

ELECTRONIC SUPPLEMENTARY INFORMATION

The first structural and spectroscopic characterisation of a ring-opened form of a *2H-naphtho[1,2-*b*]pyran: a novel photomerocyanine*

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Experimental Section - General

Unless otherwise stated, reagents were used as supplied by major chemical suppliers. ^1H (400 MHz) and ^{13}C (100 MHz) NMR spectra (Bruker Avance) were recorded in either CDCl_3 or d_6 -acetone. UV-Irradiation experiments for NMR investigations were performed using typical solution concentrations for NMR spectroscopy with irradiation using a Spectroline 8 Watt TLC inspection lamp at 365 nm with frequent manual agitation. FT-IR spectra were recorded on a Nicolet 380 FT-IR spectrophotometer equipped with a diamond ATR attachment (neat sample). Flash column chromatography was performed on chromatography silica gel (Sigma-Aldrich, 40-63 micron particle size distribution). All compounds were homogeneous by TLC using a range of eluent systems of differing polarity [Merck TLC aluminium sheets either silica gel 60 F254 (cat. No 105554) or neutral aluminium oxide 60 F254 (cat. No 105550)]. High resolution mass spectra were recorded under electrospray ionization using a linear ion trap mass spectrometer [Thermo Scientific LTQ Orbitrap XL Fourier transform mass spectrometer, EPRSC National Mass Spectrometry Service, Swansea]. UV-visible spectra were recorded in spectroscopic grade acetone solutions of the samples (10 mm pathlength quartz cuvette, PTFE capped, concentration ranges *ca.* 3×10^{-4} – 10^{-5} mol dm $^{-3}$) using an Agilent Technologies Cary 60 spectrophotometer equipped with a temperature controlled (20 °C) stirred cell. Irradiation of the stirred sample solutions was accomplished with a Spectroline 8 Watt TLC inspection lamp at 365 nm. 1,1-Bis(4-methoxyphenyl)prop-2-yn-1-ol **8a** was prepared according to literature procedures.^{S1}

Experimental Section - DFT and TD-DFT modelling

All quantum mechanical simulations have been performed with the Gaussian09 program^{S2} using the PBE0 hybrid functional,^{S3} and improving the DFT integration grid to *ultrafine* [a pruned (99,590) grid]. First geometry optimizations in acetone have been performed using the PCM (Polarizable Continuum Model) solvent model,^{S4} and the 6-31G(d) atomic basis set. In the second step, vibrational calculations were used to establish that the optimized structures corresponded to true minima of the potential energy surface. In the third stage, the first ten lowest-lying singlet excited-states have been determined within the vertical PCM-TD-DFT (Time-Dependent Density Functional Theory) approximation using the 6-31+G(d) atomic basis set and the PBE0 functional. In a fourth stage, the NMR chemical shifts have been obtained by computing shieldings for the investigated molecules and reference (TMS), at the PCM-PBE0/cc-pVTZ level of theory. In addition, we have also determined the optimal geometry and the vibrational frequencies for the first excited-states of **10** and **11**. This allowed to compute vibronic couplings within the FC approximation, with the FC Classes code of Santoro and co-workers,^{S5} and hence to determine more

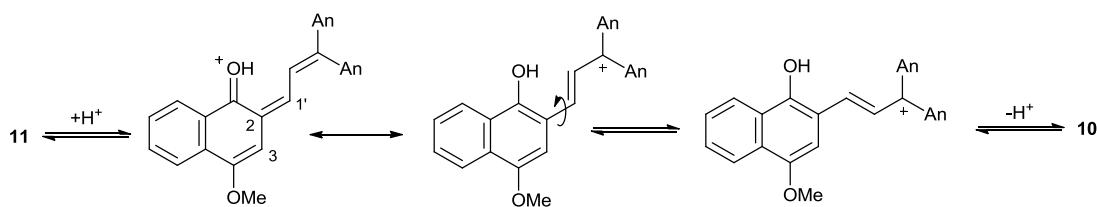
rigorously the relative absorption intensities of the two species. During the FC Classes calculations, the maximal number of integrals was increased to 10^{12} , the other parameters being kept to their default values.

References

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- S4 J. Tomasi, B. Mennucci, R. Cammi, *Chem. Rev.* 2005, **105**, 2999.
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A comment on the acid-mediated isomerization of **11** → **10**

Protonation of the carbonyl group in **11** induces polarization in the dienone system. Concomitant reduction in the C1'-C2 bond order facilitates rotational isomerization to the less hindered (*E*)-isomer **10** (Scheme ESI-1)



Scheme ESI-1

Characterisation data for the photochromic response of pyran **9**

Table ESI-1

	λ_{\max} (nm) ^a	A_o ^b	A_t ^c	Δ_{OD} ^d	K (min ⁻¹) ^e	$t_{1/2}$ (min) ^f
9	496	0.033	0.647	0.614	0.0164	42.3

Footnotes: ^a λ_{\max} = wavelength of maximum absorption after 2700 s of UV-irradiation in acetone solution at 20 °C; ^b A_o = absorbance before UV-irradiation; ^c A_t = absorbance post UV-irradiation for 2700 s; ^d Δ_{OD} = difference in absorbance (Optical Density) between A_o and A_t ; ^e K = fading rate constant at 20 °C; ^f $t_{1/2}$ = time taken for the absorbance value to fall to half of A_t .

The above data is presented in accord with such characterisation data reported by other authors for the characterisation of the photochromic response of diaryl substituted naphthopyrans.¹

- See for example: (a) C. M. Sousa, J. Berthet, S. Delbaere, P. J. Coelho, *J. Org. Chem.*, 2013, **78**, 6956; (b) C. M. Sousa, J. Berthet, S. Delbaere, P. J. Coelho, *J. Org. Chem.*, 2012, **77**, 3959; (c) K. Guo, Y. Chen, *J. Mater. Chem.*, 2010, **20**, 4193; (d) N. Malic, J. A. Campbell, R. A. Evans, *Macromolecules*, 2008, **41**, 1206; (e) W. Zhao, E. M. Carreira, *Org. Lett.*, 2006, **8**, 99; (f) M. Zayat, D. Levy, *J. Mater. Chem.*, 2003, **13**, 727.

Experimental method for the preparation of photochromic naphthopyrans

A solution of 4-methoxynaphthol (1.0g, 5.75 mmol) and the requisite 1,1-bis(4-methoxyphenyl)prop-2-yn-1-ol (1 mole equiv.), in the presence of pyridinium *p*-toluenesulfonate (PPTS) (75 mg, 0.30 mmol) and trimethyl orthoformate (1.22 g, 11.5 mmol) in 1,2-dichloroethane (50 mL), was heated under reflux for up to 4 h (the reaction time being determined by TLC examination of the reaction mixture). The cooled solvent was removed under vacuum to afford a brown gum that was either purified by flash column chromatography or by crystallisation. The following compounds were obtained in this manner:

6-Methoxy-2,2-bis(4-methoxyphenyl)-2*H*-naphtho[1,2-*b*]pyran 9 from 4-methoxy-1-naphthol and 1,1-bis(4-methoxyphenyl)prop-2-yn-1-ol (fraction 1) as a pale pink ‘fluffy’ solid (894 mg, 36.7 %) after column chromatography using 20 % EtOAc in hexane, mp = 116 – 118 °C; ν_{max} 3001, 2962, 2836, 1607, 1509, 1458, 1388, 1271, 1248, 1215, 1175, 1161, 1101, 1059, 1031, 989, 971, 828, 662, 555 cm⁻¹; δ_{H} (d_6 -acetone) 3.69 (6H, s, OMe), 3.90 (3H, s, OMe), 6.31 (1H, d, J = 9.6 Hz, 3-H), 6.69 (1H, s, 5-H), 6.77 (1H, d, J = 9.6 Hz, 4-H), 6.84 (4H, d, J = 8.6 Hz, Ar-H), 7.45 (5H, m, Ar-H), 7.54 (1H, m, Ar-H), 8.13 (1H, d, J = 8.3 Hz, 7-H), 8.33 (1H, d, J = 8.3 Hz, 10-H); δ_{C} (d_6 -acetone) 54.60, 55.17, 82.21, 102.47, 113.32, 115.81, 121.52, 121.94, 123.92, 125.57, 125.60, 126.14, 126.33, 127.92, 128.97, 137.55, 141.31, 149.60, 159.00. (Found M+H⁺, 425.1747; C₂₈H₂₄O₄ requires M+H⁺, 425.1747). Fraction 2, **(E)-2-(3',3'-bis(4-methoxyphenyl)allylidene)-4-methoxynaphthalen-1(2*H*)-one 10** as deep maroon, lustrous microcrystals (290 mg, 11.9 %), mp = 174 – 176 °C; ν_{max} 2932, 2834, 1641, 1602, 1589, 1527, 1505, 1373, 1287, 1241, 1176, 1021, 908, 833, 770, 700, 560 cm⁻¹; δ_{H} (d_6 -acetone) 3.84 (3H, s, OMe), 3.90 (3H, s, OMe), 4.02 (3H, s, 4-OMe), 6.82 (1H, s, 3-H), 6.94 (2H, d, J = 8.8 Hz, Ar-H), 7.05 (2H, d, J = 8.6 Hz, Ar-H), 7.21 (2H, d, J = 8.6 Hz, Ar-H), 7.38 (3H, m, Ar-H, 2'-H), 7.48 (1H, m, Ar-H), 7.55 (1H, d, J = 12.7 Hz, 1'-H), 7.68 (1H, m, Ar-H), 7.85 (1H, d, J = 7.9 Hz, 5-H), 8.10 (1H, d, J = 7.7 Hz, 8-H); δ_{H} (CDCl₃) 3.85 (3H, s, OMe), 3.88 (3H, s, OMe), 3.97 (3H, s, 4-OMe), 6.45 (1H, s, 3-H), 6.90 (2H, d, J = 8.5 Hz, Ar-H), 6.96 (2H, d, J = 8.4 Hz, Ar-H), 7.11 (1H, d, J = 12.6 Hz, 2'-H), 7.22 (2H, d, J = 8.4 Hz, Ar-H), 7.36 (2H, d, J = 8.5 Hz, Ar-H), 7.41 (1H, m, Ar-H), 7.61 (1H, m, Ar-H), 7.66 (1H, d, J = 12.6 Hz, 1'-H), 7.85 (1H, d, J = 7.9 Hz, 5-H), 8.21 (1H, d, J = 7.7 Hz, 8-H); δ_{C} (d_6 -acetone) 54.80, 54.85, 55.14, 97.17, 113.72, 113.81, 120.85, 122.56, 126.91, 128.18, 130.02, 130.04, 131.38, 131.72, 132.30, 133.03, 134.04, 134.60, 134.90, 150.31, 152.36, 160.19, 160.69, 183.01. (Found M+H⁺, 425.1747; C₂₈H₂₄O₄ requires M+H⁺, 425.1747).

6-Methoxy-2,2-bis(4-methoxyphenyl)-4-(4-methylphenyl)-2*H*-naphtho[1,2-*b*]pyran 12 from 4-methoxy-1-naphthol and 1,1-bis(4-methoxyphenyl)-3-(4-methylphenyl)prop-2-yn-1-ol (fraction 1) as pale fawn fluffy solid, (2.04 g, 69.2 %) after crystallisation twice from hexane and EtOAc, mp = 197 – 199 °C; ν_{max} 2980, 1607, 1508, 1350, 1249, 1173, 1037, 877, 827, 770, 696 cm⁻¹; δ_{H} (CDCl₃) 2.43 (3H, s, tolyl-Me), 3.76 (6H, s, OMe), 3.77 (3H, s, OMe), 6.13 (1H, s, 3-H), 6.54 (1H, s, 5-H), 6.82 (4H, d, J = 8.3 Hz, Ar-H), 7.24 (2H, d, J = 7.8 Hz, tolyl-H), 7.42 (2H, d, J = 7.8 Hz, tolyl-H), 7.50 (6H, m, Ar-H, 8-H, 9-H), 8.12 (1H, d, J = 8.3 Hz, 7-H), 8.37 (1H, d, J = 8.3 Hz, 10-H); δ_{C} (CDCl₃) 21.31, 55.21, 55.67, 81.97, 101.72, 113.36, 116.40, 121.89, 122.17, 125.84, 125.89, 126.20, 126.23, 126.62, 128.19, 128.72, 129.12, 135.59, 136.49, 137.56, 137.74, 142.40, 149.08, 158.78. (Found M+H⁺, 515.2209; C₃₅H₃₀O₄ requires M+H⁺, 515.2217). Column chromatography (25% EtOAc in hexane) of the liquors from the first crystallisation gave fraction 1, further **12** (300 mg, 10.2 %) identical in all aspects to the foregoing material and fraction 2, **1,1-bis(4-methoxyphenyl)-3-(4-methylphenyl)prop-2-enone 13** as deep yellow micro-crystals (180 mg, 8.7 %), mp = 93 – 95 °C; ν_{max} 2933, 2836, 1654, 1603, 1506, 1461, 1289, 1244, 1172, 1112, 1029, 827, 753, 575, 549 cm⁻¹; δ_{H} (CDCl₃) 2.39 (3H, s, Me), 3.80 (3H, s, OMe), 3.85 (3H, s, OMe), 6.81 (2H, d, J = 8.3 Hz, Ar-H), 6.89 (2H, d, J = 8.4 Hz, Ar-H), 7.01 (1H, s, 2-H), 7.13 (2H, d, J = 8.3 Hz, Ar-H), 7.19 (2H, d, J = 7.8 Hz, tolyl-H), 7.34 (2H, d, J = 8.4 Hz, Ar-H), 7.84 (2H, d, J = 7.8 Hz, tolyl-H); δ_{C} (CDCl₃) 21.67, 55.20, 55.41, 113.44, 113.79, 121.63, 128.84, 129.09, 130.29, 131.43, 131.59, 134.41, 136.23, 143.21, 154.51, 159.79, 160.72, 192.09. (Found [M+H]⁺ = 359.1636 C₂₄H₂₂O₃ requires [M+H]⁺ = 359.1642).

Preparation of 1,1-bis(4-methoxyphenyl)-3-(4-methylphenyl)prop-2-yn-1-ol **8b**

n-BuLi (2.5 M in hexanes, 41.4 mmol, 16.6 mL) was added slowly to a cold (0 °C) stirred solution of 4-ethynyltoluene (39.4 mmol, 4.58 g) in anhydrous THF (150 mL) under nitrogen. Upon completion of the addition, the cold solution was stirred for 30 min and then 4,4-dimethoxybenzophenone (35.8 mmol, 8.68 g) was added in a single portion and the mixture stirred until none of the benzophenone could be detected by TLC examination of the reaction mixture (~2h). The resulting mixture was poured into water (250 mL) containing brine (30 mL) and the organic layer was separated and the aqueous layer extracted with EtOAc (3 × 50 mL). The organic extracts were combined and washed with water (2 × 100 mL) and dried over anhyd. Na₂SO₄. Removal of the solvent afforded the *1,1-bis(4-methoxyphenyl)-3-(4-methylphenyl)prop-2-yn-1-ol* **8b** (12.1 g, 94 %) as a very viscous pale yellow oil, ν_{max} 3450, 2952, 2835, 1606, 1584, 1505, 1461, 1301, 1243, 1168, 1031, 986, 899, 8115, 742, 587, 526 cm⁻¹; δ_{H} (CDCl₃) 2.38 (3H, s, Me), 3.28 (1H, s, OH (D₂O exchangeable)), 3.80 (6H, s, OMe), 6.90 (4H, m, Ar-H), 7.15 (2H, d, J = 7.88 Hz, tolyl-H), 7.42 (2H, m, tolyl-H), 7.62 (4H, m, Ar-H); δ_{C} (CDCl₃) 21.56, 55.31, 74.18, 86.95, 91.65, 113.56, 119.58, 127.47, 129.13, 131.70, 137.85, 138.74, 158.98. Found [M-H₂O+H]⁺ = 341.1532 C₂₄H₂₂O₃ requires [M-H₂O+H]⁺ = 341.1536. This material was used directly for the preparation of naphthopyran **13**.

Figure S1 ^1H NMR spectrum of compound **8b** in commercial CDCl_3

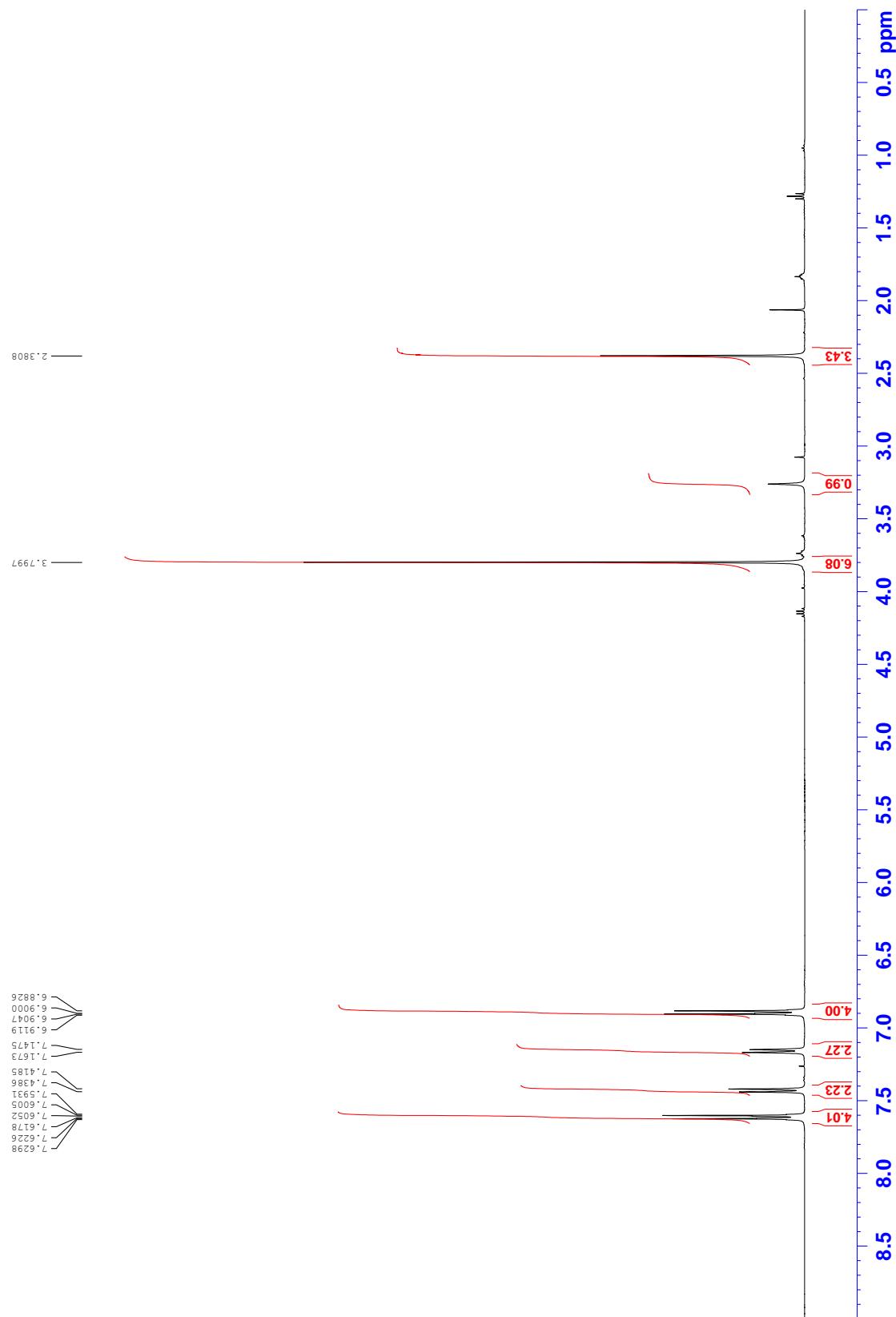


Figure S2 ^{13}C NMR spectrum of compound **8b** in commercial CDCl_3

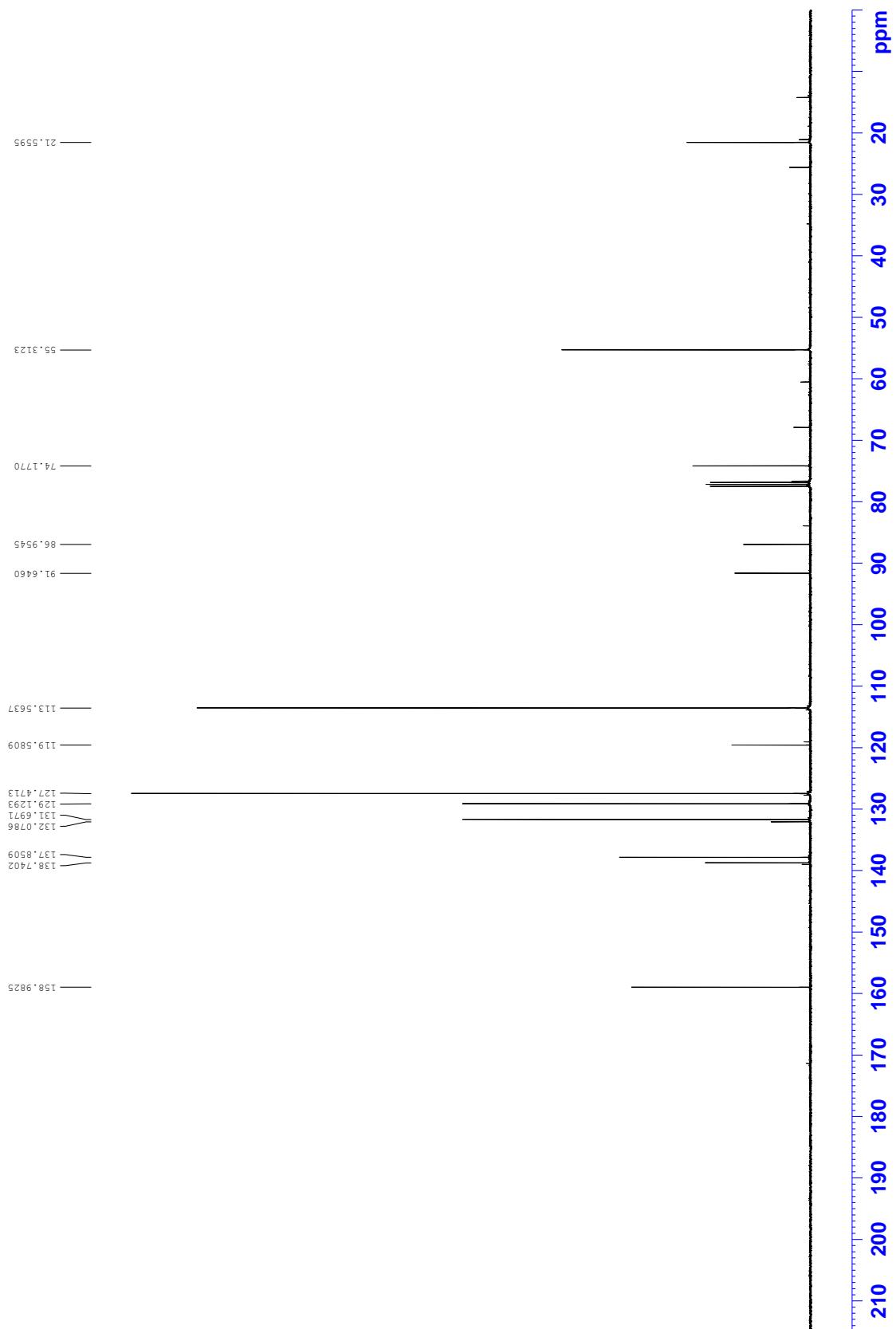
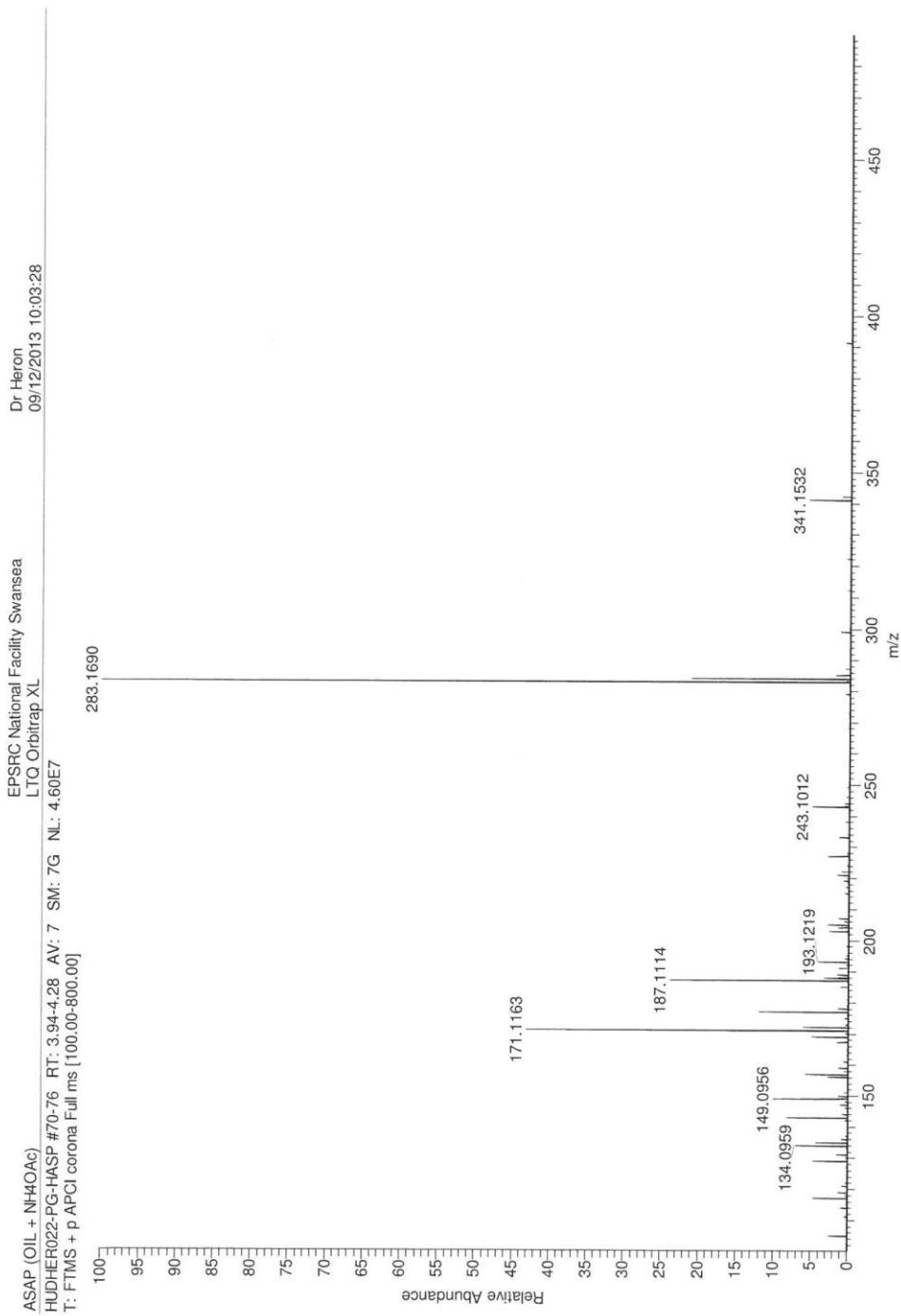


