3)	-80.42(13)
C(10)-C(9)-C(3)-C(2)	160.20(12)
C(10)-C(9)-C(3)-C(4)	38.20(14)
C(4A)-C(4)-C(3)-O(3)	-142.38(12)
C(11)-C(4)-C(3)-O(3)	92.00(13)
C(4A)-C(4)-C(3)-C(2)	-21.64(17)
C(11)-C(4)-C(3)-C(2)	-147.26(12)
C(4A)-C(4)-C(3)-C(9)	101.93(13)
C(11)-C(4)-C(3)-C(9)	-23.69(14)
C(2)-O(1)-C(8A)-C(8)	-150.22(14)
C(2)-O(1)-C(8A)-C(4A)	30.27(19)
C(5)-C(4A)-C(8A)-O(1)	-179.30(13)
C(4)-C(4A)-C(8A)-O(1)	1.9(2)
C(5)-C(4A)-C(8A)-C(8)	1.2(2)
C(4)-C(4A)-C(8A)-C(8)	-177.61(13)
C(8A)-O(1)-C(2)-C(3)	-58.59(17)
O(3)-C(3)-C(2)-O(1)	176.51(12)
C(9)-C(3)-C(2)-O(1)	-64.58(17)
C(4)-C(3)-C(2)-O(1)	53.75(17)
C(6)-C(7)-C(8)-C(8A)	-0.2(2)
O(7)-C(7)-C(8)-C(8A)	177.66(13)
O(1)-C(8A)-C(8)-C(7)	179.92(14)
C(4A)-C(8A)-C(8)-C(7)	-0.6(2)
C(19)-C(20)-C(21)-C(22)	1.6(3)
C(20)-C(21)-C(22)-C(17)	-0.5(2)
C(18)-C(17)-C(22)-C(21)	-1.4(2)
C(16)-C(17)-C(22)-C(21)	176.94(14)

Symmetry transformations used to generate equivalent atoms:

Table 7. Hydrogen bonds for zhb_yxd_1 [Å and deg.].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O(3)-H(3)...O(13) ^{#1}	0.84	2.03	2.8460(17)	163.4

Symmetry transformations used to generate equivalent atoms:

#1 -x+2,-y+1,-z+2

(+)-Brazilide A (3) crystal structure analysis

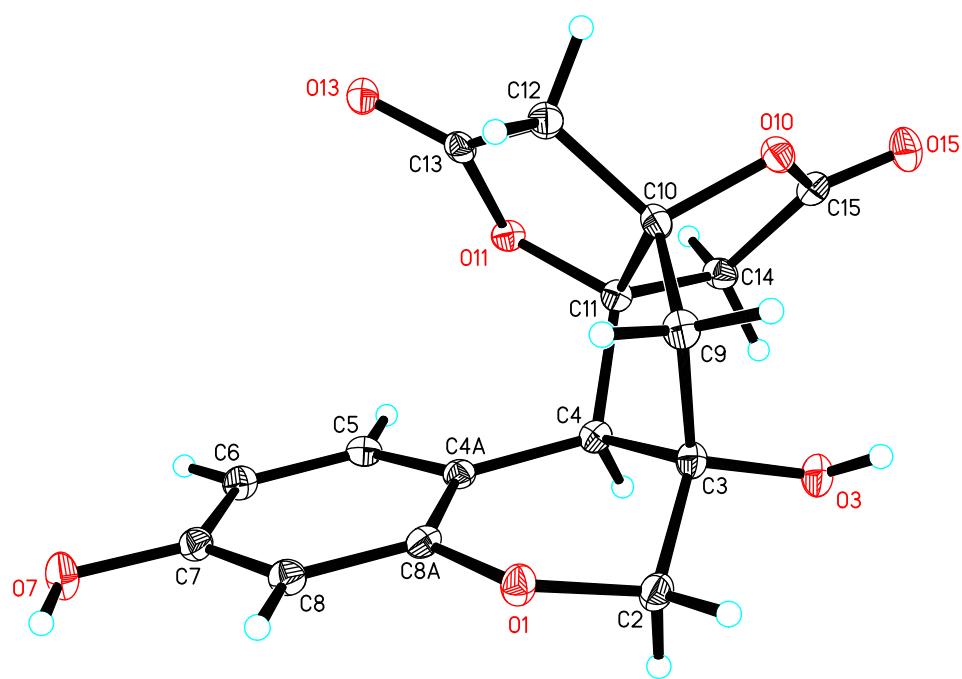


Figure 1 Absolute configuration of (+)-Brazilide A (3) (zhb_w_1)

