Supporting Information

Experimental procedures

Spiro-Fused Six-Membered N-Heterocyclic Carbene: New Scaffold toward Unique Properties and Activities

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Experimental procedures:

All experiments except for preparation of compound **3** and **1 HBF**₄ were performed in MBRAUN glove box (H₂O < 0.5 ppm, O₂ < 0.5 ppm). Solvents were dried by standard methods and distilled under argon. ¹H, ¹³C and ¹⁹F NMR spectra were recorded on Bruker AX-400 MHz or Varian Mercury-600 MHz instruments and spectral data were reported in ppm. NMR multiplicities are abbreviated as follows: s = singlet, d = doublet, t = triplet, sept = septet, m = multiplet, br = broad signal. IR spectra were recorded on a Nicolet FT-170SX spectrometer. Mass spectra (MS) were measured on spectrometer by direct inlet at 70 eV and signals were given in m/z with relative intensity (%) in brackets. High-resolution mass spectral analysis (HRMS) data were measured on the Bruker ApexII by means of the ESI technique.

Tetrafluoroborate salt **7** $\mathbf{HBF_4}^1$ was prepared according to literature procedures, other reagents were purchased from commercial sources.



Preparation of *cis,cis*-diamine compound 3 and tetrafluoroborate salt 1 HBF4:²

Aniline (2 mL, 22 mmol) was added to a solution of spiro[4,4]nonane-1,6-dione 2 (1.5200 g, 10 mmol) in dry toluene (100 mL) under argon at room temperature, then TiCl₄ (2M in dichloromethane, 20 mL, 40 mmol) was added dropwise to the solution. The reaction mixture was stirred at room temperature for 6 hours, then triethylamine (42 ml, 300 mmol) was added slowly. The reaction mixture was stirred for 10 min then cooled to 0 $^{\circ}$ C, and saturated aqueous solution of Na₂CO₃ (20 ml) was added. The organic layer was separated and aqueous layer was extracted with dichloromethane (3 × 50 mL). The combined organic layer was washed with brine (100 ml), dried over Na₂SO₄ and concentrated under vacuum to get residue. CeCl₃·7H₂O was added to a solution of the residue in methanol (15 mL) under argon at -78 °C. Ten min later, NaBH₄ (0.8360 g, 22 mmol) was added. The reaction mixture was stirred at -78 °C for 10 min and then water (1 mL) was added. The mixture was allowed to warm to room temperature and concentrated under vacuum. The residue was dissolved in water (15 mL) and then extracted with ethyl ethanoate (3 \times 30 mL). The combined organic layer was washed with brine (30 mL), dried over Na₂SO₄ and concentrated under vacuum. Purification of the residue by column chromatography on silica gel (eluting with petroleum ether/ethyl acetate 100:1) provided cis, cis-N¹,N⁶-diphenylspiro[4.4]nonane-1,6-diamine 3 (cis, cis-3 : cis, trans-3 = 5.3 : 1 determined by ¹H NMR) as yellow oil (1.4669 g, two steps, 48% yield). 3 (1.4669 g, 4.8 mmol) was dissolved in triethoxymethane (10 ml) and ammonium fluoroborate (2.0160 g, 19.2 mmol). Then catalytic amount of formic acid was added under argon at room temperature and the reaction system was refluxed for 12 hours. The reaction mixture was cooled to room temperature and concentrated under vacuum. Purification of the residue by column chromatography on silica gel (eluting with dichloromethane /methanol 50:1-30:1) provided tetrafluoroborate salt 1 HBF₄ (1.8431 g, 95% yield).

