

# Supplementary Information

## Mechanosynthesis of pharmaceutically relevant sulfonyl-(thio)ureas

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### 1. Experimental Section

Mechanosynthesis of sulfonyl-(thio)ureas was carried out in a Retsch MM400 mill at a frequency of 30 Hz using a 10 mL stainless steel milling jar and a single ball made of the same material (10 mm diameter). Gram-scale synthesis reactions were carried out in a 25 mL stainless steel milling jar. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Varian MERCURY plus-300 or plus-400 spectrometer (300 or 400 MHz) with chemical shifts ( $\delta$ ) given in parts per million (ppm). The molecular weights of the pure products were determined using high-resolution mass spectra (HRMS). FT-IR spectra were collected using a Fourier Transform-Infrared Attenuated Total Reflection PerkinElmer UATR Two spectrometer in the range 400 cm<sup>-1</sup> to 4000 cm<sup>-1</sup>.

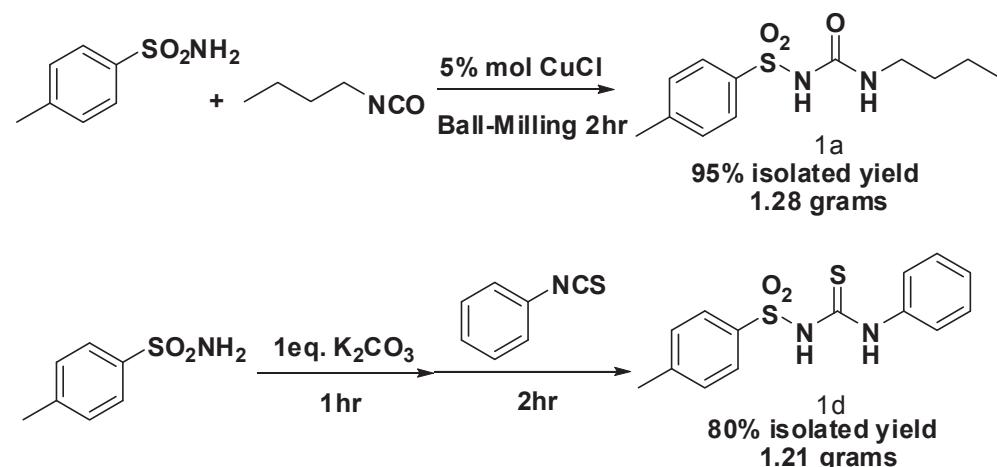
## 1.1 Mechanosynthesis by copper-catalysed coupling

A mixture of 0.50 mmol of sulfonamide, 0.50 mmol isocyanate (1 equiv), 0.025 mmol (5% mol) of CuCl and nitromethane as the grinding liquid ( $\eta = 0.25 \text{ mL mg}^{-1}$ ) was milled at a frequency of 30 Hz for 2 hrs. After the reaction, 3 mL of deionized water and 20 mg of Na<sub>2</sub>H<sub>2</sub>EDTA·2H<sub>2</sub>O were added to the green coloured crude mixture which was then milled for further 10 minutes at a frequency of 25 Hz. The resulting white product was purified by vacuum filtration and dried in air in all cases, except for glibenclamide **3b** where the formation of the sideproduct dicyclohexylurea requires chromatography.

## 1.2 Base-assisted mechanosynthesis using K<sub>2</sub>CO<sub>3</sub>

A mixture of 0.50 mmol of sulfonamide and 0.50 mmol of K<sub>2</sub>CO<sub>3</sub> (1 equiv) was milled at a frequency of 30 Hz for 1hr. Then, 0.50 mmol isocyanate or isothiocyanate (1 equiv) was added and subsequently milled for another 2 hrs at 30 Hz. 20 mL of deionized water and dilute HCl was added to the crude mixture. The pH of the resultant suspension was set to pH=3 using pH paper and left to stir for 15 mins. The product was isolated via vacuum filtration and dried in air in all cases, except for the sulfonyl-urea **1d** where the formation of the sideproduct diphenylurea requires chromatography.

## 1.3 Gram-scale synthesis

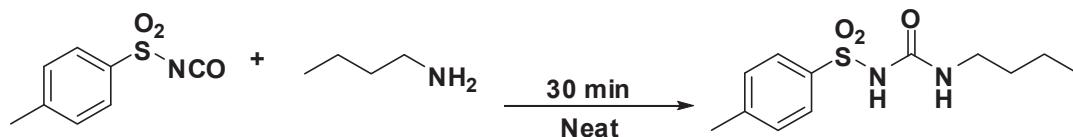


For gram-scale synthesis of tolbutamide (**1a**) 5.0 mmol of sulfonamide, 5.0 mmol isocyanate (1 equiv), 0.25 mmol (5% mol) of CuCl and nitromethane as the grinding liquid ( $\eta = 0.25 \text{ mL mg}^{-1}$ ) was milled at a frequency of 30 Hz for 2 hrs. The reaction was carried out using two 10 mm diameter stainless steel balls in a 25 mL stainless steel jar. After the reaction, 15 mL of deionised water and 200 mg of Na<sub>2</sub>H<sub>2</sub>EDTA·2H<sub>2</sub>O were added to the green coloured crude mixture which was then milled for 10 minutes more at a frequency of 25 Hz. The white product was purified via vacuum filtration and dried in air.

For the gram-scale synthesis of **1b**, a mixture of 5.0 mmol of sulfonamide and 5.0 mmol of K<sub>2</sub>CO<sub>3</sub> (1 equiv) was milled at a frequency of 30 Hz for 1 hour. Then, 5.0 mmol isothiocyanate (1 equiv) was added and subsequently milled for another 2 hrs at 30 Hz. 20

mL of deionized water and dilute HCl were added to the crude mixture. The pH of the resultant suspension was set to pH=3 using pH paper and left to stir for 15 mins. The product was purified via vacuum filtration and dried in air.

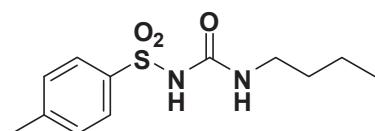
### 1.4 Pathway B for the synthesis of tolbutamide (**1a**)



The mechanosynthesis of tolbutamide **1a** (93% yield) was also explored by milling *p*-toluenesulfonyl-isocyanate and *n*-butyl-amine (retrosynthetic pathway B in the main paper). The high reactivity and moisture sensitive nature of *p*-toluenesulfonyl-isocyanate meant that reaction mixture preparation had to be done in a glovebox. Sulfonyl-isocyanates (mostly liquids) in general are less ideal, in terms of ease of handling and toxicity, as compared to sulfonamides (mostly solids).

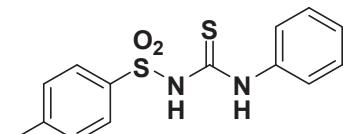
## 2. Summary of <sup>1</sup>H and <sup>13</sup>C NMR and HR-MS data

### *N*-(butylcarbamoyl)-4-methylbenzenesulfonamide **1a**



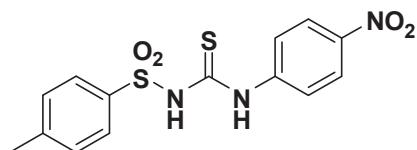
White powder (92% yield); <sup>1</sup>H-NMR (400 MHz, DMSO-*d*6) δ 0.79 (t, *J* = 7.20Hz, 3H), δ 1.11-1.18 (m, 2H), δ 1.23-1.30 (m, 2H), δ 2.37 (s, 3H), δ 2.88-2.93 (m, 2H), δ 6.41 (t, *J* = 5.80Hz, 1H), δ 7.38 (d, *J* = 8.00Hz, 2H), 7.75 (d, *J* = 7.6Hz, 2H), δ 10.45 (s, 1H); <sup>13</sup>C-NMR (300 MHz, DMSO-*d*6) δ 14.0, 19.7, 21.5, 31.7, 39.2, 127.6, 129.8, 137.9, 143.9, 151.7. HRMS: Calculated for C<sub>12</sub>H<sub>18</sub>N<sub>2</sub>NaO<sub>3</sub>S [M+Na]: 293.0930; measured: 293.0925.

### 4-methyl-*N*-(phenylcarbamothioyl)benzenesulfonamide **1b**



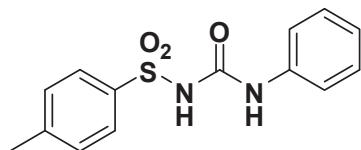
White powder (91% yield); <sup>1</sup>H-NMR (400 MHz, DMSO-*d*6) δ 2.38 (s, 3H), δ 7.16 (t, *J* = 7.26Hz, 1H), δ 7.39-7.43 (m, 4H), δ 7.28-7.36 (m, 2H), δ 7.81 (d, *J* = 8.37Hz, 2H), δ 10.14 (s, 1H); <sup>13</sup>C-NMR (300 MHz, DMSO-*d*6) δ 21.5, 119.4, 124.5, 126.2, 128.2, 129.0, 129.9, 138.5, 144.3, 178.0; HRMS: Calculated for C<sub>14</sub>H<sub>14</sub>N<sub>2</sub>NaO<sub>2</sub>S<sub>2</sub> [M+Na]: 329.0389; measured: 329.0390.

**4-methyl-N-(4-nitrophenylcarbamothioyl)benzenesulfonamide 1c**



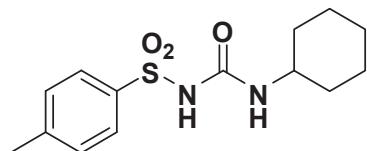
Yellow powder (80% yield); **<sup>1</sup>H-NMR** (400 MHz, DMSO-*d*6) δ 2.31 (s, 3H), δ 7.18 (d, *J* = 8.12Hz, 2H), δ 7.63 (d, *J* = 8.40Hz, 2H), 7.95-8.04 (m, 4H), δ 9.66 (s, 1H); **<sup>13</sup>C-NMR** (300 MHz, DMSO-*d*6) δ 21.1, 118.6, 124.9, 127.9, 128.3, 140.0, 142.1, 147.8, 183.0; **HRMS**: Calculated for C<sub>14</sub>H<sub>12</sub>N<sub>3</sub>O<sub>4</sub>S<sub>2</sub> [M+Na]: 350.0275; measured: 350.0272.

**4-methyl-N-(phenylcarbamoyl)benzenesulfonamide 1d**



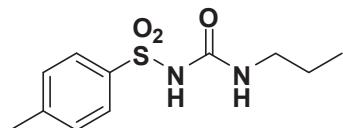
**<sup>1</sup>H-NMR** (400 MHz, DMSO-*d*6) δ 2.37, δ 6.99 (t, *J* = 4.00 Hz, 1H), δ 2.37 δ 2.37 δ 7.37-7.41 (m, 4H), δ 7.20-7.33 (m, 2H), δ 7.82 (d, *J* = 7.84Hz, 2H), δ 8.78 (s, 1H), δ 10.63 (s, 1H); **<sup>13</sup>C-NMR** (300 MHz, DMSO-*d*6) δ 21.5, 119.3, 123.5, 126.1, 127.9, 129.9, 137.7, 138.6, 144.1, 150.0 ppm. HRMS: Calculated for C<sub>14</sub>H<sub>14</sub>N<sub>2</sub>NaO<sub>3</sub>S [M+Na]: 313.0617; measured: 313.0614.

***N*-(cyclohexylcarbamoyl)-4-methylbenzenesulfonamide 1e**



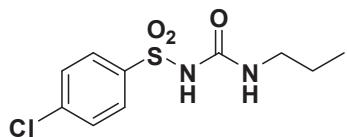
White powder (88% yield); **<sup>1</sup>H-NMR** (400 MHz, DMSO-*d*6) δ 1.02-1.25 (m, 5H), δ 1.22-1.66 (m, 5H), δ 2.37 (s, 3H), δ 3.25 (d, *J* = 7.28Hz, 1H), δ 6.30 (d, *J* = 8.40Hz, 1H) δ 7.38 (d, *J* = 8.00Hz, 2H), 7.75 (d, *J* = 7.60Hz, 2H), δ 10.26 (s, 1H); **<sup>13</sup>C-NMR** (300 MHz, DMSO-*d*6) δ 21.5, 24.6, 25.4, 32.7, 48.5, 127.6, 129.9, 137.9, 144.0, 150.9. **HRMS**: Calculated for C<sub>14</sub>H<sub>20</sub>N<sub>2</sub>NaO<sub>3</sub>S [M+Na]: 319.1087; measured: 319.1082.