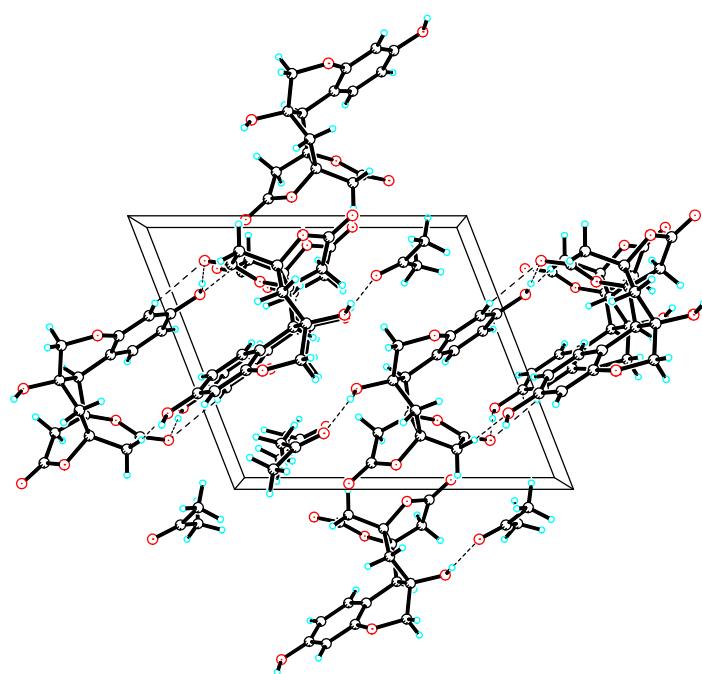


Table 1. Crystal data and structure refinement for zhb_w_1.

Identification code	cu_zhb_w_1_0m
Empirical formula	C19 H20 O8
Formula weight	376.35
Temperature	100(2) K
Wavelength	1.54178 Å
Crystal system, space group	Monoclinic, P 21
Unit cell dimensions	a = 10.9939(2) Å alpha = 90 deg. b = 6.73910(10) Å beta = 111.3280(10) deg. c = 12.6864(3) Å gamma = 90 deg.
Volume	875.55(3) Å^3
Z, Calculated density	2, 1.428 Mg/m^3
Absorption coefficient	0.949 mm^-1
F(000)	396
Crystal size	0.64 x 0.32 x 0.08 mm
Theta range for data collection	3.74 to 68.19 deg.
Limiting indices	-12<=h<=12, -7<=k<=6, -15<=l<=14
Reflections collected / unique	7274 / 2361 [R(int) = 0.0330]
Completeness to theta = 68.19	92.8 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.9280 and 0.5819
Refinement method	Full-matrix least-squares on F^2
Data / restraints / parameters	2361 / 1 / 248
Goodness-of-fit on F^2	1.097
Final R indices [I>2sigma(I)]	R1 = 0.0278, wR2 = 0.0718
R indices (all data)	R1 = 0.0279, wR2 = 0.0719
Absolute structure parameter	0.36(14)
Largest diff. peak and hole	0.153 and -0.225 e.Å^-3

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for zhb_w_1.
U(eq) is defined as one third of the trace of the orthogonalized Uij tensor.

