# **Supporting Information**

# Chiral BINOL-Bridged Imidazole Dimer Possessing Sub-Millisecond Fast Photochromism

## Takahiro Iwasaki<sup>a</sup>, Tetsuya Kato<sup>a</sup>, Yoichi Kobayashi<sup>a</sup> and Jiro Abe<sup>\*ab</sup>

<sup>a</sup>Department of Chemistry, School of Science and Engineering, Aoyama Gakuin University, 5-10-1 Fuchinobe, Chuo-ku, Sagamihara, Kanagawa 252-5258, Japan

<sup>b</sup>CREST, Japan Science and Technology Agency (JST), K's Gobancho, 7 Gobancho, Chiyoda-ku, Tokyo 102-0076, Japan.

E-mail: jiro\_abe@chem.aoyama.ac.jp

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## 1. Synthesis

All reactions were monitored by thin-layer chromatography carried out on 0.2 mm E. Merck silica gel plates (60F-254). Column chromatography was performed on the silica gel (Silica Gel 60N (spherical, neutral), 40-50  $\mu$ m, Kanto Chemical Co., Inc.). <sup>1</sup>H-NMR spectra and <sup>13</sup>C-NMR spectra were recorded on a Bruker AVANCE III 400 NanoBay. DMSO-*d*<sub>6</sub>, CDCl<sub>3</sub> and CD<sub>2</sub>Cl<sub>2</sub> were used as deuterated solvent. MASS spectra (ESI-TOF-MS) were measured by using a Bruker micrOTOF II-AGA1. All reagents were purchased from Tokyo Chemical Industry Co., Ltd., Wako Pure Chemical Industries, Ltd., and Aldrich Chemical Company, Inc. and were used without further purification. All reaction solvents were distilled on the appropriate drying reagents prior to use.

## 1.1 Synthesis of 1



Scheme S1. Synthetic procedure of 1.

1,1'-Bi-2-naphthol (2.00 g, 6.99 mmol), 4-fluorobenzaldehyde (1.80 mL, 17.1 mmol) and potassium carbonate (2.42 g, 17.5 mmol) were stirred at 100 °C in DMF (50 mL). After 42 h, the reaction mixture was cooled to room temperature and the target compound was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic phase was washed with water, dried, and evaporated. The residue was purified by column chromatography over silica gel with CHCl<sub>3</sub> as eluent to give a light yellow solid (1.28 g, 2.58 mmol, yield; 37 %). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) :  $\delta$  9.79 (s, 2H), 8.13 (d, *J* = 9.0 Hz, 2H), 8.04 (d, *J* = 8.2 Hz, 2H), 7.69 (d, *J* = 8.6 Hz, 4H), 7.50 (t, *J* = 7.6, 9.0 Hz, 2H), 7.43-7.32 (m, 4H), 7.17 (d, *J* = 8.2 Hz, 2H), 6.92 (d, *J* = 8.6 Hz, 4H). HR-MS (ESI+) calculated for C<sub>34</sub>H<sub>23</sub>O<sub>4</sub> [M+H]<sup>+</sup> 495.1596, found 495.1614.



Figure S1. <sup>1</sup>H NMR spectrum of 3 in DMSO- $d_6$ .

Compound **3** (150 mg, 0.303 mmol), benzil (220 mg, 1.05 mmol), and ammonium acetate (350mg, 4.54 mmol) were stirred at 110 °C in CH<sub>2</sub>Cl<sub>2</sub> (1.0 mL) in a sealed tube. After 14 h, the reaction mixture was cooled to room temperature and washed with water, followed by ethanol. The precipitate was filtered and washed with hexane to give a white powder (194 mg, 0.222 mmol, yield; 73 %). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$ : 12.50 (s, 2H), 8.11 (d, *J* = 8.7 Hz, 2H), 8.04 (d, *J* = 7.8, 2H), 7.94 (d, *J* = 7.8 Hz, 4H), 7.53 – 7.17 (m, 28H), 6.93 (d, *J* = 7.8 Hz, 4H); <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$ : 156.91, 151.59, 145.01, 136.88, 135.15, 133.48, 131.04, 130.21, 128.47, 128.25, 128.02, 127.86, 127.54, 126.97, 126.66, 126.34, 125.35, 125.06, 124.98, 121.65, 119.35, 118.15; HR-MS (ESI+) calculated for C<sub>62</sub>H<sub>43</sub>N<sub>4</sub>O<sub>2</sub> [M+H]<sup>+</sup> 875.3381, found 875.3297.



Figure S3. <sup>13</sup>C NMR spectrum of 4 in DMSO-*d*<sub>6</sub>.

All manipulations were carried out with the exclusion of light. Under nitrogen, a solution of potassium ferricyanide (11.8 g, 35.8 mmol) and potassium hydroxide (4.05 g, 72.2 mmol) in water (200 mL) was added to a solution of compound **4** (629 mg, 0.718 mmol) in benzene (200 mL). The reaction mixture was vigorously stirred at room temperature for 2 h. The organic layer was washed with water, dried, and evaporated. The residue was purified by column chromatography over silica gel with AcOEt/hexane = 1/3 as eluent and followed by recrystallization from CHCl<sub>3</sub>/hexane to give a slightly green crystal of **3** (129 mg, 0.148 mmol, yield; 21 %). <sup>1</sup>H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>)  $\delta$ : 8.13 (d, *J* = 8.9 Hz, 1H), 8.06-7.95 (m, 3H), 7.66-7.02 (m, 32H), 6.93 (d, *J* = 8.8 Hz, 1H), 6.75 (d, *J* = 8.2 Hz, 1H), 6.61 (d, *J* = 8.7 Hz, 1H), 6.41 (d, *J* = 8.4 Hz, 1H), 6.06 (d, *J* = 8.5 Hz, 1H), 5.94 (d, *J* = 8.8 Hz, 1H); <sup>13</sup>C NMR (100 MHz, CD<sub>2</sub>Cl<sub>2</sub>)  $\delta$ : 166.82, 166.34, 160.09, 158.96, 150.23, 149.92, 148.39, 138.14, 135.26, 135.23, 135.07, 134.75, 133.50, 133.39, 132.37, 132.34, 132.12, 131.89, 131.85, 131.36, 131.32, 131.27, 131.16, 130.77, 130.38, 130.24, 129.90, 129.83, 129.74, 128.92, 128.82, 128.70, 128.61, 128.46, 128.42, 128.27, 128.19, 128.04, 127.45, 127.33, 127.27, 126.53, 126.20, 126.13, 124.93, 124.34, 123.51, 122.47, 116.59, 115.81, 113.98, 113.17; HR-MS (ESI+) calculated for C<sub>62</sub>H<sub>41</sub>N<sub>4</sub>O<sub>2</sub> [M+H]<sup>+</sup> 873.3224, found 873.3113.



Figure S4. <sup>1</sup>H NMR spectrum of 1 in CD<sub>2</sub>Cl<sub>2</sub>.



Figure S5. <sup>13</sup>C NMR spectrum of 1 in CD<sub>2</sub>Cl<sub>2</sub> (\*Solvent Peaks).

# 1.2 Synthesis of 2 (racemate)

Scheme S2. Synthetic procedure of 2.



(*R*)-1,1'-Bi-2-naphthol (2.00 g, 6.99 mmol), 2-chloro-4-fluorobenzaldehyde (2.66 g, 2.85 mmol) and potassium carbonate (2.51 g, 5.85 mmol) were stirred at 100 °C in DMF (50 mL). After 15 h, the reaction mixture was cooled to room temperature and the target compound was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic phase was washed with water. The organic phase was dried over anhydrous sodium sulfate, filtered, and evaporated to give a yellow solid. The yellow solid was purified by column chromatography over silica gel with CH<sub>2</sub>Cl<sub>2</sub> as eluent to give a white amorphous solid (2.86 g, yield; 72 %). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.19 (d, *J* = 9.2 Hz, 2H), 8.09 (d, *J* = 8.6 Hz, 2H), 7.61 (d, *J* = 8.6 Hz, 2H), 7.54 (dd, *J* = 7.3, 7.9 Hz, 2H), 7.48 (d, *J* = 9.2 Hz, 2H), 7.41 (dd, *J* = 8.6, 9.2 Hz, 2H), 7.12 (d, *J* = 8.6 Hz, 2H), 6.83 (dd, *J* = 9.2, 2.4 Hz, 2H), 6.77 (d, *J* = 2.4 Hz, 2H); HR-MS (ESI+) calculated for C<sub>34</sub>H<sub>21</sub>Cl<sub>2</sub>O4 [M+H]<sup>+</sup> 563.0811, found 563.0819.



**Figure S6.** <sup>1</sup>H NMR spectrum of **5** in DMSO-*d*<sub>6</sub> (\*Solvent Peaks).

Compound **5** (287 mg, 0.510 mmol), benzil (324 mg, 1.54 mmol) and ammonium acetate (599 mg, 7.77 mmol) were stirred at 110 °C in CH<sub>2</sub>Cl<sub>2</sub> (1.7 mL) in a sealed tube. After 2 days, the reaction mixture was cooled to room temperature and washed with water, followed by ethanol. The precipitate was filtered and washed with hexane to give a white powder (374 mg, 0.267 mmol, yield; 77.8 %).<sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  12.5 (s, 2H), 8.18 (d, *J* = 9.2 Hz, 2H), 8.08 (d, *J* = 7.9 Hz, 2H), 7.69 (d, *J* = 8.6 Hz, 2H), 7.54–7.19 (m, 28H), 7.05 (d, *J* = 2.4 Hz, 2H), 6.98 (dd, *J* = 9.2, 2.4 Hz, 2H); HR-MS (ESI+) calculated for C<sub>62</sub>H<sub>41</sub>Cl<sub>2</sub>N<sub>4</sub>O<sub>2</sub> [M+H]<sup>+</sup> 943.2601, found 943.2634.



Figure S7. <sup>1</sup>H NMR spectrum of 6 in DMSO-*d*<sub>6</sub> (\*Solvent Peaks).