**4-methyl-N-(propylcarbamoyl)benzenesulfonamide 1f**



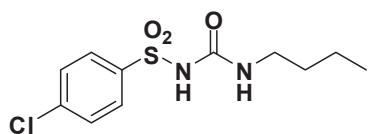
White powder (86% yield) **<sup>1</sup>H-NMR** (400 MHz, DMSO-*d*6) δ 0.73 (t, *J* = 7.00Hz, 3H), δ 1.28-1.32 (m, 2H), δ 2.37 (s, 3H), δ 2.84-2.91 (m, 2H), δ 6.42 (t, *J* = 5.80Hz, 1H), δ 7.38 (d, *J* = 8.00Hz, 2H), 7.75 (d, *J* = 7.6Hz, 2H), δ 10.46 (s, 1H); **<sup>13</sup>C-NMR** (300 MHz, DMSO-*d*6) δ 11.4, 21.4, 23.0, 41.2, 127.6, 129.7, 137.7, 143.7, 151.7; **HRMS**: Calculated for C<sub>11</sub>H<sub>16</sub>N<sub>2</sub>NaO<sub>3</sub>S [M+Na]: 279.0774; measured: 279.0764.

**4-chloro-N-(propylcarbamoyl)benzenesulfonamide 2a**



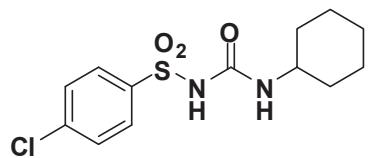
White powder (92% yield); **<sup>1</sup>H-NMR** (400 MHz, DMSO-*d*6) δ 0.73 (t, *J* = 7.48Hz, 3H), δ 1.26-1.35 (m, 2H) δ 2.85-2.89 (m, 3H), δ 6.51 (t, *J* = 5.80Hz, 1H), δ 7.67 (d, *J* = 8.36Hz, 2H), δ 7.88 (d, *J* = 7.96Hz, 2H), δ 10.66 (s, 1H); **<sup>13</sup>C-NMR** (300 MHz, DMSO-*d*6) δ 11.5, 22.9, 41.4, 129.6, 129.7, 138.5; **HRMS**: Calculated for C<sub>10</sub>H<sub>13</sub>ClN<sub>2</sub>NaO<sub>3</sub>S [M+Na]: 299.0228; measured: 299.0229.

***N*-(butylcarbamoyl)-4-chlorobenzenesulfonamide 2b**



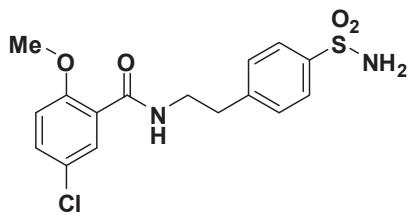
Off-white powder (92% yield); **<sup>1</sup>H-NMR** (400 MHz, DMSO-*d*6) δ 0.79 (t, *J* = 7.20Hz, 3H), δ 1.10-1.18 (m, 2H), δ 1.22-1.30 (m, 2H), δ 2.88-2.93 (m, 2H), δ 6.50 (t, *J* = 5.80Hz, 1H), δ 7.67 (d, *J* = 8.00Hz, 2H), 7.88 (d, *J* = 7.6Hz, 2H), δ 10.64 (s, 1H); **<sup>13</sup>C-NMR** (300 MHz, DMSO-*d*6) δ 13.9, 19.7, 31.1, 31.7, 129.6, 129.7, 138.4, 139.6, 151.7; **HRMS**: Calculated for C<sub>11</sub>H<sub>15</sub>ClN<sub>2</sub>NaO<sub>3</sub>S [M+Na]: 313.0384; measured: 313.0376.

**4-chloro-N-(cyclohexylcarbamoyl)benzenesulfonamide 2c**



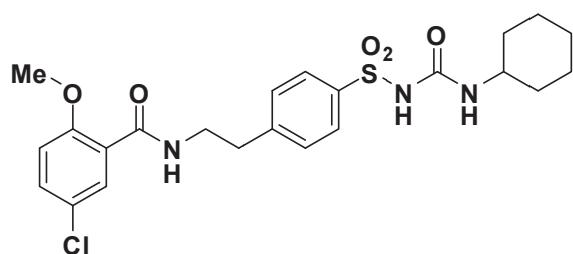
White powder (91% yield); **<sup>1</sup>H-NMR** (400 MHz, DMSO-*d*6) δ 1.03-1.22 (m, 5H), δ 1.46-1.66 (m, 5H), δ 3.26 (d, *J* = 7.28Hz, 1H), δ 6.39 (d, *J* = 8.40Hz, 1H) δ 7.66 (d, *J* = 8.28Hz, 2H), 7.88 (d, *J* = 8.64Hz, 2H), δ 10.44 (s, 1H); **<sup>13</sup>C-NMR** (300 MHz, DMSO-*d*6) δ 24.6, 25.4, 32.7, 48.6, 129.6, 129.7, 138.4; **HRMS**: Calculated for C<sub>13</sub>H<sub>17</sub>ClN<sub>2</sub>NaO<sub>3</sub>S [M+Na]: 339.0541; measured: 339.0541.

**5-chloro-2-methoxy-N-(4-sulfamoylphenethyl)benzamide 3a**



White powder (74% yield); **<sup>1</sup>H-NMR** (400 MHz, DMSO-*d*6) δ 2.89 (t, *J* = 7.04Hz, 2H), δ 3.48-3.53 (m, 2H), δ 3.79 (s, 3H) δ 7.14 (d, *J* = 8.96Hz, 2H), δ 7.29 (s, 2H), δ 7.43 (d, *J* = 8.24Hz, 2H), δ 7.47-7.50 (m, 1H), δ 7.62 (d, *J* = 2.84Hz, 1H), δ 7.74 (d, *J* = 8.12Hz, 2H), δ 8.25 (t, *J* = 5.52Hz, 1H); **<sup>13</sup>C-NMR** (300 MHz, DMSO-*d*6) δ 35.1, 56.7, 114.6, 124.8, 125.3, 126.1, 129.6, 130.0, 131.2, 142.5, 144.1, 156.1, 164.0. **HRMS**: Calculated for C<sub>16</sub>H<sub>17</sub>ClN<sub>2</sub>NaO<sub>4</sub>S [M+Na]: 391.0490; measured: 391.0478.

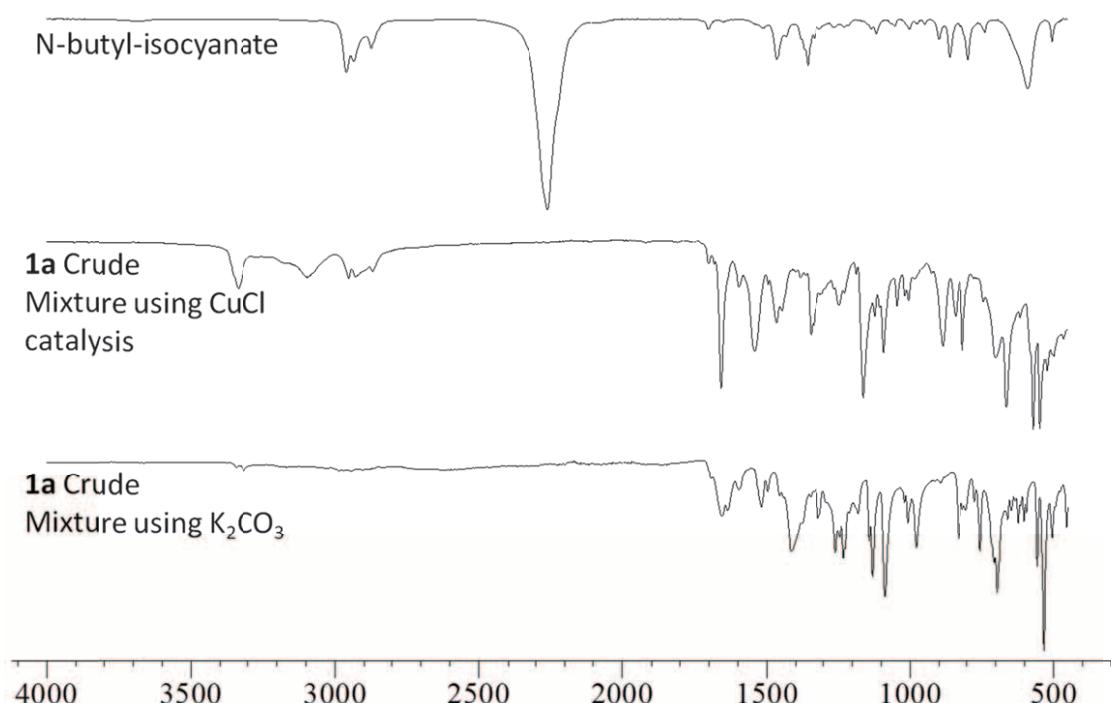
**5-chloro-N-(4-(N-(cyclohexylcarbamoyl)sulfamoyl)phenethyl)-2-methoxybenzamide 3b**



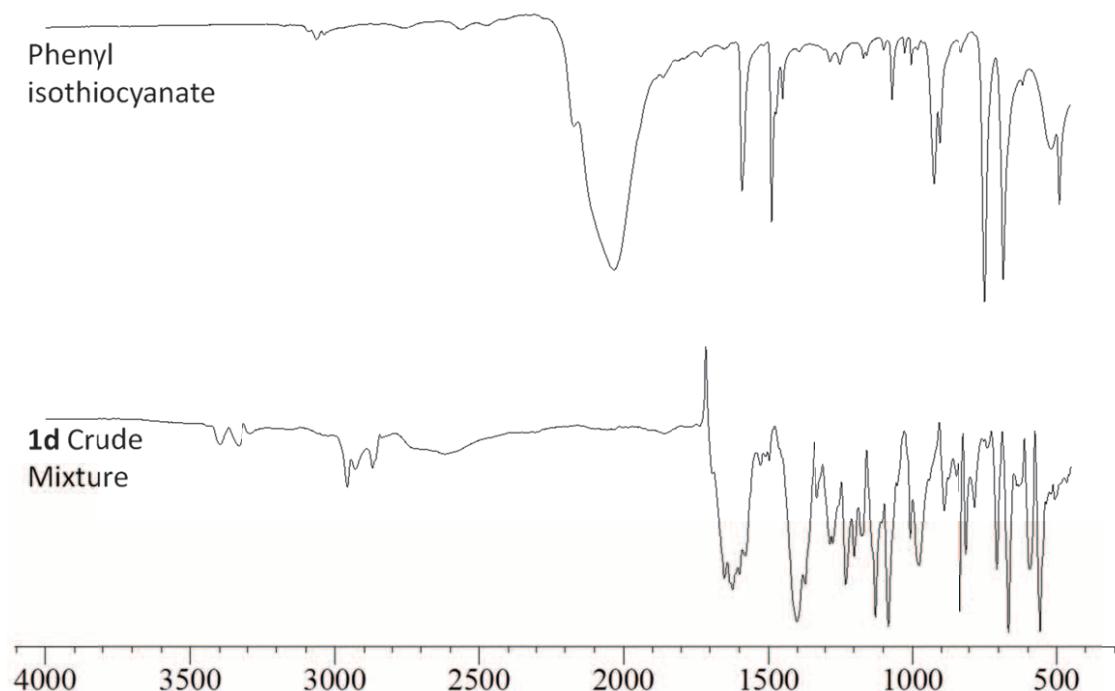
White powder; <sup>1</sup>H-NMR (400 MHz, DMSO-*d*6) δ 1.03-1.22 (m, 5H), δ 1.46-1.66 (m, 5H), 2.89 (t, *J* = 7.04Hz, 2H), δ 3.26 (d, *J* = 7.28Hz, 1H), δ 3.48-3.53 (m, 2H), δ 3.79 (s, 3H) δ 7.14 (d, *J* = 8.96Hz, 2H), δ 7.43 (d, *J* = 8.24Hz, 2H), δ 7.47-7.50 (m, 1H), δ 7.62 (d, *J* = 2.84Hz, 1H), δ 7.74 (d, *J* = 8.12Hz, 2H), δ 8.25 (t, *J* = 5.52Hz, 1H); <sup>13</sup>C-NMR (300 MHz, DMSO-*d*6) δ 24.6, 25.4, 31.2, 32.7, 35.1, 48.5, 56.7, 114.6, 124.7, 125.3, 127.7, 129.8, 129.9, 131.9, 138.6, 145.7, 150.8, 156.1, 164.0; HRMS: Calculated for C<sub>23</sub>H<sub>28</sub>ClN<sub>3</sub>NaO<sub>5</sub>S [M+Na]: 516.1330; measured: 516.1331.