	x	y	z	U(eq)
O(7)	2793(1)	619(2)	-1066(1)	29(1)
O(15)	9962(1)	1142(2)	6618(1)	30(1)
O(3)	6345(1)	3139(2)	5247(1)	25(1)
O(1)	4370(1)	4420(2)	2308(1)	24(1)
O(11)	7614(1)	-811(2)	3127(1)	21(1)
O(13)	8380(1)	-738(2)	1727(1)	24(1)
O(10)	9188(1)	2664(2)	4949(1)	22(1)
O(8)	2077(1)	1003(2)	3292(1)	35(1)
C(7)	3524(2)	793(3)	57(1)	21(1)
C(6)	4170(2)	-898(3)	621(1)	22(1)
C(5)	4927(2)	-789(3)	1757(1)	20(1)
C(4A)	5081(2)	989(3)	2373(1)	18(1)
C(4)	5931(2)	1087(3)	3622(1)	19(1)
C(11)	7388(2)	538(3)	3923(1)	20(1)
C(14)	8077(2)	-243(3)	5119(1)	23(1)
C(15)	9175(2)	1188(3)	5678(1)	22(1)
C(10)	8112(2)	2497(3)	3868(1)	20(1)
C(9)	7135(2)	4153(3)	3737(1)	22(1)
C(3)	6032(2)	3240(3)	4052(1)	20(1)
C(2)	4742(2)	4315(3)	3512(1)	23(1)
C(8A)	4376(2)	2624(3)	1796(1)	20(1)
C(8)	3621(2)	2554(3)	647(1)	22(1)
C(12)	8630(2)	2166(3)	2924(1)	23(1)
C(13)	8218(2)	101(3)	2509(1)	20(1)
C(17)	817(2)	2130(4)	1451(2)	42(1)
C(18)	1561(2)	602(3)	2295(2)	29(1)
C(19)	1629(2)	-1443(4)	1862(2)	37(1)

Table 3. Bond lengths [Å] and angles [deg] for zhb_w_1.

O(7)-C(7)	1.3627(19)
O(7)-H(7)	0.8400
O(15)-C(15)	1.192(2)
O(3)-C(3)	1.4284(19)
O(3)-H(3)	0.8400
O(1)-C(8A)	1.375(2)
O(1)-C(2)	1.4323(18)
O(11)-C(13)	1.347(2)
O(11)-C(11)	1.446(2)
O(13)-C(13)	1.209(2)
O(10)-C(15)	1.362(2)
O(10)-C(10)	1.4535(19)
O(8)-C(18)	1.214(2)
C(7)-C(8)	1.386(3)
C(7)-C(6)	1.395(3)
C(6)-C(5)	1.379(2)
C(6)-H(6)	0.9500
C(5)-C(4A)	1.407(3)
C(5)-H(5)	0.9500
C(4A)-C(8A)	1.392(2)
C(4A)-C(4)	1.521(2)
C(4)-C(3)	1.539(3)
C(4)-C(11)	1.550(2)
C(4)-H(4)	1.0000
C(11)-C(14)	1.522(2)
C(11)-C(10)	1.556(2)
C(14)-C(15)	1.505(2)
C(14)-H(14A)	0.9900
C(14)-H(14B)	0.9900
C(10)-C(9)	1.516(3)
C(10)-C(12)	1.519(2)
C(9)-C(3)	1.536(2)
C(9)-H(9A)	0.9900
C(9)-H(9B)	0.9900
C(3)-C(2)	1.517(2)
C(2)-H(2A)	0.9900
C(2)-H(2B)	0.9900
C(8A)-C(8)	1.391(2)
C(8)-H(8)	0.9500
C(12)-C(13)	1.498(3)
C(12)-H(12A)	0.9900
C(12)-H(12B)	0.9900
C(17)-C(18)	1.496(3)
C(17)-H(17A)	0.9800
C(17)-H(17B)	0.9800
C(17)-H(17C)	0.9800
C(18)-C(19)	1.495(3)
C(19)-H(19A)	0.9800
C(19)-H(19B)	0.9800
C(19)-H(19C)	0.9800
C(7)-O(7)-H(7)	109.5
C(3)-O(3)-H(3)	109.5
C(8A)-O(1)-C(2)	114.32(13)
C(13)-O(11)-C(11)	111.49(14)