Figure S3 Mass Spectral data for compound **8b**



ASAP (OIL + NH₄OAc)

HUDHER022-PG-HASP #70-76

RT: 3.94-4.28

AV: 7

SM: 7G

NL: 1.98E7

T: FTMS + p APCI corona Full ms [100.00-800.00]

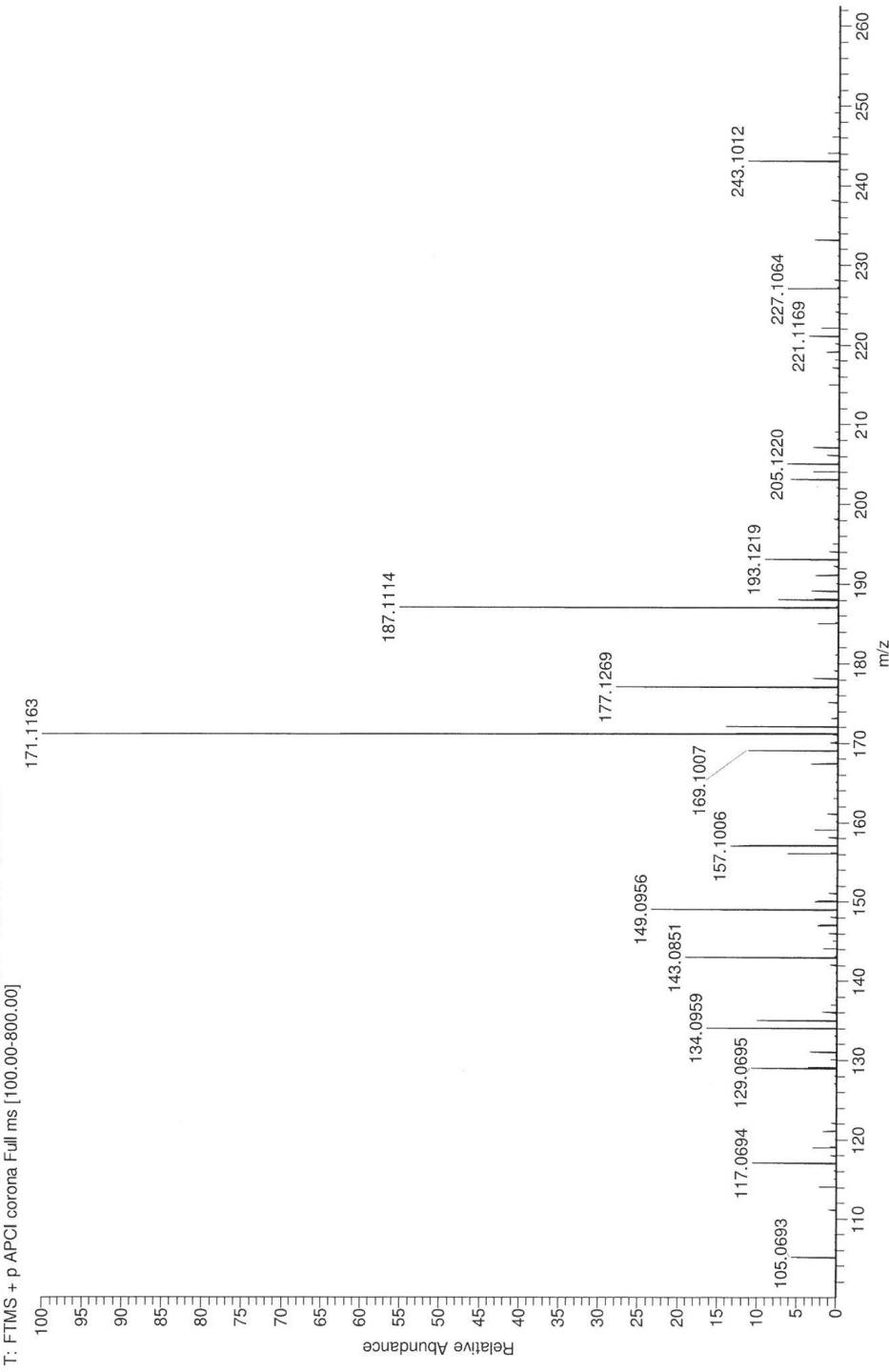
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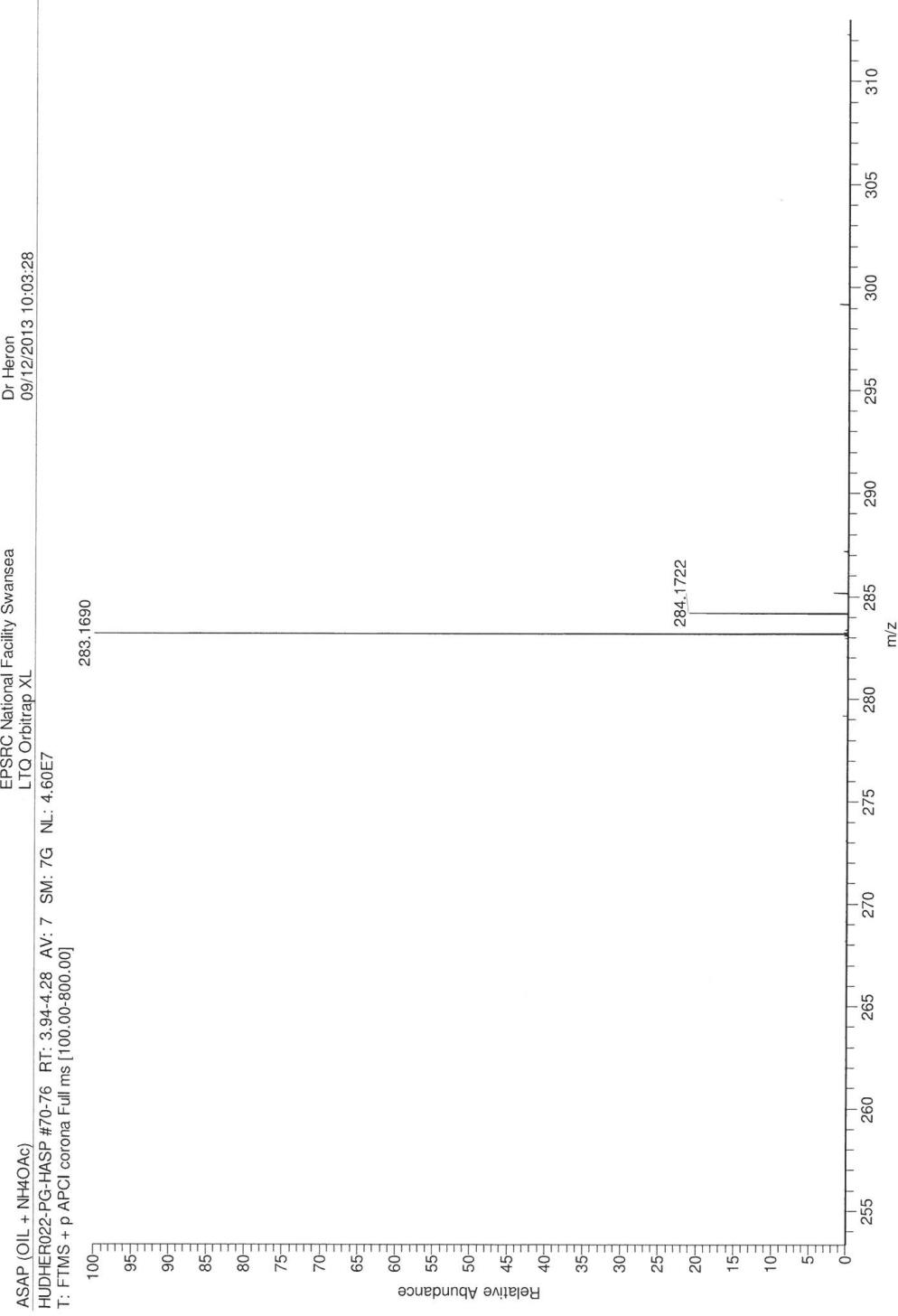
EPSRC National Facility Swansea

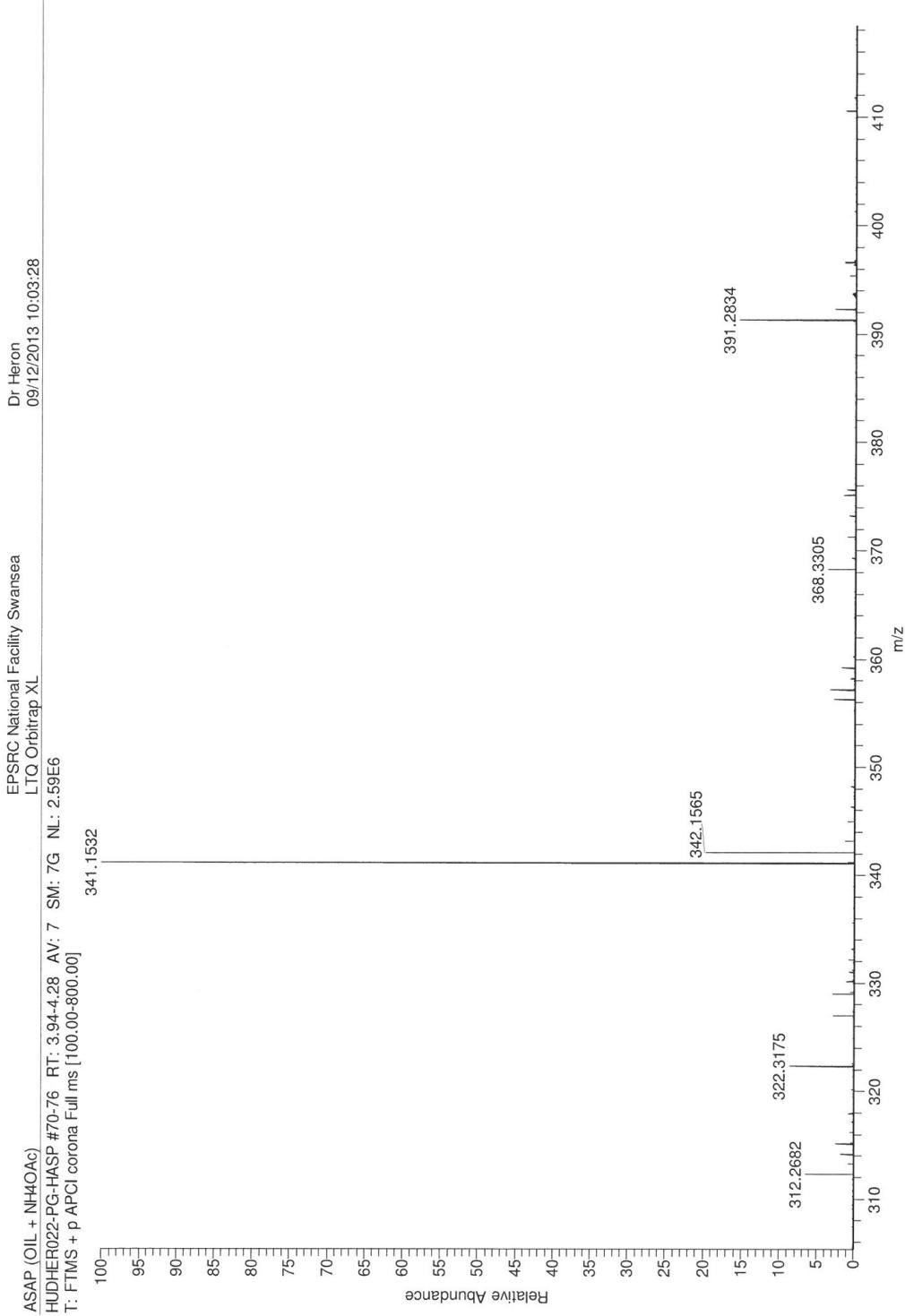
LTQ Orbitrap XL

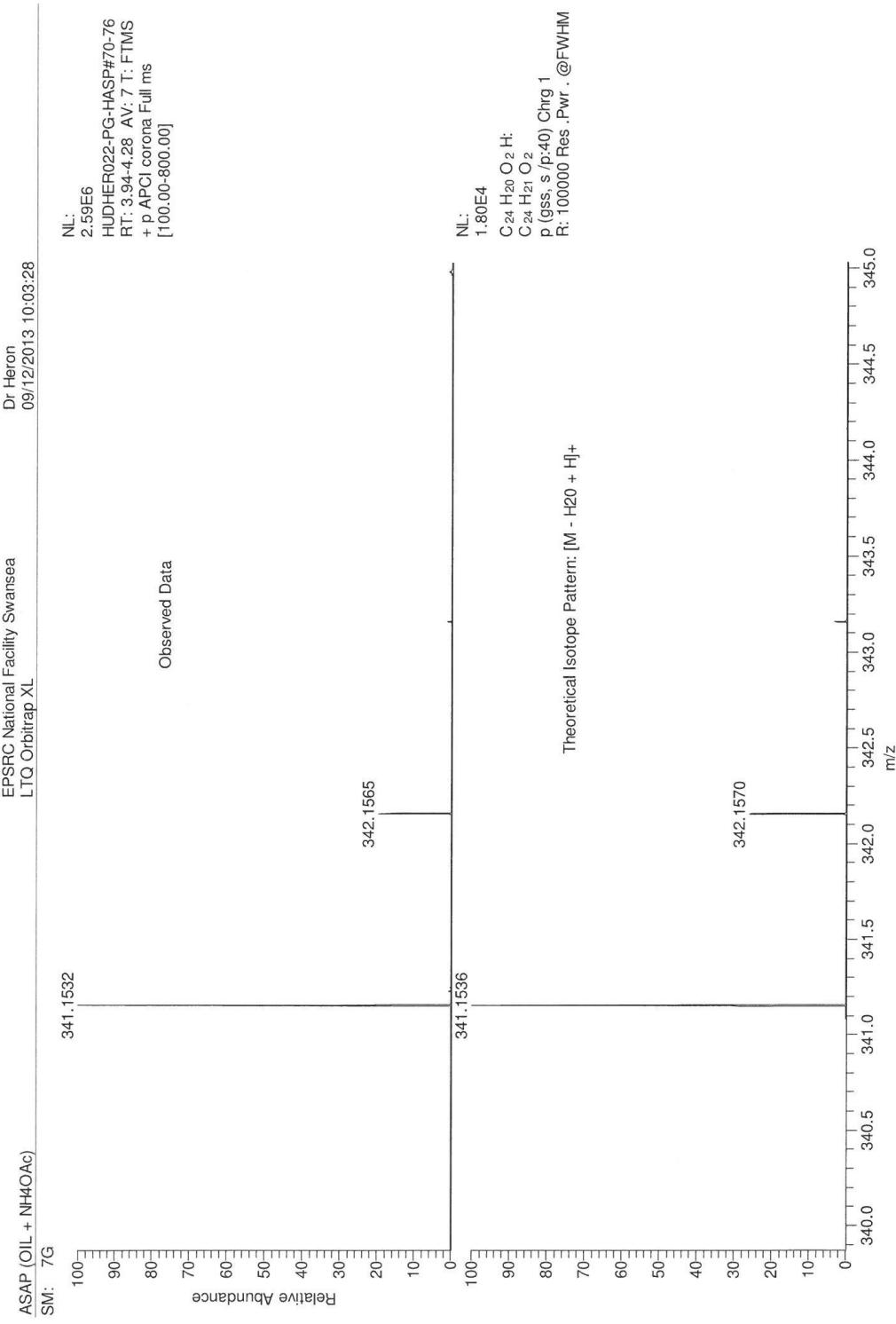
Dr Heron

09/12/2013 10:03:28









Isotope:		Min.	Max.
14 N		0.....11	
16 O		0.....10	
12 C		0.....60	
1 H		0.....70	
23 Na		0.....0	
Tolerance Window:		+/- 5.00 ppm	
Db/Ring Equiv:		-3 .. 100	
Fits:		200	
			N-Rule: Do not use
			Charge: 1
Mass	Theoretical	Delta [ppm]	RDB Composition
341.1532	341.1536	-1.2	C ₁₄ H ₂ O ₂
	341.1528	1.3	C ₈ H ₂₁ O ₂ N ₂
	341.1541	-2.7	C ₁₀ H ₂₃ O ₂ N ₂
	341.1523	2.7	C ₁₂ H ₂₉ O ₂ N ₃

Figure S4 ^1H NMR spectrum of compound **9** in d_6 -acetone

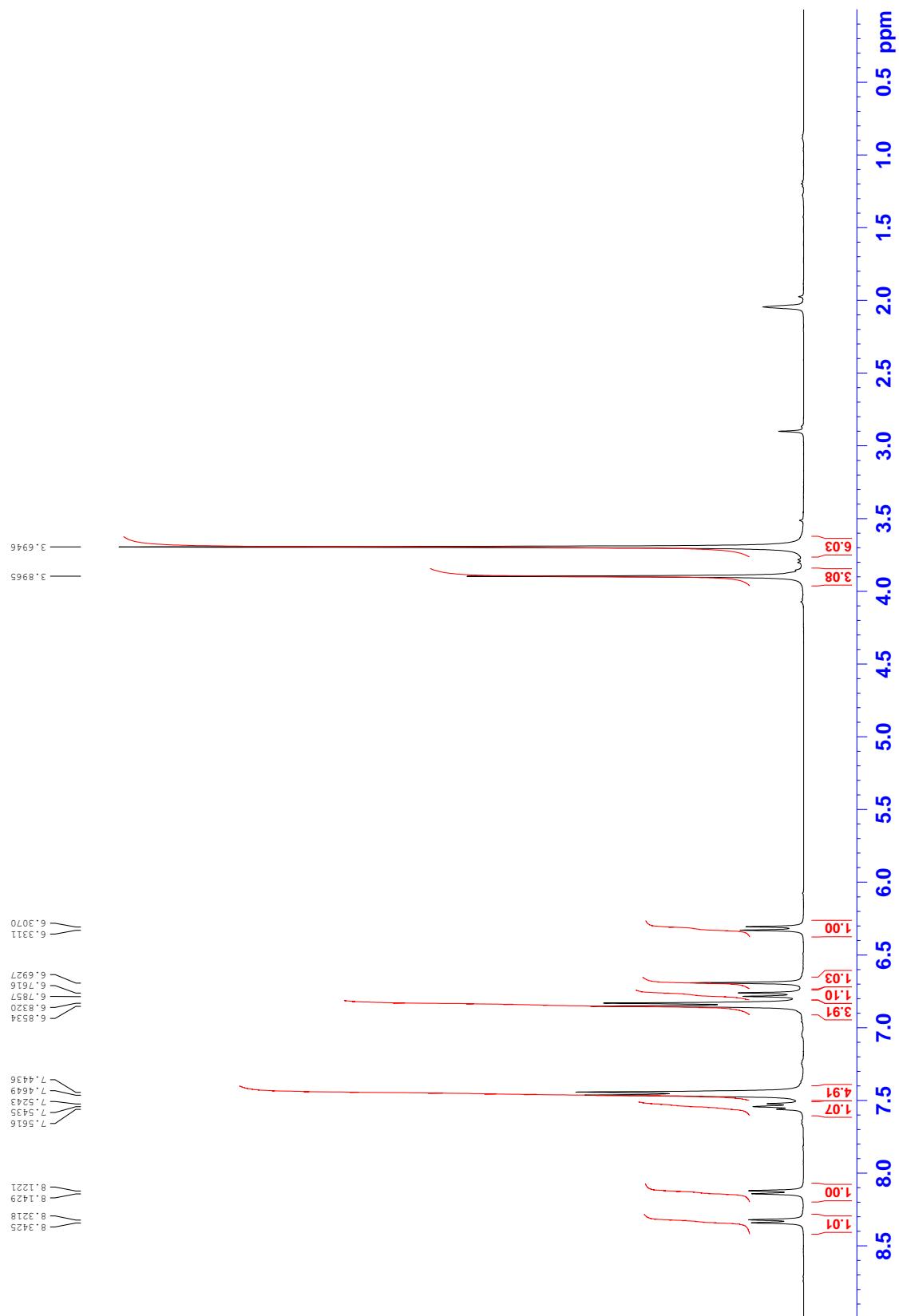


Figure S5 ^{13}C NMR spectrum of compound **9** in d_6 -acetone

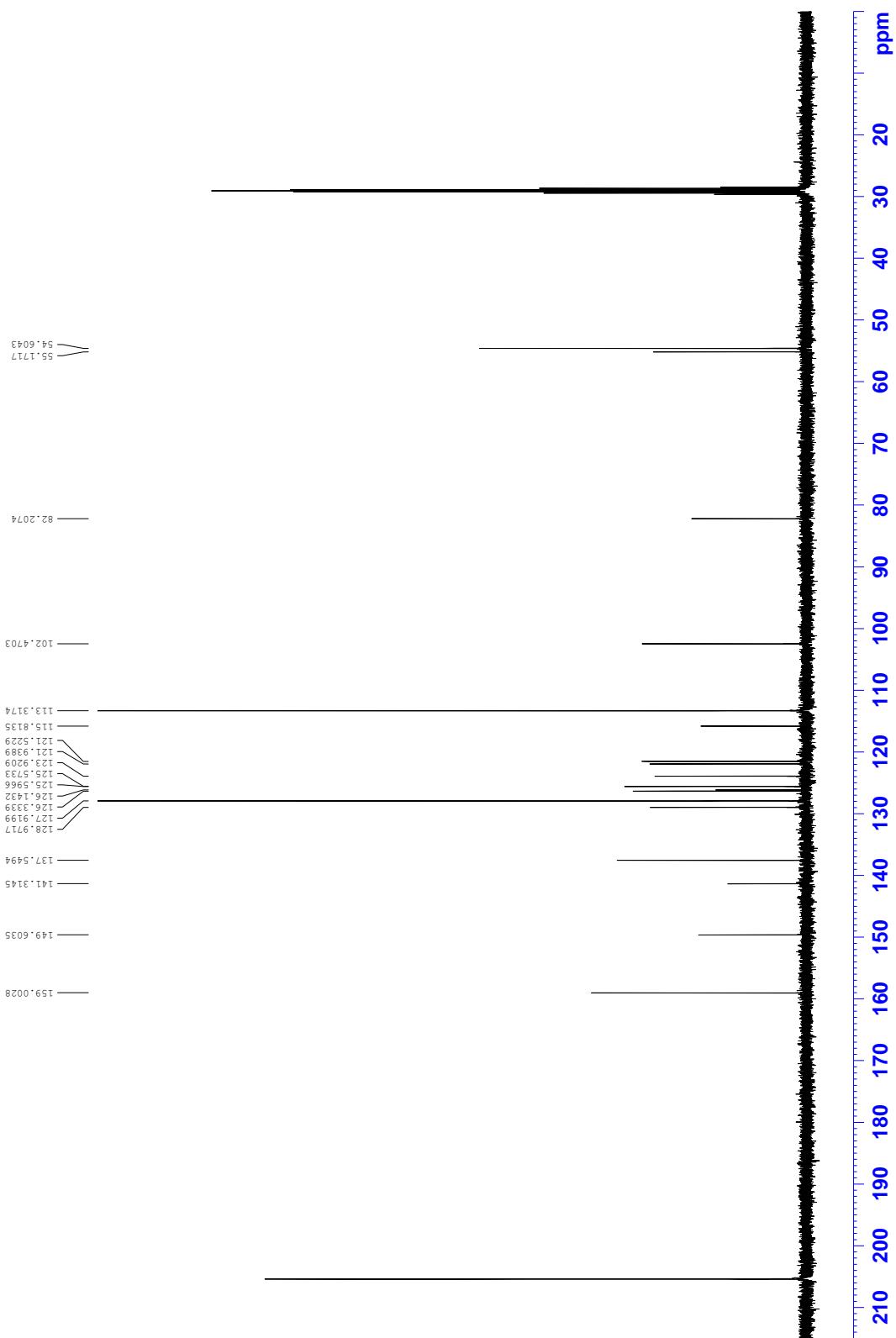


Figure S6 ^1H NMR spectrum of compound **9** in d_6 -acetone UV irradiated for 30 s

