Supplementary Information

# Conversion of 5-Hydroxymethylfurfural to a Cyclopentanone Derivative by Ring Rearrangement over Supported Au Nanoparticles

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### Material.

Nb<sub>2</sub>O<sub>5</sub> was obtained by calcination of niobium oxalate supplied from CBMM at 773 K under air. TiO<sub>2</sub> (JRC-TIO-8), Al<sub>2</sub>O<sub>3</sub> (JRC-ALO-8), CeO<sub>2</sub> (Rhodia), ZrO<sub>2</sub> (JRC-ZRO-5), and silica-alumina (JRC-SAH-1, denoted as SA) were supplied from the committee of reference catalyst, Catalysis society of Japan. Hydrotalcite (denoted as HT) was obtained from Kyowa Chemical Industry Co. Ltd. La<sub>2</sub>O<sub>3</sub>, HAuCl<sub>4</sub>, H<sub>2</sub>PtCl<sub>6</sub>, RuCl<sub>3</sub>, and NaOH were purchased from Kishida Chemical Co., Ltd. 2,5-Hydroxymethylfurfural (HMF) was obtained from Tokyo Chemical Industry Co., Ltd.

## Catalyst preparation.

Supported Au catalysts were prepared by deposition-precipitation method. A 1 M NaOH aqueous solution was added dropwise to a suspension of supports (0.99 g) in 50 mL of 1 mM HAuCl<sub>4</sub> aqueous solution, and the pH of the solution was adjusted to ca. 7. The suspension was stirred at 343 K for 1 h, followed by centrifugation. The resulting residue was washed with ca. 40 mL of water and centrifuged. The washing process was repeated 5 times. The resulting powder was dried at 353 K, and calcined under air at 573 K to prepare various metal oxides supported Au catalysts.

As references,  $Pt/Nb_2O_5$  and  $Pd/Nb_2O_5$  was prepared by impregnation method to be 1wt% of metal loading by using H<sub>2</sub>PtCl<sub>6</sub> and Pd(NO<sub>3</sub>)<sub>2</sub> as metal precursor. After the suspension containing metal precursor and Nb<sub>2</sub>O<sub>5</sub> in water was stirred under 343 K for 1h, the suspension was evaporated to dryness. Resulting powder was dried at 353 K, and calcined at 573 K under air. Ru/Nb<sub>2</sub>O<sub>5</sub> was prepared in the same method as Au/Nb<sub>2</sub>O<sub>5</sub> using RuCl<sub>3</sub> as a precursor. To a suspension of Nb<sub>2</sub>O<sub>5</sub> (0.99 g) in 50 mL of 2 mM RuCl<sub>3</sub> aqueous solution, A 1 M NaOH aqueous solution was added dropwise to adjust the pH of the solution to ca. 7. The suspension was stirred at 343 K for 1 h. After centrifugation, the residue was washed with ca. 40 mL of water. The washing process was repeated 5 times. The resulting powder was dried at 353 K, and then calcined under air at 573 K.

#### Catalytic test.

In a typical reaction, 3 mL of 0.067 M HMF aqueous solution and 10 mg of a supported catalyst were added to an autoclave with a gas in- and outlet (30mL, Taiatsu Techno Co., TVS-1 type). Prior to the reaction, the catalysts were pretreated under  $H_2$  at 473 K and subsequently under  $O_2$  at 473 K. The reaction was conducted under 8 MPa of  $H_2$  at 413 K, after 5 times of replacement of air in an autoclave by ca. 1 MPa of  $H_2$ . Sample analysis was performed using a gas chromatograph (SHIMADZU GC-14A) equipped with FID with Ultra ALLOY<sup>+</sup>-5 capillary column (Frontier Laboratories Ltd.). Products were identified by using a NMR spectrometer (Varian, INOVA-500, 500 MHz) and GC-MS (SHIMADZU-QP5000). Prior to the NMR analysis, the products were purified by column chromatography on silica gel eluted with ethyl acetate or ethyl acetate/methanol (10:1).

#### **Reaction of BHF under N<sub>2</sub>**

We checked the reaction of BHF over  $Au/Nb_2O_5$  under  $N_2$  (BHF: 3 mL aqueous solution, 0.067 M; Temp.: 413 K; Time: 4 h). As a result, 63% yield of HCPEN was obtained. This result supports Scheme 1.