### 3. Selected Fourier-transform attenuated total reflectance (FTIR-ATR) spectra

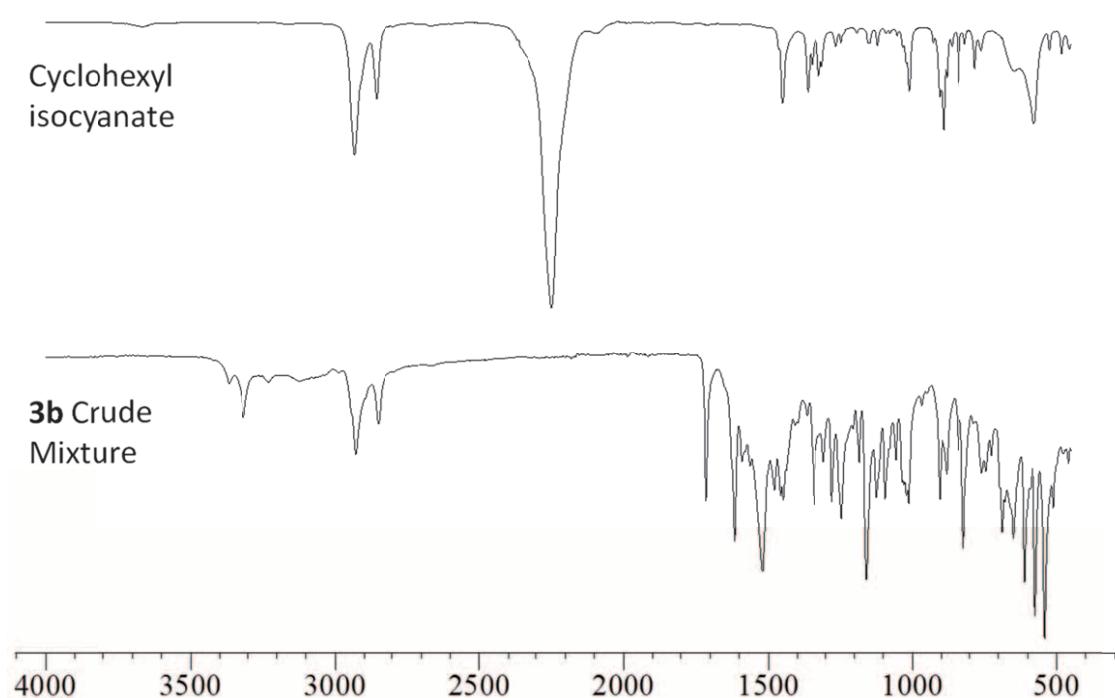
According to powder X-ray diffraction the crude reaction mixtures were amorphous and we were not able to detect product formation by solid-state diffraction methods. However, comparisons of FTIR-ATR spectra of crude reaction mixtures from both base-assisted and copper-catalysed reactions showed that all of the iso(thio)cyanate had been consumed by the disappearance of the characteristic absorption band around 2200 cm<sup>-1</sup>.



**Figure S1.** FTIR-ATR spectra for *n*-butylisocyanate reactant and the freshly prepared crude reaction mixture from the synthesis of tolbutamide (**1a**) via base-assisted and copper-catalysed routes. The disappearance of the isocyanate stretching band at 2200 cm<sup>-1</sup> is evident in both cases.

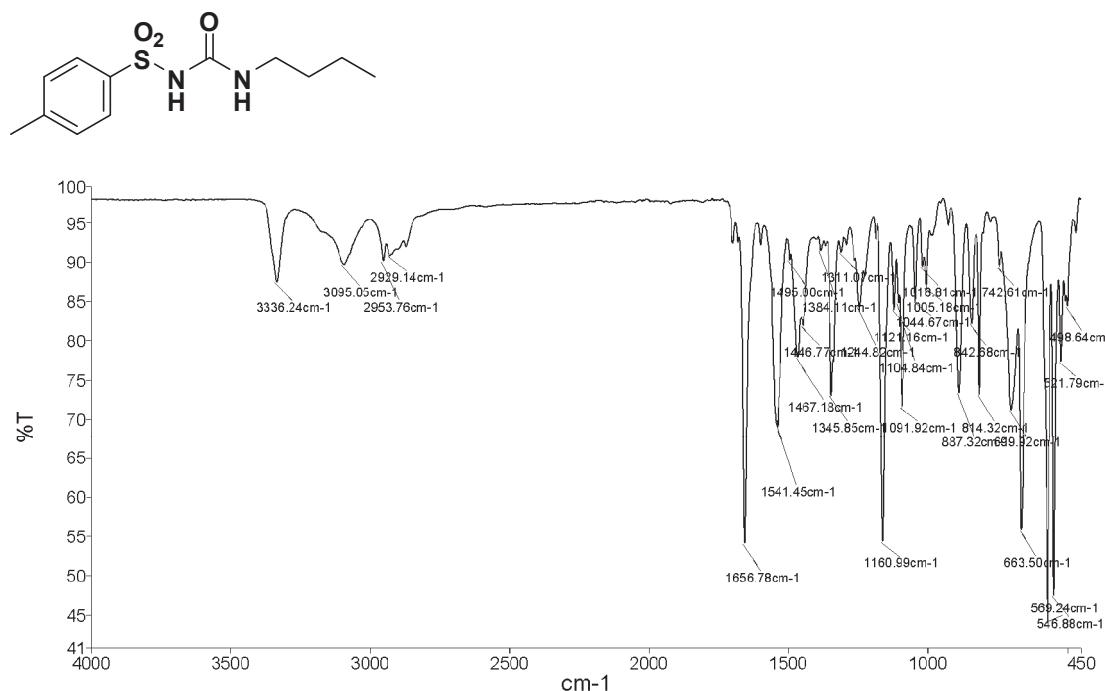


**Figure S2.** FTIR-ATR spectra for phenyl-isothiocyanate reactant and the freshly prepared crude reaction mixture from the synthesis of the sulfonyl-thiourea **1d** via base-assisted mechanochemical route. The disappearance of the thioisocyanate stretching band at ca. 2000 cm<sup>-1</sup> is evident.

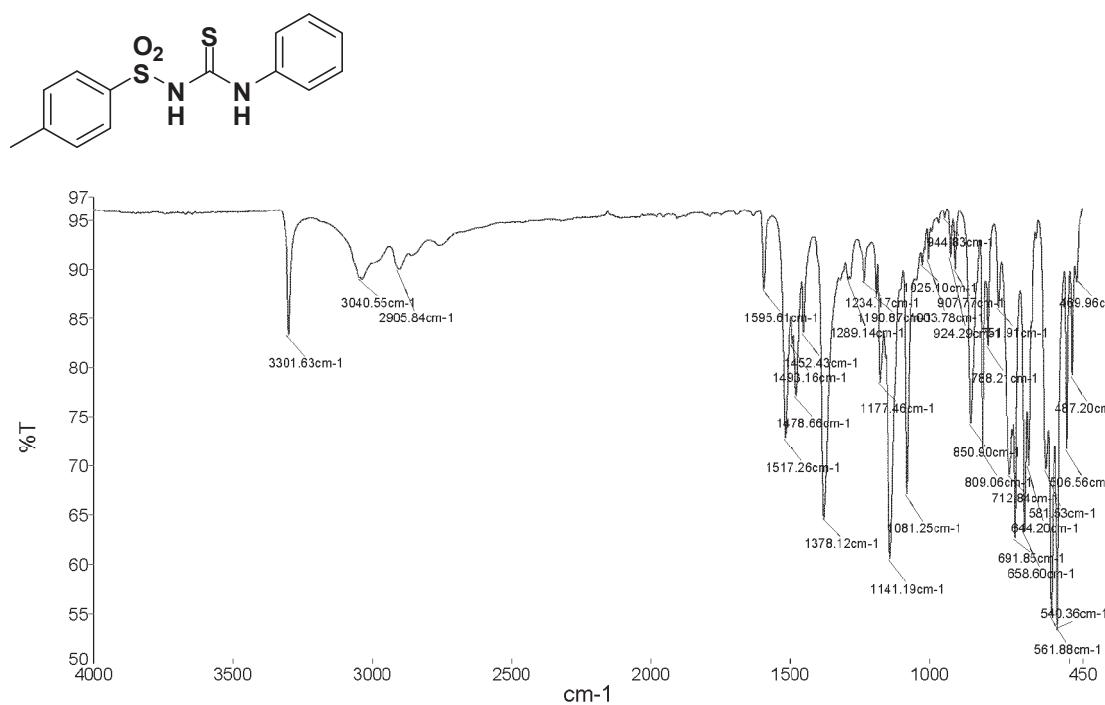


**Figure S3.** FTIR-ATR spectra for cyclohexyl-isocyanate reactant and the freshly prepared crude reaction mixture from the synthesis of glibenclamide (**3b**) via copper-catalysed mechanochemical route. The disappearance of the isocyanate stretching band at ca. 2000 cm<sup>-1</sup> is evident.

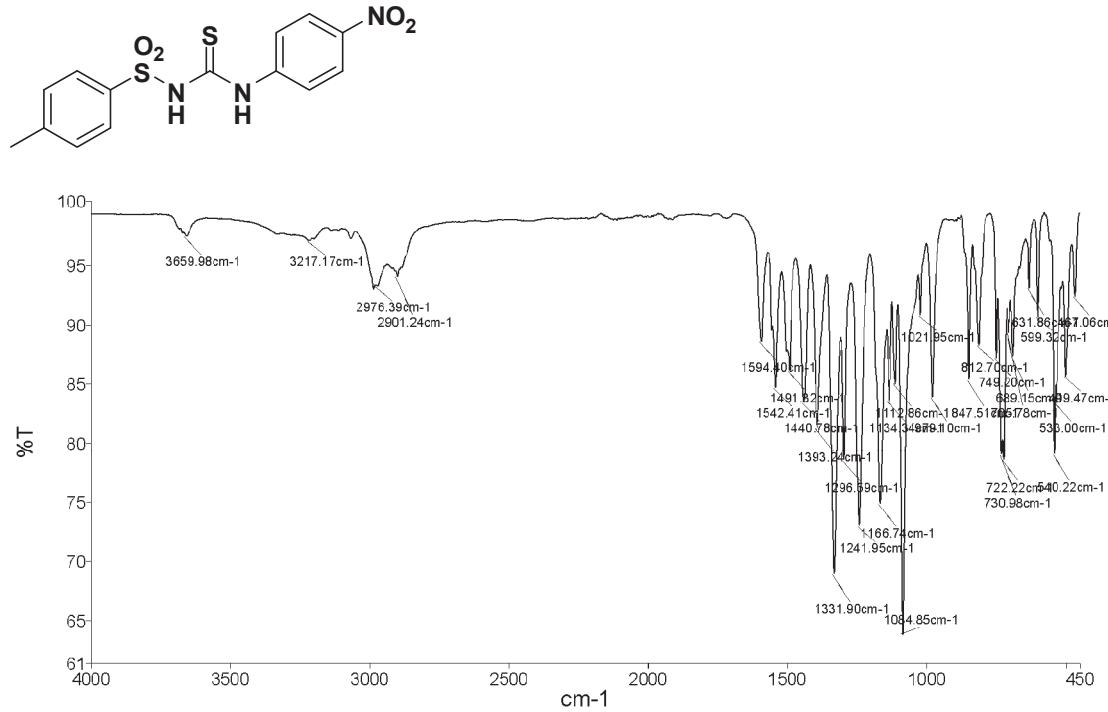
**FTIR-ATR spectrum of *N*-(butylcarbamoyl)-4-methylbenzenesulfonamide 1a**