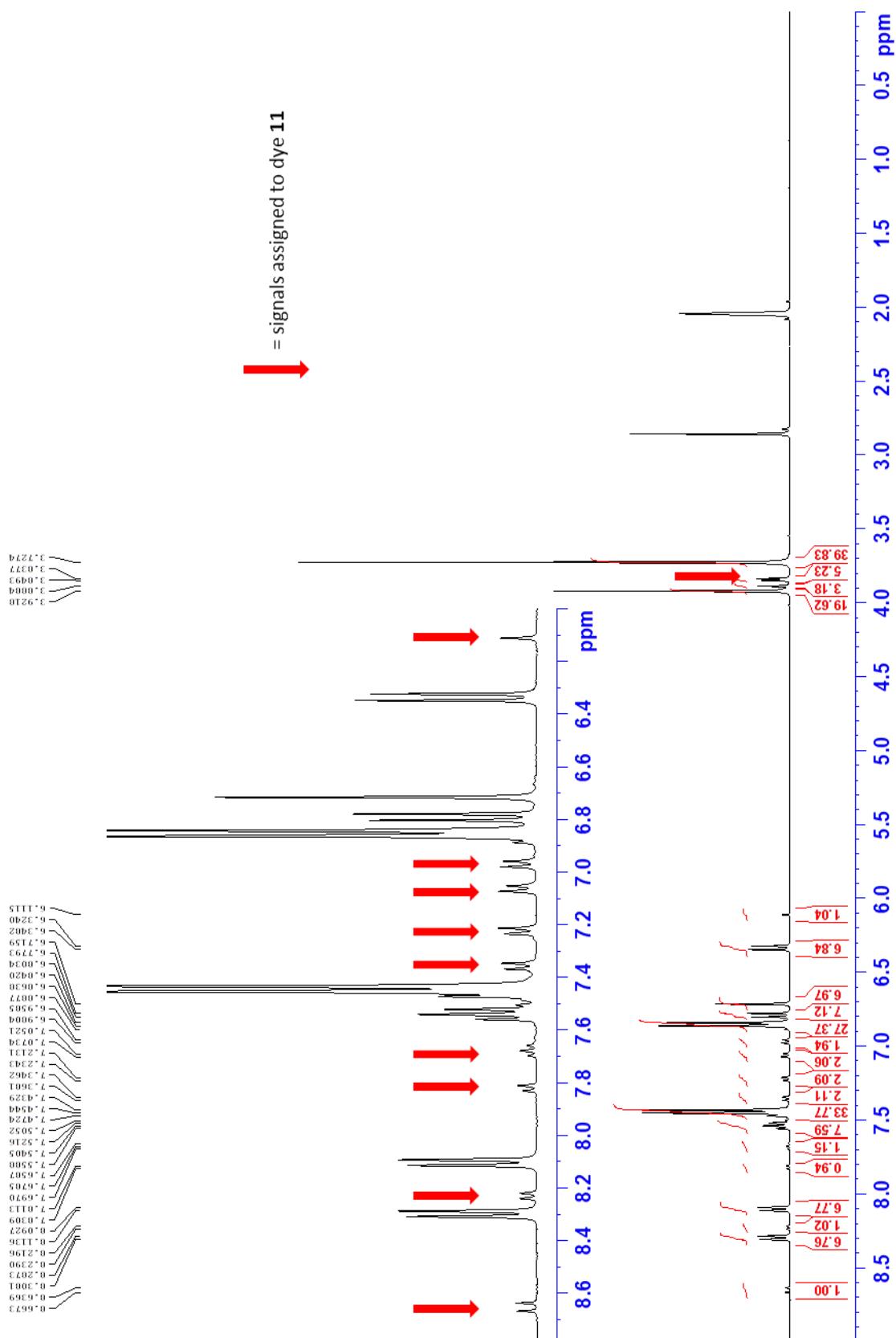


Figure S7 ^1H NMR spectrum of compound **9** in d_6 -acetone UV irradiated for 40 min

