

Supporting Information for

P2-type  $\text{Na}_{2/3}\text{Ni}_{1/3}\text{Mn}_{2/3-x}\text{Ti}_x\text{O}_2$

as a new positive electrode

for higher energy Na-ion batteries

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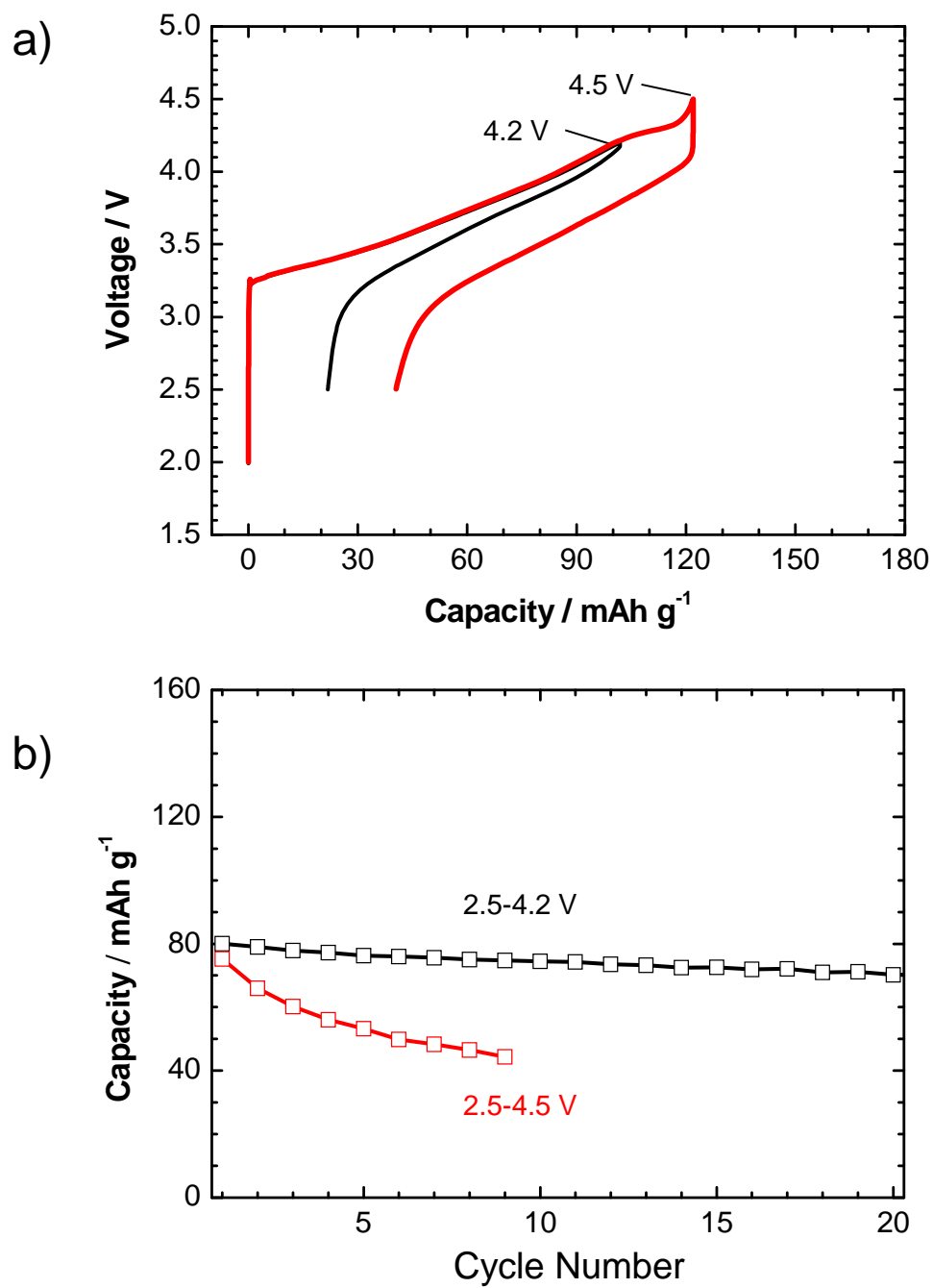


Figure S1. (a) Charge/discharge curves and (b) discharge capacity retention of Na//Na<sub>2/3</sub>Ni<sub>1/3</sub>Ti<sub>2/3</sub>O<sub>2</sub> cells with different cut-off voltage.