All manipulations were carried out with the exclusion of light. Under nitrogen, a solution of potassium ferricyanide (5.91 g, 18.0 mmol) and potassium hydroxide (2.05 g, 36.6 mmol) in water (125 mL) was added to a solution of compound **6** (344 mg, 0.363 mmol) in benzene (125 mL). The reaction mixture was vigorously stirred at room temperature for 3 h. The organic layer was washed with water, dried, and evaporated. The residue was purified by column chromatography over silica gel with AcOEt/hexane = 1/3 as eluent to give a yellow solid (209 mg, yield; 61 %). The obtained compound was the racemate as was shown in Figure S27 because the refluxes at >100 °C racemize the (*R*)-1,1'-bi-2-naphthol moiety. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.30 (d, *J* = 9.2 Hz, 1H), 8.18 (m, 3H), 7.63–6.85 (m, 32H), 6.09 (dd, *J* = 9.2, 2.4 Hz, 2H), 5.97 (dd, *J* = 9.2, 2.4 Hz, 2H); HR-MS (ESI+) calculated for C<sub>62</sub>H<sub>39</sub>Cl<sub>2</sub>N4O<sub>2</sub> [M+H]<sup>+</sup> 941.2445, found 941.2478.



Figure S8. <sup>1</sup>H NMR spectrum of 2 in DMSO-*d*<sub>6</sub> (\*Solvent Peaks).

# 1.3 Synthesis of (S)-2

Scheme S3. Synthetic procedure of (*S*)-2 (the synthesis of (*R*)-2 was the identical to that of (*S*)-2 except that (*R*)-1,1'-bi-2-naphthol was used as the starting material).



(S)-**2** 

Trityl chloride (3.89 g 13.9 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (45 mL) were added to (*S*)-1,1-bi-2-naphthol (4.00 g, 13.9 mmol), triethylamine (2.32 mL, 16.7 mmol) and CH<sub>2</sub>Cl<sub>2</sub> (45 mL) at 0 °C. The mixture was allowed to warm to room temperature and was stirred for 2 h. The reaction mixture was washed with water. The organic phase was dried over anhydrous sodium sulfate, and the solution was filtered. The evaporation of the solvent afforded a yellow solid and the recrystallization from CH<sub>2</sub>Cl<sub>2</sub>/hexane gives a white crystal of compound (*S*)-7 (5.08 g, yield; 68.7 %). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  9.56 (s, 1H), 7.94 (d, *J* = 8.9 Hz, 1H), 7.89 (dd, *J* = 6.3, 2.4 Hz, 1H), 7.74 (d, *J* = 8.0 Hz, 1H), 7.49 (d, *J* = 9.2 Hz, 1H), 7.43 (d, *J* = 8.9 Hz, 1H), 7.29–7.13 (m, 19H), 7.06 (d, *J* = 8.3 Hz, 1H), 6.86 (dd, *J* = 6.3, 2.4 Hz, 1H), 6.73 (d, *J* = 8.9 Hz, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  152.99, 151.16, 146.90, 143.99, 134.07, 133.82, 131.49, 129.64, 129.30, 129.23, 128.77, 128.39, 128.07, 127.99, 127.98, 127.73, 127.54, 127.31, 127.12, 126.97, 126.34, 125.32, 125.08, 124.40, 124.26, 124.09, 118.64, 117.82, 117.51, 115.89, 90.11; HR-MS (ESI–) calculated for C<sub>39</sub>H<sub>27</sub>O<sub>2</sub> [M–H]<sup>-</sup> 527.2011, found 527.2013.



**Figure S9.** <sup>1</sup>H NMR spectrum of (*S*)-7 in DMSO-*d*<sub>6</sub> (\*Solvent Peak).



Compound (*S*)-7 (2.00 g, 3.78 mmol), 2-chloro-4-fluorobenzaldehyde (780 mg, 4.92 mmol) and potassium carbonate (1.36 g, 9.84 mmol) were stirred at 100 °C in DMF (50 mL). After 23 h, the reaction mixture was cooled to room temperature and the target compound was extracted with AcOEt. The organic phase was washed with water. The organic phase was dried over anhydrous sodium sulfate, filtered, and evaporated to give a yellow oil solid. The yellow oil solid was purified by column chromatography over silica gel with CH<sub>2</sub>Cl<sub>2</sub>/hexane as eluent to give a white amorphous solid (1.54 g, yield; 61.1 %). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  10.08 (s, 1H), 8.28 (d, *J* = 9.0 Hz, 1H), 8.15 (d, *J* = 8.1 Hz, 1H), 7.70 (dd, *J* = 7.0, 1.4 Hz, 1H), 7.65–7.42 (m, 5H), 7.34–7.08 (m, 23H), 6.94 (d, *J* = 2.3 Hz, 1H), 6.89 (dd, *J* = 8.6, 2.2 Hz, 1H), 6.71 (d, *J* = 9.2 Hz, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  188.47, 163.06, 151.93, 150.12, 144.07, 139.28, 134.29, 133.56, 131.58, 130.74, 129.79, 128.42, 128.10, 127.96, 127.88, 127.81, 126.80, 126.77, 126.51, 126.16, 126.05, 125.68, 124.93, 123.61, 120.29, 119.70, 119.32, 118.65, 116.11, 89.64, 53.46; HR-MS (ESI+) calculated for C<sub>46</sub>H<sub>31</sub>ClNaO<sub>3</sub> [M+Na]<sup>+</sup> 689.1859, found 689.1841.



**Figure S11.** <sup>1</sup>H NMR spectrum of (S)-8 in DMSO- $d_6$  (\*Solvent Peak).



TFA (9.70 mL) was added to compound (*S*)-**8** (1.08 g, 1.62 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (200 mL) at 0 °C. The mixture was allowed to warm to room temperature and was stirred for 2 h, quenched with NaHCO<sub>3</sub> aqueous. The organic phase was washed with water, dried over anhydrous sodium sulfate, filtered, and evaporated to give a brown oil solid. The brown oil solid was purified by column chromatography over silica gel with CH<sub>2</sub>Cl<sub>2</sub>/hexane as eluent to give a white amorphous solid (544 mg, yield; 79.2 %). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  10.13 (s, 1H), 9.61 (s, 1H), 8.17 (d, *J* = 8.9 Hz, 1H), 8.09 (d, *J* = 8.2 Hz, 1H), 7.84–7.82 (m, 2H), 7.69 (d, *J* = 8.7 Hz, 1H), 7.55–7.52 (m, 1H), 7.47 (d, *J* = 8.9 Hz, 1H), 7.41–7.37 (m, 1H), 7.32–7.19 (m, 3H), 7.15 (d, *J* = 8.4 Hz, 1H), 6.97–6.91 (m, 3H); <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  188.24, 162.70, 153.04, 150.01, 137.58, 133.60, 133.54, 131.24, 131.18, 130.21, 129.60, 128.27, 127.98, 127.73, 127.49, 126.88, 126.60, 125.55, 125.04, 123.87, 122.53, 120.67, 118.29, 118.11, 116.54, 112.98; HR-MS (ESI–) calculated for C<sub>27</sub>H<sub>16</sub>ClO<sub>3</sub> [M–H]<sup>-</sup> 423.0788, found 423.0790.



**Figure S13.** <sup>1</sup>H NMR spectrum of (*S*)-9 in DMSO-*d*<sub>6</sub> (\*Solvent Peaks).



Figure S14. <sup>13</sup>C NMR spectrum of (S)-9 in DMSO-d<sub>6</sub>.

Compound (*S*)-**9** (807 mg, 1.90 mmol), 2-chloro-4-fluorobenzaldehyde (456 mg, 2.85 mmol) and potassium carbonate (808 mg, 5.85 mmol) were stirred at 100 °C in DMF (50 mL). After 18 h, the reaction mixture was cooled to room temperature and the target compound was extracted with AcOEt. The organic phase was washed with water. The organic phase was dried over anhydrous sodium sulfate, filtered, and evaporated to give a yellow solid. The yellow solid was purified by column chromatography over silica gel with CH<sub>2</sub>Cl<sub>2</sub>/hexane as eluent to give a white amorphous solid (855 mg, yield; 79.9 %). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.19 (d, *J* = 9.2 Hz, 2H), 8.09 (d, *J* = 8.6 Hz, 2H), 7.61 (d, *J* = 8.6 Hz, 2H), 7.54 (dd, *J* = 7.3, 7.9 Hz, 2H), 7.48 (d, *J* = 9.2 Hz, 2H), 7.41 (dd, *J* = 8.6, 9.2 Hz, 2H), 7.12 (d, *J* = 8.6 Hz, 2H), 6.83 (dd, *J* = 9.2, 2.4 Hz, 2H), 6.77 (d, *J* = 2.4 Hz, 2H); <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  188.23, 162.70, 153.03, 150.01, 147.73, 137.58, 133.60, 133.54, 131.24, 131.18, 130.21, 129.59, 128.27, 127.97, 127.71, 127.48, 126.88, 126.60, 126.47, 126.32, 125.59, 125.55, 125.04, 123.87, 122.53, 120.67, 118.29, 118.10, 116.54, 112.98; HR-MS (ESI+) calculated for C<sub>34</sub>H<sub>21</sub>Cl<sub>2</sub>O4 [M+H]<sup>+</sup> 563.0811, found 563.0796.



**Figure S15.** <sup>1</sup>H NMR spectrum of (*S*)-**5** in DMSO-*d*<sub>6</sub> (\*Solvent Peaks).



Figure S16. <sup>13</sup>C NMR spectrum of (S)-5 in DMSO- $d_6$ .

Compound (*S*)-5 (287 mg, 0.510 mmol), benzil (324 mg, 1.54 mmol) and ammonium acetate (599 mg, 7.77 mmol) were stirred at 110 °C in CH<sub>2</sub>Cl<sub>2</sub> (1.7 mL) in a sealed tube. After 2 days, the reaction mixture was cooled to room temperature and washed with water, followed by ethanol. The precipitate was filtered and washed with hexane to give a white powder (374 mg, 0.267 mmol, yield; 77.8 %).<sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  12.5 (s, 2H), 8.18 (d, *J* = 9.2 Hz, 2H), 8.08 (d, *J* = 7.9 Hz, 2H), 7.69 (d, *J* = 8.6 Hz, 2H), 7.54–7.19 (m, 28H), 7.05 (d, *J* = 2.4 Hz, 2H), 6.98 (dd, *J* = 9.2, 2.4 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  158.03, 150.80, 142.92, 137.56, 134.56, 134.12, 131.62, 130.96, 130.76, 130.37, 129.97, 128.90, 128.28, 128.24, 127.81, 127.72, 127.55, 127.21, 127.05, 126.95, 125.79, 125.42, 122.73, 122.28, 119.69, 118.83, 117.41; HR-MS (ESI+) calculated for C<sub>62</sub>H<sub>41</sub>Cl<sub>2</sub>N<sub>4</sub>O<sub>2</sub> [M+H]<sup>+</sup> 943.2601, found 943.2604.