Cis,cis-diamine compound 3:

¹**H** NMR (400 MHz, CDCl₃, ppm): δ 7.20 (dd, *J* = 8.4, 7.2 Hz, 4H), 6.75 (t, *J* = 7.2 Hz, 2H), 6.65 (d, *J* = 7.6 Hz, 4H), 4.26 (s, 2H), 3.74 (dd, *J* = 4.8, 2.0 Hz, 2H), 2.09-1.98 (m, 2H), 1.90-1.72 (m, 8H), 1.66-1.63 (m, 2H); ¹³**C** NMR (100 MHz, CDCl₃, ppm): δ 147.6 (C), 129.1 (CH), 117.4 (CH), 113.8 (CH), 60.5 (CH), 56.5 (C), 35.7 (CH₂),

31.2 (CH₂), 20.2 (CH₂); **MS** (EI) m/z (%) 306 (M⁺, 23) , 214 (57), 132 (70), 119 (70), 93 (100), 77 (31), 65 (11); **IR** v (cm⁻¹): 3386, 2927, 2868, 1601, 1501, 1265, 739, 703; **HRMS** (ESI) calcd for C₂₂H₂₄BF₃N₂ [M+H]⁺: 307.2169, found 307.2165, Error 1.3 ppm.

1 HBF₄:

m.p. 167-168 °C; ¹**H NMR** (400 MHz, CDCl₃, ppm): δ 7.88 (s, 1H), 7.52-7.49 (m, 4H), 7.45-7.36 (m, 6H), 4.13 (t, *J* = 5.6 Hz, 2H), 2.13-1.99 (m, 4H), 1.98-1.68 (m, 8H); ¹³**C NMR** (100 MHz, CDCl₃, ppm): δ 151.5 (CH), 140.0 (C), 130.2 (CH), 129.5 (CH), 125.0 (CH), 62.9 (CH), 50.1 (C), 35.5 (CH₂), 29.7 (CH₂), 19.9 (CH₂); **MS** (EI) *m/z* (%) 317 ([M-BF₄]⁺, 57), 316 (100), 224 (46), 194 (36), 119 (39), 93 (43), 77 (88), 51 (14); **IR** *v* (cm⁻¹): 2961, 2880, 1642, 1591, 1496, 1064, 765, 699; **HRMS** (ESI) calcd for C₂₂H₂₅BF₄N₂ [M-BF₄]⁺:317.2012, found 317.2008, Error 1.3 ppm.

NHC 1:



The synthetic reaction was performed in a glove box. To a reaction flask containing LiHMDS (0.0200g, 0.12 mmol) at room temperature, was added a solution of tetrafluoroborate salt **1 HBF**₄(0.0404 g, 0.1 mmol) in benzene (1.5 mL). The resulting mixture was stirred for 50 min, after which the mixture was filtered off by a PTFF filter. The filtrate was condensed under reduced pressure to afford NHC **1** as a white solid (0.0311 g, 98%). Colorless crystals of **1** suitable for X-ray structure determination were obtained via evaporation of a petroleum ether/toluene (1:2) solution at -40 \degree C.

¹**H** NMR (600 MHz, C₆D₆, ppm): δ 7.71 (dd, J = 8.4, 1.2 Hz, 4H), 7.18-7.15 (m, 4H), 6.96-6.94 (m, 2H), 3.36 (t, J = 6.0 Hz, 2H), 1.69-1.63 (m, 2H), 1.42-1.37 (m, 4H), 1.36-1.28 (m, 2H), 1.23-1.17 (m, 2H), 1.08-1.03 (m, 2H); ¹³C NMR (150 MHz, C₆D₆, ppm): δ 244.4 (NCN), 149.6 (C), 128.6 (CH), 123.9 (CH), 122.0 (CH), 56.6 (CH), 50.6 (C), 36.1 (CH₂), 31.2 (CH₂), 21.0 (CH₂).

RhCl(COD)(NHC) 4:



In glove box, tetrafluoroborate salt **1 HBF**₄ (0.0404 g, 0.1 mmol) was dissolved in benzene (2 mL) then LiHMDS (0.0200 g, 0.12 mmol) was added to the solution. The reaction mixture was stirred at room temperature for 50 min, and then chloro(1,5-cyclooctadiene)rhodium(I) dimer (0.0247 g, 0.05 mmol) was added. Three hours later, the reaction system was removed from the glove box and purified by column chromatography on neutral alumina (eluting with petroleum ether/ethyl acetate 20:1-10:1) provided RhCl(COD)(NHC) **4** as yellow powder (0.0440 g, 74% yield).