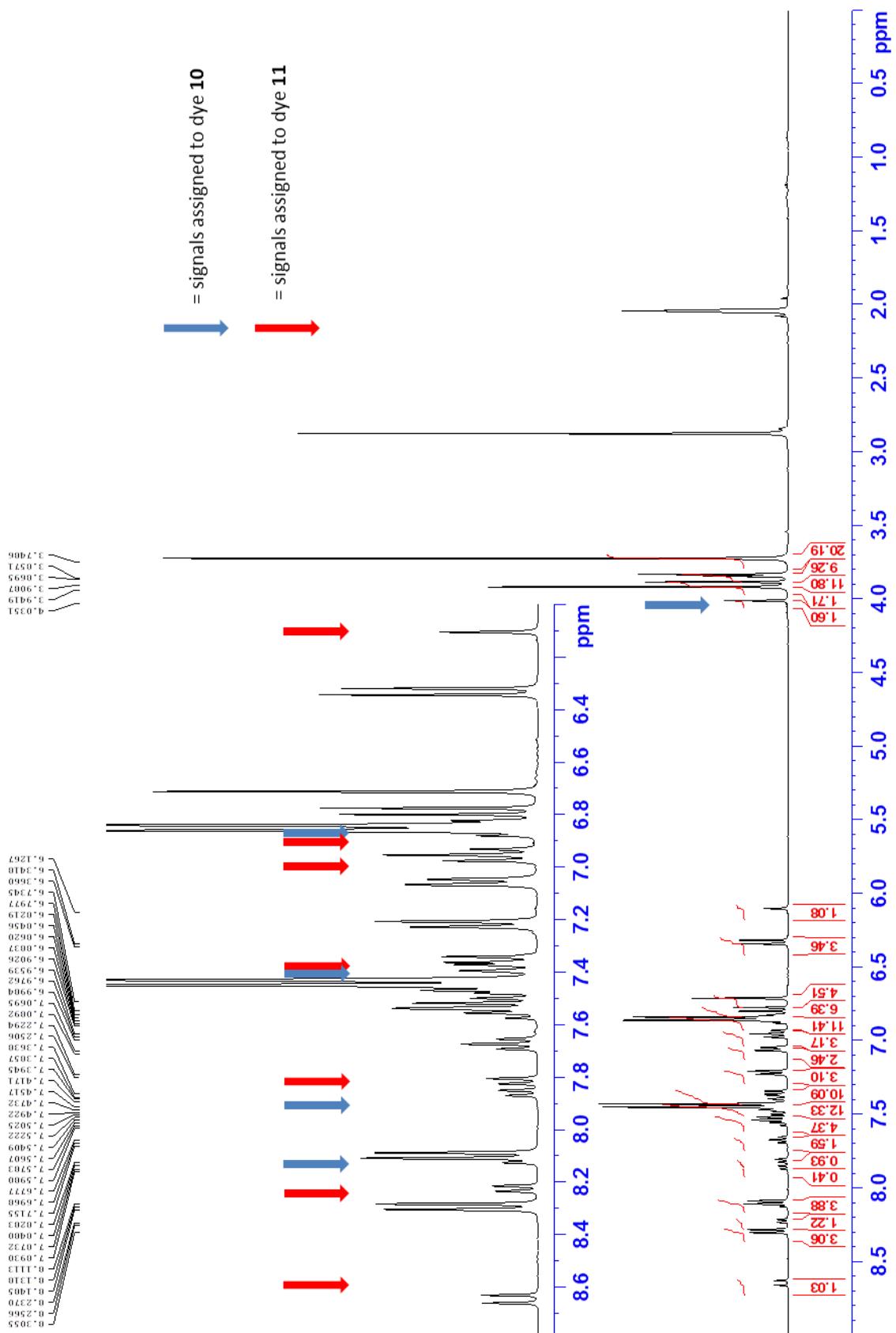


Figure S8 ^1H NMR spectrum of compound **9** in commercial CDCl_3

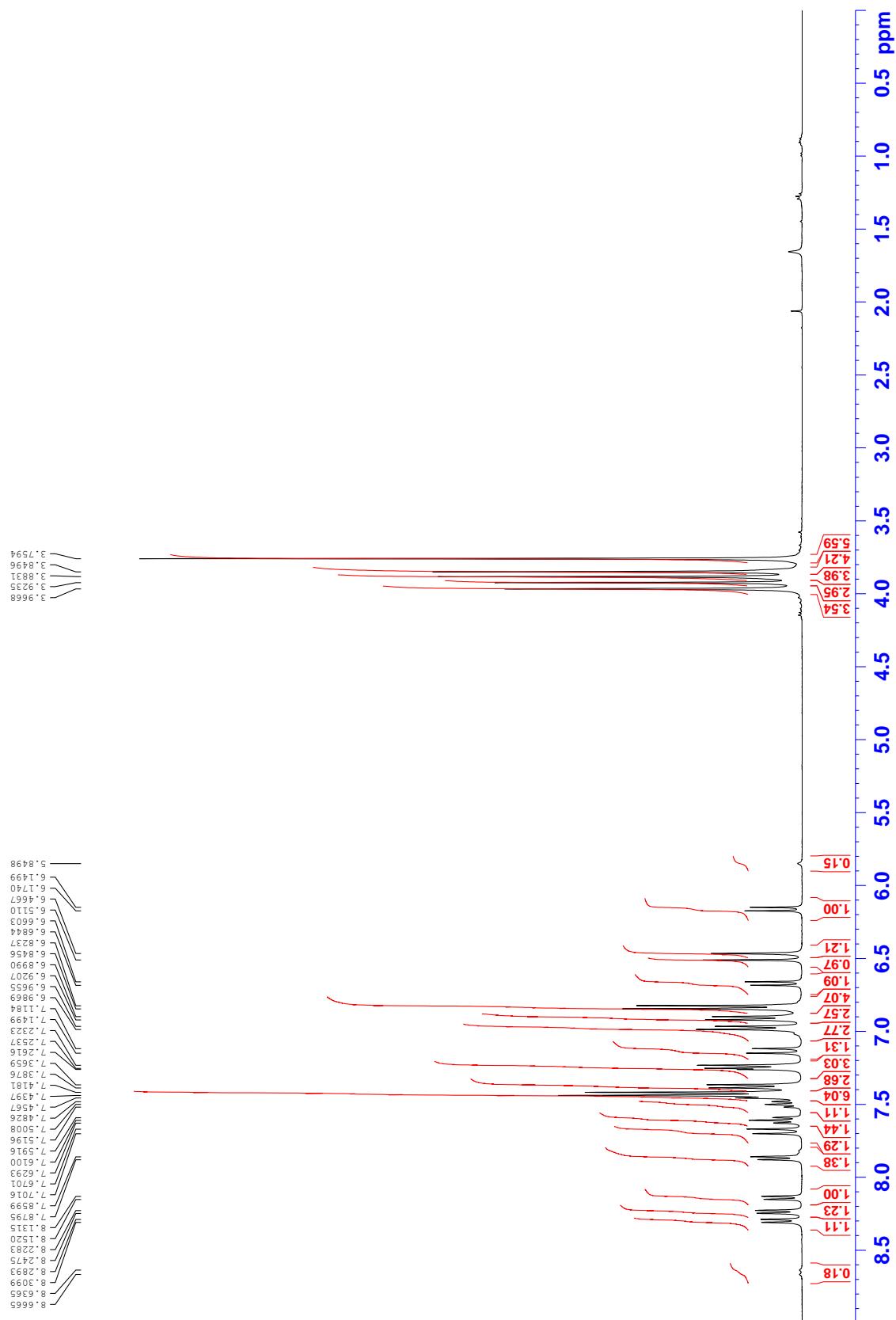


Figure S9 ^{13}C NMR spectrum of compound **9** in commercial CDCl_3

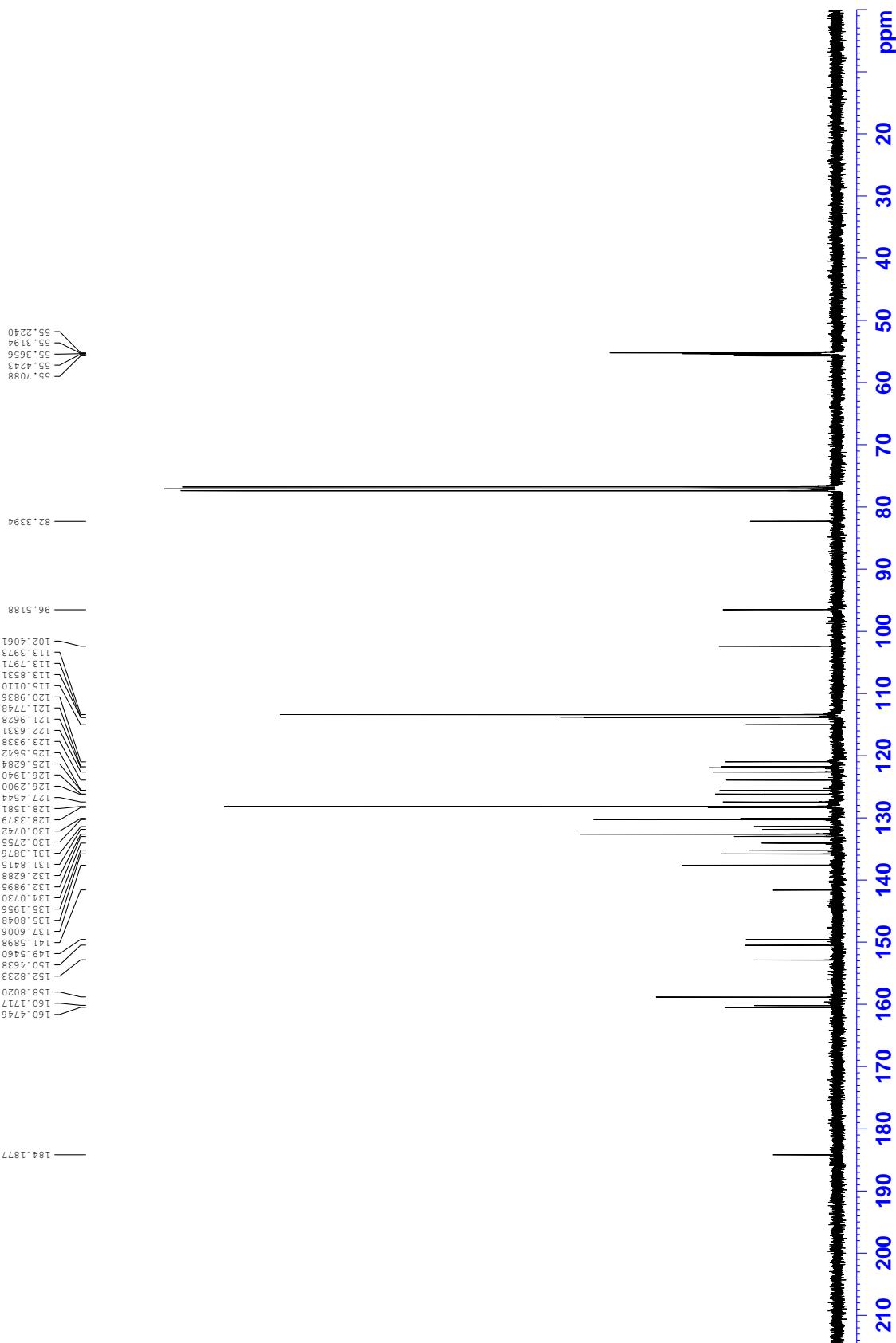


Figure S10 ^1H NMR spectrum of compound **9** in aq. K_2CO_3 washed CDCl_3

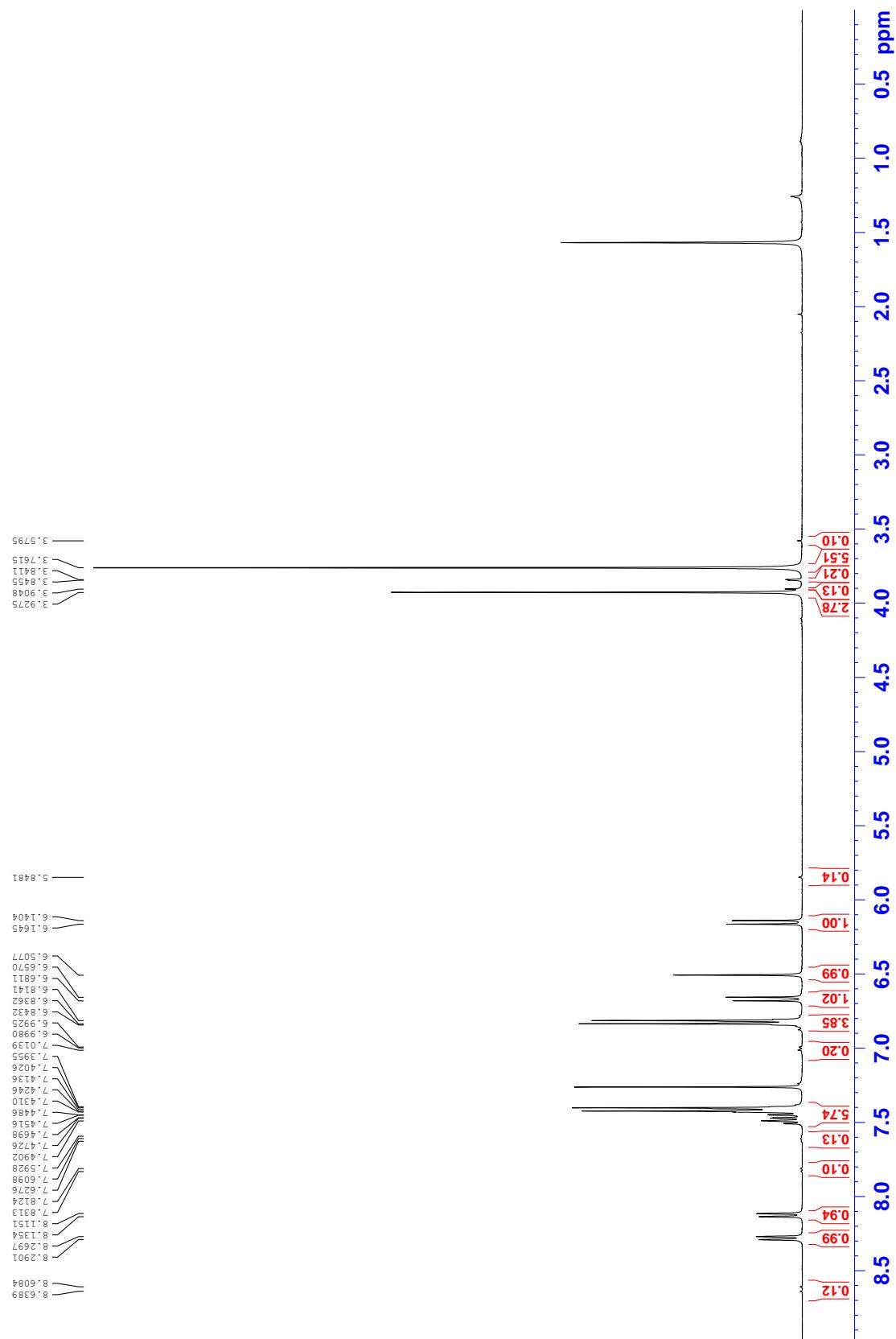


Figure S11 ^{13}C NMR spectrum of compound **9** in aq. K_2CO_3 washed CDCl_3

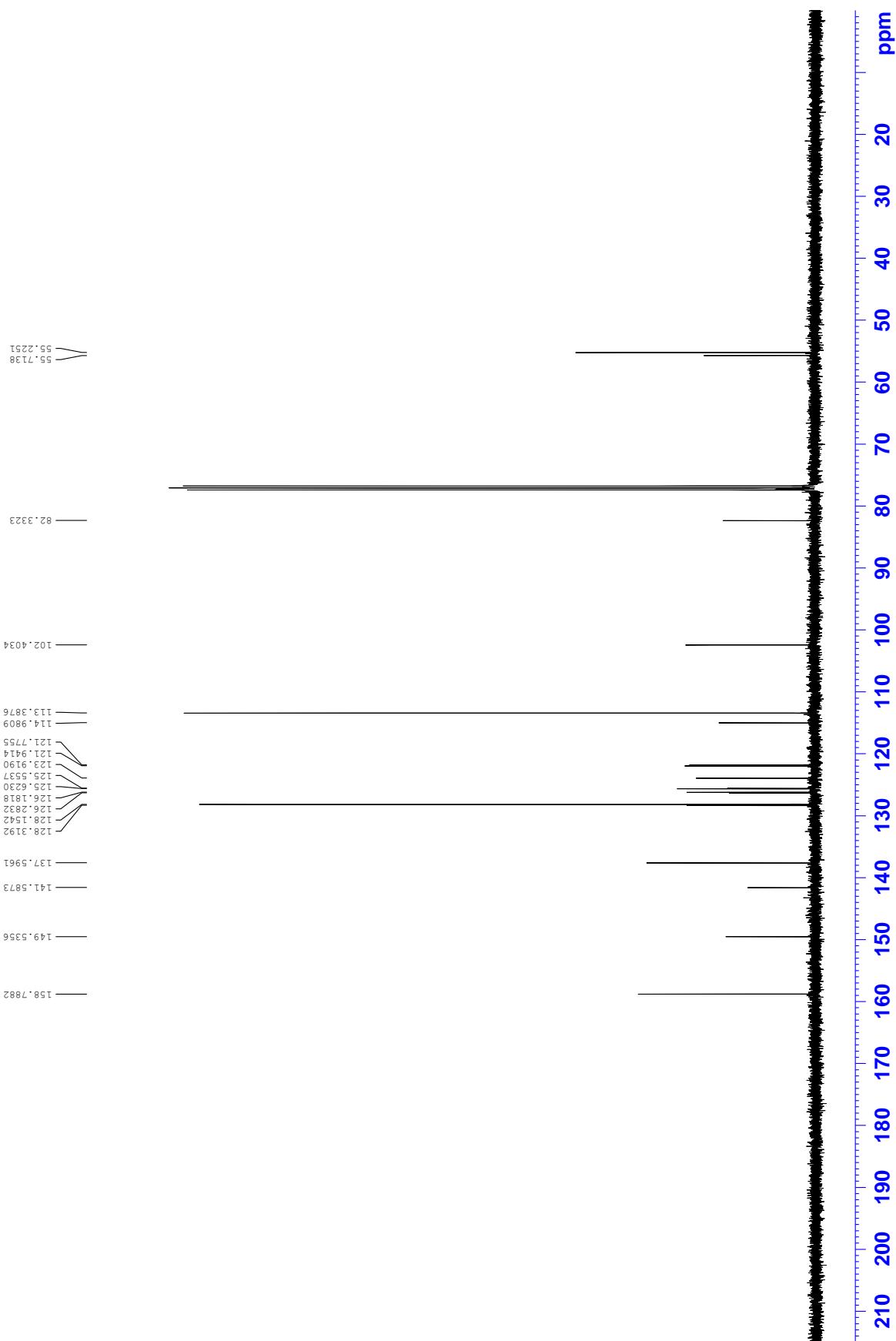
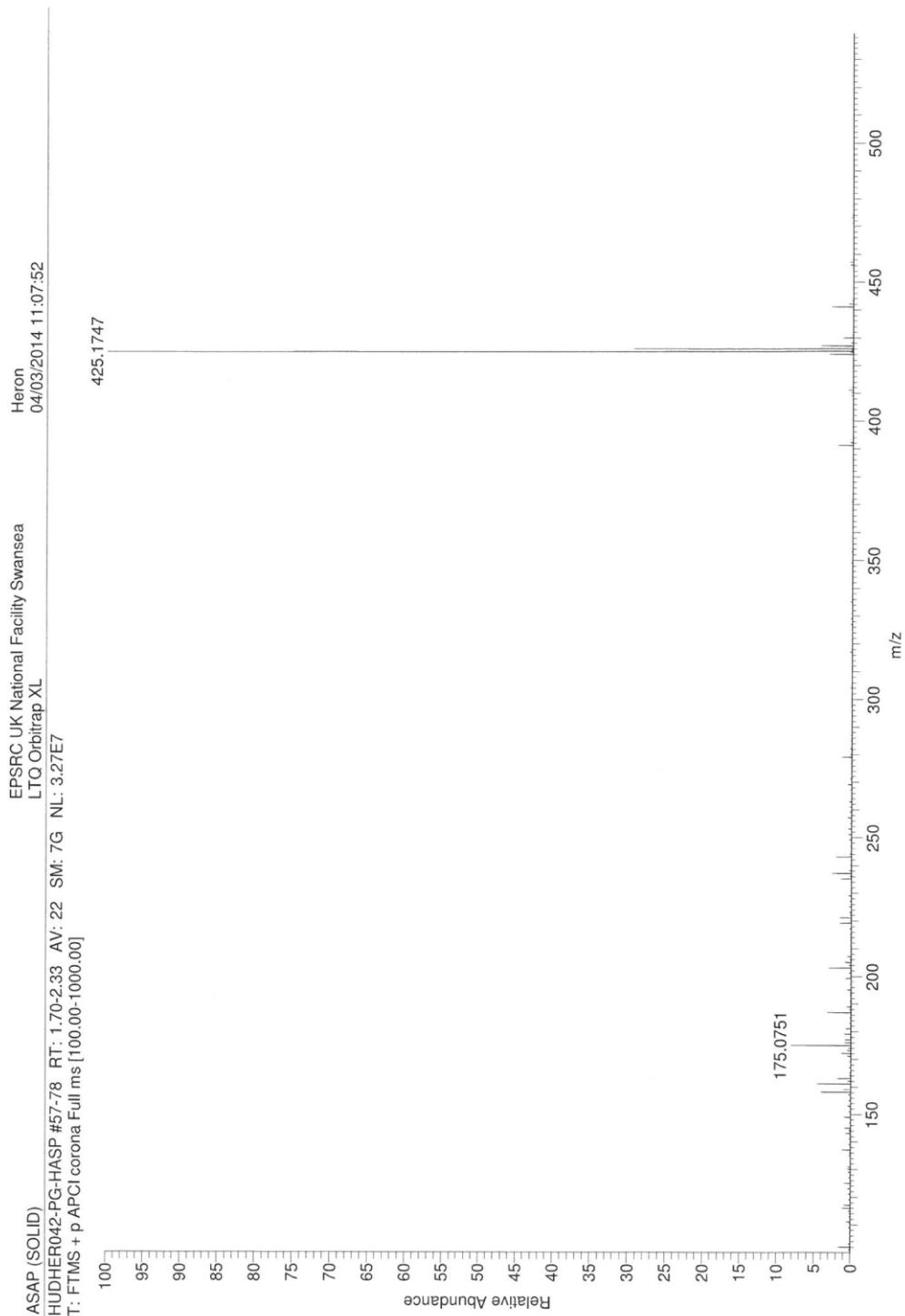
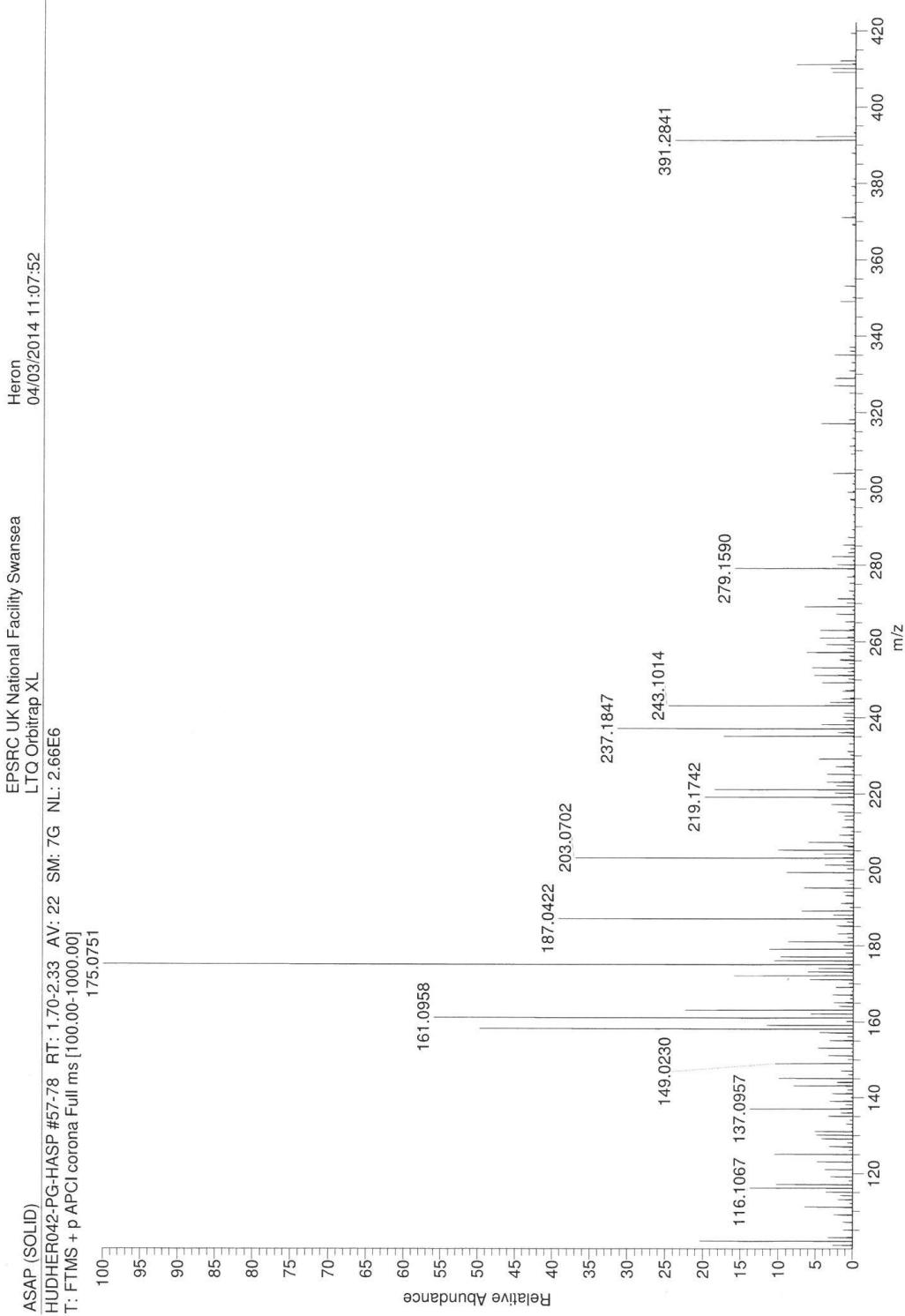
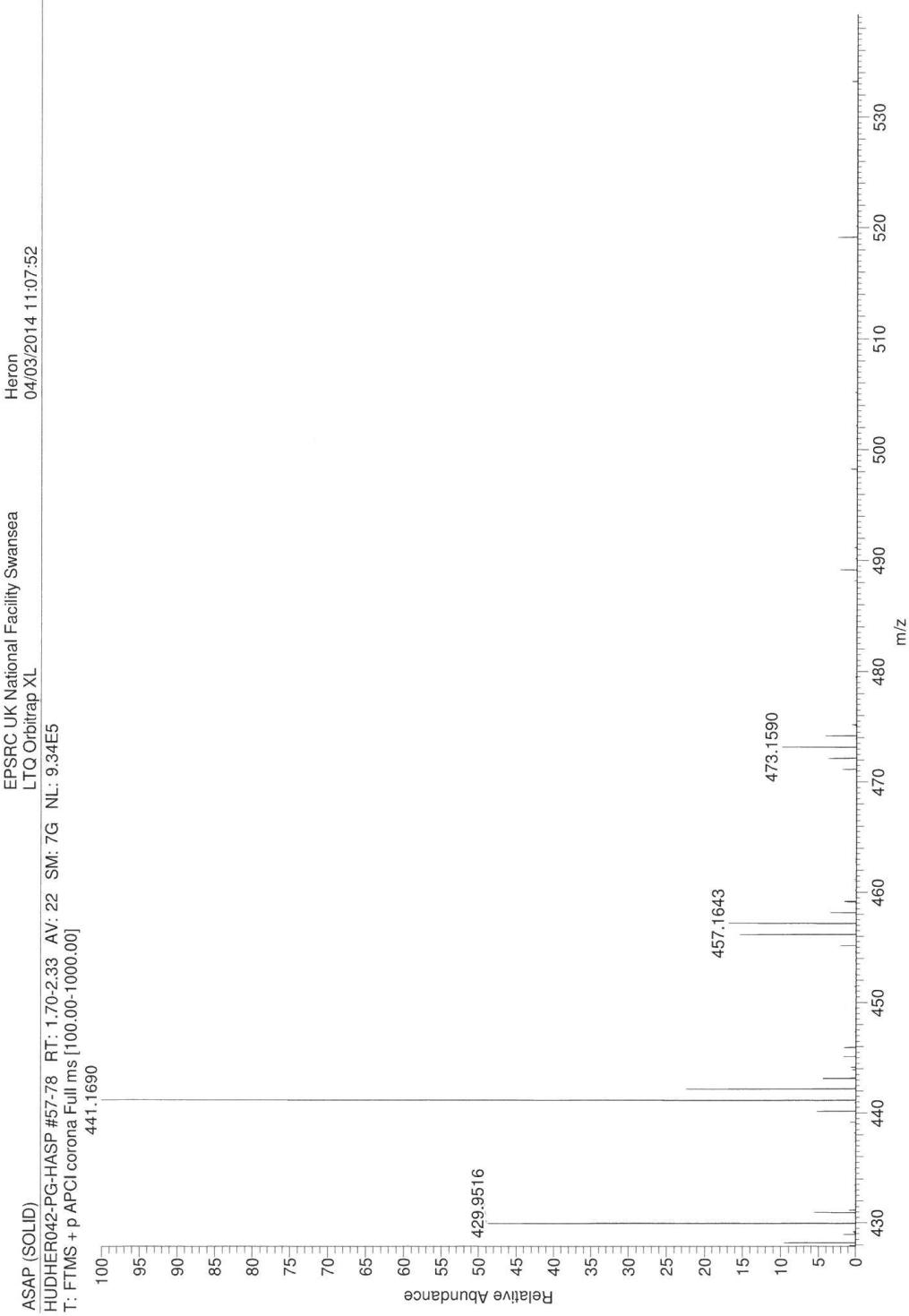
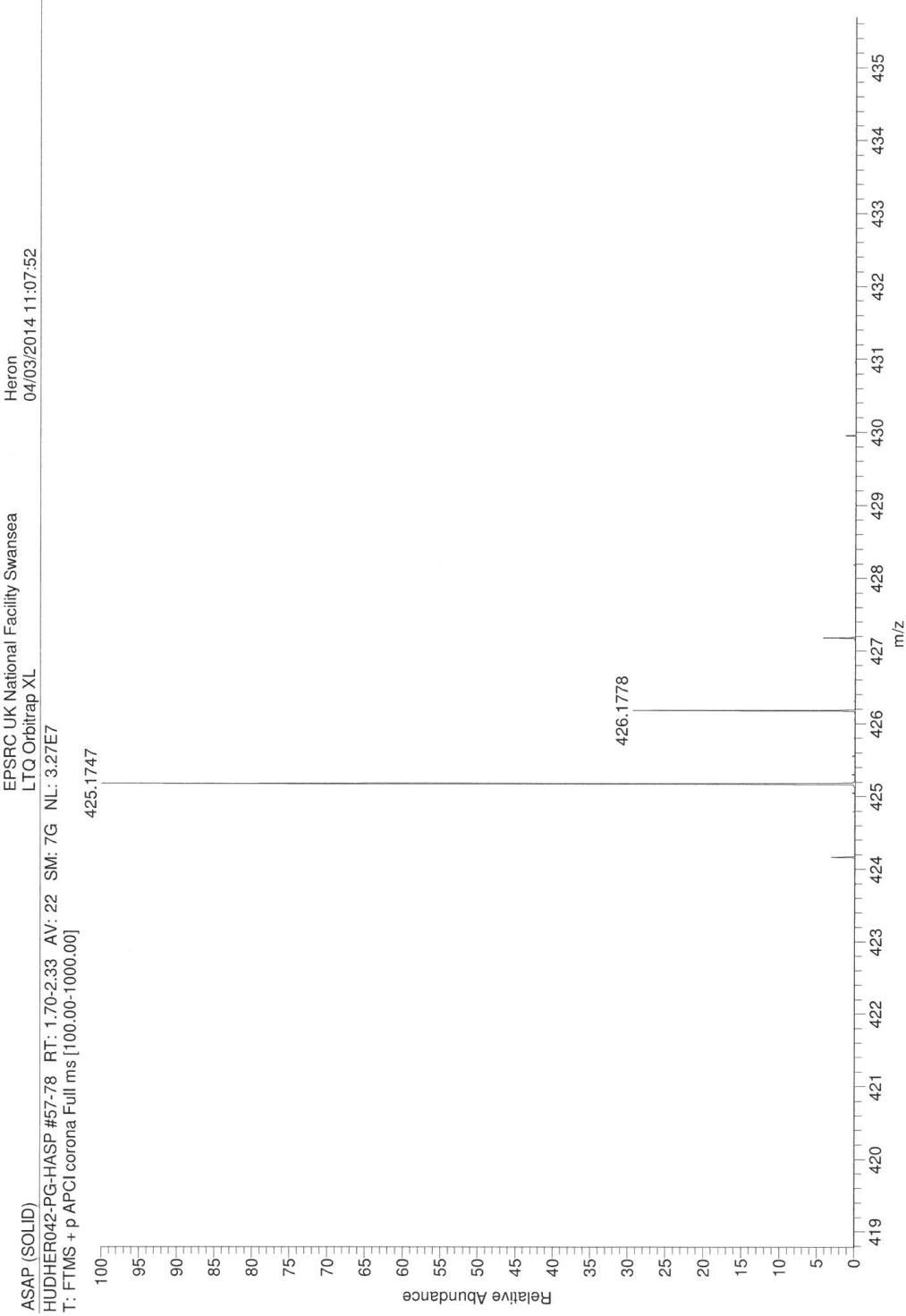


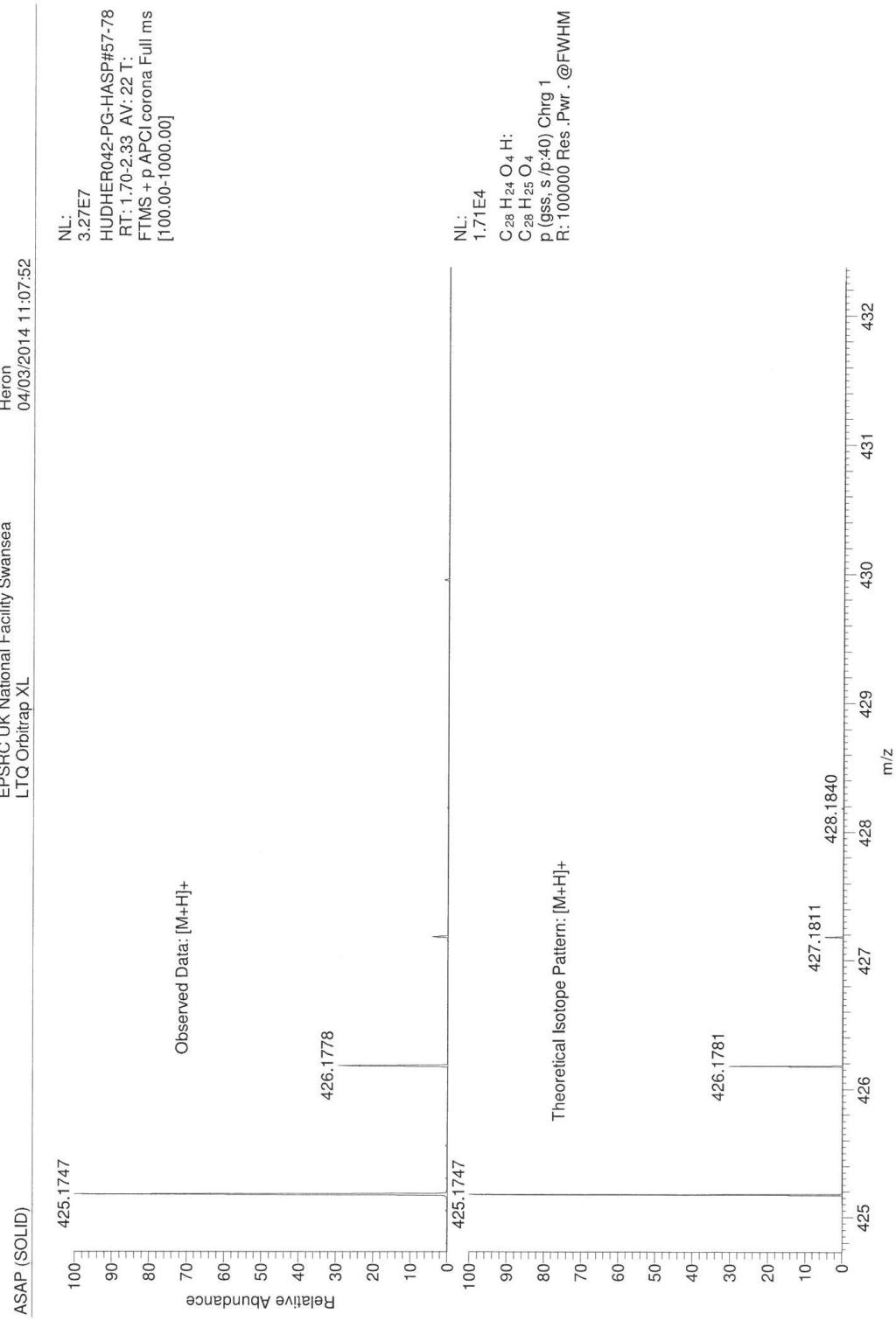
Figure S12 Mass Spectral data for compound **9**











Isotopes:		Min.	Max.
		0....15	0....15
14 N		0....15	0....15
16 O		0....15	0....15
12 C		0....60	0....60
1 H		0....70	0....70
23 Na		0....0	0....0
Tolerance Window:	+/- 5.00 PPM		
Db/Ring Equiv:	-3 . 100		
Fit:	100		
		N-Rule:	Do not use
		Charge:	1
Mass	Theoretical	Delta [PPM]	RDB
Mass			Composition
425.1747	425.1747	-0.1	16.5
425.1752	425.1752	-1.3	C ₁₈ H ₂₅ O ₂ N ₁₂
425.1752		-1.3	C ₁₃ H ₂₁ O ₂ N ₅
425.1739	425.1739	1.9	C ₁₄ H ₂₇ O ₁₀ N ₃
425.1739		1.9	C ₁₃ H ₂₁ O ₄ N ₁
425.1739		1.9	C ₁₂ H ₂₅ O ₁ N ²
425.1739		1.9	C ₁₁ H ₁₉ O ₁ N ₁₅
425.1734	425.1734	3.1	C ₁₆ H ₃₁ O ₁ N ₃
425.1761	425.1761	-3.2	C ₁₉ H ₂₁ N ₃
425.1766	425.1766	-4.4	C ₁₅ H ₂₃ O ₉ N ₉
425.1766		-4.4	C ₁₆ H ₂₉ O ₁ N ₂

Figure S13 ^1H NMR spectrum of compound **10** in d_6 -acetone

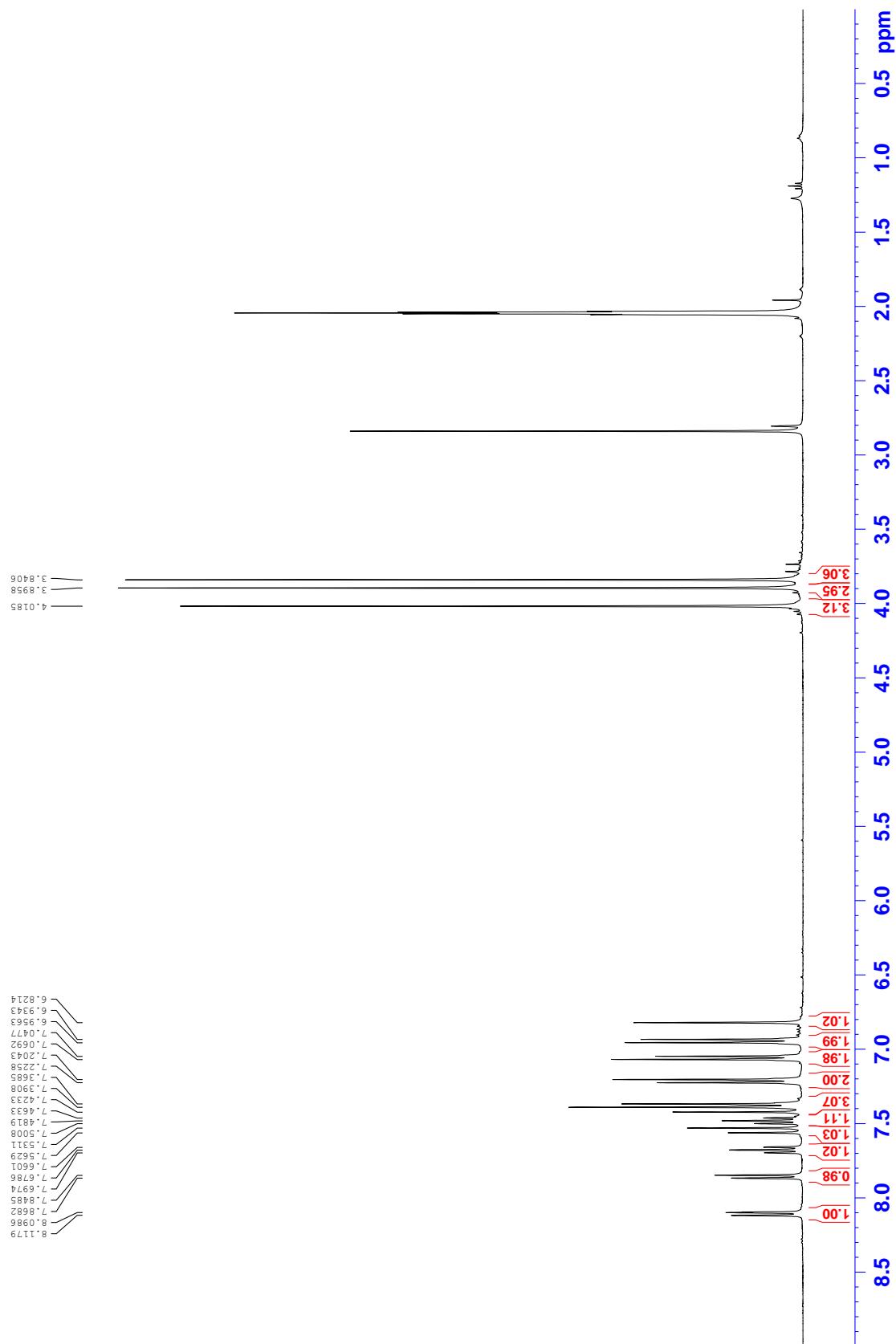


Figure S14 ^{13}C NMR spectrum of compound **10** in d_6 -acetone

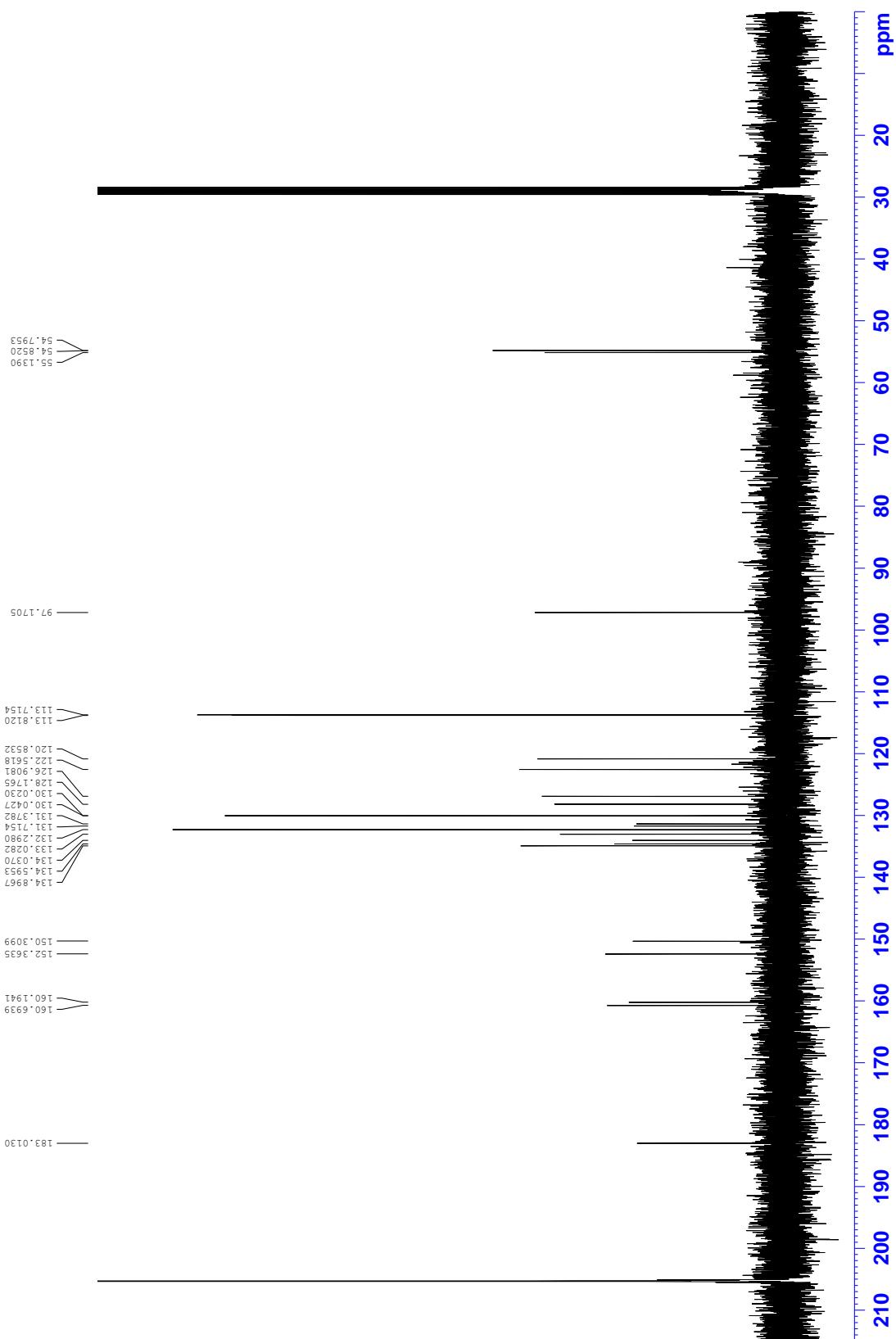
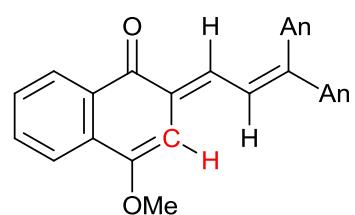
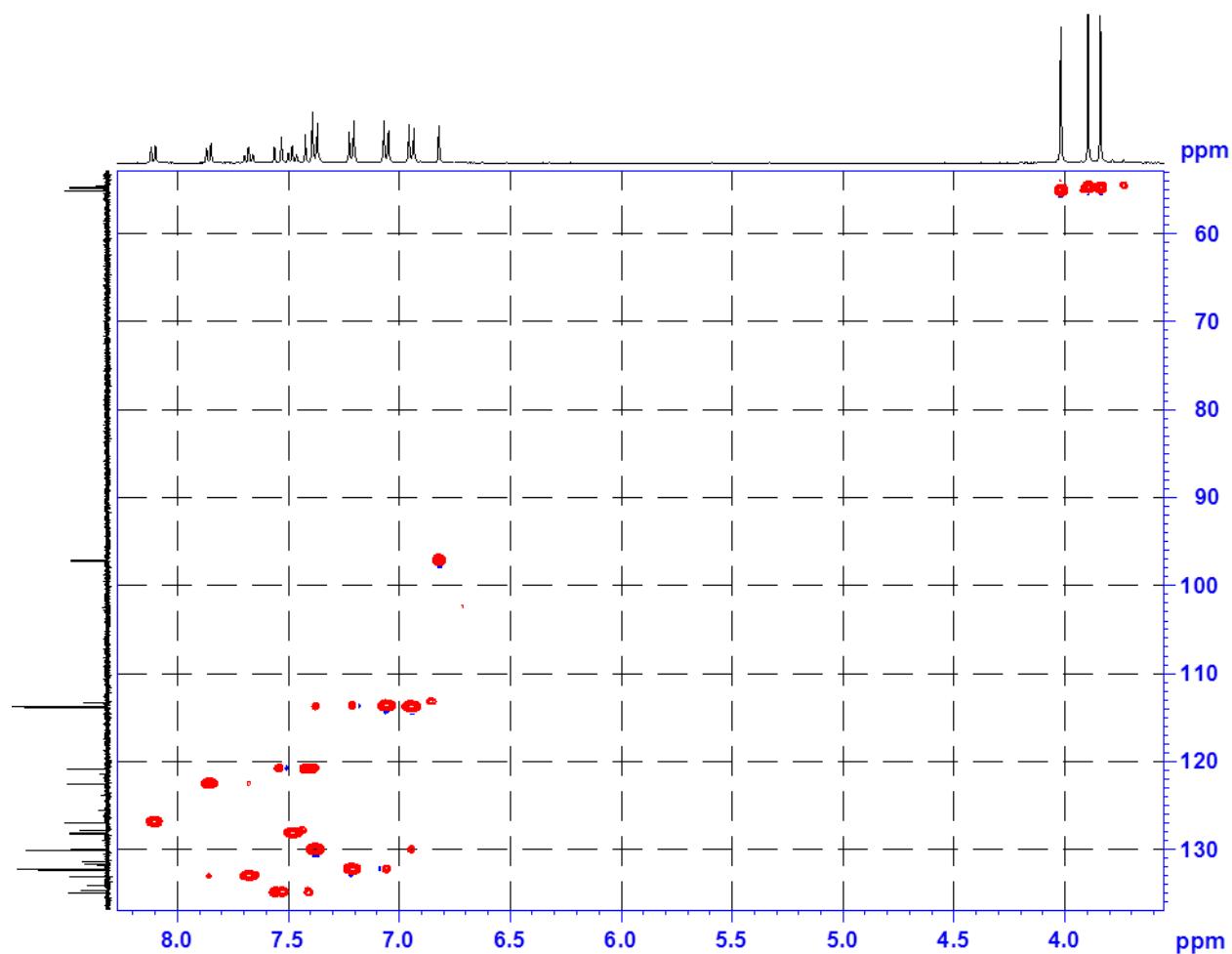
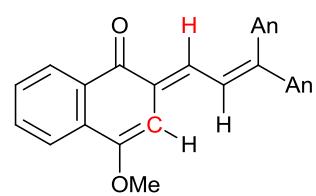
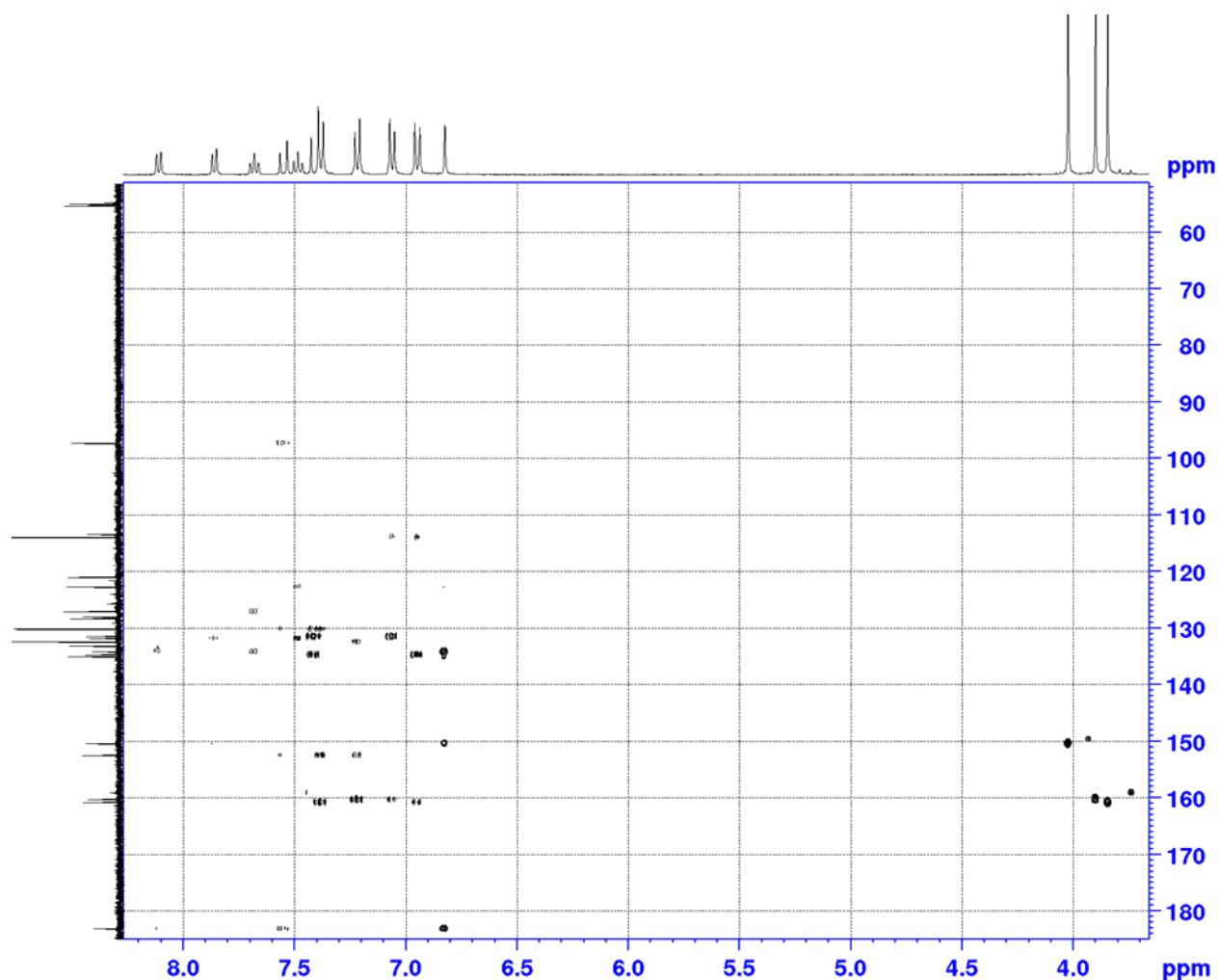