**Figure S17.** <sup>1</sup>H NMR spectrum of (*S*)-6 in DMSO-*d*<sub>6</sub> (\*Solvent Peaks).



Figure S18. <sup>13</sup>C NMR spectrum of (S)-6 in CDCl<sub>3</sub>.

All manipulations were carried out with the exclusion of light. Under nitrogen, a solution of potassium ferricyanide (3.78 g, 11.5 mmol) and potassium hydroxide (1.36 g, 24.2 mmol) in water (60 mL) was added to a solution of compound (*S*)-6 (202 mg, 0.214 mmol) in benzene (60 mL). The reaction mixture was vigorously stirred at room temperature for 3 h. The organic layer was washed with water, dried, and evaporated. The residue was purified by column chromatography over silica gel with AcOEt/hexane = 1/3 as eluent to give a yellow solid (61.0 mg, yield; 30 %). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.30 (d, *J* = 9.2 Hz, 1H), 8.18 (m, 3H), 7.63–6.85 (m, 32H), 6.09 (dd, *J* = 9.2, 2.4 Hz, 2H), 5.97 (dd, *J* = 9.2, 2.4 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  167.35, 165.01, 160.36, 160.15, 149.42, 149.42, 149.05, 144.95, 137.44, 136.34, 135.61, 134.87, 134.80, 134.28, 133.20, 133.05, 132.96, 131.78, 131.62, 131.51, 131.38, 131.03, 130.84, 130.64, 130.51, 130.27, 130.06, 129.94, 129.35, 128.97, 128.48, 128.31, 128.20, 128.16, 128.07, 127.71, 127.31, 127.20, 127.10, 126.88, 126.28, 126.03, 125.99, 125.85, 125.81, 123.82, 122.48, 121.84, 120.21, 116.90, 112.52, 112.19, 111.08; HR-MS (ESI+) calculated for C<sub>62H39</sub>Cl<sub>2</sub>N<sub>4</sub>O<sub>2</sub> [M+H]<sup>+</sup> 941.2450, found 941.2445.



**Figure S19.** <sup>1</sup>H NMR spectrum of (*S*)-**2** in DMSO-*d*<sub>6</sub> (\*Solvent Peaks).



Figure S20. <sup>13</sup>C NMR spectrum of (S)-2 in CDCl<sub>3</sub>. S24

# 2. HR-ESI-TOF-MS-spectra











**Figure S23.** HR-ESI-TOF-MS of (*S*)-2.

#### **3.** HPLC analyses for determination of purity

HPLC analyses using THF/CH<sub>3</sub>CN = 1/15 as an eluent was performed with system composed of a Mightysil RP18 (4.6 mm × 25 cm) column and JASCO PU-2080 Plus pump equipped with a UV-2075 Plus UV/VIS detector. The mobile phase was THF/CH<sub>3</sub>CN = 1/15 with a flow rate of 1.0 mL/min, inject volume; 2.0 µL.



**Figure S24.** HPLC chromatogram of the racemate of **1** (left) and **2** (right), detection wavelength; 254 nm, purity 99%, respectively.



**Figure S25.** HPLC chromatograms of (*R*)-**2**, detection wavelength; 254 (left) and 340 nm (right), purity 99%.



**Figure S26.** HPLC chromatograms of (*S*)-**2**, detection wavelength; 254 (left) and 340 nm (right), purity 99%.

#### 4. Chiral HPLC analyses for determination of enantiomeric excesses (ee)

HPLC analyses using CH<sub>2</sub>Cl<sub>2</sub>/hexane = 1/1 as an eluent was performed with a system composed of a DAICEL CHIRALPAK IC column (4.6 mm × 25 cm) and JASCO PU-2080 Plus pump equipped with a UV-2075 Plus UV/VIS detector.



**Figure S27.** HPLC charts of **2**, (*R*)-**2** and (*S*)-**2**.

#### 5. X-ray crystallographic analysis

The diffraction data of the single crystal of **1** are collected on the Bruker APEX II CCD area detector (Mo $K_{\alpha}$ ,  $\lambda = 0.71073$  nm). During the data collection, the lead glass doors of the diffractometer were covered to exclude the room light. The data refinement was carried out by the Bruker APEX II software package with SHELXT program.<sup>S1)</sup> All non-hydrogen atoms were anisotropically refined.



**Figure S28.** ORTEP representation of the molecular structure of **1** with thermal ellipsoid at 50% probability. The hydrogen atoms and solvent molecules are omitted. Oxygen and nitrogen atoms are highlighted in red and blue, respectively.

Identification code	1			
Empirical formula	C62 H40 N4 O2			
Formula weight	872.98			
Temperature	90 K			
Wavelength	0.71073 Å			
Crystal system	Triclinic			
Space group	P-1			
Unit cell dimensions	$a = 10.8574(15) \text{ Å} \qquad \alpha = 90.116(2)^{\circ}.$			
	$b = 14.598(2) \text{ Å} \qquad \beta = 105.115(2)^{\circ}.$			
	$c = 15.654(2) \text{ Å} \qquad \gamma = 110.614(2)^{\circ}.$			
Volume	2229.8(5) Å3			
Z	2			
Density (calculated)	1.300 Mg/m3			
Absorption coefficient	0.079 mm-1			
F(000)	912			
Crystal size	0.40 x 0.35 x 0.01 mm3			
Theta range for data collection	1.50 to 26.40°.			
Index ranges	-13<=h<=11, -18<=k<=12, -16<=l<=19			
Reflections collected	23502			
Independent reflections	8830 [R(int) = 0.0137]			
Completeness to theta = $26.40^{\circ}$	96.4 %			
Absorption correction	Empirical			
Refinement method	Full-matrix least-squares on F2			
Data / restraints / parameters	8830 / 0 / 613			
Goodness-of-fit on F2	Goodness-of-fit on F21.023			
Final R indices [I>2sigma(I)]	R1 = 0.0565, wR2 = 0.1105			
R indices (all data)	R indices (all data) $R1 = 0.0791$ , $wR2 = 0.1263$			
Largest diff. peak and hole	0.519 and -0.267 e.Å-3			

 Table S1. X-ray crystallographic data of 1.

#### 6. Laser flash photolysis

The laser flash photolysis experiments were performed with a Unisoku TSP-1000 time-resolved spectrophotometer. A 10 Hz Q-switch Nd:YAG laser (Continuum Minilite II) with the third harmonic at 355 nm (ca. 4 mJ per 5 ns pulse) was employed for the excitation light. A halogen lamp (OSRAM HLX64623) was used for a probe light and the transmitted light through the sample was guided into the monochrometer and a detector (Unisoku MD200 and Hamamatsu R2949 photomultiplier tube, respectively) with an optical fiber scope.

#### 7. Eyring plots

To obtain the activation parameters of the thermal back reactions of 1R and 2R, we conducted the temperature dependence of the decay profiles of the thermal back reactions of 1R and 2R. The rate constants for the thermal back reaction for 1R and 2R are tabulated in Table S2 and S3, respectively. These temperature dependences were analyzed by Eyring plots and these plots are shown in Figure S28 for 1R and Figure S29 for 2R, respectively.

<i>T</i> / °C	$k  /  \mathrm{s}^{-1}$
5	4.8
10	6.8
15	9.5
20	13.2
25	18.2
30	24.7
35	33.2
40	44.3

Table S2. First-order rate constants for the thermal back-reaction of 1 in degassed benzene

 $(2.1 \times 10^{-5} \text{ M})$  at different temperature.



Figure S29. The Eyring plot for the thermal back-reaction of colored species of 1 in degassed benzene  $(2.1 \times 10^{-5} \text{ M})$ .

T/°C	$k / s^{-1}$
5	3.49
10	4.19
15	4.98
20	5.86
25	6.68
30	7.84
35	9.11
40	10.2

Table S3. First-order rate constants for the thermal back-reaction of 2 in degassed benzene

 $(2.3 \times 10^{-5} \text{ M})$  at different temperature.



Figure S30. The Eyring plot for the thermal back-reaction of colored species of 2 in degassed benzene  $(2.3 \times 10^{-5} \text{ M})$ .

## 8. CD spectroscopy

Circular dichroism (CD) spectra were recorded on a JASCO J-820 spectropolarimeter. The CD spectra of (*R*)-2 and (*S*)-2 in acetonitrile  $(3.3 \times 10^{-5} \text{ M})$  were measured with a 10-mm quartz cell at room temperature. Photoracemization processes of (*R*)-2 and (*S*)-2 were analyzed using a Q-switch Nd:YAG laser (Continuum Minilite II) with the third harmonic at 355 nm (ca. 4 mJ per 5 ns pulse) as the excitation beam.

#### 9. UV-vis absorption spectroscopy and investigation for fatigue resistance

UV-vis absorption spectra were recorded on a Shimadzu UV-3150 spectrometer. The UV-vis absorption spectra of  $1 (2.1 \times 10^{-5} \text{ M})$  and  $2 (2.3 \times 10^{-5} \text{ M})$  in benzene were measured with a 10-mm quartz cell at 25 °C. The fatigue resistances of 1 and 2 were measured using a Q-switch Nd:YAG laser (Continuum Minilite II) with the third harmonic at 355 nm (ca. 4 mJ per 5 ns pulse) as the excitation beam.



**Figure S31.** UV-vis absorption spectra of 1 ( $2.1 \times 10^{-5}$  M) (left) and 2 ( $2.3 \times 10^{-5}$  M) (right) in benzene before and after laser shots at 25 °C.

## **10. DFT calculations**

All calculations were carried out using the Gaussian 09 program (Revision D.01)<sup>S2)</sup>. The molecular structures were fully optimized at the M062X/6-31G(d) level of the theory for **2** and the UM062X/6-31G(d) level of the theory for **2R**. The analytical second derivatives were computed using the vibrational analysis to confirm each stationary point to be a minimum. The TDDFT calculations were performed at the MPW1PW91/6-31+G(d) level of the theory for **2** and UMPW1PW91/6-31+G(d) level of the theory for **2R** for the optimized structures.