m.p. 199-200 °C (dec); ¹**H NMR** (400 MHz, CDCl₃, ppm): δ 8.00 (d, *J* = 7.6 Hz, 4H), 7.46 (dd, *J* =15.2, 7.2 Hz, 2H), 7.37-7.32(m, 2H), 4.52-4.43 (m, 2H), 3.62-3.58 (m, 2H), 2.55-2.52 (m, 1H), 2.47-2.43 (m, 1H), 2.40-2.33 (m, 1H), 2.40-2.33 (m, 2H), 4.52-4.43 (m, 2H), 3.62-3.58 (m, 2H), 3.

1H), 2.12-1.99 (m, 2H), 1.95-1.84 (m, 1H), 1.83-1.65 (m, 5H), 1.64-1.47 (m, 5H), 1.33-1.17 (m, 5H), 1.15-1.09 (m, 1H); ¹³**C NMR** (100 MHz, CDCl₃, ppm): 209.0 (d, NCN, $J_{CRh} = 46.0$ Hz), 147.1 (C), 145.3 (CH), 129.4 (CH), 128.3 (CH), 128.2 (CH), 127.8 (CH), 126.9 (CH), 126.6 (CH), 95.5 (d, CH, $J_{CRh} = 7.0$ Hz), 95.2 (d, CH, $J_{CRh} = 7.0$ Hz), 67.8 (d, CH, $J_{CRh} = 16.0$ Hz), 67.2 (d, CH, $J_{CRh} = 16.0$ Hz), 64.1 (CH), 51.8 (C), 37.4 (CH₂), 34.2 (CH₂), 32.6 (CH₂), 31.4 (CH₂), 31.3 (CH₂), 31.1 (CH₂), 27.6 (CH₂), 27.5 (CH₂), 21.8 (CH₂), 20.0 (CH₂); **MS** (EI) m/z (%) 562 (M+, <1%), 419 (22), 418(100), 317(19), 132(7), 77(17), 67(19), 54(19); **IR** ν (cm⁻¹): 2935, 2869, 1594, 1468, 1278, 759, 703; **HRMS** (ESI) calcd for C₃₀H₃₆ClN₂Rh [M-Cl]⁺: 527.1928, found 527.1952, Error 4.6 ppm.

RhCl(CO)₂(NHC) 5:



Compound 4 (0.0208 g, 0.037 mmol) was dissolved in dichloromethane (2 mL), and carbon monoxide was bubbled through the solution for 1.5 hours at room temperature. The reaction mixture was concentrated and purified by column chromatography on neutral alumina (eluting with petroleum ether/ethyl acetate 20:1-10:1) provided light yellow powder **5** (0.0180 g, 95% yield).

¹**H** NMR (400 MHz, CDCl₃, ppm): δ 7.53-7.51 (m, 2H), 7.45-7.41 (m, 4H), 7.38-7.33 (m, 4H), 3.72-3.68 (m, 2H), 2.18-2.08 (m, 2H), 1.96-1.90 (m, 3H), 1.87-1.79 (m, 2H), 1.77-1.69 (m, 5H); ¹³C NMR (150 MHz, CDCl₃, ppm): 189.9 (d, NCN, $J_{CRh} = 38.1$ Hz), 186.6 (d, CO, $J_{CRh} = 54.0$ Hz), 183.3 (d, CO, $J_{CRh} = 78.0$ Hz), 145.0 (C), 144.5 (C), 129.3 (CH), 129.2 (CH), 129.1 (CH), 128.9 (CH), 128.0 (CH), 127.8 (CH), 64.4 (CH), 63.7 (CH), 50.2 (C), 36.3 (CH₂), 35.1 (CH₂), 31.3 (CH₂), 30.4 (CH₂), 20.3 (CH₂), 20.0 (CH₂); MS (EI) *m/z* (%) 510 (M+, <1%), 419 (8), 418 (30), 366 (48), 317 (100), 194 (10), 93 (66), 77 (90), 67 (53), 55 (41); IR ν (cm⁻¹): 2070.76 (CO I), 1990.55 (CO II) (DCM) or 2059.1 (CO I), 1983.2(CO II) (KBr disk); HRMS (ESI) calcd for C₂₄H₂₄ClN₂O₂Rh [M-Cl]+: 475.0887, found 475.0900, Error 2.7 ppm.

