Electronic Supplementary Information

Enhancing Excess Electron Transport in DNA

Fazel Fakhari,^{1,2} Yun-Yun K. Chen¹ and Steven E. Rokita^{1,2*}

Contribution from the ¹Department of Chemistry and Biochemistry, University of Maryland, College Park, Maryland 20742 USA and ²Department of Chemistry, Johns Hopkins University, Baltimore, MD 21218 USA (current address)

General methods. NMR spectra were recorded on a Bruker AM400, AM500, or AM600 spectrometer and referenced to residual protons in the deuterated solvents. Chemical shifts (δ) and coupling constants (*J*) are reported in parts per million (ppm) and Hertz (Hz), respectively. Mass spectrometry analysis was performed on a JEOL AccuToF-CS ESI-MS in the ESI⁺ ionization mode and a Shimadzu Axima-CFR MALDI-TOF. HPLC purification of oligonucleotide conjugates employed a Jasco PU-980 and a reverse phase, Varian C18 Microsorb column. Photoirradiation was performed using a high-pressure Xe-arc (1000 W, Spectral Energy Co.) and a cutoff glass filter (335 nm, WG335, Schott).

General materials. All chemicals, reagents, and solvents of the highest commercial grade were used without further purification unless otherwise noted. All aqueous solutions were prepared with water purified to a resistivity of 17.8-18.0 M Ω .cm. Oligodeoxynucleotides were purchased from TriLink Biotechnologies (^{Br}dU-containing DNA) and Integrated DNA Technologies. All ^{Br}dU-containing DNA was purified by denaturing gel electrophoresis before use. DNA concentration was calculated from its absorption at 260 nm and its ϵ_{260} value provided by the vender.

Synthesis of *N*-(4-bromobutyloxy)phthalimide (1). Sodium carbonate (1.27 g, 12.0 mmol) and 1,4-dibromobutane (2.57 g, 12.0 mmol) were mixed with acetone (6 ml) in a round bottom flask and heated to reflux under a N₂ atmosphere. Hydroxyphthalimide (489 mg, 3.00 mmol) in acetone (30 ml) was then added dropwise over an hour. The reaction was refluxed for 20 h and then cooled and filtered. The filtrate was concentrated under reduced pressure. The desired product was isolated after silica gel flash chromatography (hexanes:ethyl acetate, 1:0 to 1:1) as a white powder in 95% yield (850 mg, 2.85 mmol). ¹H NMR (400 MHz, CDCl₃) δ 7.82 (m, 2H) 7.76 (m, 2H), 4.24 (t, *J* = 12.2 Hz, 2H), 3.54 (t, *J* = 13.1 Hz, 2H), 2.15 (m, 2H), 1.94 (m, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 165.1, 136.1, 130.1, 124.5, 78.6, 35.1, 30.4, 28.1. HRMS *m*/*z* calcd for C₁₂H₁₂BrNO₂ 298.0065 found 298.0045 (M + H⁺); mp 68-70 °C.

