

Electronic Supplementary Information

Laminar MoS₂ membrane for molecule separation

Luwei Sun,^a Hubiao Huang^a and Xinsheng Peng^{a,b}

^a State Key Laboratory of Silicon Materials, Department of Materials Science and Engineering, Zhejiang University, Hangzhou 310027, China.

E-mail: pengxinsheng@zju.edu.cn

^b Cyrus Tang Centre for Sensor Materials and Applications, Zhejiang University, Hangzhou 310027, China

Experimental Section

The experiment was carried out in a N₂-filled glove box. 300 mg MoS₂ powder (Alfa Aesar) was dispersed in 3 ml of 1.6 M *n*-butyl lithium in hexane (Acros) under stirring for 48 hours. The obtained black product was filtered through 0.45 µm PTFE (polytetrafluoroethylene) filter membrane and washed repeatedly by copious hexane. The purified powder was dispersed in *ca.* 100 ml ultrapure water (Millipore) under the mild ultrasonication for ~60 min. Excess lithium hydroxyl was removed by centrifuging. MoS₂ membranes were prepared by vacuum filtrating diluted MoS₂ (20-50 mg/L) solution on a polycarbonate (PC) membrane with 200 nm pores (Millipore).

Water permeance and purification were tested in through a dead-end filter holder (Millipore P158, 25 mm) for pressure testing with effective area of 2.83 cm² at room temperature. 20 ml volume 15 µM Evans blue and Cyt. C (0.125 mg/ml) solutions were filtered through the MoS₂ membrane, respectively. The permeate was collected

by a plastic tube after certain time intervals. The rejection was calculated by the corresponding intensity of the characteristic absorbance peaks of feed and permeates solution.

Atom force microscope (AFM) image was obtained on a MultiMode (Veeco Inc.) test system under tapping mode. Zeta potential was measured on Zetasizer 3000 HAS (Malvern). Scanning electron microscopy (SEM) examination was performed by a Hitachi S-4800 at an acceleration voltage of 5 kV. The X-ray photoelectron spectroscopy (XPS) was carried out by Thermo ESCALAB 250 with an Al *K* α radiation source ($h\nu = 1486.6$ eV). UV-*vis* spectra were acquired on an Agilent 3600 UV-*vis*-IR spectroscopy. Raman spectra were recorded by a HR800 Jobin Yvon Raman spectrometer with a 514 nm excitation laser under ambient condition.