Figure S15 HSQC spectrum of compound **10** in d_6 -acetone



One bond correlation between **H** at δ 6.82 and **C** at δ 97.17

Figure S16 HMBC spectrum of compound **10** in d_6 -acetone



Three bond correlation between **H** at δ 7.55 and **C** at δ 97.17

Figure S17 COSY spectrum of compound **10** in d_6 -acetone (aromatic region)

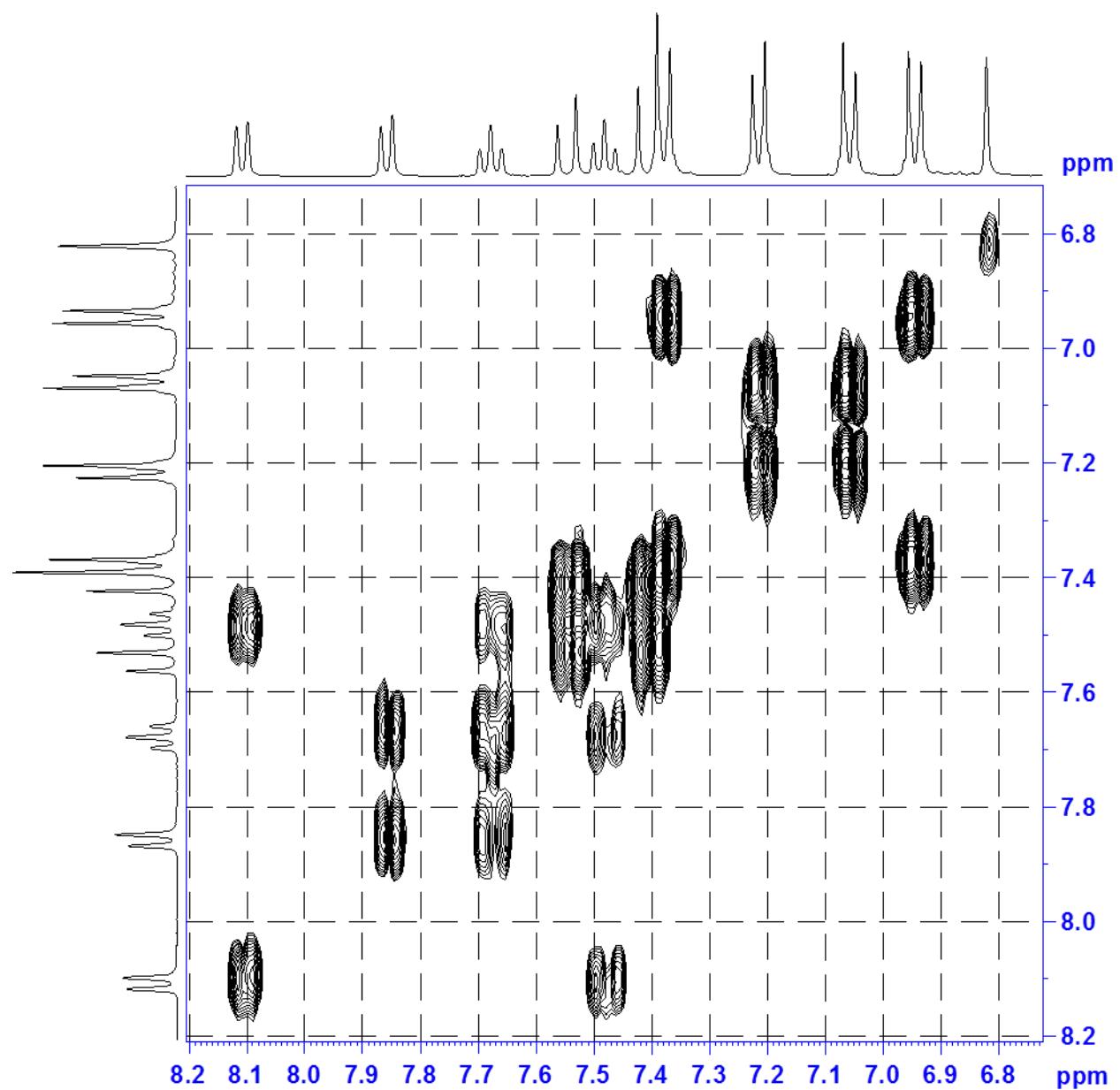
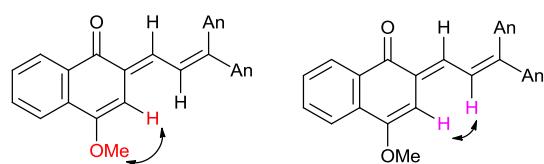
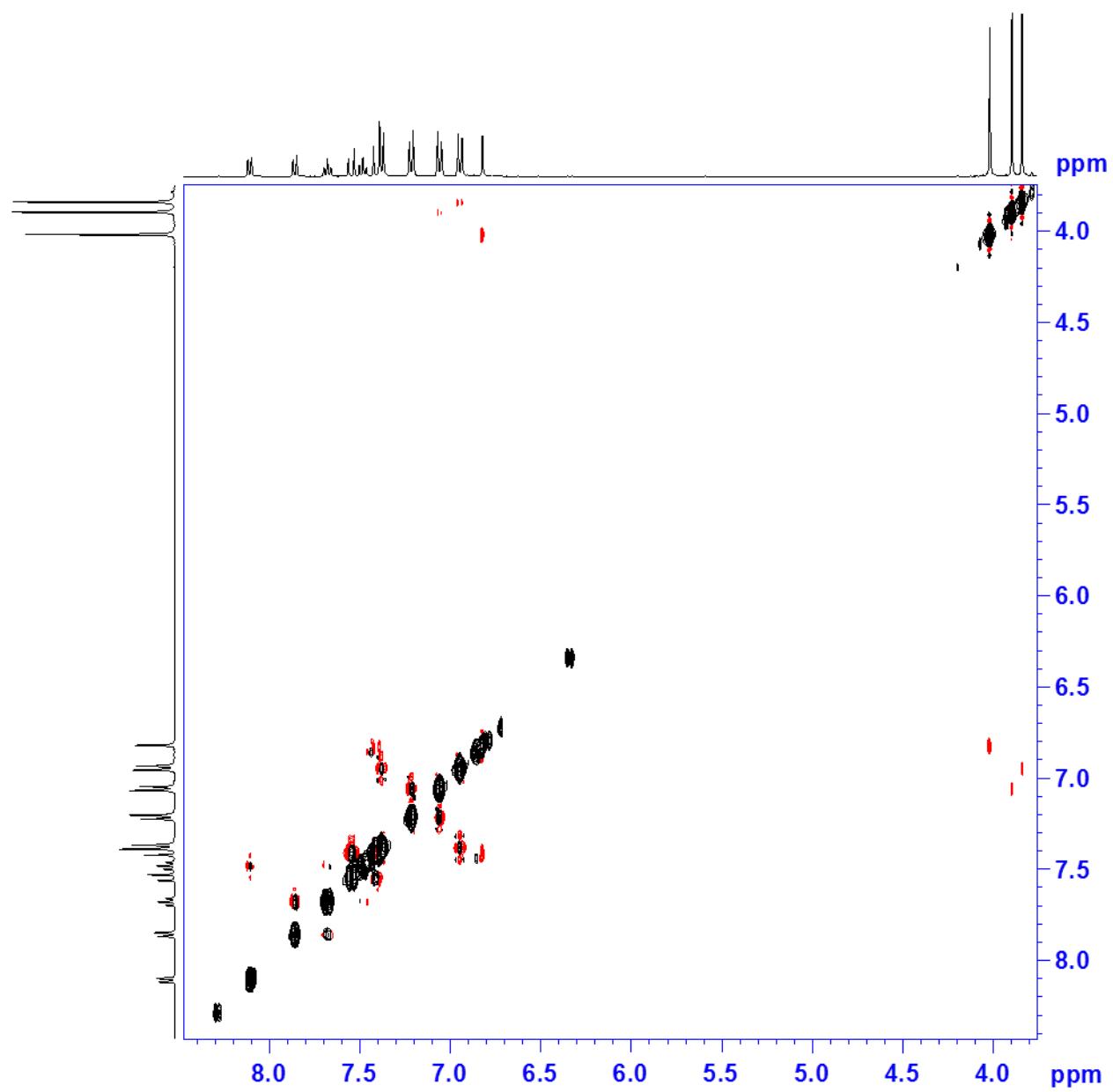


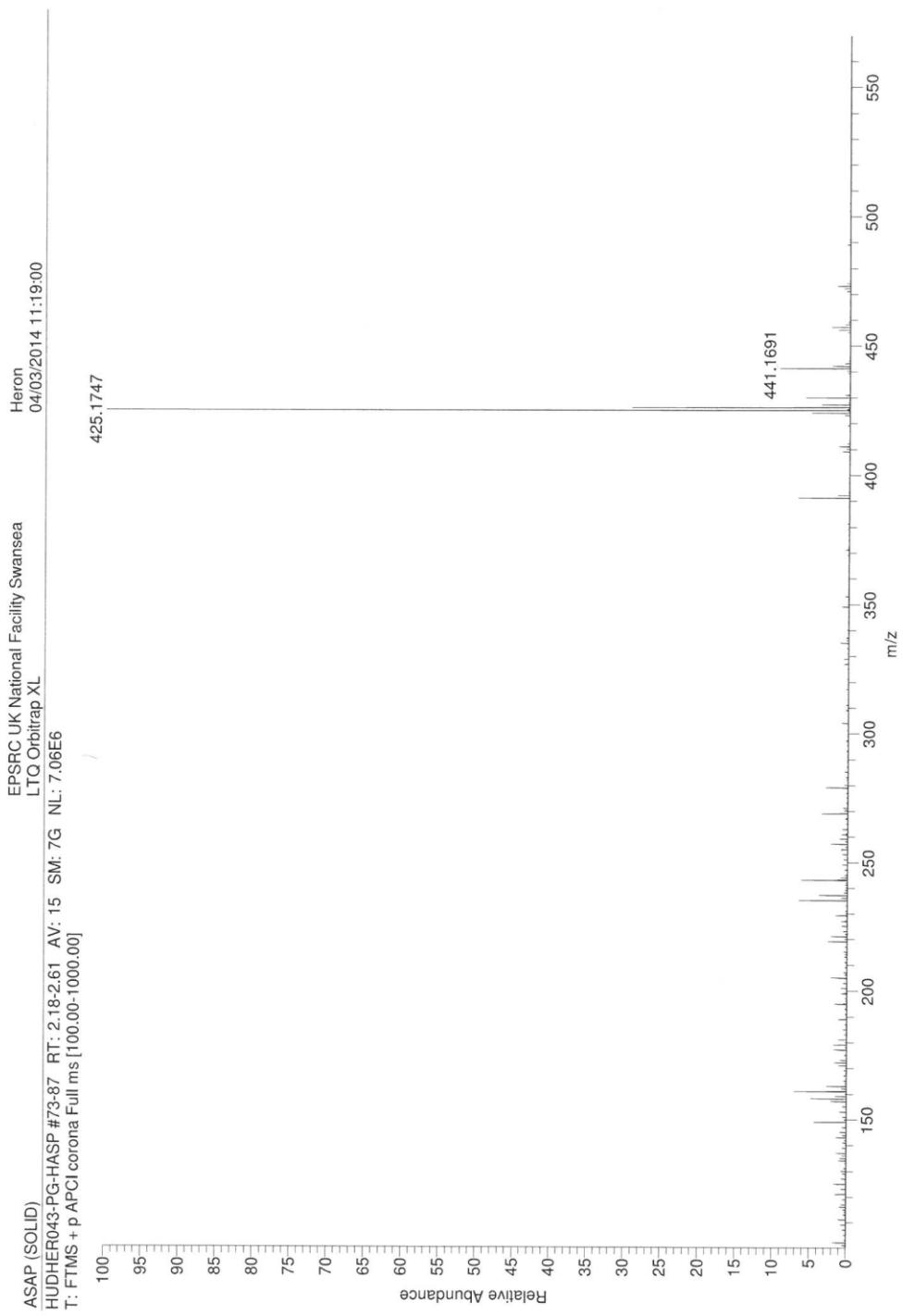
Figure S18 NOE spectrum of compound **10** in d_6 -acetone

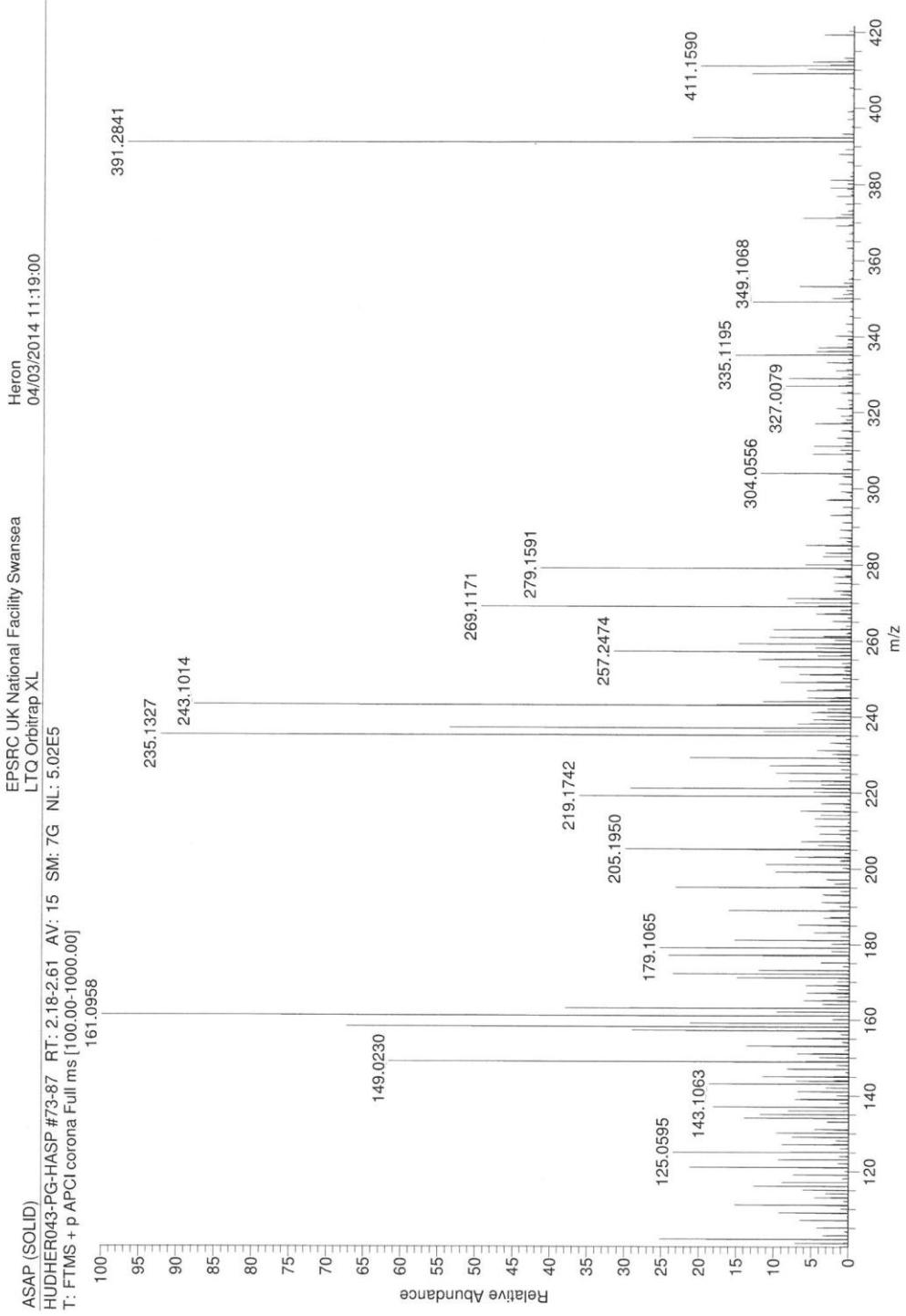


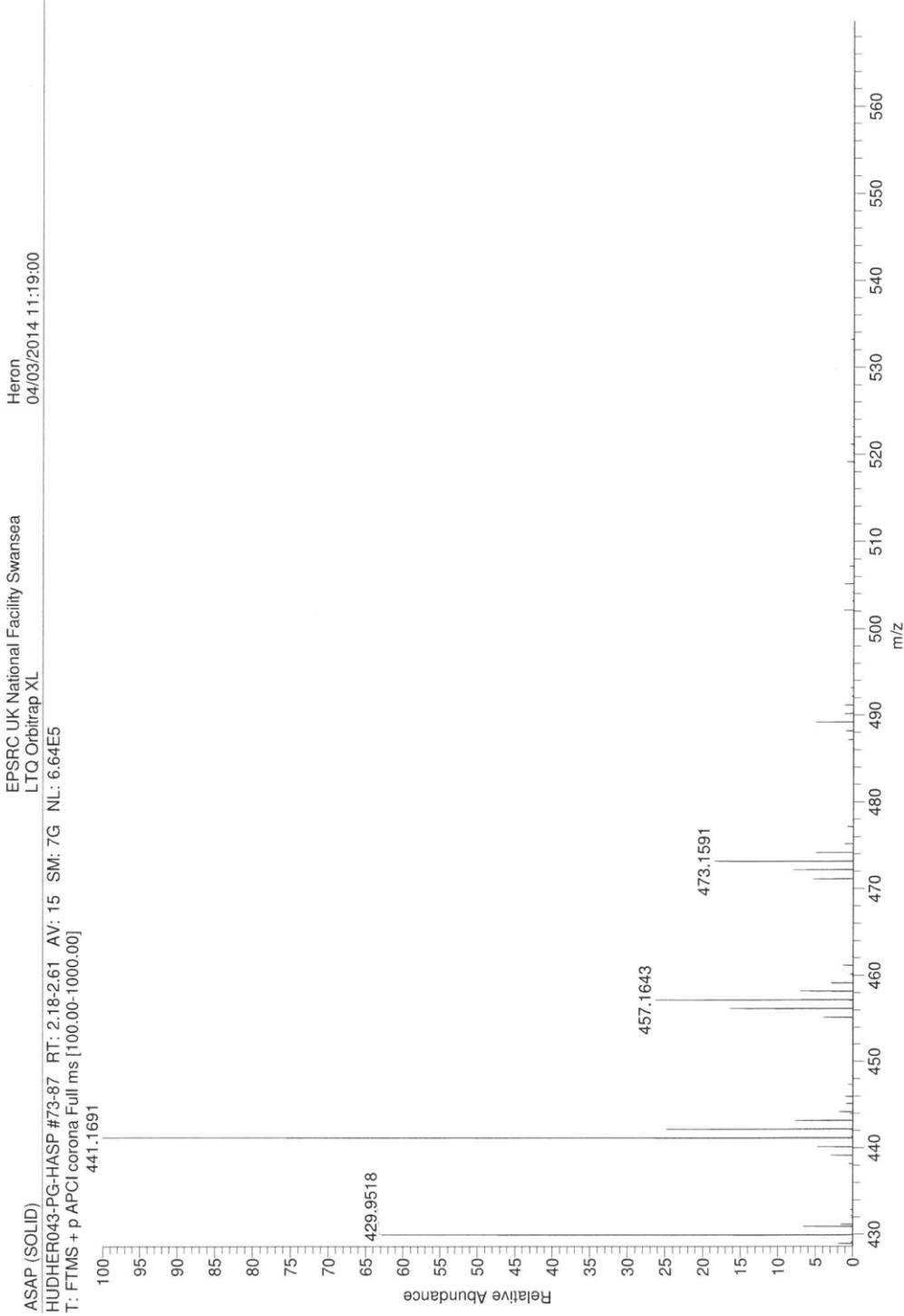
Weak correlation between **H** at δ 6.82 and **OMe** at δ 4.02

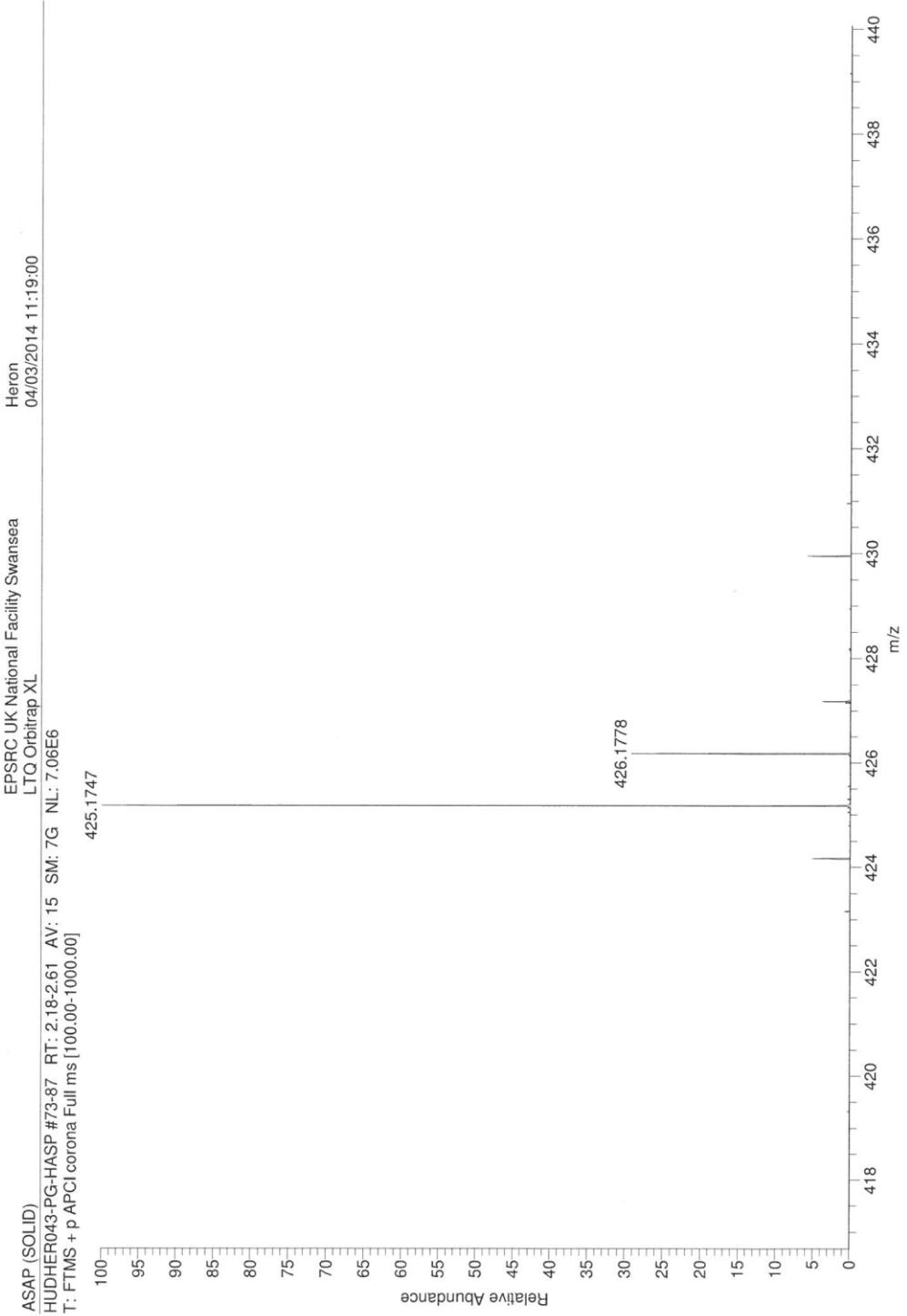
Weak correlation between **H** at δ 6.82 and **H** at δ 7.44