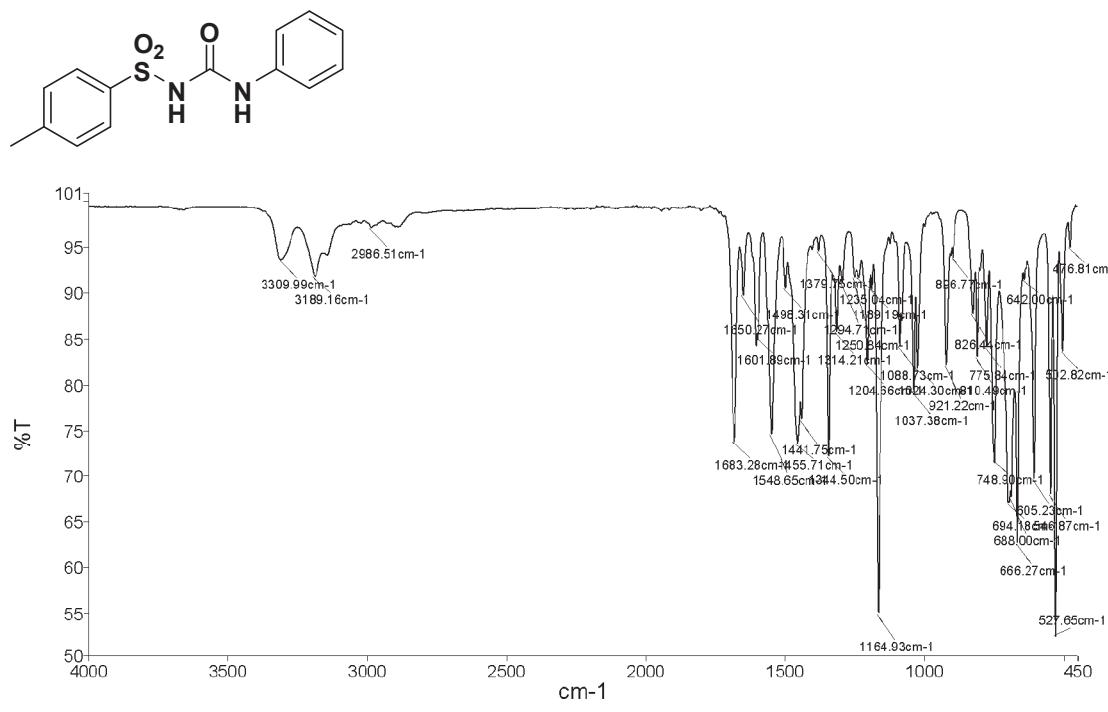
**FTIR-ATR spectrum of 4-methyl-N-(phenylcarbamothioyl)benzenesulfonamide 1b**



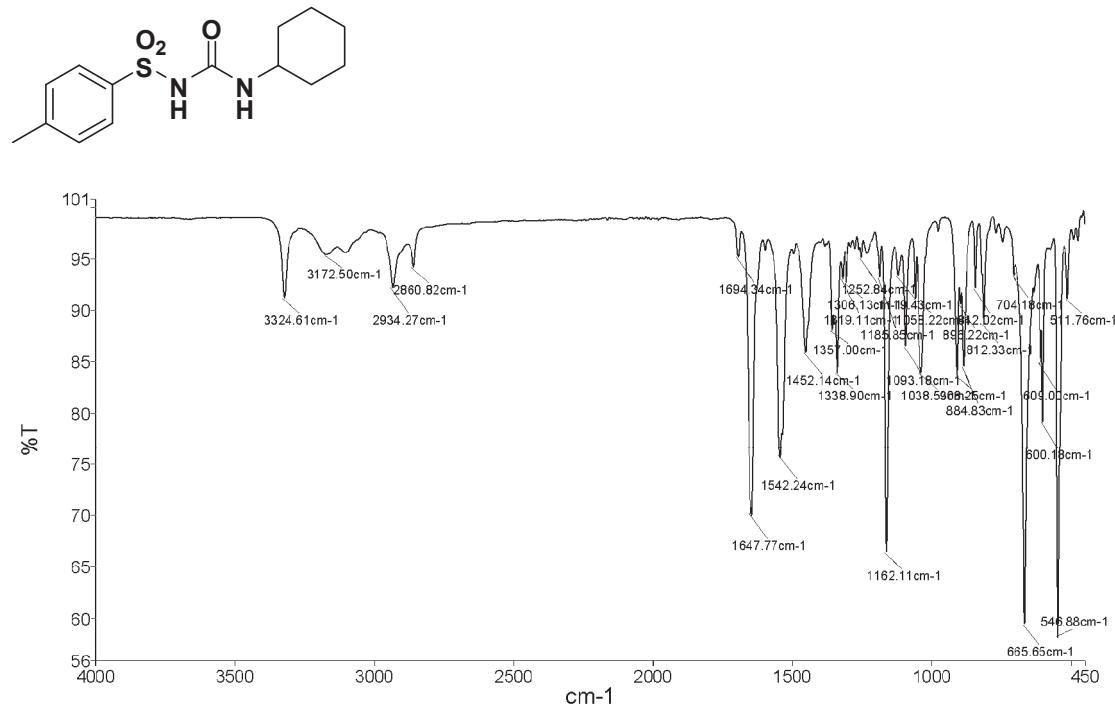
**FTIR-ATR spectrum of 4-methyl-N-(4-nitrophenylcarbamothioyl)benzenesulfonamide 1c**



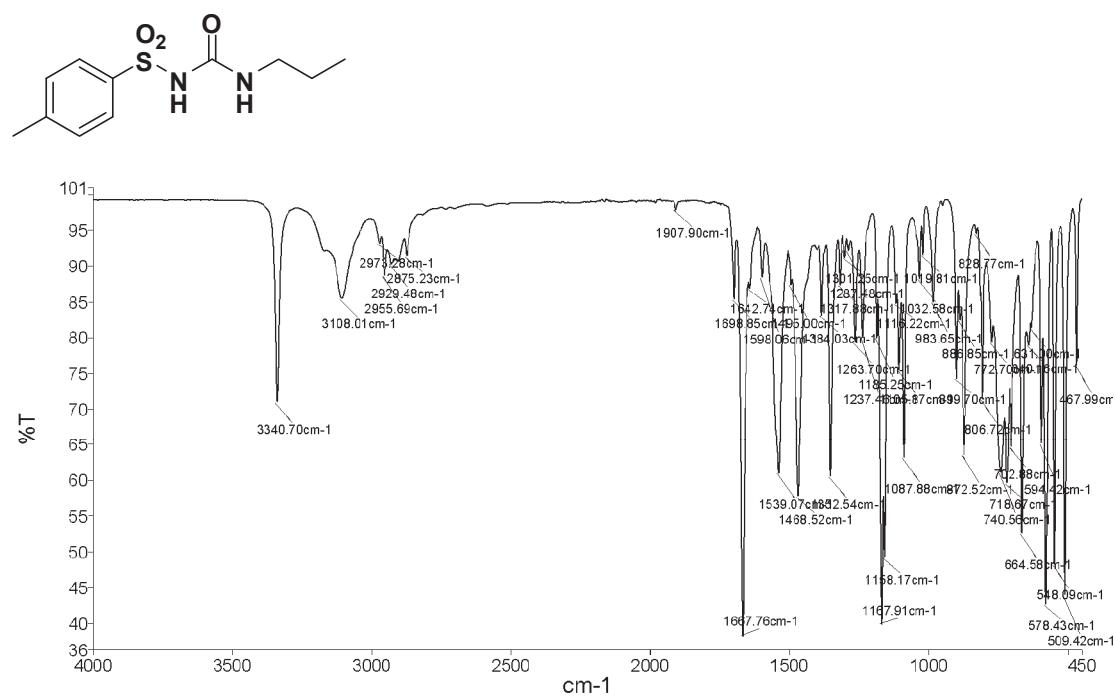
**FTIR-ATR spectrum of 4-methyl-N-(phenylcarbamoyl)benzenesulfonamide 1d**



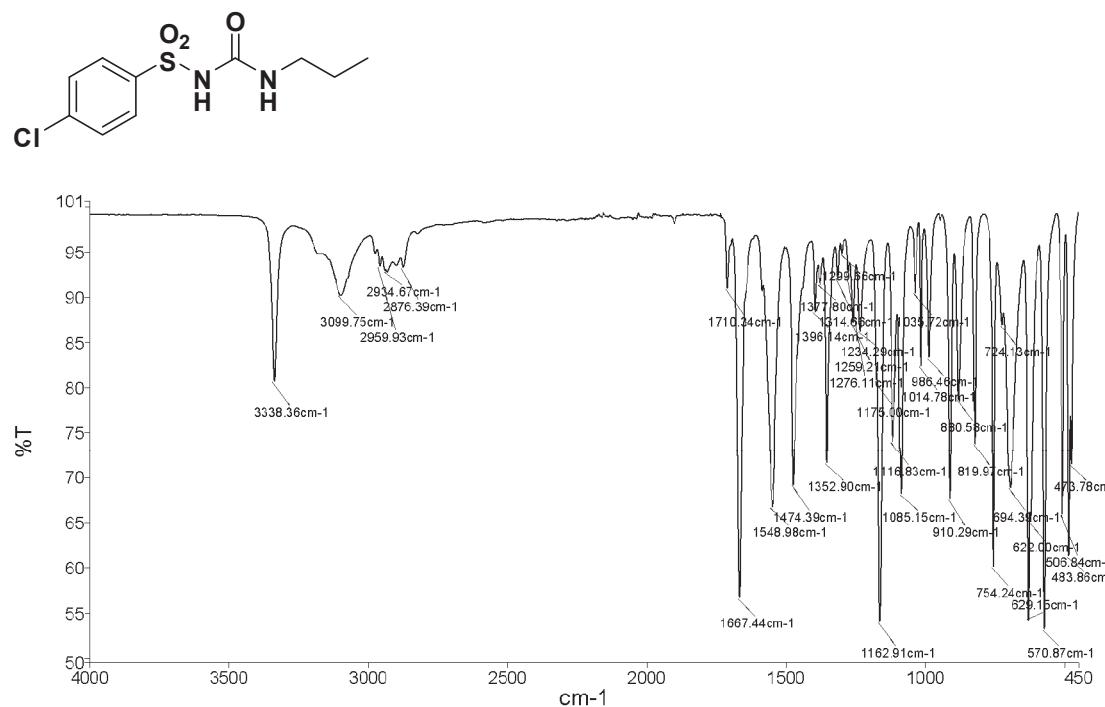
**FTIR-ATR spectrum of *N*-(cyclohexylcarbamoyl)-4-methylbenzenesulfonamide **1e****