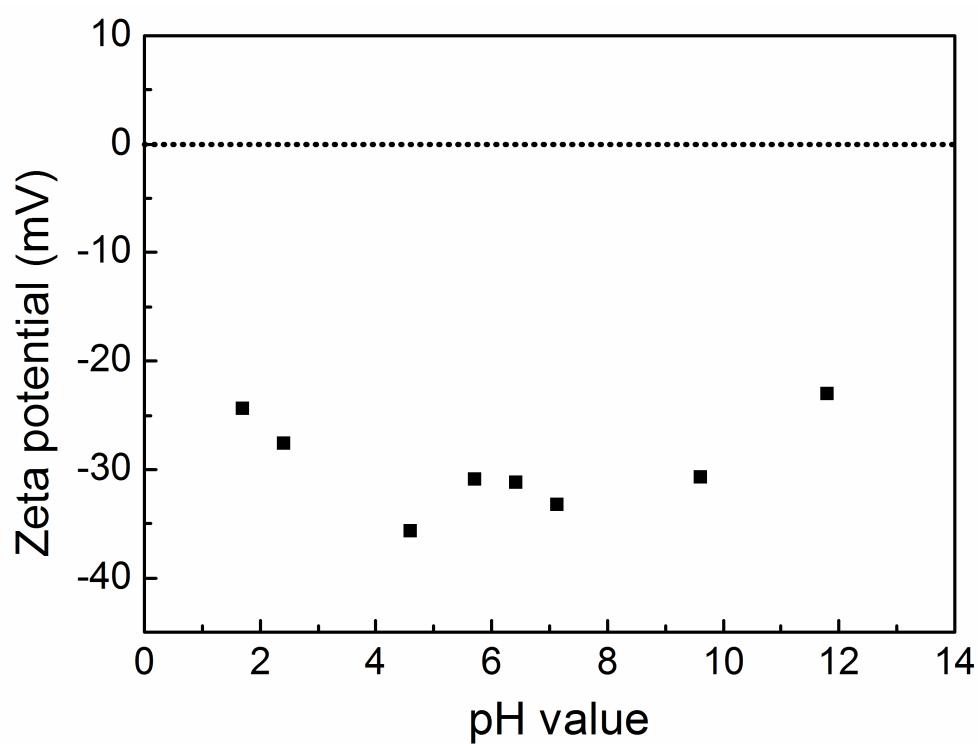


Figure S1. Zeta potential of the MoS₂ sheets dispersion solution at different pH value.

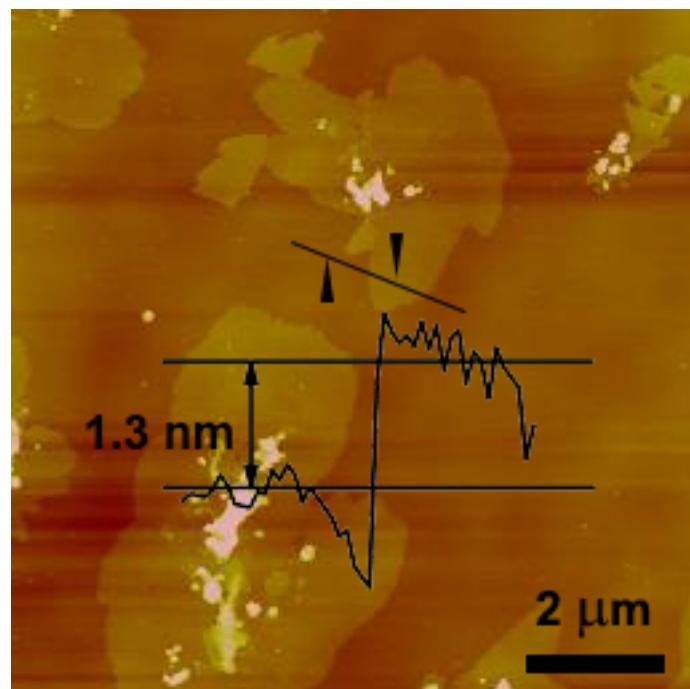


Figure S2. AFM image of MoS₂ sheet. Lateral dimension is scaled between two black arrows, as shown in inset. The MoS₂ sheet is 1.3 nm in thickness. The sharply drop in the height profile is due to silicon substrate.