Center	Atomic	Atomic	Coordinates (Angstroms)
Number	Number	Туре	X Y Z
1	17	0	0.628470 0.813532 3.508147
2	17	0	0.881086 0.940766 -2.989685
3	8	0	-3.989950 0.901268 1.510275
4	8	0	-3.666619 -0.634175 -1.651224
5	7	0	2.213950 1.079457 -0.008867
6	7	0	2.089257 -1.074957 0.976874
7	7	0	2.684124 -0.891175 -1.304792
8	7	0	2.184394 3.030688 1.068875
9	6	0	-5.525890 -0.815527 -0.171582
10	6	0	-4.544712 -1.415149 -0.923920
11	6	0	-6.458611 -1.655593 0.524947
12	6	0	-6.345858 -3.070627 0.425199

Table S4. Standard orientation of the optimized geometry for 2.

13	6	0	-7.270476	-3.894527	1.115345
14	1	0	-7.169807	-4.973301	1.027351
15	6	0	-5.313092	-3.630124	-0.373911
16	1	0	-5.239201	-4.711031	-0.457411
17	6	0	-7.503153	-1.115397	1.321327
18	1	0	-7.596744	-0.037684	1.408352
19	6	0	-4.432608	-2.823519	-1.035804
20	1	0	-3.641311	-3.231147	-1.657501
21	6	0	-8.269716	-3.344184	1.875847
22	1	0	-8.973391	-3.982536	2.400777
23	6	0	-8.382989	-1.938854	1.978298
24	1	0	-9.173671	-1.507214	2.584159
25	6	0	-5.699955	0.666242	-0.131175
26	6	0	-6.953086	2.677590	-0.834884
27	6	0	-4.954503	1.467457	0.700646
28	6	0	-6.156917	3.451262	0.051992
29	1	0	-6.340310	4.519586	0.127730
30	6	0	-5.181531	2.863073	0.804090
31	1	0	-4.565797	3.438631	1.488322
32	6	0	-6.729795	1.275238	-0.924613
33	6	0	-7.967981	3.274632	-1.624269
34	1	0	-8.125026	4.347049	-1.541920
35	6	0	-8.733880	2.518104	-2.473344
36	1	0	-9.508507	2.983955	-3.074279
37	6	0	-7.539590	0.518114	-1.812874
38	1	0	-7.375977	-0.551659	-1.894431
39	6	0	-8.512981	1.125109	-2.566508
40	1	0	-9.119947	0.529410	-3.241253
41	6	0	-1.817698	1.029226	2.401804
42	1	0	-2.189415	0.547039	3.298585
43	6	0	-2.699193	1.349173	1.372425
44	6	0	-0.508519	-1.167728	0.138862
45	1	0	-0.144769	-1.545581	1.087627
46	6	0	-0.459600	1.274831	2.229684

47	6	0	-1.856884	-1.287186	-0.165052
48	1	0	-2.534681	-1.744669	0.546621
49	6	0	0.386997	-0.498970	-0.695367
50	6	0	-2.248449	2.011371	0.231813
51	1	0	-2.934845	2.270607	-0.566435
52	6	0	-0.889894	2.247652	0.092214
53	1	0	-0.517436	2.710406	-0.817650
54	6	0	-0.114286	0.008910	-1.901642
55	6	0	-2.334998	-0.721108	-1.343082
56	6	0	-1.456552	-0.110878	-2.235989
57	1	0	-1.841364	0.327361	-3.149733
58	6	0	3.441854	2.826584	0.551908
59	6	0	1.480393	1.981393	0.746143
60	6	0	2.952513	-1.973247	0.687741
61	6	0	3.500368	1.614788	-0.101181
62	6	0	3.337129	-1.851895	-0.771031
63	6	0	1.839939	-0.326238	-0.258998
64	6	0	4.177719	-4.806693	3.626098
65	1	0	4.480022	-5.525642	4.381570
66	6	0	3.414584	-2.957794	1.684659
67	6	0	3.528594	-2.564131	3.021921
68	1	0	3.305143	-1.534314	3.284056
69	6	0	3.917374	-3.485379	3.986628
70	1	0	4.013861	-3.173995	5.021929
71	6	0	4.047366	-5.206467	2.298504
72	1	0	4.239159	-6.237393	2.018221
73	6	0	3.671355	-4.285081	1.326720
74	1	0	3.568973	-4.596898	0.291386
75	6	0	4.373751	-2.575739	-1.528271
76	6	0	4.239783	-2.685870	-2.917295
77	1	0	3.338363	-2.302626	-3.385716
78	6	0	5.253408	-3.260349	-3.672473
79	1	0	5.141779	-3.349201	-4.748690
80	6	0	6.559434	-3.592717	-1.670007

81	1	0	7.470264 -3.929593 -1.184722
82	6	0	6.415789 -3.713786 -3.049201
83	1	0	7.210655 -4.158864 -3.640323
84	6	0	5.541259 -3.028601 -0.907700
85	1	0	5.670858 -2.905101 0.162696
86	6	0	4.694578 0.979481 -0.705971
87	6	0	4.832359 0.818293 -2.089119
88	1	0	4.007256 1.096618 -2.737328
89	6	0	5.750131 0.607097 0.130997
90	1	0	5.646398 0.748293 1.203401
91	6	0	6.000287 0.282592 -2.620345
92	1	0	6.091244 0.148421 -3.694112
93	6	0	7.051904 -0.081868 -1.780080
94	1	0	7.962449 -0.500150 -2.198423
95	6	0	6.925456 0.083580 -0.404411
96	1	0	7.740313 -0.197437 0.256352
97	6	0	4.489511 3.848844 0.730327
98	6	0	5.572324 3.979237 -0.148001
99	1	0	5.658266 3.320723 -1.006037
100	6	0	4.381960 4.741819 1.803476
101	1	0	3.529175 4.652397 2.468866
102	6	0	6.532891 4.963979 0.057998
103	1	0	7.364015 5.052586 -0.635449
104	6	0	6.425894 5.838868 1.135572
105	1	0	7.176618 6.607604 1.292459
106	6	0	5.342929 5.726189 2.004604
107	1	0	5.245184 6.409538 2.843164
108	6	0	0.036934 1.837483 1.051461


Figure S32. Optimized structure of 2.

SCF Done: E(RM062X)	=	-3674.20681880 A.U.
Zero-point correction	=	0.840173 (Hartree/Particle)
Thermal correction to Energy	=	0.892764
Thermal correction to Enthalpy	=	0.893709
Thermal correction to Gibbs Free Energy	=	0.750066
Sum of electronic and zero-point Energies	=	-3673.366645
Sum of electronic and thermal Energies	=	-3673.314054
Sum of electronic and thermal Enthalpies	=	-3673.313110
Sum of electronic and thermal Free Energies =	-36	573.456753
Low frequencies8.3944 -4.4372 -0.0014	0.0002 0.0013	3.7886
Low frequencies 10.5397 16.1376 16.6243		

The Result for the TDDFT calculation for  ${\bf 2}$ 

Excitation energies and oscillator strengths:

Excited State 1: Singlet-A 2.5889 eV 478.91 nm f=0.0048 <S\*\*2>=0.000 244 -> 245 0.70233

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -3674.69470866

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 3.3013 eV 375.56 nm f=0.0468 <S\*\*2>=0.000 242 -> 245 0.14252

Excited State 3: Singlet-A 3.5666 eV 347.63 nm f=0.0044 <S\*\*2>=0.000 235 -> 245 -0.12277 239 -> 245 0.51249 240 -> 245 -0.42995

Excited State	4: Singlet-A	3.7223 eV 333.09 nm	f=0.0240 <s**2>=0.000</s**2>
237 -> 245	-0.14892		
242 -> 245	0.66210		
243 -> 245	-0.13513		

Excited State 5: Singlet-A 3.8042 eV 325.92 nm f=0.0151 <S\*\*2>=0.000 225 -> 245 0.11349 226 -> 245 0.12050 232 -> 245 -0.11062 235 -> 245 -0.15999 236 -> 245 -0.35050 238 -> 245 -0.20708 239 -> 245 0.26061

- 240 -> 245 0.40069
- Excited State 6: Singlet-A 3.8350 eV 323.29 nm f=0.0125 <S\*\*2>=0.000 244 -> 246 0.66336 244 -> 248 -0.20970

Excited State	7:	Singlet-A	3.8541 eV	321.69 nm	f=0.0004	<s**2>=0.000</s**2>
241 -> 245		0.68634				
243 -> 245		0.11487				

Excited State	8:	Singlet-A	3.8806	eV	319.50 nm	f=0.0051	<s**2>=</s**2>	=0.000
244 -> 246		0.16945						
244 -> 247		-0.29360						
244 -> 248		0.60508						

Excited State	9: Singlet-A	3.8832 eV 319.29 nm f=0.0213 <s**2>=0.000</s**2>
235 -> 245	0.12062	
236 -> 245	-0.14188	
237 -> 245	0.12076	
238 -> 245	0.57423	
239 -> 245	0.18518	
240 -> 245	0.21616	
Excited State	10: Singlet-A	3.9538 eV 313.58 nm f=0.0006 <s**2>=0.000</s**2>
244 -> 246	0.12454	
244 -> 247	0.62414	
244 -> 248	0.27435	
Excited State	11: Singlet-A	3.9781 eV 311.67 nm f=0.0210 <s**2>=0.000</s**2>
225 -> 245	-0.11472	
226 -> 245	-0.14065	
230 -> 245	0.12515	
232 -> 245	0.16552	
235 -> 245	0.37227	
237 -> 245	-0.34277	
238 -> 245	-0.18746	
239 -> 245	0.25014	
240 -> 245	0.12987	
Excited State	12: Singlet-A	4.0425 eV 306.70 nm f=0.0225 <s**2>=0.000</s**2>
235 -> 245	-0.16573	
236 -> 245	0.52101	
237 -> 245	0.26014	
239 -> 245	0.22873	
240 -> 245	0.25035	
Excited State	13: Singlet-A	4.1264 eV 300.46 nm f=0.1130 <s**2>=0.000</s**2>