C(15)-O(10)-C(10)	112.16(12)
O(7)-C(7)-C(8)	122.60(16)
O(7)-C(7)-C(6)	117.55(15)
C(8)-C(7)-C(6)	119.85(14)
C(5)-C(6)-C(7)	119.56(16)
C(5)-C(6)-H(6)	120.2
C(7)-C(6)-H(6)	120.2
C(6)-C(5)-C(4A)	122.05(16)
C(6)-C(5)-H(5)	119.0
C(4A)-C(5)-H(5)	119.0
C(8A)-C(4A)-C(5)	116.77(14)
C(8A)-C(4A)-C(4)	121.83(15)
C(5)-C(4A)-C(4)	121.34(14)
C(4A)-C(4)-C(3)	110.38(13)
C(4A)-C(4)-C(11)	115.73(13)
C(3)-C(4)-C(11)	101.70(13)
C(4A)-C(4)-H(4)	109.6
C(3)-C(4)-H(4)	109.6
C(11)-C(4)-H(4)	109.6
O(11)-C(11)-C(14)	109.25(14)
O(11)-C(11)-C(4)	113.76(12)
C(14)-C(11)-C(4)	115.01(13)
O(11)-C(11)-C(10)	106.36(12)
C(14)-C(11)-C(10)	105.11(13)
C(4)-C(11)-C(10)	106.59(14)
C(15)-C(14)-C(11)	105.73(14)
C(15)-C(14)-H(14A)	110.6
C(11)-C(14)-H(14A)	110.6
C(15)-C(14)-H(14B)	110.6
C(11)-C(14)-H(14B)	110.6
H(14A)-C(14)-H(14B)	108.7
O(15)-C(15)-O(10)	120.94(15)
O(15)-C(15)-C(14)	128.38(16)
O(10)-C(15)-C(14)	110.68(13)
O(10)-C(10)-C(9)	110.37(13)
O(10)-C(10)-C(12)	110.18(13)
C(9)-C(10)-C(12)	118.41(14)
O(10)-C(10)-C(11)	106.13(13)
C(9)-C(10)-C(11)	106.04(13)
C(12)-C(10)-C(11)	104.79(14)
C(10)-C(9)-C(3)	105.72(15)
C(10)-C(9)-H(9A)	110.6
C(3)-C(9)-H(9A)	110.6
C(10)-C(9)-H(9B)	110.6
C(3)-C(9)-H(9B)	110.6
H(9A)-C(9)-H(9B)	108.7
O(3)-C(3)-C(2)	109.07(13)
O(3)-C(3)-C(9)	112.78(13)
C(2)-C(3)-C(9)	112.81(15)
O(3)-C(3)-C(4)	106.85(13)
C(2)-C(3)-C(4)	110.76(14)
C(9)-C(3)-C(4)	104.33(14)
O(1)-C(2)-C(3)	111.28(13)
O(1)-C(2)-H(2A)	109.4
C(3)-C(2)-H(2A)	109.4
O(1)-C(2)-H(2B)	109.4
C(3)-C(2)-H(2B)	109.4
H(2A)-C(2)-H(2B)	108.0
O(1)-C(8A)-C(8)	114.91(15)

O(1)-C(8A)-C(4A)	123.07(14)
C(8)-C(8A)-C(4A)	122.02(16)
C(7)-C(8)-C(8A)	119.63(16)
C(7)-C(8)-H(8)	120.2
C(8A)-C(8)-H(8)	120.2
C(13)-C(12)-C(10)	105.15(14)
C(13)-C(12)-H(12A)	110.7
C(10)-C(12)-H(12A)	110.7
C(13)-C(12)-H(12B)	110.7
C(10)-C(12)-H(12B)	110.7
H(12A)-C(12)-H(12B)	108.8
O(13)-C(13)-O(11)	120.96(17)
O(13)-C(13)-C(12)	127.35(16)
O(11)-C(13)-C(12)	111.68(13)
C(18)-C(17)-H(17A)	109.5
C(18)-C(17)-H(17B)	109.5
H(17A)-C(17)-H(17B)	109.5
C(18)-C(17)-H(17C)	109.5
H(17A)-C(17)-H(17C)	109.5
H(17B)-C(17)-H(17C)	109.5
O(8)-C(18)-C(19)	121.38(18)
O(8)-C(18)-C(17)	121.3(2)
C(19)-C(18)-C(17)	117.37(17)
C(18)-C(19)-H(19A)	109.5
C(18)-C(19)-H(19B)	109.5
H(19A)-C(19)-H(19B)	109.5
C(18)-C(19)-H(19C)	109.5
H(19A)-C(19)-H(19C)	109.5
H(19B)-C(19)-H(19C)	109.5

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{Å}^2 \times 10^3$) for zhb_w_1.