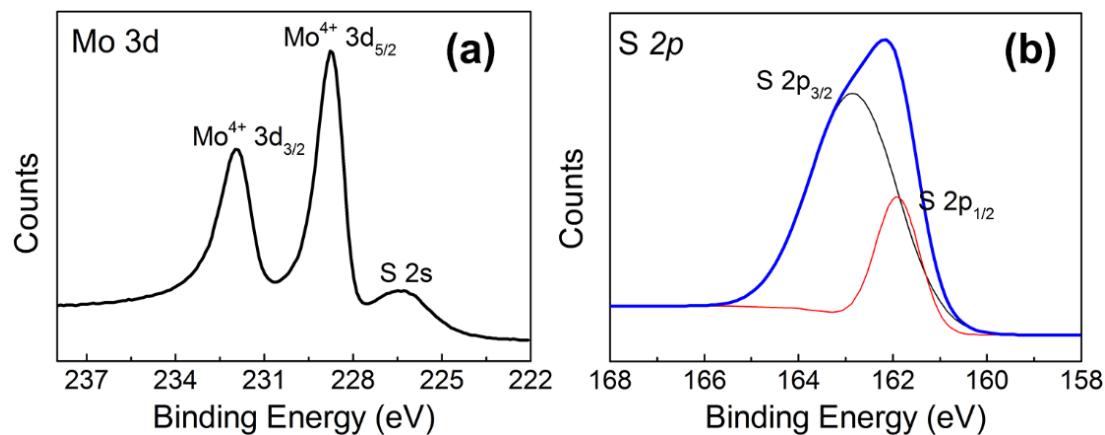


Figure S3. (a) and (b) are Mo 3d and S 2p XPS spectra, respectively. Mo⁴⁺ exhibits two peaks of 3d_{2/3} and 3d_{5/2}. A broad peak centered at 226 eV is S 2s band. S 2p band is deconvoluted two peaks corresponding to S 2p_{1/2} and 2p_{3/2}.

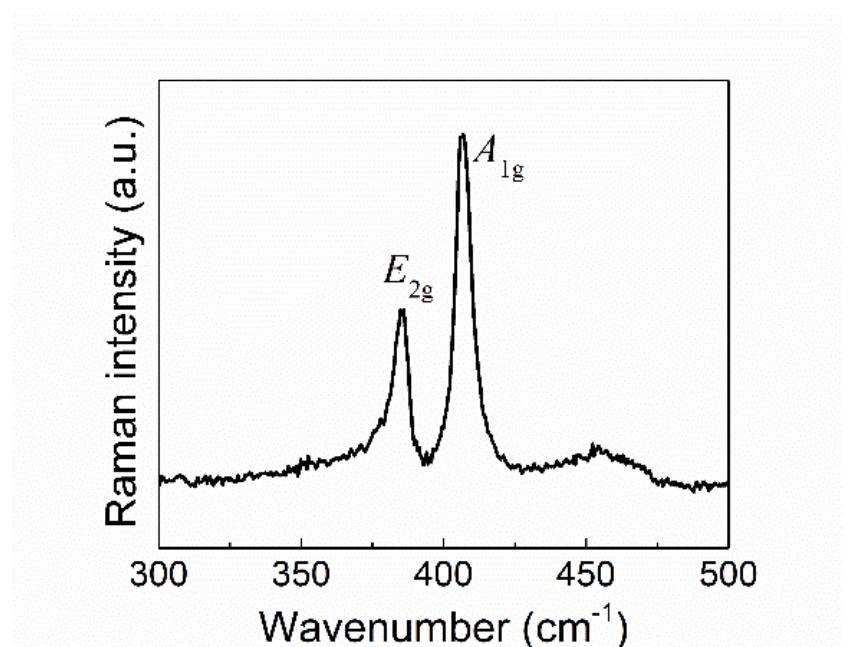


Figure S4. Raman spectrum of MoS₂ membrane

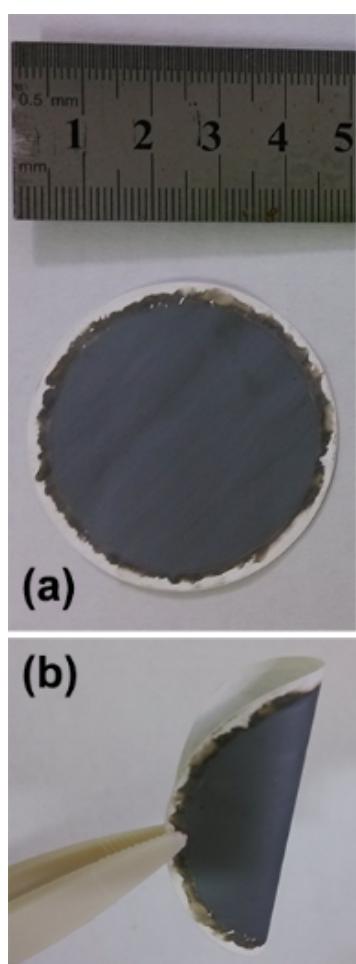


Figure S5 Photo of (a) a MoS₂ membrane with a diameter of 45 mm; and (b) its bending state.