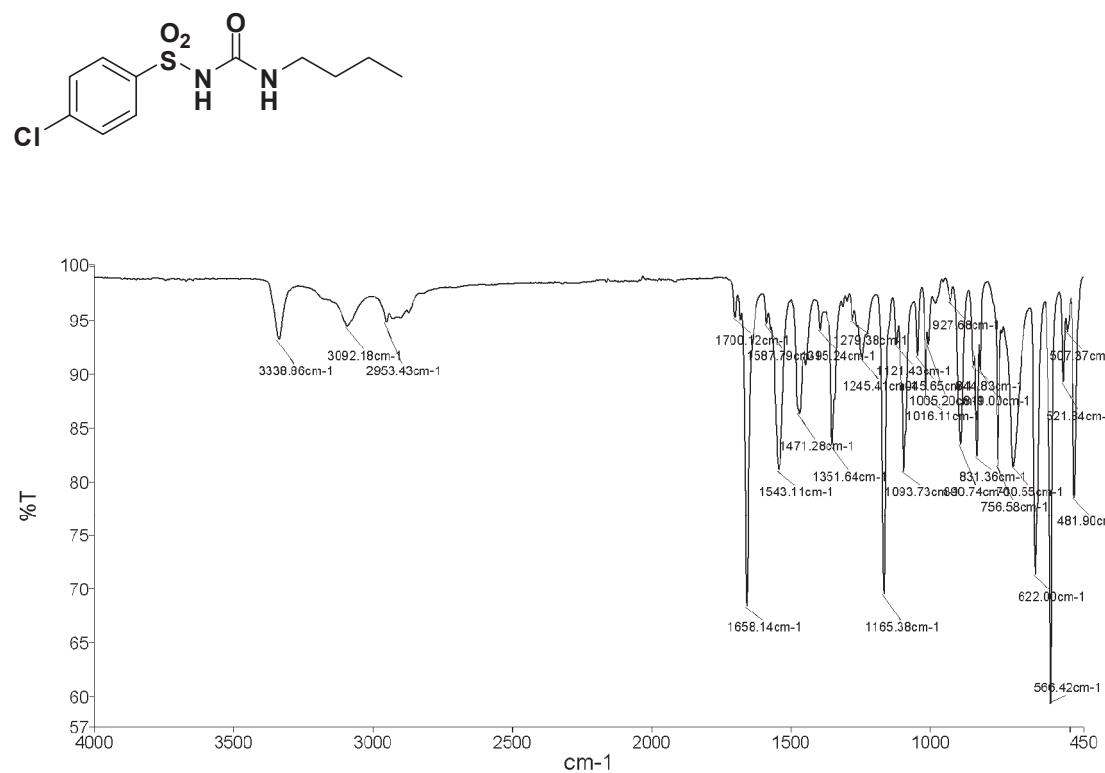
**FTIR-ATR spectrum of 4-methyl-*N*-(propylcarbamoyl)benzenesulfonamide **1f****



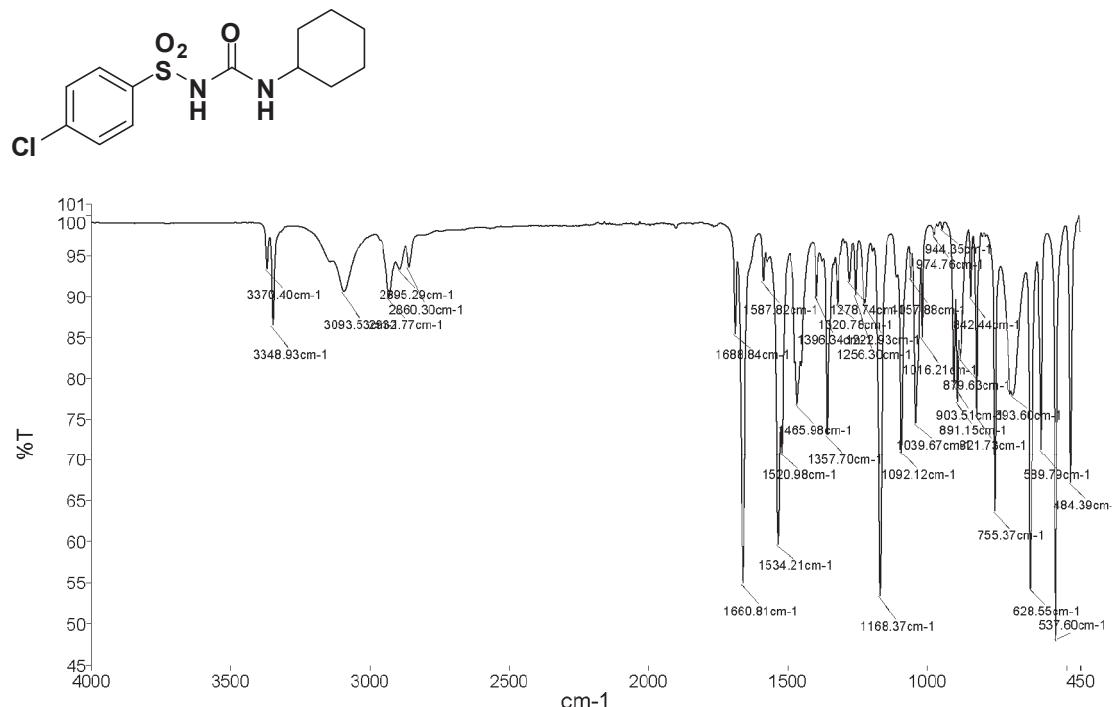
**FTIR-ATR spectrum of 4-chloro-N-(propylcarbamoyl)benzenesulfonamide 2a**



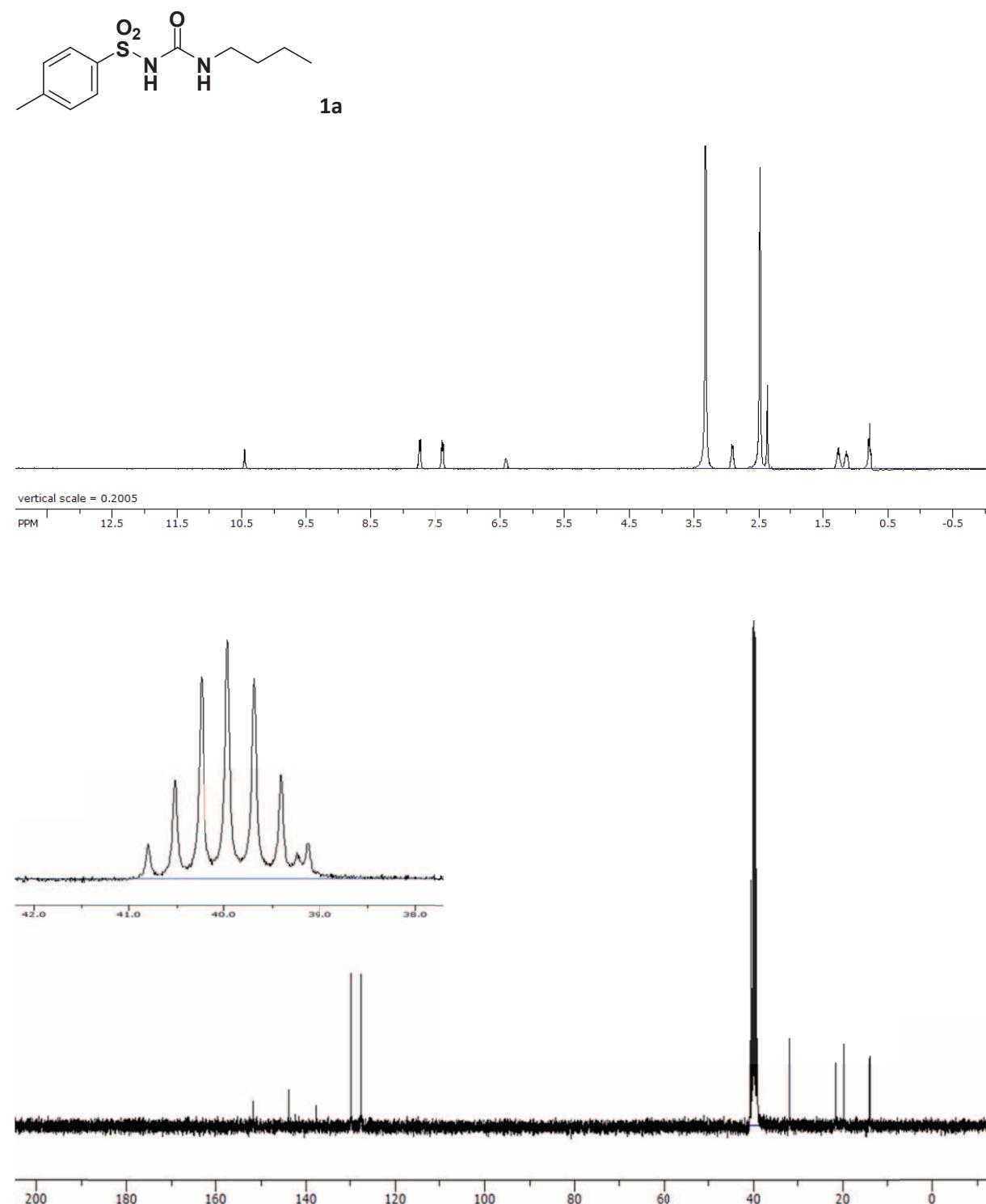
**N-(butylcarbamoyl)-4-chlorobenzenesulfonamide 2b**



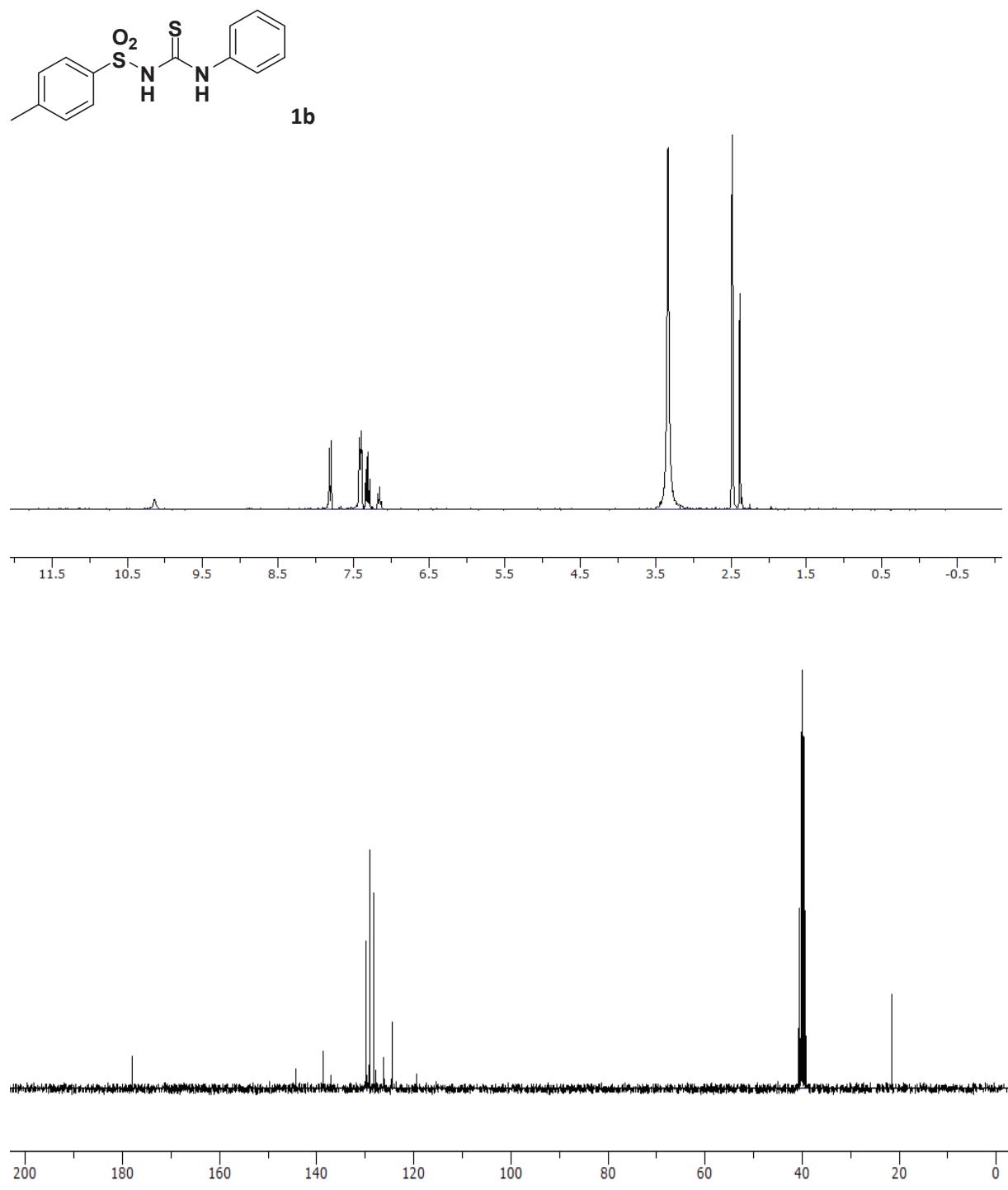
**4-chloro-N-(cyclohexylcarbamoyl)benzenesulfonamide 2c**