237 -> 245 0.17576

242 -> 246	-0.13356	
243 -> 246	0.60353	
Excited State	14: Singlet-A	4.1308 eV 300.15 nm f=0.0094 <s**2>=0.000</s**2>
229 -> 245	0.14237	
230 -> 245	0.13298	
233 -> 245	0.13891	
235 -> 245	0.18596	
236 -> 245	-0.20204	
237 -> 245	0.44354	
238 -> 245	-0.21610	
242 -> 245	0.11484	
243 -> 246	-0.24144	
Excited State	15: Singlet-A	4.1599 eV 298.04 nm f=0.0081 <s**2>=0.000</s**2>
244 -> 249	0.66673	
244 -> 250	-0.12382	
Excited State	16: Singlet-A	4.1931 eV 295.69 nm f=0.0199 <s**2>=0.000</s**2>
241 -> 246	0.11137	
241 -> 247	0.15026	
242 -> 246	-0.15761	
242 -> 247	-0.10474	
243 -> 247	0.59634	
Excited State	17: Singlet-A	4.2155 eV 294.11 nm f=0.0540 <s**2>=0.000</s**2>
242 -> 246	0.63773	
243 -> 246	0.13329	
243 -> 247	0.13092	
Excited State	18: Singlet-A	4.2279 eV 293.25 nm f=0.0069 <s**2>=0.000</s**2>
225 -> 245	0.20477	
230 -> 245	-0.22563	
231 -> 245	-0.27296	

232 -> 245	-0.30516
233 -> 245	0.14562
235 -> 245	0.38704
236 -> 245	0.12607

Excited State 19: Singlet-A 4.2689 eV 290.43 nm f=0.0100 <S\*\*2>=0.000 231 -> 245 0.16721 233 -> 245 0.47534 235 -> 245 -0.16804 244 -> 250 -0.16922 244 -> 252 0.28618 244 -> 254 -0.19229

Excited State	20: Singlet-	A 4.2923 eV	288.85 nm	f=0.0167	<S**2>=0.000
233 -> 245	0.17011				
241 -> 246	-0.36241				
241 -> 247	0.21078				
242 -> 247	0.46284				
244 -> 250	0.10439				
244 -> 252	-0.15783				

4.2954 eV 288.64 nm f=0.1051 <S\*\*2>=0.000 Excited State 21: Singlet-A 233 -> 245 -0.35871 241 -> 246 -0.15392 241 -> 247 0.10232 242 -> 247 0.19912 244 -> 250 -0.23783 244 -> 252 0.37849 244 -> 254 -0.19980

Excited State	22:	Singlet-A	4.3711 eV	283.65 nm	f=0.1093	<s**2>=0.000</s**2>	)
230 -> 245	-	-0.21504					
231 -> 245	-	-0.36046					
232 -> 245		0.51649					

Excited State	23: Singlet-A	4.4003 eV 281.76 nm f=0.0093 <s**2>=0.000</s**2>
234 -> 246	0.12017	
241 -> 246	0.49154	
241 -> 247	0.11281	
242 -> 247	0.30454	
242 -> 250	-0.10596	
243 -> 246	-0.10690	
243 -> 249	0.18431	
Excited State	24: Singlet-A	4.4082 eV 281.26 nm f=0.0593 <s**2>=0.000</s**2>
244 -> 250	-0.31449	
244 -> 251	-0.32839	
244 -> 253	-0.20452	
244 -> 254	0.45273	
Excited State	25: Singlet-A	4.4402 eV 279.23 nm f=0.0285 <s**2>=0.000</s**2>
241 -> 247	0.10037	
244 -> 250	-0.17301	
244 -> 251	0.60233	
244 -> 252	0.11248	
244 -> 253	-0.10421	
244 -> 254	0.22175	
Excited State	26: Singlet-A	4.4530 eV 278.43 nm f=0.0548 <s**2>=0.000</s**2>
241 -> 247	0.57477	
242 -> 247	-0.25317	
243 -> 247	-0.19792	
Excited State	27: Singlet-A	4.4841 eV 276.50 nm f=0.0285 <s**2>=0.000</s**2>
229 -> 245	-0.31584	
230 -> 245	0.29732	

231 -> 245 -0.15674

234 -> 245	0.32120
241 -> 246	0.14392
243 -> 249	-0.28451

Excited State 28: Singlet-A 4.5133 eV 274.71 nm f=0.0183 <S\*\*2>=0.000

231 -> 245	-0.10182
234 -> 245	0.39057
237 -> 246	0.11037
238 -> 246	-0.10819
241 -> 247	0.13122
241 -> 250	0.12652
242 -> 249	0.14104
243 -> 249	0.39083

Excited State	29: Singlet-A	4.5335 eV	273.49 nm	f=0.0086	<S**2>=0.000
229 -> 245	0.10074				
234 -> 245	-0.15883				
234 -> 247	-0.16648				
237 -> 246	0.22083				
238 -> 246	-0.20912				
241 -> 249	0.12574				
241 -> 250	0.19958				
242 -> 249	0.10848				
242 -> 253	-0.15149				
243 -> 247	0.16228				
243 -> 248	-0.20719				
243 -> 249	-0.15781				
243 -> 250	0.20609				

Excited State 3	0: Singlet-A	4.5380 eV 273.21 nr	n f=0.0108 <s**2>=0.000</s**2>
225 -> 245	0.16495		
226 -> 245	0.14985		
228 -> 245	0.34256		
229 -> 245	-0.11803		

230 -> 245	0.35940
232 -> 245	0.13717
234 -> 245	-0.18873
243 -> 249	0.27113

Excited State 31: Singlet-A 4.5454 eV 272.77 nm f=0.0080 <S\*\*2>=0.000 244 -> 249 0.14513 244 -> 250 0.42590 244 -> 252 0.39834 244 -> 254 0.22265 244 -> 256 -0.14996

Excited State	32: Singlet-A	4.5582 eV 272.00 nm f=0.0092 <s**2>=0.000</s**2>
228 -> 245	0.33747	
229 -> 245	0.37594	
231 -> 245	0.11219	
234 -> 245	0.19728	
234 -> 246	-0.10353	
237 -> 247	0.11409	
242 -> 247	0.10464	

242 -> 250	0.15813
243 -> 249	-0.12728

Excited State	33: Singlet-A	4.5652 eV 271.59 nm f=0.0360 <s**2>=0.000</s**2>	
228 -> 245	-0.15411		
229 -> 245	-0.16184		
234 -> 245	-0.22600		
234 -> 246	-0.21776		
237 -> 247	0.19334		
238 -> 247	-0.16875		
241 -> 249	0.12012		
241 -> 253	-0.13384		
242 -> 247	0.15328		
242 -> 250	0.28947		

243 -> 246 0.11144

Excited State 34: Singlet-A 4.5831 eV 270.53 nm f=0.0186 <S\*\*2>=0.000 227 -> 245 0.11540 242 -> 248 0.11452 243 -> 248 0.59511

Excited State	35: Singlet-A	4.5923 eV 269.98 nm f=0.0205 <s**2>=0.000</s**2>
222 -> 245	0.16669	
225 -> 245	0.13020	
226 -> 245	0.23518	
227 -> 245	0.39397	
228 -> 245	-0.18830	
231 -> 245	0.12620	
233 -> 245	-0.11781	
234 -> 245	0.23883	
235 -> 245	0.15303	
243 -> 248	-0.14956	

Excited State 36: Singlet-A 4.6078 eV 269.07 nm f=0.0094 <S\*\*2>=0.000 244 -> 255 0.16305 244 -> 256 0.39635 244 -> 258 0.44585 244 -> 260 -0.16994

Excited State	37: Singlet-A	4.6679 eV 2	265.61 nm	f=0.0052	<s**2>=0.000</s**2>
225 -> 245	0.17555				
226 -> 245	0.10028				
227 -> 245	-0.30212				
229 -> 245	-0.30053				
230 -> 245	-0.23427				
231 -> 245	0.37005				
232 -> 245	0.16880				
237 -> 245	0.11459				

Excited State	38: Singlet-A	4.7331 eV	261.95 nm	f=0.0113	<s**2>=0.000</s**2>
222 -> 245	0.14079				
225 -> 245	-0.30046				
226 -> 245	-0.11572				
227 -> 245	0.17278				
228 -> 245	0.32077				
229 -> 245	-0.21429				
230 -> 245	-0.18313				
233 -> 245	-0.12377				
244 -> 253	0.17922				
244 -> 257	-0.16477				
Excited State	39: Singlet-A	4.7387 eV	261.64 nm	f=0.1070	<s**2>=0.000</s**2>
225 -> 245	0.13657				

-0.14846

0.10462

0.39082

0.15747

0.16443

-0.38951

-0.11781

228 -> 245 229 -> 245

244 -> 253

244 -> 254

244 -> 256

244 -> 257

244 -> 258

Excited State	40: Singlet-A	4.7959 eV 258.52 nm f=0.0706 <s**2>=0.000</s**2>
238 -> 246	0.14386	
239 -> 248	-0.25307	
240 -> 248	0.21679	
243 -> 250	0.25467	
243 -> 252	-0.16128	
244 -> 255	0.36907	
244 -> 256	-0.19873	



MO245



MO244



MO243



MO242



MO237

**Figure S33.** The relevant molecular orbitals of **2** calculated at the MPW1PW91/6-31+G(d)//M062X/6-31G(d) level.

	Center	Atomic	Atomic	Coordinates (Angstroms)	
	Number	Number	Туре	X Y Z	
-	1	6	0	-8.743167 3.193483 1.895170	
	2	6	0	-8.755948 2.689803 0.573852	
	3	6	0	-7.840986 1.748362 0.174556	
	4	6	0	-6.860182 1.260755 1.079212	
	5	6	0	-6.847352 1.770993 2.407875	
	6	6	0	-7.807474 2.742130 2.789660	
	7	6	0	-5.886582 0.285882 0.688180	
	8	6	0	-4.962302 -0.128770 1.617107	
	9	6	0	-4.952589 0.359802 2.945872	
	10	6	0	-5.874601 1.293390 3.326429	
	11	6	0	-5.886129 -0.285917 -0.688452	
	12	6	0	-6.859568 -1.260673 -1.080154	
	13	6	0	-6.845901 -1.770941 -2.408786	
	14	6	0	-5.872500 -1.293431 -3.326683	
	15	6	0	-4.950632 -0.359954 -2.945508	
	16	6	0	-4.961119 0.128603 -1.616736	
	17	6	0	-7.841030 -1.748172 -0.176145	
	18	6	0	-8.755801 -2.689549 -0.576023	
	19	6	0	-8.742169 -3.193278 -1.897320	
	20	6	0	-7.805839 -2.742018 -2.791190	
	21	8	0	-4.074935 -1.122685 1.258621	
	22	8	0	-4.073796 1.122307 -1.257599	
	23	6	0	-2.734458 -0.928190 1.422205	
	24	6	0	-2.733315 0.928014 -1.421540	
	25	6	0	-1.952084 -2.075147 1.306873	
	26	6	0	-0.570154 -1.981771 1.341258	
	27	6	0	0.071654 -0.735610 1.511781	
	28	6	0	-0.757558 0.397830 1.634500	
	29	6	0	-2.137689 0.322126 1.601153	
	30	6	0	-1.951039 2.075063 -1.306607	
	31	6	0	-0.569087 1.981792 -1.341151	

 Table S5. Standard orientation of the optimized geometry for 2R.