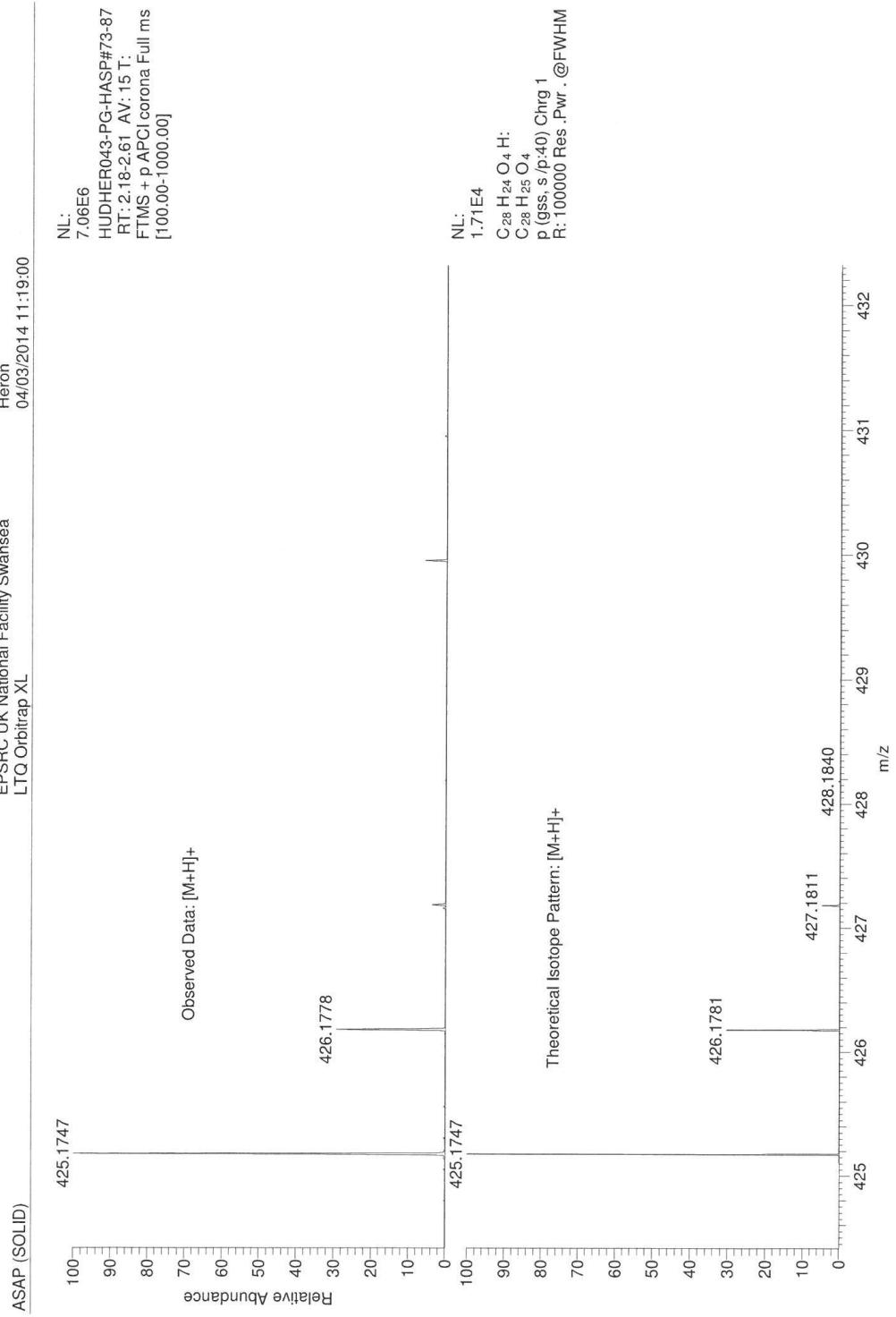
Figure S19 Mass Spectral data for compound **10**











Isotope:		Min.	Max.
14 N		015	
16 O		015	
12 C		060	
1 H		070	
23 Na		00	
Tolerance Window:		+ - 5.00 ppm	
Db/Ring Equiv:		-3 .. 100	
Fits:	100		
		N-Rule: Charge: 1	Do not use
Mass	Theoretical Mass	Delta [ppm]	RDB
425.1747	425.1747	-0.1	16.5
425.1752	425.1752	-1.3	9.5
425.1752		-1.3	C ₁₂ H ₂₁ O ₅ N ₁₂
425.1739	425.1739	1.9	4.0
425.1739		1.9	C ₁₄ H ₂₇ O ₁₀ N ₅
425.1739		1.9	C ₁₃ H ₃₁ O ₁₄ N ₁
425.1739		1.9	C ₁₂ H ₂₅ O ₉ N ₈
425.1739		1.9	C ₁₁ H ₁₉ O ₁ N ₁₅
425.1734	425.1734	3.1	10.0
425.1761	425.1761	-3.2	17.0
425.1766	425.1766	-4.4	21.5
425.1766		-4.4	C ₁₉ H ₂₁ N ₁
			C ₁₅ H ₂₃ O ₆ N ₉
			C ₁₆ H ₂₉ O ₁₁ N ₂

Figure S20 ^1H NMR spectrum of compound **9** in d_6 -acetone + $\text{CF}_3\text{CO}_2\text{H}$ after 2 h

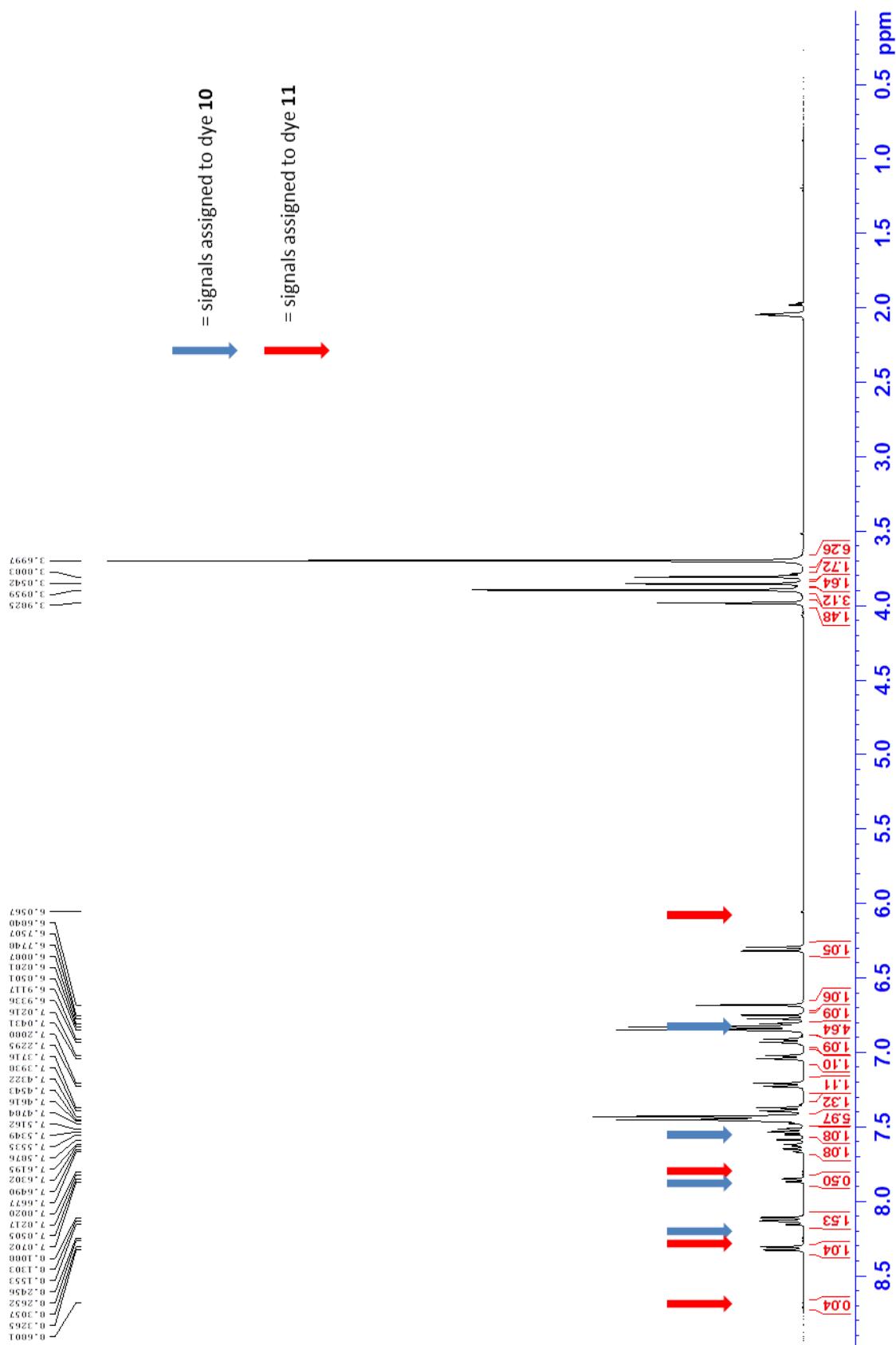


Figure S21 ^1H NMR spectrum of compound **10** in d_6 -acetone + $\text{CF}_3\text{CO}_2\text{H}$ after 2 h