NHC-BF₃ 6:



In glove box, a solution of tetrafluoroborate salt **1 HBF**₄ (0.0404 g, 0.1 mmol) in benzene/tetrahydrofuran (20/1) mixture (1.5 mL) was added to LiHMDS (0.0200 g, 0.12 mmol) at room temperature. The reaction mixture was stirred for 12 hours, then the reaction system was removed from the glove box and purified by column chromatography on neutral alumina (eluting with petroleum ether/ethyl acetate c 20:1-10:1) provided **6** as white solid (0.0365 g, 95% yield).

m.p. 171-172 °C; ¹**H NMR** (400 MHz, CDCl₃, ppm): δ 7.44-7.32 (m, 10H), 3.60 (t, *J* = 6.4 Hz, 2H), 2.15-2.08 (m, 2H), 1.95-1.81 (m, 6H), 1.77-1.69 (m, 4H); ¹³C NMR (150 MHz, CDCl₃, ppm): 176.2 (C), 144.0 (C), 130.1 (CH), 128.9 (CH), 128.1 (CH), 127.0 (CH), 65.5 (CH), 50.4 (C), 35.9 (CH₂), 31.1 (CH₂), 20.4 (CH₂); ¹⁹F NMR (376

MHz, CDCl₃, ppm): δ 133.33 (q, 3F, J_{BF} = 36.8 Hz); **MS** (EI) m/z (%) 384 (M+, <1%), 315 (100), 224 (26), 194 (28), 118 (25), 77 (84), 51 (14); **IR** ν (cm⁻¹): 2930, 2869, 1596, 1515, 1070, 945, 767, 701; **HRMS** (ESI) calcd for C₂₂H₂₄BF₃N₂ [M+Na]⁺: 406.1913, found 406.1917, Error 1.0 ppm.

NHC-BF₃8:



In glove box, a solution of tetrafluoroborate salt **7 HBF**₄ (0.0204 g, 0.1 mmol) in benzene/tetrahydrofuran (20/1) mixture (1.5 mL) was added to LiHMDS (0.0200 g, 0.12 mmol) at room temperature. The reaction mixture was stirred for 12 hours, and then removed from the glove box and purified by column chromatography on neutral alumina (eluting with dichloromethane/methanol 100:1) provided **8** as white solid (0.0191g, 99% yield).

m.p. 251-253 °C; ¹**H NMR** (400 MHz, CD₂Cl₂, ppm): δ 6.92 (s, 4H), 3.46 (t, J = 5.6 Hz, 4H), 2.36-2.31 (m, 2H), 2.29 (s, 6H), 2.28 (s, 12H); ¹³C **NMR** (150 MHz, CD₂Cl₂, ppm): δ 177.9 (C), 141.3 (C), 138.0 (C), 134.6 (C) 129.3 (CH), 48.7 (CH₂), 21.1 (CH₃), 21.0 (CH₂), 17.8 (CH₃); ¹⁹F **NMR** (376 MHz, CD₂Cl₂, ppm): δ -138.28 (q, 3F, $J_{FB} = 31.2$ Hz); **MS** (EI) m/z (%) 388 (M+, <1%), 369 (5), 320 (48), 159 (100), 146 (32), 131 (22), 91 (33), 77 (15), 65 (6); **IR** v (cm⁻¹): 2930, 2865, 1611, 1531, 1109, 1066, 940, 853; **HRMS** (ESI) calcd for C₂₂H₂₄BF₃N₂ [M+NH₄]⁺: 405.2672, found 405.2683, Error 2.7 ppm.