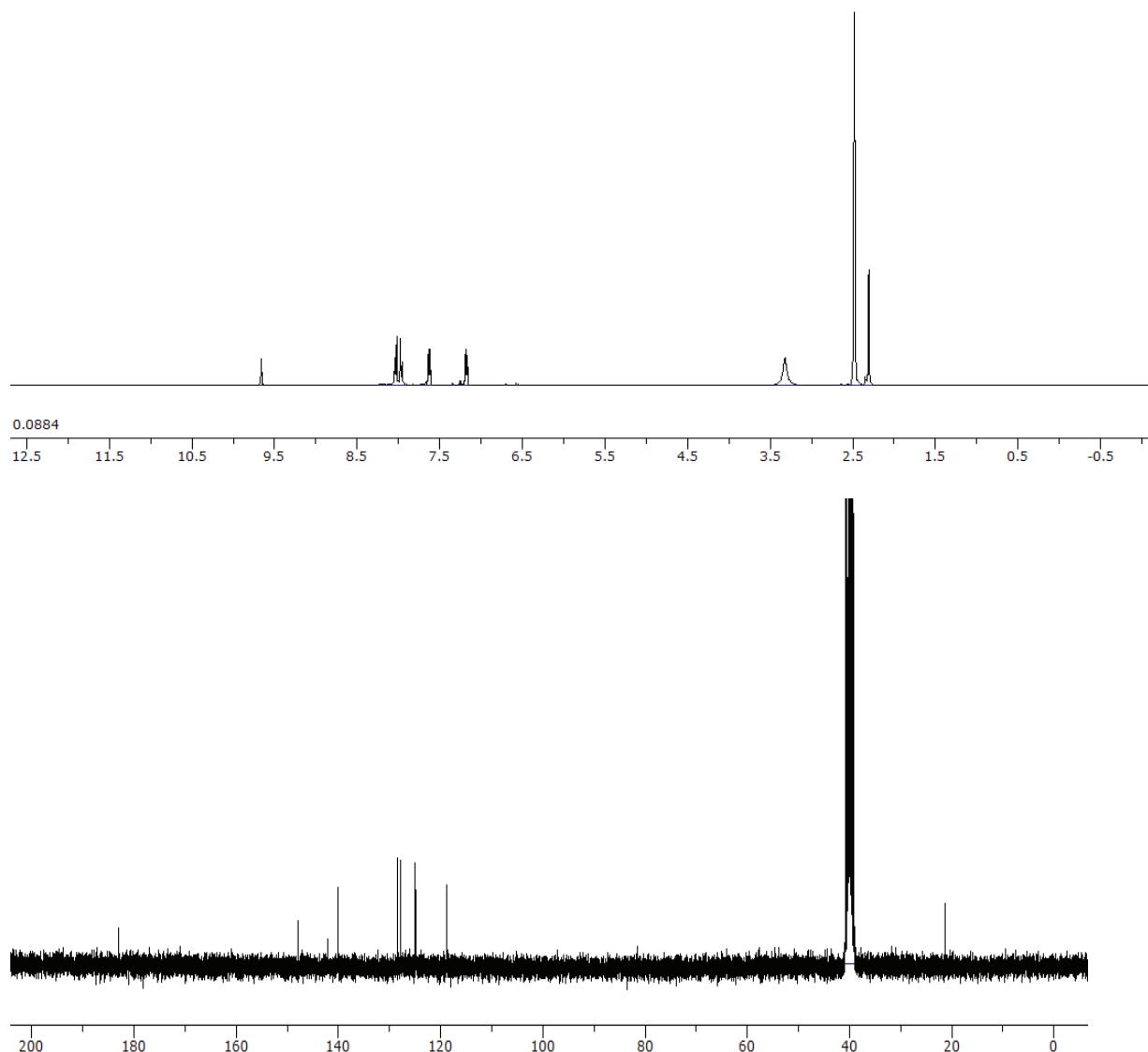
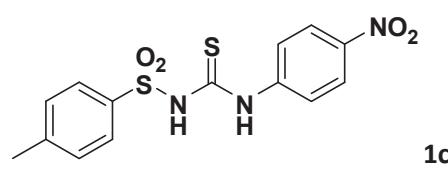
#### 4. Selected $^1\text{H}$ and $^{13}\text{C}$ NMR spectra



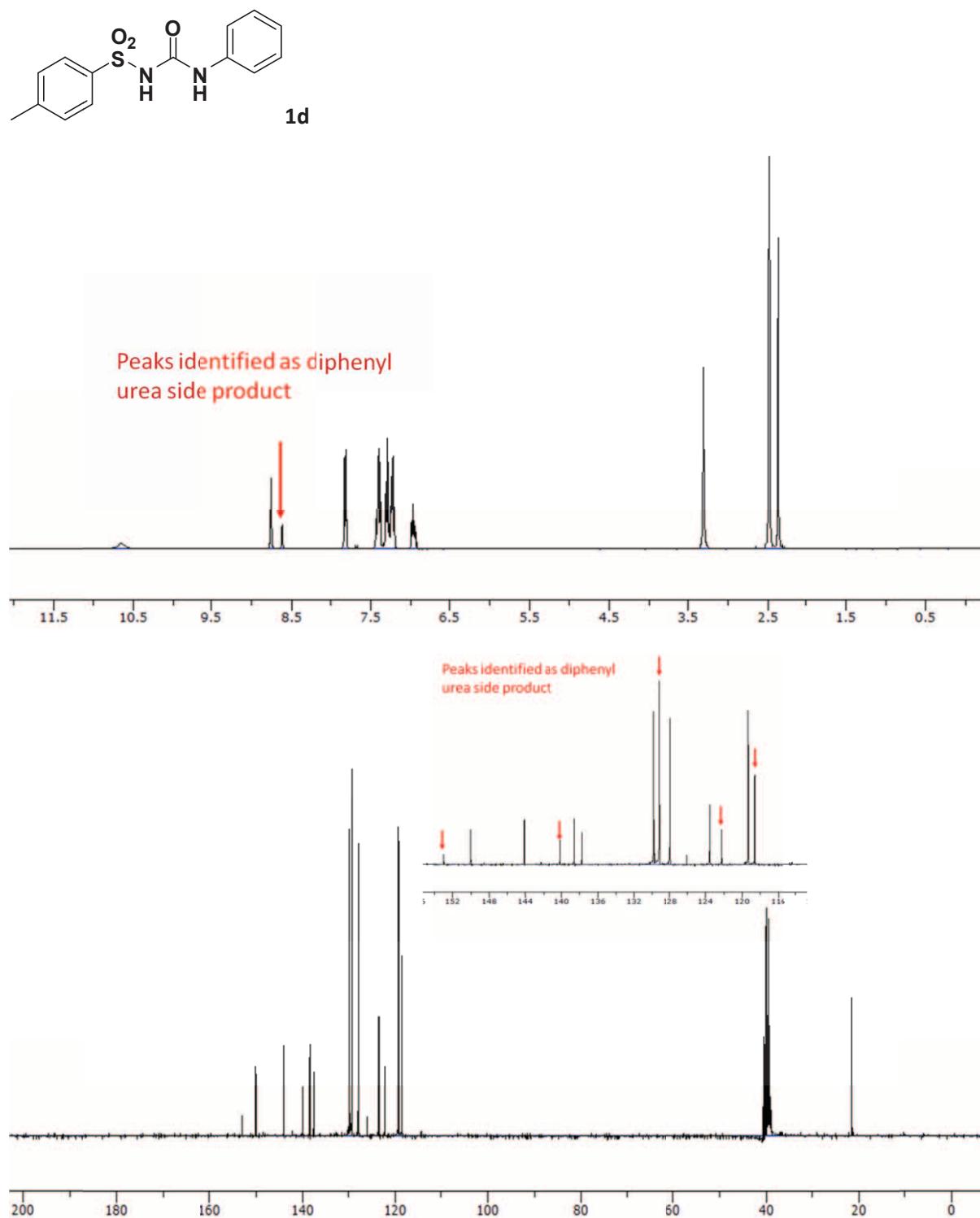
**Figure S4.**  $^1\text{H}$  (top) and  $^{13}\text{C}$  (bottom) NMR spectra for sulfonyl-urea **1a** (tolbutamide).



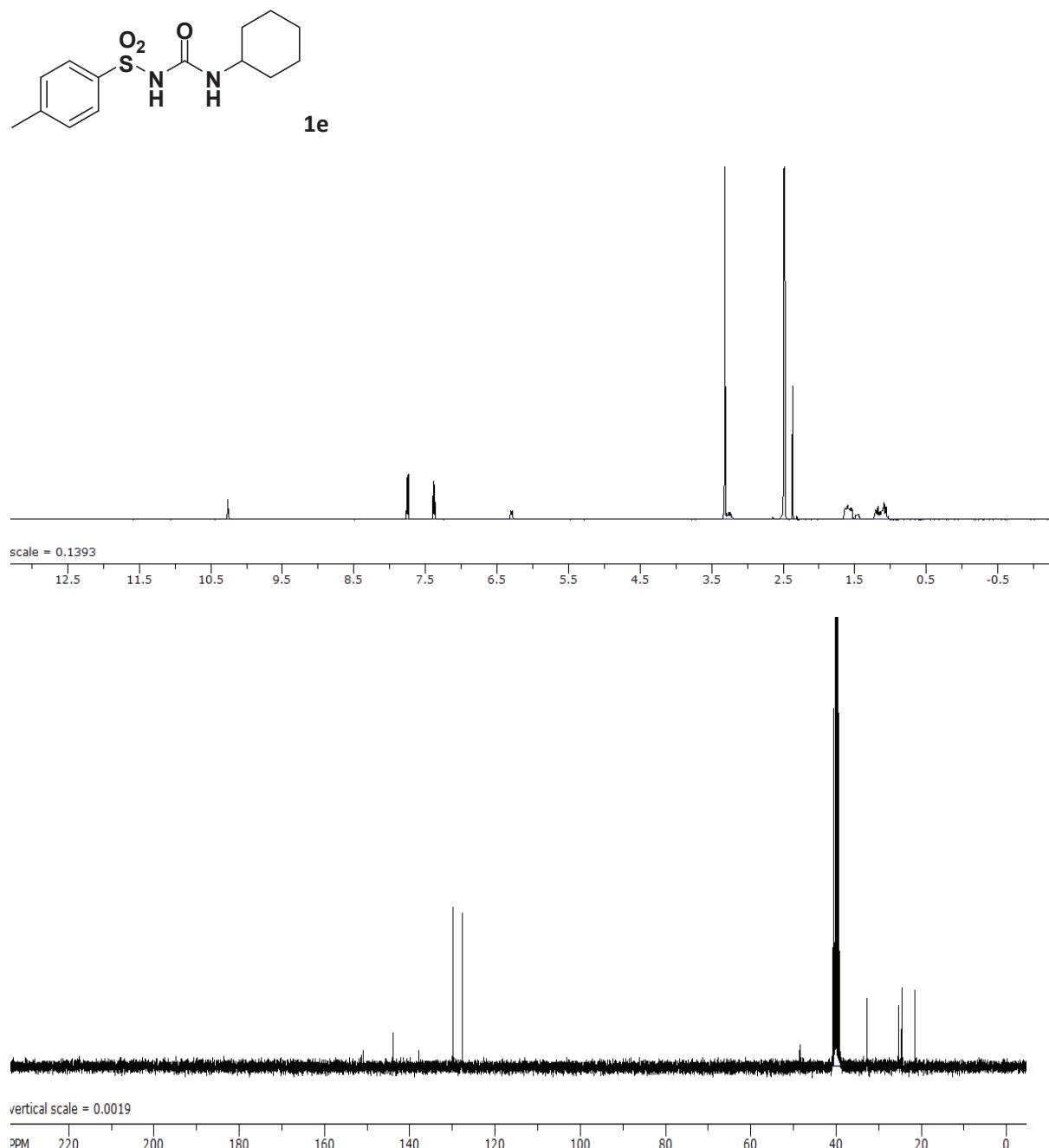
**Figure S5.** <sup>1</sup>H (top) and <sup>13</sup>C (bottom) NMR spectra for sulfonyl-thiourea **1b**.