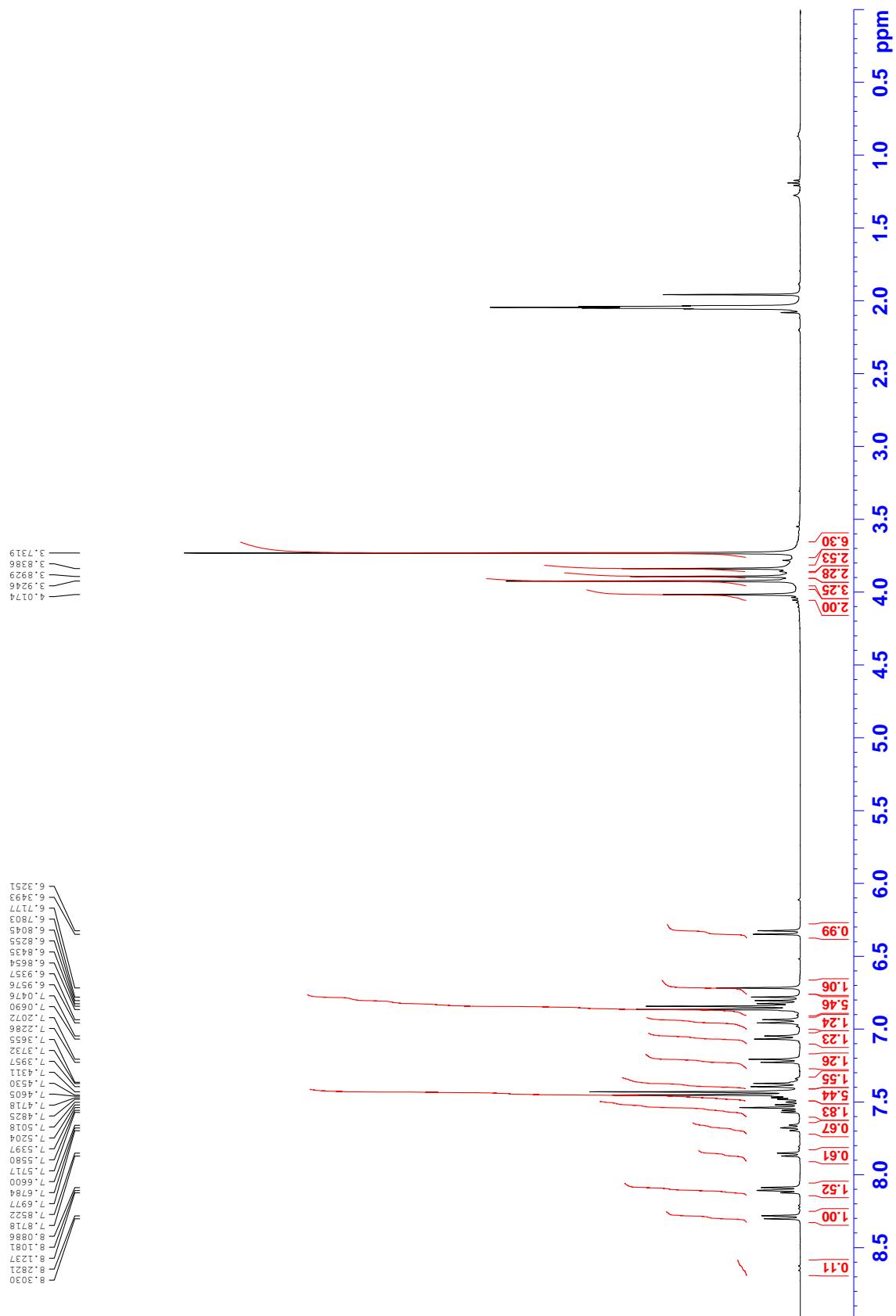


Figure S22 ^1H NMR spectrum of compound **12** in commercial CDCl_3

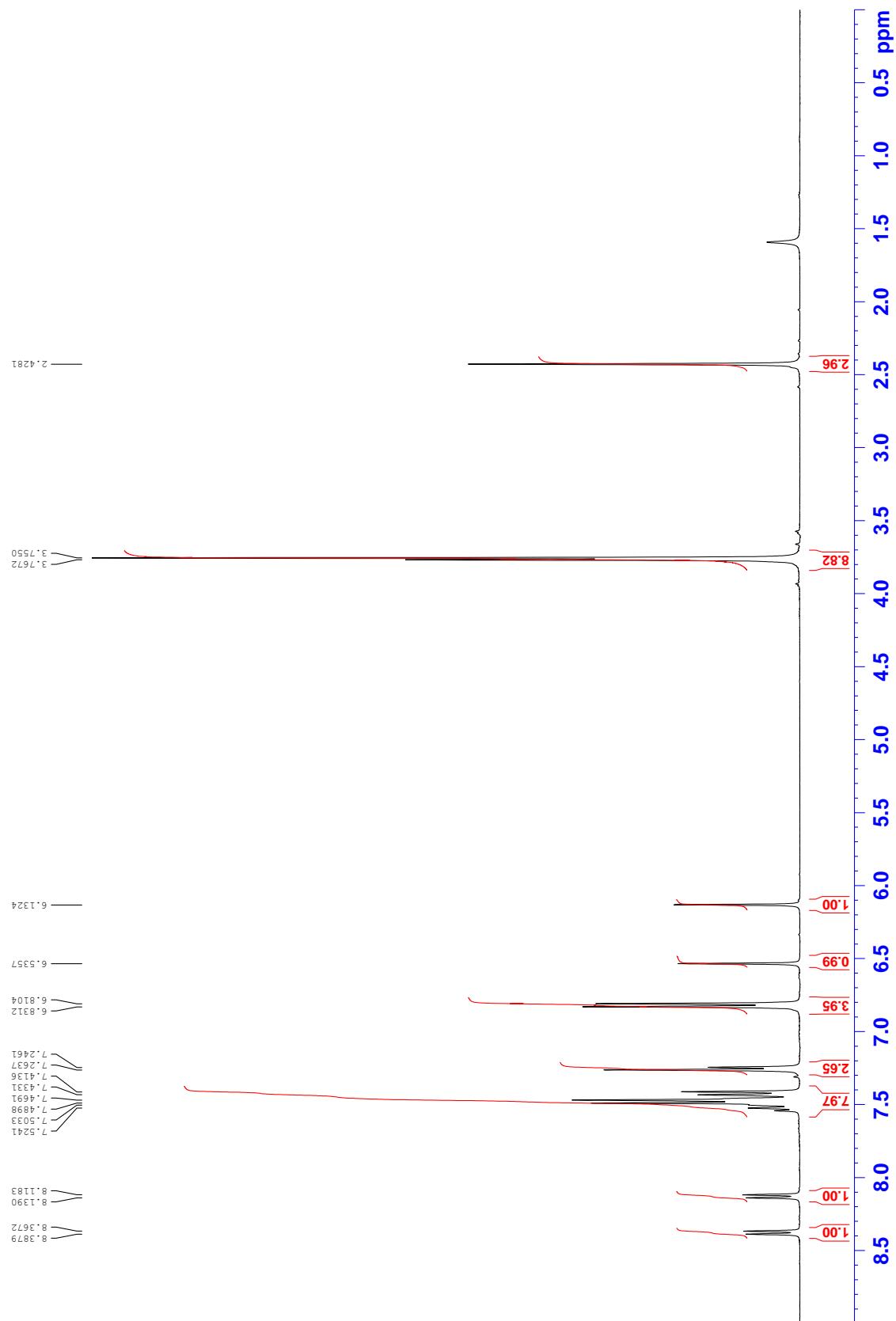


Figure S23 ^{13}C NMR spectrum of compound **12** in commercial CDCl_3

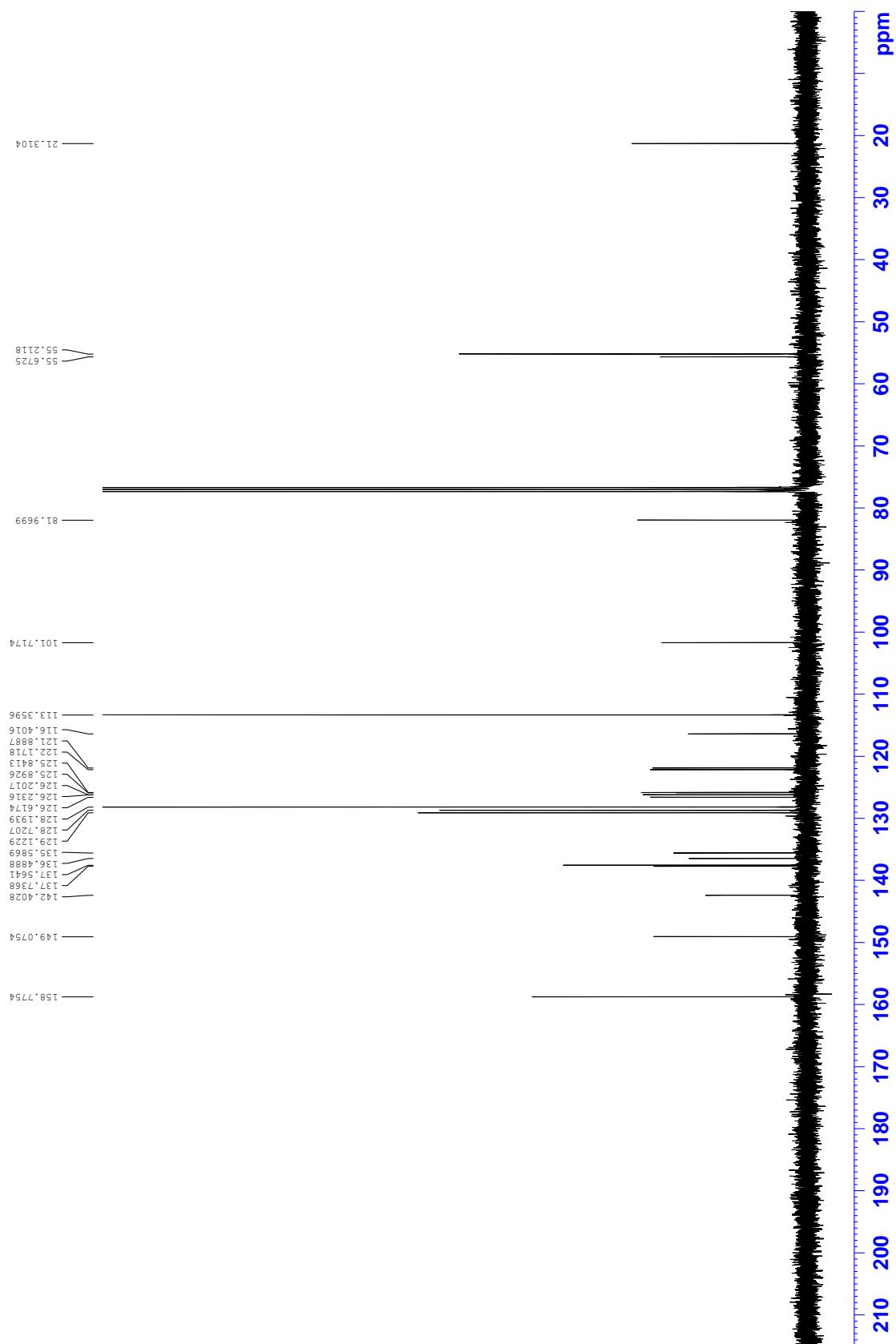
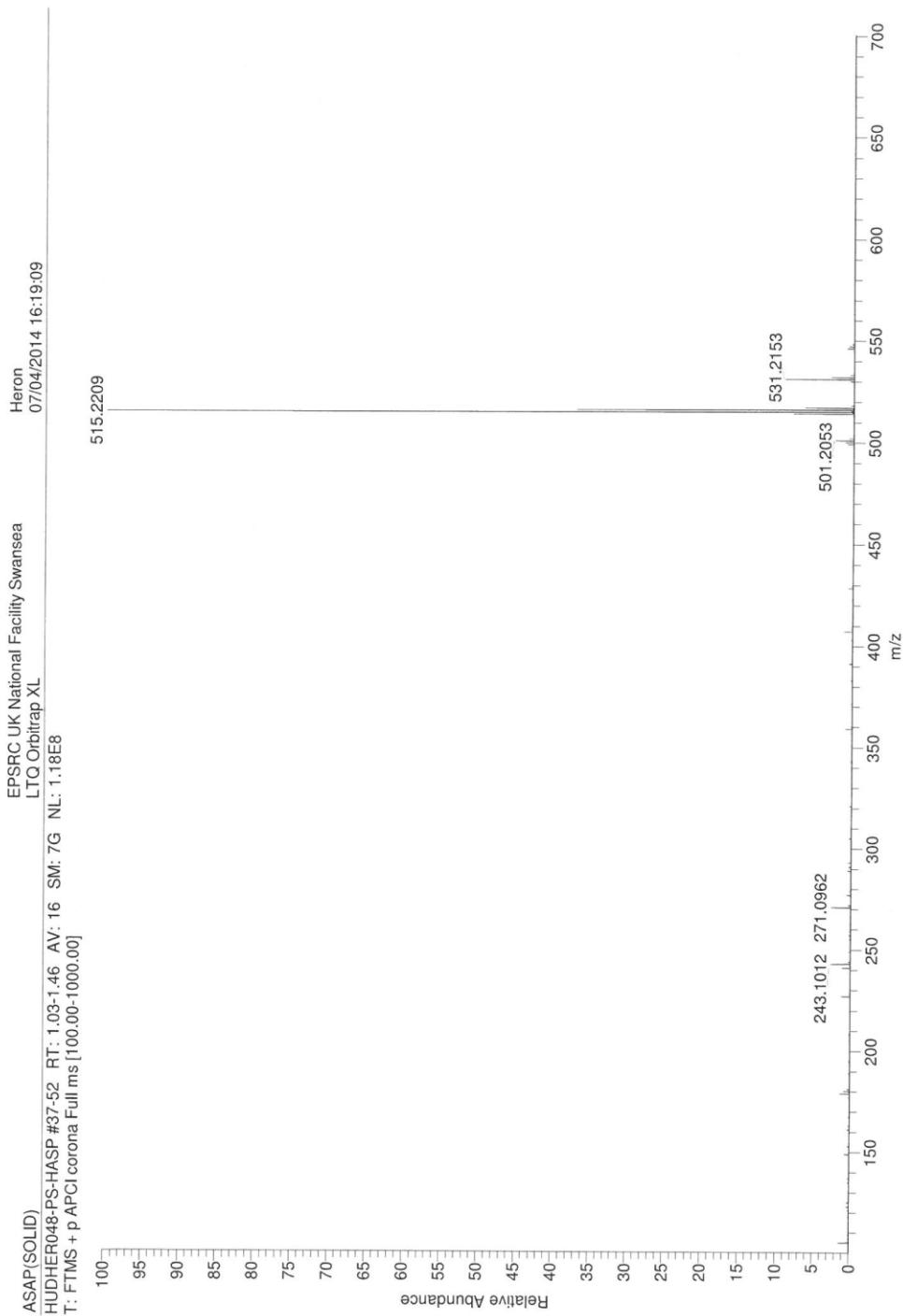
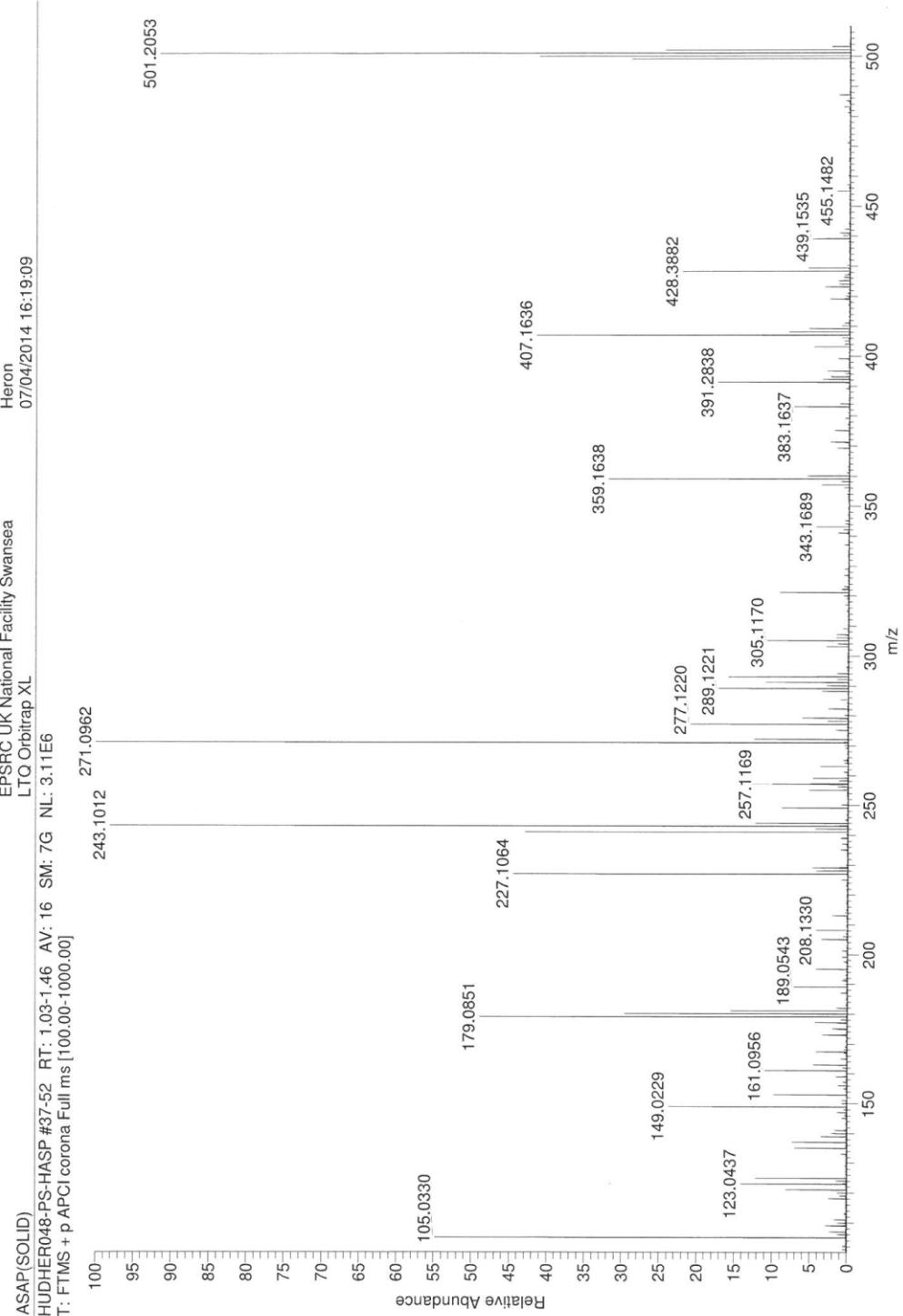
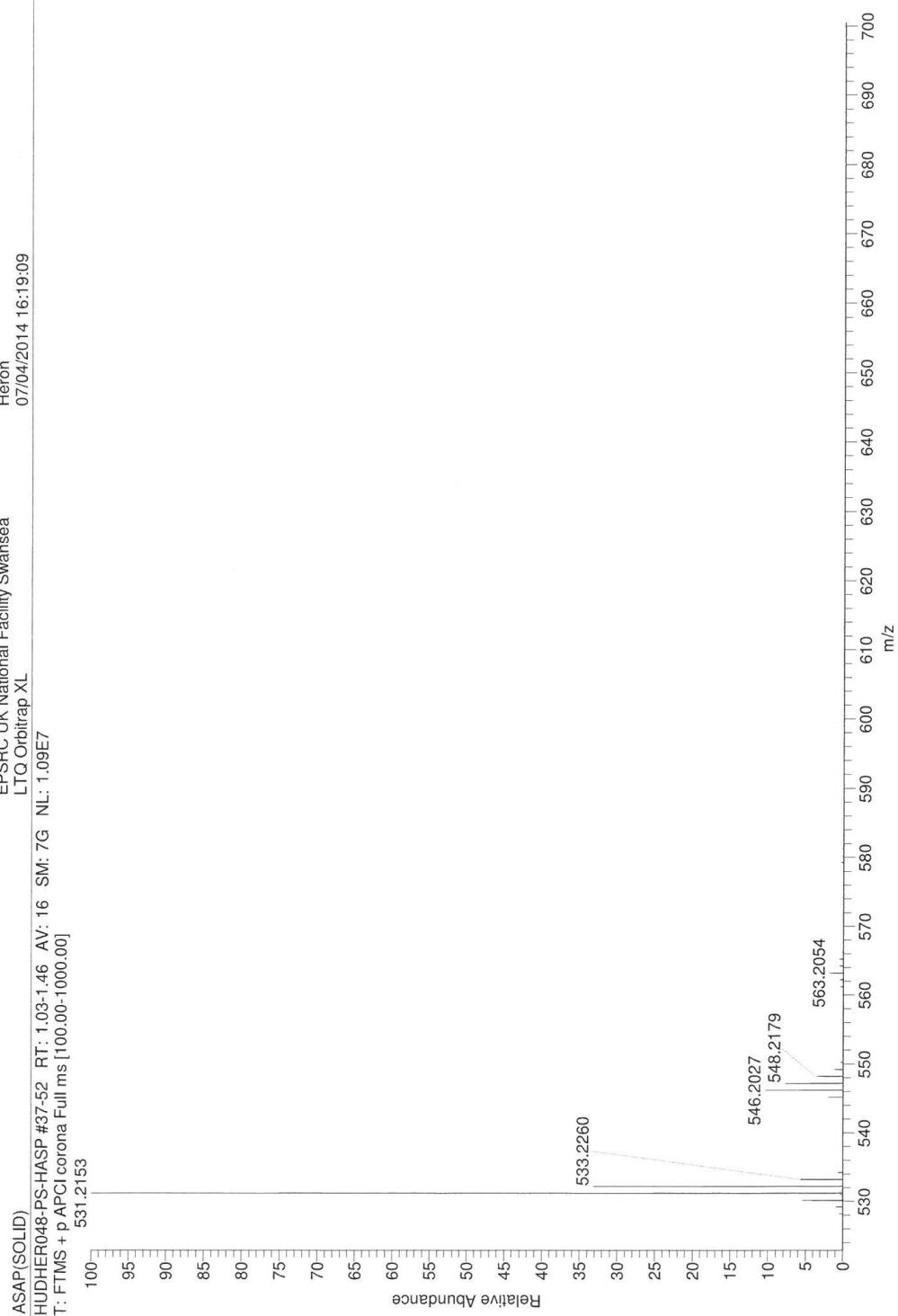
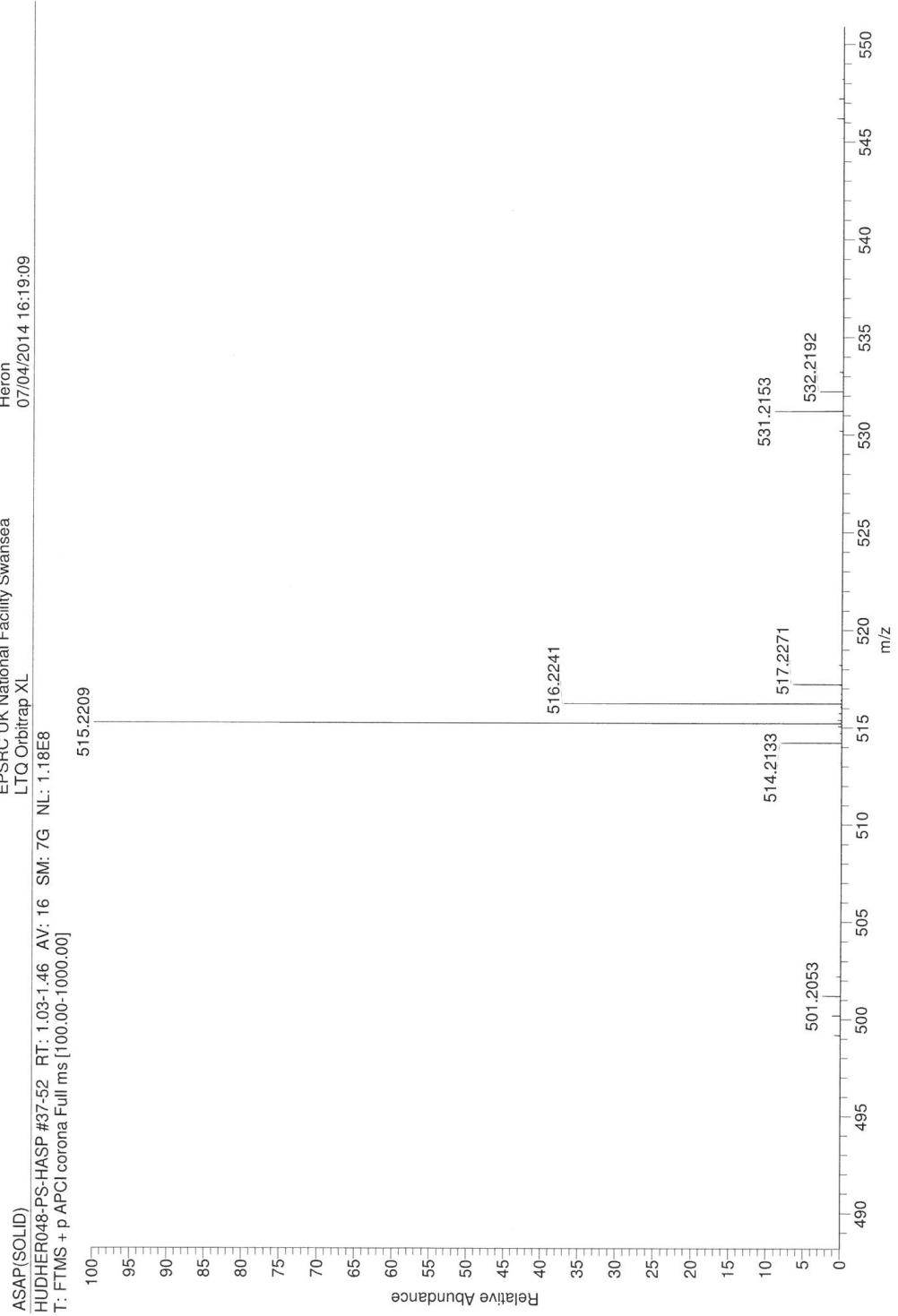


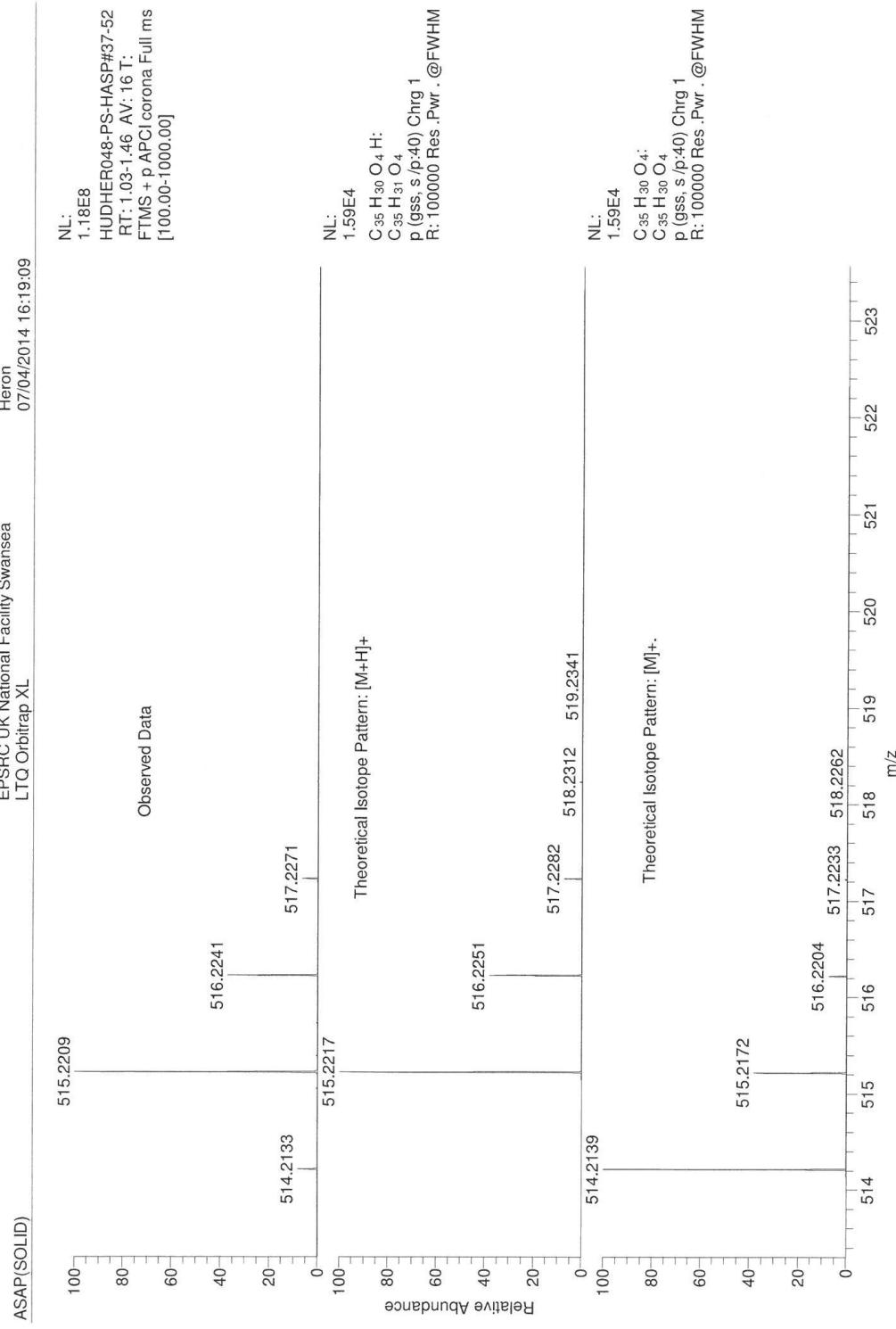
Figure S24 Mass Spectral data for compound **12**











Isotope:		Min. . . Max.	
14 N		0 . . . 14	
16 O		0 . . . 16	
12 C		0 . . . 90	
1 H		0 . . . 100	
23 Na		0 . . . 0	
Tolerance Window:		+ - 5.00 ppm	
DB/Ring Equiv:		-3 . . 100	
Fits:		100	
			N-Rule: Do not use
			Charge: 1
Mass	Theoretical Mass	Delta [ppm]	RDB
515.2209	515.2209	0.1	3.0
515.2209	515.2203	1.1	8.5
515.2217	515.2217	-1.5	21.0
515.2222	515.2222	-2.5	20.5
515.2222	515.2222	-2.5	13.5
515.2195	515.2195	2.7	8.0
515.2195	515.2190	2.7	3.5
515.2230	515.2230	-4.1	3.5
			C ₂₀ H ₇ O ₄ N ₁
			C ₁₉ H ₁₁ O ₅ N ₈
			C ₃₃ H ₂₉ O ₃ N ₃
			C ₃₅ H ₃₁ O ₄
			C ₂₀ H ₇ O ₅ N ₁₂
			C ₂₁ H ₁₃ O ₁₀ N ₃
			C ₁₈ H ₁₃ O ₃ N ₄
			C ₁₇ H ₉ O ₆ N ₁₁
			C ₁₁ H ₂₇ O ₄ N ₆
			C ₃₆ H ₂₇ N ₃