Thiourea 11:



In glove box, tetrafluoroborate salt **1 HBF**₄ (0.0202 g, 0.05 mmol) was dissolved in benzene (1 mL) then LiHMDS (0.0100 g, 0.06 mmol) was added to the solution. The reaction mixture was stirred at room temperature for 50 min, then S_8 (0.0256 g, 0.1 mmol) was added. Twelve hours later, the reaction system was removed from the glove box and purified by column chromatography on neutral alumina (eluting with petroleum ether/ethyl acetate 20:1-10:1) provided thiourea **11** as white solid (0.0167 g, 96 % yield).

m.p. 169-171 °C; ¹**H NMR** (400 MHz, CDCl₃, ppm): δ 7.41-7.37 (m, 4H), 7.30-7.25 (m, 6H), 3.76 (t, J = 6.8 Hz, 2H), 2.21-2.12 (m, 2H), 2.04-1.87 (m, 6H), 1.77-1.67 (m, 4H); ¹³**C NMR** (100 MHz, CDCl₃, ppm): 182.0 (C), 146.4 (C), 129.0 (CH), 127.9 (CH), 126.8 (CH), 68.6 (CH), 54.2 (C), 37.4 (CH₂), 32.5 (CH₂), 22.2 (CH₂); **MS** (EI) m/z (%) 348 (M⁺, 62), 347 (100), 227 (11), 93 (16), 77 (26), 67 (7), 51 (8); **IR** v (cm⁻¹): 2925, 2854, 1593, 1493, 1267, 752, 692; **HRMS** (ESI) calcd for C₂₂H₂₄N₂S [M+H]⁺: 349.1733, found 349.1730, Error 0.9 ppm.

Compound 12:



In glove box, tetrafluoroborate salt **1 HBF**₄ (0.0202 g, 0.05 mmol) was dissolved in benzene (1 mL) then LiHMDS (0.0100 g, 0.06 mmol) was added to the solution. The reaction mixture was stirred at room temperature for 50 min, then CS_2 (0.003 ml, 0.05 mmol) was added. Twelve hours later, the reaction system was removed from the glove box and purified by column chromatography on neutral alumina (eluting with petroleum ether/EtOAc 20:1-10:1) provided **12** as red solid (0.0173 g, 88% yield).

¹**H** NMR (400 MHz, CDCl₃, ppm): δ 7.50 (d, J = 6.4 Hz, 4H), 7.33-7.28 (m, 6H), 3.82 (t, J = 5.6 Hz, 2H), 2.17-2.10 (m, 4H), 2.09-1.93 (m, 4H), 1.86-1.75 (m, 4H); ¹³**C** NMR (100 MHz, CDCl₃, ppm): 229.0 (NCN),163.9 (SCS), 140.6 (C), 129.2 (CH), 128.7 (CH), 128.6 (CH), 65.7 (CH), 50.5 (C), 35.9 (CH₂), 30.4 (CH₂), 20.2 (CH₂); MS (EI) m/z (%) 392 (M⁺, 40), 347 (22), 315 (100), 224 (26), 93 (45), 77 (99), 67 (19), 51 (15); **IR** ν (cm⁻¹): 2960, 2870, 1548, 1487, 1058, 773, 697; **HRMS** (ESI) calcd for C₂₃H₂₄N₂S₂ [M+H]⁺: 393.1454, found 393.1451, Error 0.8 ppm.

Compound 13:



In glove box, tetrafluoroborate salt **1 HBF**₄ (0.0202 g, 0.05 mmol) was dissolved in benzene (1 mL) then LiHMDS (0.0100 g, 0.06 mmol) was added to the solution. The reaction mixture was stirred at room temperature for 50 min, then 3,4-dimethoxycyclobut-3-ene-1,2-dione (0.0118 g, 0.06 mmol) was added. Twelve hours later, the reaction system was removed from the glove box and purified by column chromatography on silica gel (eluting with petro-leum ether/ethyl acetate 20:1-15:1) provided **13** as rufous solid (0.0132 g, 58% yield).