32	6	0	0.072796	0.735634	-1.511295
33	6	0	-0.756357	-0.397898	-1.633708
34	6	0	-2.136477	-0.322292	-1.600296
35	6	0	1.505456	-0.524422	1.550151
36	6	0	1.506582	0.524416	-1.549535
37	7	0	2.034320	-0.733048	-1.428788
38	6	0	3.338685	-0.558740	-1.565339
39	6	0	3.585592	0.890605	-1.769676
40	7	0	2.415052	1.508612	-1.767315
41	7	0	2.413808	-1.508779	1.768392
42	6	0	3.584296	-0.890794	1.770450
43	6	0	3.337716	0.558460	1.565692
44	7	0	2.033319	0.732911	1.429110
45	6	0	4.258945	1.696758	1.591857
46	6	0	4.807126	-1.683175	1.907494
47	6	0	4.808194	1.683490	-1.906647
48	6	0	4.259425	-1.697479	-1.591525
49	6	0	3.913628	-2.863264	-0.890088
50	6	0	4.765436	-3.958903	-0.894217
51	6	0	5.960588	-3.918196	-1.611645
52	6	0	6.291464	-2.780291	-2.344328
53	6	0	5.445231	-1.677277	-2.339837
54	6	0	6.038276	1.280018	-1.370695
55	6	0	7.138661	2.127777	-1.409294
56	6	0	7.032849	3.384893	-1.999621
57	6	0	5.816407	3.792559	-2.547736
58	6	0	4.710191	2.957048	-2.491636
59	6	0	4.711090	-2.955174	2.495954
60	6	0	5.818044	-3.789820	2.551582
61	6	0	7.032947	-3.382718	1.999644
62	6	0	7.136614	-2.127135	1.405608
63	6	0	6.035501	-1.280345	1.367371
64	6	0	3.912913	2.863388	0.891920
65	6	0	4.765309	3.958555	0.896046

66	6	0	5.961319	3.916581	1.611969
67	6	0	6.292457	2.777846	2.343219
68	6	0	5.445655	1.675288	2.338733
69	1	0	-9.473629	3.937750	2.196506
70	1	0	-9.496138	3.054614	-0.131493
71	1	0	-7.854147	1.369650	-0.842616
72	1	0	-7.784285	3.121925	3.807814
73	1	0	-4.211053	-0.023858	3.639649
74	1	0	-5.880039	1.675811	4.343443
75	1	0	-5.877294	-1.675825	-4.343711
76	1	0	-4.208628	0.023600	-3.638840
77	1	0	-7.854844	-1.369430	0.841007
78	1	0	-9.496490	-3.054278	0.128841
79	1	0	-9.472473	-3.937503	-2.199131
80	1	0	-7.781985	-3.121855	-3.809313
81	1	0	-2.437187	-3.031424	1.149885
82	1	0	-0.267546	1.360303	1.741870
83	1	0	-2.742324	1.220467	1.661378
84	1	0	-2.436197	3.031349	-1.149850
85	1	0	-0.266278	-1.360365	-1.740830
86	1	0	-2.741059	-1.220689	-1.660281
87	1	0	2.989665	-2.872970	-0.318249
88	1	0	4.502691	-4.843816	-0.322871
89	1	0	6.627378	-4.775356	-1.606788
90	1	0	7.205736	-2.754417	-2.929304
91	1	0	5.693426	-0.803155	-2.933844
92	1	0	6.126153	0.302327	-0.913782
93	1	0	8.078279	1.807521	-0.969485
94	1	0	7.894367	4.045101	-2.031442
95	1	0	5.728944	4.770984	-3.010548
96	1	0	3.748785	3.272963	-2.883116
97	1	0	3.750766	-3.270696	2.890431
98	1	0	5.732335	-4.767150	3.017030
99	1	0	7.894929	-4.042337	2.031153

100	1	0	8.074849	-1.807659	0.962278
101	1	0	6.120767	-0.304411	0.906014
102	1	0	2.988293	2.874095	0.321167
103	1	0	4.502354	4.844159	0.325865
104	1	0	6.628551	4.773397	1.607019
105	1	0	7.207446	2.750906	2.927026
106	1	0	5.694195	0.800527	2.931604
107	17	0	0.292817	3.470176	-1.086718
108	17	0	0.291906	-3.469887	1.086000



Figure S34. Optimized structure of 2R.

SCF Done: E(UM062X)	=	-3674.18880187 A.U	•
S**2 before annihilation 0.8370, after	0.1535		
Zero-point correction	=	0.838224 (Hartree/Parti	cle)
Thermal correction to Energy	=	0.891260	
Thermal correction to Enthalpy	=	0.892204	
Thermal correction to Gibbs Free Energy	=	0.750214	
Sum of electronic and zero-point Energies	=	-3673.350578	
Sum of electronic and thermal Energies	=	-3673.297542	
Sum of electronic and thermal Enthalpies	=	-3673.296598	
Sum of electronic and thermal Free Energie	es =	-3673.438588	
Low frequencies3.3728 -0.0020 0.	0007 0.0018	6.1894 7.1811	
Low frequencies 13.6164 18.0737 25	5.5366		

## The Result for the TDDFT calculation for $\mathbf{2R}$

Excitation energies and oscillator strengths:

Excited State 1: 1.732-A 1.0269 eV 1207.42 nm f=0.0024 <S\*\*2>=0.500 244A -> 245A 0.70505 244B -> 245B 0.70424 This state for optimization and/or second-order correction. Total Energy, E(TD-HF/TD-KS) = -3674.73338606Copying the excited state density for this state as the 1-particle RhoCI density. 1.4051 eV 882.41 nm f=0.0534 <S\*\*2>=-0.009 Excited State 2: 0.983-A 244A -> 245A -0.70133  $244B \rightarrow 245B$ 0.70213 Excited State 3: 2.596-A 1.8925 eV 655.14 nm f=0.0000 <S\*\*2>=1.434 239A -> 245A 0.22041 240A -> 245A 0.62387 239B -> 245B 0.22133 240B -> 245B 0.62187 Excited State 4: 2.642-A 1.9760 eV 627.46 nm f=0.0051 <S\*\*2>=1.495 241A -> 245A 0.32752 242A -> 245A -0.45441 243A -> 245A 0.37723 241B -> 245B -0.32711 242B -> 245B 0.45496 243B -> 245B -0.37517 Excited State 5: 2.217-A 2.0193 eV 613.99 nm f=0.0447 <S\*\*2>=0.978 234A -> 245A 0.11661 238A -> 245A -0.11651 239A -> 245A -0.35280 240A -> 245A -0.52430 234B -> 245B -0.11681 238B -> 245B 0.10273 239B -> 245B 0.35807 240B -> 245B 0.52625 S52

Excited State	6:	2.128-A	2.1720 eV	570.82 nm	f=0.2472	<s**2>=0.883</s**2>
231A -> 245	A	0.11548				
232A -> 245	A	-0.22509				
233A -> 245	A	0.12310				
234A -> 245	A	0.16893				
236A -> 245	A	-0.10775				
237A -> 245	A	0.16252				
238A -> 245	A	0.17953				
241A -> 245	A	-0.24453				
242A -> 245	A	0.34304				
243A -> 245	A	-0.32478				
231B -> 245I	В	0.11751				
232B -> 245I	В	-0.22254				
233B -> 245I	В	0.12187				
234B -> 245I	В	0.16886				
236B -> 245I	В	-0.10499				
237B -> 245I	В	0.16535				
238B -> 245I	В	0.17589				
241B -> 245I	B	-0.24543				
242B -> 245I	В	0.34502				
243B -> 245I	В	-0.32448				

Excited State 7: 2.279-A 2.3010 eV 538.82 nm f=0.0004 <S\*\*2>=1.048

228A -> 245A	-0.16440
229A -> 245A	0.10852
231A -> 245A	-0.17570
232A -> 245A	0.41800
233A -> 245A	-0.21821
234A -> 245A	-0.21958
235A -> 245A	0.11595
236A -> 245A	0.18744
237A -> 245A	-0.25095
238A -> 245A	-0.11061

243A -> 245A	-0.13175
228B -> 245B	0.16071
229B -> 245B	-0.10872
231B -> 245B	0.17832
232B -> 245B	-0.41162
233B -> 245B	0.21449
234B -> 245B	0.21820
235B -> 245B	-0.11203
236B -> 245B	-0.18089
237B -> 245B	0.25311
238B -> 245B	0.10552
243B -> 245B	0.12402

Excited State	8:	2.420-A	2.3417 eV	529.46 nm	f=0.1591	<s**2>=1.214</s**2>
228A -> 245	A	0.16369				
231A -> 245	A	0.10694				
232A -> 245	A	-0.28802				
233A -> 245	A	0.17274				
234A -> 245	A	0.17469				
236A -> 245	A	-0.20756				
237A -> 245	A	0.15187				
240A -> 245	A	0.12100				
243A -> 245	A	0.40972				
228B -> 245	В	0.16527				
231B -> 245	В	0.11305				
232B -> 245	В	-0.29394				
233B -> 245	В	0.17559				
234B -> 245	В	0.17947				
236B -> 245	В	-0.20795				
237B -> 245	В	0.16024				
240B -> 245	В	0.12262				
243B -> 245	В	0.41023				