The anisotropic displacement factor exponent takes the form:

$$-2 \pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^{*} b^{*} U_{12}]$$

	U11	U22	U33	U23	U13	U12
O(7)	36(1)	25(1)	19(1)	-2(1)	4(1)	3(1)
O(15)	33(1)	28(1)	21(1)	0(1)	1(1)	2(1)
O(3)	32(1)	24(1)	18(1)	-2(1)	9(1)	-5(1)
O(1)	28(1)	20(1)	21(1)	-2(1)	7(1)	5(1)
O(11)	21(1)	18(1)	25(1)	0(1)	10(1)	-1(1)
O(13)	25(1)	28(1)	20(1)	-3(1)	7(1)	0(1)
O(10)	22(1)	23(1)	20(1)	0(1)	6(1)	-4(1)
O(8)	42(1)	35(1)	26(1)	0(1)	10(1)	5(1)
C(7)	21(1)	24(1)	21(1)	-1(1)	9(1)	-1(1)
C(6)	24(1)	18(1)	26(1)	-4(1)	10(1)	-2(1)
C(5)	19(1)	18(1)	26(1)	2(1)	9(1)	0(1)
C(4A)	17(1)	20(1)	20(1)	1(1)	9(1)	-2(1)
C(4)	22(1)	17(1)	20(1)	2(1)	10(1)	-2(1)
C(11)	21(1)	18(1)	20(1)	0(1)	8(1)	0(1)
C(14)	20(1)	25(1)	23(1)	5(1)	7(1)	0(1)
C(15)	24(1)	22(1)	22(1)	0(1)	10(1)	2(1)
C(10)	20(1)	22(1)	17(1)	1(1)	4(1)	-4(1)
C(9)	26(1)	19(1)	22(1)	2(1)	8(1)	-2(1)
C(3)	23(1)	21(1)	17(1)	0(1)	7(1)	-1(1)
C(2)	26(1)	23(1)	22(1)	-3(1)	10(1)	1(1)
C(8A)	21(1)	18(1)	23(1)	-2(1)	12(1)	-1(1)
C(8)	22(1)	21(1)	23(1)	2(1)	9(1)	2(1)
C(12)	24(1)	24(1)	22(1)	0(1)	9(1)	-3(1)
C(13)	16(1)	23(1)	18(1)	3(1)	3(1)	2(1)
C(17)	34(1)	61(2)	33(1)	16(1)	14(1)	1(1)
C(18)	26(1)	38(1)	28(1)	2(1)	14(1)	-1(1)
C(19)	35(1)	47(1)	32(1)	-9(1)	15(1)	-5(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{Å}^2 \times 10^3$) for zhb_w_1.

	x	y	z	U(eq)
H(7)	2399	1691	-1304	43
H(3)	6719	4193	5549	38
H(6)	4088	-2118	225	27
H(5)	5358	-1951	2135	25
H(4)	5545	232	4065	23
H(14A)	7467	-289	5532	27
H(14B)	8421	-1595	5102	27
H(9A)	6792	4646	2948	27
H(9B)	7547	5273	4247	27
H(2A)	4056	3610	3698	28
H(2B)	4818	5674	3825	28
H(8)	3175	3706	269	27
H(12A)	8251	3143	2308	28
H(12B)	9593	2289	3212	28
H(17A)	793	3373	1845	63
H(17B)	-77	1661	1050	63
H(17C)	1247	2359	908	63
H(19A)	776	-2088	1661	56
H(19B)	2287	-2218	2451	56
H(19C)	1870	-1363	1192	56

Table 6. Torsion angles [deg] for zhb_w_1.