#### **Recycle test.**

After the reaction using  $Au/Nb_2O_5$ , the suspension was filtered off, and the residue was washed with 20 mL of acetone. This washing process was repeated 3 times, and also operated with 20 mL of water. After the dryness at 353 K, the resulting  $Au/Nb_2O_5$  powder was pretreated and applied to the reaction as described above.

#### Catalyst characterization.

Au L<sub>3</sub> edge XAFS measurement was carried out on the BL9C beamline of the Photon Factory (2.5 GeV) with a Si(111) two-crystal monochromator. The XAFS spectra of supported Au catalysts were recorded in fluorescence mode using a Lytle detector, and those of Au foil and Au<sub>2</sub>O<sub>3</sub> references in transmission mode. The X-ray energy was calibrated with a spectrum of Cu foil at 8980 eV. Data reduction of XAFS spectra was performed using REX2000 program Ver. 2.5.9 (Rigaku Co.). Curve-fitting of the Fourier-filtered EXAFS spectra was performed using ab initio-calculated phases and amplitudes from the program FEFF 8.

FT-IR spectra were recorded on a JASCO FT/IR-6100 (JASCO Co.) with a 4 cm<sup>-1</sup> of resolution. A supported Au catalyst (30-50 mg) was pressed into a wafer with 20 mm diameter. The wafer was pretreated under  $10\%O_2/\text{Ar}$  flow (100 mL min<sup>-1</sup>) at 473 K for 30 min, and then exposed to pyridine vapor (12.4 µmol) under Ar flow (90 mL min<sup>-1</sup>) at 423 K to investigate the acidity of Au catalyst.

 $Au/Nb_2O_5$  before and after the reaction for 12 h were observed using HITACHI H-800 transmission electron microscope operated at 200 kV.

The loading amount of supported Au catalysts was determined using an inductivelycoupled plasma (ICP) spectrometer (Thermo Jarrel Ash IRIS/AP).

Catalyst	Au loading (wt%)	
Au/Nb <sub>2</sub> O <sub>5</sub>	0.56	
Au/ $Nb_2O_5$ after catalytic test for 12 h	0.56	
Au/Al <sub>2</sub> O <sub>3</sub>	0.60	
Au/ZrO <sub>2</sub>	0.70	
Au/TiO <sub>2</sub>	0.77	
Au/SA	0.28	
Au/La <sub>2</sub> O <sub>3</sub>	0.70	
Au/CeO <sub>2</sub>	0.69	
Au/HT	0.60	

**Table S1.** The Au loading of various supported Au catalysts determined by ICP.







# BHF HO OH GC-MS: $m/z = 128 [M]^+$ NMR chart (DMSO-d6, 500 MHz)

![](_page_7_Figure_1.jpeg)

![](_page_9_Figure_0.jpeg)

**Figure S1.** Typical TEM images and histograms of Au/Nb<sub>2</sub>O<sub>5</sub> (a) before and (b) after the reaction for 12 h. The size of gold nanoparticles before and after the reaction was  $2.5 \pm 0.7$  nm and  $2.5 \pm 0.5$  nm, respectively.

![](_page_10_Figure_0.jpeg)

**Figure S2.** (A) EXAFS spectra and (B) XANES spectra of (a) Au/Nb<sub>2</sub>O<sub>5</sub>, (b) Au/Al<sub>2</sub>O<sub>3</sub> (c) Au/ZrO<sub>2</sub>, (d) Au/TiO<sub>2</sub>, and (e) Au/SA.

Catalyst	C.N. <sup>a</sup>	$r / \text{\AA}^{b}$	$\sigma$ / Å $^{ m c}$	$R^{d}$
Au/Nb <sub>2</sub> O <sub>5</sub>	9.0	2.82	0.100	0.35
Au/Al <sub>2</sub> O <sub>3</sub>	11.9	2.84	0.096	0.04
Au/ZrO <sub>2</sub>	11.6	2.83	0.098	0.35
Au/TiO <sub>2</sub>	10.0	2.82	0.100	0.024
Au/SA	8.1	2.80	0.106	0.253

 Table S2. Results of curve-fitting analysis of Au-Au scattering of supported Au catalysts.

*r* fitting range: 2.34-3.18 Å. *k* fitting range: 3.4-14 Å<sup>-1</sup>. *a* Coordination number. *b* Interatomic distance. *c* Debye-Waller factor. *d* So called R factor:  $R = \Sigma \{k^3 \chi_{obs}(k) - k^3 \chi_{cal}(k)\}^2 / \Sigma \{k^3 \chi_{obs}(k)\}^2$ .