Figure S25 ^1H NMR spectrum of compound **13** in commercial CDCl_3

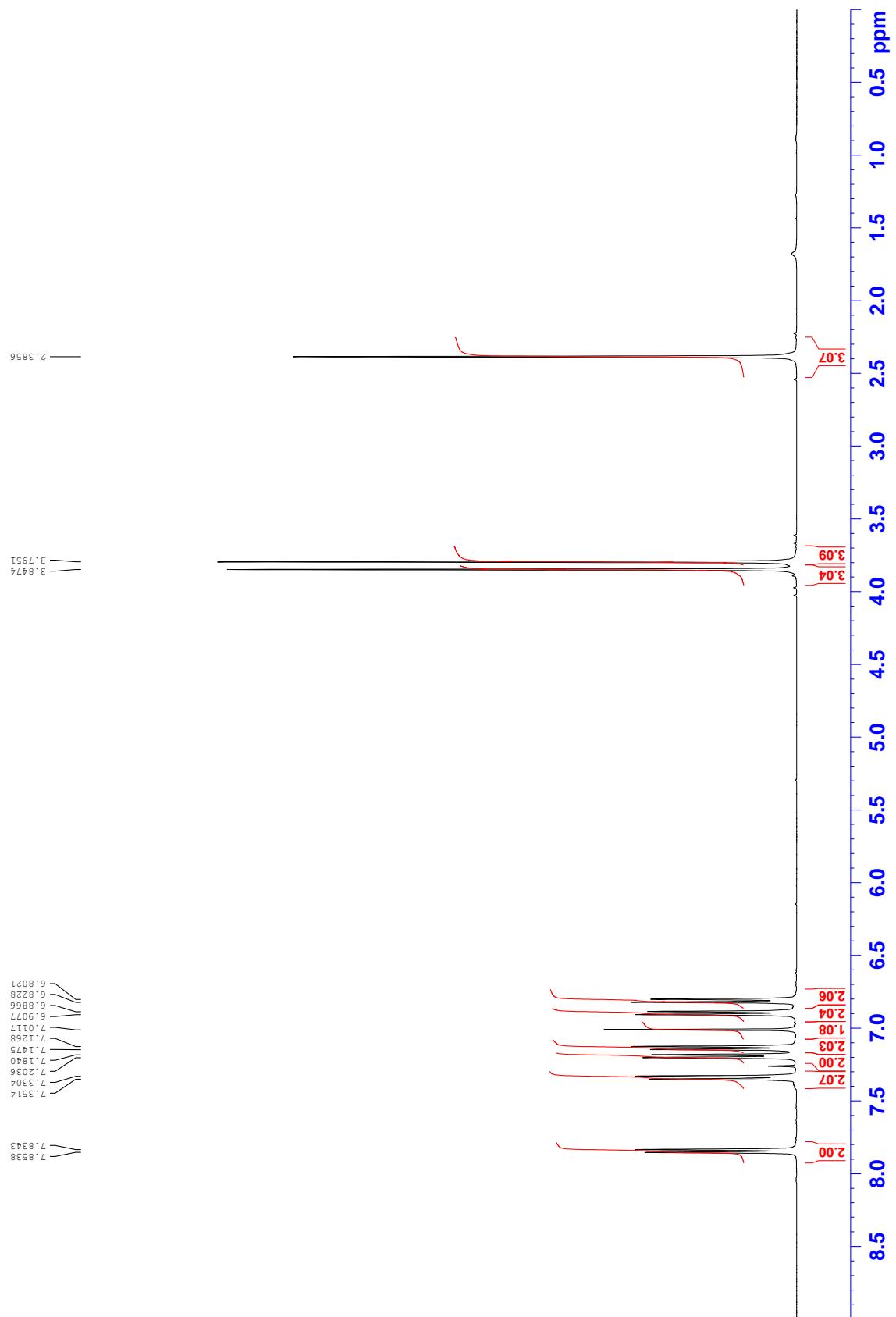


Figure S26 ^{13}C NMR spectrum of compound **13** in commercial CDCl_3

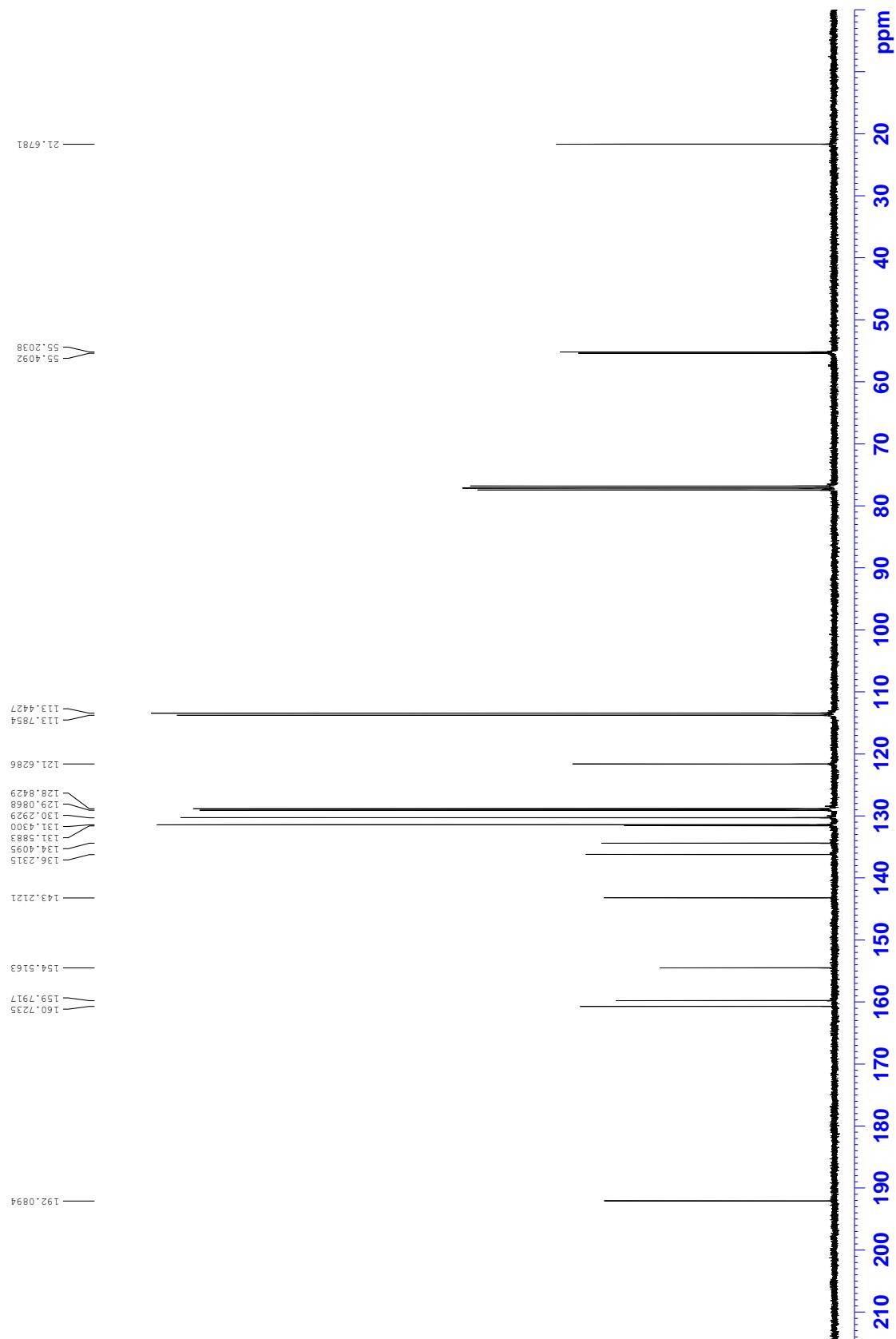
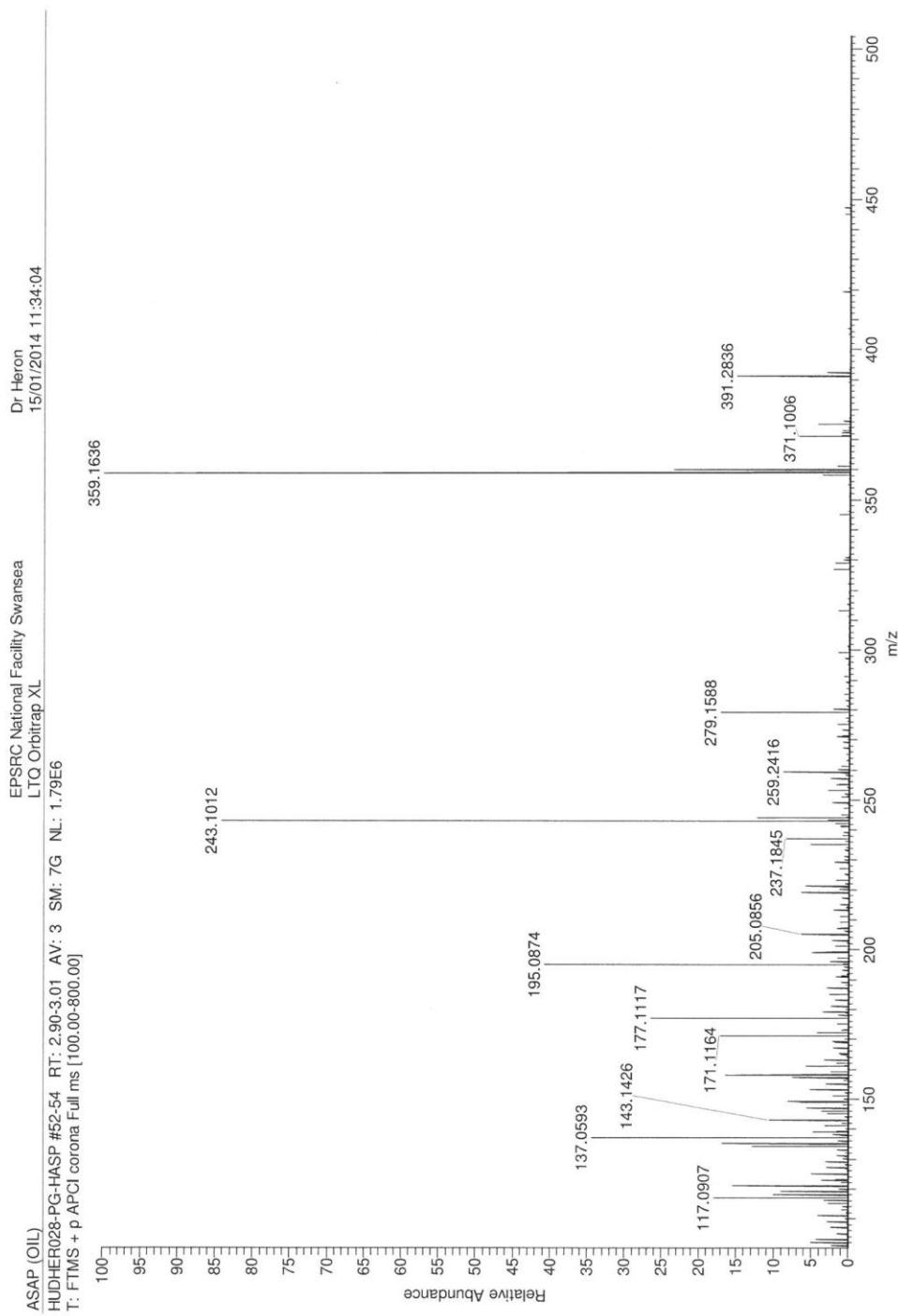
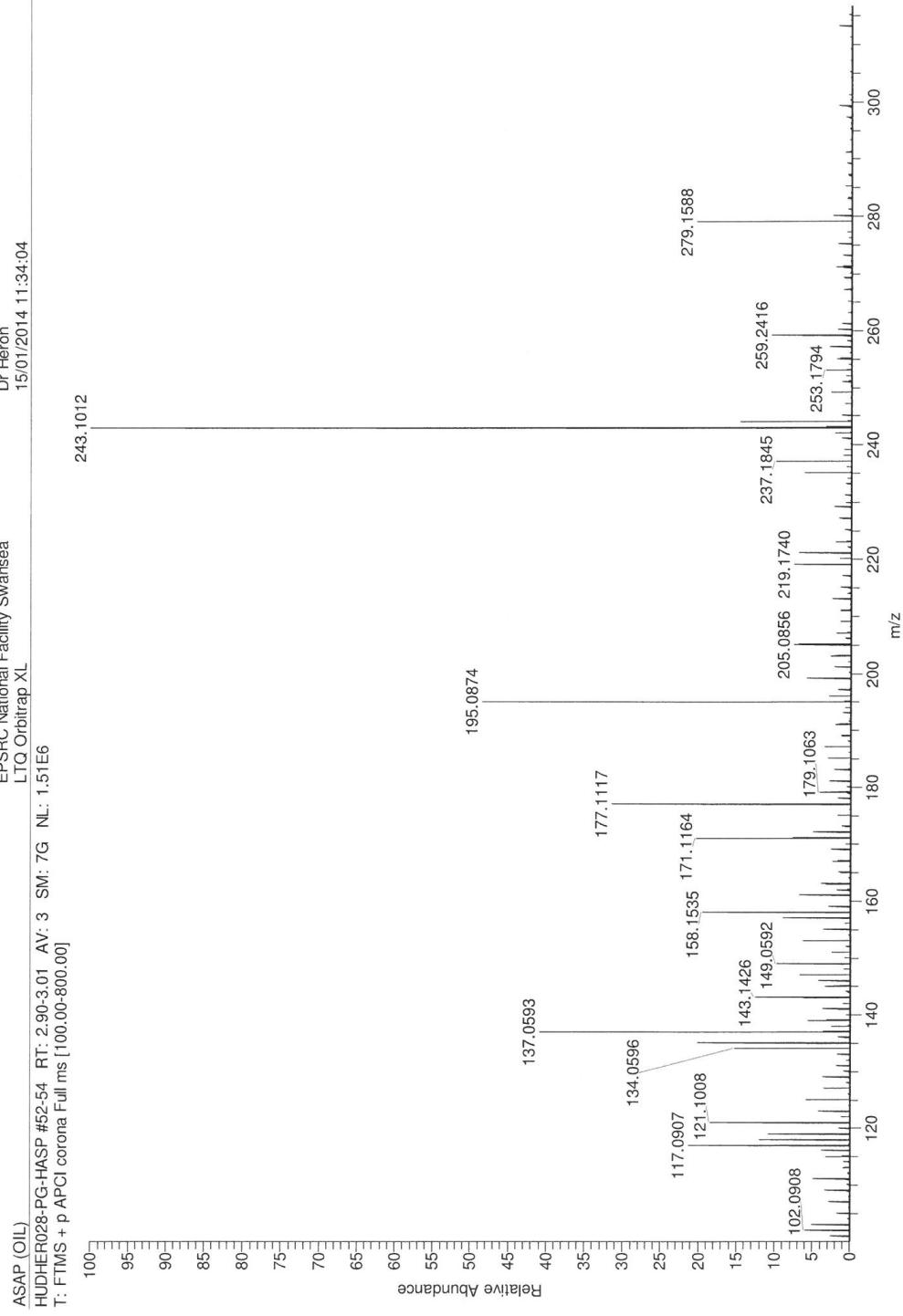
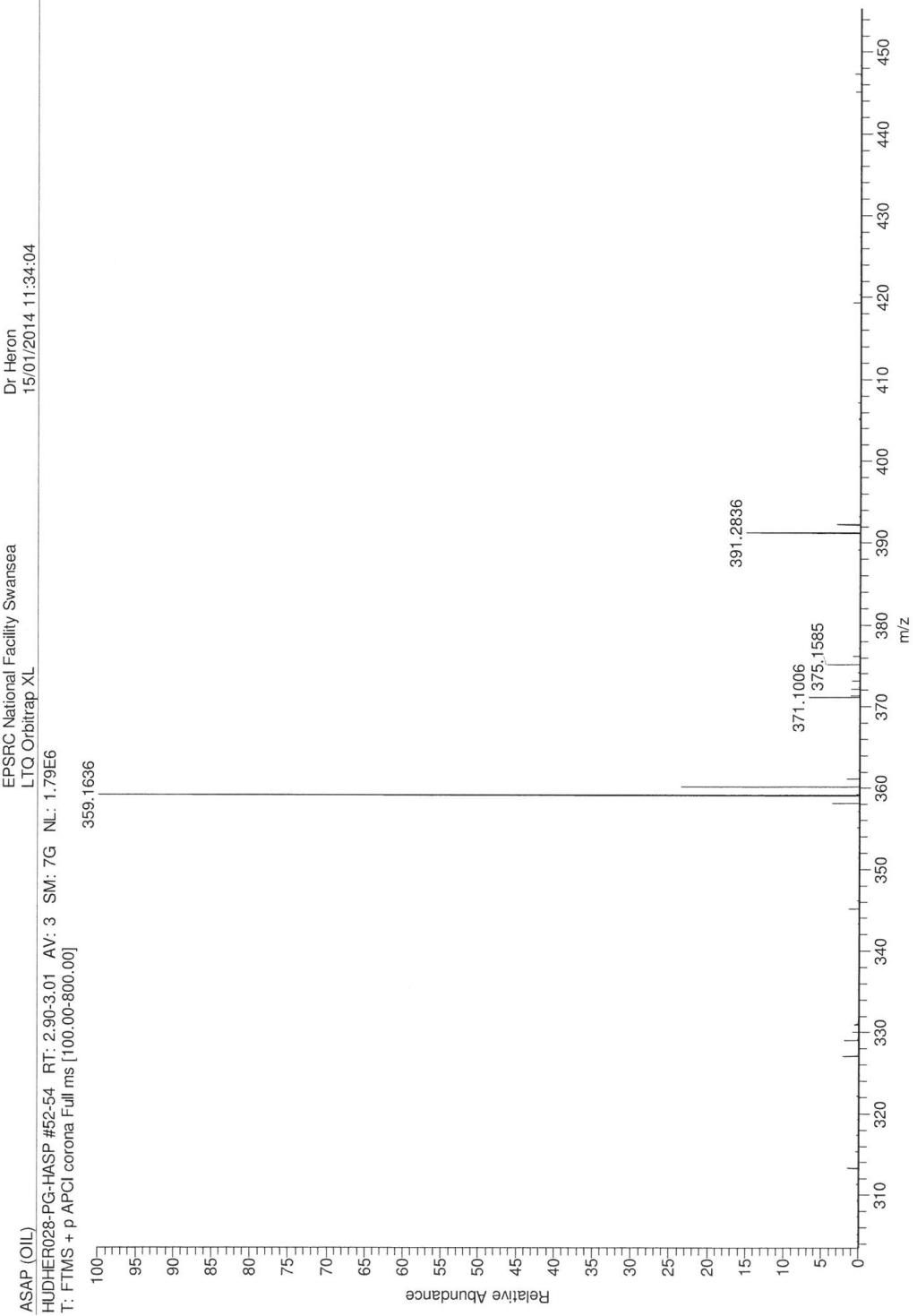
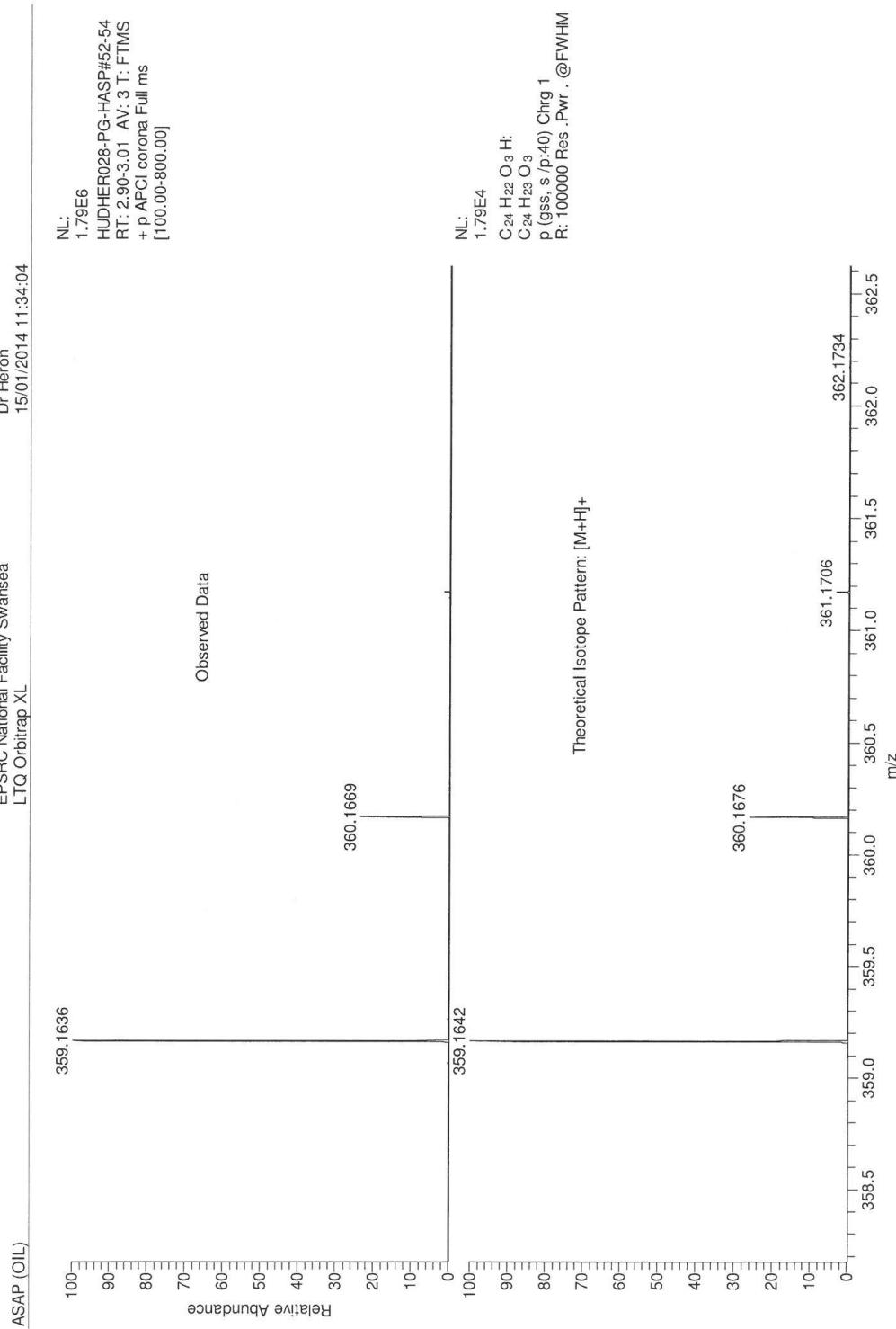


Figure S27 Mass Spectral data for compound **13**









Isotope:	Min. . . Max.			
14 N	0 . . . 12			
16 O	0 . . . 13			
12 C	0 . . . 60			
1 H	0 . . . 70			
23 Na	0 . . . 0			
Tolerance Window:	+/- 5.00 PPM			
Db/Ring Equiv:	-3 .. 100			
Fits:	100			
	N-Rule: Do not use			
	Charge: 1			
Mass	Theoretical	Delta [PPM]	RDB	Composition
359.1636	359.1633	0.7	1.5	C ₈ H ₃₃ O ₉ N ₆
	359.1642	-1.6		C ₁₄ H ₃₃ O ₂
	359.1628	2.1	13.5	C ₂₂ H ₂₁ O ₂ N ₃
	359.1647	-3.0	14.0	C ₉ H ₁₉ O ₄ N ₂
	359.1647	-3.0	6.5	C ₁₀ H ₂₅ O ₅ N ₅
	359.1620	4.5	1.0	C ₆ H ₂₁ O ₇ N ₁₁
			2.0	

Cartesian coordinates (Å) for molecule **10** (ground-state) obtained by DFT [PCM-PBE0/6-31G(d)]

E= -1380.71673921 au

6	2.499386	-1.472469	0.290691
6	3.802944	-1.845077	0.366646
6	4.877383	-0.912922	0.048774
6	4.556871	0.403601	-0.342565
6	3.150105	0.853277	-0.441342
6	2.108103	-0.144894	-0.110712
1	6.473126	-2.305126	0.427107
1	1.720409	-2.181125	0.546744
6	6.224689	-1.293335	0.125791
6	5.581331	1.305535	-0.646540
6	6.911015	0.920003	-0.567693
6	7.228848	-0.385044	-0.179834
1	5.301704	2.311430	-0.944587
1	7.700801	1.626792	-0.805631
1	8.268709	-0.693947	-0.115979
8	2.879521	2.007326	-0.785977
6	0.802046	0.281515	-0.206156
6	-0.368556	-0.500609	-0.013669
1	0.678935	1.323699	-0.491365
1	-0.243829	-1.573428	0.114475
6	-1.658583	-0.025613	-0.021970
6	-1.968911	1.415413	-0.002719
6	-1.301548	2.301078	0.853361
6	-2.965183	1.948582	-0.843139
6	-1.589045	3.662526	0.869907
1	-0.559034	1.913623	1.545178
6	-3.252295	3.300503	-0.848160
1	-3.505302	1.287782	-1.515426
6	-2.565374	4.172329	0.009537
1	-1.057143	4.308666	1.559169
1	-4.008719	3.710433	-1.510761
6	-2.784113	-0.973745	-0.032199
6	-3.995988	-0.673703	0.620298
6	-2.689107	-2.217373	-0.674198
6	-5.042559	-1.576404	0.647171
1	-4.104601	0.277833	1.132413
6	-3.734952	-3.132560	-0.662612
1	-1.786729	-2.466782	-1.225481
6	-4.922911	-2.817164	0.005710
1	-5.968494	-1.344775	1.165098
1	-3.621161	-4.075154	-1.186316
8	4.244608	-3.068483	0.734845
8	-5.999708	-3.627913	0.080865
8	-2.922347	5.472630	-0.062380
6	-2.252764	6.393497	0.779911
1	-2.681086	7.370825	0.555099
1	-2.417830	6.159392	1.838259
1	-1.176071	6.412029	0.574271
6	-5.924755	-4.894954	-0.547659
1	-5.118273	-5.503910	-0.122606
1	-6.882769	-5.380085	-0.358010
1	-5.775709	-4.795667	-1.629275
6	3.277431	-4.045619	1.064849
1	2.673453	-3.730000	1.924126
1	3.834588	-4.947008	1.322101
1	2.617384	-4.252220	0.213699

Cartesian coordinates (Å) for molecule **10** (excited-state) obtained by DFT [PCM-PBE0/6-31G(d)]

E(TD)= -1380.64376071 au

6	2.519738	-1.355567	0.376330
6	3.857321	-1.743957	0.451403
6	4.901210	-0.847818	0.054644
6	4.544074	0.447391	-0.413878
6	3.146012	0.881422	-0.498942
6	2.128566	-0.088205	-0.084487
1	6.527009	-2.200337	0.478827
1	1.754620	-2.057185	0.690735
6	6.261892	-1.211323	0.119131
6	5.562368	1.329663	-0.802260
6	6.894797	0.958260	-0.735088
6	7.245395	-0.320976	-0.270477
1	5.269348	2.313015	-1.157956
1	7.670317	1.655072	-1.041505
1	8.290590	-0.612615	-0.217209
8	2.838804	2.020395	-0.907401
6	0.765524	0.331279	-0.178112
6	-0.357740	-0.431624	0.074394
1	0.647540	1.357867	-0.515463
1	-0.217097	-1.485658	0.308920
6	-1.707708	0.004319	0.023768
6	-2.053823	1.419427	0.044258
6	-1.341648	2.350439	0.827332
6	-3.128788	1.923796	-0.726810
6	-1.667869	3.698548	0.849648
1	-0.534141	1.998510	1.462102
6	-3.455144	3.264287	-0.724199
1	-3.689263	1.244409	-1.362432
6	-2.728773	4.169712	0.065754
1	-1.101657	4.370704	1.484947
1	-4.266723	3.644121	-1.337828
6	-2.759020	-0.993652	-0.037331
6	-4.054832	-0.747032	0.484495
6	-2.533529	-2.276436	-0.586060
6	-5.041204	-1.709462	0.462146
1	-4.267086	0.212377	0.946319
6	-3.518909	-3.251236	-0.617831
1	-1.571657	-2.502053	-1.037069
6	-4.787654	-2.975809	-0.090997
1	-6.022803	-1.513905	0.883810
1	-3.298233	-4.213691	-1.066375
8	4.262691	-2.944529	0.888769
8	-5.815447	-3.846745	-0.067574
8	-3.124873	5.456370	0.003922
6	-2.414703	6.412958	0.771363
1	-2.888127	7.373025	0.564052
1	-2.486605	6.195619	1.843374
1	-1.359845	6.455048	0.476567
6	-5.607774	-5.142855	-0.601590
1	-4.813718	-5.672201	-0.062285
1	-6.551047	-5.674049	-0.471738
1	-5.358769	-5.098193	-1.668161
6	3.295631	-3.897585	1.301061
1	2.712863	-3.524968	2.150559
1	3.860919	-4.778688	1.604132
1	2.623697	-4.157930	0.475935

Cartesian coordinates (Å) for molecule **11** (ground-state) obtained by DFT [PCM-PBE0/6-31G(d)]

E= -1380.71488583 au

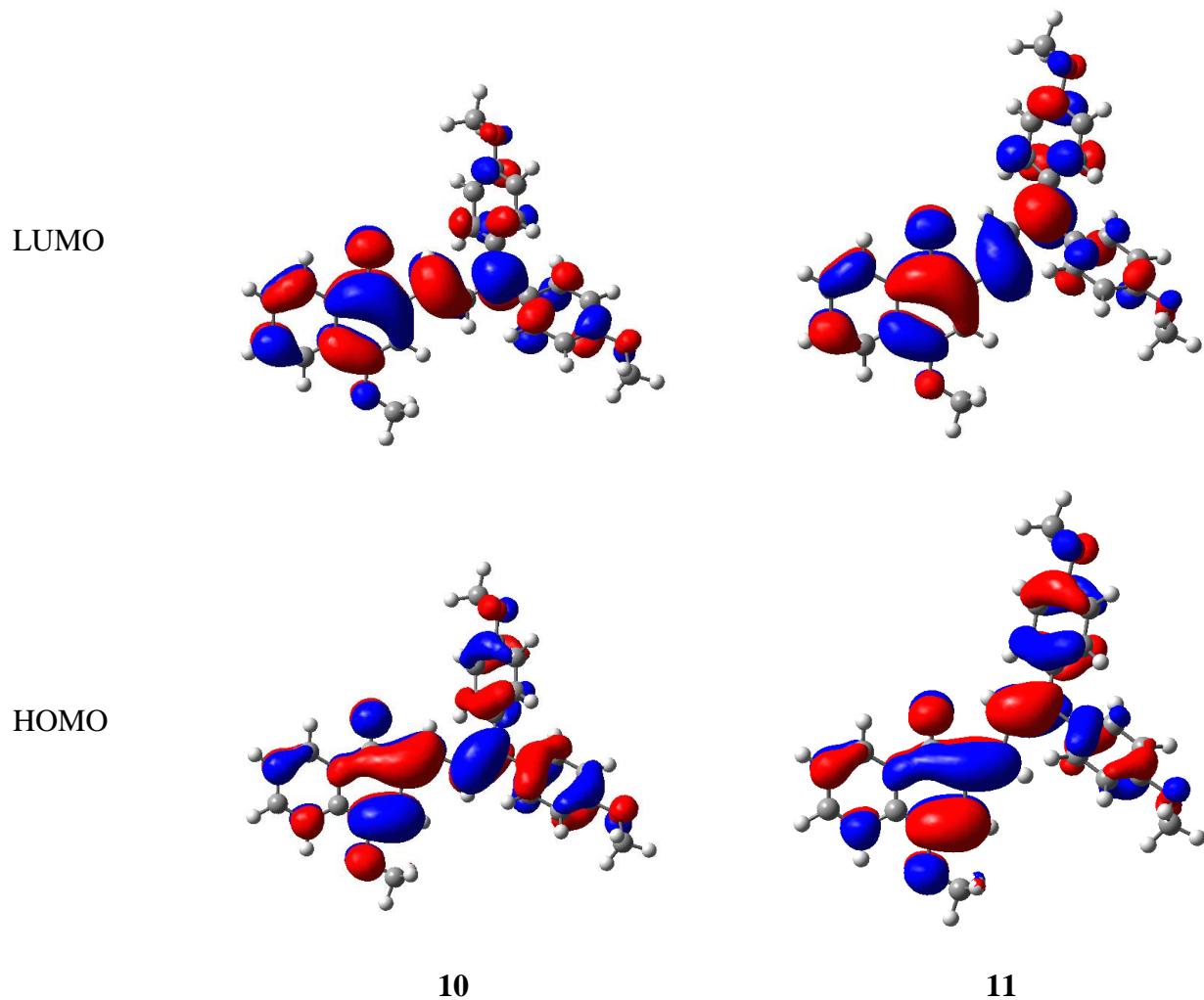
6	-3.100579	0.963911	0.443573
6	-4.435672	0.721966	0.419442
6	-4.950115	-0.585084	0.042326
6	-4.042687	-1.606014	-0.310504
6	-2.578117	-1.378723	-0.312536
6	-2.118041	-0.038246	0.101743
1	-7.023864	-0.069393	0.290793
1	-2.724582	1.940432	0.732494
6	-6.326622	-0.855043	0.020569
6	-4.529030	-2.866539	-0.675054
6	-5.890834	-3.124506	-0.691546
6	-6.789686	-2.111518	-0.341949
1	-3.807189	-3.631852	-0.943133
1	-6.258148	-4.106857	-0.974554
1	-7.858639	-2.306735	-0.353629
8	-1.812088	-2.288102	-0.651807
6	-0.781247	0.305418	0.182565
6	0.358395	-0.517724	-0.018683
1	-0.582566	1.335106	0.479941
1	0.173181	-1.575010	-0.167856
6	1.660951	-0.072430	0.003244
6	2.007835	1.360630	-0.014773
6	1.382491	2.256486	-0.891825
6	2.994956	1.875965	0.847438
6	1.702323	3.610937	-0.909428
1	0.648632	1.881075	-1.599513
6	3.313147	3.220943	0.852001
1	3.502607	1.207004	1.536786
6	2.668854	4.103101	-0.028091
1	1.203941	4.264832	-1.616264
1	4.062159	3.617419	1.531017
6	2.760696	-1.048262	0.022545
6	4.000783	-0.764865	-0.583580
6	2.613130	-2.305953	0.627783
6	5.024609	-1.693406	-0.599600
1	4.150181	0.194813	-1.069411
6	3.634965	-3.247108	0.626341
1	1.685605	-2.546555	1.139418
6	4.852542	-2.946039	0.005729
1	5.972314	-1.473618	-1.082104
1	3.479476	-4.200049	1.119749
8	-5.401333	1.618348	0.729888
8	5.910074	-3.782520	-0.055032
8	3.054584	5.395148	0.045306
6	2.431556	6.325230	-0.822100
1	2.877869	7.293503	-0.593186
1	2.619446	6.078184	-1.873658
1	1.350379	6.372335	-0.646341
6	5.781052	-5.063644	0.535268
1	4.973819	-5.638873	0.066996
1	6.732043	-5.569000	0.363977
1	5.598203	-4.989322	1.613665
6	-4.991474	2.915802	1.112195
1	-4.371187	2.885178	2.016314
1	-5.905762	3.474215	1.316138
1	-4.432903	3.408464	0.306956

Cartesian coordinates (Å) for molecule **11** (excited-state) obtained by DFT [PCM-PBE0/6-31G(d)]

E(TD)= -1380.64377307au

6	-3.067820	1.000552	0.408841
6	-4.445955	0.791145	0.377769
6	-4.970650	-0.494356	0.036335
6	-4.054434	-1.541015	-0.260904
6	-2.597945	-1.347055	-0.223316
6	-2.127372	-0.005988	0.120347
1	-7.047608	0.057131	0.220388
1	-2.687993	1.986202	0.661295
6	-6.357119	-0.747904	-0.009959
6	-4.563413	-2.805189	-0.592988
6	-5.927465	-3.039789	-0.637093
6	-6.829884	-2.003447	-0.342681
1	-3.849837	-3.592988	-0.814786
1	-6.301445	-4.025765	-0.899696
1	-7.900043	-2.187739	-0.375953
8	-1.824317	-2.297201	-0.474083
6	-0.742824	0.359247	0.184059
6	0.349237	-0.437239	-0.100788
1	-0.556020	1.380407	0.515743
1	0.130840	-1.467162	-0.363354
6	1.711911	-0.038018	-0.046536
6	2.097792	1.368049	-0.044300
6	1.423247	2.324407	-0.829627
6	3.169455	1.838058	0.752072
6	1.782969	3.664865	-0.832064
1	0.620073	1.997720	-1.483466
6	3.528216	3.170434	0.769980
1	3.701257	1.138715	1.390781
6	2.840226	4.101449	-0.024349
1	1.246090	4.356991	-1.471471
1	4.337301	3.523730	1.402531
6	2.731514	-1.067182	-0.003149
6	4.050703	-0.839257	-0.473703
6	2.449536	-2.367347	0.476884
6	5.006727	-1.832006	-0.463322
1	4.307110	0.131517	-0.886598
6	3.403731	-3.372139	0.495717
1	1.463779	-2.584812	0.877048
6	4.697743	-3.112875	0.024217
1	6.006880	-1.649081	-0.845261
1	3.138226	-4.347571	0.888490
8	-5.354896	1.737694	0.653320
8	5.699459	-4.013492	-0.005708
8	3.267788	5.377595	0.058016
6	2.600331	6.357666	-0.717444
1	3.094761	7.303395	-0.493725
1	2.688782	6.145532	-1.789319
1	1.540811	6.426845	-0.444852
6	5.435414	-5.325395	0.460685
1	4.645295	-5.805011	-0.128533
1	6.366627	-5.879089	0.337849
1	5.151028	-5.321319	1.519325
6	-4.915234	3.040586	1.003331
1	-4.308470	3.019263	1.915192
1	-5.821191	3.619922	1.180432
1	-4.342350	3.494875	0.187501

Frontier molecular orbitals obtained for **10** and **11**



Vibrationally resolved spectrum (stick+convoluted) computed for **10** and **11**

We note that the spectra are more structured than experimentally, but this is only related to the choice of a rather narrow broadening Gaussian, allowing to see that the vibrational couplings are indeed similar for the two dyes, but that the absolute intensities (ε) are not.