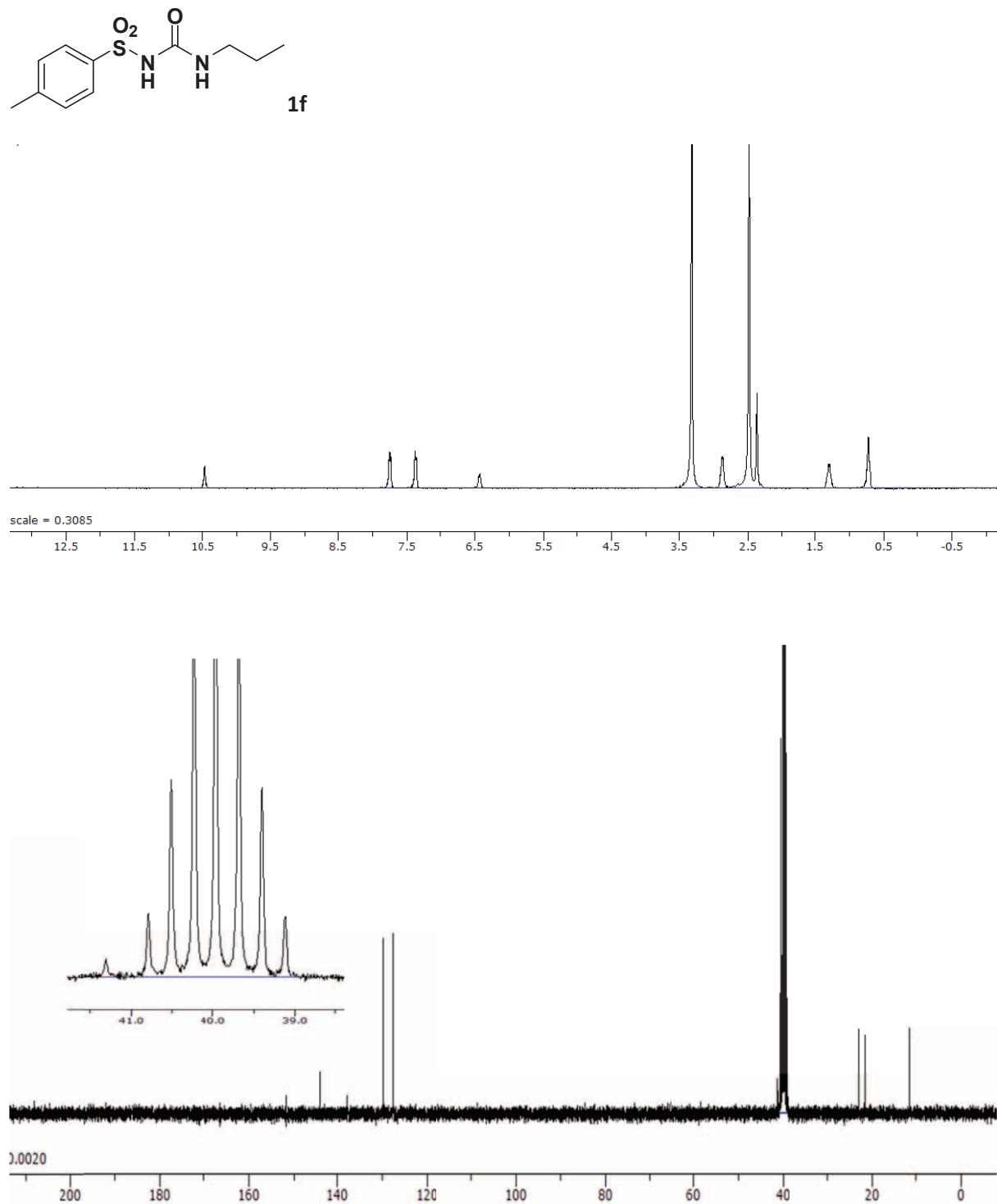
**Figure S6.**  $^1\text{H}$  (top) and  $^{13}\text{C}$  (bottom) NMR spectra for sulfonyl-thiourea **1c**.



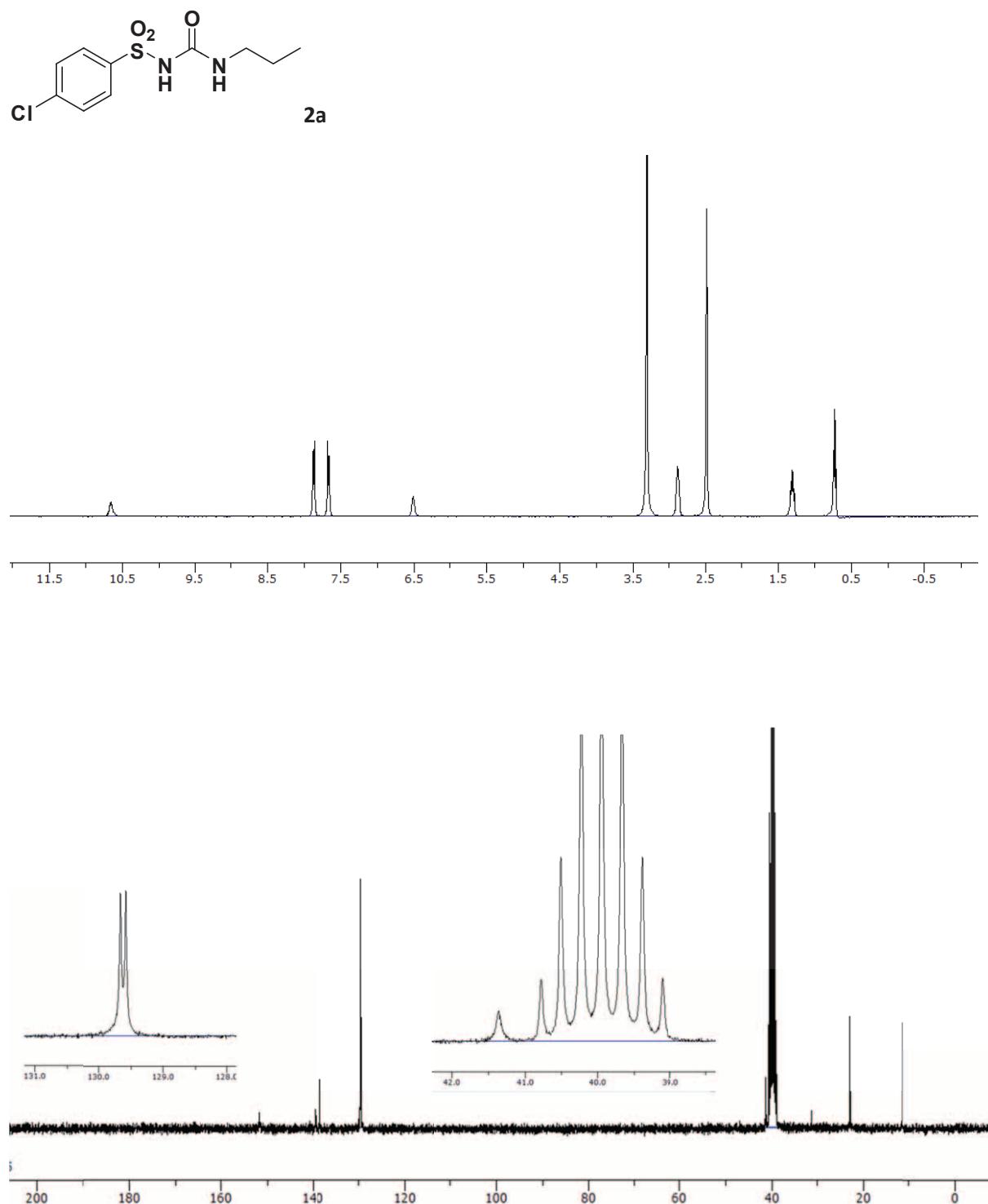
**Figure S7.** <sup>1</sup>H (top) and <sup>13</sup>C (bottom) NMR spectra for sulfonyl-urea **1d**. In this case the competing side-reaction of isocyanate coupling to form diphenylurea was difficult to avoid and the crude product contained 93% yield of **1d**, calculated from the <sup>1</sup>H NMR spectrum.



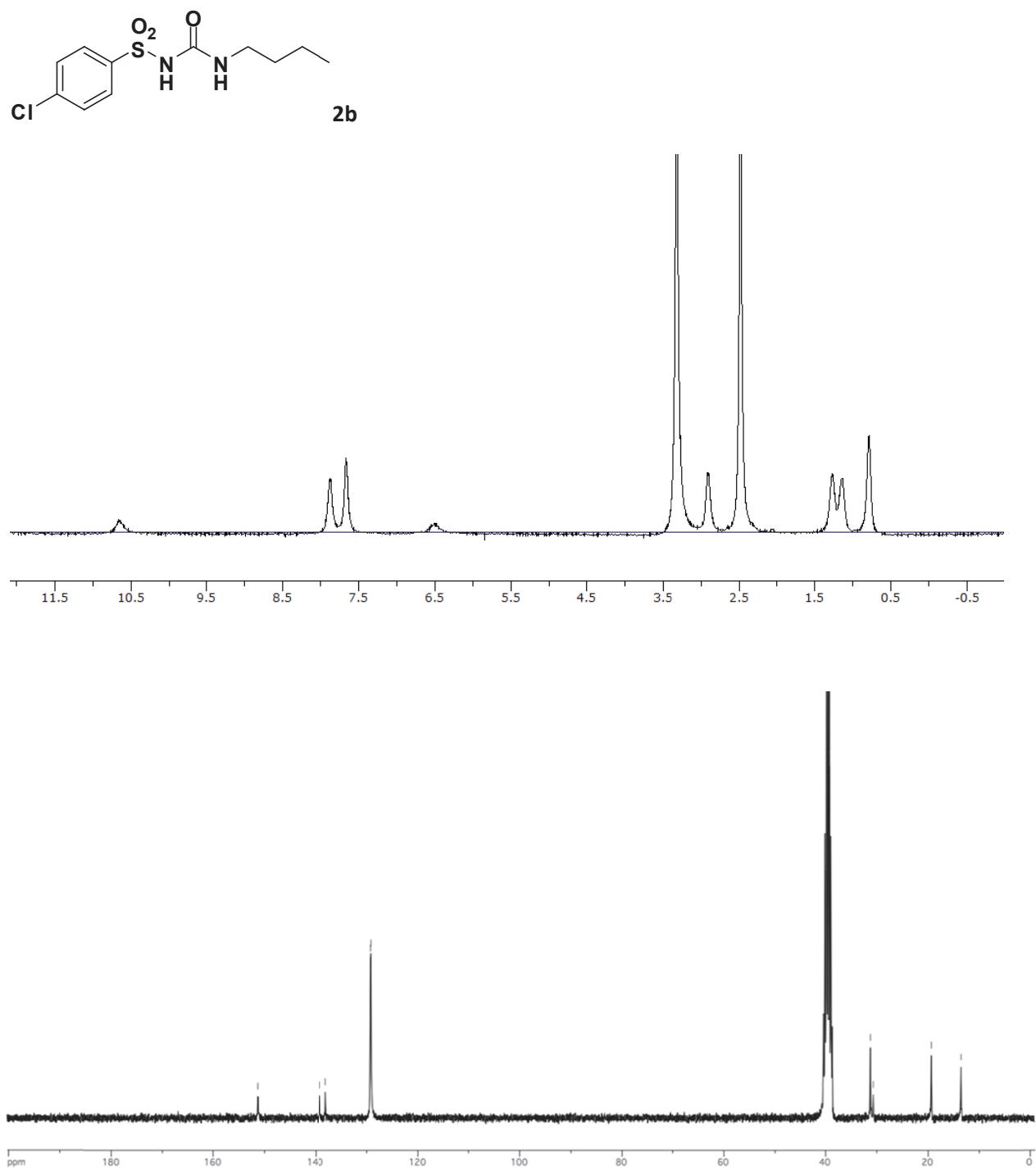
**Figure S8.** <sup>1</sup>H (top) and <sup>13</sup>C (bottom) NMR spectra for sulfonyl-urea **1e**.