O(7)-C(7)-C(6)-C(5)	179.58(14)
C(8)-C(7)-C(6)-C(5)	-1.6(2)
C(7)-C(6)-C(5)-C(4A)	-0.4(2)
C(6)-C(5)-C(4A)-C(8A)	3.2(2)
C(6)-C(5)-C(4A)-C(4)	-179.41(14)
C(8A)-C(4A)-C(4)-C(3)	-8.3(2)
C(5)-C(4A)-C(4)-C(3)	174.53(14)
C(8A)-C(4A)-C(4)-C(11)	-123.04(16)
C(5)-C(4A)-C(4)-C(11)	59.7(2)
C(13)-O(11)-C(11)-C(14)	-120.47(14)
C(13)-O(11)-C(11)-C(4)	109.51(15)
C(13)-O(11)-C(11)-C(10)	-7.50(16)
C(4A)-C(4)-C(11)-O(11)	-27.1(2)
C(3)-C(4)-C(11)-O(11)	-146.74(13)
C(4A)-C(4)-C(11)-C(14)	-154.18(15)
C(3)-C(4)-C(11)-C(14)	86.19(17)
C(4A)-C(4)-C(11)-C(10)	89.77(16)
C(3)-C(4)-C(11)-C(10)	-29.86(14)
O(11)-C(11)-C(14)-C(15)	111.29(15)
C(4)-C(11)-C(14)-C(15)	-119.38(15)
C(10)-C(11)-C(14)-C(15)	-2.49(17)
C(10)-O(10)-C(15)-O(15)	-177.25(15)
C(10)-O(10)-C(15)-C(14)	2.82(18)
C(11)-C(14)-C(15)-O(15)	-179.95(17)
C(11)-C(14)-C(15)-O(10)	-0.02(18)
C(15)-O(10)-C(10)-C(9)	110.13(15)
C(15)-O(10)-C(10)-C(12)	-117.25(15)
C(15)-O(10)-C(10)-C(11)	-4.33(17)
O(11)-C(11)-C(10)-O(10)	-111.75(13)
C(14)-C(11)-C(10)-O(10)	4.05(17)
C(4)-C(11)-C(10)-O(10)	126.55(13)
O(11)-C(11)-C(10)-C(9)	130.85(13)
C(14)-C(11)-C(10)-C(9)	-113.35(14)
C(4)-C(11)-C(10)-C(9)	9.15(15)
O(11)-C(11)-C(10)-C(12)	4.85(16)
C(14)-C(11)-C(10)-C(12)	120.64(14)
C(4)-C(11)-C(10)-C(12)	-116.85(14)
O(10)-C(10)-C(9)-C(3)	-98.92(14)
C(12)-C(10)-C(9)-C(3)	132.82(15)
C(11)-C(10)-C(9)-C(3)	15.60(15)
C(10)-C(9)-C(3)-O(3)	80.68(17)
C(10)-C(9)-C(3)-C(2)	-155.19(13)
C(10)-C(9)-C(3)-C(4)	-34.90(15)
C(4A)-C(4)-C(3)-O(3)	156.44(12)
C(11)-C(4)-C(3)-O(3)	-80.21(14)
C(4A)-C(4)-C(3)-C(2)	37.76(17)
C(11)-C(4)-C(3)-C(2)	161.11(13)
C(4A)-C(4)-C(3)-C(9)	-83.89(15)
C(11)-C(4)-C(3)-C(9)	39.46(14)
C(8A)-O(1)-C(2)-C(3)	52.17(18)
O(3)-C(3)-C(2)-O(1)	-178.82(13)
C(9)-C(3)-C(2)-O(1)	55.0(2)
C(4)-C(3)-C(2)-O(1)	-61.50(18)
C(2)-O(1)-C(8A)-C(8)	159.11(13)
C(2)-O(1)-C(8A)-C(4A)	-21.0(2)

C(5)-C(4A)-C(8A)-O(1)	175.92(13)
C(4)-C(4A)-C(8A)-O(1)	-1.4(2)
C(5)-C(4A)-C(8A)-C(8)	-4.1(2)
C(4)-C(4A)-C(8A)-C(8)	178.53(14)
O(7)-C(7)-C(8)-C(8A)	179.50(14)
C(6)-C(7)-C(8)-C(8A)	0.8(2)
O(1)-C(8A)-C(8)-C(7)	-177.83(14)
C(4A)-C(8A)-C(8)-C(7)	2.2(2)
O(10)-C(10)-C(12)-C(13)	112.83(15)
C(9)-C(10)-C(12)-C(13)	-118.83(16)
C(11)-C(10)-C(12)-C(13)	-0.95(16)
C(11)-O(11)-C(13)-O(13)	-173.07(14)
C(11)-O(11)-C(13)-C(12)	7.19(17)
C(10)-C(12)-C(13)-O(13)	176.61(15)
C(10)-C(12)-C(13)-O(11)	-3.67(17)

Symmetry transformations used to generate equivalent atoms:

Table 7. Hydrogen bonds for zhb_w_1 [Å and deg.].

D-H...A	d(D-H)	d(H...A)	d(D...A)	\angle (DHA)
O(3)-H(3)...O(8)#1	0.84	2.00	2.7983(18)	159.4
O(7)-H(7)...O(13)#2	0.84	1.92	2.7580(19)	174.6

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,y+1/2,-z+1 #2 -x+1,y+1/2,-z