¹**H** NMR (400 MHz, CDCl₃, ppm): δ 7.20-7.16 (m, 4H), 7.11-7.05 (m, 4H), 6.97-6.95 (m, 2H), 4.37 (dd, *J* = 6.0, 3.6 Hz, 1H), 4.15 (dd, *J* = 5.2, 2.0 Hz, 1H), 4.09 (s, 3H), 3.28 (s, 3H), 2.20-2.13 (m, 1H), 1.82-1.72 (m, 1H), 1.71-1.57 (m, 4H), 1.56-1.41 (m, 5H), 1.34-1.26 (m, 1H); ¹³**C** NMR (100 MHz, CDCl₃, ppm): δ 204.0 (C), 180.4 (C), 170.1 (C), 149.2 (C), 145.6 (C), 144.7 (C),129.6 (CH), 128.8 (CH), 128.5 (CH), 128.3 (CH), 126.4 (CH), 125.4 (CH), 81.2 (C), 64.2 (CH), 63.5 (CH), 60.1 (CH₃), 59.1 (CH₃), 57.1 (C), 38.3 (CH₂), 37.6 (CH₂), 32.4 (CH₂), 31.2 (CH₂), 23.0 (CH₂), 22.3 (CH₂); **MS** (EI) *m/z* (%) 458 (M⁺, 4), 401 (100), 387 (61), 371 (37), 328 (11), 77 (39), 67 (9), 55 (6); **IR** *v* (cm⁻¹): 2953, 2870, 1707, 1593, 1492, 737, 704; **HRMS** (ESI) calcd for C₂₈H₃₀N₂O₄ [M+H]⁺: 459.2278, found 459.2271, Error 1.5 ppm.

Au(NHC)Cl 14:



In glove box, tetrafluoroborate salt **1 HBF**₄ (0.0607 g, 0.15 mmol) was dissolved in benzene (2 mL) then LiHMDS (0.0301 g,0.18 mmol) was added to the solution. The reaction mixture was stirred at room temperature for 50 min, and then Au(Me₂S)Cl (0.0531 mg, 0.18 mmol) was added. Twelve hours later, the reaction system was removed from the glove box and purified by column chromatography on neutral alumina (eluting with petroleum ether/ethyl acetate 20:1-10:1) provided Au(NHC)Cl **14** as white powder (0.0821 g , 99%).

m.p. 204-206 °C; ¹**H NMR** (400 MHz, CDCl₃, ppm): δ 7.42-7.39 (m, 4H), 7.36-7.32(m, 6H), 3.72 (t, *J* = 6.0 Hz, 2H), 2.08,-2.01 (m, 2H), 2.00-1.89 (m, 2H), 1.88-1.80 (m, 2H), 1.78-1.69 (m, 6H); ¹³**C NMR** (100 MHz, CDCl₃, ppm): δ 189.4 (NCN), 146.2 (C), 129.6 (CH), 128.2 (CH), 127.6 (CH), 63.6 (CH), 49.9 (C), 35.8 (CH₂), 30.9 (CH₂), 20.2 (CH₂); **MS** (EI) *m/z* (%) 548 (M+, 100), 513 (22), 315 (42), 194 (12), 93 (17), 77 (46), 67 (9), 51 (8); **IR** *v* (cm⁻¹): 2952, 2871, 1595, 1494, 1275, 767, 699; **HRMS** (ESI) calcd for C₂₂H₂₄AuClN₂ [2M-Cl]⁺: 1061.2893, found 1061.2957, Error 6.0 ppm.

Cu(NHC)₂ CuCl₂ 15:



In glove box, tetrafluoroborate salt **1 HBF**₄ (0.0404 g, 0.1 mmol) and copper (I) chloride (0.0106 g, 0.105 mmol) were dissolved in benzene (1.5 ml) then LiHMDS (0.0200 g, 0.12 mmol) was added to the solution. The reaction mixture was stirred at room temperature for 12 hours, then the mixture was filtered off by PTFF filter. The filtrate was evaporated under reduced pressure to get crude product **15**. **15** was grown from a toluene/tetrahydrofuran (20:1) solution at -40 $^{\circ}$ C as colorless crystals (0.0294 g, 71 % yield).