Synthesis of 2a by alkylation of 1,5-diaminonaphthalene with *N*-(4-bromobutyloxy)phthalimide. The bromide 1 (298 mg, 1.00 mmol) in acetonitrile (10 ml) was added dropwise to a round bottom flask containing a mixture of 1,5-diaminonaphthalene (632 mg, 4.00 mmol), NaHCO₃ (420 mg, 5.00 mmol) and KI (83 mg, 0.50 mmol) in acetonitrile (10 ml). This mixture was refluxed under a N₂ atmosphere for 24 h. After cooling the reaction, the solvent was evaporated. The remaining dark red residue was suspended in water (10 ml), adjusted to pH 5 with 1 N HCl and extracted with dichloromethane (3 × 10 ml). The organic phases were combined, dried over MgSO4, filtered, and concentrated under reduced pressure. The desired product **2a** was isolated after silica gel flash chromatography (hexanes:ethyl acetate, 4:1 to 2:3) as an orange solid in 65% yield (244 mg). ¹H NMR (500 MHz, CD₃CN) δ 7.60 (dd, *J* = 9.5 Hz, 2H), 6.76 (d, *J* = 7.3 Hz, 2H), 7.31 (t, *J* = 8.0 Hz, 2H), 7.25 (t, *J* = 7.9 Hz, 2H) 6.97 (d, *J* = 9.5 Hz, 2H), 6.76 (d, *J* = 7.3 Hz, 2H), 4.50 (s, 3H), 3.24 (m, 4H), 1.94 (m, 4H). ¹³C NMR (500 MHz, CD₃CN) δ 165.1, 149.3, 145.0, 130.5, 126.3, 126.1, 125.9, 118.7, 115.9, 115.7, 113.1, 110.0, 78.6, 53.7, 31.4, 25.7. HRMS *m/z* calcd for C₂₂H₂₁N₃O₃ calcd 376.1616 found 376.1612 (M + H⁺); mp 122-124 °C. Synthesis of 3a by deprotection of *N*-(4-oxyphthalimidebutyl)-1,5-diaminonaphthalene. Hydrazine monohydrate (0.70 g, 14 mmol) was added to a solution of 2a (19 mg, 0.051 mmol) in EtOH:CH₂Cl₂ (1:1, 3 ml), and the mixture was stirred at room temperature for 25 min. Dichloromethane (4 ml) and sulfuric acid (2%, 8 ml) were added to the mixture, and it was filtered through celite 545[®] following a published protocol.¹ The filter cake was then washed successively with dichloromethane (2 ml) and 2% sulfuric acid (8 ml). The aqueous phase was washed with dichloromethane (2 × 5 ml). The combined organic phase was dried over MgSO4, filtered and evaporated under reduced pressure to yield an orange solid **3a** in a quantitative yield (12 mg). ¹H NMR (500 MHz, CD₂Cl₂) δ 7.31 (m, 1H), 7.24 – 7.18 (m, 2H), 7.18 – 7.13 (dd, 1H), 6.76 – 6.72 (m, 1H), 6.61-6.58 (d, *J* = 7.6 Hz, 1H), 4.61 – 3.86 (s, 3H), 3.72 (m, 2H), 3.28 (m, 2H), 1.73-1.82 (m, 4H). ¹³C NMR (126 MHz, CD₂Cl₂) δ 144.7, 143.6, 126.1, 125.6, 124.8, 124.7, 110.8, 110.1, 110.0, 104.7, 76.2, 44.7, 26.9, 26.6. HRMS *m/z* calcd for C₁₄H₁₉N₃O calcd 246.1558 found 246.1560 (M + H⁺).

Synthesis of 2b by alkylation of 1-aminoanthracene with *N*-(4-bromobutyloxy)phthalimide. The bromide 1 (298 mg, 1.00 mmol) in acetonitrile (10 ml) was added dropwise to a round bottom flask containing a mixture of 1-aminoanthracene (194 mg, 1.05 mmol), NaHCO₃ (169 mg, 2.01 mmol) and KI (83 mg, 0.50 mmol) in acetonitrile (5 ml). This mixture was refluxed under a N₂ atmosphere for 24 h. After the reaction was cooled, the solvent was evaporated. The residue was suspended in water (10 ml), adjusted to pH 5 by addition of 1 N HCl and was extracted with dichloromethane (3×10 ml). The organic phases were combined, dried over MgSO₄, filtered, and concentrated under reduced pressure. The desired product **2b** was isolated from silica gel flash chromatography (hexanes:ethyl acetate, 1:0 to 3:2) as a brown solid in 65% yield (267 mg, 0.65 mmol). ¹H NMR (400 MHz, CD₂Cl₂) δ 8.5 (s, 1H), 8.3 (s, 1H), 7.9-8.1 (m, 2H), 7.7-7.9 (m, 4H), 7.2-7.5 (m, 4H), 6.5-6.6 (m, 1H), 4.7 (s, 1H), 4.3 (m, 2H), 3.4 (m, 2H), 1.9-2.2 (m, 4H). ¹³C NMR (400 MHz, CDCl₃) 163.8, 141.7, 134.5, 132.6, 131.6, 128.9, 128.5, 127.9, 126.6, 125.8, 125.4, 124.9, 123.5, 119.3, 118.8, 117.7, 107.6, 102.4, 78.2, 44.1, 26.3, 25.6. HRMS *m/z* calcd for C₂₆H₂₂N₂O₃ 411.1663 found 411.1704 (M + H⁺).