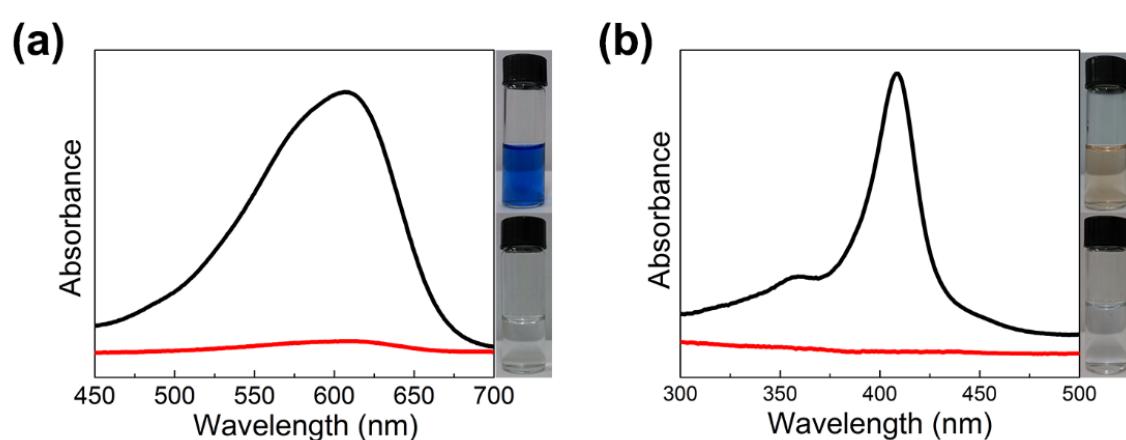


Figure S6. Photos and UV-vis absorbance spectra of feed (black line) and permeate (red line) solutions of (a) evans blue and (b) cytochrome C respectively through a $1.7 \mu\text{m} \pm 60 \text{ nm}$ thick MoS_2 membrane.

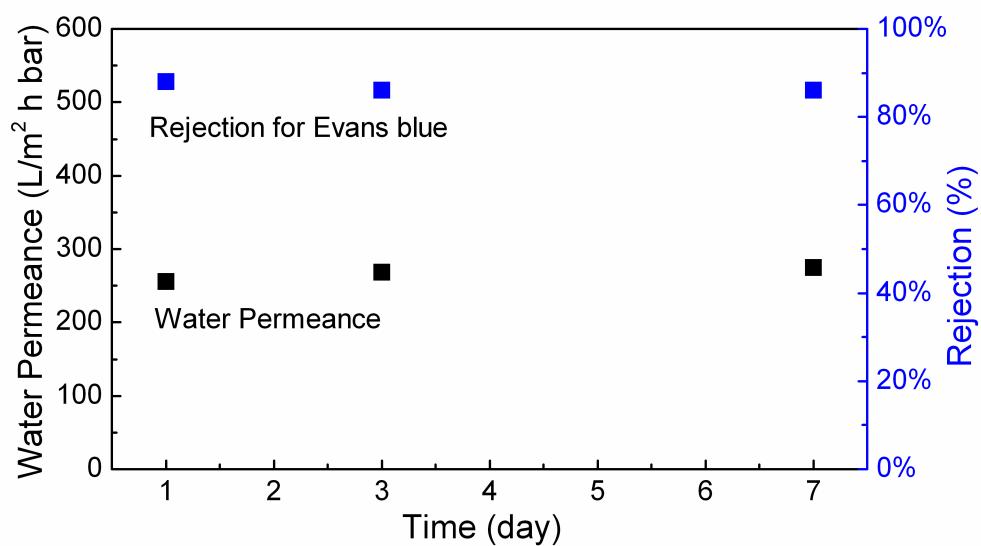


Figure S7. Water permeance (black squares) and rejection of EB (blue squares) are tested continuously through a $1.7\text{ }\mu\text{m}$ thick MoS_2 membrane after the 1, 3 and 7 days.