Excited State 9: 2.558-A 2.3859 eV 519.65 nm f=0.0082 <S\*\*2>=1.386

234A -> 245A	-0.10652
239A -> 245A	0.21324
241A -> 245A	-0.17763
242A -> 245A	0.45519
243A -> 245A	0.37767
243A -> 246A	0.10169
234B -> 245B	-0.10489
239B -> 245B	0.20080
241B -> 245B	-0.17286
242B -> 245B	0.44408
243B -> 245B	0.36367
243B -> 246B	0.10126

F	Excited State 10	): 2.075-A	2.4020 eV 516.18 nm f=0.0014 <s**2>=0.826</s**2>
	239A -> 245A	-0.33735	
	240A -> 245A	0.19453	
	241A -> 245A	0.14473	
	242A -> 245A	-0.29291	
	243A -> 245A	-0.44686	
	239B -> 245B	0.34344	
	240B -> 245B	-0.19480	
	241B -> 245B	-0.15049	
	242B -> 245B	0.30622	
	243B -> 245B	0.46147	

Excited State 11: 2.136-A 2.4679 eV 502.38 nm f=0.0190 <S\*\*2>=0.891

238A -> 245A	-0.18134
239A -> 245A	-0.38678
240A -> 245A	0.38798
242A -> 245A	0.20797
243A -> 245A	0.31065
238B -> 245B	0.16779
239B -> 245B	0.39259
240B -> 245B	-0.38848

242B -> 245B	-0.20821
243B -> 245B	-0.31084

Excited State 12:	2.079-A	2.5253 eV	490.96 nm	f=0.0603	<s**2>=0.830</s**2>
233A -> 245A	0.10379				
236A -> 245A	0.10212				
238A -> 245A	0.13338				
239A -> 245A	0.55199				
240A -> 245A	-0.24334				
242A -> 245A	-0.23526				
233B -> 245B	0.10427				
236B -> 245B	0.10032				
238B -> 245B	0.11409				
239B -> 245B	0.55748				
240B -> 245B	-0.24429				
242B -> 245B	-0.23620				

Excited State 13:	2.969-A	2.5445 eV	487.27 nm	f=0.0008	<s**2>=1.954</s**2>
241A -> 245A	0.46096				
241A -> 246A	0.22293				
242A -> 245A	0.31833				
242A -> 246A	0.14996				
242A -> 247A	-0.11762				
243A -> 247A	-0.22040				
241B -> 245B	-0.45855				
241B -> 246B	-0.22292				
242B -> 245B	-0.31629				
242B -> 246B	-0.14992				

242B -> 247B	0.11677
243B -> 247B	0.22086

Excited State 14:	2.616-A	2.5645 eV	483.46 nm	f=0.0052	<s**2>=1.462</s**2>
235A -> 245A	-0.13478				
238A -> 245A	-0.31090				

239A -> 245A	0.11927
241A -> 245A	0.34410
241A -> 247A	0.18593
242A -> 245A	0.26836
242A -> 246A	-0.12116
242A -> 247A	0.11099
243A -> 246A	-0.21723
235B -> 245B	-0.13462
238B -> 245B	-0.32026
239B -> 245B	0.10954
241B -> 245B	0.34402
241B -> 247B	0.18592
242B -> 245B	0.26746
242B -> 246B	-0.12014
242B -> 247B	0.11128
243B -> 246B	-0.21776

Excited State 15	: 2.560-A	2.5914 eV	478.45 nm	f=0.0023	<s**2>=1.389</s**2>
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232A -> 245A	-0.11809
235A -> 245A	0.11205
237A -> 245A	-0.54763
238A -> 245A	0.28704
232B -> 245B	0.12036
235B -> 245B	-0.14189
236B -> 245B	0.11723
237B -> 245B	0.56902
238B -> 245B	-0.33958

Excited State 16: 3.030-A 2.5935 eV 478.07 nm f=0.0053 <S\*\*2>=2.046 235A -> 245A 0.21457 236A -> 245A -0.27170 237A -> 245A -0.14143 238A -> 245A 0.31346 241A -> 247A 0.24285

242A -> 246A	-0.16040
242A -> 247A	0.14377
243A -> 245A	0.15904
243A -> 246A	-0.28955
235B -> 245B	0.19229
236B -> 245B	-0.25864
238B -> 245B	0.26734
241B -> 247B	0.24267
242B -> 246B	-0.16049
242B -> 247B	0.14493
243B -> 245B	0.15101
243B -> 246B	-0.28945

Excited State 17: 2.146-A 2.6183 eV 473.53 nm f=0.0001 <S\*\*2>=0.901 235A -> 245A 0.15751 237A -> 245A -0.32983 238A -> 245A 0.27034 241A -> 245A 0.43846 242A -> 245A 0.16462 243A -> 246A 0.12014 235B -> 245B 0.15104 237B -> 245B -0.31352 238B -> 245B 0.26306

241B -> 245B 0.43816 242B -> 245B 0.16469 243B -> 246B 0.12012

Excited State 18: 2.337-A 2.6583 eV 466.40 nm f=0.0132 <S\*\*2>=1.116 231A -> 245A -0.12404 232A -> 245A 0.17551 234A -> 245A -0.18093 236A -> 245A -0.33440 237A -> 245A 0.39886

238A -> 245A 0.13125

241A -> 245A	0.26746
231B -> 245B	-0.12564
232B -> 245B	0.17553
234B -> 245B	-0.17997
236B -> 245B	-0.33441
237B -> 245B	0.40035
238B -> 245B	0.12956
241B -> 245B	0.25562

Excited State 19: 3.140-A 2.6609 eV 465.96 nm f=0.0010 <S\*\*2>=2.214

236A -> 245A	-0.11061
241A -> 245A	-0.32699
241A -> 246A	0.28414
242A -> 245A	-0.18361
242A -> 246A	0.18807
242A -> 247A	-0.16129
243A -> 247A	-0.29824
241B -> 245B	0.33792
241B -> 246B	-0.28424
242B -> 245B	0.18504
242B -> 246B	-0.18629
242B -> 247B	0.15849
243B -> 245B	-0.10127
243B -> 247B	0.29969

2.7080 eV 457.85 nm f=0.0022 <S\*\*2>=0.896 Excited State 20: 2.141-A 230A -> 245A 0.13351 231A -> 245A -0.17720 232A -> 245A 0.15189 234A -> 245A -0.16016 236A -> 245A -0.43372 237A -> 245A 0.20443 238A -> 245A 0.30760

239A -> 245A -0.11555

241A -> 245A	0.13943
230B -> 245B	0.13563
231B -> 245B	0.17577
232B -> 245B	-0.15333
234B -> 245B	0.16017
236B -> 245B	0.43547
237B -> 245B	-0.20921
238B -> 245B	-0.31155
239B -> 245B	0.10322
241B -> 245B	-0.14066

Excited State 21: 2.391-A 2.7351 eV 453.30 nm f=0.0017 <S\*\*2>=1.180

230A -> 245A	-0.12367
231A -> 245A	-0.11965
233A -> 245A	0.16454
235A -> 245A	0.46345
236A -> 245A	0.33158
237A -> 245A	0.36045
239A -> 245A	-0.12413
230B -> 245B	0.10919
231B -> 245B	-0.11239
233B -> 245B	0.15128
235B -> 245B	0.38822
236B -> 245B	0.28794
237B -> 245B	0.32993
239B -> 245B	-0.10335

Excited State 22: 2.058-A 2.7504 eV 450.78 nm f= $0.0002 < S^{**}2 > =0.809$ 232A -> 245A 0.15929 235A -> 245A -0.51191 236A -> 245A -0.24268 237A -> 245A -0.15447 238A -> 245A -0.11428 239A -> 245A 0.10707

229B -> 245B	0.10123
232B -> 245B	-0.17514
235B -> 245B	0.58130
236B -> 245B	0.28447
237B -> 245B	0.19961
238B -> 245B	0.12397
239B -> 245B	-0.11783

Excited State 23: 2.520-A 2.7576 eV 449.61 nm f=0.0047 <S\*\*2>=1.337

228A -> 245A	0.11671
232A -> 245A	-0.29266
233A -> 245A	-0.19978
234A -> 245A	-0.13311
235A -> 245A	0.34453
236A -> 245A	-0.21551
238A -> 245A	-0.28220
243A -> 245A	-0.16824
228B -> 245B	0.11532
232B -> 245B	-0.28989
233B -> 245B	-0.20023
234B -> 245B	-0.13546
235B -> 245B	0.33751
236B -> 245B	-0.22193
238B -> 245B	-0.28420
243B -> 245B	-0.16913

Excited State 24: 2.265-A 2.7836 eV 445.41 nm f=0.0011 <S\*\*2>=1.033 227A -> 245A -0.13926

/		0.10/20
232A -	>245A	0.34339
233A ->	> 245A	0.24256
234A ->	> 245A	0.24874
235A ->	>245A	-0.16548
236A ->	> 245A	0.18996
238A -	>245A	0.30481

243A -> 245A	0.10264
227B -> 245B	0.13896
230B -> 245B	-0.10271
232B -> 245B	-0.34222
233B -> 245B	-0.24316
234B -> 245B	-0.25031
235B -> 245B	0.15926
236B -> 245B	-0.19524
238B -> 245B	-0.30736
243B -> 245B	-0.10342

Excited State 25: 2.471-A 2.8877 eV 429.35 nm f=0.0012 <S\*\*2>=1.276

226A -> 245A	-0.12941
229A -> 245A	0.13212
230A -> 245A	-0.12811
231A -> 245A	-0.34770
232A -> 245A	0.14724
233A -> 245A	0.36933
234A -> 245A	0.17879
236A -> 245A	-0.21567
237A -> 245A	-0.10970
238A -> 245A	-0.15350
226B -> 245B	-0.12879
229B -> 245B	0.13074
230B -> 245B	0.12543
231B -> 245B	-0.35527
232B -> 245B	0.14272
233B -> 245B	0.37355
234B -> 245B	0.17677
236B -> 245B	-0.21458
237B -> 245B	-0.11034
238B -> 245B	-0.15410