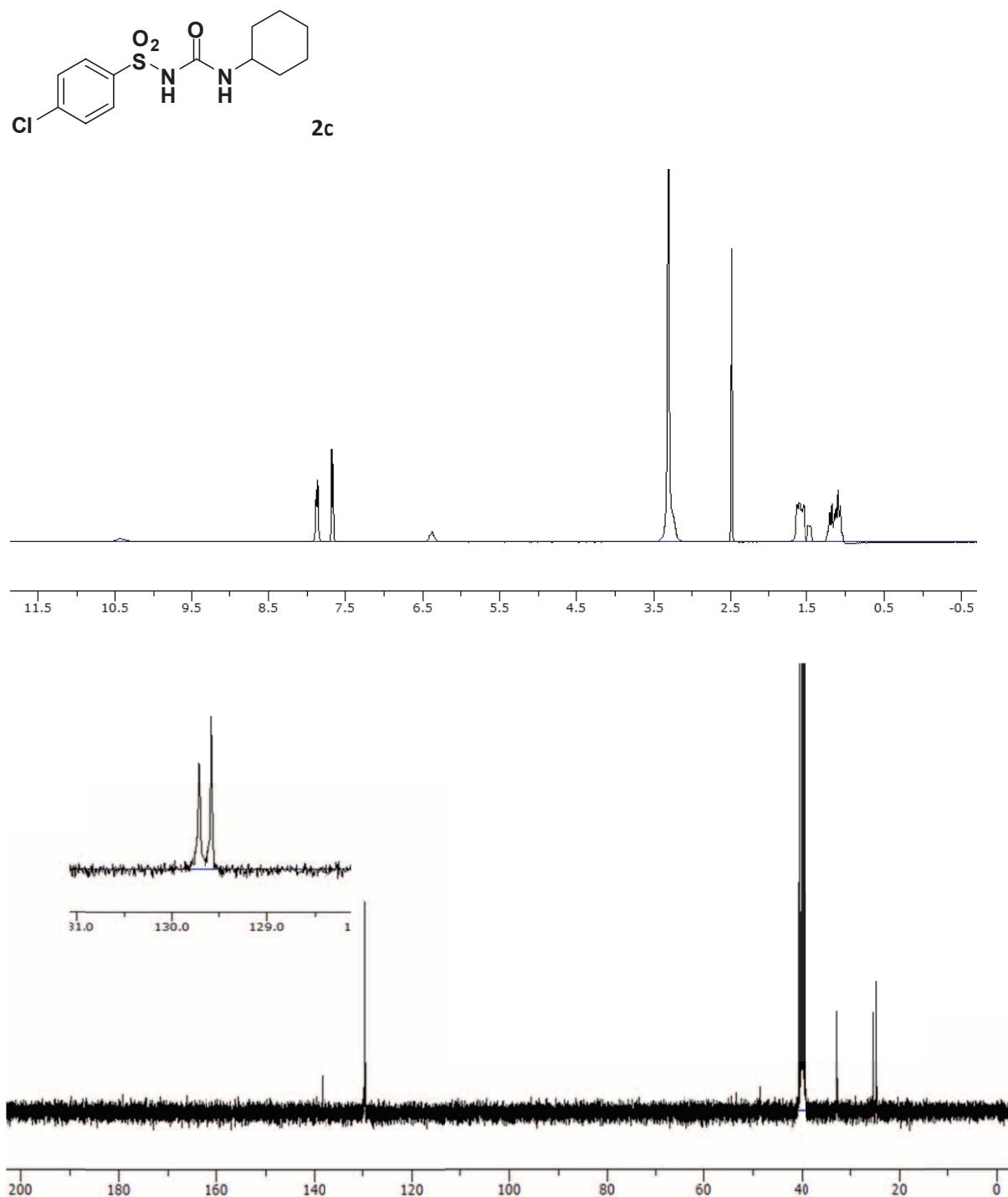
**Figure S9.** <sup>1</sup>H (top) and <sup>13</sup>C (bottom) NMR spectra for sulfonyl-urea **1f**.



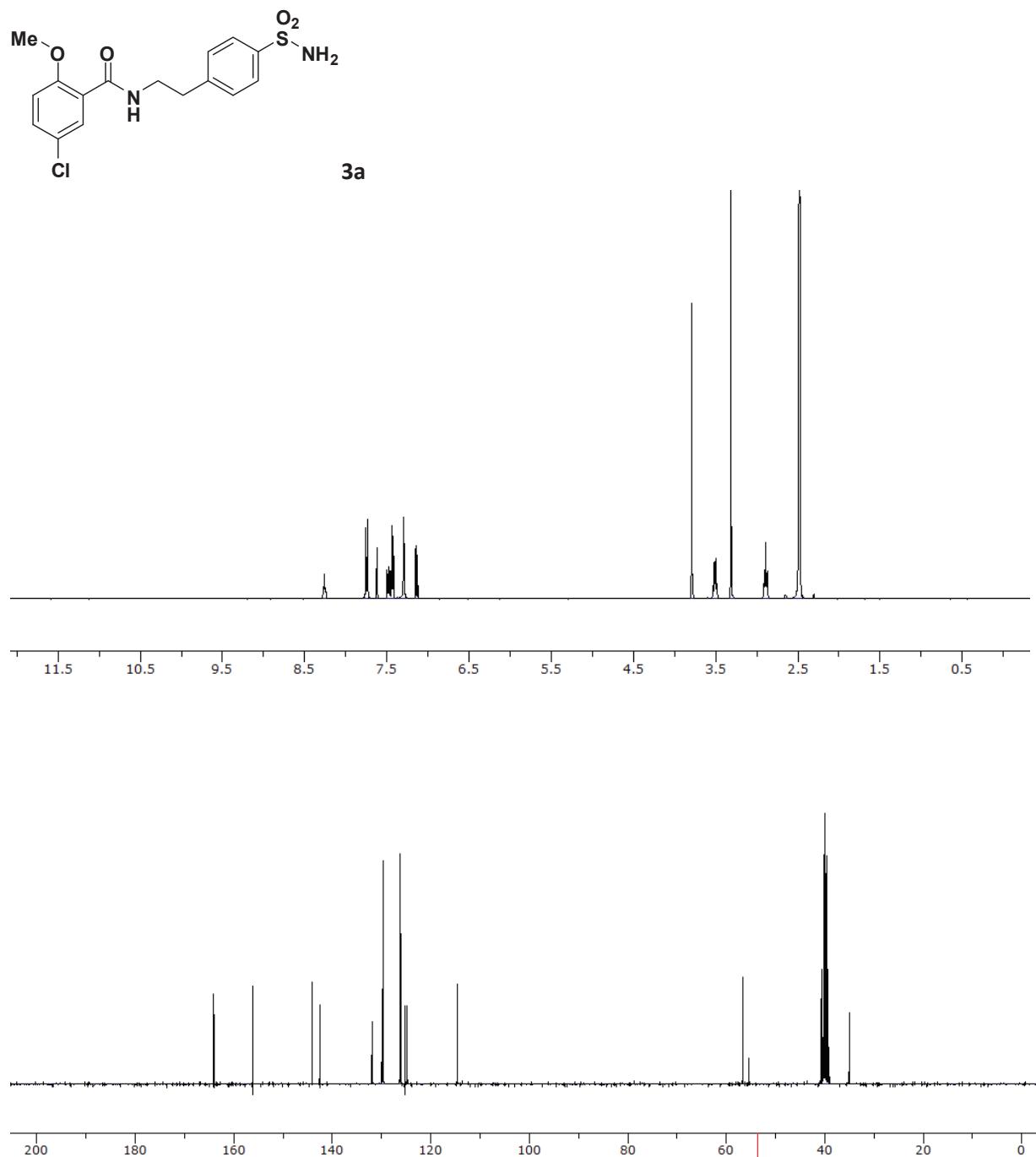
**Figure S10.** <sup>1</sup>H (top) and <sup>13</sup>C (bottom) NMR spectra for sulfonyl-urea **2a** (chlorpropamide).



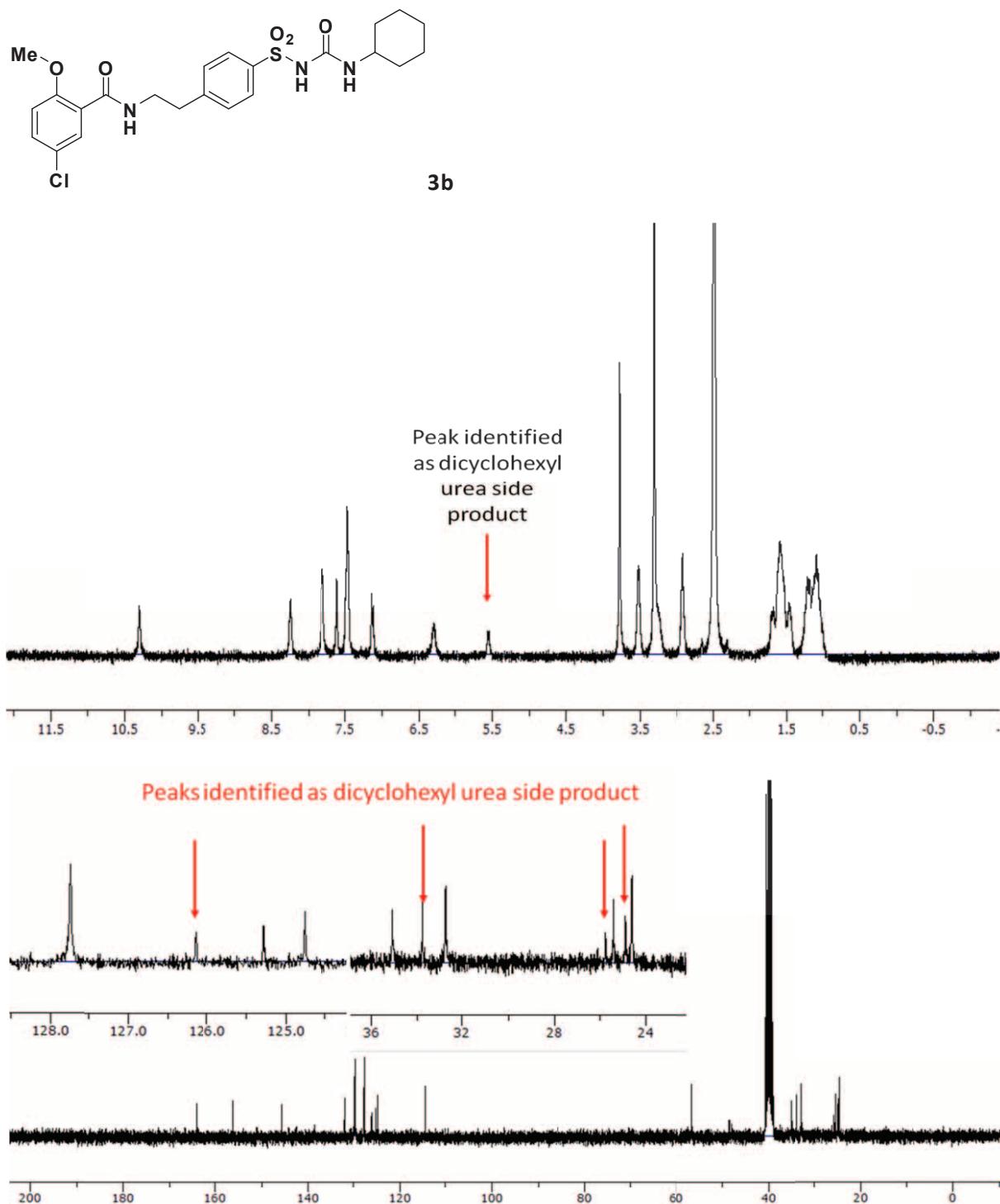
**Figure S11.** <sup>1</sup>H (top) and <sup>13</sup>C (bottom) NMR spectra for sulfonyl-urea **2b**.



**Figure S12.** <sup>1</sup>H (top) and <sup>13</sup>C (bottom) NMR spectra for sulfonyl-urea **2c**.



**Figure S13.** <sup>1</sup>H (top) and <sup>13</sup>C (bottom) NMR spectra for the glibenclamide precursor **3a**, obtained by mechanochemical amide synthesis.



**Figure S14.**  $^1\text{H}$  (top) and  $^{13}\text{C}$  (bottom) NMR spectra for glibenclamide (**3b**) obtained by mechanochemical amide coupling and mechanochemical copper-catalysed coupling.