Synthesis of 3b by deprotection of *N*-(4-oxyphthalimidebutyl)-1-aminoanthracene. Hydrazine monohydrate (0.70 g, 14.0 mmol) was added to a solution of 2b (25 mg, 0.063 mmol) in EtOH:CH₂Cl₂ (1:1, 3 ml), and the mixture was stirred at room temperature for 40 min. Dichloromethane (4 ml) and sulfuric acid (2%, 8 ml) were added to the mixture, and it was filtered through celite 545[®] following a published protocol.¹ The filter cake was then washed successively with dichloromethane (2 ml) and 2% sulfuric acid (8 ml). The aqueous phase was washed with dichloromethane (2 × 5 ml). The combined organic phase was dried over MgSO₄, filtered and evaporated under reduced pressure to yield an orange solid 3b in a quantitative yield (17 mg). ¹H NMR (600 MHz, CD₃OD) δ 7.7-8.0 (m, 4H), 7.4 (m, 2H), 7.2 (m, 2H), 6.5 (m, 1H), 4.6 (s, 1H), 3.7 (m, 2H), 3.3 (m, 2H), 1.7-2.0 (m, 4H). ¹³C NMR (500 MHz, CD₂Cl₂) δ 143.0, 133.4, 132.5, 131.8, 129.2, 128.6, 127.2, 126.4, 126.0, 124.5, 124.2, 120.4, 119.6, 108.0, 73.6, 44.9, 28.0, 26.7. HRMS *m*/z calcd for C₁₈H₂₀N₂O 281.1607 found 281.1590.

Conjugation of electron donors, 3a and 3b, to DNA. Oligonucleotides containing internal abasic sites were generated from uridine containing analogues **OD7'** and **OD8'** (see **Chart S1**) using uracil-DNA deglycosylase as described previously.² Oligonucleotides containing terminal abasic sites (**OD5'** and **OD6'**) were purchased from Trilink Biotechnologies. Solutions of **3a** and **3b** (50 μ L, 50 mM in HPLC grade acetonitrile) were alternatively combined with oligonucleotides containing an abasic DNA (20 μ L, 1 mM in water) and incubated at 37 °C for 1 h. Excess donor was removed with a BioSpin 6 column (prewashed with water, 3 times), and the conjugates were isolated from reverse phase (C-18) HPLC using a gradient of 10% acetonitrile in 45 mM triethylammonium acetate (pH 6.0) to 30% acetonitrile in 35 mM trietyl-ammonium acetate (pH 6.0) over 20 min (1 ml/min) in yields about 30% (based on A₂₆₀ of starting material) (See for example, **Figure S1**). MALDI (with a matrix of 3,5-dimethoxy-4-hydroxycinnamic acid) *m/z* calcd for **OD1'a** and **OD3'a** C₁₈₅H₂₃₆N₆₅O₁₀₇P₁₇ 5608.8, found 5608.3 and 5609.2, calcd for **OD1'b** and **OD3'b** C₁₈₉H₂₃₇N₆₄O₁₀₇P₁₇ 5643.4 found 5643.2 and 5644.1, calcd for **OD2'a** and **OD4'a** C₁₈₅H₂₃₆N₆₅O₁₀₆P₁₇

5592.8 found 5591.7 and 5592.4, calcd for **OD2'b** and **OD4'b** C₁₈₉H₂₃₇N₆₄O₁₀₆P₁₇ 5627.4 found 5627.9 and 5627.8, respectively.

Photochemical initiation of EET in DNA. DNA strands containing ^{Br}dU were radiolabeled at the 5'-terminus using $[\gamma^{-32}P]$ ATP (Amersham Bioscience) and T4 polynucleotide kinase (New England Biolabs) following standard procedures. The radiolabeled strand $(0.2 \ \mu\text{M})$ and the complementary conjugate strand (1.1 - 1.6 equivalents) was annealed in 10 mM sodium phosphate pH 7 and 100 mM NaCl by heating to 90 °C followed by slow cooling to room temperature. Aliquots of this solution $(10 \ \mu\text{L} \text{ each})$ were then irradiated in microcentrifuge tubes under aerobic conditions $(10 \ ^{\circ}\text{C})$. DNA was precipitated with sodium acetate (3 M, pH 5.5, 5 μ L) and ethanol (200 μ L), treated with 10% piperidine (15 μ L) at 90 °C for 30 min, and dried under reduced pressure. The resulting residue was resuspended in loading buffer (DMF, 0.1% bromphenol blue, 0.1% xylene cyanol FF) and analyzed by electrophoresis using a 20% denaturing polyacrylamide gel (acrylamidebisacrylamide 19:1, 7 M urea). Strand scission was quantified and reported relative to the total material in each lane as measured by phosphorimagery and its software ImageQuant (GE Healthcare Life Sciences).