Excited State 26: 2.510-A 2.9147 eV 425.38 nm f=0.0025 <S\*\*2>=1.325

227A -> 245A	0.29624
229A -> 245A	0.26108
231A -> 245A	-0.29894
233A -> 245A	0.37794
234A -> 245A	-0.13741
235A -> 245A	-0.17013
227B -> 245B	-0.29645
229B -> 245B	-0.25615
231B -> 245B	0.29736
233B -> 245B	-0.37045
234B -> 245B	0.13862
235B -> 245B	0.17365

225A -> 245A	0.18045
226A -> 245A	-0.20958
227A -> 245A	-0.22526
229A -> 245A	0.15300
230A -> 245A	-0.12028
231A -> 245A	-0.19095
233A -> 245A	0.14726
234A -> 245A	0.25235

Excited State 27: 2.183-A 2.9445 eV 421.07 nm f=0.0049 <S\*\*2>=0.942

- 235A -> 245A 0.14537
- 236A -> 245A -0.26492
- 238A -> 245A -0.23904 239A -> 245A 0.16013 225B -> 245B -0.17724 226B -> 245B 0.20825 227B -> 245B 0.22253
- 229B -> 245B -0.15077 230B -> 245B -0.11967 231B -> 245B 0.19847 233B -> 245B -0.15040
- 234B -> 245B -0.25119

235B -> 245B	-0.14139
236B -> 245B	0.26736
238B -> 245B	0.24499
239B -> 245B	-0.14867

Excited State 28: 2.370-A 2.9677 eV 417.78 nm f=0.0040 <S\*\*2>=1.154

225A -> 245A	-0.23132
226A -> 245A	0.45724
227A -> 245A	0.12864
228A -> 245A	0.19492
230A -> 245A	0.13481
232A -> 245A	0.25475
233A -> 245A	0.18317
239A -> 245A	-0.12111
225B -> 245B	-0.23800
226B -> 245B	0.47384
227B -> 245B	0.13309
228B -> 245B	0.20010
230B -> 245B	-0.13838
232B -> 245B	0.26444
233B -> 245B	0.19278
239B -> 245B	-0.12569

Excited State 29: 2.171-A 2.9891 eV 414.78 nm f=0.0010 <S\*\*2>=0.929

225A -> 245A	-0.24746
226A -> 245A	0.36235
228A -> 245A	0.22397
229A -> 245A	-0.13770
232A -> 245A	0.27758
233A -> 245A	0.22724
235A -> 245A	0.18670
238A -> 245A	-0.17255
225B -> 245B	0.23572
226B -> 245B	-0.34670

228B -> 245B	-0.21539
229B -> 245B	0.13800
232B -> 245B	-0.27337
233B -> 245B	-0.21994
234B -> 245B	-0.10203
235B -> 245B	-0.18721
238B -> 245B	0.17071

Excited State 30:	2.394-A	3.0190 eV	410.68 nm	f=0.0021	<s**2>=1.183</s**2>
225A -> 245A	0.11842				
226A -> 245A	-0.18308				
227A -> 245A	-0.12672				
228A -> 245A	0.16013				
230A -> 245A	0.20417				
231A -> 245A	0.33252				
232A -> 245A	0.30795				
234A -> 245A	0.19133				
235A -> 245A	0.24011				
237A -> 245A	0.10657				
225B -> 245B	0.12329				
226B -> 245B	-0.19222				
227B -> 245B	-0.12665				
228B -> 245B	0.15802				
230B -> 245B	-0.20197				
231B -> 245B	0.32988				
232B -> 245B	0.30948				
234B -> 245B	0.19115				
235B -> 245B	0.24283				
237B -> 245B	0.10350				
238B -> 245B	-0.10365				

Excited State 31: 2.431-A 3.0865 eV 401.70 nm f=0.0005 <S\*\*2>=1.228 225A -> 245A 0.16963 226A -> 245A -0.19567

229A -> 245A	-0.10539
230A -> 245A	0.58915
236A -> 245A	0.17374
225B -> 245B	-0.16794
226B -> 245B	0.19702
229B -> 245B	0.10204
230B -> 245B	0.59066
236B -> 245B	-0.17615

Excited State 32: 2.210-A 3.1132 eV 398.26 nm f=0.0005 <S\*\*2>=0.971

225A -> 245A	0.12892
226A -> 245A	-0.20119
227A -> 245A	0.10065
228A -> 245A	0.16132
230A -> 245A	0.44906
231A -> 245A	-0.11935
233A -> 245A	0.15288
234A -> 245A	-0.28231
236A -> 245A	0.15281
225B -> 245B	0.12845
226B -> 245B	-0.20279
227B -> 245B	0.10691
228B -> 245B	0.16151
230B -> 245B	-0.44425
231B -> 245B	-0.10977
233B -> 245B	0.15606
234B -> 245B	-0.28457
236B -> 245B	0.15154

Excited State 33: 2.467-A 3.1355 eV 395.41 nm f=0.0011 <S\*\*2>=1.271 225A -> 245A 0.21886 226A -> 245A -0.23978 227A -> 245A 0.23972 228A -> 245A 0.14561

230A -> 245A	-0.16684
231A -> 245A	0.35698
232A -> 245A	0.16643
233A -> 245A	0.10676
234A -> 245A	-0.21779
235A -> 245A	0.15003
225B -> 245B	-0.21421
226B -> 245B	0.23648
227B -> 245B	-0.24013
228B -> 245B	-0.14142
230B -> 245B	-0.17639
231B -> 245B	-0.35208
232B -> 245B	-0.16893
233B -> 245B	-0.10668
234B -> 245B	0.21309
235B -> 245B	-0 15057

Excited State	34:	2.151-A	3.1837 eV	389.43 nm	f=0.0001	<s**2>=0.907</s**2>
	<i>c</i>	=	0.100/0/	<i>c c c c c c c c c c</i>	1 0.0001	5 = 0.207

226A -> 245A	-0.11310
227A -> 245A	0.20159
228A -> 245A	0.16802
230A -> 245A	-0.35801
231A -> 245A	0.29316
233A -> 245A	0.11342
234A -> 245A	-0.26159
244A -> 248A	0.15301
226B -> 245B	-0.11372
227B -> 245B	0.20500
228B -> 245B	0.16844
230B -> 245B	0.36510
231B -> 245B	0.28635
233B -> 245B	0.11637
234B -> 245B	-0.26144
244B -> 248B	0.15297

Excited State 35: 2.810-A 3.2121 eV 386.00 nm f=0.0008 <S\*\*2>=1.724

224A -> 245A	-0.11010
230A -> 245A	0.11717
244A -> 248A	0.52096
224B -> 245B	-0.11034
230B -> 245B	-0.11621
244B -> 248B	0.52082

Excited State 36: 2.651-A 3.2324 eV 383.56 nm f=0.0024 <S\*\*2>=1.507

227A -> 245A	-0.16763
229A -> 245A	-0.25333
230A -> 245A	-0.10972
231A -> 245A	-0.13222
233A -> 245A	0.14373
234A -> 245A	-0.17488
238A -> 245A	0.10003
244A -> 246A	-0.16495
244A -> 249A	-0.33986
244A -> 252A	0.14927
227B -> 245B	0.16781
229B -> 245B	0.24147
230B -> 245B	-0.10135
231B -> 245B	0.12975
233B -> 245B	-0.14134
234B -> 245B	0.17115
244B -> 246B	0.16682
244B -> 249B	0.34282
244B -> 252B	-0.15016

Excited State 37: 2.362-A 3.2459 eV 381.97 nm f=0.0010 <S\*\*2>=1.145 227A -> 245A -0.12655 228A -> 245A 0.29215 229A -> 245A 0.49251

230A -> 245A	0.11106
231A -> 245A	0.21009
233A -> 245A	-0.13317
234A -> 245A	-0.10421
244A -> 249A	-0.14563
227B -> 245B	0.12238
228B -> 245B	-0.29399
229B -> 245B	-0.46134
230B -> 245B	0.10263
231B -> 245B	-0.20505
233B -> 245B	0.12676
234B -> 245B	0.10857
244B -> 249B	0.14521

Excited State 38:	2.092-A	3.2644 eV	379.81 nm	f=0.0179	<s**2>=0.8</s**2>	344
226A -> 245A	0.12550					
228A -> 245A	0.13251					
229A -> 245A	0.59783					
233A -> 245A	-0.13192					
226B -> 245B	0.12954					
228B -> 245B	0.15773					
229B -> 245B	0.62315					

233B -> 245B -0.14160

Excited State 39: 2.647-A 3.2915 eV 376.68 nm f=0.1839 <S\*\*2>=1.501

227A -> 245A	0.28445
228A -> 245A	-0.20203
229A -> 245A	-0.10903
230A -> 245A	0.16378
234A -> 245A	0.18613
239A -> 245A	0.10686
242A -> 249A	-0.11173
244A -> 246A	-0.18770
244A -> 249A	-0.21231

244A -> 252A	0.10529
227B -> 245B	0.28458
228B -> 245B	-0.20734
229B -> 245B	-0.10737
230B -> 245B	-0.16454
234B -> 245B	0.18739
239B -> 245B	0.10318
242B -> 249B	-0.11195
244B -> 246B	-0.18623
244B -> 249B	-0.21014
244B -> 252B	0.10397

Excited State 40: 2.742-A 3	3.2934 eV 3'	76.46 nm 🗄	f=0.0354	<s**2>=1.</s**2>	.630
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220A -> 245A	-0.12387
224A -> 245A	-0.18847
225A -> 245A	-0.10504
227A -> 245A	-0.12225
228A -> 245A	-0.10870
231A -> 245A	0.12928
237A -> 255A	0.10254
242A -> 248A	0.10988
244A -> 248A	-0.36582
244A -> 250A	-0.21359
220B -> 245B	0.12326
224B -> 245B	0.18694
225B -> 245B	0.10370
227B -> 245B	0.12387
228B -> 245B	0.10317
231B -> 245B	-0.12981
237B -> 255B	-0.10182
242B -> 248B	-0.10989
244B -> 248B	0.36597
244B -> 250B	0.21352





MO245a





MO244α



MO244β



 $MO243\alpha$ 



MO243β



MO242β

Figure S35. The relevant molecular orbitals of 2R calculated at the UMPW1PW91/6-31+G(d)//UM062X/6-31G(d) level. S71

## 11. References

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