m.p. 99-101 °C; ¹**H NMR** (600 MHz, C₆D₆, ppm): δ 7.20-7.15 (m, 16H), 7.07-7.04 (m, 4H), 3.12 (t, J = 4.0 Hz, 4H), 1.47-1.41 (m, 4H), 1.41-1.36 (m, 8H), 1.36-1.27 (m, 4H), 1.26-1.16 (m, 4H), 1.12-1.09 (m, 4H); ¹³**C NMR** (150 MHz, C₆D₆, ppm): 198.5 (C), 146.8 (C), 129.7 (CH), 128.3 (CH), 126.2 (CH), 61.4 (CH), 49.7 (C), 35.5 (CH₂), 30.7 (CH₂), 20.2 (CH₂); **MS** (EI) m/z (%) 695 ([M-CuCl₂]⁺, <1%), 414 (88), 379 (15), 315 (100), 224 (32), 195 (24), 132 (12), 93 (31), 77 (60), 67 (11), 51 (11); **IR** v (cm⁻¹): 2959, 2866, 1593, 1490, 1351, 1275, 766, 702; **HRMS** (ESI) calcd for C₄₄H₄₈N₄Cu₂Cl₂ [M-CuCl₂]⁺: 695.3170, found 695.3185 Error 2.2 ppm.

Reference

- 1 M. Mayr, K. Wurst, K.-H, Ongania, M. R. Buchmeiser, Chem. Eur. J. 2004, 10, 1256.
- 2 a) A. R. Katritzky, A. Denisenko, S. N. Denisenko, M. Arend, J. Heterocycl. Chem. 2000, 37, 1309; b) S. Saba,
 A. Brescia, M. K. Kaloustian, Tetrahedron Lett. 1991, 32, 5031.

2. X-Ray ellipsoid plots of 4, 11, 14, and 15



X-ray structures of **4** with thermal ellipsoids was drawn at 30% probability level. The crystal structure has been deposited at the Cambridge Crystallographic Data Centre and allocated the deposition number: 973602.



X-ray structures of **11** with thermal ellipsoids was drawn at 30% probability level. The crystal structure has been deposited at the Cambridge Crystallographic Data Centre and allocated the deposition number: 973596.



X-ray structures of **14** with thermal ellipsoids was drawn at 30% probability level. The crystal structure has been deposited at the Cambridge Crystallographic Data Centre and allocated the deposition number: 973598.



X-ray structures of **15** with thermal ellipsoids was drawn at 30% probability level. The crystal structure has been deposited at the Cambridge Crystallographic Data Centre and allocated the deposition number: 973600.

	1	2'
C3-C4	1.528	1.520
C3-C2	1.530	1.520
C4-C16	1.544	1.514
C2-C11	1.533	1.514
C16-C15	1.526	1.521
C11-C12	1.539	1.521
C15-C14	1.504	1.486
C13-C12	1.546	1.486
C14-C3	1.532	1.524
C13-C3	1.534	1.524
C2-C3-C4	110.65	117.55
C2-C3-C13	100.28	103.31
C4-C3-C14	105.46	103.31
C3-C4-N(O)2	110.10	105.23
C16-C4-N(O)2	111.05	108.94
C3-C2-N(O)1	109.08	105.23
C11-C2-N(O)1	115.20	108.94

Table 1. Selected Bond Lengths (Å) and Angles (deg) of NHC 1 and Bis-(*p*-nitrobenaoate) of Diol 2'.

3. Selected Bond Lengths and Angles of NHC 1 , Bis-(p-nitrobenaoate) of Diol 2' and 6-Mes

Table 2. Selected Angles (deg) of NHC ${\bf 1}$ and ${\bf 6}\text{-}{\bf Mes}.$

6-Mes

	1	6-Mes	
C2-C3-C4	110.65	108.39	00
C3-C4-N2	110.10	108.88	à
C3-C2-N1	110.20	108.88	
C4-N2-C1	127.78	126.27	·
C2-N1-C1	124.61	126.27	X Y A
N1-C1-N2	115.56	114.65	































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S21



















Ph'

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¹H NMR



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Ph N Ph AuCl 14 1 ³ C NMR

S33











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