OD1' 5'-CGT CAT GYA ATG TAC TGC

 OD2' 5'-CGT CAT AYA ATG TAC TGC

 OD3' 5'-AAT GTA CTG CCG TCA TGY

 OD4' 5'-AAT GTA CTG CCG TCA TAY

 OD5' 5'-AAT GTA CTG CCG TCA TGZ

 OD6' 5'-AAT GTA CTG CCG TCA TAZ

 OD6' 5'-AAT GTA CTG CCG TCA TAZ

 OD7' 5'-CGT CAT GUA ATG TAC TGC

 OD8' 5'-CGT CAT AUA ATG TAC TGC



Chart S1. Nucleotide sequences of oligonucleotides and their conjugates.

Figure S1. A sample separation of an oligonucleotide containing an abasic site (**OD5'**) and its DN conjugate (**OD3'a**) by reverse phase (C-18) HPLC using a gradient of 10% acetonitrile in 45 mM triethylamine acetate buffer (pH 6.0) to 30% acetonitrile in 35 mM trietylamonium acetate (pH 6.0) over 20 min (1 ml/min).

Figure S2. Absorption spectra of DN- and AA-containing oligonucleotide conjugates. A change in lamps within the spectrophotometer is evident at ca. 375 nm. The integrated area under the absorption traces above 330 nm for DN:AA was 1.0:0.9.



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_		$T_{\rm m}(^{\rm o}{\rm C})$	
	abasic	DN	AA
Duplex	parent	conjugate	conjugate
DNA1	61	65	67
DNA2	60	65	66
DNA3	59	64	65
DNA4	58	64	65

Figure S3. Sample melting of DNA duplexes
(△) parent DNA3 containing an abasic site,
(□) DNA3a containing a conjugated DN and
(■) DNA3b containing a conjugated AA.
DNA samples were pre-annealed in 50 mM sodium phosphate pH 7 and 50 mM NaCl.

Table S1. Melting temperatures for DNA duplexes containing abasic sites and their DN- and AA- conjugates under conditions described in Figure S3. A_{260} was monitored with a Varian Cary 100 UV-Vis spectrophotometer while heating samples (0.5 °C/min) T_m values were defined by the midpoint transition.



Figure S4. Initial rates of strand scission by photoinduced EET from donor (DN and AA) to acceptor (^{Br}dU) in duplex DNA. Red unfilled and blue filled squares correspond to data observed from the internal and terminal conjugates, respectively. Duplex DNA (0.2 μ M, 90 nCi, ^{Br}dU strand) in sodium phosphate (10 mM, pH 7.0) and NaCl (100 mM) were irradiated ($\lambda > 335$ nm) for the indicated times. Initial rates were calculated from the linear fit of strand scission (% vs. total) resulting from electron capture by ^{Br}dU (See **Figure S5**). Each experiment was repeated at least three times and error bars represent standard deviations.

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Figure S5. Phosphoimagery of 20% denaturing polyacrylamide gels of DNA after UV irradiation (> 335 nm, 10 °C) and subsequent treatment with 10% piperidine at 90 °C for 30 min. (A) DN- and AA-conjugates **DNA1a,b** (B) DN- and AA-conjugates **DNA2a,b**, (C) DN- and AA-conjugates **DNA3a,b** and (D) DN- and AA-conjugates **DNA4a,b**. In each study, the ^{Br}dU containing strand was radiolabeled at the 5'-terminus with γ -³²P-ATP. Lanes 1-6 correspond to DNA with conjugated DN and lanes 7-12 correspond to DNA with conjugated AA as the electron donor. Arrows indicate the scission site based on electron capture by ^{Br}dU.

References:

- 1) S. K. Jackson, A. Karadeolian, A. B. Driega and M. A. Kerr, Stereodivergent methodology for the synthesis of complex pyrrolidines, *J. Am. Chem. Soc.*, 2008, **130**, 4196-4201.
- 2) A. Fakhari M. and S. E. Rokita, A new solvatochromic fluorophore for exploring nonpolar environments created by biopolymers, *Chem. Commun.*, 2011, **47**, 4222-4224.