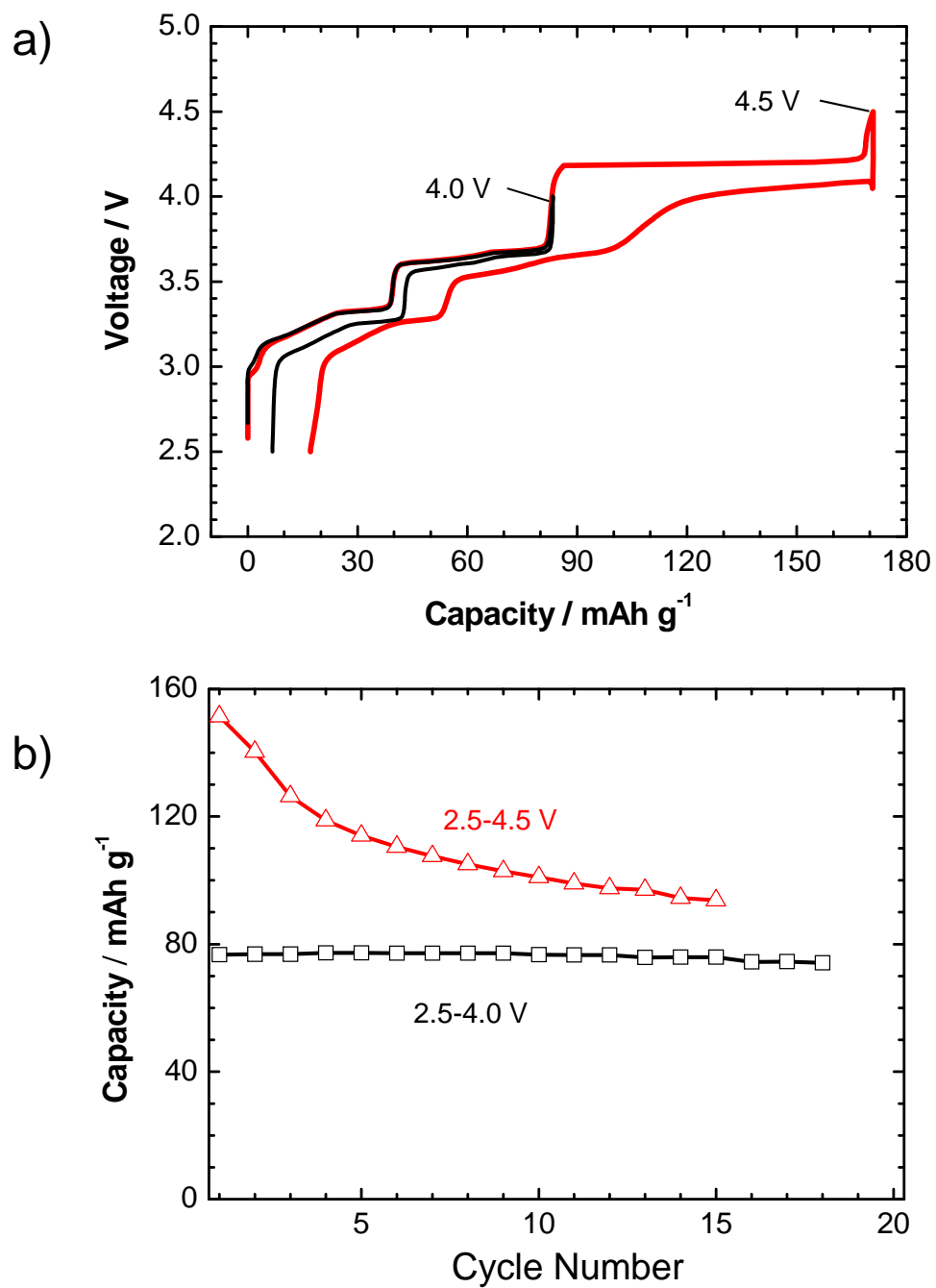


Figure S2. (a) Charge/discharge curves and (b) discharge capacity retention of Na//Na<sub>2/3</sub>Ni<sub>1/3</sub>Mn<sub>2/3</sub>O<sub>2</sub> cells with different cut-off voltage.

Table S1. Comparison of the electrochemical properties and the ratio of lattice volume changes of samples.

	Energy Density <sup>a</sup> /Wh kg <sup>-1</sup>	Capacity retention <sup>b</sup> /%	d-space(fully charged) <sup>c</sup> /Å	Cell volume change <sup>d</sup> /%
Na <sub>2/3</sub> Ni <sub>1/3</sub> Mn <sub>2/3</sub> O <sub>2</sub>	556	66.7	4.43	-23.2
Na <sub>2/3</sub> Ni <sub>1/3</sub> Mn <sub>1/2</sub> Ti <sub>1/6</sub> O <sub>2</sub>	470	93.9	5.03	-12.7
Na <sub>2/3</sub> Ni <sub>1/3</sub> Mn <sub>1/3</sub> Ti <sub>1/3</sub> O <sub>2</sub>	437	86.7	5.10	-12.1

<sup>a</sup> Calculated from Figure 2.

<sup>b</sup> Capacity retention after 10 cycle.

<sup>c</sup> Inter-layer distance after fully charging (to 4.5 V) was calculated from ex-situ XRD results.

<sup>d</sup> Cell volume change is estimated by difference of lattice volume of fully-charged and pristine.